

**Exploring diamondlike lattice thermal conductivity crystals via feature-based transfer learning**

Shenghong Ju

*Department of Mechanical Engineering, The University of Tokyo, 7-3-1 Hongo, Bunkyo, Tokyo 113-8656, Japan  
and National Institute for Materials Science (NIMS), 1-2-1 Sengen, Tsukuba, Ibaraki 305-0047, Japan*

Ryo Yoshida

*Research Organization of Information and Systems, The Institute of Statistical Mathematics (ISM),  
10-3 Midori-cho, Tachikawa, Tokyo 190-8562, Japan  
and National Institute for Materials Science (NIMS), 1-2-1 Sengen, Tsukuba, Ibaraki 305-0047, Japan*

Chang Liu and Stephen Wu

*Research Organization of Information and Systems, The Institute of Statistical Mathematics (ISM),  
10-3 Midori-cho, Tachikawa, Tokyo 190-8562, Japan*

Kenta Hongo

*Japan Advanced Institute of Science and Technology (JAIST), 1-1 Asahidai, Nomi, Ishikawa 923-1292, Japan;  
National Institute for Materials Science (NIMS), 1-2-1 Sengen, Tsukuba, Ibaraki 305-0047, Japan;  
and PRESTO, Japan Science and Technology Agency (JST), Kawaguchi, Saitama 332-0012, Japan*

Terumasa Tadano

*National Institute for Materials Science (NIMS), 1-2-1 Sengen, Tsukuba, Ibaraki 305-0047, Japan*Junichiro Shiomi *Department of Mechanical Engineering, The University of Tokyo, 7-3-1 Hongo, Bunkyo, Tokyo 113-8656, Japan;  
National Institute for Materials Science (NIMS), 1-2-1 Sengen, Tsukuba, Ibaraki 305-0047, Japan;  
Core Research for Evolutional Science and Technology (CREST), Japan Science and Technology Agency (JST),  
4-1-8, Kawaguchi, Saitama 332-0012, Japan;  
and RIKEN Center for Advanced Intelligence Project, 1-4-1 Nihonbashi, Chuo-ku, Tokyo 103-0027, Japan*

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Ultrahigh lattice thermal conductivity materials hold great importance since they play a critical role in the thermal management of electronic and optical devices. Models using machine learning can search for materials with outstanding higher-order properties like thermal conductivity. However, the lack of sufficient data to train a model is a serious hurdle. Herein we show that big data can complement small data for accurate predictions when lower-order feature properties available in big data are selected properly and applied to transfer learning. The connection between the crystal information and thermal conductivity is directly built with a neural network by transferring descriptors acquired through a pretrained model for the feature property. Successful transfer learning shows the ability of extrapolative prediction and reveals descriptors for lattice anharmonicity. The resulting model is employed to screen over 60 000 compounds to identify novel crystals that can serve as alternatives to diamond. Even though most materials in the top list are superhard materials, we reveal that superhard property does not necessarily lead to high lattice thermal conductivity. Large hardness means high elastic constants and group velocity of phonons in the linear dispersion regime, but the lattice thermal conductivity is determined also by other important factors such as the phonon relaxation time. What is more, the average or maximum dipole polarizability and the van der Waals radius are revealed to be the leading descriptors among those that can also be qualitatively related to anharmonicity.

DOI: [10.1103/PhysRevMaterials.5.053801](https://doi.org/10.1103/PhysRevMaterials.5.053801)**I. INTRODUCTION**

The power densities of microelectronic devices and their components continually increase due to advances in the fab-

rication and integration of advanced materials and structures. Hence, the large thermal density must be quickly removed to guarantee reliable performance. Material innovations in heat spreaders and sinks and thermal interface materials are at the core of the thermal-management challenge. A key element to such innovations is materials with a high lattice thermal conductivity ( $\kappa_L$ ) either as bulk crystals or fillers for composites.

\*To whom correspondence should be addressed: [shiomi@photon.t.u-tokyo.ac.jp](mailto:shiomi@photon.t.u-tokyo.ac.jp)

Although metals are generally suitable thermal conductors, insulators have the highest thermal conductivities. In many thermal management applications involving heat spreaders and sinks, electrical insulation is necessary to avoid electric current leakage. Diamond, which has a thermal conductivity of about  $2000 \text{ Wm}^{-1} \text{ K}^{-1}$  at room temperature, is a representative bulk material [1]. It is widely used as heat spreaders or sinks for laser diodes and power electronics in the form of bulk or composites to prevent overheating. One drawback is that it sustains thermal damage via oxidation or graphitization at high temperature [2,3], significantly altering the thermal properties of the heat spreader or sink.

Cubic and hexagonal boron nitrides have been investigated as alternative materials [4]. Considering the surface affinities with various other materials for composite syntheses and integration, other alternatives should be useful. Although physics insights suggest that some materials exhibit a fairly high  $\kappa_L$  such as SiC, BeO, BP, AlN, BeS, GaN, Si, AlP, and GaP, materials with  $\kappa_L$  approaching or exceeding  $1000 \text{ Wm}^{-1} \text{ K}^{-1}$  are rare.

Single-crystal compounds are obvious candidates as alternative high- $\kappa_L$  materials to diamond [5]. However, only a few materials have quantified thermal conductivity values due to the difficulty of synthesizing single crystals that can be measured in a standardized fashion. Moreover, a material search is extremely cumbersome. Herein we propose utilizing computational techniques to efficiently search for high- $\kappa_L$  materials. In recent decades, the development of lattice dynamics methods using interatomic force constants obtained from density functional theories has enabled first-principles calculations of the  $\kappa_L$ . Simultaneously, databases containing tens of thousands of crystal compounds have been constructed. Examples include MATERIALS PROJECT [6], AFLOW [7], ICSD [8], and ATOMWORK [9]. However, performing first-principles calculations for all the crystals in the databases is extremely time consuming and unrealistic.

Another option is high throughput screening [10] based on machine learning. High throughput screening can speed up discovery of new materials. It has been applied in many fields such as catalysis, battery technologies, thermoelectric materials, chemical probes, polymers, and magnetic materials. Motivated by realizing high-performance thermoelectric materials, efforts to apply  $\kappa_L$  to crystals have centered on screening ultralow  $\kappa_L$  crystals [11–13]. Carrete *et al.* [13] screened 79 000 half-Heusler compounds and found that materials with large atomic radii elements have a lower  $\kappa_L$ . Seko *et al.* [12] screened 54 779 crystals based on the Gaussian process regression and reported 221 materials with a low  $\kappa_L$ . Roekeghem *et al.* [11] extended the screening of mechanically stable compounds at high temperatures using finite-temperature phonon calculations.

One challenge when screening crystals with high and low thermal conductivities is the large gap between the “big data” required for credible machine learning and the “small data” currently available. Although bridging this gap is a general problem in materials informatics [14–16], it is especially intense when searching for materials with a preferred thermal conductivity because it involves both the harmonic phonon properties, which are fairly easy to calculate, and anharmonic

properties, which are much more expensive to evaluate. As shown in Fig. 1, “big data” are available for the harmonic phonon property of the three-phonon scattering phase space but not for the thermal conductivity (Fig. 1). Only “small data” are available for the thermal conductivity due to the heavy calculation required for the anharmonic phonon property. To overcome the difficulty of limited data availability, we employ a transfer learning strategy which is an increasingly popular technique of machine learning [17]. Because openly accessible big data are less available in material research, the ability of transfer learning to learn on small data has attracted much attention [18–20]. Yamada *et al.* [21] demonstrated the potential power of transfer learning in various applications in materials science, including organic and inorganic chemistry. In this study, we show that a prediction model with fairly high accuracy can be derived from the limited data set by effectively reusing features learned on the harmonic phonon property as features in the model of the anharmonic phonon property.

In this work, we develop a feature-based transfer learning method to overcome the gap. This method begins with a broad search over the entire structure database of crystal compounds for the feature harmonic property, which should be correlated with thermal conductivity. Subsequently, a focused search of the selected candidates is conducted for a high thermal conductivity. Here, we choose the scattering phase space ( $P_3$ ) of the three-phonon scattering process as the feature property because this can be quickly extracted from the harmonic calculation.

## II. METHODS

### A. Anharmonic lattice dynamics

The conventional method of calculating  $\kappa_L$  is formulated on the basis of anharmonic lattice dynamics calculations using the interatomic force constants obtained by first principles, which are well described elsewhere [22]. The  $\kappa_L$  was calculated by solving the Boltzmann transport equation (see Supplemental Material, Note A [23])

$$-\mathbf{v}_{qs} \nabla T \left( \frac{\partial n_{qs}}{\partial T} \right) + \left. \frac{\partial n_{qs}}{\partial t} \right|_{\text{scattering}} = 0, \quad (1)$$

where  $n$  is the phonon distribution function,  $qs$  is the phonon mode, and  $\mathbf{v}$  is the group velocity. Although the single-mode relaxation time approximation (see Supplemental Material, Note B [23]) is often used to calculate  $\kappa_L$ , it significantly underestimates  $\kappa_L$  for high-thermal-conductivity crystals. Thus, Eq. (1) must be solved iteratively or directly (see Supplemental Material, Fig. S2 and Note C [23]).

As discussed above, a key feature in the current process is  $P_3$ , which quantifies the phonon scattering channels. The total  $P_3$  is calculated as

$$P_3 = \frac{1}{N_q} \sum_{qs} \frac{1}{3m^3} (2P_3^{(+)}(qs) + P_3^{(-)}(qs)), \quad (2)$$

where  $m$  is the number of phonon branches and

$$P_3^{(\pm)}(qs) = \frac{1}{N_q} \sum_{q's', q''s''} \delta(\omega_{qs} \pm \omega_{q's'} - \omega_{q''s''}) \delta_{\mathbf{q} \pm \mathbf{q}', \mathbf{q}'' + \mathbf{G}}. \quad (3)$$

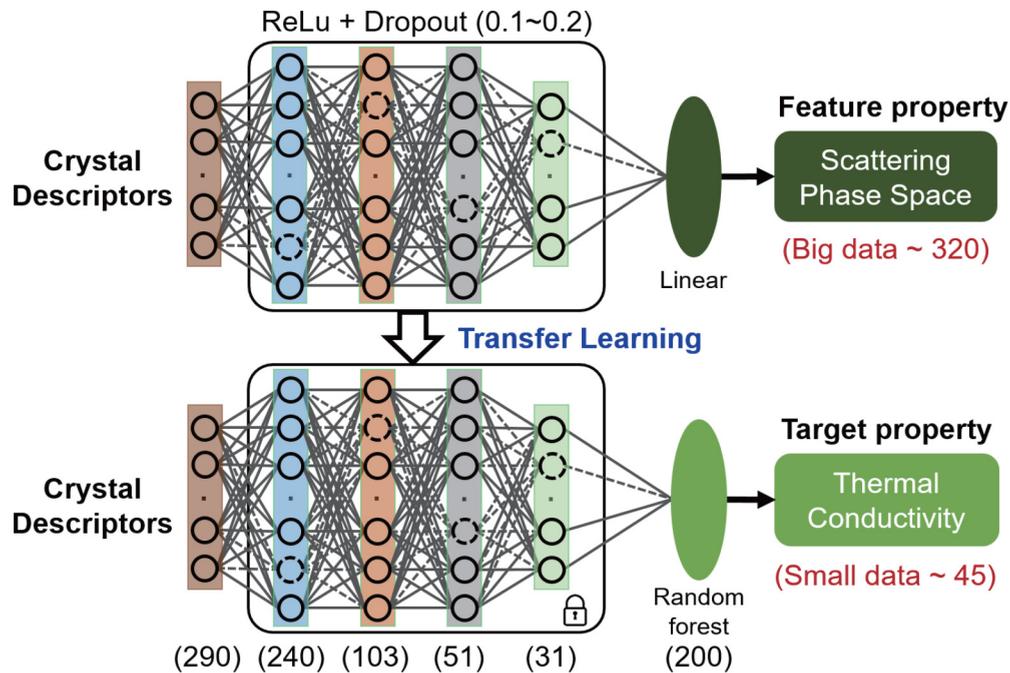


FIG. 1. Schematics of feature-based transfer learning. The transfer learning bridges “big data” (harmonic three-phonon scattering phase space of 320 crystals) and “small data” (thermal conductivity of 45 crystals) to search for ultrahigh lattice thermal conductivity crystals. All neurons (circles) are activated by ReLU (rectified linear unit). Dropout (dashed circles and lines) range (0.1 or 0.2) in each hidden layer is randomly chosen. Numbers at the bottom indicate the number of neurons or trees used in each layer of the neural network and random forest model.

Equation (3) indicates that  $P_3$  can be calculated solely from harmonic interatomic force constants.

### B. $P_3$ and $\kappa_L$ data collections

The candidates were approximately 60 000 inorganic crystals in the MATERIALS PROJECT [6] database. Because this study focused on  $\kappa_L$ , materials with a band gap smaller than 0.1 eV, molecular crystals such as  $O_2$ ,  $H_2$ ,  $H_2O$ ,  $H_2O_2$ , and crystals with hydrogen atoms were excluded. We collected the atom displacement from the phonon database [24] and calculated force data of 320 crystals by first principles (see Supplemental Material, Table S1 [23]). The ALAMODE package [25] was then used to fit the harmonic interatomic force constants and calculate the  $P_3$  values. Aside from the  $P_3$  data, we also collected thermal conductivity data for 45 materials (see Supplemental Material, Table S2 [23]).

### C. Transfer learning

For the given target property ( $\kappa_L$ ), which has limited training data, models on the proxy feature property ( $P_3$ ) are pretrained using the sufficient data to capture the features relevant to the commonality between  $\kappa_L$  and  $P_3$ . Repurposing the pretrained feature extractor on the target task can realize an outstanding prediction ability even with the exceedingly small amount of data. This study focuses on a specific type of transfer learning using artificial neural networks (Fig. 1).

Transfer learning was performed via XENONPY, a self-developed open-source Python package [26]. We used XENONPY to calculate 290 compositional features for a given chemical composition using information about the 58 per-

element features, such as the atomic weight, electronegativity, van der Waals radius, and so on. We pretrained a fully connected pyramid neural network using the 320 instances of  $P_3$  and the 290-dimensional descriptor vectors. All neurons were activated by ReLU (rectified linear unit), and a linear model was placed on the output layer, which defines the transformation from the 10 neurons in the last hidden layer to the  $P_3$ . We produced 1000 pretrained models on the  $P_3$  with randomly generated network structure; the number of hidden layers, which ranged between 4 and 6; the number of neurons in each layer; and the dropout probability, which was either 0.1 or 0.2. Subsequently, the best model on the  $P_3$  was selected based on the 10-fold cross validation looped within the 320 instances. Except for the output layer, the subnetwork of the selected model was used as both a feature extractor and an input descriptor in the prediction model of the  $\kappa_L$ . Finally, the random forest (the number of trees = 200) model was selected and trained using the 45 instances (see Supplemental Material, Table S2 [23]) of the  $\kappa_L$  and the 10-dimensional descriptors acquired through the pretraining process.

## III. RESULTS AND DISCUSSION

### A. Performance of transfer learning

The prediction model connecting the input materials and  $P_3$  was initially trained based on the 320 collected instances of  $P_3$  and 290 descriptors. The subnetwork of the pretrained model was transferred to train the model connecting the crystal structures and  $\kappa_L$  by replacing the linear output layer with the random forest model using the 45  $\kappa_L$  data. The training

and validation results of the pretrained and transferred models are shown in Figs. 2(a) and 2(b).

Using the models for the  $\kappa_L$  and  $P_3$  in the high-throughput screening with  $\sim 60\,000$  crystals in MATERIALS PROJECT[6], we identified the top-14 crystals with the smallest  $P_3$  (see Supplemental Material, Table S3 [23]) from the top-100 prediction list whose  $\kappa_L$  were validated using the first-principles based anharmonic lattice dynamics calculations (Fig. 3 and Table I). The calculation details can be seen in the Supplemental Material, Table S7 and Note D [23]. The transferred model successfully predicts the 14 crystals even though their  $\kappa_L$  lie in the ultrahigh region of  $1000\text{--}3000\text{ Wm}^{-1}\text{K}^{-1}$  [Fig. 2(c)]. It should be noted that the  $\kappa_L$  of the 45 training crystals reside in the region smaller than  $370\text{ Wm}^{-1}\text{K}^{-1}$ , which is much lower than the prediction [Fig. 2(d)]. This indicates that the transferred model exhibits “extrapolative predictive power.” In general, ordinary machine learning is “interpolative,” and its

prediction ability is applicable only in a neighboring region of the given training instances.

Indeed, a neural network directly trained using the 45 samples performs rather poorly for the 14 crystals as the predicted  $\kappa_L$  never exceeded  $600\text{ Wm}^{-1}\text{K}^{-1}$  [Fig. 2(c) and see Supplemental Material, Table S4 [23]]. The pretraining process using the 320 instances on  $P_3$  would contribute to the acquisition of the extrapolative ability. The pretrained neural network on  $P_3$  is able to represent material structures that are applicable to a broader input space than the one spanned by much fewer instances of  $\kappa_L$ . This is because the 320 source  $P_3$  data contain instances that account for the structure-property relationships relevant to the ultrahigh thermal conductivity.

**B. Top high-thermal-conductivity crystals**

We identified the top-14 materials by feature-based transfer learning. They are comprised of boron arsenides

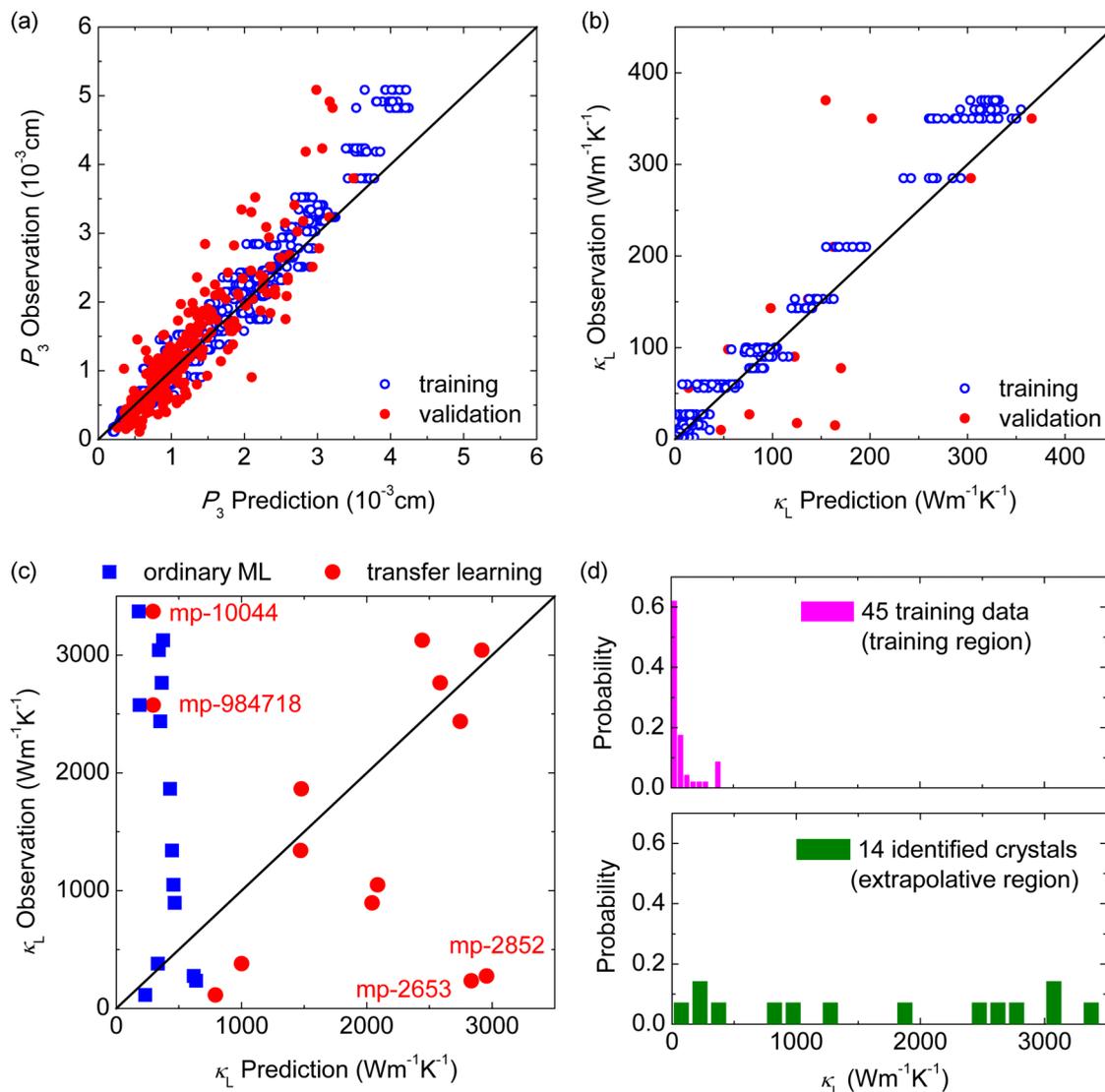


FIG. 2. Performance of transfer learning. Training and validation for (a) the pretrained  $P_3$  model (mean absolute error =  $0.000\,237\text{ cm}$ ) and (b) transferred  $\kappa_L$  model (mean absolute error =  $30.8528\text{ Wm}^{-1}\text{K}^{-1}$ ). Hollow and solid dots denote the results of training and testing in the cross validation for the best prediction model. (c) Comparison between ordinary machine learning and transfer learning. (d)  $\kappa_L$  distribution of 45 training and 14 identified crystals.

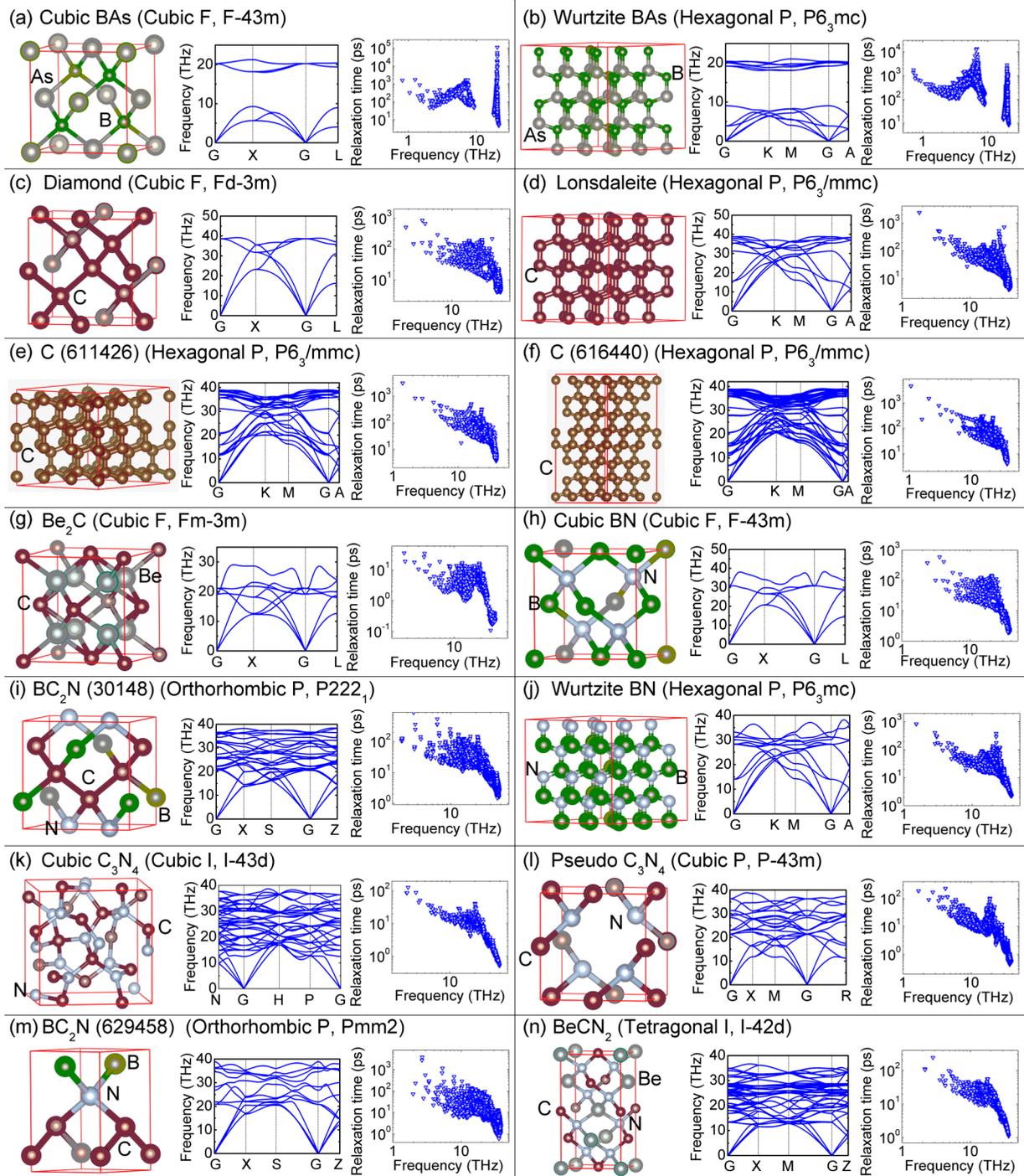


FIG. 3. Top-14 crystals with the lowest calculated  $P_3$ . Crystal structures, phonon dispersions, and phonon relaxation times of the top-14 materials with the lowest calculated  $P_3$ . Parentheses indicate the crystal structure.

(BAs), carbon (C), beryllium carbide (Be<sub>2</sub>C), boron nitride (BN), heterodiamond (BC<sub>2</sub>N), carbon nitride (C<sub>3</sub>N<sub>4</sub>), and ternary BeCN<sub>2</sub>. They all have thermal conductivities above 100 Wm<sup>-1</sup> K<sup>-1</sup>. In fact, 10 exceed 500 Wm<sup>-1</sup> K<sup>-1</sup>. The top two crystals with the lowest calculated  $P_3$  are cubic and wurtzite BAs, which have thermal conductivity over 1000 Wm<sup>-1</sup> K<sup>-1</sup>.

Recently, many studies have investigated cubic BAs due to their high predicted thermal conductivity (3170 Wm<sup>-1</sup> K<sup>-1</sup>)

[27], which is comparable with diamond. However, these initial experiments measured the thermal conductivity of BAs crystals around 186–350 Wm<sup>-1</sup> K<sup>-1</sup> [28,32,33]. This difference is attributed mainly to the difficulty in the fabrication of single crystals of boron-related materials as well as the complicated synthesis due to the high volatility and toxicity of arsenide atoms. Although the four-phonon scattering process is important for high-thermal-conductivity materials at high temperatures [34], the theoretical thermal conductivity

TABLE I. Top-14 materials with the lowest  $P_3$  values and their thermal conductivities calculated by the iterative Boltzmann transport equation solution.  $xx$ ,  $yy$ , and  $zz$  indicate the lattice directions.

Name	Structure	$P_3$ ( $10^{-4}$ cm)	Thermal conductivity ( $\text{Wm}^{-1} \text{K}^{-1}$ )				
			$xx$	This work $yy$	$zz$	Calc. Ref. $xx/yy$ ( $zz$ )	Expt. Ref. $xx/yy$ ( $zz$ )
Cubic BAs	$F-43m$	0.6397	3411	3411	3411	3170 [27]	351 [28]
Wurtzite BAs	$P63mc$	0.9064	2947	2947	1881	2380 (1210) [29]	
Diamond	$Fd-3m$	1.0005	3048	3048	3048	3450 [27]	3000 [1]
Lonsdaleite	$P63/mmc$	1.0335	2533	2533	2122	1500 (1270) [30]	
C (611426)	$P63/mmc$	1.2437	2842	2842	2675		
C (616440)	$P63/mmc$	1.2569	2583	2583	4214		
Be <sub>2</sub> C	$Fm-3m$	1.2596	117	117	117		
Cubic BN	$F-43m$	1.3300	1876	1876	1876	1800 [29]	768 [31]
BC <sub>2</sub> N (30148)	$P2221$	1.3670	895	910	804		
Wurtzite BN	$P63mc$	1.4394	1359	1359	1305	1230 (1040) [29]	
Cubic C <sub>3</sub> N <sub>4</sub>	$I-43d$	1.4490	234	234	234		
Pseudo C <sub>3</sub> N <sub>4</sub>	$P-43m$	1.4529	275	275	275		
BC <sub>2</sub> N (629458)	$Pmm2$	1.5127	1392	972	784		
BeCN <sub>2</sub>	$I-42d$	1.5472	351	351	440		

of BAs remains as high as  $2000 \text{ Wm}^{-1} \text{K}^{-1}$ . In fact, recent experiments realized cubic BAs with a thermal conductivity as high as  $1000 \text{ Wm}^{-1} \text{K}^{-1}$  [35–37].

Cubic BAs should be a reasonable demonstration of the effectiveness of the current screening. In this paper, we focus on the thermal conductivity at room temperature. Because the three-phonon scattering rate is much higher than the four-phonon scattering rate, employing three-phonon  $P_3$  is reasonable to find high-thermal-conductivity materials.

Another interesting feature of the top list is that it contains allotropes of known high-thermal-conductivity cubic materials (diamond, BN, and BAs). These include lonsdaleite, hexagonal diamonds, wurtzite BN, and wurtzite BAs, which have not been studied previously in terms of thermal conductivity. These materials are potential alternatives to their cubic counterparts. Although both cubic BN and wurtzite BN can be formed by compressing hexagonal BN, wurtzite BN is formed at much lower temperatures (around 2000 K) than cubic ones (3000–4000 K) [38]. Recently, a single-phase wurtzite BN bulk crystal was synthesized directly from a hexagonal BN bulk crystal under 10 GPa and 850 °C [39]. If the wurtzite structure has similar merits in other species, then BAs may be the first to benefit.

The list of top-100 materials with the smallest  $P_3$  also contains some typical high-thermal-conductivity materials like SiC (mp-ID: 8062), GaN (mp-ID: 830), and AlN (mp-ID: 1700) (see Supplemental Material, Table S3 [23]). Additionally, the top-100 list includes different crystal structures of GaN (mp-ID: 804) and AlN (mp-ID: 1330), which may display high thermal conductivities. Layered structure materials such as hexagonal BN and graphite have high thermal conductivities in the in-plane direction, but their out-of-plane thermal conductivity is very low due to the weak atomistic interaction. Hexagonal BN (mp-ID: 984) is in the top-100 prediction list as well as other BNs with layered structures. Examples include mp-ID: 7991, 685145, 13150, 604884, 629015, and 569655. Moreover, the list has some graphite structures (mp-ID: 568806, 632329, 990448, 568286 and 569304). In this

work, we evaluate the thermal conductivity with the scattering phase space, which is a scalar parameter that tends to suggest crystals with a high thermal conductivity in all three-lattice directions. The experimental conditions are the main reason why two-dimensional (2D) materials such as hexagonal BN and graphite do not appear in our top-14 list.

### C. Hardness versus thermal conductivity

An interesting feature of the screening results is that most of the top-14 list are superhard materials, including diamond, carbon nitride, born nitride, and heterodiamond. The Vickers hardness of diamond is around 115 GPa [40], which is the highest among reported superhard crystals, including cubic BN (62 GPa) [41], cubic BC<sub>2</sub>N (76 GPa) [40], C<sub>3</sub>N<sub>4</sub> (37–90 GPa) [42], and BeCN<sub>2</sub> (37 GPa) [43]. Be<sub>2</sub>C has a Knoop hardness of  $2410 \text{ kg mm}^{-2}$ , whereas diamond has a Knoop hardness of  $7000 \text{ kg mm}^{-2}$  [44].

The shear modulus is roughly proportional to the hardness. In the past two decades, it has been used as a guide for theoretical predictions of hard materials. Figure 4(a) plots the calculated shear modulus versus the average  $\kappa_L$ . Materials with a superhard property do not necessarily result in a high  $\kappa_L$ . The shear modulus of cubic and wurtzite BAs is only around 123 GPa which is the lowest among the top-14 materials, but their thermal conductivities exceed  $2500 \text{ Wm}^{-1} \text{K}^{-1}$ . Another example is superhard cubic silicon nitride (Si<sub>3</sub>N<sub>4</sub>). Although its reported Vickers hardness is around 35 GPa [45], the calculated thermal conductivity is only around  $81 \text{ Wm}^{-1} \text{K}^{-1}$  [46], which is much lower than that of BeCN<sub>2</sub> with the same order of hardness.

Figures 4(b)–4(d) show the average group velocity, heat capacity, and phonon relaxation time of the top-14 materials, respectively. Cubic and wurtzite BAs have group velocities lower than other hard materials, but they have the highest relaxation times. Consequently, they have a high  $\kappa_L$  comparable with diamond. Although superhardness means large elastic constants and group velocities of phonons in the

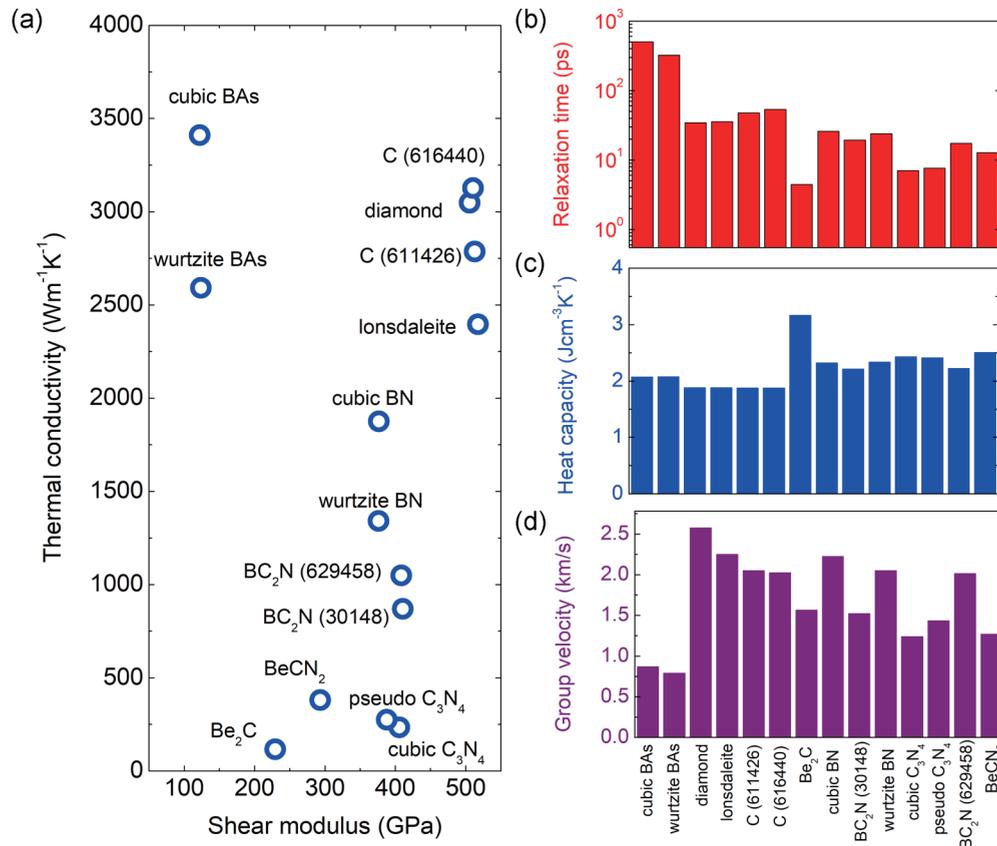


FIG. 4. Hardness versus thermal conductivity and comparison of parameters related with thermal conductivity. (a) Thermal conductivity versus shear modulus for the top-14 materials. (b)–(d) Average group velocity, heat capacity, and relaxation time of the top-14 materials.

linear dispersion regime,  $\kappa_L$  is also determined by other factors such as the phonon relaxation time (i.e.,  $P_3$  and anharmonic scattering amplitude). In addition, the properties of phonons with nonlinear dispersions largely influence the  $\kappa_L$ . This highlights the necessity and importance of developing rapid screening models to explore high- $\kappa_L$  crystals from databases.

Superhard materials consisting of B, C, and N atoms like BN, BC<sub>2</sub>N, and C<sub>3</sub>N<sub>4</sub> are advantageous over diamond in terms of stability and oxidation because the covalent bond energies between B-N (−117.19 eV) and C-N (−141.74 eV) are stronger than that of C-C (−103.64 eV) [47]. Mixing diamond with BN as a starting material may create new BCN alloy compounds under high pressure and temperature, which are more stable thermally and chemically than diamond and harder than BN.

Among ternary BCN compounds, heterodiamond in the form of BC<sub>2</sub>N has gained some attention. Heterodiamond has various structural forms ranging from layered graphitelike and diamond structures. However, the top-14 list includes two cubic BC<sub>2</sub>N (mp-ID: 30148 and 629458) structures with thermal conductivities around 784–1392 Wm<sup>-1</sup>K<sup>-1</sup>. Polycrystalline cubic BC<sub>2</sub>N materials have been synthesized from hexagonal BN at 20 GPa and 2200–2250 K [48] and from graphitelike BC<sub>2</sub>N above 18 GPa and 2100–2200 K [40,41]. The measured hardness of synthesized cubic BC<sub>2</sub>N is higher than that of a cubic BN single crystal but lower than diamond [40,41]. The current finding that cubic BC<sub>2</sub>N has a high potential to be

an ultrahigh thermal conductor is a motivation to improve the synthesis techniques of heterodiamond.

C<sub>3</sub>N<sub>4</sub> is another interesting superhard material with a reported hardness around 37–90 GPa [42]. Here we found two cubic structures of C<sub>3</sub>N<sub>4</sub> in the top-14 list. One is pseudo C<sub>3</sub>N<sub>4</sub> (mp-ID: 571653) and the other is cubic C<sub>3</sub>N<sub>4</sub> (mp-ID: 2852) (Fig. 3). The former has a defect zinc-blende structure with a hole in the central region of the unit cell. The latter has as many as 14 atoms in the primitive unit cell (see Supplemental Material, Fig. S3 [23]), which gives rise to complex phonon modes with many branches. Despite their apparent defective and complex structures, their thermal conductivities exceed 200 Wm<sup>-1</sup>K<sup>-1</sup>. Efforts have been made to synthesize different phases of carbon nitrides [49]. Martin-Gil *et al.* [49] synthesized a pseudo C<sub>3</sub>N<sub>4</sub> by a chemical precursor route under 800 °C and claims the process is scalable.

Lonsdaleite, which is also a wurtzite structure, has been studied previously as its hardness is comparable or even harder than diamond [50]. Its thermal conductivity (2122–2533 Wm<sup>-1</sup>K<sup>-1</sup>) is also comparable with diamond. Recently, polycrystalline lonsdaleite has been successfully synthesized in a diamond anvil cell at 100 GPa and 400 °C [51] using graphitic layers, which provide a low-energy barrier for progressive transformation from graphite to lonsdaleite. The synthesis temperatures are well below those previously reported for lonsdaleite [50].

The other two hexagonal diamonds (mp-ID: 611426 and 616440) combine the structure features of diamond and

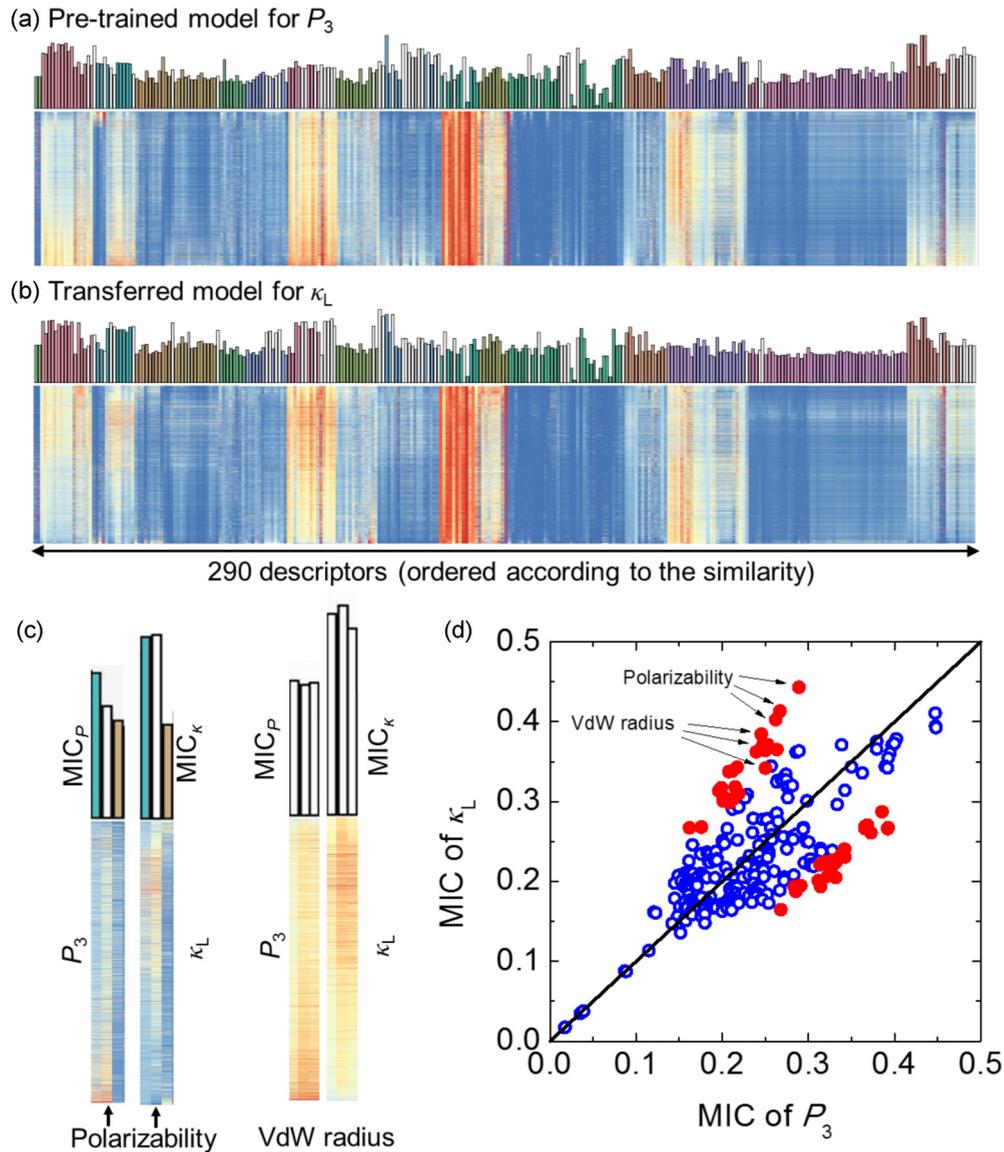


FIG. 5. Descriptor-property heat map and maximal information coefficient (MIC) scores. Descriptor-property heat map of (a) the pretrained model for  $P_3$  and (b) transferred model for  $\kappa_L$ . Top bar plots show the MIC scores for each descriptor. (c) Enlarged heat map for the descriptors of polarizability and the van der Waals (VdW) radius. (d) Distribution of the MIC scores for the 290 descriptors with respect to  $P_3$  and  $\kappa_L$ . Solid red dots denote 57 key descriptors with the MIC scores exhibiting significant differences between  $P_3$  and  $\kappa_L$  (see Supplemental Material, Tables S5 and S6 [23]).

lonsdaleite (see Supplemental Material, Fig. S4 [23]). These can be described as diamond-lonsdaleite superlattices. Their thermal conductivities are comparable with diamond.

#### D. Knowledge gained from transfer learning

The comparison between the pretrained and transferred models provides some important physical indications. Here, the pretrained model only involves the harmonic phonon properties, whereas the transferred model also requires anharmonic properties (i.e., the magnitude of the three-phonon scattering obtained from cubic interatomic force constants). The success of transfer learning means that the correlation between  $P_3$  and  $\kappa_L$  can be learned from that between the basic crystal structure information and  $P_3$ , revealing an underlying

commonality of the descriptors corresponding to the harmonic and anharmonic properties.

To understand the differences in how to recognize structure-property relationships for the pretrained and transferred models, we created a descriptor-property heat map (Fig. 5). For each model, the 290 descriptor vectors of the  $\sim 60\,000$  candidates to be screened are displayed onto the heat map. The candidate materials in the heat map are sorted according to a descending order of the predicted values of  $P_3$  or  $\kappa_L$ . This visualization reveals the presence of key descriptors relevant to the pattern recognition inherent in the trained model. Irrelevant or relevant descriptors might exhibit random or nonrandom patterns such as a linear trend along with the ordered predicted properties from top to bottom.

We investigated the underlying mechanisms responsible for the successful transfer from  $P_3$  to  $\kappa_L$ . We aimed to interpret key features that distinguish between the pretrained and post-transferred models. To quantitatively assess the dependency (relevance or association) between each descriptor and the predicted  $P_3$  or  $\kappa_L$ , we introduced the maximal information coefficient (MIC). MIC is a common measure of nonlinear correlations in bivariate random variables [52].

The bar plots in Fig. 5 show the MIC scores of the 290 descriptors with regard to  $P_3$  and  $\kappa_L$ . While most of the realized MICs do not change significantly between  $P_3$  and  $\kappa_L$ , some descriptors exhibit either a significant increase or decrease in the MICs. By looking at the descriptors discriminating the mechanisms regulating  $P_3$  and  $\kappa_L$  where the differences of  $MIC_P$  and  $MIC_\kappa$  are  $\pm 0.09$  (see Supplemental Material, Tables S5 and S6 [23]), we can identify descriptors that are relevant to the difference between  $P_3$  and  $\kappa_L$ .

The descriptors related with the average or maximum dipole polarizability and van der Waals (vdW) radius over all atoms in the crystal compounds are relevant to  $\kappa_L$  but not to  $P_3$ . The polarizability generally correlates with the interactions between electrons and the nucleus. Atoms with a larger number of electrons or atomic radius tend to have a high polarizability. In crystal compounds with a large electronic polarizability, the displacement or force perturbation can be easily transferred and persists over a long range via the orbital electron interaction with the nucleus. A typical example is rocksalt IV-VI materials like PbTe crystal, where the resonant bonding [53] and the corresponding large anharmonic interatomic-force constants are manifested due to the long-range polarization.

The maximum vdW radius that characterizes the non-bonded interactions between atoms also affects the anharmonicity in crystals. The vdW radius is related with the polarizability via the relation  $V_w = \alpha / (4\pi\epsilon_0)$ , where  $\alpha$  is the polarizability,  $\epsilon_0$  is the relative permittivity, and  $V_w$  is the vdW volume, which is given by the vdW radius. Since the descriptors correlated with  $\kappa_L$  but not with  $P_3$  should govern the linear output layer between the subnetwork pretrained by  $P_3$  and the final  $\kappa_L$ , it makes sense that polarizability and the vdW radius related to anharmonicity are the descriptors. However, it is meaningful to quantitatively identify that the average or maximum dipole polarizability and the vdW radius are the leading descriptors among those that can also be qualitatively related to anharmonicity. This provides a search direction. In addition to a small  $P_3$ , the low average or maximum dipole polarizability and vdW radius are necessary to achieve a high  $\kappa_L$ .

#### IV. CONCLUSIONS

In summary, we screened over 60 000 crystal compounds with phonon  $P_3$  as the feature quantity and identified a set of semiconducting compounds with high thermal conductivities. Screening was performed based on our developed

feature-based transfer learning, which bridges the gaps between the “big data” required for credible machine learning and the “small data” of thermal conductivity. Transfer learning directly models the connection between the basic crystal information and the thermal conductivity with a neural network by transferring descriptors acquired through pretraining for  $P_3$ . The successful prediction of high-thermal-conductivity crystals demonstrates the advantage of extrapolative prediction via transfer learning, and reveals the descriptors that are dominantly correlated with the anharmonic phonon properties.

The final, obtained materials in the top-14 list by feature-based and transfer learning screening all show high thermal conductivities, including boron arsenides (BAs), carbon (C), boron nitride (BN), and heterodiamond (BC<sub>2</sub>N). They have thermal conductivities on the order of  $1000 \text{ Wm}^{-1} \text{ K}^{-1}$ , validating the accuracy and high efficiency of the developed screening method. Although most of these are superhard materials with a large group velocity, the results are nontrivial. It has been observed that for some superhard materials, a large dispersion could lead to a large  $P_3$  and limit thermal conductivity. Because screening via  $P_3$  prefers to search crystals with a high thermal conductivity in three different lattice directions, 2D materials such as hexagonal BN and graphite do not appear in our top-14 list due to the weak atomistic interaction in the out-of-plane direction.

The screening also identified known materials that have yet to be studied in the context of heat transport. These include two types of novel carbon crystals with mixed phases of diamond and lonsdaleite and two phases of BC<sub>2</sub>N. These materials may be advantageous over well-explored high-thermal-conductivity materials in terms of thermodynamic stability and facileness of synthesis. These findings should contribute to next-generation thermal management technology by broadening the alternatives of high-thermal-conductivity materials and adding degrees of freedom to their surface affinities with various other materials for composite syntheses and integration.

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