

Unusual structural rearrangement and superconductivity in infinite layer cuprate superlattices

D. Samal^{1,2,*}, Nicolas Gauquelin^{3,*}, Yayoi Takamura⁴, Ivan Lobato³, Elke Arenholz^{5,§}, Sandra Van Aert³, Mark Huijben⁶, Zhicheng Zhong⁷, Jo Verbeeck³, Gustaaf Van Tendeloo³, and Gertjan Koster^{6,||}

¹*Institute of Physics, Sachivalaya Marg, Bhubaneswar 751005, India*

²*Homi Bhabha National Institute, Anushakti Nagar, Mumbai 400085, India*

³*EMAT, University of Antwerp, Groenenborgerlaan 171, 2020 Antwerp, Belgium*

⁴*Department of Materials Science and Engineering, University of California–Davis, Davis, California 95616, USA*

⁵*Advanced Light Source, Lawrence Berkeley National Laboratory, Berkeley, California 94720, USA*

⁶*Faculty of Science and Technology, MESA+ Institute for Nanotechnology, University of Twente, 7500 AE Enschede, The Netherlands*

⁷*Key Laboratory of Magnetic Materials Devices and Zhejiang Province Key Laboratory of Magnetic Materials and Application Technology, Ningbo Institute of Materials Technology and Engineering, Chinese Academy of Sciences, Ningbo 315201, China*



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Epitaxial stabilization of thermodynamically metastable phases and advances in atomic control of complex oxide thin-film growth can be used effectively for realizing novel phenomena and as an alternative for bulk synthesis under extreme thermodynamic conditions. Here, we investigate infinite layer (IL) based cuprate superlattices, where 7–8 unit cells of $\text{Sr}_{0.6}\text{Ca}_{0.4}\text{CuO}_2$ (SCCO) are sandwiched between ultrathin spacer layers of SrTiO_3 (STO), SrRuO_3 , or BaCuO_2 (BCO) and only observe superconductivity in the pure [SCCO/BCO] superlattice (SL) without spacer layers. Apparently, the insertion of an additional STO spacer layer in the latter SL prevents the occurrence of superconductivity. The observed superconductivity in [SCCO/BCO] SL is discussed in terms of a structural model involving the interplay between the CuO_2 plane and the CuO chain similar to the bulk $\text{YBa}_2\text{Cu}_3\text{O}_7$ superconductor. The structural origin was found by the identification of a metastable IL- BaCuO_2 variant, which deviates highly from its parent bulk crystal structure and exhibits a relatively larger out-of-plane lattice parameter (around 7 Å) when sandwiched with SCCO in the form of [SCCO/BCO] SL. However, this variant is absent when STO spacer layers are introduced between SCCO and BCO layers. X-ray absorption spectra of the Cu L edge for BCO exhibits a slightly higher energy satellite peak as compared to the $3d^9L$ Zhang-Rice character observed in SCCO. This result indicates the existence of contrasting plane and chain-type Cu-O blocks in SCCO and BCO, respectively, which is further corroborated using annular bright field scanning transmission electron microscopy imaging. This work unravels an unexpected structure of BCO which helps in realizing superconductivity in [SCCO/BCO] SL and provides a wider perspective in the growth and design of cuprate-based hybrid structures.

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I. INTRODUCTION

The structural paradigm of high- T_c cuprate superconductors is built on current-carrying CuO_2 planes separated by charge-reservoir (CR) blocks (blocking/balance layers) that essentially dope these CuO_2 planes [1–4]. While the CuO_2 planes described by Cu $3d_{x^2-y^2}$ orbitals bonded to O $2p_{xy}$ orbitals in a square-planar geometry remain similar for all cuprates, the CR chemistry varies significantly from system to system. Charge carriers are generally added to the CuO_2 plane either by alteration of the CR layer through cation substitution (Sr^{2+} ions in $\text{La}_{2-x}\text{Sr}_x\text{CuO}_4$) [5], by controlling oxygen content as in the case of $\text{YBa}_2\text{Cu}_3\text{O}_{7-\delta}$ (YBCO) [6], or by electric field induced doping effect [7–10]. Particularly, in the case of YBCO, self-doping between CuO

chains and quasi-2D CuO_2 planes is at the heart of realizing superconductivity where the CuO chains are known to act as CR blocks [11,12]. The infinite layer (IL) compound $A\text{CuO}_2$ ($A = \text{Ca}, \text{Sr}, \text{or Ba}$) appears to be the simplest structure that holds one of the key ingredients (CuO_2 planes) for realizing high- T_c superconductivity. Its crystal structure can be viewed as an oxygen-deficient perovskite where each CuO_2 plane is separated by a plane of alkaline-earth-metal ions. The Cu^{2+} ions are equatorially coordinated to four oxygen ions to form a network of corner-shared CuO_4 plaquettes in a two-dimensional (2D) CuO_2 plane. A schematic of this structure is shown in Fig. 1(a). Despite its simple crystal structure, the chemical instability of IL compounds limits their synthesis to high-pressure conditions in bulk [13,14]. The IL compounds ($n = \infty$) are the end member of a series $A_{n+1}\text{Cu}_n\text{O}_{2n+2}$, to which most of the cuprate superconductors belong, and the $A_2\text{CuO}_4$ (214, $A = \text{La}, \text{Sr}, \text{Ba}$) compound is the first undoped member in this series. Mixtures of $\text{La}^{3+}:\text{Sr}^{2+}$ or $\text{La}^{3+}:\text{Ba}^{2+}$ as well as oxygen vacancies lead to the desired doped (superconducting) phases. In this structure, the AO layer can exist in a rocksalt configuration (T phase) or a fluorite-like structure (T' phase),

*These authors contributed equally to this work.

†Corresponding author: dsamal@iopb.res.in

‡Corresponding author: Nicolas.Gauquelin@uantwerpen.be

§Present address: Physical Sciences Division, Pacific Northwest National Laboratory, Richland, WA 99352, USA.

||Corresponding author: g.koster@utwente.nl

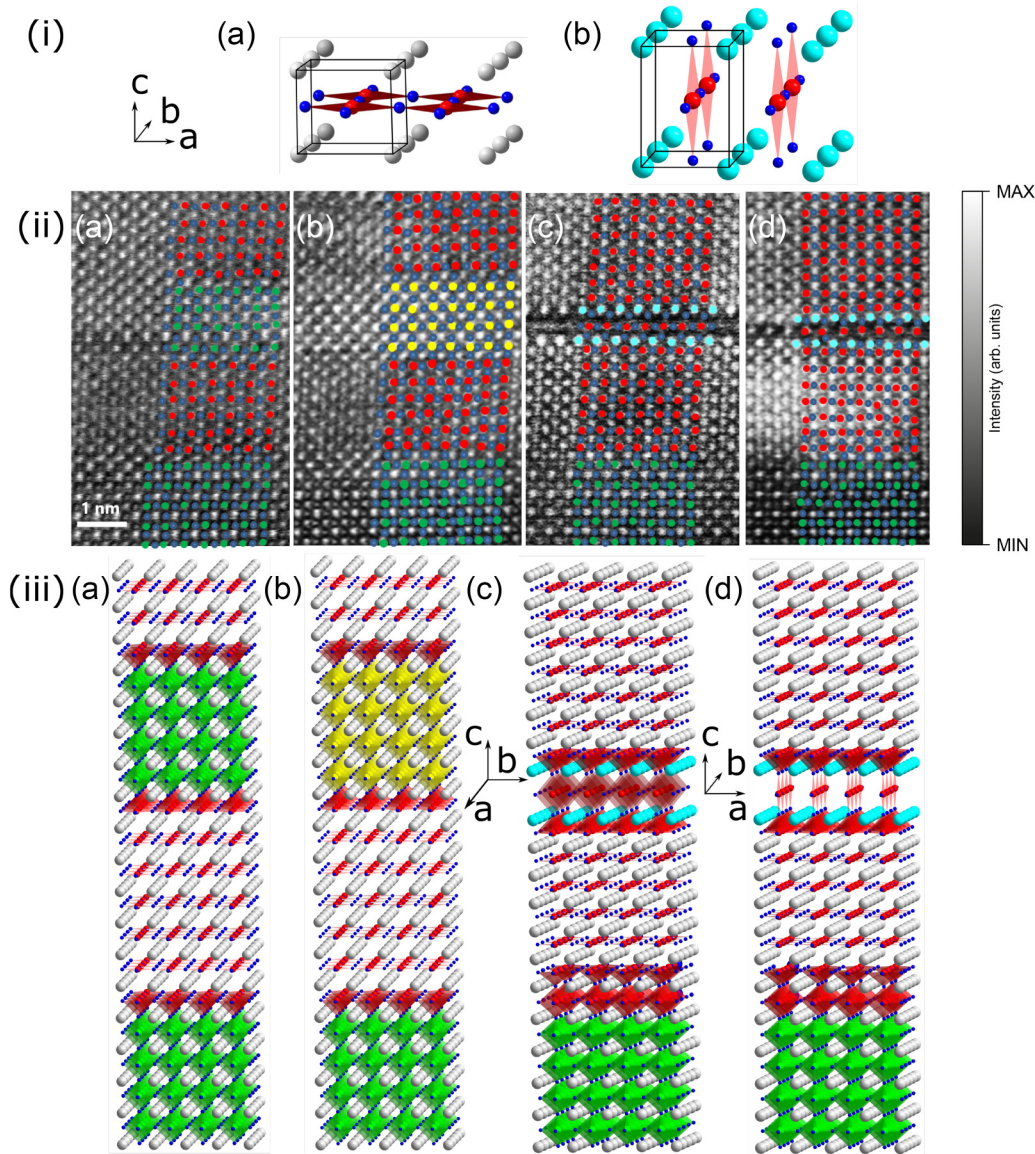


FIG. 1. Structural analysis of the SLs. (i) Structural model on the same scale of (a) the infinite layer compound (Sr, Ca) CuO₂ and (b) the expected chain-type structure of BaCuO₂. (ii) Raw inverted-contrast ABF-STEM image of (a) [SCCO/STO], (b) [SCCO/SRO], and (c) [SCCO/BCO] SLs imaged along the [100] zone axis and (d) [SCCO/BCO] imaged along the [010] zone direction. (iii) SLs with corresponding 3D structural models as determined by STEM analysis: (a) [SCCO/STO], (b) [SCCO/SRO], (c) and (d) [SCCO/BCO] (two different orientations of the [SCCO/BCO] SL showing the chain-type structure of the BCO layer along the *a* and *b* direction, respectively). The MO_{*x*} polyhedra are shown in color around the central atom (Sr/Ca atoms in white, oxygen in blue, copper in red, Ti in green, Ba in light blue, Ru in yellow, MO_{*x*} polyhedra in the color of the B = Ti, Ru, Cu central atom). The apparent skewness in the ABF images is due to scan distortions and drift of the sample during acquisition. The scanning direction is along the horizontal direction (width of the image).

whereas the Cu ion is coordinated by an elongated oxygen octahedron of which the energetics is determined by Jahn-Teller physics. Very recently, a superconducting Ba₂CuO_{4-*y*} phase (*T_c* around 73 K) was isolated with unusual high hole concentration (Cu^{*p*}, *p* ~ 2.4–2.6, which is twice that found in other cuprates) and having oxygen vacancies in the plane [15]. Furthermore, the CuO₆ octahedron in Ba₂CuO_{4-*y*} was highly compressed exhibiting a shorter Cu-O apical distance of 1.86 Å as compared to the traditional cuprates (2.4 Å for La₂CuO₄ and 1.95 Å for Sr₂CuO₃ [16]), leading to the significant admixing of *d*_{3z²-r²} orbital character near *E_F* to the usual *d*_{*x*²-*y*²} orbital. In a following discussion

in literature [17], this structural model seemed appropriate and subsequently, density functional theory calculations have captured the essential physics [16,18]. However, experimentally high pressures and strongly oxidizing conditions are required to stabilize such unique cuprates and for exploring these novel phases.

In the early 1990s, it was demonstrated that metastable tetragonal infinite layer phases could be fabricated as epitaxial strained thin films under suitable growth conditions, enabled by the *pseudomorphic* stabilizing effect of the substrates, thereby avoiding the extreme bulk synthesis conditions [19–21]. Being metastable, these films typically contain

many defects, but upon further optimization of growth conditions, the quality and reproducibility have improved. These works have paved the way for the design of layer-by-layer IL based cuprate heterostructures with the prime aim to search for high- T_c superconductivity in artificial hybrid structures as well as hints of any subtle structure-property relationship in these materials [22–26]. Even more success in the ability to design artificial structures has been demonstrated in layered systems using existing superconductors, such as $(\text{LaSr})_2\text{CuO}_4$ and the Bi cuprates as building blocks [7,27–29] in atomic layering strategies based on molecular beam epitaxy. Next to technological advances in atomic control of artificial structure design, advancement in scanning transmission electron microscopy (STEM) instrumentation and techniques such as aberration correction of the probe-forming lens (condenser) has made it possible to achieve subangstrom sized probes to study materials in an atomic-column by atomic-column fashion. High-angle annular dark field (HAADF) combined with annular bright field (ABF) imaging have made it possible to study and understand the structural ordering of the cationic (due to its contrast being proportional to the atomic number of the atom) and oxygen sublattices in these materials, respectively [30–34].

In the context of recent research on IL cuprates, the study by Samal *et al.* [35] on SrCuO_2 demonstrated a remarkable change in the oxygen sublattice, i.e., a change of oxygen position from an equatorial to an apical site (from a planar to a chain-type structure) upon lowering the film thickness below about 5 unit cells (u.c.). The schematic structure of this “chain-type” structure is shown in Fig. 1(b). This phenomenon was attributed to an electrostatic instability associated with the polar nature of IL cuprates that consists of opposite charged atomic planes stacked along the c direction [35,36]. The experimental realization that the geometry of the CuO_4 plaquette can be controlled precisely from an in-plane to out-of-plane configuration at the atomic scale [35,37–39] in thin-film structures opens the possibility to design structures with specific functionalities, e.g., current-carrying action layer and/or charge-reservoir balance layer, without resorting to chemical doping/substitution. Remarkably, the observations by Di Castro *et al.* [40,41] of high-temperature superconductivity (up to 40 K) at the interface between CaCuO_2 and SrTiO_3 (STO) insulating oxides in $\text{CaCuO}_2/\text{STO}$ superlattices (SLs) grown in highly oxidizing conditions has drawn the attention of the field. Their work hypothesizes that under highly oxidizing conditions, extra oxygen ions can be incorporated in the interfacial Ca plane, providing an apical oxygen site for the Cu ions and hosts holes in the CuO_2 plane. The role of the CR block in this case was attributed to a reconstructed CaO_x composition of the Ca plane at the interface. Regarding other possible effects such as relative band alignment at the CaCuO_2 (CCO)/STO interface and the electrostatic built-in potential induced charge transfer to interface due to polar character of CCO was also invoked to account for any doping effects that can give rise to superconductivity in $\text{CaCuO}_2/\text{STO}$ SLs. However, the hard x-ray photoelectron spectroscopy study by Aruta *et al.* [42] revealed that direct charge transfer between CCO and STO bands is not feasible under the band alignment condition. Further it was found that the built-in electrostatic potential is suppressed even for

a three-unit-cells-thick CCO block by oxygen redistribution in the alkaline-earth-metal interface planes. Therefore, both these above effects do not contribute to the realization of superconductivity through doping. But under a highly oxidizing growth condition, the oxygen coordination at the interface may be increased, resulting in hole doping of a cuprate block and thus the appearance of superconductivity. Similar effects of superoxygenation have shown increased T_c in YBCO-based films [43,44]. Furthermore, the recent discovery of superconductivity in doped infinite layer nickelate epitaxial thin films makes the structure even more interesting [45,46]. These studies indicate that instead of high-pressure bulk synthesis, thin-film epitaxy can be effectively used to stabilize thermodynamically metastable phases for manipulating their electronic properties more precisely at the atomic scale.

In this study, we investigated the oxygen coordination of Cu ions and the occurrence of superconductivity for a series of artificially grown hybrid IL- BaCuO_2 SLs. We have fabricated SLs by combining single unit cell BaCuO_2 (BCO) layers with 8 u.c. plane-type $\text{Sr}_{0.6}\text{Ca}_{0.4}\text{CuO}_2$ (SCCO) layers in which the role of an additional ultrathin spacer layer of the band insulator STO is studied. These $[(\text{Sr}_{0.6}\text{Ca}_{0.4}\text{CuO}_2)_8/\text{STO}_{m=0,2}/(\text{BCO})_1/\text{STO}_{m=0,2}]_{10}$ SLs were grown by pulsed laser deposition (PLD). As reference samples containing no Ba cations, we also fabricated $[(\text{Sr}_{0.6}\text{Ca}_{0.4}\text{CuO}_2)_8/(\text{SrRuO}_3)_4]_{10}$ and $[(\text{Sr}_{0.6}\text{Ca}_{0.4}\text{CuO}_2)_8/(\text{STO})_4]_{10}$ SLs. The SrRuO_3 (SRO) layer was chosen to be 4 u.c. thin, which is the critical limit for itinerant ferromagnetic character in SRO thin films, below which it tends to become insulating [47,48]. Even if the CuO_2 planar structure in the SCCO layers is preserved in all SLs, we observe the occurrence of superconductivity in the case of $[(\text{SCCO})_8/(\text{BCO})_1]_{10}$ SL, in contrast to the insulating behavior observed for $[(\text{SCCO})_8/(\text{STO})_4]_{10}$ and $[(\text{SCCO})_8/(\text{SRO})_4]_{10}$. Further, when the close proximity between the SCCO and BCO layers is disrupted by inserting an ultrathin STO spacer layer as in the case of $[(\text{Sr}_{0.6}\text{Ca}_{0.4}\text{CuO}_2)_8/\text{STO}_{m=2}/(\text{BCO})_1/\text{STO}_{m=2}]_{10}$, the superconductivity disappears. In a more general context, the prime goal of this study was to design IL cuprate-based SLs that can be driven into the superconducting state by manipulating the sublayer structural/electronic characteristics, and subsequently elucidate the possible physical mechanism. By employing electron microscopy combined with x-ray absorption spectroscopy, we identified that the BCO layers have a Cu-O chain configuration (oxygen vacancy in the CuO_2 planes), and a remarkable larger out-of-plane lattice parameter (around 7 Å) when present in nonbuffered [SCCO/BCO] SL. Note that earlier a similar kind of structure was noticed by Koster *et al.* [26] where the BaCuO_2 phase was converted to Ba_2CuO_x during pulsed laser interval thin-film deposition. However, x-ray absorption spectroscopy and transport measurements were never performed on those structures to explore its origin.

II. EXPERIMENT

A. Thin-film growth

The SLs were fabricated on (001)-oriented TiO_2 -terminated STO substrates by a reflection high-energy

electron diffraction (RHEED) assisted PLD (KrF laser with $\lambda = 248$ nm) at a pulse rate of 1 Hz, and a laser fluence of 2 J/cm^2 . The most homogeneous part of the laser beam was selected using a 4×15 mm rectangular mask and an image of the mask was created on the stoichiometric targets with a lens. Before deposition the target was preblasted for 2 min at a pulse rate of 5 Hz and laser fluence 2 J/cm^2 . The substrate temperature during the growth was set at 650°C . *In situ* RHEED was used to monitor the growth. The observed RHEED intensity oscillations shown in Fig. S1 of the Supplemental Material (SM) [49] during growth indicate the successful control on the unit cell scale due to the layer-by-layer growth mode. To grow a complete stack of SL structures, the number of laser pulses required for each layer of a specific material was calibrated and then the growth was performed in accordance with the periodicity of the SL structure. Streaky 2D RHEED patterns were observed during the film growth (Fig. S2 in the SM [49]). After the deposition, the samples were cooled in a high oxygen pressure (~ 1 bar). Two growth recipes were applied: one set of samples was grown in an oxygen pressure of 0.3 mbar, while the second set was grown in a mixture of oxygen and $\sim 5\%$ ozone at a pressure of about 0.3 mbar (hereafter abbreviated as O_3 grown) to enhance the oxidation. The O_3 -grown samples were subsequently annealed *ex situ* at 375°C for 2 h in a mixture of oxygen and $\sim 5\%$ ozone at a pressure of 0.3 mbar (hereinafter abbreviated as O_3 annealed).

B. Structural characterization

The structural characterization on the SLs was carried out by x-ray diffraction and STEM. STEM was performed on the X-Ant-Em instrument at the University of Antwerp. Two independent cross-sectional samples were prepared along the [100] and [010] directions of the STO substrate to investigate in-plane anisotropy using a FEI Helios 650 dual-beam focused ion beam device and exposure to oxygen was avoided using a Kammrath and Weiss transfer box and a Gatan vacuum transfer holder as proved successful on atmosphere-sensitive materials [50–53]. All the SLs under STEM investigation were O_3 annealed. An ~ 25 -nm gold layer was sputtered on top of samples prior to the preparation of the TEM cross-section specimen in order to prevent a change of the oxygen stoichiometry during the preparation process. Satisfactory samples were prepared using very low-energy ion beam thinning subsequent to a protection of the sample surface by sputtering of a 10-nm-thick carbon protection layer, followed by e-beam deposition of platinum as a first step to the focused ion beam lamella preparation procedure. The FEI Titan G3 electron microscope was equipped with an aberration corrector for the probe-forming lens as well as a high-brightness gun operated at 200 kV acceleration voltage with a beam current of around 20 pA for all experiments to minimize beam damage. The STEM convergence semiangle used was 21 mrad, providing a probe size of $\sim 0.8 \text{ \AA}$. The collection of semiangles ranges from 11 to 29 mrad and 29 to 160 mrad for ABF and annular dark field imaging, respectively. No further data processing was applied and the raw data is presented. The line scans presented in Fig. 2 were obtained by using

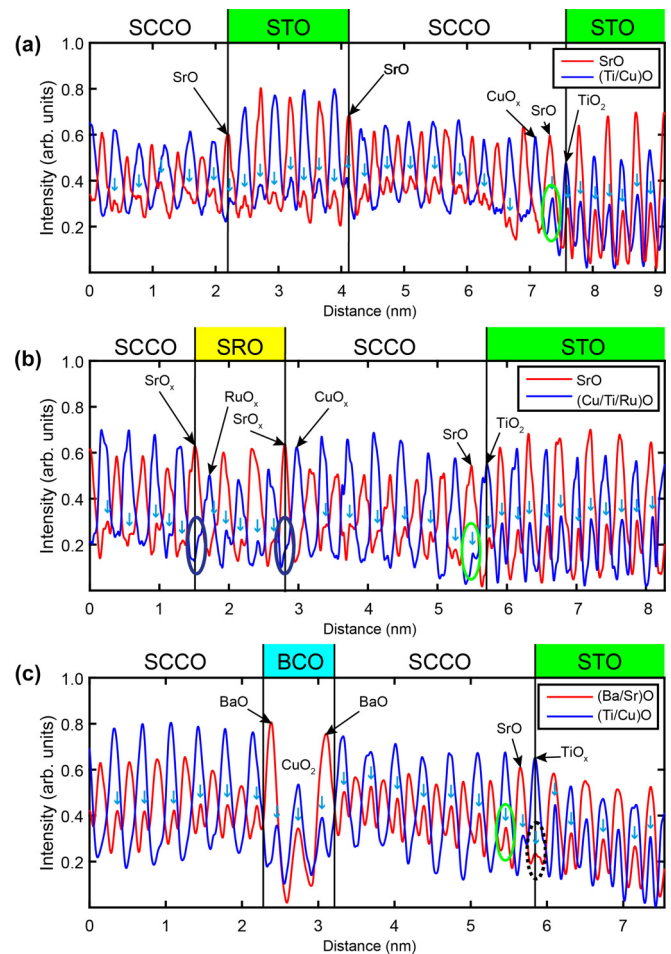


FIG. 2. Comparison of line scans over the STEM–ABF images averaged across 20 of each type of atomic layer on the Sr/Ca/Ba-O (in red) and Ru/Cu/Ti-O (in blue) planes of the (a) [SCCO/STO], (b) [SCCO/SRO], and (c) [SCCO/BCO] SLs, respectively, showing the layering and chemistry in each layer. Oxygen atomic positions are marked with blue arrows. Green circles show the presence of oxygen in the first SrO_x layer of the SCCO, dark blue circles show the low oxygen content of the SrO_x planes at the interfaces between SRO and SCCO, and the black dashed circle shows the oxygen depletion in the last TiO_{2-x} plane of the STO substrate for the [SCCO/BCO] SL. For each SL, the first $(\text{Sr}, \text{Ca})\text{O}_x$ plane of the first SCCO layer contains oxygen on top of the terminal TiO_2 layer of the STO substrate. For all SLs, the planar structure of SCCO is evidenced by the absence of oxygen in the blue profile for the Sr layers. The red (peaks) planes not directly labeled in the STO, SCCO, and SRO are, respectively, SrO, (Sr/Ca), and SrO. The blue (peaks) planes not directly labeled in the STO, SCCO, and SRO are, respectively, TiO_2 , CuO_2 , and RuO_2 .

the following procedure: first all atomic-column positions in the STEM–ABF image were identified and later refined by performing a Gaussian fitting [54]. Next, all atomic column positions of each atomic layer were identified and grouped by similar planes according to A (in red) or B (in blue) sites [Fig. S5(b) in the SM [49]] of the perovskite structure and used to obtain an integrated line scan through the image by using a window of 3 pixels (75 picometers, i.e., similar size to the electron probe) for each line. Finally, the integrated line scans were averaged over each type of atomic layer. Using

this averaging, the intensity profile obtained can be used to reveal nonstoichiometry on the oxygen-containing planes and making the contrast more trustworthy and more interpretable as it is proportional to Z^γ , in which γ is a real number between 1.0 and 2.0 depending on the microscope settings [55].

C. Electrical transport measurements

Transport measurements was performed using a four-probe van der Pauw method with ultrasonic bonding of aluminum wire as electrodes, in a Quantum Design physical property measurement system.

D. X-ray absorption measurements

X-ray absorption (XA) spectra at room temperature were acquired at beamline 4.0.2 at the Advanced Light Source in total electron yield mode by monitoring the sample drain current. To avoid charging effects during the measurement, SLs were grown on Nb-doped (0.05 at. %) STO substrates. In our experiment, the linearly polarized x rays were incident upon the sample in a grazing incidence geometry, and the E vector was applied either in plane ($E_{\perp c}$ axis) or canted out of plane by a 30 deg angle relative to the [001] direction ($\sim E \parallel c$ axis) of the sample, as schematically represented in Fig. 6(a).

III. RESULTS AND DISCUSSION

The designed structures $[(\text{SCCO})_8/(\text{STO})_4]_{10}$, $[(\text{SCCO})_8/(\text{SRO})_4]_{10}$, and $[(\text{SCCO})_8/(\text{BCO})_1]_{10}$ show distinct satellite peaks as seen from x-ray diffraction (Fig. S3 in the SM [49]), which demonstrates the structural quality of the SLs. According to STEM measurements, the roughness of the layer thicknesses is equal to ± 1 u.c., which is in agreement with state-of-the-art PLD growth [56–58] and the structure of each SL is discussed separately in the following section. A model with colored dots overlaid on Figs. 1(ii)(a)–1(ii)(d) is detailed in the caption, and for clarity, a three-dimensional (3D) structural model for each SL derived from STEM analysis is shown in Figs. 1(iii)(a)–1(iii)(d). For the sake of brevity, we represent $[(\text{SCCO})_8/(\text{STO})_4]_{10}$, $[(\text{SCCO})_8/(\text{SRO})_4]_{10}$, and $[(\text{SCCO})_8/(\text{BCO})_1]_{10}$ SLs as [SCCO/STO], [SCCO/SRO], and [SCCO/BCO], respectively, in the succeeding section. Figure 2 presents the profiles of intensity for the three SLs taken from a region where no defects were observed over an area of a few unit cells wide in the ABF image. As described in the methods part, the contrast from the ABF image is used for localization of the oxygen column and not for identifying the heavy atom columns Sr/(Sr,Ca)/Ba (red profiles) or Ru/Ti/Cu (blue profile). For simplicity, the Sr and Ca layers of the SCCO will be assimilated and named as Sr layers since we cannot determine the Ca content of those layers from intensity. Pure oxygen atomic positions are marked with blue lines and used as an indication of the oxygen content. Green circles show the presence of oxygen in the first SrO_x layer of the SCCO layer. For each SL, the first (Sr, Ca) O_x plane of the first SCCO layer contains oxygen on top of the TiO_2 -terminated layer of the STO substrate. For all SLs, the planar structure of the SCCO is evidenced by the absence of oxygen in the blue profile for the Sr layers.

The [SCCO/STO] SL structure is represented in Figs. 1(ii)(a) and S4(a) in the SM [49]. The SCCO and STO layers grow nicely in an epitaxial fashion and have thicknesses in the range of 7–8 and 3–4 u.c., respectively. According to the STEM measurements presented in Fig. 2(a), the interfacial SrO layer between the SCCO and STO layers are rich in oxygen since both SCCO/STO interfaces contain oxygen. The planar structure of the SCCO layers and their oxygen content is confirmed by the contrast in the ABF-STEM images [Fig. 2(a)] (no contrast linked to oxygen atomic columns in apical position in the ABF image).

Figures 1(ii)(b) and S4(b) in the SM [49] represent the crystal structure of the [SCCO/SRO] SL. We can notice that the SCCO and SRO layers have thicknesses in the ranges 7–8 and 3–4 u.c., respectively, although some Ruddlesden-Popper defects (not shown) do exist perpendicular to the film. Furthermore, some waviness of the SRO layers [as seen in Fig. S4(b) in the SM [49]] is observed mainly in the third SRO layer, which might be linked to the nonhomogeneous character of the SCCO layers as the presence of chain- and planelike structures are observed in some regions. The CuO_2 region within the SCCO layer has on average a planar character, but we can notice the presence of some contrast on the oxygen columns in-between the infinite layers, i.e., in the SrO_x planes [suggesting the possible presence of higher coordination (five to six) Cu sites]. Contrary to the previous case, the interfacial SrO_x layer between SCCO and SRO layers seems slightly underoxygenated as can be seen in Fig. 2(b) (dark blue circles). This statement should be taken with caution due to the nonlinear contrast dependence of ABF.

Figures 1(ii)(c) and 1(ii)(d), and S4(c) in the SM [49] represent the crystal structure of the [SCCO/BCO] SL. We can notice that the SCCO and BCO layers have thicknesses in the ranges 7–8 and 1–2 u.c., respectively. Many steps can be seen as shown in Figure S5(a) (red arrows) of the SM [49] suggesting that the two-u.c. thicknesses are an optical illusion due to the projection character of STEM imaging. The BCO layer is mostly 1 u.c. thick and the SCCO has the expected planar structure with CuO_2 planes running along the film as in the case of [SCCO/STO] SL. The structure of the ultrathin BCO layer seen here is far from the expected bulk structure, nonetheless a few conclusions can be made: the CuO_2 has chain configuration as confirmed by ABF imaging (O presents only when looking along the [100] direction of BCO). Surprisingly, the out-of-plane spacing between the oxygen ions from the BaO plane in the BCO layer is rather large (around 7 Å) and the Ba atoms are displaced 0.5 Å farther than the oxygen on each side of the central CuO chain (Fig. 3). This result is similar to the buckling of the BaO plane observed in YBCO [59,60]. In fact, according to the structural refinement of YBCO by x-ray crystallography, the Ba atom is vertically displaced by 0.32 Å (0.499 Å) relative to the apical O for fully oxygenated (underoxygenated) YBCO [61,62]. The origin of this incredibly large out-of-plane lattice spacing observed in the BCO layer will be the subject of our future work. The interface with the substrate is SrO_{1-x} for all samples, as some oxygen ions are present within the Sr plane of the first unit cell of the first SCCO layer inducing a slight overstoichiometry as shown by green circles in Fig. 2(c). However, we can notice a slight depletion in oxygen of the first TiO_{2-x}

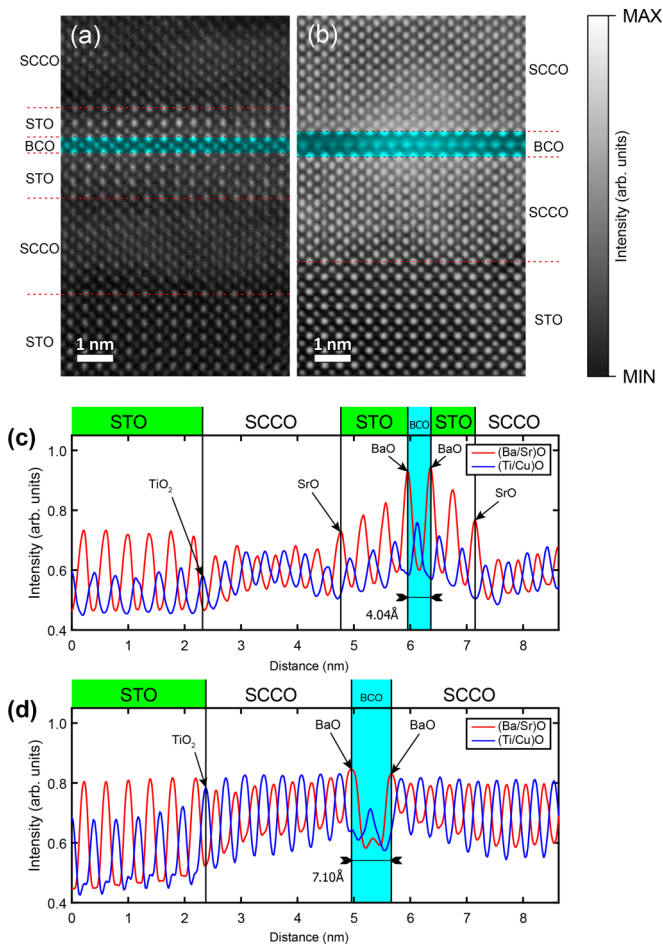


FIG. 3. STEM-HAADF images resulting from averaging 20 low-dose images of the (a) [SCCO/STO/BCO/STO] and (b) [SCCO/BCO] SLs with the corresponding averaged line profiles [(c) and (d)] acquired as described for Fig. 2. The intensity is displayed as gray scale and is proportional to the average atomic number in each column. We can easily note the difference in the out-of-plane lattice spacing of the BCO layer between the STO buffered insulating case (c) and the nonbuffered superconducting case (d). This difference could be related to the structure difference between a perovskite environment (c) and the chain-type structure (d). The red (peaks) planes not directly labeled in STO, SCCO, and BCO are, respectively, SrO, (Sr/Ca), and BaO. The blue (peaks) planes not directly labeled in STO, SCCO, and BCO are, respectively, TiO₂, CuO₂, and CuO₂.

plane of the substrate for the [SCCO/BCO] SL (dashed black circle). Statements about stoichiometry should be taken as an indication due to the projection character of the technique and the sample preparation step during which the stoichiometry of the sample can be altered [11,56]. Lastly, on the buffered [SCCO/BCO] SL where a spacer STO layer is inserted in-between the SCCO and BCO layers, we can observe the out-of-plane lattice parameter for the BCO layer to be 4.04 Å [Fig. 3(c)] in contrast to the nonbuffered [SCCO/BCO] SL in which it is found to be 7.10 Å [Fig. 3(d)]. Electron energy loss spectroscopy (EELS) was used for chemical characterization and to show the good epitaxial quality of the complex SL stacking consisting of a [SCCO (8 u.c.)/STO (2 u.c.)/BCO (1 u.c.)/STO (2 u.c.)] nominal composition (each interface at a

roughness of ± 1 u.c.) as shown in Fig. S6 of the SM [49], and this SL hereinafter is abbreviated as [SCCO/STO/BCO/STO]. To avoid strong beam damage, the data was acquired with a 69 mrad collection angle, a 19 mrad convergence angle, and an acceleration voltage of 80 kV with a beam current of <40 pA and an acquisition time of 100 ms/pixel. We suggest that due to the proximity of the SrO planes of the STO layer, the BCO layer stoichiometry might be closer to BaCuO_{3-x} inducing a shrinkage of the lattice parameter in the growth direction, but this is in the process of being verified using *ab initio* calculations. Even in this complex architecture, the SCCO layer still presents the same planar structure as in all other cases presented above.

In Fig. 4, we show the electrical transport properties of the designed SLs. The [SCCO]/STO SLs show relatively higher sheet resistance (R_S) as compared to the [SCCO/SRO] SL. In addition, with O₃ growth, R_S drops significantly for both kinds of SLs as compared to those grown under O₂ process gas, which could be related to a relative increase of the oxygen content due to O₃ activation. With O₃ annealing, R_S drops further, but no signature of any superconducting transition is observed down to 2 K. This is unlike the case reported by Di Castro *et al.* [40] where superconductivity was observed in CaCuO₂/STO SLs synthesized under high pressure and strong oxidizing conditions. Interestingly, a spectacular electronic effect emerges when the STO or SRO spacer layer in the designed heterostructure is replaced by a chain-type BCO layer. Figure 4(b) shows the results from transport study on [SCCO/BCO] SLs synthesized under various conditions. The salient features observed are as follows: (i) the [SCCO/BCO] SL grown under O₂ process gas exhibits the onset of superconductivity around 15 K; (ii) the [SCCO/BCO] SL grown under a mixture of O₂ and O₃ process gas (earlier abbreviated as O₃ grown) shows enhanced metallic conduction compared to that grown under O₂ process gas; however, the superconducting transition remains similar to that of [SCCO/BCO] SL; (iii) with O₃ annealing, the metallic behavior is further enhanced and the onset of the superconducting transition temperature increases up to ~ 50 K; however, the superconducting transition region gets widened as can be seen from temperature-dependent $R_S(T)/R_S(80\text{ K})$ and (dR_S/dT) plots in Fig. S7 of the SM [49]. This result suggests spatial inhomogeneities in the charge-carrier distribution [63] or to a certain crystallographic disorder across different SCCO/BCO interfaces after O₃ annealing. Further, it is to be noted from Fig. 4(b) that there is an enhancement of sheet resistance close to the onset of superconducting transition which could be related to thermodynamic fluctuations of the single-electron density of states (normal phase) at the Fermi level close to T_c that suppresses the one-electron conductivity [64,65]. As expected, the application of a magnetic field suppresses the onset of superconductivity from 50 to ~ 42 and 37 K with $H = 3$ and 9 T, respectively, as shown in Fig. 5(a). To understand if the close proximity of the chain-type BCO layer with the plane-type SCCO layer, as determined above by STEM imaging, is crucial to promote superconductivity in the designed hybrid structure, we investigated SLs where the BCO and SCCO layers were separated by 2 unit cells of STO i.e.,

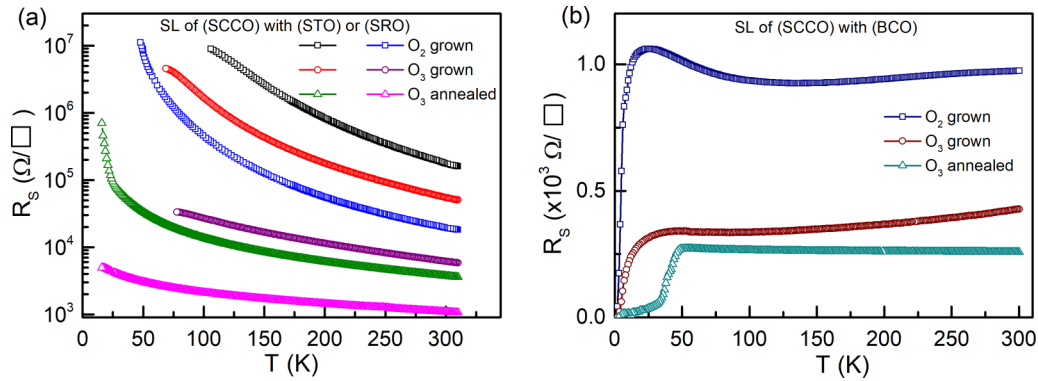


FIG. 4. Temperature-dependent sheet resistance for (a) $[(\text{SCCO})_8/(\text{STO})_4]_{10}$, $[(\text{SCCO})_8/(\text{SRO})_4]_{10}$, and (b) $[(\text{SCCO})_8/(\text{BCO})_1]_{10}$ SLs under various conditions (O_2 grown, O_3 grown, and O_3 annealed; respectively, squares, circles, and triangles).

$[(\text{SCCO}(8 \text{ u.c.})/\text{STO}(2 \text{ u.c.})/\text{BCO}(1 \text{ u.c.})/\text{STO}(2 \text{ u.c.}))]_{10}$, which is abbreviated earlier as $[\text{SCCO}/\text{STO}/\text{BCO}/\text{STO}]$. Surprisingly, no superconductivity is observed in $[\text{SCCO}/\text{STO}/\text{BCO}/\text{STO}]$ SL; rather, it is found to be insulating [Fig. 5(b)]. This observation points to the fact that BCO and SCCO layers must be in direct contact for the superconductivity to be observed. The results from Di Castro *et al.* [41] on $\text{CaCuO}_2/\text{SrTiO}_3$ hybrids revealed that the doping of holes as a function of the distance from the CR interface is confined to within about 1–2 u.c. Therefore, it seems pertinent that incorporating a 2-u.c.-thick spacer layer between the SCCO and BCO building blocks could prohibit the observed superconductivity. Further, it is to be noted that if the superconductivity would have arisen from BCO chains only, then it would have possibly been seen in $[\text{SCCO}/\text{STO}/\text{BCO}/\text{STO}]$ SL (although BCO in this case has a different out-of-plane lattice parameter than that in superconducting $[\text{SCCO}/\text{BCO}]$ SL). From transport measurements it is not straightforward to infer specifically as to which layer hosts superconductivity in $[\text{SCCO}/\text{BCO}]$ SL. However, through a detailed structural and x-ray absorption spectroscopy investigation, it is realized that SCCO and BCO layers adopt distinct plane- and chain-type structural arrangement. Based on the conventional wisdom of cuprate physics related to the interplay between chain- and plane-type blocks, we conjecture that a direct coupling between the adjacent planelike SCCO and chain-type BCO layers in

$[\text{SCCO}/\text{BCO}]$ SL is necessary to induce superconductivity more likely in the SCCO planes than in the BCO chains.

To obtain information about the orbital occupancy at Cu $3d$ and O $2p$ sites that could provide insight about the origin of the observed superconductivity, we performed soft XA spectroscopy measurements on various (O_3 -annealed) cuprate SLs. XA spectra probes the density of unoccupied states by exciting electrons from core shell to an empty valence state upon irradiation with x rays of the appropriate resonance energy and they are sensitive to orbital occupancy, symmetry, valence state, and charge transfer/carrier doping. By changing the direction of the x-ray polarization (E vector) of the linearly polarized x ray relative to the sample surface, one can probe the angular dependence of the empty valence states. Ideally, the Cu^{2+} ion ($3d^9$ configuration) in a square-planar geometry hosts a single hole in the $3d_{x^2-y^2}$ orbital as shown schematically in the inset of Fig. 6(b). To elucidate the hole symmetry at the Cu site in the planar SCCO layer, we examined the polarization dependence of spectra taken at the Cu $L_{2,3}$ edge that correspond to the transitions $2p_{1/2}^6 3d^9 \rightarrow 2p_{1/2}^5 3d^{10}$ (at about 930 eV) and $2p_{3/2}^6 3d^9 \rightarrow 2p_{3/2}^5 3d^{10}$ (at about 950 eV), respectively [35,40,66–69], for $[\text{SCCO}/\text{STO}]$ and $[\text{SCCO}/\text{SRO}]$ SLs as shown in Figs. 6(b) and 6(c). The polarization effect shows a much stronger absorption when E is aligned in plane as compared to when it is parallel to the growth direction of the sample. The observed large asymmetry

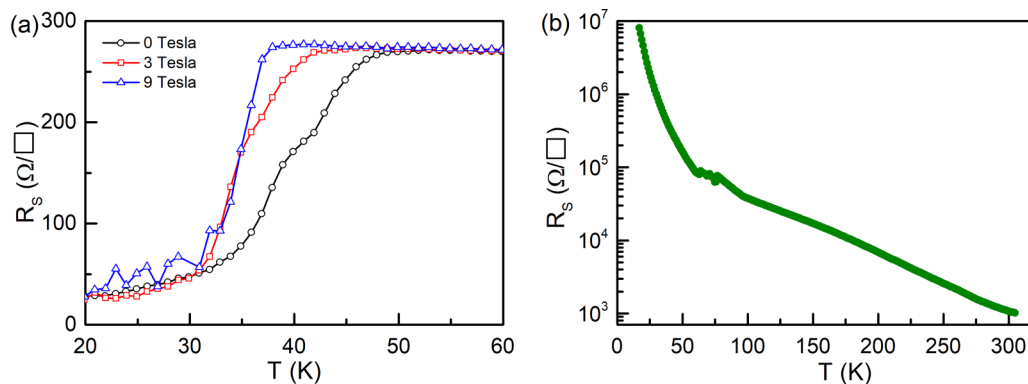


FIG. 5. Temperature-dependent sheet resistance for (a) $[(\text{SCCO})_8/(\text{BCO})_1]_{10}$ SL (O_3 annealed) under the application of a magnetic field. (b) Zero field temperature-dependent sheet resistance for $[(\text{SCCO})_8/(\text{STO})_2/(\text{BCO})_1/(\text{STO})_2]_{10}$ SL (O_3 annealed).

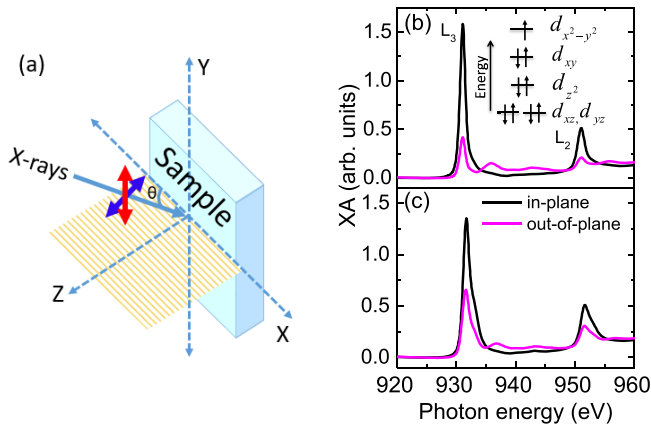


FIG. 6. Schematic (a) XA measurement geometry. XA spectra for (b) [(SCCO)₈/(STO)₄] and (c) [(SCCO)₈/(SRO)₄] SLs, respectively. The inset in (b) schematically shows the square-planar crystal-field energy levels with electronic occupation at the Cu site in the SCCO layer.

of $\sim 57\%$ and $\sim 35\%$ for [SCCO/STO] and [SCCO/SRO] SLs, respectively [asymmetry = $(I_{\parallel} - I_{\perp})/I_{\parallel} + I_{\perp}$] where I_{\parallel} and I_{\perp} represent the absorption intensity at the L edge measured with polarizations $E_{\perp c}$ axis and $E_{\parallel c}$ axis of the film plane, respectively) suggests that most of the holes are present in the $d_{x^2-y^2}$ orbital at the Cu sites in the SCCO layer. This result implies a square-planar structure of the CuO₂ layer and is in agreement with the STEM results. Further,

the less pronounced asymmetry for the [SCCO/SRO] SL as compared to the [SCCO/STO] SL can be related to the presence of higher coordination (five to six) Cu sites as established from the STEM study. Besides the central white line L_3 peak around 930 eV corresponding to the $2p_{1/2}^6 3d^9 \rightarrow 2p_{1/2}^5 3d^{10}$ transition, a second significantly weaker peak is observed ~ 4.6 eV higher in energy (more pronounced for the $E_{\perp c}$ axis), which is attributed to the small content of monovalent Cu ions ($3d^{10}$), similar to the studies on Cu₂O [70] and oxygen-deficient YBa₂Cu₃O_x [11,68,71] where it is attributed to the $3d^{10} \rightarrow 3d^9 4s^1$ transition.

To discern the subtle electronic structure changes between the SLs involving a BCO layer and non-BCO layers, we show a magnified view of the fine structure in close vicinity to the L_3 peak (Fig. 7). In addition to the L_3 peak associated with the undoped Cu site, a slightly higher energy satellite characteristic peak around 932.7 eV is observed for [SCCO/BCO] and [SCCO/STO/BCO/STO] SLs, while only a broad shoulderlike feature around 932.5 eV is observed for [SCCO/SRO] SL (for [SCCO/STO] SL this shoulderlike feature is not that prominent). The feature around 932.5 eV is a signature for the Zhang-Rice singlet (ZRS), associated with a transition from $2p^6 3d^9 L \rightarrow 2p^5 3d^{10} L$, where L denotes an additional oxygen ligand hole [11,25,67,69]. The higher energy state corresponding to the ZRS as compared to the $3d^{10}$ transition is attributed to the interaction between the ligand and the core hole that raises the energy required to promote the core electron to the unoccupied state [67]. The spectral intensity corresponding to

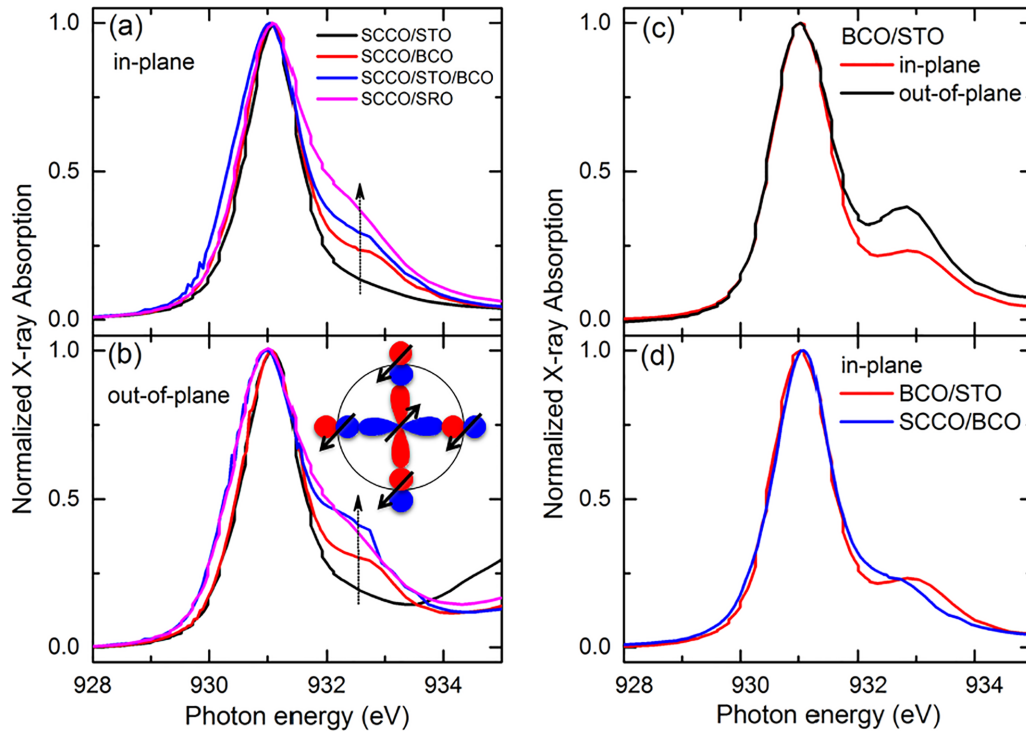


FIG. 7. XA spectra normalized to the maximum height intensity of the L_3 peak for [(SCCO)₈/(STO)₄], [(SCCO)₈/(BCO)₁], [(SCCO)₈/(STO)₂/(BCO)₁/(STO)₂], and [(SCCO)₈/(SRO)₄] SLs for the polarization in the (a) in-plane and (b) out-of-plane directions. The arrow indicates the region of the ligand hole state. The inset shows the schematic of the ZRS structure. (c) XA spectra of the L_3 peak for the [(BCO)₁/(STO)₃] SL in both polarization directions. (d) Difference in the spectral intensity shape of the L_3 peak belonging to ligand states between the coupled (chain and plane) [SCCO/BCO] SL and the chain-type only BCO layer.

ZRS sensitively depends on the specific arrangement of the Cu-O network and doping level. Since the [SCCO/BCO] and [SCCO/STO/BCO/STO] SLs were made out of two distinct cuprate blocks (SCCO and BCO), it was difficult to accurately distinguish the characteristic ligand hole contribution from each sublayer assuming a few-nanometer probing depth for oxides by XA measurements. Clarification of this connection was obtained by examining a [(BCO)₁/(STO)₃] SL hybrid as shown in Fig. 7(c). Clearly, in addition to the main $3d^9$ -related peak (nominal hole at the undoped Cu²⁺ site), we observe a stronger satellite peak at a slightly higher energy as discussed above for the [SCCO/BCO] and [SCCO/STO/BCO/STO] SLs and is analogous to the observation made in slightly underdoped YBCO compounds. We can further notice a polarization dependence of this peak similar to what is observed in YBCO [68,71,72] (more along the *c* axis than the *b* axis) revealing the $3d_{z^2-y^2}$ (chain-type) character of the BCO layer as deduced from the STEM images (Figs. 1 and 3). The presence of a slightly higher energy satellite peak in the chain-type compound has already been evidenced by Gauquelin *et al.* [11] in STEM-EELS experiments. Thus, we can decompose the complex multielectron states surrounding the $3d^9$ -related peak in [SCCO/BCO] and [SCCO/STO/BCO/STO] SLs into $3d^9 + 3d^9 L^{-1} + 3d^8 L^{+1}$, where the ligand hole configurations L^{+1} and L^{-1} as proposed by Magnuson *et al.* [12] represent for chains (additional electron on the oxygen) and planes (missing electron on the oxygen), respectively. It has to be noted that while the electronic state surrounding the $3d^9$ -related peak for the BCO layer with having a higher energy satellite peak and chain-type configuration (STEM structural analysis) can be attributed to a $3d^9 + 3d^8 L^{+1}$ configuration (equivalent to the CuO chain in YBCO), the same for the SCCO layer with having broad shoulder and the planelike configuration (STEM structural analysis) can be attributed to a $3d^9 + 3d^9 L^{-1}$ configuration corresponding to the ZRS (equivalent to the CuO₂ planes in YBCO). We, therefore, conclude that the self-doping at the interface between the BCO and SCCO sublayers might trigger the observed superconductivity in [SCCO/BCO] SL. Finally, we would like to mention that in the context of high-*T_c* cuprates, the hybridization between Cu $d_{x^2-y^2}$ and in-plane ligand orbitals ($O 2p_{xy}$) govern the essential physics. However, we note the signature of the ZRS structure for both polarizations ($E \perp c$ and $E \parallel c$) indicating the presence of both in-plane and out-of-plane ligand hole characteristics. Although the out-of-plane ligand holes have already been observed in some of the high-*T_c* cuprates, their role realizing superconductivity remains an important question/outstanding problem [25,66,71], and was rather thought to be detrimental for superconductivity. Remarkably, the work by Aruta *et al.* [25] on (Cu_{1- δ} C _{δ})Ba₂CuO_{*x*}/CaCuO₂ superconducting SLs demonstrated that there exists a relatively high density of out-of-plane holes from polarized XA spectra. These holes originating from the out-of-plane orbitals ($O 2p_z$) associated with the apical oxygen in the CR block (Cu_{1- δ} C _{δ} Ba₂CuO_{*x*}) are doped into CuO₂ planes of IL CaCuO₂ layer at the interface. However, when CaCuO₂ thickness in the above SL structure was increased, the superconductivity got suppressed, possibly because of a lack of apical oxygen ions over several CuO₂ atomic planes in the CaCuO₂ layer. There-

fore, it appears that itinerant holes with out-of-plane ($O 2p_z$) ligand symmetry are relevant for realizing superconductivity in cuprates. Indeed, a recent study on superconducting Ba₂CuO_{4- δ} [15], with having a considerable admixture of $d_{3z^2-r^2}$ orbital character, apparently suggest that the holes with out-of-plane orbital character can also cooperate to produce superconductivity.

IV. CONCLUSIONS

Based on our experimental evidence, we identify that IL-BaCuO₂ crystal when sandwiched with planar IL-SCCO layers in the form of a [SCCO/BCO] SL, deviates highly from its parent bulk crystal structure. We find it to have chain-type structure with a relatively larger out-of-plane lattice parameter (around 7 Å). However, with the insertion of stable perovskite STO in-between the SCCO and BCO layers, the out-of-plane lattice parameter for the BCO layer turns out to be 4.04 Å (IL bulk $c_{\text{BCO}} \sim 3.888$ Å) as expected for a compressive in-plane strained (bulk $a_{\text{STO}} < \text{bulk } a_{\text{BCO}}$) and tensile out-of-plane strained IL-BaCuO₂ layer. The distinct plane- and chain-type structural arrangement of SCCO and BCO layers is evident from STEM and XA investigation. Remarkably, the [SCCO/BCO] SL is found to exhibit superconductivity; on the contrary, this effect disappears when the close proximity between SCCO and BCO layers is disrupted by employing a thin STO spacer layer. Further, ([SCCO/STO] and [SCCO/SRO]) SLs do not exhibit any sign of superconductivity. All these observations seemingly point to the fact that a direct coupling between the adjacent planelike SCCO and chain-type BCO layers is necessary to induce the self-doping responsible for the observed superconductivity. This is analogous to the structural model of YBCO where the current-carrying CuO₂ planes are doped by the adjacent charge-reservoir CuO chains. In the context of cuprate hybrid structures, it has been reported [41] that hole doping is confined to about 1–2 u.c. from the charge-reservoir interface. Therefore, it is pertinent to assume that insertion of an insulating 2-u.c.-thick spacer layer between the SCCO and the BCO building blocks prohibits the occurrence of superconductivity as reflected in the present study. In a nutshell, by exploiting the structural and electronic properties of cuprate blocks at sublayer level, we demonstrate the possibility to design a synthetic heterostructure that hosts superconductivity. The stabilization of an unexpected structure of BCO subject to the sequencing of layers in the superlattice structure is an important revelation in the growth and design of IL cuprate-based hybrid structures for future research and calls for further investigation to understand the underlying thermodynamics behind such structural transformation in ultrathin limit.

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