



High-throughput search for potential permanent magnet materials

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High-performance permanent magnets have a wide range of applications in various fields of the information age. One distinct feature of permanent magnet materials is significant magnetic anisotropy, which is mainly affected by spin-orbit coupling (SOC). Focusing on materials containing 3d transition elements with specific Wyckoff positions, where certain partially occupied orbital multiplets can significantly enhance the effect of SOC, we perform a highly efficient search for permanent magnet materials in the inorganic crystal structure database. According to common standards of permanent magnets, we identify 19 potential permanent magnet materials. Among these candidates, 14 materials have already been discussed in previous studies, and we finally propose five new permanent magnet candidates. As examples, the detailed magnetic properties of two candidates among these five new proposed materials are presented. We believe that these potential permanent magnet materials deserve further experimental study.

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I. INTRODUCTION

Permanent magnets are widely used in many areas such as transportation, energy, information and communication technology, which are required for the implementation of a large number of technologies, such as electric vehicle motors and generators, windmills, speakers, and relays [1,2]. The main requirements of these materials are large saturation magnetization, high Curie temperature (T_C), and large uniaxial magnetic anisotropy [1,2]. In the past few decades, some representative high-performance permanent magnets materials were discovered and widely used (such as the Nd-Fe-B alloy [3–6], SmCo₅ [7,8]). Most of these materials contain rare-earth elements [3,9]. However, there is a growing need to reduce or eliminate the use of rare-earth magnets, which are relatively expensive, by finding rare-earth-free alternatives [1,2]. Therefore, various rare-earth-free permanent magnet materials were proposed recently, including bulk materials [10–13], nanostructure [14–19], and thin films [20–22], but it is still desirable to explore new high-performance permanent magnet materials containing cheaper and less critical elements [1,2].

High-throughput density functional theory (DFT) methods, which can efficiently screen large numbers of compounds, were widely used in past work recently. For example, it has been successfully performed for the discovery of non-magnetic topological materials and thousands of topological materials candidates were proposed [23–25]. However, the *ab initio* calculation of magnetic topological compounds would be difficult to characterize the ground state when spin-orbit coupling (SOC), which plays an important role in topological properties [26], typically leads to complex

magnetic structures. Meanwhile, unlike in nonmagnetic systems, the Coulomb interaction is of substantial importance in most magnetic systems and the Coulomb repulsion is usually incorporated by the parameter U in first-principles calculations [27]. The first-principles predictions for magnetic materials usually depends on the value of U [28,29]. Therefore, compared to the time-reversal-invariant nonmagnetic topological materials, the predictions on magnetic topological materials are relatively few [30]. The same as the case of magnetic topological materials, both the SOC effect and Coulomb interaction are also important for permanent magnetic materials, which makes the efficient search of permanent magnetic materials difficult. At present, people usually perform comprehensive high-throughput DFT methods for rare-earth-free permanent magnetic materials based on their respective screening criteria and range [31–35]. Kusne *et al.* performed high-throughput computing and machine learning to search for potential high-performance permanent magnets in Fe-Co- X (X is the transition-metal element) alloys system and suggested Fe₇₈Co₁₁Mo₁₁ to be a rare-earth-free permanent magnet [31]. Meanwhile, Vishina *et al.* presented an application of high-throughput and data-mining approach to predict new permanent magnets containing both 3d and 5d transition-metal elements based on the Inorganic Crystal Structure Database (ICSD) [36], and the most promising candidates they identified were Pt₂FeNi, Pt₂FeCu, and W₂FeB₂ [32]. Furthermore, they made an extended search for materials containing two different 3d elements in the chemical formula and predicted one rare-earth-free permanent magnet candidate Co₃Mn₂Ge finally [33].

Generally, the 3d transition metal materials would provide high saturation magnetization and large Curie temperature [32,38], meeting two main requirements of permanent magnets. However, it is commonly believed that SOC does not play an essential role in 3d transition metal materials

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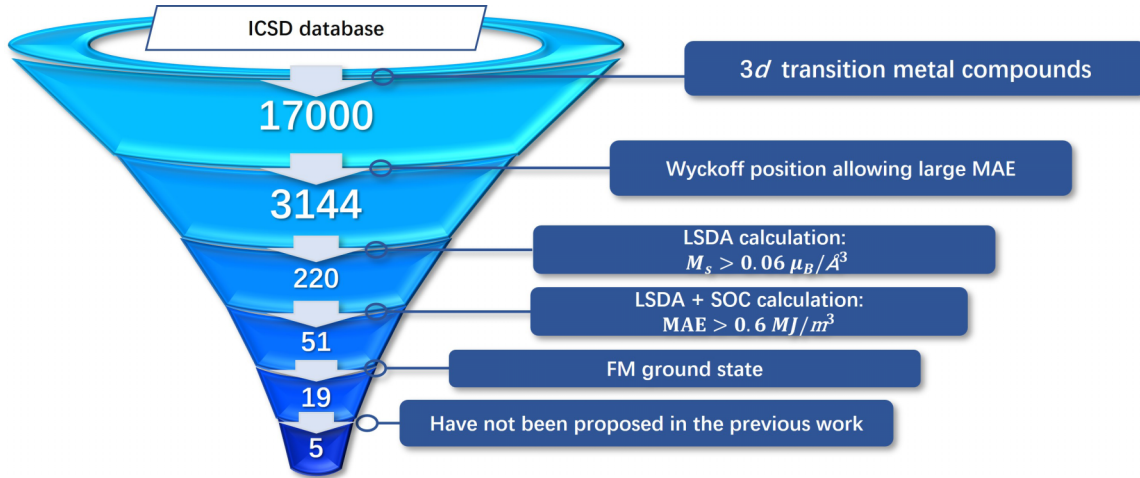


FIG. 1. Steps of high-throughput search and screening with the number of structures left after each step. In the first step of screening, 17 000 compounds containing 3d transition metal elements are selected from the ICSD database. Second, we extract these materials in the corresponding space groups, where the 3d transition elements correspond to the specific Wyckoff position allowing large SOC [37]. A total of 3144 different materials are identified. Third, LSDA calculations with FM configuration are performed and 220 materials with the calculated $M_s > 0.06 \mu_B/\text{\AA}^3$ are preserved. At the fourth step, LSDA + SOC calculations are performed and 51 candidates meet the filter criteria $MAE > 0.6 \text{ MJ/m}^3$. Finally, we perform LSDA + SOC (+ U) calculations by considering several different conventional magnetic configurations with different typical magnetic directions to explore their magnetic ground state. Materials with AFM configuration are excluded and 19 FM candidates are left, which are listed in Tables I and VI. Among them, five new candidate materials were not discussed before as candidates of permanent magnets to our best knowledge.

[38,39], resulting in the relatively small magnetic anisotropy energy (MAE). To overcome this shortcoming, the compounds containing both 3d and 5d transition-metal elements were explored in the previous work [32], which significantly limited the search range of potential permanent magnets materials. It was proposed that, due to certain partially occupied orbital multiplets, the cooperative effect of the electron correlation can significantly enhance the effective SOC in light 3d transition metal ions [37]. Therefore, one can screen 3d transition metal materials with potentially huge MAEs by simply checking whether the materials belong to the corresponding space groups and whether the 3d transition elements occupy the specific Wyckoff positions [37].

In this work, we perform a high-throughput screening for potential permanent magnet materials in ICSD [36]. We focus on 3d transition metal materials allowing large SOC according to the certain partially occupied orbital multiplets, and first-principles calculations are then performed to estimate the saturation magnetization, MAE, and magnetic ground state of the screened materials. According to recognized standards of permanent magnets [1,2], we suggest 19 materials to be potential candidates. While 14 materials among them were already discussed in previous studies, we finally obtain five new potential permanent magnets materials FeB_4 , FeB_2 , Fe_2Sc , Mn_5PB_2 , and Fe_2Ge , which are worthy of further experimental investigations. In addition, we present the detailed magnetic properties in two of the new permanent magnet candidates FeB_4 and FeB_2 as representatives.

II. STEPS OF HIGH-THROUGHPUT SCREENING

As mentioned above, both the SOC effect and Coulomb interaction U play important roles in permanent mag-

nets, which makes it difficult to perform a comprehensive high-throughput first-principles search based on all 3d transition metal materials with various magnetic configurations, spin orientations, and different values of the Coulomb parameter U . Here we present a highly efficient search for permanent magnetic materials. This screening is mainly divided into five steps (see Fig. 1) in the following.

(1) During the first step of the material collection stage, we select 3d transition metal materials from the ICSD database [36]. As mentioned above, compared to 4d and 5d elements, 3d elements usually provide high saturation magnetization and large Curie temperature [32,38], and the materials containing 4d and 5d transition elements usually providing magnetism (4d: from Nb to Rh, 5d: from Re to Ir) are excluded. To avoid missing potential materials, we selected all experimental and theoretical materials in the ICSD database. Meanwhile, structures containing rare-earth elements which are expensive were also excluded. There are a total of 17 000 materials after deduplication at the first step.

(2) Second, the 3d transition metal materials allowing large MAE are selected. When there are certain partially occupied orbital multiplets in 3d materials, the SOC effect could be enhanced by Coulomb correlation [37]. For a d -shell ion exposed to the crystal field, only four types of orbital multiplets are allowed by the crystallographic symmetries, including three doublets $E_1 \{d_{xz}, d_{yz}\}$, $E_2 \{d_{xy}, d_{x^2-y^2}\}$, $E_3 \{d_{z^2}, d_{x^2-y^2}\}$, and one triplet $T \{d_{xz}, d_{yz}, d_{xy}\}$ [40]. Among them, the first-order SOC effect is absent in E_3 , therefore the magnetic anisotropy for the orbital multiplet E_3 should be small. Meanwhile, in the orbital triplet $T \{d_{xz}, d_{yz}, d_{xy}\}$, though the first-order perturbation of the SOC splits the degenerate orbitals, the perturbed energies are independent of the orientation of the magnetization, indicating that the

TABLE I. New proposed materials by screening with desirable properties for permanent magnets. Listed are the material's ICSD, formula, space group, the corresponding 3d transition elements and Wyckoff positions, the calculated MAE (MJ/m³), saturation magnetic moment M_s ($\mu_B/\text{\AA}^3$) and magnetic hardness parameter κ .

ICSD number	Material formula	Space group	Transition elements	Wyckoff position	MAE (MJ/m ³)	M_s ($\mu_B/\text{\AA}^3$)	κ
633 415	Fe ₂ Sc	194	Fe	2 a	0.925	0.0653	1.42
670 859	FeB ₄	194	Fe1	2 b	2.397	0.0693	2.15
			Fe2	2 c			
613 892	FeB ₂	191	Fe	1 a	5.714	0.0973	2.36
109 111	Mn ₅ PB ₂	140	Mn	1 a	0.617	0.1119	0.68
53 460	Fe ₂ Ge	194	Fe	2 a	0.669	0.1219	0.65

first-order SOC effect on the magnetic anisotropy for this orbital triplet T is negligible. Therefore, we select compounds with 3d transition elements corresponding to the specific Wyckoff position allowing E_1 or E_2 orbital multiplets. At this step, about 81.5% of the materials are excluded and 3144 materials are left.

(3) Saturation magnetization (M_s) is one of the most important criteria of the permanent magnet. In this step, we perform local spin-density approximation (LSDA) calculations for the remaining 3144 materials with ferromagnetic (FM) configuration to obtain their M_s . Note that Coulomb interaction usually enhances the magnetic moments in 3d magnetic systems [38,39], therefore, our filter criteria $M_s > 0.06 \mu_B/\text{\AA}^3$ [2] from LSDA calculations is strictly to avoid missing potential materials. Meanwhile, SOC usually have little effect on the magnetic moments of 3d magnetic ions [39], thus in this step SOC is also not considered to reduce the computational cost. This screening step leaves about 220 candidates.

(4) In the fourth step, we perform LSDA + SOC calculations with FM configuration to obtain MAE of these potential candidates. Usually one needs to estimate the MAE by performing first-principles calculations with spin orientations along a set of different typical directions. However, for certain partially occupied orbital multiplets in 3d materials [37], the prediction of the easy axis and hard axis of magnetization can be obtained by symmetry analysis. Here we take the degenerate $E_1 \{d_{xz}, d_{yz}\}$ orbitals in the crystal field as an example, where the SOC interaction could be written as $\lambda L \cdot S$. Since

spin splitting typically overwhelms the SOC effect in 3d systems, we treat SOC as a perturbation with the spin-conserved part only and then the orbital degeneracy is split into states $|l = 2, m = \pm 1\rangle$. When the orbital doublet is half-filled, the occupied state takes the energy of $-\frac{1}{2}\lambda\hbar^2|\cos\theta|$ [41], where θ is the angle between the magnetic moment and the z -axis. The energy of the occupied state is minimized when $\theta = 0$ or π , making out-of-plane magnetization favored. Similarly, the MAE for magnetic materials with half-filled orbital doublet $E_2 \{d_{xy}, d_{x^2-y^2}\}$ is $-\lambda\hbar^2|\cos\theta|$, which also favors the (001) direction to be the easy axis [41]. It is worth mentioning that Coulomb correlation increases the orbital polarization and the effective SOC when the orbital multiplets are partially occupied [37]. With the reinforcement of the orbital polarization and SOC effect, the magnetocrystalline anisotropic energy can significantly increase. Based on LSDA + SOC calculations, the MAE is calculated by the total energy differences between spin orientations along the easy axis and hard axis, and the results of MAE are double-checked by using the force theorem [42,43]. The calculated results based on these two methods are usually similar and we show the ones by energy differences approach if not specified in the following. Note that the Coulomb correlation could further enhance the SOC effect, resulting in a significant gain of the MAE [37]. Therefore, we adopt the appropriate filter criteria to be MAE > 0.6 MJ/m³ [2], and in this step 51 materials are left.

(5) Finally, the magnetic structures of these remaining 51 candidates are studied. We perform LSDA + SOC (+ U) calculations by considering several different conventional

TABLE II. Relative total energies (E_{tot} in meV/f.u.) of different magnetic ordering states from LSDA (+ U) and LSDA + SOC (+ U) with (001) spin orientation. Here total energy of FM configuration is set to be 0.

	U (eV)	FM	AFM-1	AFM-2	AFM-3	FIM-1
$E_{\text{tot}}(\text{LSDA})$	0.0	0.00	279.54	667.07	457.89	383.41
	0.5	0.00	342.39	663.51	483.29	362.90
	1.0	0.00	368.06	561.34	467.67	309.25
	1.5	0.00	379.13	430.70	470.95	250.38
	2.0	0.00	397.80	368.47	358.63	192.95
$E_{\text{tot}}(\text{LSDA} + \text{SOC})$	0.0	0.00	275.61	660.72	456.15	382.36
	0.5	0.00	336.39	632.09	589.25	360.88
	1.0	0.00	359.67	558.92	466.92	308.23
	1.5	0.00	393.34	435.06	482.85	256.38
	2.0	0.00	412.86	385.72	382.36	203.42

TABLE III. Spin exchange parameters (in meV) of FeB₄ evaluated from LSDA (+ U) calculations. Meanwhile, the calculated Curie temperature (in K) based on MFT and MC method with different values of U are also summarized in the table.

	Distance (Å)	LSDA	LSDA + U (= 0.5 eV)	LSDA + U (= 1.0 eV)	LSDA + U (= 1.5 eV)	LSDA + U (= 2.0 eV)
J_1	2.961	-55.59	-52.79	-46.78	-39.25	-29.89
J_2	5.128	-9.31	-15.05	-19.80	-22.54	-21.94
J_3	5.128	-6.23	-5.77	-3.57	-1.86	-1.70
J_4	3.290	-69.88	-85.60	-92.02	-94.78	-99.45
T_C (MFT)		1546	1758	1797	1755	1664
T_C (MC)		945	1186	1217	1227	1126

magnetic configurations with spin orientations along the predicted easy axis to explore their magnetic ground state. Meanwhile, the effect of Coulomb interaction parameter U is also considered. All materials known or suggested to be antiferromagnetic (AFM) are then excluded. Among them, 19 candidates whose magnetic ground states are suggested to be FM remain as potential rare-earth-free permanent magnet materials, which are listed in Tables I and VI and will be discussed in the results section.

III. METHOD

The first-principles calculations were carried out by using the full potential linearized augmented plane-wave method as implemented in the WIEN2K package [44]. The k mesh of the Brillouin zone is set by $60/a$ along each direction, where a denotes the length of the lattice constant in Å. The self-consistent calculations are considered to be converged when the difference in the total energy of the crystal does not exceed 0.01 mRy. We adopt the local spin-density approximation (LSDA) [45] as the exchange-correlation potential and include the SOC using the second-order variational procedure [46]. Meanwhile, the effect of Coulomb repulsion is taken into account by utilizing the LSDA + U scheme [27].

Monte Carlo (MC) simulations are performed with the METROPOLIS algorithm for the Heisenberg model [47–49]. The size of the cell in the MC simulation in both FeB₄ and FeB₂ are $16 \times 16 \times 16$ -unit cells with periodic boundary conditions. At each temperature we carry out 400 000 sweeps to prepare the system and sample averages are accumulated over 800 000 sweeps.

IV. RESULTS

Through our high-throughput search above, we propose 19 candidate materials in the ICSD database that meet the properties of permanent magnets as summarized in Tables I and Appendix. Among these candidates, 14 materials were already discussed in previous studies as potential permanent magnet applications. Some of them were already confirmed by experimental works, such as Fe-Pt alloys [50,51], Fe₁₆N₂ [52–55], and Co₅Y [56–58]. We list these known permanent magnet materials in Appendix, and it can be seen that our calculated results are in good agreement with previous works, which also confirms the high efficiency and reliability of our screening method. Finally, to the best of our knowledge, five new candidates are found by our high-throughput calculation and listed in Table I.

As shown in Table I, these five candidates are in three space groups $I4/mcm$ (SG 140), $P6/mmm$ (SG 191), and $P6_3/mmc$ (SG 194). Based on LSDA calculations, their saturation magnetic moments are around 0.06–0.12 $\mu_B/\text{Å}^3$, which are larger than one of most famous permanent magnets Nd₂Fe₁₄B (0.04 $\mu_B/\text{Å}^3$) [5,6]. Meanwhile, their MAE based on LSDA + SOC calculations are distributed from 0.6 to 5.7 MJ/m³, where FeB₄ and FeB₂ have the relatively larger MAE (2.4 MJ/m³ for FeB₄ and 5.7 MJ/m³ for FeB₂), which has the same order of magnitude as Nd₂Fe₁₄B (4.8 MJ/m³) [5,6]. Moreover, the magnetic hardness parameter could be evaluated from the expression $\kappa = \sqrt{E_{\text{MAE}}/\mu_0 M_s^2}$ and $\kappa > 1$ is an empirical criterion for a material to have a chance of resisting self-demagnetization when made into any possible shape [2]. Their magnetic hardness parameters are distributed from 0.6 to 2.3. According to the screening results, a pair of iron-boron materials (FeB₂ and FeB₄) are included in the candidates. Considering the possible potential of Fe-B alloys as permanent magnet materials, we also extended the calculation of all Fe-B alloys existing in the ICSD database. Their space groups, MAE, and Wyckoff positions occupied by Fe ions are listed in Table VII of the Appendix. In the other Fe-B alloys in the ICSD database, the anisotropy of Fe single ions in these alloys is significantly lower than that of the two screened materials, which also demonstrate the reliability and efficiency of our screening method. It is worth mentioning that FeB₄ comes from theoretical work [59] and warrants further experimental investigations. Besides, there has been experimental work [60] on the magnetic properties of Fe₂Sc, which is suggested to have a large MAE. We will present FeB₄ and FeB₂ in detail and discuss their electronic structure and magnetic properties in the following.

TABLE IV. Relative total energies (E_{tot} in meV/f.u.) of different magnetic ordering states from LSDA (+ U) and LSDA + SOC (+ U) with (001) spin orientation. Here total energy of FM configuration is set to be 0.

	U (eV)	FM	AFM-1	AFM-2	FIM-1
E_{tot} (LSDA)	0.0	0.00	26.08	57.27	45.14
	0.5	0.00	65.13	93.06	76.07
	1.0	0.00	103.41	127.65	98.61
E_{tot} (LSDA + SOC)	0.0	0.00	23.94	57.12	45.00
	0.5	0.00	61.99	93.41	76.77
	1.0	0.00	99.76	127.05	98.29

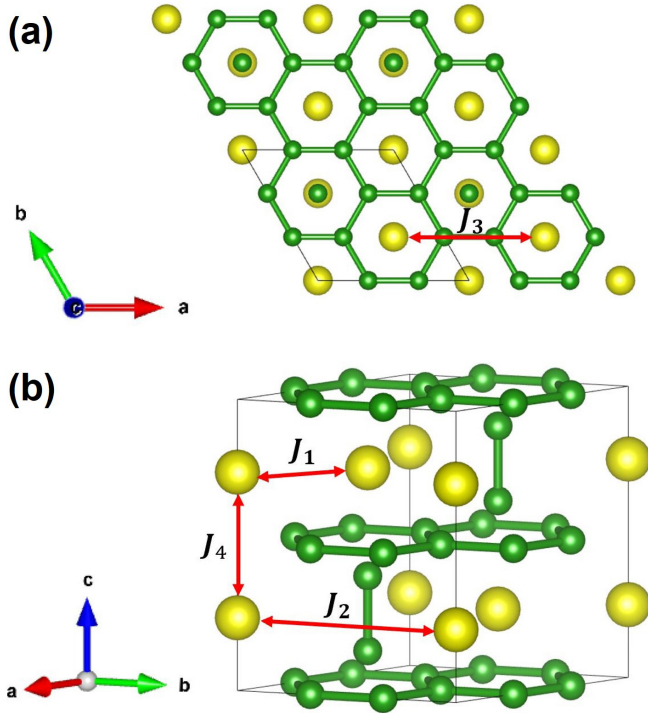


FIG. 2. (a). Top view of FeB₄. The yellow and green balls represent Fe and B, respectively. (b). The exchange interactions J_1 , J_2 , J_3 , and J_4 are shown. Within (001)-planes, J_1 denote the nearest interactions between Fe1 and Fe2, J_2 denote the nearest interactions between Fe1 ions, J_3 denote the nearest interactions between Fe2 ions. J_4 represent the couplings between two nearest (001)-planes.

A. FeB₄

As shown in Fig. 2, FeB₄ crystallizes in the hexagonal structure ($P6_3/mmc$, SG 194) [59], where Fe atoms located at two nonequivalent crystallographic sites: Fe1 atoms occupy the $2b$ position: $(0, 0, 1/4)$, and Fe2 atoms occupy the $2c$ position: $(1/3, 2/3, 1/4)$. The site-symmetry group of both Fe ions are all $(-62m)$, which permits the E_1 and E_2 orbital multiplets and is expected to have large MAE.

From LSDA calculations for the FM configuration, FeB₄ is suggested to be a metal. The band structure from LSDA

calculation is shown in Figs. 3(a) and 3(b), where the main contribution around the Fermi level comes from the $3d$ orbitals of Fe ions. The calculated magnetic moments on the Fe1 and Fe2 sites are $2.66 \mu_B$ and $2.42 \mu_B$, respectively, indicating $M_S = 0.0693 \mu_B/\text{\AA}^3$. According to the site symmetry of Fe ions, we perform the LSDA + SOC calculations for (001) and (100) spin directions and estimate the MAE as 2.397 MJ/m^3 with the easy-axis magnetization along the (001) direction as predicted. The band structure from the LSDA + SOC calculation with (001) spin direction is shown in Fig. 3(c). In addition to the energy differences approach, we also calculate the MAE by using the force theorem [42,43] to be about 2.409 MJ/m^3 , which is well consistent with the value from energy differences, confirming the reliability of our symmetry analysis and the accuracy of our first-principles calculations. The value of Coulomb parameter U in iron borides was fixed at 0.4 eV in the previous work [61], and we consider the effect of Coulomb interaction with the appropriately wider range of U from 0.0 to 2.0 eV . As shown in Fig. 3(d), the MAE is significantly enhanced by the correlation effect in FeB₄ as expected [37].

Finally, to confirm its FM ground state, we consider several different magnetic configurations of FeB₄, as shown in Fig. 4. In addition to the FM configuration, we also consider three AFM states and one ferrimagnetic (FIM-1) state: AFM-1, where the Fe atoms couple antiferromagnetically along the c axis; AFM-2, where Fe1 and Fe2 have opposite spin orientations; AFM-3, where both the Fe1 and Fe2 atoms couple antiferromagnetically along the a and b axes; FIM-1 states, where the Fe1 atoms couple ferromagnetically while the Fe2 atoms couple antiferromagnetically along the a and b axes. As shown in Table II, whether SOC is considered, the FM configuration has the lowest total energy with different values of U . Note that SOC has little effect on the calculated energy differences as shown in Table II, and the calculated magnetic moments for the different magnetic configurations are similar (with a difference around than $\pm 0.1 \mu_B$), allowing us to estimate the Heisenberg exchange couplings by the energy-mapping analysis [62,63] based on LSDA (+ U) calculations in the following. As shown in Fig. 2, we consider four magnetic interactions labeled as J_1 – J_4 here. The total energies per unit cell for these five spin configurations are

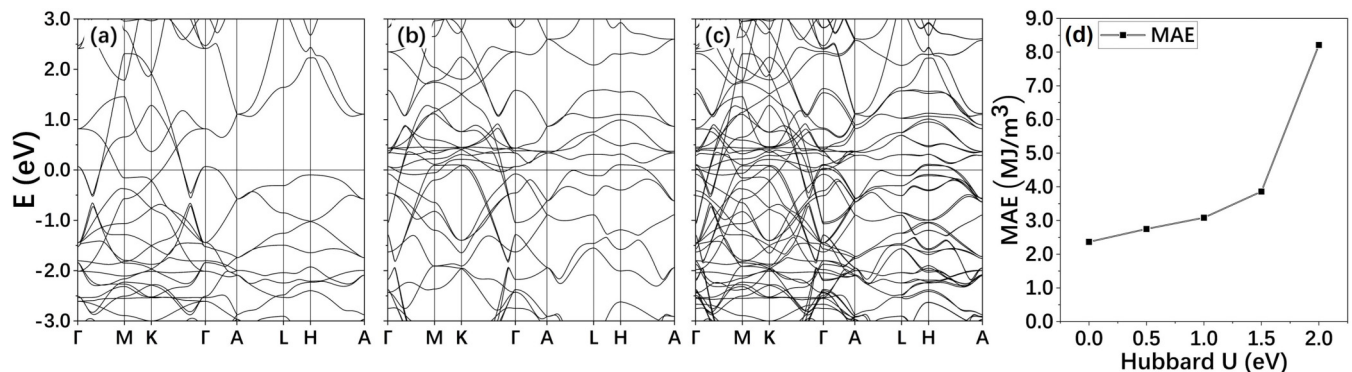


FIG. 3. (a,b) Electronic structure of FeB₄ from LSDA calculation with FM configuration for the spin-up and spin-down channels, respectively. (c) Electronic structure of FeB₄ from LSDA + SOC calculation with spin orientations along the (001) direction. (d) The calculated MAE value with increasing U .

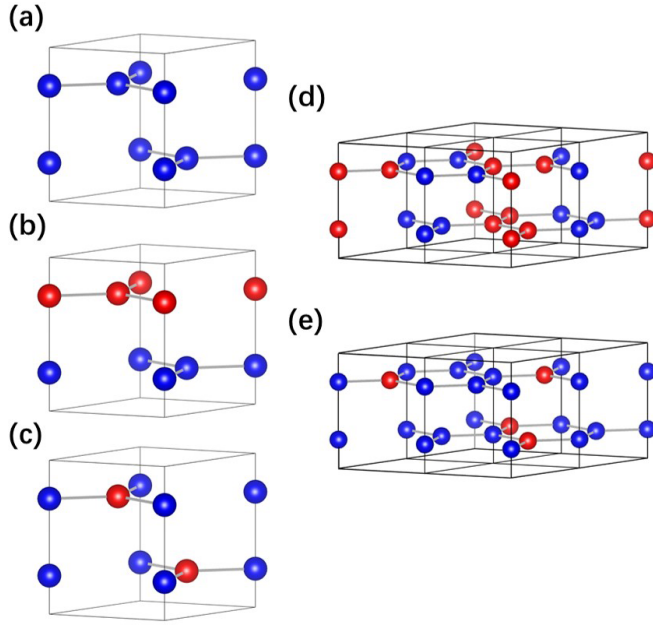


FIG. 4. The magnetic configurations of FeB_4 which we considered in DFT calculations. For clarity only Fe atoms are shown, while the blue and red ones represent Fe ions with up and down spin, respectively. (a)–(e) FM, AFM-1, AFM-2, AFM-3, and FIM-1 configurations, respectively.

expressed as

$$\begin{aligned}
 E_{\text{FM}} &= E_0 + 6J_1 + 6J_2 + 6J_3 + 2J_4, \\
 E_{\text{AFM-1}} &= E_0 + 6J_1 + 6J_2 + 6J_3 - 2J_4, \\
 E_{\text{AFM-2}} &= E_0 - 6J_1 + 6J_2 + 6J_3 + 2J_4, \\
 E_{\text{AFM-3}} &= E_0 - 2J_2 - 2J_3 + 2J_4, \\
 E_{\text{FIM-1}} &= E_0 + 6J_2 - 2J_3 + 2J_4.
 \end{aligned}
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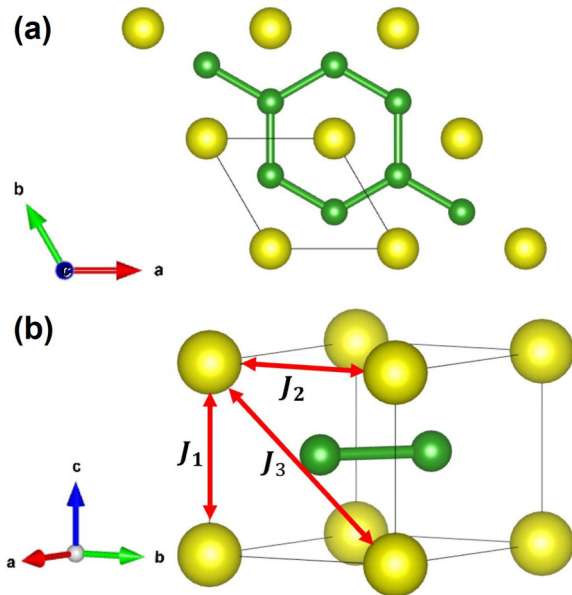


FIG. 5. (a) Top view of FeB_2 . The yellow and green balls represent the Fe and B, respectively. (b) The nearest-neighbor, nearest-neighbor exchange, and third-nearest neighbor interactions for Fe magnetic moments are shown as J_1 , J_2 , and J_3 , respectively.

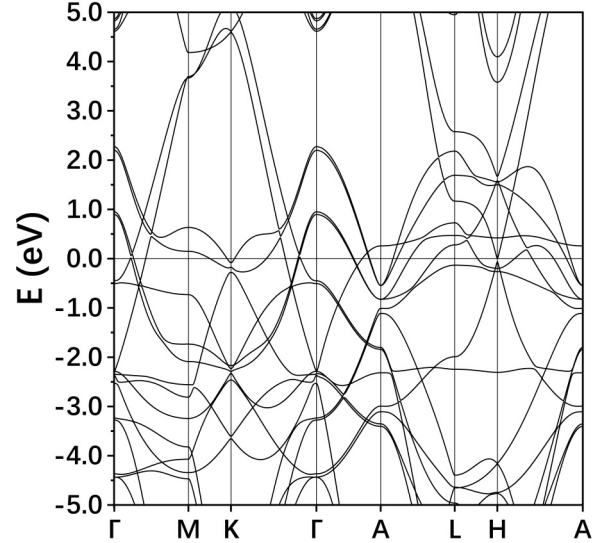


FIG. 6. The electronic structure of FeB_2 based on LSDA + SOC with (001) spin orientation.

Thus, the values of J_1 to J_4 can be evaluated by mapping the relative total energies of the five spin states and the obtained magnetic interactions are summarized in Table III. These four magnetic interactions are all ferromagnetic, in which J_1 and J_4 play a leading role. With the increase of U , the strength of these exchange parameters have somewhat differences, but the dominant terms do not change. Based on the spin exchange parameters in Table III, we calculate Curie temperatures by the mean-field theory (MFT) approximation [64] and MC simulations [47–49], which are also summarized in Table III. The calculated Curie temperature based on LSDA calculation is very high (1546 K) for MFT approximation, while the value from MC simulations is somewhat smaller (945 K). Meanwhile, the calculated Curie temperatures are not greatly affected by the value of U . In summary, FeB_4 is suggested to be a good candidate for potential high-temperature permanent magnet materials.

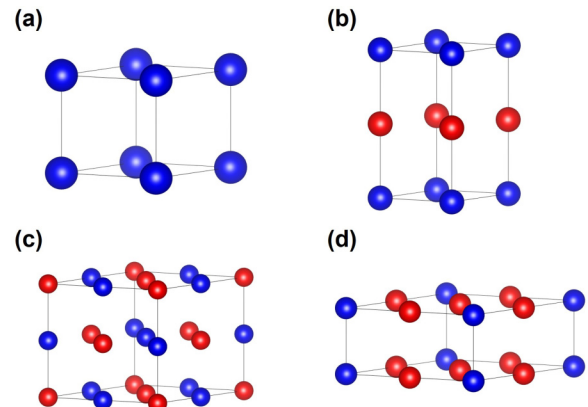


FIG. 7. The magnetic configurations of FeB_2 considered in DFT calculations. For clarity only Fe atoms are shown, while the blue and red ones represent Fe ions with up and down spin, respectively. (a)–(d) FM, AFM-1, AFM-2, and FIM-1 configurations, respectively.

TABLE V. Spin exchange parameters (in meV) of FeB₂ evaluated from LSDA (+ U) calculations. Meanwhile, the calculated Curie temperature (in K) based on MFT and MC method with different values of U are also summarized in the table.

	Distance (Å)	LSDA	LSDA + U (= 0.5 eV)	LSDA + U (= 1.0 eV)	LSDA + U (= 1.5 eV)	LSDA + U (= 2.0 eV)
J_1	3.035	-2.16	-5.01	-11.49	-16.54	-26.52
J_2	3.045	-11.42	-16.17	-19.46	-23.17	-25.08
J_3	4.299	-1.81	-4.59	-6.70	-8.15	-8.21
T_C (MFT)		366	627	852	1044	1168
T_C (MC)		355	588	828	964	1054

B. FeB₂

FeB₂ crystallizes in the hexagonal structure (space group $P6/mmm$) [65] as shown in Fig. 5. Fe atoms occupy the $1a$ position: (0, 0, 0), and the site-symmetry group of Fe is ($6/mmm$), which also permits the E_1 and E_2 orbital multiplets allowing large MAE.

As one of the most important criterias of the permanent magnet, the magnetic moment on the Fe site is estimated to be $2.21 \mu_B$ ($M_S = 0.0937 \mu_B/\text{Å}^3$) from LSDA calculations for FM configuration. Meanwhile, the MAE is calculated to be 5.394 MJ/m^3 with the easy-axis magnetization along the (001) direction by the energy differences approach, while based on the force theorem [42,43] the MAE is 5.688 MJ/m^3 . The band structure from LSDA + SOC calculation with (001) spin direction is shown in Fig. 6. In a similar way to the FeB₄ case, we perform first-principles calculations for several different magnetic configurations of FeB₂, including

FM configuration, two AFM configurations, and one FIM configuration as shown in Fig. 7. As shown in Table IV, the FM configuration has the lowest total energy with different Hubbard U . The magnetic moments are comparable for different magnetic configurations with the same Hubbard U value (with a difference around $\pm 0.3 \mu_B$), suggesting that the energy-mapping method is an effective approach to roughly calculate interactions. As shown in Fig. 5, we consider four magnetic interactions labeled as J_1 to J_3 . The total energies per unit cell for these four spin configurations are expressed as

$$\begin{aligned}
 E_{\text{FM}} &= E_0 + J_1 + 3J_2 + 6J_3, \\
 E_{\text{AFM-1}} &= E_0 - J_1 + 3J_2 - 6J_3, \\
 E_{\text{AFM-2}} &= E_0 - J_1 - J_2 + 2J_3, \\
 E_{\text{FIM}} &= E_0 + J_1.
 \end{aligned} \tag{2}$$

TABLE VI. Materials for permanent magnets by screening which have already been discussed in previous studies. Listed are the material's ICSD, formula, space group, the corresponding $3d$ transition elements and Wyckoff positions, the calculated MAE (MJ/m^3) from LSDA + SOC calculations and saturation magnetic moment M_s ($\mu_B/\text{Å}^3$) from LSDA calculations. The data in the previous theoretical and experimental work are also shown for comparison.

ICSD number	Material formula	Space group	Transition elements	Wyckoff positon	This work		Previous work			
					MAE	M_s	Experimental work		Theoretical work	
							MAE	M_s	MAE	M_s
150 640	FePt	123	Fe	$1a$	11.977	0.119			10.0 [51]	
53 259	Pt ₂ FeCu	123	Cu	$1a$	5.291	0.073			5.83 [32]	0.075 [32]
42 564	Pt ₂ FeNi	123	Fe	$1a$	2.358	0.088			2.42 [32]	0.089 [32]
			Ni	$1c$						
102 620	CoPt	123	Co	$1a$	5.098	0.085	4.0 [66], 6.9 [67]		9.0 [68]	
634 628	MnGa	123	Mn	$1d$	2.256	0.092			2.0 [69]	0.087 [70]
181 719	FePd	123	Fe1	$1a$	1.833	0.118			1.28 [71]	0.114 [71]
			Fe2	$1c$					1.713 [72]	
180 896	FePd	123	Fe	$1a$	0.938	0.119		1.04 [72]	0.96 [72]	
76 636	Fe ₁₆ N ₂	139	Fe1	$4d$	1.264	0.188		0.78 [52]	1.043 [54]	0.192 [54]
			Fe2	$4e$						0.266 [53]
99 787	Fe ₃ Pt	139	Fe	$4e$	1.344	0.167			0.086 [73]	0.172 [73]
102 731	YCo ₅	191	Co	$2c$	2.294	0.086	6.5 [74]		4.37 [56]	0.099 [57], 0.096 [56]
52 972	Co ₃ Mn ₂ Ge	194	Mn	$4f$	1.465	0.147	1.18 [33]	0.087 [33]	1.44 [33]	0.147 [33]
633 778	Fe ₂ Ta	194	Fe	$2a$	2.023	0.062			1.25 [75]	0.059 [75]
643 689	Mn ₃ V ₂ Si ₃	193	V	$4d$	0.798	0.078			0.85 [33]	0.063 [33]
625 559	YCo ₃	194	Co1	$2a$	1.051	0.060	1.03 [76]	0.029 [76]		
			Co2	$2b$						
			Co3	$2c$						

Thus, the values of J_1 to J_3 can be evaluated by mapping the relative total energies of the four spin states and the obtained magnetic interactions are summarized in Table V. Based on the calculated spin exchange parameters from LSDA calculations, the Curie temperatures are estimated to be 366 K and 355 K using the MFT approximation [64] and MC method [47–49], respectively. With the increasing value of U , the Curie temperature by both the MFT approximation and MC method become higher as shown in Table V. According to its high saturation magnetization and large uniaxial MAE, FeB₂ is also suggested to be a good candidate for potential permanent magnet material.

V. CONCLUSION

In conclusion, we present a highly efficient search for permanent magnet candidates in the ICSD. Focusing on 3d materials where certain partially occupied orbital multiplets can significantly enhance the effect of SOC, we first screen 3d transition metal materials with potentially huge MAEs by simply checking whether the materials belong to the corresponding space groups and whether the 3d transition elements occupy the specific Wyckoff positions. Then first-principles high-throughput screening is performed based on the calculated saturation magnetic moment, magnetic anisotropy energy, and magnetic ground state. In addition to many materials discussed in earlier studies, finally five new candidates are proposed as potential permanent magnet materials. As examples, we present an investigation of detailed magnetic properties in FeB₄ and FeB₂, and discuss the effect of Coulomb interaction U on them. Our results show that 3d transition elements compounds may also have large MAE and we believe that these potential rare-earth-free permanent magnet materials deserve further experimental study.

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TABLE VII. The space group, Wyckoff positions of Fe ions, and the calculated MAE (MJ/m³) of Fe-B alloys in ICSD database (* materials for discussion in main text).

Formula	SG	Wyckoff position	MAE
FeB ₂ [77]	62	4c	-0.0138
FeB ₂ * [65]	191	1a	5.7143
FeB ₄ [78]	58	2a	0.0935
FeB ₄ [59]	58	2b	0.0074
FeB ₄ [59]	64	4a	-0.0001
FeB ₄ [59]	142	8b	0.0000
FeB ₄ [59]	194	2d	-1.7967
FeB ₄ * [59]	194	2b, 2c	2.4084
FeB ₄ [59]	216	4a	-0.0018
FeB [79]	62	4c	-0.4758
FeB [77]	141	8e	0.3763
Fe ₂ B [80]	121	8i	-0.2615
Fe ₂ B [81]	140	8h	-0.3131
Fe ₂ B ₇ [82]	55	4h, 4g	-0.1112
Fe ₃ B [83]	62	4c	-0.4710
Fe ₃ B [84]	82	8g	-0.1831
Fe ₃ B [83]	182	6g	0.1505
Fe ₂₃ B ₆ [85]	225	4a, 8c, 48h, 32f	0.0001

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APPENDIX

In the Appendix, we list these suggested permanent magnet materials which were already discussed in earlier studies in Table VI. Meanwhile, we also list the space groups, MAE, and Wyckoff positions occupied by Fe ions of Fe-B alloys existing in the ICSD database in Table VII.

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