

Stable Rb-B compounds under high pressure

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As a frontier issue of physics and material, the structures and related properties of borides have been extensively investigated in fundamental science. The search for pressure-induced stable compounds has become a feasible approach to acquire borides that are inaccessible at atmospheric pressure. Combined with state-of-the-art swarm intelligence structure prediction and first-principles calculations, we systematically explored the Rb-B system and uncovered a series of unprecedented RbB, Rb₂B₃, RbB₃, RbB₆, RbB₈, and RbB₁₀ under high pressure. It is found that the catenation of boron evolves from linear chain to layered sheets, clusterlike units, and further to three-dimensional tunnel structures with increasing boron content. Among them, RbB₆ and RbB₈ are expected phonon-mediated superconductors with T_c of ~ 12 K and superhard material with a hardness of ~ 37 GPa at ambient pressure, respectively. Additionally, RbB₈ is a suitable precursor for obtaining the superconducting α -B₁₆ boron allotrope by removing Rb due to its better stability than isomorphous SrB₈. The current results provide insights into the design of unforeseen borides and illustrate intriguing B-B bonding features originating from Rb \rightarrow B charge transfer under pressures.

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I. INTRODUCTION

Boron-based materials often show fascinating structural complexity due to the electron-deficient nature of boron [1,2]. Heretofore, hundreds of binary metal borides have been reported and attracted wide attention because of their outstanding properties for potential applications. Among these borides, MgB₂ was found to be an archetypal Bardeen-Cooper-Schrieffer (BCS) superconductor with the highest superconducting critical temperature (T_c) of 39 K at ambient pressure [3], which has triggered a flurry of explorations on the superconducting borides [4]. To date, many metal boride superconductors have been synthesized [5–10] or predicted [11–16], including YB₆ (7.2 K) [5], OsB₂ (2.1 K) [6], RuB₂ (1.6 K) [6], FeB₄ (2.9 K) [7], YB₁₂ (4.5 K) [8], ZrB₁₂ (6 K) [8,9], MoB₂ (32 K) [10], CaB₂ (~ 50 K) [11], LaB₈ (14 or ~ 20 K) [12,13], BeB₆ (24 K) [14], MgB₆ (9.5 K) [15], and LiBC (65 K) [16]. In addition, some transition-metal borides have been suggested as superhard materials owing to their high valence-electron density and strong covalent bonds [17–19], such as ReB₂ [20], OsB₂ [21], WB₄ [22], CrB₄ [23], FeB₄ [7], ZrB₁₂ [24], Zr_{0.5}Y_{0.5}B₁₂ [8], YB₆ [25], ScB₃, and ScB₆ [26], most of whose hardness values have been measured to > 40 GPa. Furthermore, highly symmetrical cubic

MB₆ compounds ($M =$ alkaline-earth and rare-earth metals) also exhibit other compelling behaviors such as low work function (LaB₆ and NdB₆) [27,28], topological insulator state (SmB₆ and PuB₆) [29,30], and semiconductivity (CaB₆, SrB₆, and BaB₆) [31], making them of broad interest for industrial application.

Compared with the borides mentioned above, however, alkali metal borides are rarely investigated [32–37], which is mainly restricted by the difficult syntheses of these materials. Notably, theoretical simulations play a key role in accelerating the discovery of structures and broadening the understanding of traditional physics and chemistry knowledge. Many theoretical studies have been carried out on the structural stability and properties of alkali metal borides at ambient and high pressures. For instance, lithium borides, with a variety of stoichiometries, exhibit abundant boron motifs, i.e., from polyhedral boron framework to graphitelike sheets, zigzag chains, dimers, and eventually isolated B ions with increasing Li content [38,39]. A semiconducting $I2_12_12_1$ -Na₂B₃₀ structure was calculated to be more stable than the known *Imma*-Na₂B₃₀ and possess a Vickers hardness value of 37.4 GPa at ambient pressure [40]. The negative linear compressibility and superionicity of NaB₃ have been revealed in recent theoretical work [41]. Cubic NaB₆ and KB₆ have been reported to possess superconductivity with predicted T_c of 15.7 and 14.6 K, respectively [42]. More interestingly, the removal of metal elements from binary compounds that are predicted to be stable under high pressure is considered an efficient route to obtain boron allotropes. For example, based on the precursors *I4/mmm*-NaB₄, *Pm*-Na₂B₁₇, and *C2/m*-KB₄, three metastable boron allotropes *I4/mmm*-B₄, *Pm*-B₁₇, and two-dimensional *C2/m*-structured boron have been proposed as

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superconductors at ambient pressure with estimated T_c of 19.8, 15.4, and 17.9 K, respectively [43,44], which ignited a research boom in elemental superconductors.

In view of the above descriptions, alkali metal borides are expected to be a potential system with abundant structural types, providing a platform for searching for borides with appealing properties. Intuitively, rubidium (Rb) possesses a low electronegativity that should be conducive to transferring its electron to nonmetals (such as B) and forming corresponding compound(s). However, the relatively large atomic radius of Rb increases the distances between the B frameworks therein and weakens their interactions, resulting in no stable Rb-B compound at ambient pressure [45]. Accordingly, the Rb-B compounds are suggested to be stable at high pressure [45,46], which has been an effective tool for reducing atomic volumes and generating materials that are generally inaccessible under ambient conditions. Therefore, it remains of interest to further explore the crystal structures of compressed Rb-B compounds that may exhibit unanticipated physical properties and, equally, their important underlying mechanism.

In this paper, we have performed a theoretical investigation on the Rb-B binary system at a pressure range of 0–100 GPa. As a result, six stable stoichiometries of RbB, Rb₂B₃, RbB₃, RbB₆, RbB₈, and RbB₁₀ were uncovered through the swarm intelligence structure prediction method. Among all these predicted structures, five stoichiometric Rb-B phases (e.g., *P6₃/mmc*-RbB, *C2/m*-RbB₃, *Pm-3m*-RbB₆, *Cmmm*-RbB₆, and *Immm*-RbB₈) are dynamically stable at atmospheric pressure and exhibit appealing structures and bonding behaviors with different boron concentrations. Furthermore, electron-phonon coupling (EPC) calculations indicate that metastable *Pm-3m*-RbB₆ and *Immm*-RbB₈ are superconductors with the estimated T_c of 7.3–11.6 and 4.8–7.5 K; more simulations suggest that some Rb-B compounds with three-dimensional structures are potentially superhard materials, with the highest Vickers hardness value of 36.9 GPa found in *Immm*-RbB₈. It is encouraging that *Immm*-RbB₈ could be a more suitable precursor to experimentally obtain the *o*-B₁₆ boron allotrope at ambient pressure due to its superior stability. The present results expand the scope of research on the borides and showcase a feasible route to design unforeseen Rb-B compounds as potential functional materials.

II. COMPUTATIONAL METHODS

Our structural prediction approach is based on a global minimization of free energy surfaces merging *ab initio* total-energy calculations through Crystal structure AnaLYsis by Particle Swarm Optimization (CALYPSO) methodology as implemented in its same-name CALYPSO code [47–49], which has been benchmarked on various known systems with several successful structure predictions [50,51]. The search for low-energy crystalline structures of Rb_xB_y ($x = 1, y = 1–12; x = 2, y = 1, 3, 5, 7, \text{ and } 9$) was carried out using simulation cells with 1 and 2 f.u. at selected pressures of 0, 25, 50, and 100 GPa. In our simulations, each generation contained 40 structures, and the structures in the first generation were produced randomly. During structure evolution, 60% of structures in the first generation with lower enthalpies are selected to produce

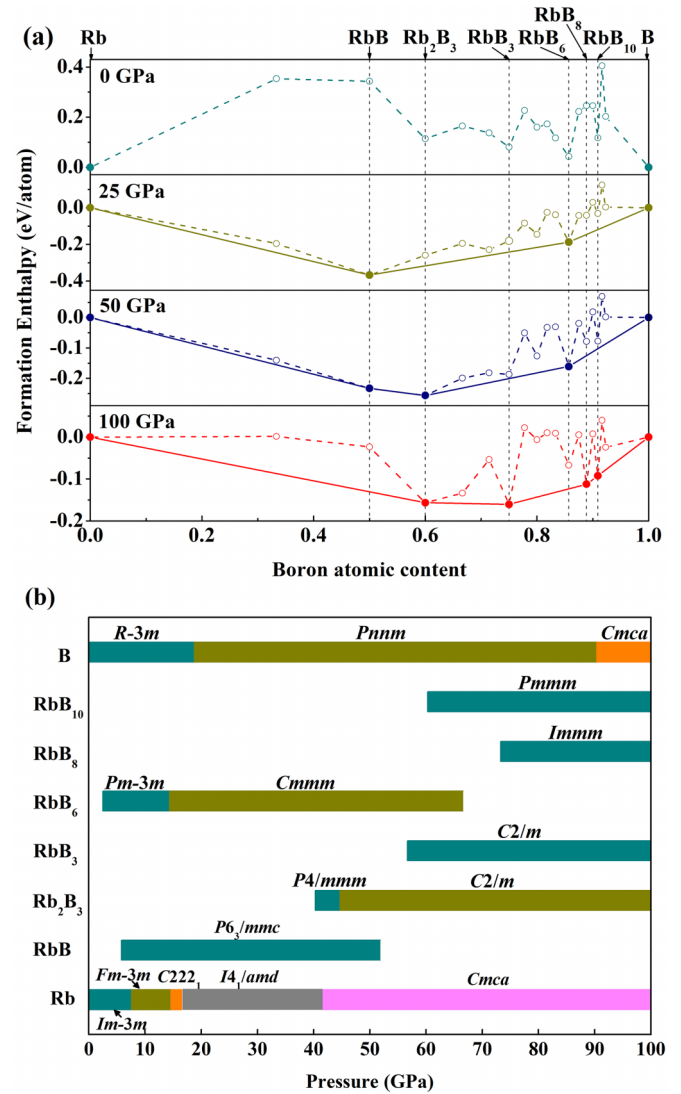


FIG. 1. (a) Thermodynamic stability of various Rb-B compounds with respect to elements Rb and B at selected pressures. The energetically stable stoichiometries are shown using solid symbols connected by solid lines on the convex hulls. (b) Pressure-composition phase stability diagram of the Rb-B system.

the structures in the next generation by the particle swarm optimization method, and 40% of the structures in the new generation are randomly generated. In most cases, structural searching simulations for each calculation are stopped after generating ~ 1000 structures (e.g., ~ 20 – 30 generations). The total number of predicted structures is at least $\sim 136\,000$ in the range of 0–100 GPa.

Our structural relaxations and electronic property calculations were performed within the Vienna *Ab initio* Simulation Package code within the framework of density functional theory [52]. The Perdew-Burke-Ernzerhof generalized gradient approximation [53] was chosen for the exchange-correlation functional. The ion-electron interaction is described by the projector augmented-wave (PAW) potential [54] with $4s^2 4p^6 5s^1$ and $2s^2 2p^1$ electrons as valence for Rb and B atoms, respectively. A plane-wave energy cutoff of 600 eV and Monkhorst-Pack k meshes with grid spacing of 0.20 \AA^{-1}

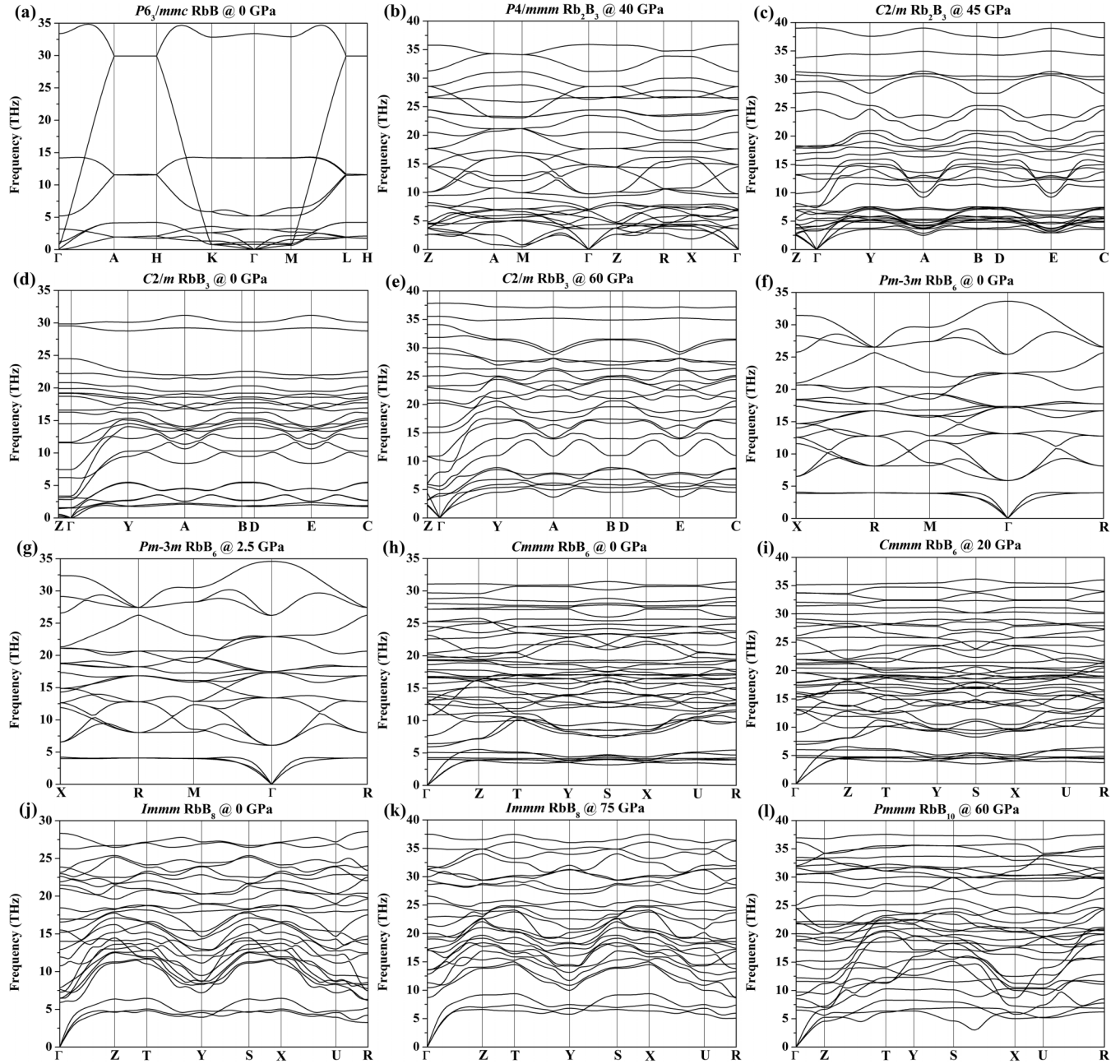


FIG. 2. Phonon dispersion curves of the predicted Rb-B phases.

for Brillouin zone sampling were employed to ensure total energy convergence to better than 1 meV/atom. The structures were fully relaxed until the energy and force were $< 10^{-6}$ eV and 0.01 eV/Å, respectively. Phonon calculations were carried out by using the linear response theory, which is implemented in the PHONOPY software [55].

The EPC calculations for superconducting properties were performed with density functional perturbation theory through the QUANTUM ESPRESSO code [56]. PAW pseudopotentials were used for Rb and B, with a kinetic energy cutoff of 80 Ry. The T_c was estimated by the standard Allen-Dynes modified McMillan equation [57]:

$$T_c = \frac{\omega_{\log}}{1.2} \exp \left[\frac{-1.04(1 + \lambda)}{\lambda(1 - 0.62\mu^*) - \mu^*} \right],$$

where λ , ω_{\log} , and μ^* represent the EPC strength, logarithmically averaged phonon frequency, and Coulomb repulsion coefficient, respectively. Vickers hardness was estimated by the microscopic hardness models [58,59].

III. RESULTS AND DISCUSSION

A. Structural stability

First, we investigated the thermodynamic stability of the binary Rb-B system by the calculation of their formation enthalpies relative to the dissociation products of Rb [60–63] and B [64] and constructed the convex hulls at pressures of 0, 25, 50, and 100 GPa, as shown in Fig. 1(a). In the view of energy, the thermodynamically stable structures are located on the solid lines of the convex hulls, while those on the dotted

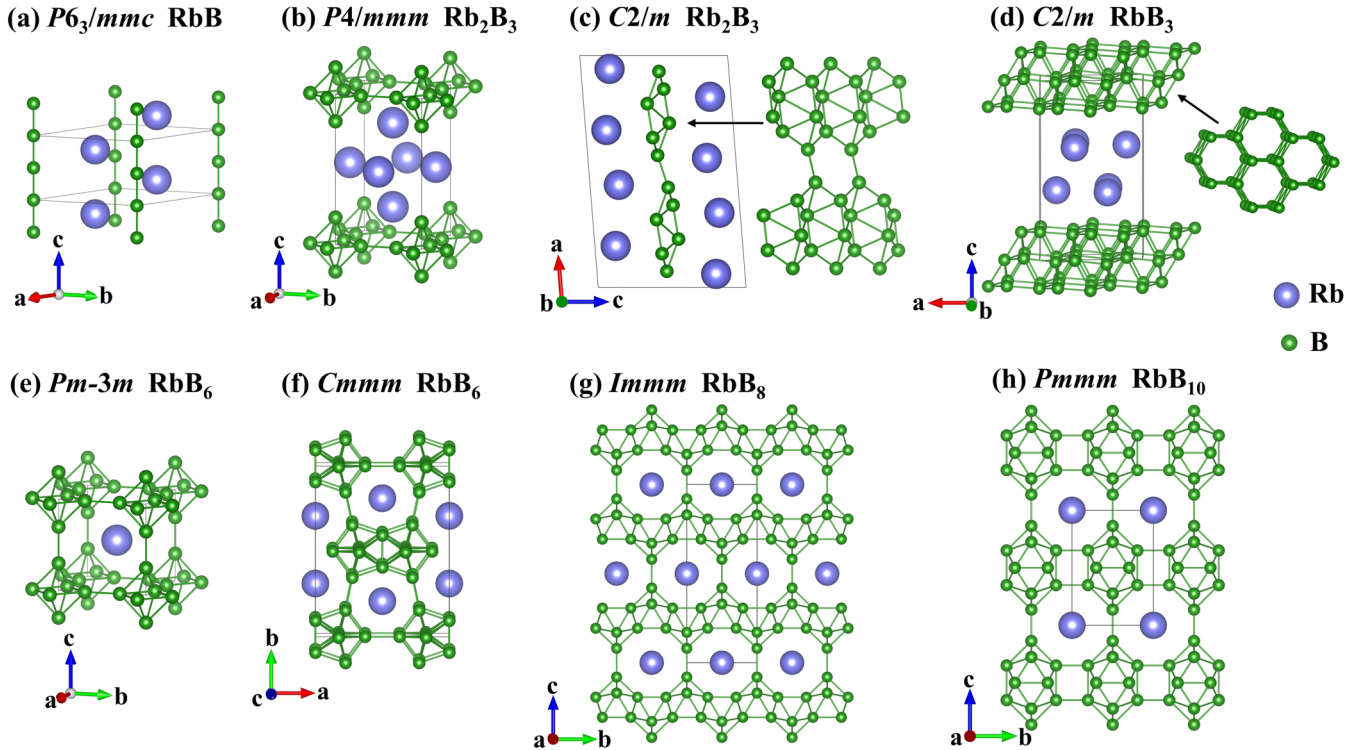


FIG. 3. Crystal structures of the Rb-B compounds under corresponding pressures. The Rb and B atoms are shown as big purple and small green spheres, respectively. (a) $P6_3/mmc$ -RbB at 6 GPa, (b) $P4/mmm$ -Rb₂B₃ at 40 GPa, (c) $C2/m$ -Rb₂B₃ at 45 GPa, (d) $C2/m$ -RbB₃ at 60 GPa, (e) $Pm-3m$ -RbB₆ at 2.5 GPa, (f) $Cmmm$ -RbB₆ at 15 GPa, (g) $Immm$ -RbB₈ at 75 GPa, and (h) $Pmmm$ -RbB₁₀ at 60 GPa.

lines are thermodynamically unstable and tend to decompose into elements or other energetically stable binary compounds. Our result that no stoichiometry is stable at zero pressure is consistent with the previous research [45,46]. At 25 GPa, two stable compounds of RbB and RbB₆ lay on the convex hull, and RbB has the most negative enthalpy of formation. With increasing pressure to 50 GPa, Rb₂B₃ becomes thermodynamically stable as well. Upon further compression, three additional unexpected stoichiometries of RbB₃, RbB₈, and RbB₁₀ appear, while RbB and RbB₆ are no longer stable. To accurately determine the pressure range of each stable Rb-B phase, we thus build the pressure-composition phase diagram, as shown in Fig. 1(b). The $P6_3/mmc$ structure of RbB is calculated to be stable between 5.8 and 52.0 GPa; $P4/mmm$ -Rb₂B₃ is predicted to be stable in a small pressure range between 40.3 and 44.7 GPa and transform into an energetically favored $C2/m$ structure > 44.7 GPa; and $C2/m$ -RbB₃ is stabilized at 56.7 GPa and remains stable to at least 100 GPa. Additionally, the cubic $Pm-3m$ phase of RbB₆ becomes stable > 2.5 GPa, then converts to a $Cmmm$ phase at 14.4 GPa, and finally dissociates into RbB₃ and RbB₈ at 66.7 GPa. For the stoichiometries with a higher B content, $Immm$ -RbB₈ and $Pmmm$ -RbB₁₀ are calculated to be stable > 73.3 and 60.3 GPa, respectively. Moreover, we have examined the stability of stoichiometric Rb₃B₂₀, which was previously reported to be stable at moderate pressure relative to the elements Rb and B [45]. In our simulations, however, Rb₃B₂₀ is unstable in the pressure range of 0–100 GPa once the abovementioned Rb-B compounds are considered.

We further performed phonon calculations to determine the dynamical stability of these proposed structures, as shown in Fig. 2. No imaginary phonon modes were found in the entire Brillouin zone of the studied structures, demonstrating these structures are dynamically stable under corresponding pressure. Notably, $P6_3/mmc$ -RbB, $C2/m$ -RbB₃, $Pm-3m$ -RbB₆, $Cmmm$ -RbB₆, and $Immm$ -RbB₈ are dynamically stable at 0 GPa, indicating that they are likely to be quenched to ambient pressure for potential applications.

B. Crystal structures

To fundamentally understand these unprecedented Rb-B compounds under high pressures, the evolution of their crystal structure features with increasing B content is presented in Fig. 3. Hexagonal $P6_3/mmc$ -RbB shows the same configuration as the known $P6_3/mmc$ -LiB and KB [38,39,44], where B atoms exhibit one-dimensional chains along the c axis with a B-B bond length of 1.59 Å at 6 GPa [Fig. 3(a)]. Rb₂B₃ adopts a tetragonal ($P4/mmm$) structure at low pressure, in which B₆ octahedra occupy the vertex positions. High-pressure Rb₂B₃ and RbB₃ are monoclinic ($C2/m$) structures, and both can be viewed as two-dimensional sandwichlike structures with alternating buckled graphene-like boron slabs and Rb layers [Figs. 3(c) and 3(d)]. The low-pressure phase of RbB₆ adopts a CsCl-like structure with $Pm-3m$ space group, in which the Rb atoms and B₆ octahedra replace the Cs and Cl sites. In this structure, the intraoctahedral and interoctahedral B-B distances are 1.80 and 1.70 Å at 2.5 GPa [Fig. 3(e)]. At elevated pressure, $Pm-3m$ -RbB₆ is calculated to transform

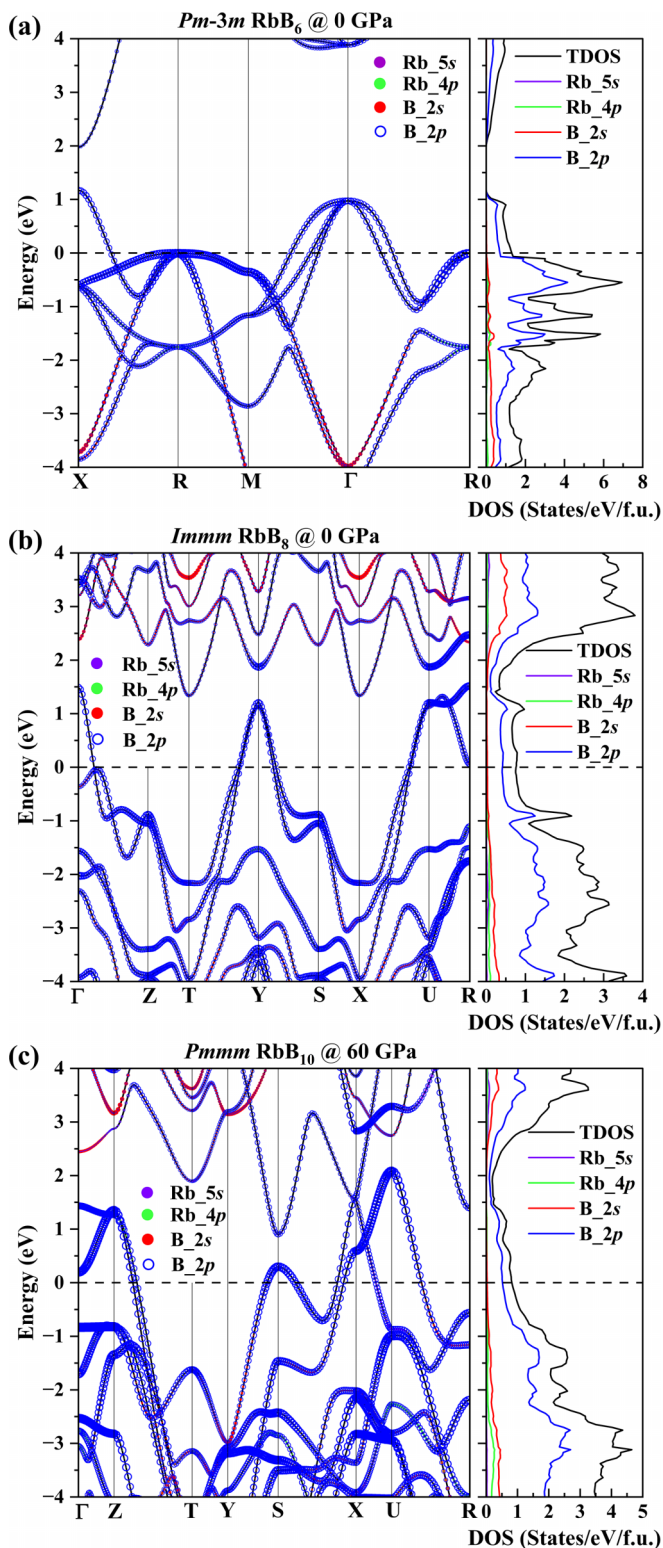


FIG. 4. Orbital-resolved band structures and density of states (DOS) for (a) $Pm\text{-}3m\text{-RbB}_6$ at 0 GPa, (b) $Immm\text{-RbB}_8$ at 0 GPa, and (c) $Pmmm\text{-RbB}_{10}$ at 60 GPa. The horizontal dashed lines indicate Fermi energy levels. The size of the circles and dots represents the weights of the corresponding orbitals.

into an orthorhombic ($Cmmm$) structure. In this structure, two decahedral B_7 clusters are gathered by sharing an edge, forming a B_{12} cluster with B-B bond lengths of 1.66–1.89 Å at 15 GPa, where Rb atoms occupy interstitial positions in the lattice [Fig. 3(f)]. B-rich stoichiometric RbB_8 possesses an orthorhombic structure with $Immm$ symmetry, which is isostructural to that of SrB_8 [65]. In this structure, B atoms form an open framework with B-B bond lengths of 1.65–1.85 Å at 75 GPa, while the Rb atom is located at the center of the boron framework [Fig. 3(g)]. B-richest stoichiometric RbB_{10} hosts an orthorhombic structure with $Pmmm$ symmetry, consisting of a three-dimensional B framework with B-B bond lengths of 1.59–1.83 Å at 60 GPa, forming open channels parallel to the a axis to capture the Rb atoms [Fig. 3(h)]. In this Rb-B system, the arrangement of B atoms evolves from one-dimensional chains to two-dimensional layers, finally becoming a three-dimensional tunnel configuration with increased B element concentration. More structure information of predicted structures is summarized in Table S1 in the Supplemental Material [66].

C. Electronic properties

To further dig into the nature of their bonding characteristics, the electron localization functions (ELFs) [67,68] were calculated. Generally, high ELF values (>0.7) correspond to localized electrons in core or bonding regions (covalent bonds) or lone pairs, whereas small values (<0.2) correspond to unshared electron interactions, such as ionic bonds or van der Waals bonds. The ELF values between 0.7 and 0.2 correspond to an electron localization like that of the electron-gas and indicate metallic bonds. As illustrated in Fig. S1 in the Supplemental Material [66], high ELF values are distributed between the neighboring B atoms, indicating that all B-B bonds are covalent. In contrast, no electron localization between Rb and B atoms demonstrates an ionic feature of the Rb-B bond, which is further evidenced by Bader charge analysis [69], where Rb apparently donates its electron (Table S2 in the Supplemental Material [66]). For all these stable Rb-B phases, each Rb atom transfers ~ 0.5 electrons to the B framework. Notably, for $C2/m\text{-RbB}_3$, $Pm\text{-}3m\text{-RbB}_6$, $Cmmm\text{-RbB}_6$, and $Immm\text{-RbB}_8$, some B atoms also transfer a low number of electrons (~ 0.1 e/atom) to other B atoms.

Since boron-rich metal borides usually have interesting electronic properties, we calculated the electronic band structures and projected density of states (PDOS) of these predicted Rb-B compounds, as shown in Fig. S2 in the Supplemental Material [66]. Except for semiconducting $Cmmm\text{-RbB}_6$, all the candidate structures exhibit metallic features with several bands crossing the Fermi level (Fig. S2 in the Supplemental Material [66]). $Cmmm\text{-RbB}_6$ exhibits semiconducting characteristics with a band gap of ~ 0.82 eV at 15 GPa (Fig. S2(f) in the Supplemental Material [66]). For $P6_3/mmc\text{-RbB}$, $P4/mmm\text{-Rb}_2B_3$, $C2/m\text{-Rb}_2B_3$, and $C2/m\text{-RbB}_3$, the densities of states (DOSs) at the Fermi level are dominated by the Rb atom, which is not beneficial to superconductivity; while for the more B-rich $Pm\text{-}3m\text{-RbB}_6$, $Immm\text{-RbB}_8$, and $Pmmm\text{-RbB}_{10}$, the contribution to the total DOS is mainly from B atoms, and the contribution from Rb atoms is negligible (Figs. S2(e), S2(g), and S2(h) in the Supplemental Material

TABLE I. Superconducting parameters of various Rb-B compounds.

Compounds	Space group	Pressure (GPa)	λ	ω_{log} (K)	N_{Ef} (states/spin/Ry/f.u.)	T_c (K) $\mu^* = 0.1$	T_c (K) $\mu^* = 0.13$
RbB ₆	<i>Pm-3m</i>	0	0.52	797	12.1	11.6	7.3
RbB ₆	<i>Pm-3m</i>	2.5	0.52	806	11.9	11.2	7.0
RbB ₆	<i>Pm-3m</i>	14	0.51	826	11.4	10.8	6.7
RbB ₈	<i>Immm</i>	0	0.54	464	5.2	7.5	4.8
RbB ₈	<i>Immm</i>	30	0.41	604	4.7	2.9	1.3
RbB ₈	<i>Immm</i>	75	0.31	783	4.4	0.5	0.1
RbB ₁₀	<i>Pmmm</i>	60	0.40	514	5.6	2.0	0.9

[66]). This fact suggests that these more B-rich phases have the potential to exhibit high- T_c superconductivity, especially *Pm-3m*-RbB₆, which has a total DOS approximately twice as large as that of *Immm*-RbB₈ or *Pmmm*-RbB₁₀. Additionally, orbital-resolved band structures and DOSs for *Pm-3m*-RbB₆ at 0 GPa, *Immm*-RbB₈ at 0 GPa, and *Pmmm*-RbB₁₀ at 60 GPa are also shown in Fig. 4. It can be clearly seen that the DOS near the Fermi level for these three structures is mainly from the contribution of B-2*p* orbitals.

D. Superconductivity

Motivated by the appearance of boron-dominated electronic DOS at the Fermi energy level for predicted *Pm-3m*-RbB₆, *Immm*-RbB₈, and *Pmmm*-RbB₁₀, we subsequently explored the superconductivity of these structures at different pressures through the Allen-Dynes modified McMillan equation using typical Coulomb pseudopotential parameters (μ^*) from 0.1 to 0.13 [57], as shown in Table I. As expected, the estimated T_c of *Pm-3m*-RbB₆ is ~ 11.6 (7.3) K with μ^* of 0.1 (0.13) at 0 GPa, which is the highest value in the Rb-B system. Although *Immm*-RbB₈ and *Pmmm*-RbB₁₀ have higher B content than *Pm-3m*-RbB₆, the fewer electrons near the Fermi energy level cannot be effectively coupled with the lattice vibrations, resulting in both possessing relatively low superconductivity.

The phonon dispersions, projected phonon DOS (PHDOS), Eliashberg spectral function $\alpha^2 F(\omega)/(\omega)$, and the integrated EPC strength $\lambda(\omega)$ for *Pm-3m*-RbB₆ and *Immm*-RbB₈ at 0 GPa are shown in Fig. 5. For *Pm-3m*-RbB₆ [Fig. 5(a)], the phonon dispersion curves and PHDOS show that frequencies < 5 THz are dominated by Rb due to its large mass, while the high-frequency part is almost entirely driven by the vibrations of B atoms. It is noteworthy that the high-frequency part contributes $\sim 97\%$ to the total λ and mainly originates from the vibrations of B atoms. There are several strongly coupled phonon modes at 12–15 and 25–27 THz along the *X-R-M* path and around the Γ points, which contribute ~ 13 and $\sim 4\%$ to the total λ , respectively. More obviously, the acoustic modes from 15 to 20 THz along the entire Brillouin zone make great contributions to the EPC, which is also reflected in the Eliashberg spectral function $\alpha^2 F(\omega)/(\omega)$, corresponding to a significant amount ($\sim 46\%$) of the total λ , and thus result in the relatively high superconducting T_c (11.6 K for $\mu^* = 0.1$) of *Pm-3m*-RbB₆. To probe the underlying causes, we further analyze several representative vibrational modes with relatively large EPC in this frequency range (e.g., E_g , B_{2g} , and T_{2g} modes at *X*, *M*, and Γ points, respectively; see Fig. S3 in the Supplemental Material [66]). These modes are all composed of bending vibrations of the diagonal atoms in the B₆ octahedron. Considering the dominant DOS of B-2*p*

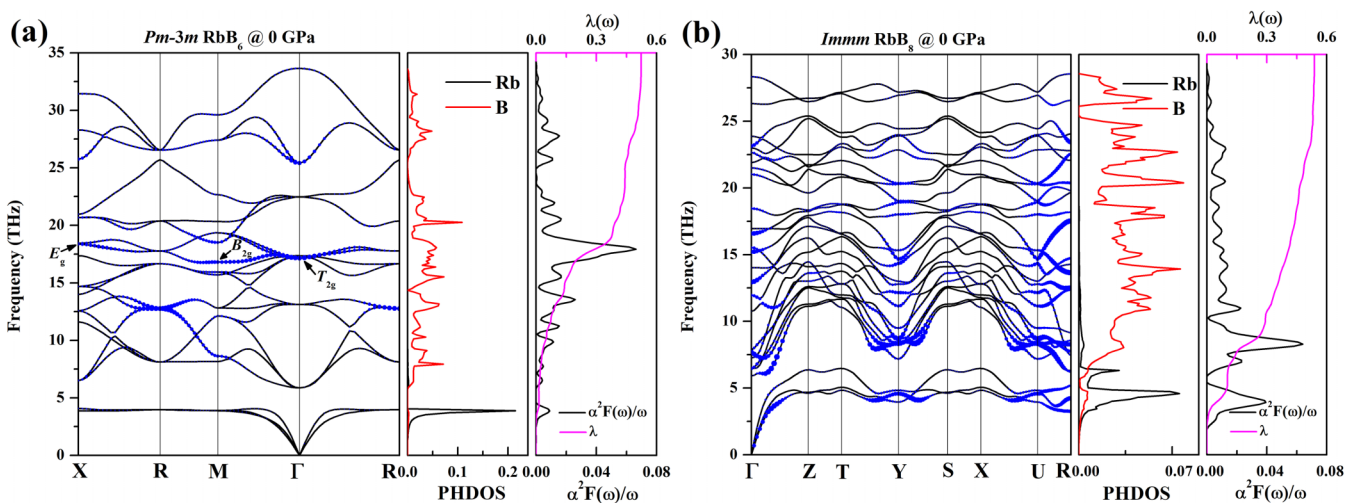


FIG. 5. Phonon dispersion relations, projected phonon density of states (PHDOS), Eliashberg function $\alpha^2 F(\omega)/(\omega)$, and integrated electron-phonon coupling (EPC) strength $\lambda(\omega)$ of (a) *Pm-3m*-RbB₆ and (b) *Immm*-RbB₈ at 0 GPa. The size of the blue dots represents the magnitude of the EPC. Three representative strongly coupled phonon modes at 15–20 THz are marked with arrows in (a).

TABLE II. Hardness of various Rb-B compounds at ambient pressure.

Compounds	Space group	Hardness (GPa)	
		$H_{v, \text{Chen}}$	$H_{v, \text{Tian}}$
RbB ₆	<i>Pm-3m</i>	19.7	18.6
RbB ₆	<i>Cmmm</i>	31.1	28.7
RbB ₈	<i>Immm</i>	36.9	34.4

orbitals at the Fermi level, we therefore conclude that the superconductivity of *Pm-3m*-RbB₆ mainly originates from the coupling between the electrons of B-2*p* orbitals and bending vibrations of the diagonal atoms in B₆ octahedron. For *Immm*-RbB₈ [Fig. 5(b)], coupling of electrons to phonons mainly originates from high-frequency vibrations in the range of 6–30 THz, which contribute 81% to the total λ . These frequencies are dominated by the vibrations of the B atoms, as in *Pm-3m*-RbB₆. The resulting T_c reaches a maximum value of 7.5 K using $\mu^* = 0.1$ at 0 GPa. Furthermore, the effects of pressure on the T_c were examined for *Pm-3m*-RbB₆ and *Immm*-RbB₈. The results show a decrease in T_c from 11.6 (7.5) K at 0 GPa to 10.8 (0.5) K at 14 (75) GPa for *Pm-3m*-RbB₆ (*Immm*-RbB₈) with a Coulomb potential μ^* of 0.1, which is attributed to the lattice hardening and the decrease of the total DOS at Fermi level with increasing pressure (see Table I).

E. Hardness

Because of the strong covalent B-B bonding, the Rb-B compounds are expected to possess high hardness. We first evaluated the mechanical properties of the predicted B-rich phases with the Voigt-Reuss-Hill approximation [70] at 0 GPa. Based on the empirical Vickers hardness methods [58,59], we calculated the hardness of these three B-rich phases, as listed in Table II. *Immm*-RbB₈ is a candidate for hard material with the highest hardness value in the Rb-B system. Its estimated Vickers hardness value is 36.9 GPa,

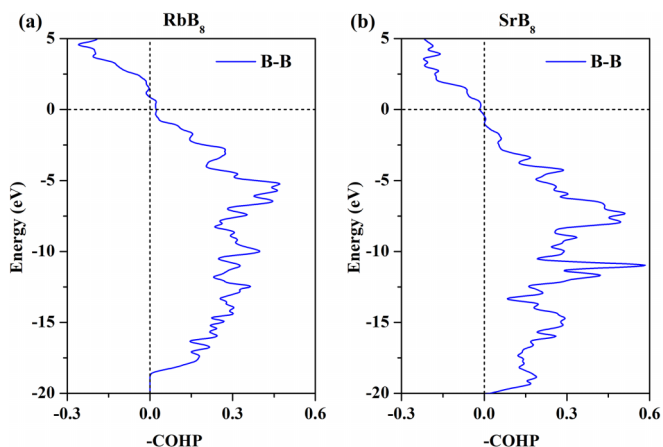


FIG. 6. Weighted average crystal orbital Hamiltonian population (COHP) of the seven B-B bonds in (a) RbB₈ and (b) SrB₈ at 75 GPa. The values of COHP > 0 and < 0 signify bonding states and antibonding states, respectively.

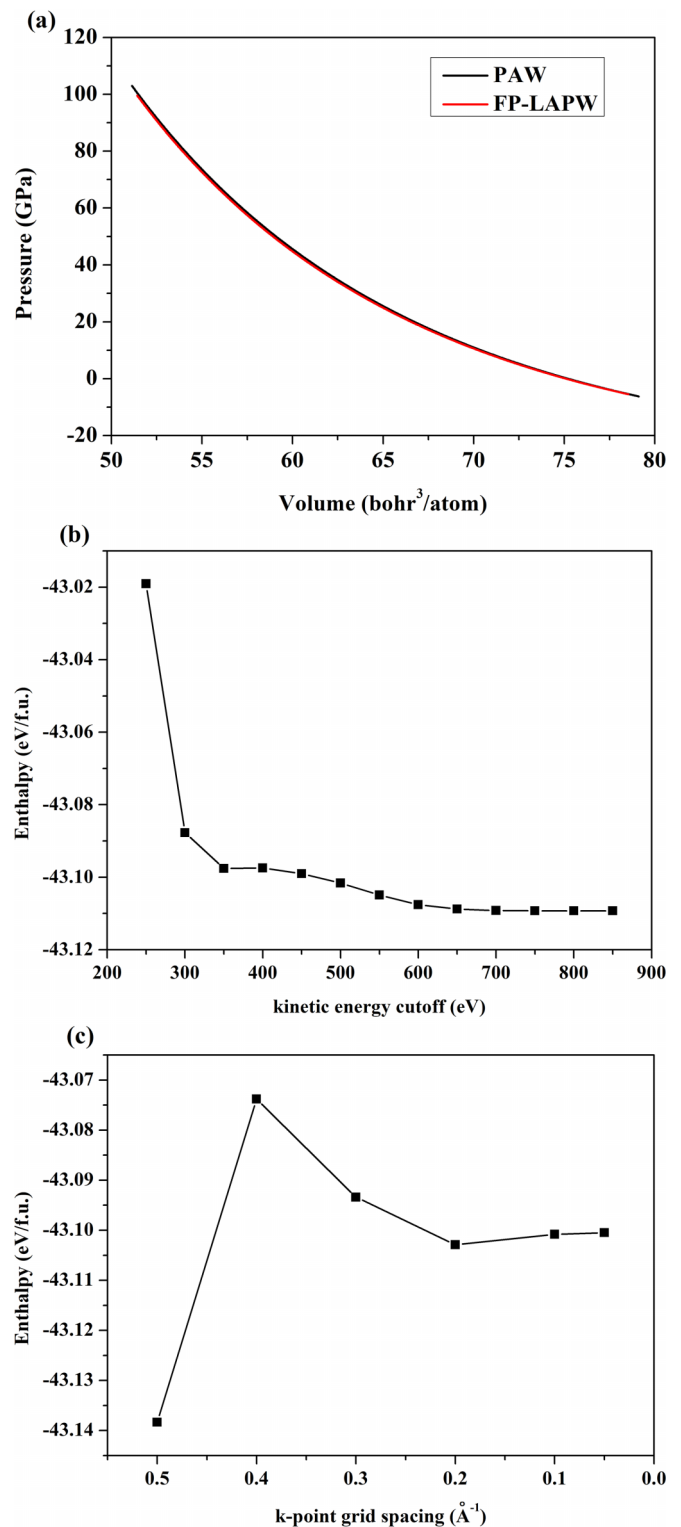


FIG. 7. (a) Comparison of the fitted Birch-Murnaghan equations of state for RbB₆ with *Pm-3m* symmetry by using the calculated results from the projector augmented-wave (PAW) pseudopotentials and the full-potential linearization augmented plane wave (FP-LAPW) methods. Convergence test of total energy relative to (b) plane-wave cutoff energy and (c) k -point grid spacing for *Pm-3m* RbB₆ at 0 GPa.

TABLE III. Calculated elastic constants C_{ij} (GPa), bulk modulus B (GPa), and shear modulus G (GPa), for $Pm-3m$ -RbB₆, $Cmmm$ -RbB₆, and $Immm$ -RbB₈ at 0 GPa.

Phases	C_{11}	C_{22}	C_{33}	C_{44}	C_{55}	C_{66}	C_{12}	C_{13}	C_{23}	B	G
$Pm-3m$ RbB ₆	300.7	300.7	300.6	90.4	90.4	90.4	47.8	47.8	47.8	132.1	103.4
$Cmmm$ RbB ₆	339.2	330.7	321.5	117.4	123.4	153.6	46.8	43.9	42.7	139.7	135.4
$Immm$ RbB ₈	405.5	461.8	305.2	180.9	188.5	144.5	6.1	77.1	81.5	166.5	166.6

which is comparable with that of $I2_12_12_1$ -Na₂B₃₀ (37.4 GPa) [40] and $R3m$ -YB₆ (37 GPa) [25] and slightly lower than that of ZrB₁₂ (40 GPa) [24] and WB₄ (\sim 43 GPa) [22] at the load of 0.49 N.

F. Boron allotrope from RbB₈

Following the scheme of synthesis under high pressure and removal of metal atoms from precursors at ambient pressure, some allotropes have been successfully synthesized [71,72] or predicted [43,44,65,73]. Among them, an α -B₁₆ boron allotrope may be available by removing the Sr atoms in channel-like SrB₈ [65]. However, since our calculation shows that SrB₈ is dynamically unstable at ambient pressure (Fig. S4 in the Supplemental Material [66]), it is still a challenge to obtain α -B₁₆ boron allotrope from SrB₈ precursor. Here, we found that RbB₈, sharing the same boron structure as SrB₈, is not only dynamically stable at ambient pressure but also possesses superior thermodynamic stability. To further understand its stability mechanism, we used the crystal orbital Hamilton population (COHP) [74] to analyze the bonding character of the states around the Fermi level of RbB₈ and SrB₈. The COHP analysis aims to measure weighted population of wave functions on two atomic orbitals of a pair of selected atoms, which is implemented in the LOBSTER-4.0.0 program [75]. The values of COHP > 0 and < 0 signify bonding states and antibonding states, respectively. The COHPs in Fig. 6 show that the antibonding character of occupied states around the Fermi level is unambiguous between B-B atoms in SrB₈, owing to Sr donating more electrons to occupy the antibond orbit. For Rb that has only one valence electron, no occupied antibonding states are exhibited in RbB₈, which results in a decrease in energy of the system and is a more suitable precursor for the synthesis of the α -B₁₆ boron allotrope.

IV. CONCLUSIONS

We have systematically studied the crystal structures and physical properties of the Rb-B compounds at the pressure range of 0–100 GPa using the swarm intelligence structure search method combined with first-principles calculations. Six thermodynamically stable phases of RbB, Rb₂B₃, RbB₃, RbB₆, RbB₈, and RbB₁₀ have been established. Among these predicted structures, $Pm-3m$ -RbB₆ and $Immm$ -RbB₈ exhibit superconductivity with estimated T_c of 11.6 and 7.5 K at ambient pressure. Further analysis of electronic and phonon properties indicates that the superconducting mainly originates from the coupling between the B-2*p* orbitals and the high-frequency vibrations of the B atoms. In addition, the predicted Vickers hardness (36.9 GPa) indicates that $Immm$ -RbB₈ is a potential hard material. Encouragingly, given the

robust stability of $Immm$ -RbB₈, it is a more suitable precursor to obtain the α -B₁₆ boron allotrope experimentally. The current findings not only deepen the understanding of the high-pressure behavior of the alkali metal borides family but also stimulate future experimental and theoretical investigations on the Rb-B compounds.

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APPENDIX

The full-potential linearization augmented plane wave (FP-LAPW) method as implemented in the WIEN2K code [76] was used to check the equation of state for $Pm-3m$ -RbB₆. The result of FP-LAPW is highly consistent with that of pseudopotential calculation, as shown in Fig. 7(a), confirming the reliability of the adopted pseudopotentials in this paper. To ensure good convergence of total energy, we have performed the convergence test of total energy relative to plane-wave cutoff energy and k -point grid spacing for $Pm-3m$ RbB₆ at 0 GPa, as shown in Figs. 7(b) and 7(c). The plane-wave cutoff energy of 600 eV and Monkhorst-Pack k meshes with grid spacing of 0.20 \AA^{-1} are sufficient to ensure good convergence of total energy.

We have evaluated the mechanical properties of the predicted B-rich phases with the Voigt-Reuss-Hill approximation [70] at 0 GPa. The results in Table III show that $Pm-3m$ -RbB₆, $Cmmm$ -RbB₆, and $Immm$ -RbB₈ are all mechanically stable.

The cubic $Pm-3m$ -RbB₆ structure satisfies the following mechanical stability criteria:

$$C_{11} > 0, C_{44} > 0, C_{11} > |C_{12}|, (C_{11} + 2C_{12}) > 0.$$

The orthorhombic $Cmmm$ -RbB₆ and $Immm$ -RbB₈ structures satisfy the following mechanical stability criteria:

$$C_{11} > 0, C_{22} > 0, C_{33} > 0, C_{44} > 0, C_{55} > 0, C_{66} > 0,$$

$$[C_{11} + C_{22} + C_{33} + 2(C_{12} + C_{13} + C_{23})] > 0,$$

$$(C_{11} + C_{22} - 2C_{12}) > 0,$$

$$(C_{11} + C_{33} - 2C_{13}) > 0,$$

$$(C_{22} + C_{33} - 2C_{23}) > 0.$$

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