

Single-crystal growth and physical properties of 50% electron-doped rhodate $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ Z. W. Li,¹ H. Guo,¹ Z. Hu,¹ T. S. Chan,² K. Nemkovski,³ and A. C. Komarek^{1,*}¹Max-Planck-Institute for Chemical Physics of Solids, Nöthnitzer Str. 40, D-01187 Dresden, Germany²National Synchrotron Radiation Research Center (NSRRC), 101 Hsin-Ann Road, Hsinchu 30077, Taiwan³Jülich Centre for Neutron Science JCNS at Heinz Maier-Leibnitz Zentrum (MLZ), Forschungszentrum Jülich GmbH, Lichtenbergstraße 1, 85748 Garching, Germany

(Received 30 June 2017; published 15 September 2017)

Centimeter-sized single crystals of $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ were grown by the floating zone method at oxygen pressures of 20 bar. The quality of our single crystals was confirmed by x-ray Laue, powder and single crystal x-ray diffraction, neutron and x-ray absorption spectroscopy measurements. At $\sim 50\%$ electron doping we observe RhO_3 octahedral rotations of $\sim 8.2^\circ$ within the octahedral basal plane which are incompatible with space group $I4/mmm$. Our single crystal was further characterized by susceptibility, electrical transport, and, finally, specific heat measurements showing a temperature-dependent Debye temperature.

DOI: [10.1103/PhysRevMaterials.1.044005](https://doi.org/10.1103/PhysRevMaterials.1.044005)

I. INTRODUCTION

Over the last decades, strongly correlated transition metal oxides have attracted enormous attention [1–6]. In $4d$ transition metal oxides especially the discovery of the possible occurrence of spin-triplet p -wave superconductivity in ruthenates triggered a lot of research activity [7,8]. However, the rhodium oxides that also crystallize in the K_2NiF_4 -type structure were less investigated so far. Besides of a comparative study of Sr_2RhO_4 single crystals, isotypic with Sr_2RuO_4 mainly polycrystalline $\text{Sr}_{2-x}\text{La}_x\text{RhO}_4$ samples were studied over a wider doping range $x = 0 \sim 1.0$ [9–11]. Regarding the recent observations of charge order and nanophase separation in the $3d$ cobaltate system $\text{La}_{2-x}\text{Sr}_x\text{CoO}_4$ [12–15], also the availability of single crystals of the $4d$ rhodium analog with hole-doping levels away from the end member Sr_2RhO_4 and, especially, around 50% electron doping ($\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$), would be of interest for future studies of the charge correlations.

In these rhodates, $\text{Sr}_{2-x}\text{La}_x\text{RhO}_4$, the crystal structure changes from orthorhombic for $x = 0 \sim 0.1$ to tetragonal for $x = 0.1 \sim 0.6$ and back to orthorhombic for $x = 0.6 \sim 1.0$. A doping dependent metal-insulator transition and a complex magnetic behavior that may be connected to the structural changes have been observed [9]. Furthermore, a quasi-two-dimensional Fermi liquid state has been reported in floating zone grown single crystals of the undoped parent compound Sr_2RhO_4 [10,11]. More recently, unconventional magnetism has been reported in Ga- or Ca-doped LaSrRhO_4 [16]. In these systems the occupation of the intermediate spin (IS) state ($S = 1$) of the Rh^{3+} ions might be the origin of this unconventional magnetic behavior. Note that the low spin state (LS) ($S = 0$) is commonly expected for the Rh^{3+} ions.

In the present work, we report the growth of centimeter-sized high-quality single crystals of 50% electron-doped Sr_2RhO_4 , i.e., $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$, by the flux feeding floating zone (FFFZ) method. When comparing with the expected value for a 50% Rh^{3+} ($S = 0$) and 50% Rh^{4+} ($S = 1/2$) solid solution, we find an enhancement of the effective magnetic

moment (mainly) in the in-plane direction. Moreover, a temperature-dependent Debye temperature can be observed for $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$. We discuss a possible thermal occupation of higher spin states of the Rh ions as a possible mechanism that might be responsible for the interesting physics revealed in this system.

II. RESULTS AND DISCUSSION

The $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ single crystal was grown following the procedure for the growth of $\text{Sr}_{2-x}\text{Ba}_x\text{RuO}_4$ [8]. First, powders of SrCO_3 (99.99%, Alfa Aesar), La_2O_3 (99.99%, Alfa Aesar), and $\text{Rh}_2\text{O}_3 \cdot x\text{H}_2\text{O}$ ($\text{Rh}\text{-}\% = 81.08\%$, Alfa Aesar) were mixed together and ground thoroughly before reaction at 1200°C in air for 24 hours. The resulting powder was packed into latex tubes and pressed under ~ 100 MPa hydrostatic pressure into rods of ~ 6 mm diameter and ~ 140 mm (~ 35 mm) length for the feeding (seed) rod. The obtained feed and seed rods were sintered at 1300°C for 48 hours. Similar to the growth of ruthenates [8] an excess of 5% Rh_2O_3 needs to be added into the feed rod (with nominal composition $\text{Sr}_{1.5}\text{La}_{0.5}\text{Rh}_{1.1}\text{O}_x$) in order to serve as self-flux and in order to compensate the heavy evaporation during the growth. These rods were, then, mounted in a high-pressure optical mirror furnace (High Pressure Crystal Growth Furnace, Scientific Instrument Dresden GmbH) equipped with one single Xenon lamp. A stable growth was maintained with a mixture of Ar and O_2 gas with a ratio of 1:1 flowing with a speed of 1 L/min at a pressure of 20 bar. The growth speed was set to 7.5 mm/h. The feed and seed rod were counter rotated with a speed of 30 rotations per minute. The growth direction of this single crystal is roughly the $[1\ 1\ 0]$ direction. After floating zone growth, the several-centimeter-sized as-grown crystal with 6–7 mm in diameter was annealed at 800°C in flowing O_2 for 7 days in order to remove any possible oxygen vacancies [11].

For powder XRD measurements parts of the grown single crystals have been ground into fine powders. The XRD measurements have been performed with a 2θ step of 0.01° with $\text{Cu } K_{\alpha 1}$ radiation on a Bruker D8 Discover A25 powder x-ray diffractometer. The FullProf program package [17] was used for Le Bail fits.

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Neutron scattering experiments were performed at the DNS spectrometer [18] at the Heinz Maier-Leibnitz Zentrum using an incident neutron wavelength of $\lambda = 4.2 \text{ \AA}$. Using Mo K_α radiation, single-crystal x-ray diffraction measurements have been performed on a *Bruker D8 VENTURE* single-crystal x-ray diffractometer equipped with a bent graphite monochromator for about $3\times$ intensity enhancement and a *Photon CMOS* large area detector. A spherical sample with roughly $100\text{-}\mu\text{m}$ diameter has been measured and a multiscan absorption correction has been applied to the data (minimum and maximum transmission: 0.6203 and 0.7510, respectively). For space group $I4/mmm$, 42677 reflections (h : $-10 \rightarrow 10$, k : $-7 \rightarrow 10$, and l : $-34 \rightarrow 31$) have been collected with an internal R value of 3.62%, a redundancy of 67.53 and 99.21% coverage up to $\sin(\Theta)/\lambda = 1.359$. For the final structure refinement with space group $P2_1/c$, the integration was redone and the refinement was based on a set of 80499 reflections (h : $-18 \rightarrow 18$, k : $-14 \rightarrow 13$, and l : $-31 \rightarrow 34$) which have been collected with an internal R value of 3.15%, a redundancy of 14.48, 100% coverage up to $2\Theta = 97.34^\circ$, and 99.1% coverage up to $2\Theta = 144.94^\circ$. Within this refinement, four different twin domains with the following volume fractions and twin laws have been refined: $V_1 = 0.105(7)$, $V_2 = 0.403(4)$, $V_3 = 0.152(4)$, $V_4 = 0.340(4)$, $M_2 = (0.000\ 2.000\ 0.000, 0.500\ 0.000\ 0.000, 0.500\ 1.000 - 1.000)$, $M_2 = (1.000\ 0.000\ 0.000, 0.000 - 1.000\ 0.000, 1.000\ 0.000 - 1.000)$, and $M_3 = (0.000 - 2.000\ 0.000, 0.500\ 0.000\ 0.000, -0.500 - 1.000\ 1.000)$. The obtained structural parameters are listed in Table II. The JANA2006 program suite [19] was used for all crystal structure refinements.

The valence state of Rh was identified by x-ray absorption spectroscopy (XAS) using bulk sensitive fluorescence yield at the Rh- L_3 edge measured at the 16A beamline at the National Synchrotron Radiation Research Center in Taiwan. Using a SQUID magnetometer (MPMS, Quantum Design) the magnetization was measured as a function of temperature with magnetic field applied along both crystallographic [110] and [001] directions. For resistivity measurements, the crystal was cut into a bar shape with typical dimensions of $\sim 1 \times 1 \times 2 \text{ mm}^3$. Four probe contacts were made by connecting Au wires using silver paste and the measurements were performed by using a Physical Property Measurement System (PPMS, Quantum Design). The crystal was cut into a typical dimension of $\sim 2 \times 2 \times 0.5 \text{ mm}^3$ along the a , b , and c axes and the specific heat was measured by using the relaxation method in our PPMS system. Note that all directions for susceptibility and resistivity measurements are given in tetragonal notation.

A. Structure

In Fig. 1, the powder x-ray diffraction (XRD) pattern of our $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ single crystal (crushed to powder) is shown. All peaks can be indexed with the tetragonal structure model with space group $I4/mmm$. This is consistent with literature reports on polycrystalline powder samples [9]. The lattice constants amount to $a = b = 3.8989(8) \text{ \AA}$ and $c = 12.667(3) \text{ \AA}$. Laue diffraction measurements confirm the single crystallinity of our $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ crystals, see Fig. 2. Moreover, neutron diffraction measurements that are penetrating and probing the

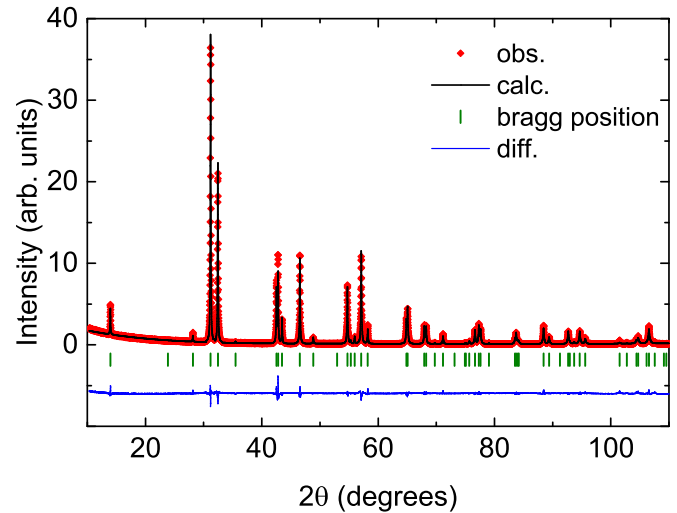


FIG. 1. Powder x-ray diffraction pattern of the grown crystal $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ collected at room temperature. The solid line represents a Le Bail fit of the data using the FULLPROF software package [17]. The calculated Bragg peak positions according to space group $I4/mmm$ are indicated by vertical bars and the difference between the experimental and calculated intensities are shown as a solid blue line at the bottom.

entire single crystal further confirm the single crystallinity, see Fig. 3.

However, within a single-crystal x-ray diffraction measurement—see Fig. 4—we find evidence for additional oxygen octahedral rotations around the c axis. Although less reflections are considered for the higher symmetric space group $I4/mmm$, a refinement with such a structure model yields enhanced values for the goodness of fit (GoF) and strongly elongated probability ellipsoids of the basal oxygen ions in a transversal direction to the Rh-O bonds, see Fig. 5(a). Therefore we tried to overcome these problems by applying

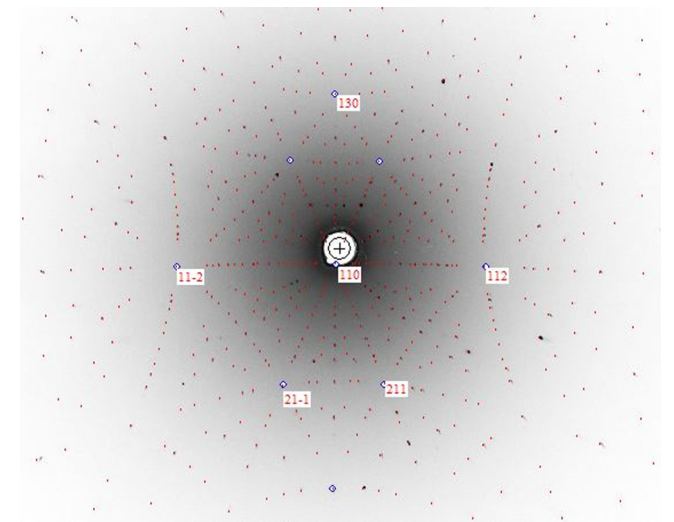


FIG. 2. Laue diffraction pattern of $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$. The red spot is a simulation (space group $I4/mmm$) using the software ORIENTEXPRESS.

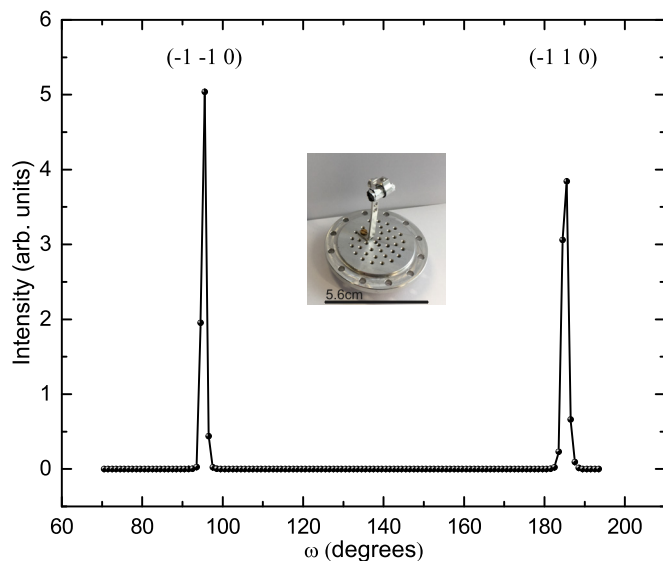


FIG. 3. Large rocking scans measured in neutron scattering experiments confirming the single crystallinity of $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$. The inset shows a picture of the measured single crystal mounted in an Al sample holder and the black bar indicates the length scale (56 mm).

several known structure models for layered perovskites based on space groups $Fm\bar{3}m$, $Bm\bar{3}b$, $P4_2/ncm$, $I4_1/acd$, and $I4_1/a$. But the strongly elongated probability ellipsoids that indicate RhO_3 octahedral rotations around the c axis remain essentially unchanged for all these space groups (including the subgroups $I4_1$ and $C2/c$). Moreover, GoF and R values stay always enlarged for all these structure models, see Table I. Especially, also the crystal structure of the end member Sr_2RhO_4 , which crystallizes with the symmetry of space group $I4_1/acd$ [20], does not yield any satisfactory description of our single-crystal x-ray data.

Therefore we started to search for solutions based on lower symmetric space groups. Finally, we were able to describe the crystal structure of $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ with space group $P2_1/c$ and to obtain acceptable GoF and R values, see Table I. Especially, the extremely strong elongation of the basal oxygen ellipsoids in a direction transversal to the Rh-O bonds, which was present in all other refinements (with anisotropic ADP) has almost vanished for the structure model based on the monoclinic space group $P2_1/c$, see Fig. 5. The $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ crystal structure contains RhO_3 octahedra that are elongated in apical direction [i.e., the

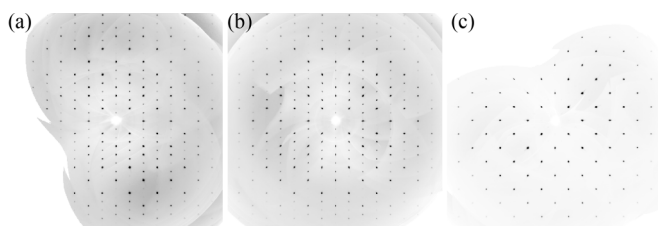


FIG. 4. Single-crystal x-ray diffraction intensities of $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ within the (a) $0KL$ (b) $H0L$ and (c) $HK0$ planes of reciprocal space (space group $I4/m\bar{3}m$).

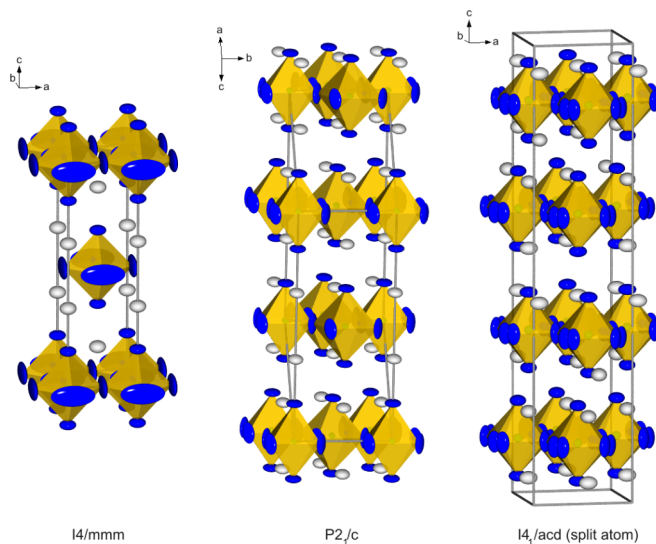


FIG. 5. Representation of the crystal structure of $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ as obtained from refinements with space group $I4/m\bar{3}m$, $P2_1/c$ ($a = 6.9155 \text{ \AA}$, $b = 5.5194 \text{ \AA}$, $c = 12.682 \text{ \AA}$, and $\beta = 156.48^\circ$) and with a split atom model based on space group $I4_1/acd$. Grey/green/blue ellipsoids denote the probability ellipsoids of Sr(La)/Rh/O atoms. As can be seen, for the previously reported space group $I4/m\bar{3}m$ of $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ the basal oxygen probability ellipsoids are extremely enlarged in direction perpendicular to the Rh-O bonds. This problem could be overcome only for the monoclinic symmetry of space group $P2_1/c$ or with a split atom model as reported in literature [21] for the end member Sr_2RhO_4 with space group $I4_1/acd$.

Rh-O distances amount to $\sim 2.080 \text{ \AA}$ in apical direction and $\sim 1.967(1) \text{ \AA}$ and $\sim 1.976(1) \text{ \AA}$ in basal direction]. Moreover, the basal Rh-O-Rh bond angles amount to $\sim 163.6(1)^\circ$ and, thus, indicate basal octahedral rotations of about 8.2° , whereas the apical oxygens are only tilted by less than 2° . Since mainly oxygen displacements are involved, these structural distortions are difficult to detect with powder x-ray diffraction, which might explain why these distortions could not be observed before (Fig. 1 and Ref. [9]). Moreover, this single-crystal x-ray diffraction measurement is able to corroborate a close to stoichiometric composition of our single crystal, i.e., the refinement of La, Sr, and Rh occupancies indicates the following composition: $\text{La}_{0.481(8)}\text{Sr}_{1.519(8)}\text{Rh}_{0.991(2)}\text{O}_4$. The details of the crystal structure are listed in Table II.

Finally, we would like to note that there was a structural study on Sr_2RhO_4 , which proposes the occurrence of microdomains in Sr_2RhO_4 with clockwise and anticlockwise rotations of the RhO_3 octahedra around the c axis that are allowed in space group $I4_1/acd$ [21]. A structure model with a statistical (random) rotation of the RhO_3 octahedra around the c axis was able to describe the neutron diffraction data of Sr_2RhO_4 properly [21]. This structure model has been realized by splitting the basal oxygen site according to the two different possible rotation directions—clockwise and anticlockwise—of the RhO_3 octahedra which are allowed within space group $I4_1/acd$ [21] (but not, e.g., in space group $I4/m\bar{3}m$ that has been previously reported for $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ [9]). The occupation of these split oxygen sites $\text{O}2$ and $\text{O}2'$ indicates

TABLE I. Refinement results of single-crystal x-ray diffraction measurements of $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$. Here, the GoF and R values obtained for refinements with different space groups are listed together with the largest value of any U_{ii} ; the symbol \dagger indicates that U_{iso} was listed because only a refinement with isotropic displacement parameters for the oxygen atoms turned out to be stable (all heavier atoms were still refined with anisotropic ADP). The bold values are strongly enlarged values, which are not in favor for the corresponding structure model. For all these space groups—apart from $P2_1/c$ —the basal oxygen ions have one extremely enlarged anisotropic displacement parameter U_{ii} . Finally, we were able to describe the crystal structure of $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ with space group $P2_1/c$, see also Fig. 5. However, also a split atom model \ddagger is able to overcome the highly enlarged U_{ii} . Nevertheless, the weighted R value stays significantly larger than for the monoclinic solution $P2_1/c$.

	GoF	R value (%)	R_w value (%)	largest U_{ii} (\AA^2)
$I4/mmm$	9.85	2.90	6.70	0.133(6)
$Fmmm$	7.43	2.91	6.69	0.076(14)
$Bmcb$	5.43	3.57	13.54	0.076(7)[†]
$P4_2/ncm$	5.18	4.64	13.73	0.065(6)
$I4_1/acd$	4.68	3.73	12.92	0.053(3)
$I4_1/a$	3.90	3.22	11.11	0.058(14)
$I4_1$	3.29	3.21	9.87	0.070(15)
$Ibca$	4.01	3.39	11.53	0.069(4)
$C2/c$	3.48	3.26	10.43	0.097(14)
$P2_1/c$	1.97	2.62	6.68	0.018(1)
split atom:				
$I4_1/acd^{\ddagger}$	3.70	3.08	10.22	0.032(1)

the total sizes of domains with clockwise and anticlockwise rotation. In principle, this split-atom model with the absence of fully long-range ordered rotations of the RhO_3 octahedra could also be a good description of the crystal structure of $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ because the major shortcoming of any crystal structure model that is listed in Table I is the extreme elongation of the probability ellipsoids of the basal oxygen ions, which strongly indicates a wrong description of the RhO_3 octahedral rotations around the c axis for all these models. Hence we also refined a split-atom model similar to the one in Ref. [21] for $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ —see Table I. Besides our monoclinic solution ($P2_1/c$), also this split-atom model with statistical rotations of the RhO_3 octahedra around the c axis [21] does not suffer from these extremely enlarged probability ellipsoids of the basal oxygen ions, see also Fig. 4. In this split-atom model, the RhO_3 octahedra are statistically rotated by $\pm\sim 8.0^\circ$ around the c axis, which is close to the value of the octahedral rotations in our monoclinic solution ($\sim 8.2^\circ$). However, the weighted R values are significantly better for our monoclinic structure model, see Table I. Hence the monoclinic solution appears to be more likely, and, maybe also the parent compound Sr_2RhO_4 needs to be reanalyzed. One main qualitative difference of these two structure models is the absence of any RhO_3 octahedral tilts (i.e., rotations either of the apical or of the basal oxygen ions around any rotational axis perpendicular to the tetragonal c axis) for the split-atom model based on space group $I4_1/acd$. Such tilts are known

TABLE II. Refinement results of single-crystal x-ray diffraction measurements of $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$. Here, the structural parameters of the refinement with space group $P2_1/c$ ($a = 6.9155 \text{ \AA}$, $b = 5.5194 \text{ \AA}$, $c = 12.682 \text{ \AA}$, $\beta = 156.48^\circ$) are listed, see also Fig. 5(b). Goodness of fit R and weighted R values amount to 1.97, 2.62%, and 6.68%, respectively.

Atom	occup.	x	y	z
Sr1	0.760(4)	0.00237(17)	−0.00066(9)	0.35698(9)
La1	0.240(4)	0.00237(17)	−0.00066(9)	0.35698(9)
Rh1	0.9909(18)	0	0	0
O1	1	−0.5733(5)	0.2855(2)	−0.7879(5)
O2	1	0.0052(16)	0.0016(9)	0.1662(8)
Atom	U_{11} (\AA^2)	U_{22} (\AA^2)	U_{33} (\AA^2)	
Sr1	0.00907(5)	0.00949(6)	0.00737(9)	
La1	0.00907(5)	0.00949(6)	0.00737(9)	
Rh1	0.00261(4)	0.00297(4)	0.00143(7)	
O1	0.0090(5)	0.0079(4)	0.0093(10)	
O2	0.0176(6)	0.0179(6)	0.0137(8)	
Atom	U_{12} (\AA^2)	U_{13} (\AA^2)	U_{23} (\AA^2)	U_{iso} (\AA^2)
Sr1	−0.00033(6)	0.00781(6)	−0.00026(10)	0.0076(3)
La1	−0.00033(6)	0.00781(6)	−0.00026(10)	0.0076(3)
Rh1	0.00001(7)	0.00179(4)	0.00002(10)	0.0026(2)
O1	−0.0060(2)	0.0067(6)	−0.0048(6)	0.015(3)
O2	0.0000(6)	0.0152(6)	−0.0001(9)	0.013(3)

to occur in layered perovskites with K_2NiF_4 structure (e.g., for space group $Bmcb$) and these kind of tilts are allowed for the monoclinic space group $P2_1/c$ in addition to the rotations of the RhO_3 octahedra around the tetragonal c axis. All these octahedral rotations are important for an understanding of the electronic structure of $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$, similar as was pointed out for the end member Sr_2RhO_4 in Ref. [22].

B. X-ray absorption spectroscopy

It is well known that XAS spectra at the $4d$ transition metal $L_{2,3}$ edge are very sensitive to the valence state—an increase of the valence state of the metal ion by one causes a shift of the XAS $L_{2,3}$ spectra by one or more eV towards higher energies [23]. The XAS spectrum of $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ is just in between that of the Rh^{3+} and Rh^{4+} references $\text{LaCo}_{0.5}\text{Rh}_{0.5}\text{O}_3$ and Sr_2RhO_4 , respectively. A superposition of 52.5% of the Rh^{3+} and 47.5% of the Rh^{4+} reference spectra is nicely able to reproduce the XAS spectrum of $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$, thus, indicating a Rh valence state of roughly 3.475+, which is close to its nominal value of 3.5+ and which would correspond to a small oxygen deficiency of the order of $\delta \sim -0.0125$ in our $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_{4+\delta}$ single crystal—see Fig. 6. However, the deviation of the measured valence from its nominal value is also of the order of the estimated error bar of $\sim 2\%$ in our XAS measurements.

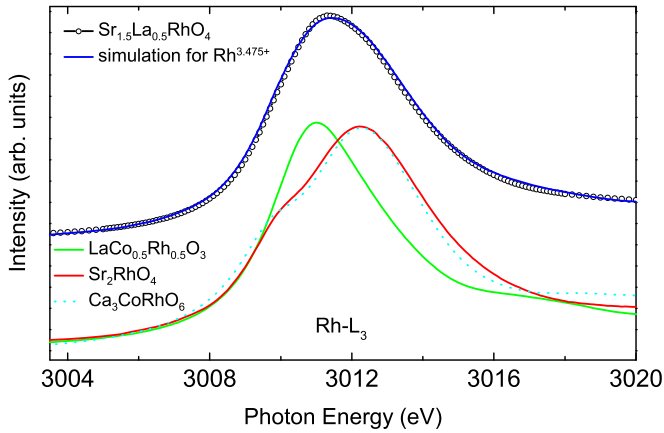


FIG. 6. X-ray absorption spectroscopy measurements. Below, the spectrum of $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ [(black circles) the Rh^{3+} and Rh^{4+} references $\text{LaCo}_{0.5}\text{Rh}_{0.5}\text{O}_3$ (green line)] and Sr_2RhO_4 (red line) are shown. Also the spectrum of $\text{Ca}_3\text{CoRhO}_6$ (dashed cyan line) as another Rh^{4+} reference is shown. A superposition of the Rh^{3+} and Rh^{4+} reference spectra (blue line) with weights of 52.5% and 47.5%, respectively, is able to model the XAS spectrum of $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ (black circles) convincingly. Hence the Rh valence in our $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ single crystal amounts to $\sim 3.475+$, which is close to the nominal value of 3.5+ in this compound. Note that the deviation from its nominal value is of the order of the estimated error bar of $\sim 2\%$ in these kind of measurements.

C. Magnetic properties

The magnetic susceptibility $\chi(T)$ measured in an external field of $\mathbf{H} = 5000$ Oe is shown in Fig. 7. $\chi_{[110]}$ and $\chi_{[001]}$ are the magnetic susceptibilities measured with $\mathbf{H} \parallel [110]$ and $\mathbf{H} \parallel [001]$ directions, respectively. We observe a moderate anisotropy of $\chi_{[110]}/\chi_{[001]} = 2 \sim 6$ over the whole measured temperature range. No hysteresis between zero-field cooled (ZFC) and field-cooled (FC) curves can be observed. The solid line in Fig. 7 is a fit of $\chi(T)$ containing a Curie-Weiss and a

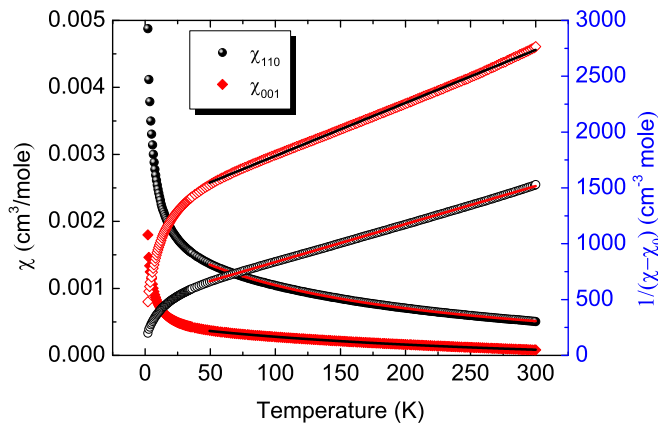


FIG. 7. Temperature dependence of the magnetic susceptibilities $\chi(T)$ (filled symbols) of $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ measured with an external field of $\mathbf{H} = 5000$ Oe applied along the crystallographic $[110]$ and $[001]$ directions. Inverse magnetic susceptibilities $1/(\chi - \chi_0)$ (open symbols) are shown to the right axis. Solid lines between 50 to 300 K are Curie-Weiss-like fits to the experimental data.

temperature independent term χ_0 :

$$\chi(T) = \chi_0 + \frac{C}{T - \theta} \quad (1)$$

with the Curie constant $C = N\mu_{\text{eff}}^2/(3k_B)$, Weiss temperature θ , number of spins N , Boltzmann constant k_B , and the effective magnetic moment per formula unit μ_{eff} . The resulting fitting parameters for $\chi_{[110]}(T)$ amount to $C_{[110]} = 0.295(3) \text{ cm}^3 \text{ K/mole}$, $\chi_{0,[110]} = -1.48(5) \times 10^{-4} \text{ cm}^3/\text{mole}$, and $\theta_{[110]} = -147(1) \text{ K}$. For $\chi_{[001]}(T)$, these values amount to $C_{[001]} = 0.211(5) \text{ cm}^3 \text{ K/mole}$, $\chi_{0,[001]} = -2.81(6) \times 10^{-4} \text{ cm}^3/\text{mole}$, and $\theta_{[001]} = -277(5) \text{ K}$. The effective magnetic moments calculated from these parameters are $\mu_{\text{eff},[110]} = 1.54(1) \mu_B$ and $\mu_{\text{eff},[001]} = 1.30(2) \mu_B$ for $[110]$ -direction and $[001]$ -direction, respectively. The determined effective magnetic moments here are slightly smaller than that reported earlier from polycrystalline samples, $\mu_{\text{eff}} = 1.738 \mu_B$ [9]. In $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$, only the Rh ions with oxidation state 3.5+ contribute to the magnetism. Due to the localized nature of $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ (see our resistivity measurements below), it is reasonable to assume a statistical distribution of Rh^{4+} ($S = 1/2$) and Rh^{3+} ($S = 0$) ions, thus, yielding $\langle S(S+1) \rangle = 3/8$. Assuming a g factor of 2, the effective magnetic moment amounts to $\mu_{\text{eff,calc.}} = 1.225 \mu_B$. This calculated value is somewhat smaller than the above two experimentally obtained values from our Curie-Weiss fit. This discrepancy is either indicative for a thermal population of an intermediate or high spin (HS) state of the Rh ions [16] or for a somewhat larger g factor [16,24].

D. Resistivity

The in-plane and out-of-plane resistivity $\rho_{[110]}(T)$ and $\rho_{[001]}(T)$ of $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ was measured with an applied electrical current along the crystallographic $[110]$ and $[001]$ directions respectively, see Fig. 8(a). The room temperature values of the electrical resistivity amount to $\rho_{[110]}(300 \text{ K}) = 0.75 \times 10^{-2} \Omega \text{ cm}$ and $\rho_{[001]}(300 \text{ K}) = 0.44 \Omega \text{ cm}$. With decreasing temperature, the resistivity measured in both directions increases to values of five orders of magnitude larger. This reveals the insulating nature of $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$. We fitted the experimental data between 50 to 300 K with the two dimensional Mott variable-range hopping model [25], $\rho(T) = \rho_0 \exp[(T_0/T)^{1/(d+1)}]$, where $d = 2$ for a two dimensional layered system, see Fig. 8(a) (solid line). This is different to studies on polycrystalline samples in literature where plural types of conduction mechanisms were observed, i.e., neither a simple activated mechanism nor a variable-range hopping mechanism was reported for $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ [9]. The resulting parameters from our fits to $\rho_{[110]}(T)$ and $\rho_{[001]}(T)$ amount to $\rho_{0,[110]} = 9.9(2) \times 10^{-4} \Omega \text{ cm}$, $T_{0,[110]} = 2098(37) \text{ K}$ and $\rho_{0,[001]} = 3.88(9) \times 10^{-2} \Omega \text{ cm}$, $T_{0,[001]} = 4078(58) \text{ K}$, respectively. For clarity, we plot the resistivity data as a function of $T^{-1/3}$ in Fig. 8(b). Solid lines were calculated using the fitting parameters above. Only at low temperatures, $\sim T \leq 15 \text{ K}$, the data points start to deviate from the theoretical model, thus, suggesting that electron-electron interactions and/or magnetic scattering start to play some role at low temperatures.

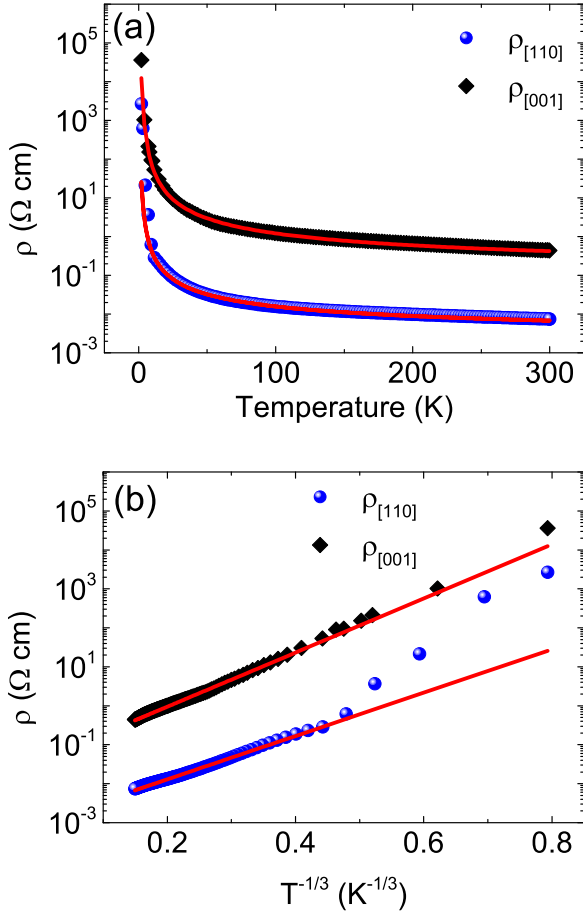


FIG. 8. (a) Resistivity as a function of temperature and (b) resistivity vs $T^{-1/3}$ on a logarithmic scale for $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ measured with the electrical current flow along the crystallographic [110]/[001] directions as indicated in the legends. Solid lines represent fits to experimental data points with the two dimensional Mott variable-range hopping theory in the temperature range between 50 to 300 K (see text).

E. Specific heat

In Fig. 9, the zero-field specific heat $C_P(T)$ of our $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ single crystals is shown. No obvious anomaly can be observed in the entire temperature range, which is in agreement with our magnetization and resistivity measurements. Our measured values of C_P at high temperatures, e.g., $C_P(267 \text{ K}) = 147 \text{ J mole}^{-1} \text{ K}^{-1}$, are significantly lower than the classical Dulong-Petit prediction of the lattice specific heat $C_V = 3nR = 175 \text{ J mole}^{-1} \text{ K}^{-1}$, where n is the number of atoms per formula unit and R is the molar gas constant. This suggests that $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ has a Debye temperature Θ_D distinctly higher than 267 K, which is consistent with published data for the undoped parent compound Sr_2RhO_4 ($\Theta_D(\text{Sr}_2\text{RhO}_4) = 421 \text{ K}$) [11]. As shown in the inset of Fig. 9, $C_P(T)/T$ versus T^2 exhibits a linear region at low temperatures (between 10 to 20 K). Therefore we fitted the data in this range by the formula $C_P(T)/T = A + \beta T^2$, where the weak upturn below $\sim 10 \text{ K}$ and the small but non-vanishing first term $A = 1.0(1) \times 10^{-2} \text{ J mole}^{-1} \text{ K}^{-1}$ could arise from the slowing down of critical spin fluctuations. The second term corresponds to the low-temperature limit

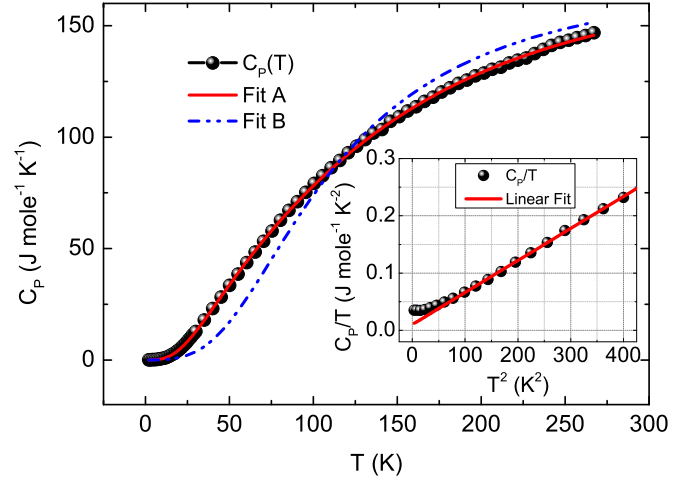


FIG. 9. Specific heat C_P of $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ single crystal as a function of temperature measured in zero magnetic field. The red solid and blue dash-dotted curves are fits to the experimental data using the Debye lattice specific heat model with temperature-dependent and constant Debye temperatures, respectively (see text). Inset shows C_P/T vs T^2 in the low-temperature range, $T \leq 20 \text{ K}$. The straight red line represents a linear fit to the data between 10 to 20 K.

of the Debye lattice specific heat with fitted value of $\beta = 5.59(5) \times 10^{-4} \text{ J mole}^{-1} \text{ K}^{-4}$. From this value of β , we can estimate the Debye temperature using the expression $\Theta_D = (12\pi^4 nR/5\beta)^{1/3} = 289.8(9) \text{ K}$. This value is much smaller than one would expect from the high-temperature limit data, suggesting that the system temperature has a non-negligible effect on the Debye temperature of $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$. To quantify this effect, we tried to fit the entire data set using the Debye lattice specific heat model [26]:

$$C_V(T) = 9R \left(\frac{T}{\Theta_D} \right)^3 \int_0^{\Theta_D/T} \frac{x^4 e^x}{(e^x - 1)^2} dx. \quad (2)$$

In the fit, we used the analytic Padé approximate function, given in Ref. [27], which accurately represents the above integral. As shown by the blue dash-dotted line [$\Theta_D = 456(6) \text{ K}$] in Fig. 9, a reasonable fit can not be obtained with a constant Debye temperature. In order to estimate the temperature dependence of the $\Theta_D(T)$, we assume that the system exhibits a ground state with Θ_{D0} at base temperature. With increasing temperature, the system is thermally activated to a state separated by ΔE with probability of $e^{-\frac{\Delta E}{k_B T}}$ and we further assume that this activated state has a simple form of Debye temperature $\Theta_{D1} = \Theta_{D0} + \text{constant}$. Hence it follows that

$$\Theta_D(T) = \Theta_{D0} \left(1 + A e^{-\frac{\Delta E}{k_B T}} \right). \quad (3)$$

As shown by the red solid curve in Fig. 9, an excellent fit can be obtained by our simple two-states model with parameters of $\Theta_{D0} = 291(4) \text{ K}$, $\Delta E = 4.9(2) \text{ meV}$ and $A = 1.14(2)$. The origin of this behavior is subject of future studies.

III. CONCLUSIONS

In summary, we have grown centimeter-sized single crystals of $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ by the FFFZ method. We have

characterized our crystals by x-ray, neutron, susceptibility, electrical transport, and specific heat measurements. We find indications either for an enhanced in-plane effective magnetic moment or for an anisotropic g factor and we observe a temperature-dependent Debye temperature for our $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ single crystal. Moreover, we have strong indications that the crystal structure of $\text{Sr}_{1.5}\text{La}_{0.5}\text{RhO}_4$ can not be described with space group $I4/mmm$ as reported in literature. After comparing different models, the best

description of the crystal structure can be achieved with space group $P2_1/c$. The symmetry lowering arises from both, RhO_3 octahedral rotations of $\sim 8.2^\circ$ around the former tetragonal c axis and octahedral tilts of slightly less than 2° .

ACKNOWLEDGMENT

We thank L. H. Tjeng for helpful discussions.

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- [1] G. Senatore and N. H. March, *Rev. Mod. Phys.* **66**, 445 (1994).
- [2] J. G. Bednorz and K. A. Müller, *Z. Phys. B* **64**, 189 (1986).
- [3] H. Krauth, K. Schlenga, D. Nardelli, I. Pallecchi, M. Tropeano, R. Hott, T. Wolf, and M. Putti, in *Superconducting Materials, Applied Superconductivity: Handbook on Devices and Applications*, edited by P. Seidel (Wiley-VCH Verlag GmbH & Co. KGaA: Weinheim, Germany, 2015), pp. 105–191.
- [4] R. E. Cohen, *Nature* **358**, 136 (1992).
- [5] Y. K. Liu, Y. W. Yin, and X. G. Li, *Chin. Phys. B* **22**, 087502 (2013).
- [6] C. N. R. Rao, *J. Phys. Chem. B* **104**, 5877 (2000).
- [7] A. P. Mackenzie and Y. Maeno, *Rev. Mod. Phys.* **75**, 657 (2003).
- [8] Z. W. Li, C. F. Liu, M. Skoulatos, L. H. Tjeng, and A. C. Komarek, *J. Cryst. Growth* **427**, 94 (2015).
- [9] T. Shimura, M. Itoh, Y. Inaguma, and T. Nakamura, *Phys. Rev. B* **49**, 5591 (1994).
- [10] R. S. Perry *et al.*, Sr_2RhO_4 : A new, clean correlated electron metal, *New J. Phys.* **8**, 175 (2006).
- [11] I. Nagai, N. Shirakawa, N. Umeyama, and S. Ikeda, *J. Phys. Soc. Jpn.* **79**, 114719 (2010).
- [12] Y. Drees, D. Lamago, A. Piovano, and A. C. Komarek, *Nat. Commun.* **4**, 2449 (2013).
- [13] Y. Drees, Z. W. Li, A. Ricci, M. Rotter, W. Schmidt, D. Lamago, O. Sobolev, U. Rütt, O. Gutowski, M. Sprung, A. Piovano, J. P. Castellán, and A. C. Komarek, *Nat. Commun.* **5**, 5731 (2014).
- [14] H. Guo, W. Schmidt, L. H. Tjeng, and A. C. Komarek, *Phys. Status Solidi (RRL)* **9**, 580 (2015).
- [15] Z. W. Li, Y. Drees, A. Ricci, D. Lamago, A. Piovano, M. Rotter, W. Schmidt, O. Sobolev, U. Rütt, O. Gutowski, M. Sprung, J. P. Castellán, L. H. Tjeng, and A. C. Komarek, *J. Supercond. Nov. Magn.* **29**, 727 (2016).
- [16] N. Furuta, S. Asai, T. Igarashi, R. Okazaki, Y. Yasui, I. Terasaki, M. Ikeda, T. Fujita, M. Hagiwara, K. Kobayashi, R. Kumai, H. Nakao, and Y. Murakami, *Phys. Rev. B* **90**, 144402 (2014).
- [17] J. Rodríguez-Carvajal, *Phys. B* **192**, 55 (1993).
- [18] Y. Su, K. Nemkovskiy, and S. Demirdis, *J. Large-Scale Res. Facil.* **A27**, 1 (2015); W. Schweika and P. Böni, *Physica B (Amsterdam)* **297**, 155 (2001).
- [19] V. Petříček, M. Dušek, and L. Palatinus, *Z. Kristallogr.* **229**, 345 (2014).
- [20] M. Itoh, T. Shimura, Y. Inaguma, and Y. Morii, *J. Solid State Chem.* **118**, 206 (1995).
- [21] T. Vogt and D. J. Buttrey, *J. Solid State Chem.* **123**, 186 (1996).
- [22] B. J. Kim, J. Yu, H. Koh, I. Nagai, S. I. Ikeda, S.-J. Oh, and C. Kim, *Phys. Rev. Lett.* **97**, 106401 (2006).
- [23] Z. Hu, M. S. Golden, S. G. Ebbinghaus, M. Knupfer, J. Fink, F. M. F. de Groot, and G. Kaindl, *Chem. Phys.* **282**, 451 (2002).
- [24] M. Hagiwara, S. Kimura, N. Nishihagi, T. Suzuki, M. Nohara, H. Takagi, and K. Kindo, *J. Low Temp. Phys.* **159**, 11 (2009).
- [25] N. F. Mott, *Adv. Phys.* **21**, 785 (1972).
- [26] C. Kittel, *Phonons II. Thermal Properties*, in *Introduction to Solid State Physics*, edited by S. Johnson (Wiley, New York, USA, 2015), pp. 107–117.
- [27] R. J. Goetsch, V. K. Anand, A. Pandey, and D. C. Johnston, *Phys. Rev. B* **85**, 054517 (2012).