





Metal-organic kagome systems as candidates to study spin liquids, spin ice or the quantum anomalous Hall effect

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We present the results of first-principle calculations using the Vienna *Ab initio* Simulation Package (VASP) for a class of organometallics labeled $TM_3C_6O_6$ ($TM = Sc, Ti, V, Cr, Fe, Co, Ni,$ and Cu) in the form of planar, two-dimensional, periodic freestanding layers. These materials, which can be produced by on-surface coordination on metallic surfaces, have a kagome lattice of TM ions. Calculating the structural properties, we show that all considered materials have local magnetic moments in the ground state, but four of them (with $Fe, Co, Ni,$ and Cu) show spin-crossover behavior or switch between magnetic and nonmagnetic states by changing the lattice constant, which could be valuable for possible epitaxy routes on various substrates. Surprisingly, we find a very large richness of electronic and magnetic properties, qualifying these materials as highly promising metal-organic topological quantum materials. We find semiconductors with nearest-neighbor ferromagnetic (FM) or antiferromagnetic (AFM) couplings for $V,$ and $Sc, Ti,$ and $Cr,$ respectively, being of potential interest to study spin ice or spin liquids on the 2D kagome lattice. Other TM ion systems combine AFM couplings with metallic behavior (Fe and Ni) or are ferromagnetic kagome metals like $Cu_3C_6O_6$ with band crossings at the Fermi surface. For the latter compound, the spin-orbit coupling is shown to be responsible for small gaps which makes them a candidate material to observe the quantum anomalous Hall effect.

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I. INTRODUCTION

The kagome Heisenberg antiferromagnet is a well-known paradigm of highly frustrated magnetism [1–5]. Furthermore, the electronic and transport properties on the kagome lattices have attracted enormous interest recently, since they are characterized by symmetry-protected band-crossing points similar to the Dirac point in graphene [6]. So, the kagome systems are promising representations for topological quantum materials that became particularly clear with the recent discovery of ferromagnetic kagome metals [7]. Together with spin-orbit coupling one expects topological [8] and Chern insulating phases, which are interesting for the quantum anomalous Hall effect (QAHE) [9]. These exotic states can be traced back to several peculiarities of the kagome lattice consisting of corner-sharing triangles, namely frustration, and the possibilities for Dirac points and of local excitations of magnetic or electronic nature leading to dispersionless (flat) excitation branches. Such flat-band systems have attracted much attention in several areas of physics, and many interesting phenomena that are related to flat bands have been found, see, e.g., the reviews Refs. [10–13]. Interestingly, for partial filling of a nearly dispersionless band one may expect fractional topological quantum states [14,15] in analogy to the fractional quantum Hall effect. It is remarkable that also the

antiferromagnetic state in the kagome compounds Mn_3Ge or Mn_3Sn allows the anomalous Hall effect [16–19].

Whereas numerous topological quantum materials were already found and investigated in the class of inorganic layered crystals, there is much less progress for organometallic lattices despite existing proposals [20]. However, the kagome systems $Mn_3C_6O_6$ and $Cu_3C_6O_6$, as well as $Fe_3C_6O_6$, have recently been synthesized by on-surface coordination reaction on noble-metal surfaces [21–23]. These metal-organic coordination networks consist of C_6O_6 rings and a dense kagome lattice of $3d$ transition metal (TM) ions. In the present theoretical study, we extend the search of exotic systems to all $3d$ TM ions going from Sc to Cu . We study here freestanding monolayers, first as reference systems for the adsorbed case. But, most of them are also thermodynamically and kinetically stable without substrate [24]. It is promising that we found good candidates in the material class of $TM_3C_6O_6$ for realizations of several unconventional kagome systems.

As already mentioned, the kagome Heisenberg antiferromagnet is highly frustrated. In the classical limit its ground state is massively degenerated, i.e., it has an extensive ground-state manifold [25]. Quantum fluctuations may select coplanar ground states [26,27]. In the extreme quantum limit (spin quantum numbers $S = 1/2$ and $S = 1$) the ground state is magnetically disordered. Although there is a plethora of theoretical studies for $S = 1/2$, see, e.g., Refs. [28,29] and references therein, the nature of the quantum ground state is still under debate. Favored candidates are a gapless $U(1)$

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Dirac spin liquid [30,31] and a gapped Z2 spin liquid [32,33]. For $S = 1$ there are many fewer studies available; however, the absence of magnetic order seems to be well established; see, e.g., Refs. [34–37] and references therein. Candidates for the ground states without magnetic long-range order are a chiral spin liquid, a hexagonal-singlet solid, or a trimerized state [35–37]. For larger spin $S > 1$ there are indications for a magnetically ordered ground state [34,38,39].

In addition to the extensively discussed nature of the spin-liquid ground state, the intriguing magnetization process of the kagome Heisenberg antiferromagnet has attracted much attention. For $S = 1/2$ magnetization exhibits plateaus at $1/3$, $5/9$, and $7/9$ of the saturation magnetization [40,41] and a macroscopic jump at the saturation field due to the very existence of a flat one-magnon band [42]. The $1/3$ plateau as well as the jump at the saturation are also present for $S > 1/2$ but both shrink with increasing S [42,43]. A spectacular feature of the quantum kagome Heisenberg antiferromagnet is the appearance of a magnon crystal phase just below the saturation field [44,45].

The above-outlined theoretical predictions for kagome Heisenberg antiferromagnet at zero and finite magnetic fields go hand in hand with numerous experimental studies; see Refs. [4,46] for an overview. Among many candidate materials Herbertsmithite is a near-perfect $S = 1/2$ kagome Heisenberg antiferromagnet compound showing characteristic features of a spin liquid [2,47]. There are also experimental indications for the plateaus and the magnon crystallization [48,49] predicted by theory. Turning to ferromagnetic (FM) nearest-neighbor interactions, they are interesting as well. Classical spins on the kagome lattice with a FM nearest-neighbor interaction and site-dependent single-site anisotropy have a residual entropy at zero temperature and, therefore, they are called spin ice [50].

In the present study we investigate freestanding $TM_3C_6O_6$ monolayers by *ab initio* band-structure calculations taking into account the Coulomb interaction in the $3d$ shell and clarify its structural, electronic, and magnetic properties. Most importantly, we found candidates for realizations of very different exotic kagome systems: the semiconducting Sc system with antiferromagnetic (AFM) interactions between spins $1/2$ allowing the spin liquid state, several AFM kagome systems with different values of the local spin going from $S = 1/2$ to $S = 5/2$ with the exception of $S = 3/2$ (V system) for which the interaction is FM and could give rise to spin ice, and finally $Cu_3C_6O_6$, which is predicted to be a FM kagome metal and is a candidate for the QAHE.

We also find spin-crossover (SCO) complexes and transitions between magnetic and nonmagnetic states in the studied material class. These SCO complexes [51–58] have gathered considerable attention for their potential use as intrinsic switch to build nanoscale electronic components. Indeed, in our case, they are composed of a central transition-metal ion surrounded by C_6O_6 units whose spin state can be potentially switched by applying external stimuli such as temperature, light, pressure, magnetic or electric fields, or current [59–70]. In this paper, we theoretically explore the synergy between the two research fields (SCO and topological quantum materials) by studying the spin properties of the $TM_3C_6O_6$ structures. The results uncover $TM_3C_6O_6$ as a fascinating material class

with a surprisingly rich behavior mentioned above. That is outlined below based on our calculations of (i) the total energies for different lattice constants and spins states, (ii) the various densities of states, (iii) the total and local magnetic moments and the magnetic couplings, and (iv) the band structures without and with spin-orbit (SO) coupling.

II. METHOD

We used the Vienna *Ab initio* Simulation Package (VASP) [71] at the level of the spin-polarized generalized-gradient approximation (SGGA) in the form of the Perdew, Burke, and Ernzerhof functional. The SGGA functional is used to investigate the structural properties (lattice constants, atomic positions, and spin-crossover transitions). It is well known, however, that the standard density-functional theory (DFT) has difficulties to describe correctly the electronic density of states, including the gap values, especially for $3d$ transition-metal ions. For that reason, we apply here also the SGGA with a Hubbard term U (SGGA+ U). The SGGA+ U corrections were introduced by Liechtenstein *et al.* [72]. We take in the following $U = 5$ eV and an exchange energy of $J = 0.90$ eV. The necessity of the SGGA+ U method for metal-organic compounds with transition-metal ions is proven by many examples such as *TM*-TCNB [73,74] or *TM*-TCNQ [75] *TM*-ZQ [76–80], where *TM* is a transition metal, and the organic molecules are tetracyanobenzene (TCNB), tetracyanoquinone (TCNQ), and zwitterionic quinone (ZQ).

The interaction between the valence electrons and ionic cores was described within the framework of the projector augmented-wave (PAW) method [81–84]. The electronic wave functions were expanded in plane waves with a kinetic energy cutoff of 480 eV; the monolayers were relaxed and the convergence criterion for the energy deviations was 10^{-7} eV. The Gaussian smearing method was used in these calculations and a width of $\sigma = 0.01$ eV was adopted in most calculations. The Brillouin zone was usually determined by a set of $6 \times 6 \times 1$ k points in the unit cell using the Monkhorst-Pack points [85]. The geometry optimization was performed without spin-orbit coupling. But, a relativistic calculation was used for $Cu_3C_6O_6$ to investigate the possibility of a QAHE state. That high-precision band-structure calculation was performed with $18 \times 18 \times 1$ k points and $\sigma = 0.001$ eV.

A. Structural properties

The first step in our calculations is to perform the structural optimization of the $TM_3C_6O_6$ networks studied by calculating the total energy as a function of lattice constant a and to determine the fundamental state. For that purpose, we use the SGGA functional, but the influence of the Hubbard U correction on the structural properties is only small. The iteration process is repeated until the calculation of the total energy converges. Figure 1 shows the total energies for the FM state and the nonmagnetic (NM) configurations for those situations where we found a SCO situation, i.e., for Fe and Ni, or a transition between magnetic and nonmagnetic states like for Co and Cu. Since the Hubbard U term has no influence for a nonmagnetic state, we have to use the SGGA method (without U) to compare magnetic and nonmagnetic states in Fig. 1.

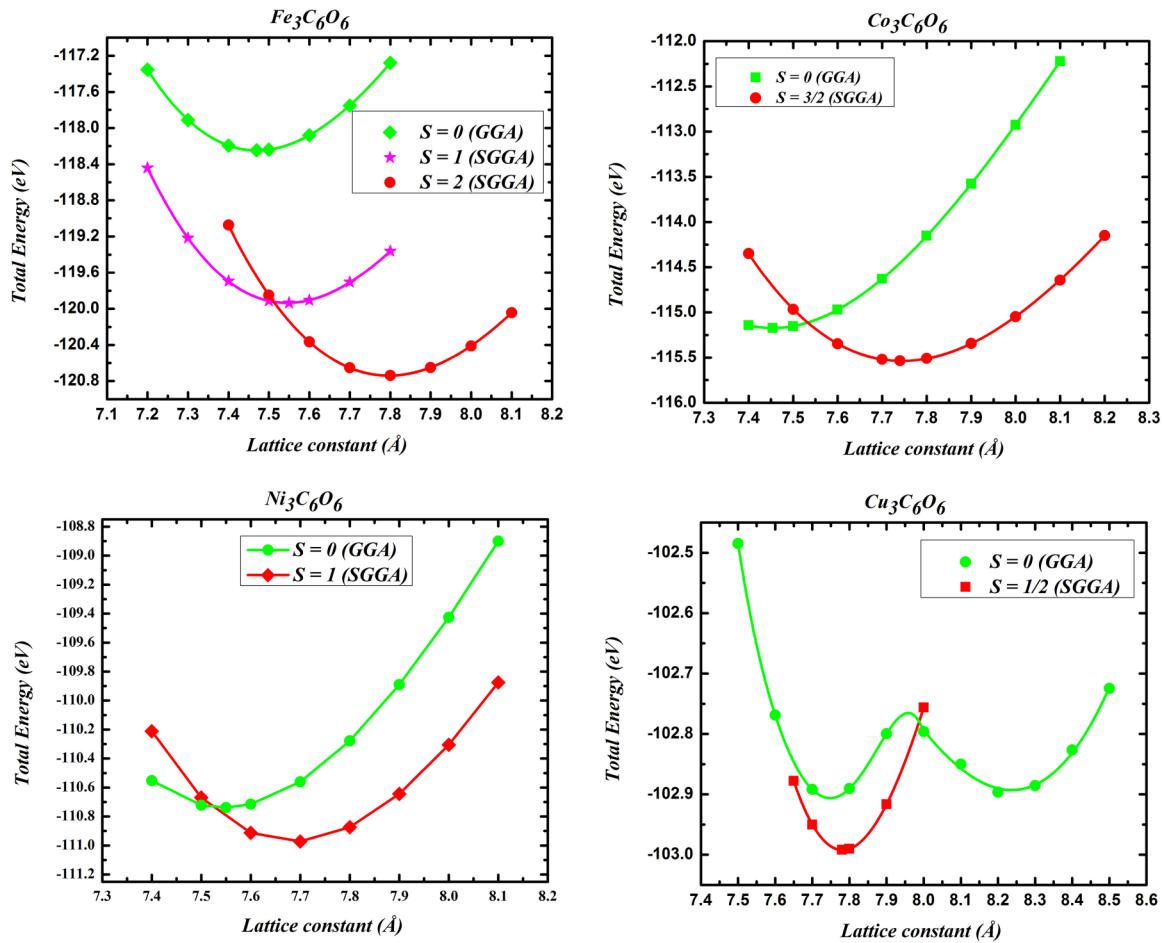


FIG. 1. Total energy vs lattice constant for freestanding TM₃C₆O₆ structures calculated at DFT level with (SGGA) and without (GGA) magnetic state *S* of *TM* atom (Fe, Co, Ni, and Cu).

The crystal structure of TM₃C₆O₆ is shown in Fig. 2(a). The gray, red, and green spheres represent carbon, oxygen, and transition-metal atoms, respectively. The lattice constant is optimized and the resulting lattice constants are presented in Table I including the *U* correction, and in Appendix A without *U*. Also, the metal-oxygen distances are listed there and one can see that they remain nearly constant throughout the 3*d* series. We find two compounds, namely Co₃C₆O₆ and Cu₃C₆O₆, with two minima: a nonmagnetic state at the equi-

librium lattice constant of 7.45 Å (8.25 Å) for the Co (Cu-) system, and a magnetic state with *S* = 3/2 per Co (*S* = 1/2 per Cu) by variation of the lattice constant. The magnetic state has an energy minimum for a lattice constant of 7.74 Å (7.78 Å). There are transitions between states of different spin for Fe between *S* = 2 and *S* = 1 and for Ni between *S* = 1 and *S* = 0, which one can qualify as SCO transitions due to the competition between crystal field and Coulomb energies. Notice that the crystal structure of nonmagnetic Cu₃C₆O₆ at

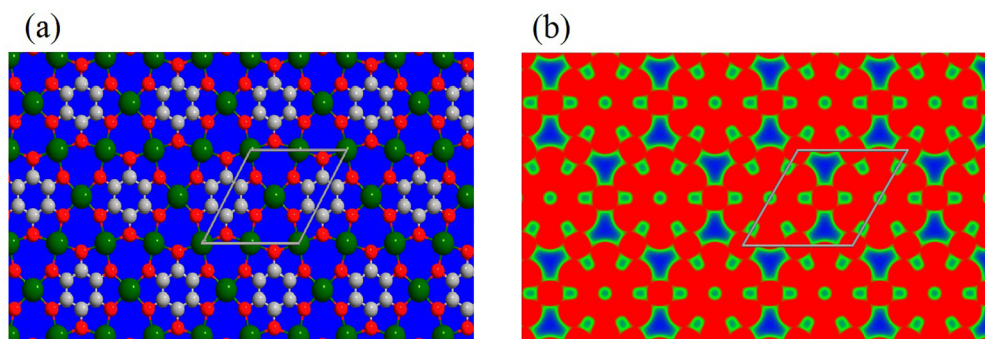


FIG. 2. Geometrical structure (a) and charge density (b) between 0 and 0.5 Å⁻³ (from blue over green to red) of a freestanding structure of Fe₃C₆O₆ (C: gray, O: red, and Fe: green). Elementary surface cell is also indicated.

TABLE I. Distance between *TM* atoms and O atoms (d_{TM-O} , in Å); lattice constant (a , in Å), energy differences $E_{ex,1/2} = E_{AFM(1/2)} - E_{FM}$ per *TM* atom, in meV; exchange-coupling constants J_1 and J_2 as explained in the text in meV; total magnetic moments (M per *TM* atom, in μ_B); local magnetic moments of d orbital at *TM* atoms (M_d per *TM* atom, in μ_B); local magnetic moments at *TM* atoms (M_m per *TM* atom, in μ_B); energy band gaps [spin up (E_a) and spin down (E_b), in eV]; and total energy gaps (E_g , in eV), for 2D $TM_3C_6O_6$ with PAW-SGGA+ U method. Exchange constants J_1 and J_2 are put in parentheses for $Cu_3C_6O_6$ since a Heisenberg model description is questionable there.

	d^1	d^2	d^3	d^4	d^5	d^6	d^7	d^8	d^9
$U = 5$ eV	Sc ₃ C ₆ O ₆	Ti ₃ C ₆ O ₆	V ₃ C ₆ O ₆	Cr ₃ C ₆ O ₆	Mn ₃ C ₆ O ₆	Fe ₃ C ₆ O ₆	Co ₃ C ₆ O ₆	Ni ₃ C ₆ O ₆	Cu ₃ C ₆ O ₆
$J = 0.9$ eV	($S = 1/2$)	($S = 1$)	($S = 3/2$)	($S = 2$)	($S = 5/2$)	($S = 2$)	($S = 3/2$)	($S = 1$)	($S = 1/2$)
d_{TM-O}	2.14	2.09	2.08	2.05	2.10	2.04	2.03	2.03	2.03
a	8.06	7.92	7.90	7.82	7.98	7.82	7.76	7.72	7.75
$E_{ex,1}$	-17.91	-5.36	2.17	-4.02	-6.08	-15.60	-11.93	-31.32	24.86
$E_{ex,2}$	-33.91	-10.24	5.45	-7.55	-10.87	-28.92	21.31	-50.53	30.50
J_1	48.00	3.66	-1.09	0.66	0.57	2.50	-11.08	14.41	(16.92)
J_2	2.87	0.18	0.19	0.05	0.08	0.21	7.53	4.54	(-62.67)
M	1	2	3	4	5	4	3	2	1
M_d	0.415	1.428	2.36	3.37	4.58	3.61	2.66	1.67	0.397
M_m	0.454	1.495	2.39	3.41	4.65	3.65	2.68	1.69	0.394
E_a	1.15	2.07	1.60	1.71	1.22	1.84	1.68	0	2.23
E_b	2.25	2.20	2.22	2.22	1.50	0	1.13	0	0
E_g	0.43	1.42	1.60	1.67	1.22	0	1.13	0	0

the second minima with a lattice constant of 8.25 Å is slightly different from Fig. 2(a). The C₆O₆ rings are rotated such that two Cu-O bonds are stronger than the two others [22].

B. Magnetic properties

In Fig. 2(b) we show the charge density for the cut plane which is the x - y plane, and in Fig. 3 we show the spin density. The charge density is defined as the sum of the electron densities for spin up and spin down ($\rho_{spin\ up} + \rho_{spin\ down}$). The magnetism distributions of the $TM_3C_6O_6$ monolayers can be intuitively studied by the analysis of the spin density, which is defined as the difference between the electron densities for spin up and spin down ($\rho_{spin\ up} - \rho_{spin\ down}$). The change in the magnetization distribution when one goes from Sc to Cu is quite interesting. In the beginning of the 3d series (Sc, Ti, V, and Cr) the spin density is nearly exclusively concentrated on the metal sites, which generates a kagome lattice. For Fe, Co, and Ni, some moments appear at the oxygen sites that bridge the metals. And finally, for $Cu_3C_6O_6$, the main magnetic moment is located on the carbon ring.

To describe the magnetic moments at the *TM* sites, we calculated the exchange couplings J_1 and J_2 to nearest- and second-nearest neighbors on the kagome lattice [see Fig. 4(a)] in the Heisenberg Hamiltonian:

$$H = \sum_{\langle i,j \rangle} J_{ij} \vec{S}_i \vec{S}_j = \frac{1}{2} \sum_{i,j} J_{ij} \vec{S}_i \vec{S}_j = \frac{1}{2} \sum_{i,g} J_g \vec{S}_i \vec{S}_{i+g},$$

where $\langle i, j \rangle$ means that one sums over each bond only once, and where the exchange couplings depend only on $g = j - i$. For that purpose, we estimate the energy difference between the ferromagnetic state and the antiferromagnetic one after relaxation of all atomic positions $E_{ex} = E_{AFM} - E_{FM}$, defined per magnetic ion. If that energy difference is positive (negative), it indicates that the ferromagnetic (antiferromagnetic) state is preferred. For a ferromagnetic arrangement of the spins S on the kagome lattice, we obtain $E_{FM} = 2(J_1 + J_2)S^2$,

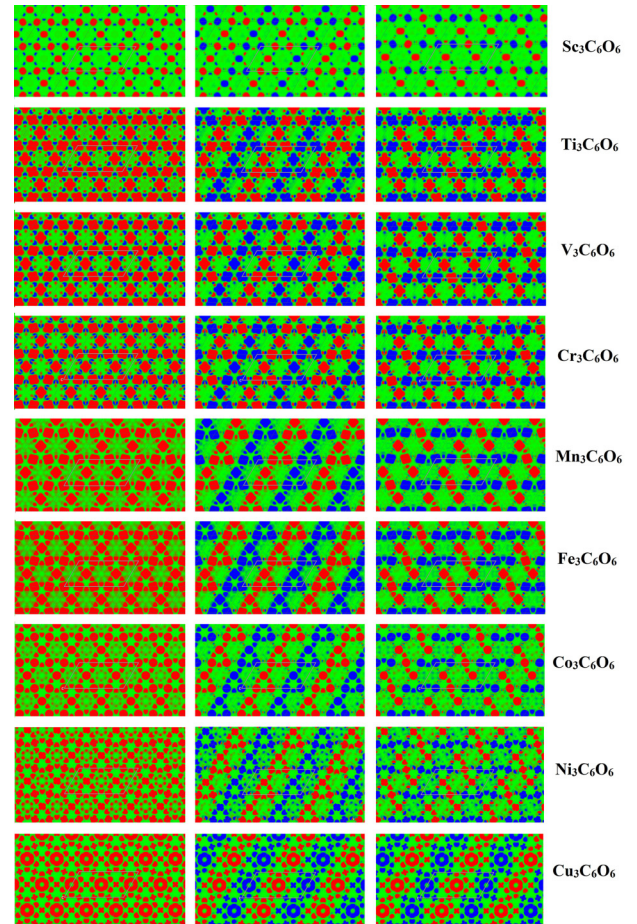


FIG. 3. Spin-density distribution of $TM_3C_6O_6$ obtained with SGGA+ U correction. Compared are ferromagnetic arrangement (left column) with two antiferromagnetic arrangements (AFM1: middle column and AFM2: right column) for *TMs* of the 3d series from Sc to Cu. Spin density varies between -0.02 and 0.02 Å⁻³ (blue is negative and red is positive).

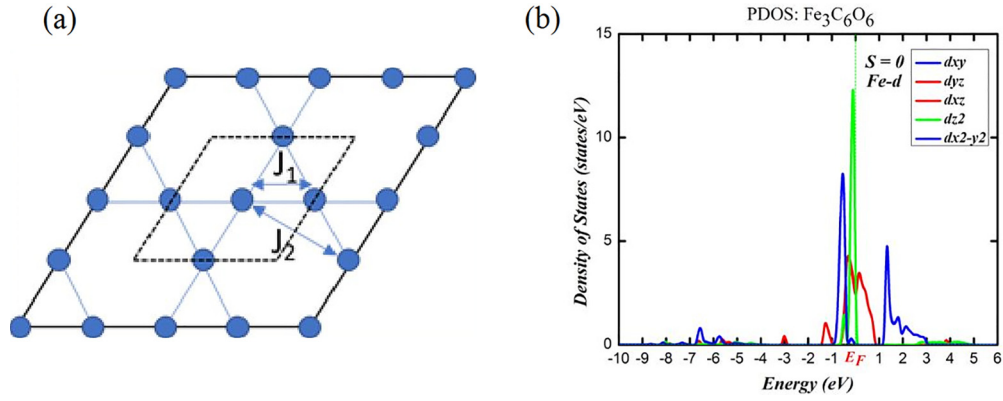


FIG. 4. Schematic representation of kagome lattice with exchange couplings J_1 and J_2 (a) and partial, orbital-resolved 3d DOS for $\text{Fe}_3\text{C}_6\text{O}_6$ in GGA (not spin polarized and without U correction) to illustrate ligand field splitting (b).

whereas the two antiferromagnetic arrangements that are visible in Fig. 3 give $E_{\text{AFM1}} = \frac{2}{3}(J_1 - J_2)S^2$ and $E_{\text{AFM2}} = -\frac{2}{3}(J_1 + J_2)S^2$. Taking the two energy differences $E_{\text{ex},1} = E_{\text{AFM1}} - E_{\text{FM}}$ and $E_{\text{ex},2} = E_{\text{AFM2}} - E_{\text{FM}}$, we calculate the corresponding exchange couplings reported in Table I.

With the SGGA+ U method, the exchange couplings are antiferromagnetic and of short range for the transition metals Sc, Ti, Cr, and Fe. We find a rather large value of $J_1 = 48$ meV for Sc, whereas J_1 varies between 0.66 meV (Cr) and 3.66 meV (Ti) for the other three compounds. In all cases, the second-neighbor coupling is at least a factor of 10

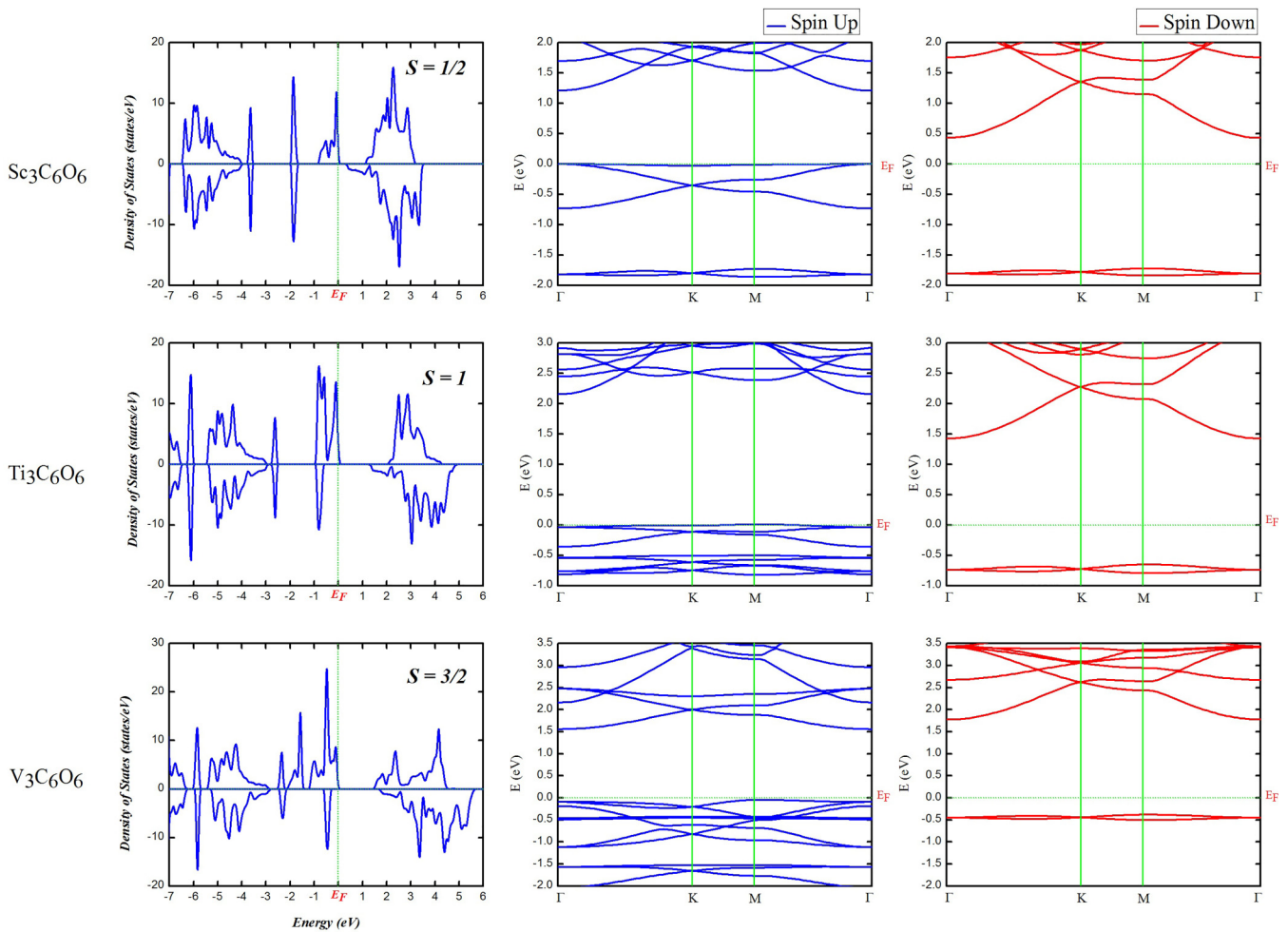


FIG. 5. Total DOS (left column) and band structures with spin up (middle column) and spin down (right column) for $\text{Sc}_3\text{C}_6\text{O}_6$, $\text{Ti}_3\text{C}_6\text{O}_6$, and $\text{V}_3\text{C}_6\text{O}_6$ monolayers, calculated with PAW-SGGA+ U method. Green dashed lines represent Fermi level at 0 eV.

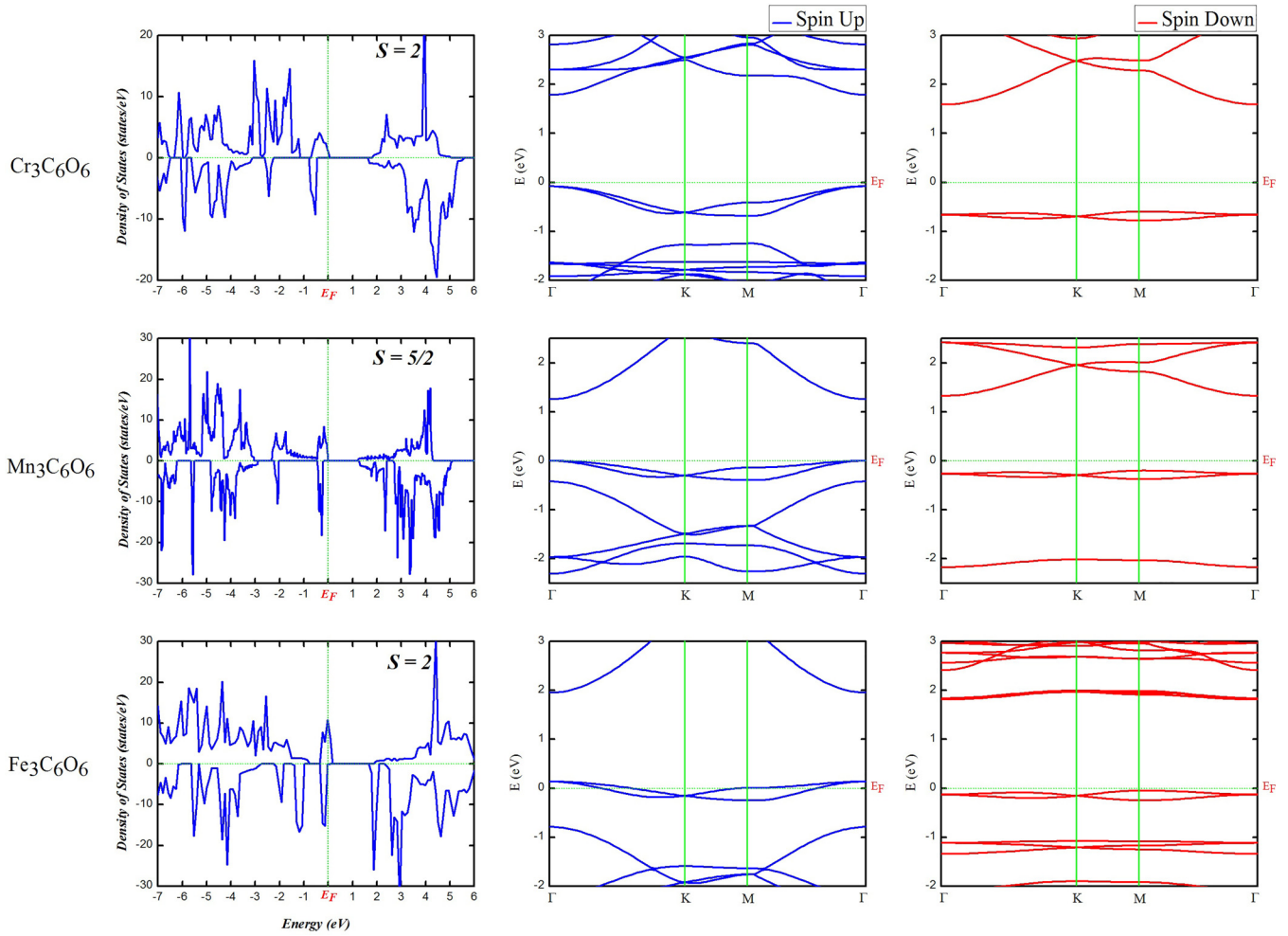


FIG. 6. Total DOS and band structures of $TM_3C_6O_6$ monolayers with PAW-SGGA+ U method, where TM is Cr, Mn, and Fe, respectively.

smaller. We find it remarkable that the SGGA+ U values of exchange couplings are confirmed by the SGGA calculations (see Appendix A). In two cases, we find the energy of the ferromagnetic configuration lower than both antiferromagnetic solutions, for the V- and the Cu system. Furthermore, the second-neighbor coupling considerably exceeds the first-neighbor one for $Cu_3C_6O_6$ and we cannot exclude further reaching exchange couplings, which means that a short-range Heisenberg model is probably not applicable in that situation. But, the metallic state and the strong energy gain of the FM configuration are strong indications for a FM ground state of $Cu_3C_6O_6$. For the $Co_3C_6O_6$ structure, the ferromagnetic state is energetically located in between both antiferromagnetic ones ($E_{ex,1} = -11.93$ meV and $E_{ex,2} = 21.31$ meV). It means that the exchange coupling to nearest neighbors $J_1 = -11.08$ meV is ferromagnetic, but the second-neighbor exchange coupling $J_2 = 7.53$ meV turns to an antiferromagnetic exchange (see Table I).

C. Electronic properties

The spin-resolved band-structure calculations for the $TM_3C_6O_6$ structures were carried out in the high-symmetry directions of the first Brillouin zone and they are shown together with the total densities of states (DOS) in Figs. 5–7.

Figures 8–10 show the partial DOS. We present the results with the SGGA+ U functional which we think to be more relevant than the SGGA results as outlined above. The SGGA+ U gap values which are visible in Figs. 5–7 are also given in Table I and they can be compared with the SGGA gaps in Table III in Appendix A. It is important to note that the kagome lattice described here [Fig. 4(a)] has the same point-group symmetry D_{6h} as graphene. As a consequence, the numerous band crossings at the K point are guaranteed by symmetry, which qualifies all materials of the $TM_3C_6O_6$ class to be potentially interesting topological quantum materials. Note also, that the TM -O lattice in Fig. 2(a) coincides exactly with the Cu-O lattice in the planes of Herbertsmithite [47], a naturally occurring material that is widely studied to observe spin liquid behavior.

The local magnetic moment is mostly determined by the filling of the 3d levels which is also important for the electronic properties. The splitting of those 3d levels is best visible in a GGA calculation (nonmagnetic solution without U correction) which is free of exchange or Coulomb shifts [Fig. 4(b)]. Following the analysis in Ref. [9] we can distinguish the d_z^2 band, a doubly degenerate d_{xz}/d_{yz} band and the $d_{x^2-y^2}/d_{xy}$ complex which splits into a lower-lying bonding and a higher-lying antibonding part due to the metal-metal interaction in the plane. The order of these levels is rather

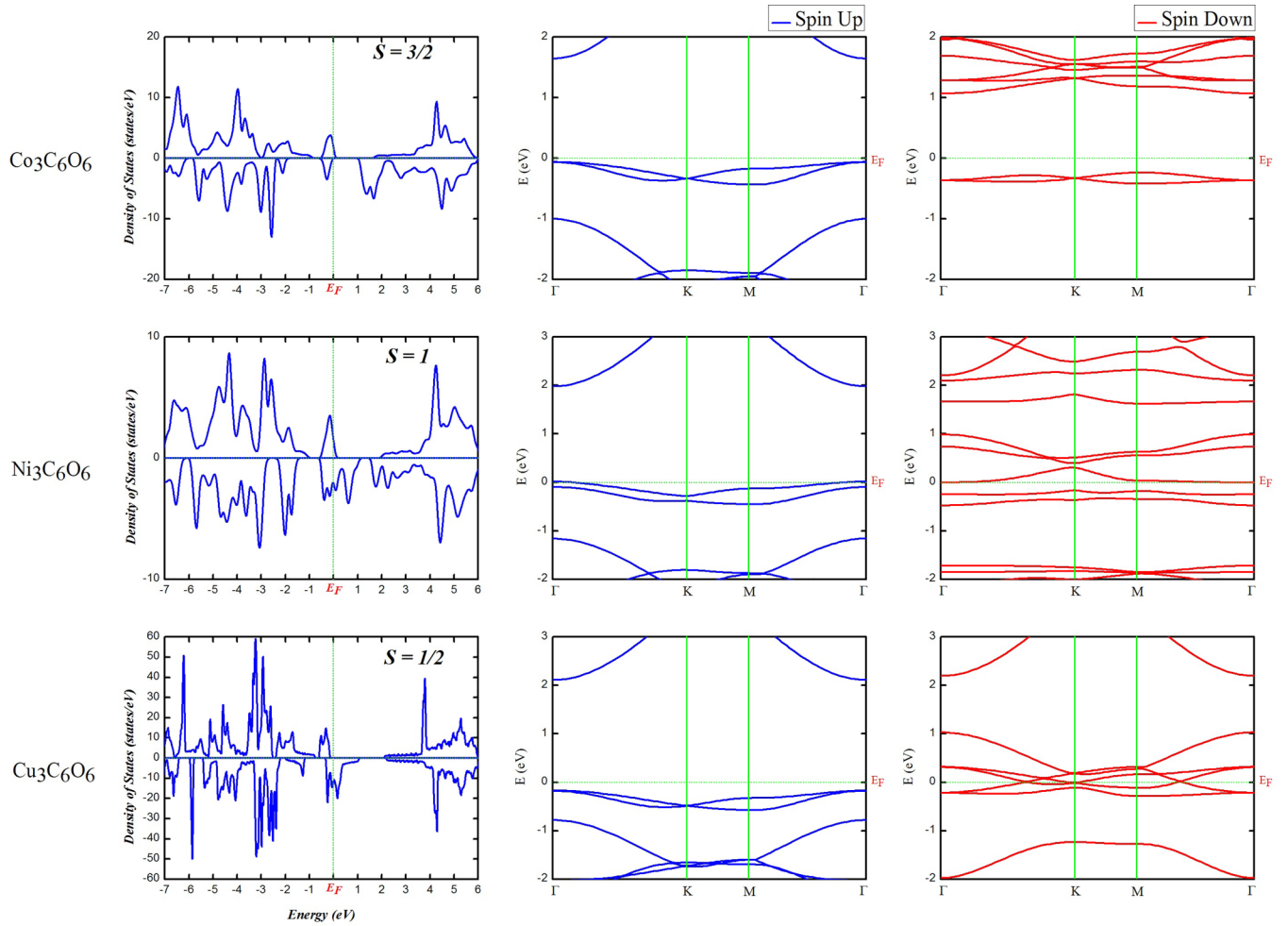


FIG. 7. Total DOS and band structures of $TM_3C_6O_6$ monolayers with PAW-SGGA+ U method, where TM is Co, Ni, and Cu, respectively.

constant throughout the $3d$ series, just the d_z^2 band moves a little bit from one system to another.

As one can easily observe in Figs. 6 and 7, the Fe, Ni, and Cu structures are predicted to be metallic. Therefore, they could be interesting for technological applications as metal-organic conducting layers. One might wonder why $Fe_3C_6O_6$ is metallic, whereas its neighboring compounds in the $3d$ series with Mn and Co are both insulating. Furthermore, the gap of the Fe case is not very sensitive to the Hubbard U parameter in the $3d$ shell. That is explained since the relevant band at the Fermi level has very few $3d$ contributions, both for the Fe and the Co cases, but that band is half filled for $Fe_3C_6O_6$ and completely occupied for $Co_3C_6O_6$.

The $Cu_3C_6O_6$ case is highly interesting, since it is metallic and shows band crossings at the Fermi level without spin-orbit coupling for parallel spin arrangements. It has ferromagnetic second-neighbor couplings which dominate with respect to the nearest-neighbor ones and we predict a free-standing $Cu_3C_6O_6$ layer to be a ferromagnetic kagome metal. According to Fig. 7, it has three band-crossing points at the Fermi level along the path Γ - K - M - Γ . Regarding more in detail (see Appendix B), it becomes clear that the crossing between Γ and K is connected with the crossing between Γ and M , whereas the band crossing at K is isolated. Finally, we have a Fermi line around Γ and two points at K and K' in the Brillouin

zone [shown in Fig. 11(a)]. As expected, the SO coupling with the magnetization perpendicular to the plane opens small gaps of about 10 meV at the crossing points as shown in Fig. 11(b). It should be mentioned that the strong electron correlation in the $3d$ shell, treated in the SGGA+ U approximation, is not able to open a gap. Since the gap opens only due to SO coupling, $Cu_3C_6O_6$ cannot be a Mott-Hubbard insulator and, similar to the kagome lattice $Cs_2LiMn_3F_{12}$ [9], we expect also $Cu_3C_6O_6$ to be a Chern insulator and to show the QAHE. Different from previously discussed materials, $Cu_3C_6O_6$ has not only isolated Fermi points but also a closed Fermi line without SO coupling.

III. SUMMARY AND DISCUSSION

We summarize the possible electronic states and magnetic couplings of the $TM_3C_6O_6$ material class in Table II. Possible spin-crossover or magnetic-nonmagnetic transitions for the Fe, Co, Ni, and Cu compounds (see Fig. 1) are not mentioned in this table. Due to the SCO transition the variation of the total energy with the lattice constant is smaller than without it, which should allow epitaxial growth on very different lattices, as it was observed for Fe-zwitterionic quinone on Ag(111) and Au(110) surfaces [77]. There is a series of semi-conducting compounds with dominating antiferromagnetic

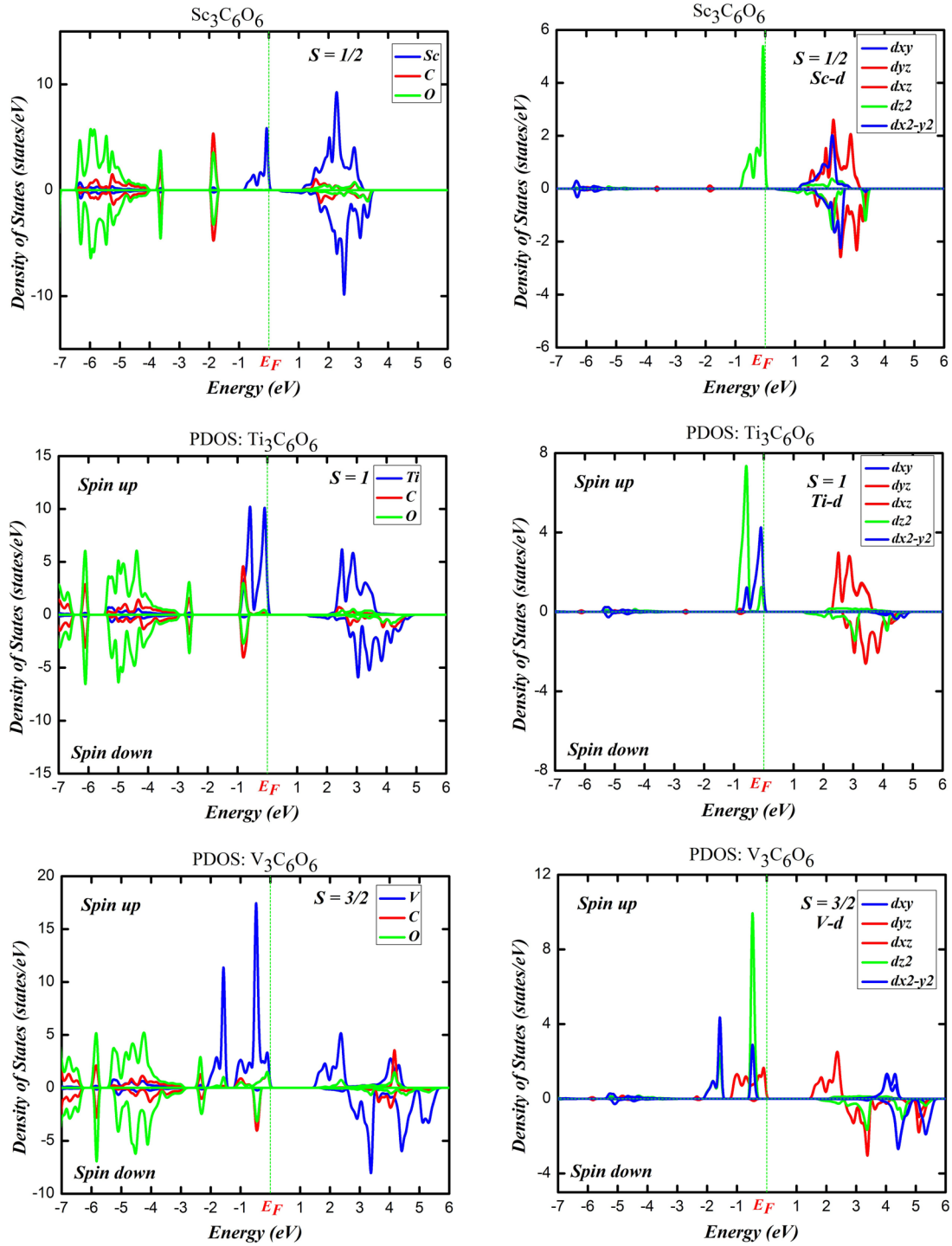


FIG. 8. Projected DOS for *TM*, C, and O atoms (left column) and *d* orbitals on *TM* atoms (right column) for $\text{Sc}_3\text{C}_6\text{O}_6$, $\text{Ti}_3\text{C}_6\text{O}_6$, and $\text{V}_3\text{C}_6\text{O}_6$ monolayers obtained with PAW-SGGA+*U* method.

nearest-neighbor exchange but varying local spins between $S = 1/2$ and $S = 5/2$, which is very interesting since it could allow experimental studies of the antiferromagnetic kagome lattice for different spins. Three compounds are predicted to be metallic with the interesting perspectives to investigate the interaction between magnetic order and transport properties, especially the anomalous Hall effect. As we argued, the anomalous Hall effect can even be quantified for $\text{Cu}_3\text{C}_6\text{O}_6$.

For a final conclusion about the possible magnetic states, also local magnetic anisotropies should be investigated in further studies. As one can see by comparing Tables I and III, the structural properties and the values of the local magnetic moments are much less influenced by the Hubbard *U* correction than the electronic properties, especially the gap values. Therefore, the present theoretical predictions have to be verified by experimental methods.

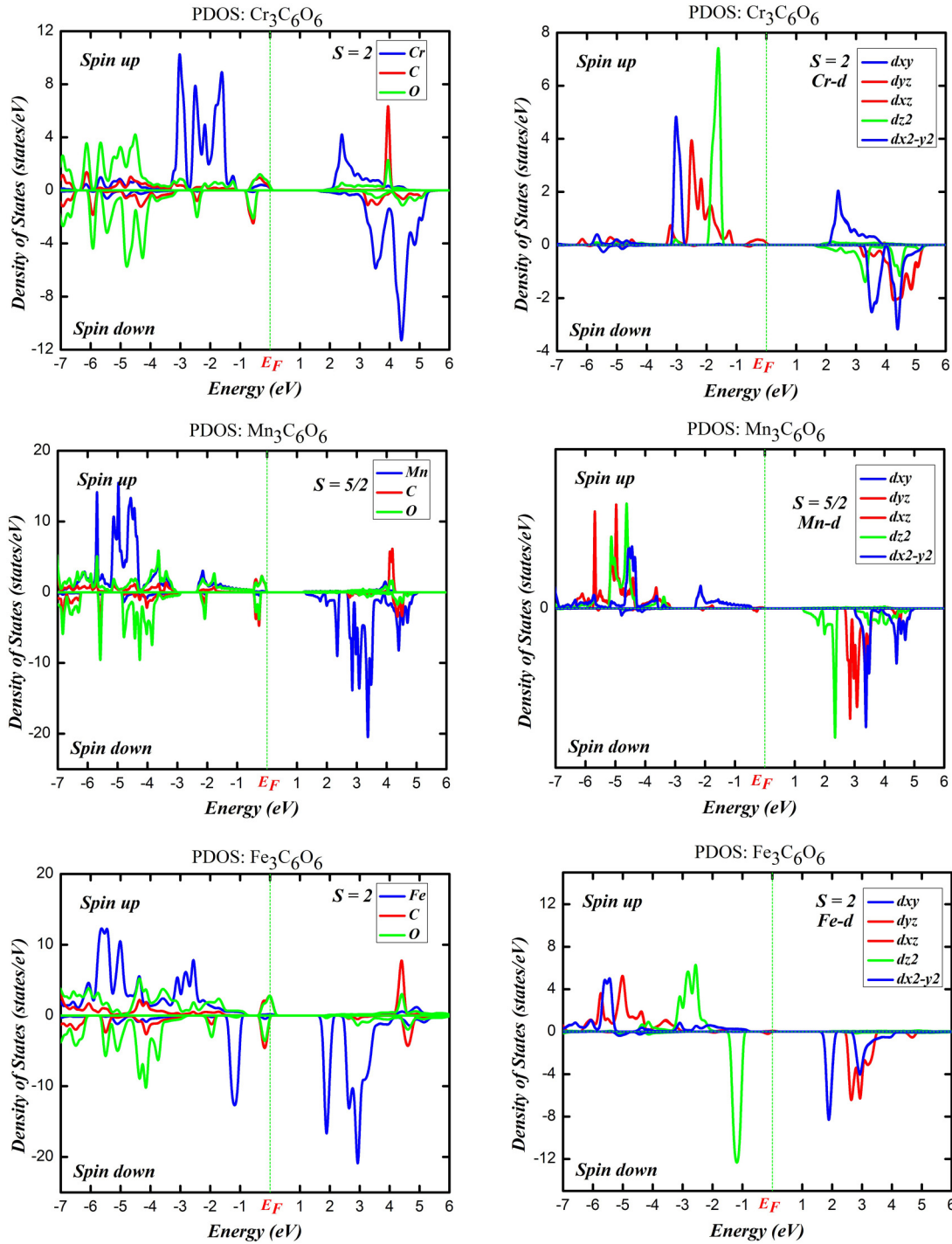


FIG. 9. Projected DOS of atoms (TM , C, and O) and d orbitals on TM atoms in 2D $TM_3C_6O_6$ materials with PAW-SGGA+ U method for TMs Cr, Mn, and Fe.

IV. CONCLUSION AND PERSPECTIVES

We investigated systematically with the help of first-principles calculations the structural, electronic, and magnetic properties of organometallic networks, which are built of C_6O_6 ligands (L) and $3d$ transition metals and detected a surprisingly rich behavior. Similar organometallic networks with other organic molecules are also known [86,87]. We investigated the TM_3L material class; for calculations of the TM_3L_2

class with a larger distance between TM ions and correspondingly weaker magnetic couplings, see the recent publication [88]. We have found that for $Fe_3C_6O_6$, $Co_3C_6O_6$, $Ni_3C_6O_6$, and $Cu_3C_6O_6$ the magnetic state changes as a function of the lattice parameter. These transitions may be triggered by different substrates or other external stimuli like pressure or temperature.

Three of the calculated monolayers (for the Fe [23], Mn [22], and Cu [21] systems) could already be synthesized on

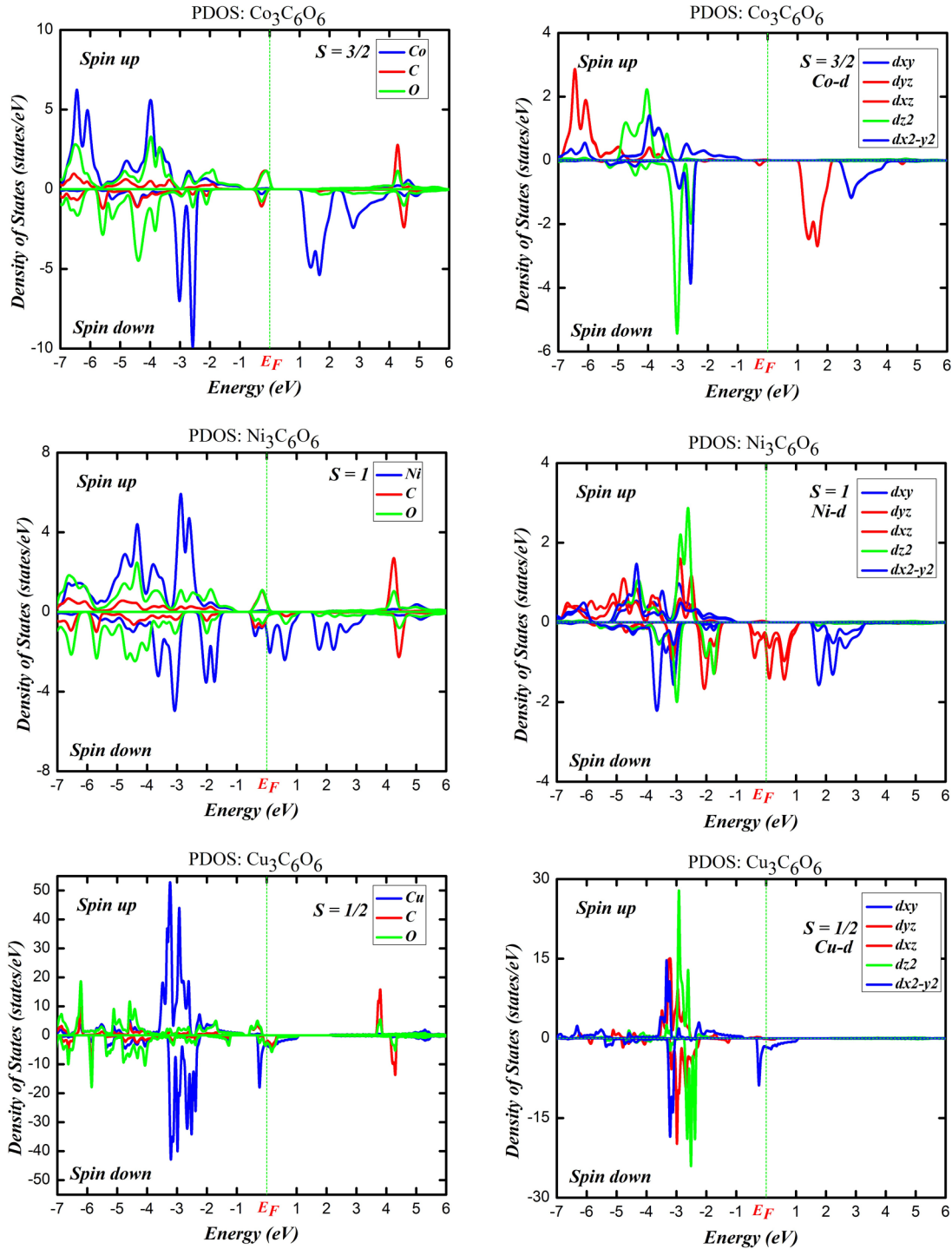


FIG. 10. Projected DOS of atoms (*TM*, C, and O) and *d* orbitals on *TM* atoms in 2D $TM_3C_6O_6$ materials with PAW-SGGA+*U* method for *TMs* Co, Ni, and Cu.

noble-metal substrates. There is no visible reason which could hinder synthesizing also all the other investigated materials, and it would be interesting to try insulating substrates. A metallic substrate is expected to influence the electronic structure in several respects: it may fix the lattice constant, induce a charge transfer between substrate and monolayer, and lead to a slight buckling of the monolayer. However, as it was calculated for $Mn_3C_6O_6$ [22], the local magnetic moments are expected to remain with slightly changed magnetic in-

teractions. Investigating the magnetic properties of $Sc_3C_6O_6$ monolayers, one could expect signatures of spin liquid behavior since we predict it to be an $S = 1/2$ kagome Heisenberg antiferromagnet. Possible experimental methods could be x-ray dichroism, spin-resolved tunneling spectroscopy, or magneto-optical Kerr studies. Also interesting is the comparison with $Ti_3C_6O_6$, which is also a kagome antiferromagnet but with $S = 1$. The ferromagnetic nearest-neighbor couplings in $V_3C_6O_6$ are also interesting. Together with an appropriate

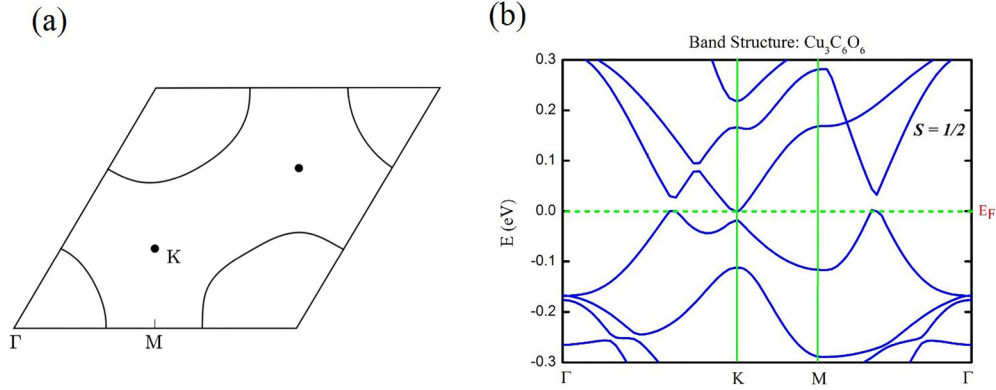


FIG. 11. Fermi surface without SO coupling consisting of a circle around Γ and two isolated points in the Brillouin zone (a), and band structure including SO coupling which opens a gap for $\text{Cu}_3\text{C}_6\text{O}_6$ network calculated with PAW-SGGA+ U method (b).

magnetic anisotropy (still to be calculated), they could lead to spin ice behavior.

Two-dimensional metal-organic conducting systems are important for technical applications and we predict metallic behavior for the Fe, Ni, and Cu metal-organic compounds. From a fundamental point of view, a band crossing at the Fermi level allows us to study topological quantum states that we found to be realized in $\text{Cu}_3\text{C}_6\text{O}_6$ (without SO coupling) in connection with ferromagnetic exchange couplings. The SO coupling leads to small gaps of about 10 meV and this material is a candidate for a Chern insulating phase and the QAHE. Comparing the freestanding layer of $\text{Cu}_3\text{C}_6\text{O}_6$ investigated here with the one synthesized on Ag(111) [22], there is unfortunately a charge transfer for the adsorbed layer that destabilizes the magnetic solution with respect to the nonmagnetic one with larger lattice constant. So, it would be interesting to look for alternative synthesis routes of this promising material using for instance insulating substrates.

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The authors declare that they have no conflict of interest.

APPENDIX A

The *ab initio* results without Hubbard U correction are listed in Table III.

APPENDIX B

To clarify the Fermi surface of $\text{Cu}_3\text{C}_6\text{O}_6$ we calculated a series of band structures along the lines $S1 \dots S6$ starting at Γ and ending at $\vec{b}_1 + x\vec{b}_2$ with $x = \{0; 0.1; 0.2, 0.3, 0.4, 0.5\}$ (Fig. 12). Please note that $(1/2)\vec{b}_1$ and $(1/2)\vec{b}_2$ correspond to the M point and $(2/3)\vec{b}_1 + (1/3)\vec{b}_2$ corresponds to the K point in the Brillouin zone.

TABLE II. Summary of electronic and magnetic properties of $\text{TM}_3\text{C}_6\text{O}_6$ material class. Mentioned are either dominating magnetic couplings or both of them if they are of comparable strength. Results for $\text{Mn}_3\text{C}_6\text{O}_6$ are taken from Ref. [22].

System	Local spin	Magnetic couplings (dominant)	Electronic properties
$\text{Sc}_3\text{C}_6\text{O}_6$	1/2	AFM ^a (first neighbor)	SC
$\text{Ti}_3\text{C}_6\text{O}_6$	1	AFM (first neighbor)	SC
$\text{V}_3\text{C}_6\text{O}_6$	3/2	FM (first neighbor)	SC
$\text{Cr}_3\text{C}_6\text{O}_6$	2	AFM (first neighbor)	SC
$\text{Mn}_3\text{C}_6\text{O}_6$	5/2	AFM (first neighbor)	SC
$\text{Fe}_3\text{C}_6\text{O}_6$	2	AFM (first neighbor)	M
$\text{Co}_3\text{C}_6\text{O}_6$	3/2	FM (first) and AFM (second neighbor)	SC
$\text{Ni}_3\text{C}_6\text{O}_6$	1	AFM (first and second neighbor)	M
$\text{Cu}_3\text{C}_6\text{O}_6$	1/2	FM (no Heisenberg)	M

^aAbbreviations used in table. AFM: antiferromagnetic, SC: semiconductor, FM: ferromagnetic, and M: metal.

TABLE III. Same physical parameters as in Table I but without U correction.

	d^1	d^2	d^3	d^4	d^5	d^6	d^7	d^8	d^9
Without U	$\text{Sc}_3\text{C}_6\text{O}_6$ ($S = 1/2$)	$\text{Ti}_3\text{C}_6\text{O}_6$ ($S = 1$)	$\text{V}_3\text{C}_6\text{O}_6$ ($S = 3/2$)	$\text{Cr}_3\text{C}_6\text{O}_6$ ($S = 2$)	$\text{Mn}_3\text{C}_6\text{O}_6$ ($S = 5/2$)	$\text{Fe}_3\text{C}_6\text{O}_6$ ($S = 2$)	$\text{Co}_3\text{C}_6\text{O}_6$ ($S = 3/2$)	$\text{Ni}_3\text{C}_6\text{O}_6$ ($S = 1$)	$\text{Cu}_3\text{C}_6\text{O}_6$ ($S = 1/2$)
d_{TM-O}	2.14	2.09	2.04	2.03	2.10	2.04	2.02	2.03	2.04
a	8.06	7.92	7.80	7.76	7.94	7.80	7.74	7.70	7.78
E_{ex-1}	-24.77	-7.49	39.96	-8.37	-26.07	-52.96	-59.65	-33.60	3.92
E_{ex-2}	-44.47	-12.83	110.95	-14.97	-46.35	-91.60	31.14	-39.32	3.1
J_1	59.11	4.01	-23.66	1.24	2.43	7.25	-30.26	4.29	2.46
J_2	7.61	0.81	5.17	0.17	0.35	1.34	25.07	10.46	-7.11
M	1	2	3	4	5	4	3	2	1
M_d	0.404	1.337	2.30	3.27	4.36	3.49	2.44	1.36	0.378
M_m	0.442	1.410	2.33	3.31	4.45	3.52	2.46	1.37	0.376
E_a	1.21	0.83	0	0.75	0.64	1.76	0	0	1.99
E_b	2.21	2.20	2.37	1.95	1.08	0	0	0	0
E_g	1.05	0.82	0	0.75	0	0	0	0	0

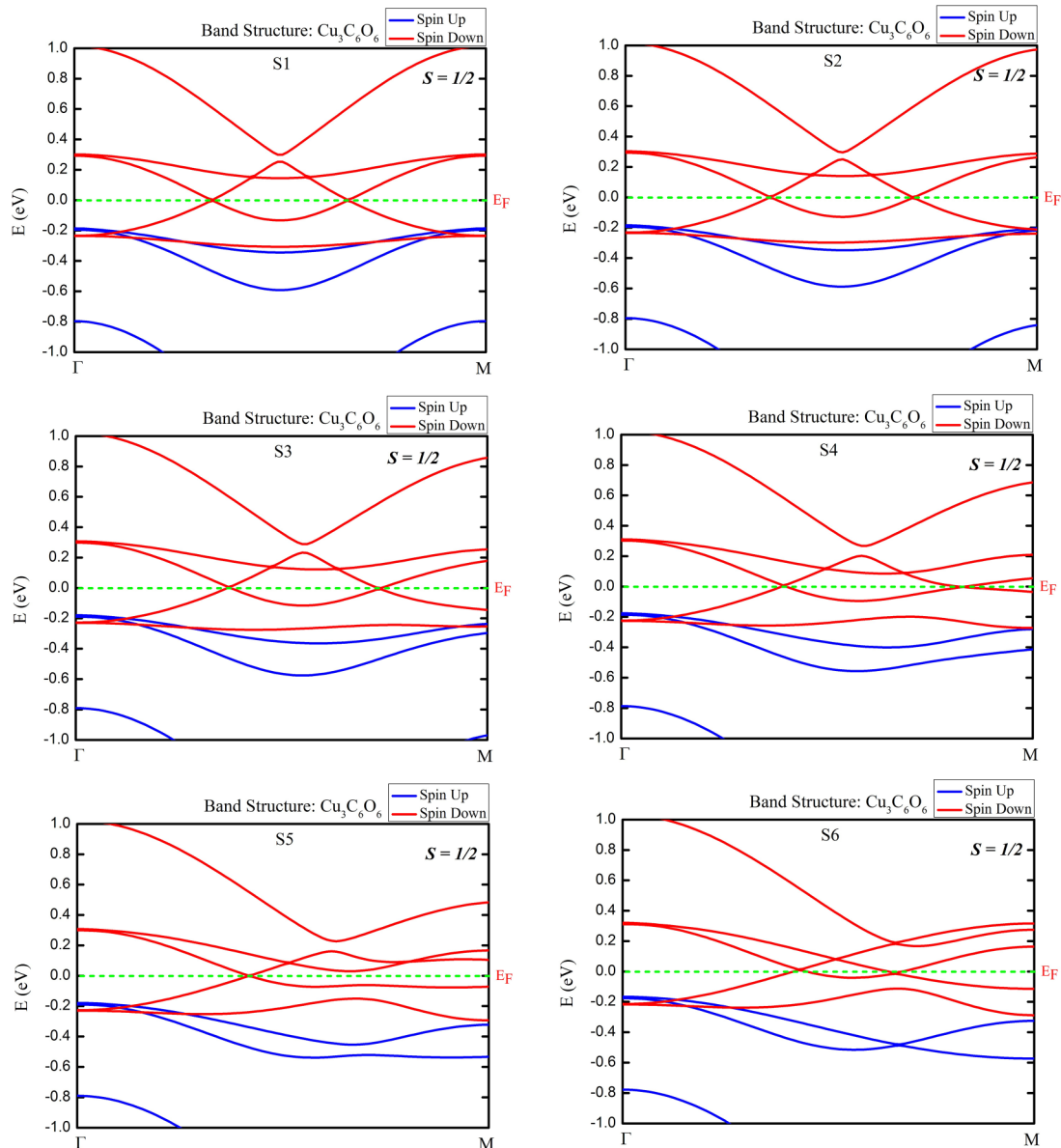


FIG. 12. Band structure without SO coupling of $\text{Cu}_3\text{C}_6\text{O}_6$ along lines S1 ... S6 explained in the text.

- [1] J. O. Sofo and C. A. Balseiro, *Phys. Rev. B* **45**, 377 (1992).
- [2] T. H. Han, J. S. Helton, S. Chu, D. G. Nocera, J. A. Rodriguez-Rivera, C. Broholm, and Y. S. Lee, *Nature (London)* **492**, 406 (2012).
- [3] Y. Zhou, K. Kanoda, and T.-K. Ng, *Rev. Mod. Phys.* **89**, 025003 (2017).
- [4] P. Mendels and F. Bert, *C. R. Phys.* **17**, 455 (2016).
- [5] L. Savary and L. Balents, *Rep. Prog. Phys.* **80**, 016502 (2017).
- [6] I. I. Mazin, H. O. Jeschke, F. Lechermann, H. Lee, M. Fink, R. Thomale, and R. Valentí, *Nat. Commun.* **5**, 4261 (2014).
- [7] L. Ye, M. Kang, J. Liu, F. Von Cube, C. R. Wicker, T. Suzuki, C. Jozwiak, A. Bostwick, E. Rotenberg, D. C. Bell, L. Fu, R. Comin, and J. G. Checkelsky, *Nature (London)* **555**, 638 (2018).
- [8] H. M. Guo and M. Franz, *Phys. Rev. B* **80**, 113102 (2009).
- [9] G. Xu, B. Lian, and S. C. Zhang, *Phys. Rev. Lett.* **115**, 186802 (2015).
- [10] E. J. Bergholtz and Z. Liu, *Int. J. Mod. Phys. B* **27**, 1330017 (2013).
- [11] O. Derzhko, J. Richter, and M. Maksymenko, *Int. J. Mod. Phys. B* **29**, 1530007 (2015).
- [12] D. Leykam, A. Andreanov, and S. Flach, *Adv. Phys. X* **3**, 1473052 (2018).
- [13] Y. Hatsugai, *Ann. Phys.* **435**, 168453 (2021).
- [14] E. Tang, J. W. Mei, and X. G. Wen, *Phys. Rev. Lett.* **106**, 236802 (2011).
- [15] E. J. Bergholtz, Z. Liu, M. Trescher, R. Moessner, and M. Udagawa, *Phys. Rev. Lett.* **114**, 016806 (2015).
- [16] S. Nakatsuji, N. Kiyohara, and T. Higo, *Nature (London)* **527**, 212 (2015).
- [17] A. K. Nayak, J. E. Fischer, Y. Sun, B. Yan, J. Karel, A. C. Komarek, C. Shekhar, N. Kumar, W. Schnelle, J. Kübler, C. Felser, and S. S. P. Parkin, *Sci. Adv.* **2**, e1501870 (2016).
- [18] K. Kuroda, T. Tomita, M. T. Suzuki, C. Bareille, A. A. Nugroho, P. Goswami, M. Ochi, M. Ikhlas, M. Nakayama, S. Akebi, R. Noguchi, R. Ishii, N. Inami, K. Ono, H. Kumigashira, A. Varykhalov, T. Muro, T. Koretsune, R. Arita, S. Shin, T. Kondo, and S. Nakatsuji, *Nat. Mater.* **16**, 1090 (2017).
- [19] S. Dasgupta, *Phys. Rev. B* **106**, 064431 (2022).
- [20] Z. F. Wang, Z. Liu, and F. Liu, *Nat. Commun.* **4**, 1471 (2013).
- [21] R. Zhang, J. Liu, Y. Gao, M. Hua, B. Xia, P. Knecht, A. C. Papageorgiou, J. Reichert, J. V. Barth, H. Xu, L. Huang, and N. Lin, *Angew. Chem. Int. Ed.* **59**, 2669 (2020).
- [22] N. Shaiek, H. Denawi, M. Koudia, R. Hayn, S. Schäfer, I. Berbezier, C. Lamine, O. Siri, A. Akremi, and M. Abel, *Adv. Mater. Interfaces* **9**, 2201099 (2022).
- [23] M. Hua, B. Xia, M. Wang, E. Li, J. Liu, T. Wu, Y. Wang, R. Li, H. Ding, J. Hu, Y. Wang, J. Zhu, H. Xu, W. Zhao, and N. Lin, *J. Phys. Chem. Lett.* **12**, 3733 (2021).
- [24] Z. Gao, H. Ma, S. Yuan, H. Ren, Z. Ge, H. Zhu, W. Guo, F. Ding, and W. Zhao, *Appl. Surf. Sci.* **601**, 154187 (2022).
- [25] J. T. Chalker, P. C. W. Holdsworth, and E. F. Shender, *Phys. Rev. Lett.* **68**, 855 (1992).
- [26] A. Chubukov, *Phys. Rev. Lett.* **69**, 832 (1992).
- [27] C. L. Henley and E. P. Chan, *J. Magn. Magn. Mater.* **140-144**, 1693 (1995).
- [28] A. M. Läuchli, J. Sudan, and R. Moessner, *Phys. Rev. B* **100**, 155142 (2019).
- [29] J. Schnack, J. Schulenburg, and J. Richter, *Phys. Rev. B* **98**, 094423 (2018).
- [30] Y. Iqbal, F. Becca, S. Sorella, and D. Poilblanc, *Phys. Rev. B* **87**, 060405(R) (2013).
- [31] H. J. Liao, Z. Y. Xie, J. Chen, Z. Y. Liu, H. D. Xie, R. Z. Huang, B. Normand, and T. Xiang, *Phys. Rev. Lett.* **118**, 137202 (2017).
- [32] S. Yan, D. A. Huse, and S. R. White, *Science* **332**, 1173 (2011).
- [33] S. Depenbrock, I. P. McCulloch, and U. Schollwöck, *Phys. Rev. Lett.* **109**, 067201 (2012).
- [34] O. Götze, D. J. J. Farnell, R. F. Bishop, P. H. Y. Li, and J. Richter, *Phys. Rev. B* **84**, 224428 (2011).
- [35] H. J. Changlani and A. M. Läuchli, *Phys. Rev. B* **91**, 100407(R) (2015).
- [36] S. Nishimoto and M. Nakamura, *Phys. Rev. B* **92**, 140412(R) (2015).
- [37] Z.-X. Liu, H.-H. Tu, Y.-H. Wu, R.-Q. He, X.-J. Liu, Y. Zhou, and T.-K. Ng, *Phys. Rev. B* **97**, 195158 (2018).
- [38] J. Oitmaa and R. R. P. Singh, *Phys. Rev. B* **93**, 014424 (2016).
- [39] T. Liu, W. Li, and G. Su, *Phys. Rev. E* **94**, 032114 (2016).
- [40] S. Nishimoto, N. Shibata, and C. Hotta, *Nat. Commun.* **4**, 2287 (2013).
- [41] S. Capponi, O. Derzhko, A. Honecker, A. M. Läuchli, and J. Richter, *Phys. Rev. B* **88**, 144416 (2013).
- [42] J. Schulenburg, A. Honecker, J. Schnack, J. Richter, and H.-J. Schmidt, *Phys. Rev. Lett.* **88**, 167207 (2002).
- [43] M. E. Zhitomirsky, *J. Phys.: Conf. Ser.* **592**, 012110 (2015).
- [44] M. E. Zhitomirsky and H. Tsunetsugu, *Phys. Rev. B* **70**, 100403(R) (2004).
- [45] J. Schnack, J. Schulenburg, A. Honecker, and J. Richter, *Phys. Rev. Lett.* **125**, 117207 (2020).
- [46] D. S. Inosov, *Adv. Phys.* **67**, 149 (2018).
- [47] M. R. Norman, *Rev. Mod. Phys.* **88**, 041002 (2016).
- [48] R. Okuma, D. Nakamura, T. Okubo, A. Miyake, A. Matsuo, K. Kindo, M. Tokunaga, S. Takeyama, and Z. Hiroi, *Nat. Commun.* **10**, 1229 (2019).
- [49] R. Okuma, D. Nakamura, and S. Takeyama, *Phys. Rev. B* **102**, 104429 (2020).
- [50] A. S. Wills, R. Ballou, and C. Lacroix, *Phys. Rev. B* **66**, 144407 (2002).
- [51] *Spin Crossover in Transition Metal Compounds I-III*, edited by P. Gütllich and H. A. Goodwin (Springer, Berlin, Heidelberg, 2004).
- [52] A. B. Gaspar, V. Ksenofontov, M. Seredyuk, and P. Gütllich, *Coord. Chem. Rev.* **249**, 2661 (2005).
- [53] V. Martínez, A. B. Gaspar, M. Carmen Muñoz, G. V. Bukin, G. Levchenko, and J. A. Real, *Chem. Eur. J.* **15**, 10960 (2009).
- [54] A. Galet, A. B. Gaspar, M. C. Muñoz, G. V. Bukin, G. Levchenko, and J. A. Real, *Adv. Mater.* **17**, 2949 (2005).
- [55] G. Agustí, M. C. Muñoz, A. B. Gaspar, and J. A. Real, *Inorg. Chem.* **47**, 2552 (2008).
- [56] J. Dugay, M. Giménez-Marqués, T. Kozlova, H. W. Zandbergen, E. Coronado, and H. S. J. Van Der Zant, *Adv. Mater.* **27**, 1288 (2015).
- [57] X. T. Zhang, D. Sun, B. Li, L. M. Fan, B. Li, and P. H. Wei, *Cryst. Growth Des.* **12**, 3845 (2012).
- [58] J. A. Real, E. Andrés, M. C. Muñoz, M. Julve, A. Bousseksou, and F. Varret, *Science* **268**, 265 (1995).
- [59] M. Seredyuk, A. B. Gaspar, V. Ksenofontov, M. Verdaguer, F. Villain, and P. Gütllich, *Inorg. Chem.* **48**, 6130 (2009).
- [60] T. Miyamachi, M. Gruber, V. Davesne, M. Bowen, S. Boukari, L. Joly, F. Scheurer, G. Rogez, T. K. Yamada, P. Ohresser, E. Beaupaire, and W. Wulfhekel, *Nat. Commun.* **3**, 938 (2012).

- [61] S. Rat, M. Piedrahita-Bello, L. Salmon, G. Molnár, P. Demont, and A. Bousseksou, *Adv. Mater.* **30**, 1705275 (2018).
- [62] N. Baadji, M. Piacenza, T. Tugsuz, F. Della Sala, G. Maruccio, and S. Sanvito, *Nat. Mater.* **8**, 813 (2009).
- [63] F. Prins, M. Monrabal-Capilla, E. A. Osorio, E. Coronado, and H. S. J. Van Der Zant, *Adv. Mater.* **23**, 1545 (2011).
- [64] T. G. Gopakumar, F. Matino, H. Naggert, A. Bannwarth, F. Tuczek, and R. Berndt, *Angew. Chem., Int. Ed.* **51**, 6262 (2012).
- [65] P. Gütllich, Y. Garcia, and H. A. Goodwin, *Chem. Soc. Rev.* **29**, 419 (2000).
- [66] A. Bousseksou, G. Molnár, L. Salmon, and W. Nicolazzi, *Chem. Soc. Rev.* **40**, 3313 (2011).
- [67] H. J. Shepherd, G. Molnár, W. Nicolazzi, L. Salmon, and A. Bousseksou, *Eur. J. Inorg. Chem.* **2013**, 653 (2013).
- [68] A. Hauser, *Chem. Phys. Lett.* **124**, 543 (1986).
- [69] M. Marchivie, P. Guionneau, J. A. K. Howard, G. Chastanet, J. F. Létard, A. E. Goeta, and D. Chasseau, *J. Am. Chem. Soc.* **124**, 194 (2002).
- [70] A. Enriquez-Cabrera, A. Rapakousiou, M. Piedrahita Bello, G. Molnár, L. Salmon, and A. Bousseksou, *Coord. Chem. Rev.* **419**, 213396 (2020).
- [71] J. Hafner, *J. Comput. Chem.* **29**, 2044 (2008).
- [72] A. I. Liechtenstein, V. I. Anisimov, and J. Zaanen, *Phys. Rev. B* **52**, R5467 (1995).
- [73] M. Mabrouk, A. Savoyant, L. Giovanelli, S. Clair, R. Hayn, and R. Ben Chaabane, *J. Phys. Chem. C* **121**, 4253 (2017).
- [74] M. Mabrouk, R. Hayn, H. Denawi, and R. Ben Chaabane, *J. Magn. Magn. Mater.* **453**, 48 (2018).
- [75] Y. Ma, Y. Dai, W. Wei, L. Yu, and B. Huang, *J. Phys. Chem. A* **117**, 5171 (2013).
- [76] H. Denawi, M. Abel, and R. Hayn, *J. Phys. Chem. C* **123**, 4582 (2019).
- [77] H. Denawi, M. Koudia, R. Hayn, O. Siri, and M. Abel, *J. Phys. Chem. C* **122**, 15033 (2018).
- [78] H. Denawi, E. Nardi, M. Koudia, O. Siri, M. Abel, and R. Hayn, *J. Phys. Chem. C* **124**, 1346 (2019).
- [79] H. Denawi, M. Abel, O. Siri, and R. Hayn, *J. Magn. Magn. Mater.* **537**, 168183 (2021).
- [80] H. Denawi, M. Abel, A. Boukortt, O. Siri, and R. Hayn, *J. Magn. Magn. Mater.* **560**, 169561 (2022).
- [81] P. E. Blöchl, *Phys. Rev. B* **50**, 17953 (1994).
- [82] G. Kresse and J. Furthmüller, *Comput. Mater. Sci.* **6**, 15 (1996).
- [83] G. Kresse and D. Joubert, *Phys. Rev. B* **59**, 1758 (1999).
- [84] G. Kresse and J. Furthmüller, *Phys. Rev. B* **54**, 11169 (1996).
- [85] H. J. Monkhorst and J. D. Pack, *Phys. Rev. B* **13**, 5188 (1976).
- [86] J. Liu and N. Lin, *ChemPlusChem* **88**, e202200359 (2023).
- [87] Q. Cui, G. Qin, W. Wang, K. R. Geethalakshmi, A. Du, and Q. Sun, *J. Mater. Chem. A* **7**, 14510 (2019).
- [88] S. Kang and J. Yu, *Phys. Chem. Chem. Phys.* **24**, 22168 (2022).