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NaV₂O₄: A Quasi-1D Metallic Antiferromagnet with Half-Metallic Chains

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 NaV_2O_4 crystals were grown under high pressure using a NaCl flux, and the crystals were characterized with x-ray diffraction, electrical resistivity, heat capacity, and magnetization. The structure of NaV_2O_4 consists of double chains of edge-sharing VO_6 octahedra. The resistivity is highly anisotropic, with the resistivity perpendicular to the chains more than 20 times greater than that parallel to the chains. Magnetically, the intrachain interactions are ferromagnetic and the interchain interactions are antiferromagnetic; 3D antiferromagnetic order is established at 140 K. First-principles electronic structure calculations indicate that the chains are half-metallic. Interestingly, the case of NaV_2O_4 seems to be a quasi-1D analogue of what was found for half-metallic materials.

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Quasi-1D metals have intrigued condensed matter physicists for decades, and they still continue to fascinate both due to their propensity toward Fermi surface instabilities such as density waves and to their ability to form Luttinger liquid and other non-Fermi-liquid ground states. Efforts to elucidate quasi-1D phenomena are an important topic of ongoing experimental and theoretical research.

Recent examples of novel quasi-1D behavior in vanadates include β -Na_{0.33}V₂O₅, which was found to undergo a superconducting transition (believed to be non-BCS) by applying pressure, and $BaVS_3$, which showed Luttinger liquid behavior in its photoemission spectra [1]. With these examples in mind, we paid particular attention to the $3d^{1}$ system and related compounds to search for a novel electronic state. Specifically, we chose the electrically insulating compound CaV2O4 comprising double chains of edgesharing VO₆ octahedra (known as the CaFe₂O₄ structure [2]) as a subject, and we conducted aliovalent substitution of Ca^{2+} by Na^{1+} to increase the population of the $3d^1$ state. Here we report the initial synthesis of NaV₂O₄ using a high-pressure method. X-ray diffraction confirmed that this mixed-valent $V^{3.5+}$ (3 $d^{1.5}$) compound is indeed isostructural to CaV₂O₄. Its quasi-1D metallic nature and an antiferromagnetic (AF) transition ($T_N \sim 140$ K) were confirmed by magnetic and transport measurements. Intriguingly, the compound maintains a quasi-1D metallic state down to 40 mK, and the metallicity coexists with magnetic order. This is surprising because in the presence of magnetic interactions a quasi-1D metal is generally unstable against the formation of a spin density wave. Moreover, a small amount of disorder can localize the charge carriers, transforming a one-dimensional metal into an insulator [3]. Even though the electronic system of NaV_2O_4 is not sufficiently anisotropic enough to warrant the occurrence of quasi-1D instabilities, as far as we know there are very few materials exhibiting metallic nature below a well-defined PACS numbers: 72.80.Ga, 71.27.+a, 73.63.Nm

AF transition with an anisotropic crystal structure. The metallic ground state of NaV_2O_4 is therefore quite surprising.

Single crystals of NaV₂O₄ were grown by a NaCl flux technique in a high-pressure furnace that was able to maintain 6 GPa throughout the crystal growth [4]. The crystals grew as shiny black needles or platelets as shown in Fig. 1(a). Single crystal diffraction data were obtained on a Bruker SMART APEX (Mo $K_{\alpha} \lambda = 0.71069$ Å) diffractometer and were analyzed using the SHELXL-97 analysis software [5]. The *R* factors in the final refinement, 2.24% (R_p) and 5.83% (R_{wp}), indicate good convergence. The study clearly confirmed the CaFe₂O₄ structure for the crystal [2]. A brief summary of the structural parameters appears in Table I, and Figs. 1(b) and 1(c) show, respectively, a schematic view of the crystal structure and an illustration of the VO₆ double chains that run along the *b* axis.



FIG. 1 (color online). (a) The crystal of NaV_2O_4 (1.779 mg) used for magnetization measurements. (b) Schematic structural view of NaV_2O_4 , drawn from the x-ray result and (c) a local structural view of the double chain. The thick lines indicate the conventional orthorhombic unit cell, and the red, blue, purple circles indicate V, O, Na atoms, respectively.

Magnetization measurements were performed on the 1.779 mg crystal shown in Fig. 1(a) using a Quantum Design Magnetic Properties Measurement System between 1.8 K and 350 K and in an applied field of 50 kOe. Electrical resistivity measurements were performed both parallel and perpendicular to *b* from 2 K to 300 K using a Quantum Design Physical Properties Measurement System and from 40 mK to 4 K using a ³He/⁴He dilution refrigerator. The crystals were large enough for 4-wire measurements to be used in all cases. 20–30 μ m Pt wires were affixed to the samples using silver epoxy. Voltage leads were separated by 0.2–0.5 mm for measurements parallel to *b* and 0.1–0.2 mm for measurements were performed in fields to 90 kOe at temperatures of 50 K, 150 K, and 200 K.

Specific heat measurements were performed on a polycrystalline sample [4] by a relaxation technique between 2 K and 160 K, and with and without an applied magnetic field of 70 kOe. Additional data were collected on a collection of 10 crystals with Ag paint thermal connections between them in a dilution refrigerator between 70 mK and 3 K. Seebeck coefficient measurements were performed in a Quantum Design Physical Properties Measurement System using constantan as a reference material.

Since the vanadium ions have a nonintegral valence $(V^{+3.5}; 3d^{1.5})$, we carefully examined the crystallography data for each vanadium site (see Table I) to investigate possible nonuniformity of charge distribution. The two site surroundings, however, show no significant differences. For example, the average distance to the ligand O was 1.97(1) Å and 1.98(2) Å, and the bond valence sum was +3.4 and +3.3 for the V1 site and the V2 site, respectively [6]. Moreover, the distortion factor (defined as the ratio of the longest to the shortest distance) of each of the VO₆ octahedron was 1.020 and 1.024 for V1 and V2, respectively, being much lower than the factors of other isostructural materials [7]. It should be stressed here that the t_{2g} electronic system shows a clear AF ordering (shown later); however, the x-ray results are not suggestive of expected orbital ordering or crystal-field splitting [8]. Low temperature x-ray study might provide further aspects to the issue.

Magnetic susceptibility (χ) of NaV₂O₄ is plotted in Fig. 2(a), and inverse susceptibility $(1/l\chi)$ is plotted in Fig. 2(b). A clear transition, consistent with antiferromag-

netism, is evident near 140 K. To parametrize the data, a Curie-Weiss fit was performed at high temperatures (>250 K). Fits were performed with and without a temperature-independent paramagnetic term, χ_0 . For $\chi_0 =$ 0, the best fit yielded $1.99 \mu_B/V$ for the effective moment per vanadium ion, and +118(3) K for the Weiss temperature, Θ_W , suggesting predominantly ferromagnetic (FM) interactions. Interestingly, the fitted moment of $1.99 \mu_B/V$ is close to that expected for a purely localized model, $2.35\mu_B/V$ [9]. For $\chi_0 \neq 0$, the fit yielded $2.70\mu_B/V$ and +55(8) K at $\chi_0 = -9.70 \times 10^{-4}$ emu/mole of V. It should be kept in mind, however, that the applicability of the Curie-Weiss law to an itinerant system is not theoretically well justified, although in practice often yields useful results. Although the positive Θ_W from the Curie-Weiss fits suggests FM interactions, but the shape of the transition is clearly antiferromagnetic. Furthermore, isothermal magnetization data [Figs. 2(c) and 2(d)] did not show any trace of FM behavior such as a spontaneous magnetization or magnetic hysteresis. We therefore tentatively conclude that strong FM intrachain interactions coexist with weaker AF interchain interactions, resulting in 3D AF order near 140 K.

The behavior of the susceptibility below the AF transition indicates that the ordered moments are collinear with the *b* axis, as the susceptibility with the field applied parallel to *b* drops sharply below the transition. Further checks were made to confirm this conclusion. For example, the $H \perp b$ data for the crystal rotated 90 degrees along the *b* axis (not shown) were found equivalent to the $H \perp b$ data obtained without the rotation. Moreover, the *M* versus *H* data in Figs. 2(c) and 2(d) show magnetic anisotropy below the ordering temperature: the curves at 150 K and 300 K did not show orientation dependence, while the 10 K curves did. All these observations are consistent with the moments lying along the chains (*b* axis).

The temperature dependence of the electrical resistivity of 4 different crystals is plotted in Fig. 3(a). The reproducibility of the data from sample to sample is quite good. The resistivity was measured along the chains $(i \parallel b)$ and transverse to the chains $(i \perp b)$ from 2–300 K. Measured along the chains, the resistivity is metallic with a noticeable downturn below the 140 K magnetic transition. The $i \parallel b$ resistance at 40 mK was nearly the same as it was at 4 K,

TABLE I. Atomic coordinates and anisotropic displacement parameters of NaV₂O₄ at 293(2) K. Space group: *Pnma*, a = 9.1304(11) Å, b = 2.8844(4) Å, c = 10.6284(13) Å, z = 4, V = 279.91(6) Å³. The parameters y, U_{12} , U_{23} are 0.25, 0, and 0, respectively, for every atom.

Atom	x	Z	$100U_{11}$	$100U_{22}$	$100U_{33}$	$100U_{13}$
Na	0.24271(13)	0.34620(10)	1.28(6)	1.35(6)	1.22(6)	0.02(4)
V1	0.08309(5)	0.60374(4)	0.94(2)	0.81(3)	0.58(3)	0.013(15)
V2	0.06330(5)	0.11198(4)	0.93(2)	0.86(3)	0.59(3)	-0.013(15)
01	0.2910(2)	0.64828(16)	1.09(9)	0.97(9)	0.90(9)	0.05(6)
O2	0.3872(2)	-0.02053(17)	0.97(9)	0.97(9)	0.74(9)	-0.05(7)
03	0.3872(2)	0.21757(18)	1.09(9)	1.05(9)	0.74(8)	0.04(7)
O4	0.0785(2)	-0.07120(18)	1.08(9)	0.91(9)	0.67(8)	0.03(6)



FIG. 2. (a), (b) Temperature dependence of the magnetic susceptibility χ and $1/\chi$; (c), (d) isothermal magnetization of a crystal of NaV₂O₄ [see Fig. 1(a)]. Open and closed symbols in the susceptibility curves represent zero-field cooling and field cooling data, respectively.

suggesting no further phase transitions (including superconductivity) down to that temperature. In contrast, the $i \perp b$ results are considerably less metallic with a broad hump near 140 K, constituting a remarkably anisotropic feature. No thermal hysteresis was observed across the transition, suggesting that the transition is continuous rather than first order.

The inset of Fig. 3(a) shows transverse MR data obtained with current applied parallel to the chains in magnetic fields to ± 90 kOe. The MR ratio was calculated using the formula MR = $\rho/\rho_0 - 1$, where ρ_0 stands for the zero-field resistivity. Above the magnetic transition temperature, NaV₂O₄ shows a small degree of negative MR (<0.01) that is consistent with normal metallic behavior. Below the magnetic transition, the sign of the MR changes and its magnitude increases, indicating that the energy dependence of the scattering or the electronic structure changes at the magnetic transition.

Figures 3(b) and 3(c) show the specific heat and thermopower of NaV₂O₄, respectively. Both quantities clearly show an anomaly at the 140 K magnetic ordering temperature. A Sommerfeld coefficient $\gamma =$ 6.91(4) mJ mol V⁻ K⁻² and Debye temperature $\Theta_D =$ 498.0(8) K were inferred from a least squares fit to C_p/T versus T^2 as shown in the inset to Fig. 3(b). The experimental γ is comparable to the calculated result $(\sim 4 \text{ mJ mol V}^{-1} \text{ K}^{-2})$ for the FM solution and $\sim 10 \text{ mJ mol V}^{-1} \text{ K}^{-2}$ for the nonmagnetic solution, see below), indicating that heavy quasiparticles are not being formed in NaV_2O_4 unlike the related compound LiV_2O_4 [10]. This is not surprising, as the magnetic entropy is eliminated via magnetic ordering rather than being transferred to the Fermi liquid. Below ~ 1 K, a strong upturn of specific heat was observed (not shown), which was most likely a nuclear Schottky contribution, hyperfine enhanced due to the magnetic ordering at higher temperature.



FIG. 3. (a) Temperature dependence of electrical resistivity, (b) specific heat, and (c) thermopower of NaV₂O₄. [Inset, (a)] MR ratios of ρ_{\parallel} at $H \perp b$, and [inset, (b)] a C_p/T vs T^2 fit of the low temperature specific heat data used to extract the Sommerfeld coefficient and Debye temperature.

Quantitative analysis of the magnetic entropy was accomplished by first subtracting the lattice specific heat of $CaSc_2O_4$ [11], an isostructural, nonmagnetic analogue, from the zero-field specific heat of NaV_2O_4 . After subtraction, the integrated entropy associated with the magnetic transition was ~6 J mol⁻¹ K⁻¹; this corresponds to only 40% of $R(\ln 3 + \ln 2)$, expected if whole spins of V³⁺ $(3d^2, S = 1)$ and V⁴⁺ $(3d^1, S = 1/2)$ are localized and ordered at low temperature. It is likely that much of the entropy is removed via short-range ordering above the magnetic transition temperature.

The electronic structure of NaV_2O_4 was studied using the local spin density approximation of density functional theory [12]. We used the WIEN2K package [13], which is based on the full-potential augmented-plane-wave method. Experimental lattice parameters and atomic coordinates



FIG. 4. (a) Nonmagnetic and (b) the ferromagnetic DOS of NaV_2O_4 .

were used for the calculations. The atomic radii were chosen as 2.1, 1.9, and 1.7 a.u. for Na, V, and O, respectively. The cutoff wave number K for wave functions in interstitial region was set to RK = 7.5, where R is the smallest atomic sphere radius. The integration over Brillouin zone was performed by a tetrahedron method with 132 k points in the irreducible Brillouin zone. The nonmagnetic density of states (DOS) is plotted in Fig. 4(a). Note that V d bands (-0.5 to 4 eV) are sharply separated into e_g bands (1.8 to 4 eV) and t_{2g} bands (-0.5 to 1.5 eV), reflecting the small degree of the VO₆ distortion. This result suggests that the t_{2g} orbitals determine the low energy properties of NaV₂O₄. Moreover, because the t_{2g} electrons remain highly itinerant along the chain direction, neither a spin-Peierls transition nor a spin density wave is likely to explain the magnetic transition at ~ 140 K.

The calculated Fermi energy (E_F) lies near a major peak in the DOS, which suggests a magnetic instability. Indeed, we found a stable FM DOS solution [shown in Fig. 4(b)], which gained ~ 0.15 eV per V from the nonmagnetic total energy. The exchange splitting lifts the minority t_{2g} bands above E_F and empties the bands completely, resulting in a half-metallic (HM) state. The estimated magnetic moment is $3.0\mu_B$ per primitive cell owing to the integer filling of the minority bands, suggesting the whole t_{2g} electrons, $d^1 + d^2$, fully contribute to the magnetism. The estimation is likely consistent with the result from the susceptibility study; however, further studies of a role of spin fluctuations is needed to clarify nature of the magnetism. In addition, an antiparallel order state was tested as well; however, a stable DOS solution was not obtained. These results suggest strong instability toward a FM ordering in the present system, being consistent with the metallic transport property observed at low temperature. However, in contrast to the band structure calculation, bulk ferromagnetism was not experimentally observed. The facts thus suggest that the interchain magnetic coupling is rather AF while the intrachain coupling is FM, preventing the appearance of net magnetic moment.

The most realistic magnetic model of NaV₂O₄ that accounts for the results presented here is that the integer magnetic moments order ferromagnetically along the chains, with the chains coupled antiferromagnetically. The spatial distribution of the spin current in that case would be a 1D analog of what has been predicted by a first-principles calculation in the layered compound Na_xCoO₂ ($x \ge 0.75$) in a condition, in which HM layers are coupled antiferromagnetically [14]. In the HM chain model, the resistivity kink at ~140 K would be due to a Fisher-Langer singularity [15]. Transport perpendicular to the chains is hindered because there are no available states for the electrons to hop into, and thus the anisotropy and the nonmetallic character of the resistivity observed in the $i \perp b$ measurements can be readily understood.

In summary, we have synthesized and characterized the basic physical properties of a new mixed-valent vanadium oxide, NaV₂O₄, which shows remarkable magnetic and electrical properties. The degree of the 1D anisotropy (ρ/ρ_{\parallel}) is >20 at low temperature, which is large for a transition metal oxide (c.f. ~6 in the double rutile-chains compound BaRu₆O₁₂ [16]). First-principles calculations indicate that HM ferromagnetism is found on the chains, whereas the behavior of the magnetization is consistent with AF order, with moments collinear with the chains.

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