Growth and characterization of off-stoichiometric LaVO₃ thin films

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LaVO₃ (LVO) has been proposed as a promising material for photovoltaics because its strongly correlated *d* electrons can facilitate the creation of multiple electron-hole pairs per incoming photon, which would lead to increased device efficiency. In this paper, we intentionally grow off-stoichiometric LVO films by changing the growth conditions such as laser fluence. Our aim is to study how deviating La:V stoichiometries affect the electronic properties of LVO thin films. We find that the off-stoichiometry clearly alters the physical properties of the films. Structural characterization shows that both La-rich and V-rich films have different levels of structural distortion, with La-rich (V-rich) films showing a larger (smaller) out-of-plane lattice parameter compared to what one would expect from epitaxial strain effects alone. Both types of films show deviation from the behavior of bulk LVO in optical measurement, i.e., they do not show signatures of the expected long range orbital order, which can be a result of the structural distortions or the presence of structural domains. In transport measurements, La-rich films display clear signatures of electronic phase separation accompanying a temperature induced metal-insulator transition, while V-rich films behave as Mott insulators. The out-of-plane lattice parameter plays a crucial role in determining the transport properties, as the crossover from Mott-insulating to disorder-induced phase-separated behavior occurs around a lattice parameter value of 3.96 Å, quite different from what has been previously reported.

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I. INTRODUCTION

Strongly correlated electron systems have attracted intense research interests over decades due to their wide range of functionalities. Properties including colossal magnetoresistance [1], high- T_C superconductivity [2], and orbital ordered states [3,4] arise from the complicated nature of d electrons in transition metals [5]. Orthovanadates RVO_3 (R = rare earth or yttrium) are strongly correlated insulators (SCIs) the properties of which can be affected by the choice of the R elements [6,7]. LaVO₃ (LVO), as a narrow band-gap SCI, exhibits a variety of properties that are of both fundamental and technological interest. For example, Coulter et al. [8] theoretically proposed that some SCIs can become novel photovoltaic materials with enhanced solar cell efficiency, because the strong Coulomb interactions in the d electrons in vanadium atoms can allow the material to utilize the excess energy from a "hot" electron to create multiple electron-hole carriers through the process called impact ionization (II) [8-11]. For LVO, the calculated II rate is of the order of 10^{15} s⁻¹, which is orders of magnitude faster than the rate of electron-phonon interaction $(\approx 10^{12} \text{ s}^{-1})$ in this material [8,12]; the band gap of bulk LVO is 1.1 eV [12,13], which makes it theoretically ideal to realize the II process with solar energy. In addition to that, LVO also exhibits a variety of intriguing physical properties such as orbital order [6], structural transitions [14], and metal-insulator transitions, and it shows promise as a basis for a *p*-type transparent conducting oxide [15–17]. Most importantly, numerous thin-film works have clearly shown that these functional properties can be tuned by applying external perturbations, such as applied current and structural distortions induced by epitaxial strain and doping [18–22]. This makes LVO a promising candidate for strain- and doping-driven engineering of materials properties.

Bulk LVO is orthorhombic (Pbnm) or pseudocubic at room temperature, with lattice constants $a_p \approx b_p \approx c_p \approx$ 3.924 Å [7,19]. The level of distortion of a RVO_3 lattice can be evaluated by the rotation and tilting of the VO₆ octahedra or by the angle of V-O-V bonds, and such distortions will affect how the d orbitals order inside vanadium atoms. Generally, for orthorhombic RVO_3 , the d orbitals adopt a so-called C-type orbital ordering (C-type OO) configuration [Fig. 1(a) right], where the d_{vz} and d_{zx} orbitals are aligned out of phase in the *ab* plane, but in phase along the c direction; if a RVO_3 has a more distorted monoclinic structure, an alternative G-type orbital ordering (G-type OO) [Fig. 1(a) left] can form; in this type of ordering, d_{vz} and d_{zx} are aligned out of phase in all directions [23]. Among the RVO_3 family, LVO is the least distorted crystal, it has the largest V-O-V bonding angle [6,7] and it undergoes a structural transition to a monoclinic phase when it is cooled to below 141 K [14]. As a result, the d orbitals will order to the G-type OO state at low temperature. Previous reports showed that optical absorption is effective in detecting the structural phase transition associated with the formation of G-type OO in LVO [14,24]. This is because the energies to

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FIG. 1. (a) Two possible orbital configurations in the RVO_3 family. Left, G-type OO; right, C-type OO; for LVO, these have corresponding peaks in absorption spectra at 1.8 and 2.4 eV, respectively. (b) Optical measurement setup to detect types of orbital ordering. A sample stage that is angled at 45° with regard to the incoming light is located inside of a cryostat. A horizontally polarized white light source is used to illuminate the samples. The transmitted light intensity is measured by a visible-near infrared CCD detector.

excite electron hopping between neighboring vanadium sites are different in the C-type OO and G-type OO phases. In the case of bulk LVO, its C-type OO absorption peak is at 2.4 eV while the G-type OO peak is at 1.8 eV, respectively.

Here we present the study on using pulsed laser deposition (PLD) to grow off-stoichiometric LaVO₃ thin films on SrTiO₃ (STO) (001) substrates. We investigate how the presence of La:V off-stoichiometry affects the physical properties of LVO thin films. In line with studies on other complex oxides (e.g., manganites [1] and cuprates [25]), one expects to find rich physics upon straining, doping, or distorting the structure of orthovanadate thin films as well. According to existing literature, PLD is the most common technique to grow LVO films [13,18–20,26,27]. Previous reports show that growth conditions can control the oxygen stoichiometry of the films, sometimes leading to the formation of unwanted LaVO₄ parasitic phases [13,18,28]. Some have investigated the optical and transport properties of PLD-grown LVO thin films, including ultrathin films [13,29]. However, these works only looked at stoichiometric thin films. Off-stoichiometric LVO thin film studies are all done on molecular beam epitaxy (MBE) grown films, where the La:V stoichiometry is shown to alter the structural properties of the films [12,21]. In this paper, we report on the use of PLD to grow off-stoichiometric LVO thin films by means of controlling laser fluence. Characterizations on the off-stoichiometric LVO films show that the lattice mismatch between the films and the substrate causes the films to be strained. From structural characterizations using x-ray diffraction and transmission electron microscopy (TEM), we learn that the amount of elongation of the out-of-plane lattice constant depends on the La:V stoichiometry. The V-rich (La-rich) films have a smaller (larger) unit cell volume compared to the expected unit cell volume based on the reported Poisson ratio for bulk LVO and the epitaxial strain [30,31]. Furthermore, the film unit cells are not single-oriented, similar to previous reports [30]. From absorption measurements [see Fig. 1(b)] we find the coexistence of C-type and G-type absorption peaks. We do not see significant changes in the spectral weight of G-type OO peak as the films are cooled down. This indicates that the long range G-type orbital ordering that was observed at low temperature in bulk crystals is absent in our thin films. The transport measurements show a clear stoichiometry dependence. The V-rich films show correspondence to bulk LVO, but the La-rich films have unexpected metal-insulator transitions as well as non-Ohmic current-voltage curves pointing to electronic phase separation. These observations clearly indicate that the electronic properties of off-stoichiometric thin films are largely affected by the out-of-plane lattice parameter, i.e., the amount of distortion to the LVO unit cell.

II. EXPERIMENTAL

LVO films are grown by the PLD technique, in which a KrF excimer laser ($\lambda = 248$ nm) is focused on a ceramic target $LaV_{1,2}O_4$ (half of the La-rich films were grown using a $LaV_{1.1}O_4$ target). The $LaV_{1.2}O_4$ targets are prepared from La₂O₃ (99.9% purity) and V₂O₅ (99.9% purity) powders with the ratio of 1:1.2. Excess V₂O₅ powder is used to compensate for losses of the highly volatile vanadium element during the target sintering process (see Supplemental Material [32] and Ref. [33] for more details on the target fabrication recipe). The STO substrates are pretreated using deionized water in a leaching and thermal annealing method to obtain atomically flat and TiO₂ terminated surfaces [34]. The laser fluence ranges from ≈ 0.6 to ≈ 2.0 J/cm², while the repetition rate of the laser is kept at 1 Hz. All films are deposited at the order of 10^{-7} -mTorr base pressure to stabilize the perovskite phase of LaVO₃. The temperature of the substrate is kept in the range of 600–650 °C during growth [18].

The morphology of the LVO films is characterized by atomic force microscopy (AFM) using an Asylum Research microscope. The structure is studied using a Scintag DMC-105 x-ray diffractometer, a Rigaku SmartLab SE x-ray diffractometer (both with Cu K α_1 , $\lambda = 0.154$ nm), and the Advanced Photon Source at Argonne National Laboratory (photon energy = 12.398 keV, $\lambda = 1$ Å) at beamline 33-BM-C [35]. Detailed information on the microstructure and the chemical composition are investigated using a probeaberration-corrected cold field emission JEM-ARM200cF at 200 kV, which is equipped with an Oxford energy dispersive spectroscopy (EDS) detector for composition analysis. Electrical resistivity $\rho(T)$ measurements are carried out from 300 to 2 K using a Quantum Design physical property measurement system (PPMS) and Keithley external electronics model 2182 and model 6221. The samples for PPMS measurements are cut to strips (length, 5 mm; width, 0.5-1.2 mm) and we use a four-point probe measurement geometry via evaporated gold contacts. The optical properties are characterized using absorption measurements [see Fig. 1(b) for a schematic of the setup]. For the absorption measurements we used an Ocean Optics USB2000 Spectrometer with a quartz-tungstenhalogen lamp as the source. The CCD detector is a Si-based photodiode array that is uncooled (integrated into the USB spectrometer by Ocean Optics). The resolution and range of the spectrometer are 3 meV and 1.5-3.1 eV, which covers the



FIG. 2. AFM and TEM images for LVO films with different stoichiometry. Top row: V-rich film LVO-V1. (a) AFM image showing a surface with unit cell terraces (RMS roughness: 2.8 Å). (b) Low magnification bright field TEM image of the cross-section view. The streaky lines inside the film are contrast from the antiphase domains. (c) Atomic resolution HAADF-STEM image. Insets: Fast Fourier transforms (FFTs) from the [001] domain (left) and [110] domain (right). Bottom row: La-rich film LVO-L1. (d) AFM image showing a surface with unit cell terraces (RMS roughness: 3.5 Å). (e) Low magnification bright field TEM image of the cross-section view. (f) Atomic resolution HAADF-STEM image. Inset: FFTs from domain 1 (left) and domain 2 (right).

range of our interest, the LVO *d-d* transitions. The incoming light is polarized to the horizontal direction and the sample stage is angled at 45° with regard to it. An LVO film and a bare STO substrate are fixed to the sample stage in order to take their spectra at each temperature. This setup is intentionally designed this way so that the light can interact with the film both along the *c* axis and *ab* plane. The backsides of all samples measured in this optical setup are hand-polished until optically smooth using a 3- μ m diamond lapping film. The relative absorbance A_{λ} is calculated with

$$A_{\lambda} = -\log_{10} \frac{\text{LVO}_{\lambda} - \text{Dark}_{\lambda}}{\text{STO}_{\lambda} - \text{Dark}_{\lambda}}$$
(1)

where LVO_{λ} and STO_{λ} are the transmitted light intensity of LVO film and STO substrate, respectively. Dark_{λ} is the dark current noise from the detector.

III. RESULTS AND DISCUSSION

A. Morphology and structure

First, we discuss the influence of the stoichiometry on the morphology and microstructure of the thin films. The results from AFM measurements show that the surface morphology of the LVO films is not very sensitive to stoichiometry. AFM images in Figs. 2(a) and 2(d) show that both types of films have unit cell terraces with comparable RMS roughness values, 2.8 Å for LVO-V1 and 3.5 Å for LVO-L1. The microstructure of the film does depend on stoichiometry, as evidenced by the TEM images in Figs. 2(b), 2(c), 2(e) and 2(f). The La:V stoichiometry of the films was determined

using the EDS and/or the growth conditions used and is 45:55 for the V-rich and 55:45 for the La-rich films. The cross-section view of the V-rich films [Figs. 2(b) and 2(c)] revealed that there are high density antiphase nanodomains with different *c*-axis orientation relative to the substrate. The orientations of these domains are determined to be [001] (*c* axis pointing to the viewer) and [110] (*c* axis pointing to the top surface) by doing Fourier transformation of the images [20,36]. However, the La-rich film shows different microstructures, and it has lower density of antiphase domains, but larger grains with tilted *c* axis forming low angle grain boundaries instead [Figs. 2(e) and 2(f)]. The more random tilts of the *c* axis in the La-rich films is in line with the reduced structural coherence observed in these films in the RSM measurements, which we will discuss next.

Representative reciprocal space maps (RSMs) on a V-rich film (LVO-V2) and a La-rich film (LVO-L1) reveal that our films are strained onto the STO substrates (Fig. 3). The lattice constants extracted from these RSMs are $a_p = 3.908(1)$ Å, $b_p = 3.905(1)$ Å, and $c_p = 3.955(2)$ Å for LVO-V2, and $a_p = 3.907(1)$ Å, $b_p = 3.911(1)$ Å, and $c_p = 3.969(4)$ Å for LVO-L1. (See Supplemental Material [32] for how the lattice constants are extracted and calculated.) The contraction in the *a* and *b* axes and the elongation in the *c* axis (out-of-plane direction) in these films are expected because there is a compressive in-planar strain of about 0.5% induced by the lattice mismatch between STO (3.905 Å) [37] and LVO. Furthermore, LVO-L1 has weaker film peak intensity than LVO-V2. Since the two films are similar in thickness (around 200 nm), such a difference indicates that



FIG. 3. Reciprocal space mapping for LVO-V2 and LVO-L1. (a), (b) RSM for LVO-V2 around its (103) and (013) peaks, respectively. (c), (d) RSM for LVO-L1 around its (103) and (013) peaks, respectively. The STO and LVO peaks are labeled in each panel. The white cross-hairs indicate the peak position for stoichiometric bulk LVO. The red "x" marks the position of maximum intensity for each film peak.

the structure in the La-rich film is less coherent, which is consistent with the structural domains observed in TEM (see Fig. 2). Apart from the change in the dimensions of the LVO unit cell due to the epitaxial strain effect, we notice that our LVO films with different La:V stoichiometry have differences in their out-of-plane lattice constants. Figure 4 summarizes the out-of-plane lattice constant c extracted from $\theta - 2\theta$ scans on our films (see Supplemental Material [32] for details). The c parameter for the La-rich films is approximately 1% greater than for the V-rich films. Such a difference is subtle but consistent, and in agreement with previous studies on the nonstoichiometric bulk LVO samples [38,39].

To further characterize the lattice structure in our films, we performed synchrotron x-ray diffraction measurements on



FIG. 4. Summary of the out-of-plane lattice constants for the films discussed in the paper. Each film was grown using nominally the same growth conditions, except for the laser fluence, which was set to $J \sim 0.6$ and 1 J/cm² (or higher), for La-rich and V-rich films, respectively. The La:V stoichiometry was determined using EDS (La:V = 55:45 and 45:55) or inferred from the growth conditions. The film thickness was around 200 nm unless indicated otherwise, i.e., L3: 350 nm, as determined from TEM and from the number of pulses used during growth. The dashed line is the expected lattice parameter based on a Poisson ratio of v = 0.398 [30,31] and epitaxial strain of $\varepsilon_{\parallel} = -0.5\%$.

LVO-L1 and LVO-V2. From the scans, we obtain half order peaks (see Fig. 5) such as even-odd-odd $([0\frac{1}{2}\frac{3}{2}])$, odd-even-odd $([\frac{1}{2}0\frac{3}{2}])$, and odd-odd-odd $([\frac{1}{2}\frac{1}{2}\frac{3}{2}], [-\frac{1}{2}\frac{1}{2}\frac{1}{2}])$. Based on these observed peaks, we conclude an $a^+a^-c^-$ octahedral tilting pattern in both films. This is in agreement with the pattern reported by Rotella *et al.* [19] $(a^-a^+c^-)$ in their stoichiometric LVO thin films.

B. Absorption

Temperature dependent absorbance is done on various films and representative spectra (for LVO-L2 and LVO-V2) at selected temperatures are shown in Fig. 6. To analyze the absorption peaks, spectra are fitted to the Gaussian function with the fitting background set to be a constant value (the minimum of each spectrum) (Fig. 6 insets). The fit results can be found in Table I. From Fig. 6 and Table I, we can see both films have a temperature dependence when being cooled down. The G-type OO absorption peak slowly gains spectral weight, but not as drastic as what others reported in bulk LVO in the low temperature regime [14,24]. The energies for G-type and C-type OO peaks in both types of films are similar to bulk values. Though there seems to be some



FIG. 5. Half order peaks of LVO-L1. With the diffraction peaks shown, we are able to determine an $a^+a^-c^-$ tilting pattern in our films.

Temperature (K)	LVO-L2				LVO-V2			
	G-type OO		C-type OO		G-type OO		C-type OO	
	Energy (eV)	% weight						
250	1.78	3.73	2.40	96.27	1.77	3.04	2.39	96.96
200	1.79	6.21	2.40	93.79	1.78	4.86	2.39	95.14
150	1.80	10.01	2.39	89.99	1.80	6.54	2.35	93.46
100	1.81	11.05	2.35	88.95	1.82	5.63	2.39	94.37
25	1.83	15.80	2.33	84.20	1.82	4.88	2.39	95.12

TABLE I. Peak analysis of absorption spectra at different temperatures.

temperature dependent energy shifts, these are more likely due to fitting error because other absorption peaks outside our measurement range (i.e., transitions between the O 2p and V 3d bands) can contribute to the spectrum as a temperature dependent "background" [13]. From these measurements it is clear that our LVO-L2 and LVO-V2 do not display evidence of OO at low temperatures, but the two peaks of d-d transitions are distinguishable (see Table I) at all temperatures. Here we note that Zhang et al. [12] reported room temperature absorption spectra on stoichiometric and off-stoichiometric LVO thin films grown using MBE. They found that both the stoichiometric and the La-rich films kept a pronounced two-peak feature (peaks at 1.7 and 2.2 eV, respectively), but for the V-rich films the two peaks wash out with increasing V concentration. Hence, our films may be closer to optimal stoichiometry in comparison and the optical properties appear



FIG. 6. Temperature-dependent absorption spectra on (a) LVO-L2 (integration time: 250 ms) and (b) LVO-V2 (integration time: 400 ms) at 250, 150, and 25 K. Insets are Gaussian fittings of the 25-K spectra.

to be quite insensitive to changes in stoichiometry of the order of $\approx 5\%$.

Furthermore, the apparent absence of orbital order in our films at low temperature has multiple possible causes. The presence of structural domains as seen in our TEM images will hinder formation of long range orbital order, even if the individual domains are ordered. It is also possible that the extension of the out-of-plane direction plays a role in suppressing the formation of orbital order. The excess or deficiency of the vanadium alters the V-O-V bond lengths and angles [38,39], which could lead to a more distorted bonding scheme possibly disrupting orbital ordering.

C. Resistivity

Bulk LVO is a Mott insulator [40,41]. All V-rich films described in this paper behave as Mott insulators. Figure 7(a) plots the temperature dependent resistance of a representative V-rich film (LVO-V2): its high temperature resistance can be fitted with the thermal activation model with an activation energy of 73.3 meV [Fig. 7(a) inset]; in the low *T* region, the resistance can be explained using the Mott variable range hopping model of a three dimensional material [42,43]. All



FIG. 7. Resistance measurement on a representative V-rich film LVO-V2 measured with 0.001 mA, showing that its low temperature resistance can be well fitted to the Mott variable range hopping model of a three dimensional material ($\approx T^{-1/4}$); the resistance at high temperature can be explained using the thermal activation model (see inset).



FIG. 8. Temperature dependent resistance measurement for a representative La-rich film LVO-L3 measured with 0.001 mA. There are two turnovers in the full temperature range scan. Part of its metallic region can be fitted to Fermi-liquid model $\rho \sim T^2$ (see inset).

V-rich films indicated in Fig. 4 with the blue symbols show this behavior.

The situation is more complicated in La-rich films. Existing studies on bulk LVO show that samples become more insulating as they become more La rich (or V deficient) [39], but this is not what we observe in our films. In Fig. 8, the temperature dependent resistivity of LVO-L3 shows two turnovers; the first one around 250 K is indicating that the film changes from an insulatinglike state to a metalliclike state; the second one at low T shows a drastic increase in resistance. Part of the region between two turnovers can be fitted to $\rho \sim T^2$, which indicates a Fermi-liquid-like behavior [44,45]. All La-rich films, indicated by red symbols in Fig. 4, show this behavior (see Supplemental Material [32] for more R vs T data).

There have been reports on stoichiometric LVO thin films showing a similar temperature dependence of the resistivity. For example, He et al. reported the resistance of LVO changing from a high temperature conducting phase to a low Tinsulating phase in a 30-nm-thick film, attributing the observation to the LVO-STO interface effect [29,46]. But the LVO-STO interface likely does not affect our resistivity since our films are almost an order of magnitude thicker. Rather, it appears that the lattice parameter extension, and associated changes to the V-O-V bonding scheme, induced by V deficiency is the crucial parameter controlling the transport properties. Additionally, the unexpected turnovers in our Larich films could be due to the fact that V vacancies cause the formation of defects and disorders inside the film, and lead to impurity-induced electron localization. Thus, a phononrelated Peierls transition [47] or disorder-related Anderson localization [48] can play a role in the observed turnovers in resistivity. Moreover, off-stoichiometry is often accompanied by the change in valence state. Since the film is La rich, it is likely that the valence state of vanadium is not purely 3+. A mixed valence state can create a dopinglike effect. Therefore the film behaves like a metal at high temperature because



FIG. 9. The upturn in resistance at low T can be suppressed by increasing driving current. Inset: A scheme showing the insulating phase (colored in red) is not formed homogeneously, allowing current to percolate when higher currents are applied (colored in green).

of excess charge carriers provided by the dopant, but at low temperature all electrons are bonded to chemical impurities and structural disorders so it becomes insulating again, similar to behaviors reported for Sr-doped LVO films [49,50].

We also find that we can suppress the upturn in resistance at low T by increasing applied current (Fig. 9), i.e., the I-V curves are nonlinear at low temperature (see Supplemental Material [32]). We can rule out heating as a source of nonlinearity in the low temperature I-V curves by performing a Joule heating analysis similar to the one reported in Padhan et al. [51] (see Supplemental Material [32] for details). Razavi et al. reported similar upturns in resistivity and nonlinear I-V characteristics at low temperature; they attributed it to the formation of random diode networks in their LVO thin films [52]. They reported that the observed nonlinear behavior happened only in the films with relatively small values for c (close to that of the STO substrate) and it would disappear when the films become insulating as the lattice parameter of the film became larger (1% larger than that of STO). In their study the effect of possible off-stoichiometry was not discussed. All our Larich films that have an out-of-plane lattice parameter of about 3.96 Å or larger show similar diodelike behavior. It is likely that the randomly distributed impurities and defects in the La-rich films help to make it conductorlike in the temperature range of 100 to 250 K, but when the temperature is low enough they cause the localization of electrons resulting in scattered insulating regions. Once the randomly formed insulating regions become large enough to reach across the measured strip, the transport behavior will be dominated by the insulating state. However, given enough applied bias across this region, much like overcoming the contact potential in the forward bias mode of a metal-semiconductor junction, the current will start percolating through the strip and lower the measured resistance.

IV. CONCLUSION

We have successfully grown off-stoichiometric LVO thin films on STO substrates using PLD technique by adjusting the laser fluence during growth. Clearly the off-stoichiometry in LaVO₃ films affects its structural, optical, and transport performance. The absorption spectra show little dependence on stoichiometric variation, however both types of films show absence of long range orbital order, which is different from bulk. This is either due to the presence of structural domains in the films or it may be due to the off-stoichiometry distorting the V-O-V bonding scheme of the LVO films. While the V-rich films behave as Mott insulators, the resistivity measurements in La-rich films show unexpected metal-insulator transitions and clear signatures of electronic phase separation at low temperature. It is clear that structural distortions due to the extension of the out-of-plane lattice parameter as well as the presence of doping and defects play a role in the changed transport properties of the La-rich films. The low temperature metal-insulator transition and the associated current-driven melting of electronic phase separation in La-rich films can make it useful in some temperature- or current-driven novel devices. But when it comes to employing the material as an absorber in solar cells, one should adjust growth conditions so that the product tends to be V rich rather than La rich. However, the possible effects of charge trapping due to de-

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fects associated with the observed antiphase boundaries could hamper the use of these V-rich films in photovoltaic devices, which needs further investigation.

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