Dynamics of hydrogen guests in ice XVII nanopores

Leonardo del Rosso,¹ Milva Celli,^{1,*} Daniele Colognesi,^{1,*} Svemir Rudić,² Niall J. English,³ Christian J. Burnham,³ and Lorenzo Ulivi^{1,†}

¹ISC-CNR, via Madonna del Piano 10, I-50019 Sesto Fiorentino, Italy

²ISIS Facility, STFC Rutherford Appleton Laboratory, Chilton, Oxfordshire OX11 0QX, United Kingdom ³School of Chemical and Bioprocess Engineering, University College Dublin, Belfield, Dublin 4, Ireland (Received 28 June 2017; published 27 November 2017)

The present high-resolution inelastic neutron scattering experiment on ice XVII, containing molecular hydrogen with a different ortho/para ratio, allows one to assign the H₂ motion spectral bands to rotational and center-of-mass translational transitions of either *para*- or *ortho*-H₂. Due to its structure, ice XVII confines H₂ molecules to move in spiral channels of molecular size. Reported data demonstrate that H₂ molecules rotate almost freely in these nanometric channels, though showing larger perturbation than in clathrate hydrates, and perform a translational motion exhibiting two low-frequency excitations. The agreement between the experimental spectra and the corresponding molecular dynamics results clearly enables one to portray a picture of the confined motions of a hydrophobic guest within a metastable ice framework, i.e., ice XVII.

DOI: 10.1103/PhysRevMaterials.1.065602

Ice XVII is a newly discovered form of solid water obtained from the so-called C_0 phase of the H_2 - H_2 O binary compound, quenched at a temperature T=77 K and ambient pressure, after letting the hydrogen molecules diffuse out of the crystal [1]. It is a pure water crystal, metastable at ambient pressure if maintained below 130 K. Its crystal structure is intrinsically porous and presents accessible channels where hydrogen molecules have been located during the production and where other molecules (belonging to hydrogen or another gas) can be absorbed again, confined in an essentially one-dimensional geometry [2]. The diameter of these channels, measured from the center of the oxygen atoms, is about 6.10 Å. This exotic and low-density water crystal adds to the list of solid structures of water possibly stable at negative pressure [3].

The study of the dynamics of hydrogen molecules in nanoconfinement, that is intrinsically quantum mechanical, is of great importance from both a practical and fundamental point of view. In hydrogen clathrate hydrates, which possess a similar chemical composition, although a different structure and stoichiometry, the hydrogen molecules confined in nearly spherical cages [4] perform an almost free rotation and a deeply nonharmonic center-of-mass (c.m.) vibrational motion (rattling), both of which have been experimentally investigated by inelastic neutron scattering [5–9] and Raman scattering [10–13]. In this paper we discuss the results of a combined experimental and simulation study on the dynamics of the H₂ guests in D₂O ice XVII and of the D₂O host lattice.

Samples for the present inelastic neutron scattering (INS) experiment were produced at ISC–CNR using D_2O and H_2 , as described in Ref. [1]. This isotopic composition is chosen in order to exploit the intrinsic advantage of the large incoherent neutron scattering cross section of the proton (compared to

both D and O), thus allowing for a relatively simple access to the self-dynamics of molecular hydrogen. Measurements have been performed at T=15 K on TOSCA, a high-energy resolution (ΔE) spectrometer at ISIS (U.K.) characterized by $1.5\% \lesssim \Delta E/E_i \lesssim 2.5\%$, with E_i being the incoming neutron energy. Raw time-of-flight data have been transformed into energy-transfer spectra, taking into account both the correction for the kinematic factor and the normalization for the incoming neutron flux.

We have measured the spectra for three different gascharged samples and one reference sample of pure deuterated ice XVII. Initially, we probed the material as prepared, i.e., the metastable H_2 - D_2O compound in the C_0 phase, quenched at low temperature and ambient pressure. This material, as for structure and composition, does not differ much from ice XVII when refilled with H₂, apart for small possible nitrogen impurities [2] which do not give a visible signal in this experiment. After a thermal treatment at about 120 K that removes all the guest molecules (described in Ref. [1]), we have recorded a spectrum of pure deuterated ice XVII at T = 15 K. Subsequently, we have measured two spectra of ice XVII loaded, respectively, with normal (n-H₂) and paraenriched hydrogen (p-rich H₂). Due to the limited working pressure of our aluminum sample cell, a quite low temperature (i.e., T = 20 K) has been chosen for the gas loading processes. According to previous work [1,14], this produces ice XVII with a H₂/D₂O molar ratio of about 25%, that is, half filling of the H₂ crystallographic sites.

The measured spectra, before any analysis, are shown in Fig. 1 on a logarithmic horizontal axis. Many features due to translational and librational excitations of the lattice are already recognizable in the spectrum of empty ice XVII (blue line). The other two spectra from the hydrogen-charged samples (black n-H₂, red p-rich H₂), present, in addition, several narrow and intense bands due to the molecular hydrogen dynamics. The two main H₂-molecule rotational features are readily identified in the spectrum and assigned, namely, the strong $J=0 \rightarrow 1$ rotational band of para-hydrogen (p-H₂), i.e., a triplet at \simeq 14 meV, and the $J=1 \rightarrow 2$ rotational band of ortho-hydrogen (o-H₂) around 29 meV.

^{*}Present address: Istituto di Fisica Applicata "Nello Carrara", IFAC-CNR, Sesto Fiorentino FI, Italy.

[†]Present address: Istituto di Fisica Applicata "Nello Carrara", IFAC-CNR, Sesto Fiorentino FI, Italy and Michigan Technological University, Houghton, MI 49931; l.ulivi@ifac.cnr.it

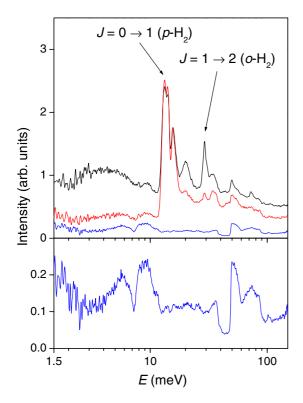


FIG. 1. Bottom panel: INS spectrum of deuterated ice XVII. Top panel: The two upper traces are the spectra of ice XVII refilled with normal (black line) and para-enriched (red line) hydrogen. The blue line represents the same spectrum as in the bottom panel replotted here for an easier comparison. All spectra have been recorded at $T=15~\rm K$. The logarithmic horizontal scale implies that the widths of the bands in different spectral regions are not readily comparable by eye.

For the identification of the c.m. excitations a subtraction procedure is applied, similar to that described in Ref. [5]. First, the hydrogen contribution was obtained subtracting the weak reference spectrum (i.e., that of empty ice XVII plus the aluminium cell, adjusted by considering self-shielding attenuation). The resulting H_2 spectrum can be modeled considering the neutron scattering cross section for the two hydrogen-molecule spin isomers, and its dependence on the rotational transitions [15]. After neglecting the coherent part of the neutron scattering, the double differential scattering cross section becomes proportional to the self part of the dynamical structure factor for the c.m. motion, $S_{\text{self}}(Q,\omega)$, and can be written as [16]

$$\frac{d^2\sigma}{d\Omega d\omega} = \frac{k_{\rm f}}{k_{\rm i}} S_{\rm self}(Q,\omega) \otimes \sum_{IJ'} \delta(\omega - \omega_{JJ'}) \nu(J,J',Q), \quad (1)$$

where k_i and k_f stand for the initial and final neutron wave vectors, respectively, ω is simply $\hbar^{-1}E$, and the symbol \otimes represents a convolution product. This expression holds whenever the hypothesis of decoupling between rotational and c.m. motions is satisfied. The Dirac delta functions reported in the previous equation are centered at the rotational transition energies $\hbar \omega_{JJ'}$ while $\nu(J,J',Q)$ (called molecular form factor) depends on both the momentum transfer Q and the rotational transition $J \to J'$ undergone by the molecule. Therefore,

the expected neutron spectrum is made of a comb of c.m. excitations, replicated and shifted by the energy of any possible rotational excitation of the single molecule. The molecular form factors v(J,J',Q) can be easily calculated assuming either a rigid rotor [17] or a rotating harmonic oscillator [18] model.

Molecular hydrogen trapped in ice XVII channels is a nonequilibrium mixture of o-H₂ and p-H₂, in a concentration which is essentially invariant in time, since the conversion rate is extremely low. At the low-temperature values typical of the present experiment, only the lowest rotational state for each species is populated (namely, J = 0 for p-H₂ and J = 1for o-H₂) and so few transitions contribute to the spectrum in the frequency region of interest, namely, the rotationally elastic $J = 1 \rightarrow 1$ and the inelastic $J = 1 \rightarrow 2$ transition of o-H₂, plus the inelastic transition $J = 0 \rightarrow 1$ of p-H₂. We remark that the $J = 0 \rightarrow 0$ transition of the p-H₂ molecule, being weighted by the mere coherent cross section, does not contribute appreciably to the observed spectrum. As a consequence, we assume that the spectral intensity recorded in the energy range below 10 meV is solely due to ortho molecules. Being the strong $J = 0 \rightarrow 1$ band around 14 meV due only to p-H₂, we can extract, by a linear combination of the two measured spectra (samples with n-H₂ and para-rich H_2), the spectra of pure p- H_2 and o- H_2 , represented in Fig. 2.

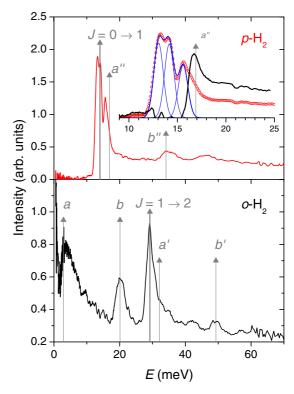


FIG. 2. Spectra of pure p-H $_2$ (top panel) and pure o-H $_2$ (bottom panel). The two observed c.m. excitations are marked with a and b, while we use a', b' and a'', b'' for the combinations of these excitations with $J=1\to 2$ and $J=0\to 1$ rotation, respectively. In the inset we show the decomposition of the $J=0\to 1$ rotational lines into three components. The residuals (scaled by a factor of 2 and plotted as a black line in the inset) highlight the presence of the intensity due to the a'' mode.

This decomposition enables one to highlight two intense bands due to the translational excitations of the molecule c.m., which appear at low energy in the spectrum of o-H₂, as combinations with the rotationally elastic $J=1 \rightarrow 1$ line. These bands, marked with letters a and b in the figure, are quite broad (about 4.0 and 3.5 meV, respectively). Band a is asymmetric, extending from 2.2 to 6.2 meV, while b has a more symmetrical shape. The position of these band is estimated at 2.9(8) and 20.1(6) meV. According to Eq. (1), the same c.m. excitations should give rise to similar combinations with the other rotational lines, shifted by an equal amount. This is indeed the case, as can be seen by observing Fig. 2, where the combination bands are evident exactly at the positions where they are expected, marked a'b' for o-H₂ (bottom panel) and a''b'' for p-H₂ (top panel).

In the inset of Fig. 2 we show the fitting of the $J=0 \rightarrow 1$ rotational line, exhibiting three separated components. The splitting of the rotational band into a triplet, commonly observed for hydrogen molecules in a confined geometry, is due to the lifting of the threefold degeneracy of the J=1 rotational level, as a consequence of the potential energy anisotropy with respect to the orientation of the H_2 molecule. This rotational triplet is nicely fitted by the sum of three Gaussians, whose energy positions are 13.05, 14.16, and 15.59 meV. Comparing these values with those measured for the same transition of H_2 in clathrate hydrates [5] (namely, 13.64, 14.44, and 15.14 meV), we observe here a larger splitting (i.e., 2.54 meV instead of 1.50 meV), proving a stronger potential energy anisotropy.

This fitting procedure also highlights the extra intensity due to the a'' band. Concerning the c.m. translational motion of the hydrogen molecule inside the channel, even though this has a spiral shape [2], it can be pictured as locally cylindrical. Therefore two vibrational frequencies are expected, the lower corresponding to the vibration along the cylinder axis, say, z, and the higher to the xy degenerate mode. This is the way we have assigned the a and b bands, observed in the spectra. In order to verify this assignment, we have performed classical molecular dynamics (MD) and calculated the spectrum of the c.m. motion.

The MD computation was performed at T = 50 K, with the water potential model TIP4P-2005 [19] and Alavi's H₂ model [20], using Partridge-Swenke and Morse intramolecular flexibility for water and H₂, respectively. The average structure of empty ice XVII is known from an accurate Rietveld refinement of neutron diffraction data [2] and is described by the space group P6₁22. Less certain but at any rate very similar is the structure of the samples containing H₂. A neutron diffraction measurement on the recovered C_0 phase [21] determined a lower symmetry space group ($P3_112$). On the contrary, Strobel et al. [22], from x-ray diffraction measurements in a diamond anvil cell, recently claimed to have determined that also the high-pressure phase C_0 exhibits a hexagonal $P6_122$ structure. The differences between the atomic coordinates of the two structures are numerically minimal, but are substantial to determine the potential energy shape for one H₂ molecule hypothetically located in the channel, as clearly discussed in Ref. [2]. Specifically, the ice lattice configuration based on the lower symmetry space group $P3_112$ generates three stable positions for the H₂ molecule per each hexagonal unit cell.

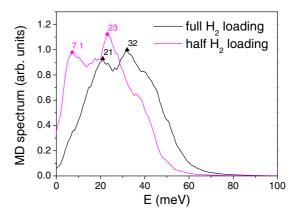


FIG. 3. Spectra of the c.m. velocity autocorrelation function, obtained by MD calculation, for completely hydrogen filled (black line) and randomly half-filled (magenta line) ice XVII. Symbols mark the positions of the maxima of each curve.

For the MD calculation aimed to study the c.m. dynamics, we start from an initial configuration made by 384 water molecules, located on the lattice sites of a supercell $4 \times 2 \times 4$ times larger than the conventional orthorhombic unit cell for the space group $P6_122$. The assumed lattice constants are a=6.326 Å and c=6.080 Å for the hexagonal cell (corresponding to an orthorhombic cell $6.326 \times 10.957 \times 6.080$ Å³). The O atoms of the water molecules in the supercell are placed on their crystallographic positions. However, the D atoms in deuterated ice XVII are configurationally disordered. Different random distributions of the framework water deuterons were generated by a computer routine, each consistent with the Bernal-Fowler ice rules [23] and the periodic boundary conditions outside the supercell.

A proton configuration with a negligible dipole moment was selected for use in further calculations.

The spectra calculated for the H₂ c.m. motion are shown in Fig. 3, assuming full hydrogen loading (i.e., three H₂ molecules per primitive cell) or half loading (i.e., randomly filling half of the H₂ sites). The comparison with the experimental values shows a semiquantitative agreement, considering that our samples are in a situation close to half filling. The disagreement of the low-frequency value can be easily imputed to the approximations (classical motion, empirical interaction potentials, structural model, etc.) in the computation and, partly, to the large uncertainty of the experimental band energy. It is interesting to note that, by increasing the loading, the low-energy mode essentially disappears and the intensity of a mode at higher energy increases. This supports the assignment of the modes at about 7 and 23 meV to the motion along and across the spiral channels, respectively. The energy of the mode along the channels is much more sensitive to the amount of loading, which is very reasonable.

By means of the same MD computation, we have calculated the D-projected density of phonon states (DOPS), which is compared in Fig. 4 with the same quantity extracted from our neutron scattering spectra. The analysis of the experimental data has been done analogously to what presented in Ref. [24]. The first steps are straightforward and consist in the subtraction of the empty container contribution and the correction for

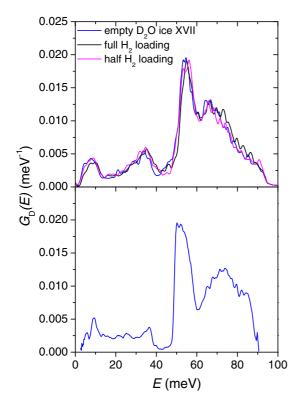


FIG. 4. Experimentally determined (lower panel) and calculated (upper panel) density of phonon states, projected on the deuteron nucleus. The calculation has been performed for different hydrogen fillings. Experimental data are derived from the spectra in Fig. 1 (bottom panel).

self-shielding attenuation, operated via the analytical approach suggested by Agrawal [25].

Multiple scattering contaminations made of two inelastic scattering events have been found to be practically negligible in the energy transfer interval of interest (i.e., 2 meV < E < 100 meV). Finally, in order to determine and remove the multiphonon terms in these spectra, we have used a well-known self-consistent procedure [26] already tested with a positive experience on a number of systems. Such a procedure, performed in the framework of the incoherent approximation, would have been totally justified for a polycrystalline H₂O-based material and would have led directly to the proton-projected density of phonon states, due to the much larger cross section, and lower mass, of the H nucleus than that of the O nucleus. For a deuterated sample such as ours, however, the rationale of this approximation is based on well-established results obtained on various forms of D₂O ice [27].

Despite the various approximations involved (i.e., a purely incoherent, harmonic, isotropic, and single-site treatment of the multiphonon terms), the results reveal a satisfactory convergence of the method, allowing for a sound extraction of the one-phonon component of the self-dynamic structure factor, and, finally, for the evaluation of the deuterium-projected density of phonon states in ice XVII, $G_D(E)$, which is reported in Fig. 4. It is worth noting that, in order to obtain an accurate result from a quantitative point of view, the oxygen contribution to the experimental estimate of $G_D(E)$

has been duly evaluated and subtracted. This contribution has been found to be fully negligible in the librational part of the spectrum, while in the lattice phonon part it turned out to amount to 27.7% of the deuterium one. Due to the fact that the lattice phonon corresponds to a rigid motion of D_2O molecules as single units, then the D and O contributions to the density of phonon states in the corresponding spectral zone exhibit exactly the same spectral shape and so the aforementioned corrections to $G_D(E)$ can be simply performed by scaling the intensity of the lattice part of $G_D(E)$ (in the energy transfer range from 0 to 42 meV) by a factor 0.783, obtained by taking into account the different cross section of the D_2O molecule and D nucleus.

The calculated DOPS does not depend appreciably on the hydrogen loading, and shows some bands originated by different vibrational modes. By looking at Fig. 4, one can observe that the lattice phonon region (i.e., E < 42 meV) of $G_D(E)$ is different from what can be observed in basically all other ice forms [27], but is quite similar to the lattice band in deuterated sII clathrate hydrate (filled with Ne) [24], even though the cleft of the main acoustic peak seems less deep and more asymmetric in ice XVII. In the librational region (i.e., 45 meV < E < 90 meV), one also notes some differences between $G_D(E)$ of ice XVII and that of sII clathrate hydrate. In particular, the first steep peak, located at about 50 meV, appears more asymmetric and centered at a slightly lower frequency in ice XVII, followed by a dip, at around 60 meV, which looks shallower and less pronounced.

Altogether, the comparison with the computed $G_D(E)$ shows a reasonably good agreement, similar to what obtained for clathrate hydrates of various structures, as discussed in Refs. [28,29].

In this work we have accurately measured the quantum dynamics of a single H₂ molecule in the confined geometry of one single nanometric ice XVII channel. The splitting of the rotational and translational bands is a consequence of the water environment, whose anisotropy appears stronger than in the clathrate hydrate confinement. As for the c.m. translational dynamics, the comparison between the measured spectra and the MD calculations supports the identification of the lowest-frequency band as the vibration along the channel direction, while the higher mode corresponds to the motion across the spiral channel. Both frequencies are significantly influenced by the hydrogen loading, but this dependence is rather more marked for the vibration along the channel. The present data about the guest motion, together with a substantial agreement between the measured and calculated D-projected DOPS of the host molecules, allow one to characterize the main features of the vibrational dynamics of this novel inclusion compound.

We gratefully thank Andrea Donati for his skillful technical support in the sample preparation. We acknowledge the PRIN project ZAPPING, No. 2015HK93L7, granted by the Italian MIUR (Ministero dell'Istruzione, dell'Università e della Ricerca). This work has been performed within the Agreement No. 0018318 (02/06/2014) between STFC and CNR, concerning collaboration in scientific research at the spallation neutron source ISIS.

- L. del Rosso, M. Celli, and L. Ulivi, Nat. Commun. 7, 13394 (2016).
- [2] L. del Rosso, F. Grazzi, M. Celli, D. Colognesi, V. Garcia-Sakai, and L. Ulivi, J. Phys. Chem. C 120, 26955 (2016).
- [3] Y. Huang, C. Zhu, L. Wang, X. Cao, Y. Su, X. Jiang, S. Meng, J. Zhao, and X. C. Zeng, Sci. Adv. 2, e1501010 (2016).
- [4] E. D. Sloan, Clathrate Hydrates of Natural Gases (Dekker, New York, 1997).
- [5] L. Ulivi, M. Celli, A. Giannasi, A. J. Ramirez-Cuesta, D. J. Bull, and M. Zoppi, Phys. Rev. B 76, 161401(R) (2007).
- [6] M. Xu, L. Ulivi, M. Celli, D. Colognesi, and Z. Bačić, Phys. Rev. B 83, 241403(R) (2011).
- [7] M. Xu, L. Ulivi, M. Celli, D. Colognesi, and Z. Bačić, Chem. Phys. Lett. 563, 1 (2013).
- [8] M. Celli, A. Powers, D. Colognesi, M. Xu, Z. Bačić, and L. Ulivi, J. Chem. Phys. 139, 164507 (2013).
- [9] D. Colognesi, M. Celli, L. Ulivi, M. Xu, and Z. Bačić, J. Phys. Chem. A 117, 7314 (2013).
- [10] A. Giannasi, M. Celli, L. Ulivi, and M. Zoppi, J. Chem. Phys. 129, 084705 (2008).
- [11] T. A. Strobel, E. D. Sloan, and C. A. Koh, J. Chem. Phys. 130, 014506 (2009).
- [12] A. Giannasi, M. Celli, M. Zoppi, M. Moraldi, and L. Ulivi, J. Chem. Phys. 135, 054506 (2011).
- [13] M. A. S. Zaghloul, M. Celli, N. M. Salem, S. M. Elsheikh, and L. Ulivi, J. Chem. Phys. 137, 164320 (2012).
- [14] L. del Rosso, M. Celli, and L. Ulivi, Challenges 8, 3 (2017).

- [15] S. W. Lovesey, Theory of Neutron Scattering from Condensed Matter (Clarendon, Oxford, U.K., 1984), p. 24.
- [16] D. Colognesi, M. Celli, and M. Zoppi, J. Chem. Phys. 120, 5657 (2004)
- [17] J. A. Young and J. U. Koppel, Phys. Rev. 135, A603 (1964).
- [18] M. Zoppi, Physica B **183**, 235 (1993).
- [19] C. J. Burnham and N. J. English, J. Phys. Chem. C 120, 16561 (2016).
- [20] S. Alavi, J. A. Ripmeester, and D. D. Klug, J. Chem. Phys. 123, 024507 (2005).
- [21] V. S. Efimchenko, M. A. Kuzovnikov, V. K. Fedotov, M. K. Sakharov, S. V. Simonov, and M. Tkacz, J. Alloys Compd. 509, S860 (2011).
- [22] T. A. Strobel, M. S. Somayazulu, S. V. Sinogeikin, P. Dera, and R. J. Hemley, J. Am. Chem. Soc. 138, 13786 (2016).
- [23] J. D. Bernal and R. H. Fowler, J. Chem. Phys. 1, 515 (1933).
- [24] M. Celli, D. Colognesi, L. Ulivi, M. Zoppi, and A. J. Ramirez-Cuesta, J. Phys.: Conf. Ser. 340, 012051 (2012).
- [25] A. K. Agrawal, Phys. Rev. A 4, 1560 (1971).
- [26] J. Dawidowski, F. J. Bermejo, and J. R. Granada, Phys. Rev. B 58, 706 (1998).
- [27] J.-C. Li, J. Chem. Phys. 105, 6733 (1996).
- [28] M. Celli, D. Colognesi, A. Giannasi, L. Ulivi, M. Zoppi, V. G. Sakai, and A. J. Ramirez-Cuesta, Adv. Sci. Technol. 72, 196 (2010).
- [29] C. J. Burnham and N. J. English, J. Chem. Phys. **144**, 164503 (2016).