## Ferroelectricity Driven by Y $d^0$ -ness with Rehybridization in YMnO<sub>3</sub>

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(Received 8 November 2006; published 21 May 2007)

We investigated electronic structure of hexagonal multiferroic YMnO<sub>3</sub> using the polarization dependent x-ray absorption spectroscopy (XAS) at O K and Mn  $L_{2,3}$  edges. The spectra exhibit strong polarization dependence at both edges, reflecting anisotropic Mn 3d orbital occupation. Moreover, the O K edge spectra show that Y 4d states are strongly hybridized with O 2p ones, resulting in large anomalies in Born effective charges on off-centering Y and O ions. These results manifest that the Y  $d^0$ -ness with rehybridization is the driving force for the ferroelectricity, and suggest a new approach to understand the multiferroicity in the hexagonal manganites.

DOI: 10.1103/PhysRevLett.98.217601

PACS numbers: 77.84.-s, 71.70.Ch, 75.50.Ee, 78.70.Dm

Recently, multiferroicity, in which magnetism and ferroelectricity coexist, takes much attention due to its exotic magnetoelecric phenomena [1–3], which enable us to control magnetism through an electric field or electricity through a magnetic field and may open up a new concept of potential technological applications. Especially,  $RMnO_3$ , which is crystallized either in hexagonal structure for R = Ho, Er, Tm, Yb, Lu, or Y, Sc with smaller ionic radius or in the orthorhombic one for R = La, Ce, Pr, Nd, Eu, Gd, Tb, Dy, or Bi with larger ionic radius [4,5], exhibits different types of multiferroicity [3,4].

Hexagonal RMnO<sub>3</sub> exhibits multiferroicity with high ferroelectric and low antiferromagnetic transition temperature ( $T_E > 600$  K,  $T_M \sim 90$  K). The hexagonal structure  $(P6_3 cm)$  brings out soft mode phonons required for ferroelectricity [6], but its driving force has been puzzling. A prototype ferroelectricity was observed in the so-called " $d^0$ -ness" systems such as BaTiO<sub>3</sub>, in which Ti<sup>4+</sup>( $d^0$ ) ions make off-center movements in TiO<sub>6</sub> octahedra to lower the energy through enhanced Ti 3d–O 2p hybridization (rehybridization) [7,8]. RMnO<sub>3</sub>, however, contains  $Mn^{3+}$  ( $d^4$ ) magnetic ions, and the ferroelectricity demands a mechanism other than the  $d^0$ -ness, such as lone pair electrons in BiMnO<sub>3</sub> or a spin frustration of magnetic order in  $TbMnO_3$  [3]. However, neither the lone pair nor the spin frustration could hold for the ferroelectricity of hexagonal RMnO<sub>3</sub>. The system contains bipyramid MnO<sub>5</sub>, two ontop oxygens  $(O_T)$  and three in-plane oxygens  $(O_P)$ , and its unusual electronic structure has been suspected to be responsible for the ferroelectricity [9,10].

Under the MnO<sub>5</sub>  $D_{3h}$  site symmetry, 3*d* orbital is split into two doublets,  $e_{1g}$  (yz/zx) and  $e_{2g}$  ( $xy/x^2 - y^2$ ), and one singlet,  $a_{1g}$  ( $z^2$ ). The Mn<sup>3+</sup>  $d^4$  ground state configuration becomes  $e_{1g}^2 e_{2g}^2$ , and Mn 3d states have strong anisotropic bonding with O 2p ones along the hexagonal c axis, which is the ferroelectric direction. Then "directional" Mn  $d^0$ -ness was suggested as the origin of the ferroelectricity [9]. However, actual polarization was found to be induced from off-center movements of  $Y-O_P$ sites rather than  $Mn-O_T$  sites [11]. Meanwhile, van Aken et al. claimed based on local density band calculations that  $Y-O_P$  off-center movements are stabilized entirely due to electrostatic energy since Y-O bonds are almost ionic and so do not contribute large anomalies in Born effective charges [12], which are directly related to the polarization, on the acting Y- $O_P$  sites [13]. However, the predicted site movements are considerably different from the actual ones, and the estimated stabilization energy is much smaller than  $T_E \sim 1000$  K. The differences could be reduced by optimizing the lattice coordinates in an LSDA + U (local spin density approximation plus Hubbard U) method [10,14], but the predicted electronic structure is far away from the experimental one.

In this Letter, we scrutinized the electronic structure and Mn-O and Y-O bond characteristics in YMnO<sub>3</sub> using the polarization dependent x-ray absorption spectroscopy (XAS) at O K edge, which reflects the unoccupied conduction band. The strict selection rules of the polarization dependent XAS enable us to confidently identify the hybridized Mn 3d and Y 4d states. The detailed spectral features were found to be away from the band calculation results but well described in terms of the cluster model calculations for MnO<sub>5</sub> and YO<sub>8</sub>. The analysis for the Mn 3d region in the O K edge XAS, which shows strong anisotropic hybridization, describes consistently the Mn  $L_{2,3}$  edge polarization dependent XAS results. Interestingly,

the Y 4*d* region in the O *K* edge XAS also reveals strong anisotropic Y 4*d*-O 2*p* hybridization, especially along the *c*-axis polarization direction. The strong Y 4*d*-O 2*p* hybridization contributes large anomalies in Born effective charges on the off-centering Y-O<sub>*P*</sub> sites. Further, the XAS spectra at O *K*- and Mn  $L_{2,3}$  edges of hexagonal DyMnO<sub>3</sub>/Pt(111) and orthorhombic DyMnO<sub>3</sub>/SrTiO<sub>3</sub>(001) show very similar polarization dependence to those of hexagonal YMnO<sub>3</sub> and orthorhombic LaMnO<sub>3</sub>, respectively. These results manifest that in hexagonal *R*MnO<sub>3</sub>, the *R d*<sup>0</sup>-ness with rehybridization is commonly responsible for the ferroelectricity.

A high quality YMnO<sub>3</sub> single crystal was grown by using the floating zone method [15], and oriented through the Laue diffraction. As shown in the inset of Fig. 1(a), Mn<sup>3+</sup> and Y<sup>3+</sup> ions have MnO<sub>5</sub> and YO<sub>8</sub> local structures with bipyramidal  $D_{3h}$  and trigonal  $D_{3d}$  site symmetries, respectively. A clean YMnO3 crystal was prepared to expose a hexagonal  $[0001] \times [10\overline{1}0]$  plane, in which the c axis is in the plane, to the incoming light, and was introduced into an experimental chamber under  $4 \times$  $10^{-10}$  Torr. The measurements were performed at the 2A elliptically polarized undulator beam line in the Pohang light source (PLS). Taking advantage of controllability of photon polarization, we could obtain the polarization dependent XAS spectra without changing any experimental geometry. The sample temperature was kept at room temperature, and the energy resolution was set to be about



FIG. 1 (color online). (a) O K edge polarization dependent XAS spectra of hexagonal YMnO<sub>3</sub> collected in FY mode in comparison with the CI model calculations and the band calculations taken from Ref. [13] for O 2p PDOS. The crystal structure is shown in the inset. The solid bars represent the intensities of the delta functions. (b) Schematic crystal field splittings for MnO<sub>5</sub> ( $D_{3h}$ ) and YO<sub>8</sub> ( $D_{3d}$ ).

0.15 eV. The O K edge spectra were simultaneously collected in both total electron yield (TEY) and fluorescence yield (FY) modes, confirming that surface degradation effects were negligible.

Figure 1(a) shows the O K edge polarization dependent XAS spectra ( $E \parallel c$  and  $E \parallel ab$ ) of YMnO<sub>3</sub>. The spectra, which directly show transferred O 2p partial density of states (PDOS) due to hybridization to the unoccupied conduction bands, are roughly divided into Mn 3d, Y 4d, and Mn 4sp/Y 5sp regions. The spectrum exhibits strong polarization dependence in the Mn 3d and Y 4d regions, indicating strong Mn 3d-O 2p and Y 4d-O 2p bond anisotropy, respectively. The Mn 3d region consists of four major features, which can be identified as  $a_{1g}$  ( $z_1^2$ ),  $e_{1g}(yz_1/zx_1), e_{2g}(xy_1/x^2 - y_1^2)$ , and  $a_{1g}(z_1^2)$  as denoted in Fig. 1(a). The more detailed features were analyzed by using the cluster model calculations including the configuration interaction (CI) and atomic full multiplets [16], as compared in the figure. In the model calculation, Mn 3d-O 2p hybridizations were presented by the Slater-Koster transfer integrals,  $V_{pd\sigma}$  and  $V_{pd\pi}$  with  $V_{pd\sigma} = -2V_{pd\pi}$ [17]. The Harrison's rule  $(d^{-3.5})$  was applied for different interatomic distances to the transfer integrals [18]. The bipyramid MnO<sub>5</sub> local structure leads about 3 times larger effective hybridization strength for the  $z^2$  orbital than for other orbitals. The  $Mn^{3+}$  (3d<sup>4</sup>) ionic ground state was found to be S = 2 with empty  $z^2$  orbital. Considering the O 1s core-hole energy difference (0.4 eV) of  $O_T$  and  $O_P$  in the final state, we found that the energies relative to the  $a_{1g}$  $(z_1^2)$  lowest energy state are 0.4 eV  $(e_{1g})$ , 1.1 eV  $(e_{2g})$ , and 1.9 eV  $(a_{1g})$ , very consistent with the previous reports [19].

The Y 4d region consists of three major features, corresponding to a doublet  $e_g^{\pi}$ , a singlet  $a_{1g}$ , and a doublet  $e_g^{\sigma}$ orbital as denoted in Fig. 1(a). Y ions are surrounded by six  $O_T$  and two  $O_P$  ions, i.e.,  $YO_8$ , with a  $D_{3d}$  trigonal site symmetry. The crystal field splittings are schematically depicted in Fig. 1(b). The spectral features were also analyzed by using the cluster model calculations. The calculated spectra presented in Fig. 1(a) agree well with the experiments for both  $E \parallel c$  and  $E \parallel ab$ . Y 4d-O 2p hybridization is so strong that it contributes even larger intensity (O 2p PDOS) than the Mn 3d-O 2p one, in contrast to the band calculations [13], which predict that Y 4d-O 2p hybridization is negligible. As also compared in the figure, the O 2p PDOS quoted from the band calculations is far away from the spectra even in the Mn 3dregion, indicating that the one-electron approximation may not be appropriate for this material. Further, the spectra show that the  $a_{1g}$  state, which mainly hybridizes with  $O_P$ ions along the c axis, has quite large transferred O 2pweight. Considering the fact that the off-center movements mostly occur at Y-O<sub>P</sub> sites, such a large amount of transferred charge must contribute large anomalies in Born effective charges on the ions.

Figure 2 shows local structures of  $MnO_5$  and  $YO_8$  in ferroelectric and centrosymmetric lattices [20]. The  $MnO_5$ 



FIG. 2 (color online). Local structures of  $MnO_5$  and  $YO_8$  ( $Y^1O_8$  and  $Y^2O_8$ ) cages in ferroelectric and the centrosymmetric phases.

cage simply rotates with the ferroelectric distortions without off centering. The large off-center movements occur at  $Y^1O_8$  and  $Y^2O_8$  cages.  $Y-O_T$  bond lengths barely change but  $Y^1-O_P$  and  $Y^2-O_P$  bond lengths change by huge amounts, 23% (0.54 Å) and 17% (0.39 Å) of the ordinary Y-O distance, 2.3 Å, as in  $Y_2O_3$ , and according to the Harrison's rule for the *p-d* hybridization,  $|V_{pd}| \propto d^{-3.5}$ [18], the  $O_P$ -Y- $O_P$  hybridization strengths,  $|V_{pd}|^2$ , are enhanced by 73% and 34%, which contribute 8% and 4% enhancements to the total *p-d* hybridization strengths of  $Y^1O_8$  and  $Y^2O_8$ , respectively. These enhancements are comparable to 5% enhancement in BaTiO<sub>3</sub> (TiO<sub>6</sub>).

Using the Harrison's bond-orbital model [18], one can roughly estimate Born effective charges which are defined as sums of static charges and additional dynamic contributions. The static charge is reduced from the ionic value by the amount of transferred charges due to hybridization. The dynamic contribution originates in additional charge transfer due to a variation of the interatomic distance. For a p-dorbital hybridization, the amount of transferred charge  $(\Delta Z)$  corresponds to  $\sim |V_{pd}^{\text{eff}}|^2/(\epsilon_d - \epsilon_p) \propto d^{-7}$ , and the dynamic contribution does to  $-7 \times \Delta Z$  in the first order perturbation; i.e., both are proportional to the amount of transferred O 2p electrons through the hybridization [8]. Here  $|V_{pd}^{\text{eff}}|^2$  and  $\epsilon_d - \epsilon_p$  denote the effective summation of the *p*-*d* hybridizations for different orbitals and the charge transfer energy, respectively. The O K edge XAS shows that the transferred O 2p PDOS through Y 4d-O 2phybridization is much larger (more than  $\sim 40\%$ ) than that through Mn 3d-O 2p one, and is expected to be larger than that through Ti 3d-O 2p one in BaTiO<sub>3</sub> [21]. It is because the stronger Y 4d-O 2p hybridization sufficiently compensates the larger charge transfer energy. Considering the large  $|V_{pd}^{\text{eff}}|^2/(\epsilon_d - \epsilon_p)$  and the hybridization enhancement due to the ferroelectric movements in the Y 4d-O 2pbonds, the dynamic contribution in Born effective charges on Y and  $O_P$  ions should be comparable or even larger than those on Ti and O<sub>||</sub> ions in BaTiO<sub>3</sub>. Here we cannot exactly estimate the Born effective charges of Y and  $O_P$  [22], but one can easily expect that those are at least similar to those of Ti (  $\sim +7)$  and  $O_{\parallel}$  (  $\sim -6)$  in BaTiO\_3 [8]. Hence the polarization in YMnO<sub>3</sub> is certainly dominated by the large dynamic contributions in Born effective charges, rather than the electrostatic charges, and the ferroelectric instability must be attributed to the Y  $d^0$ -ness with rehybridization, in an analogy to the Ti  $d^0$ -ness in BaTiO<sub>3</sub>. Such large Born effective charges are consistent with the high  $T_E$ . The small ferroelectric polarization (5.5  $\mu$ C cm<sup>2</sup>) in comparison with that in BaTiO<sub>3</sub> (25  $\mu$ C cm<sup>2</sup>), could be attributed to the "ferrielectric" character; the opposite polarizations between  $Y^1O_8$  and  $Y^2O_8$  with 1:2 ratio.

To examine the cluster model analysis for Mn 3*d* states, we analyzed polarization dependent XAS spectra at Mn  $L_{2,3}$  edges with about the same parameter set [23]. The experimental spectra at Mn  $L_{2,3}$  edges were compared with those of the model calculation results in Fig. 3. The calculation well reproduces the detailed spectral features not only for both polarizations (*E* || *c* and *E* || *ab*) but also for the difference spectrum. This result confirms that the electronic structure can be well explained by means of the cluster model analysis including CI and atomic full multiplets.

Now we try to extend our understanding. Figure 4 shows (a) Mn  $L_{2,3}$ - and (b) O K edge polarization dependent XAS spectra for the hexagonal YMnO<sub>3</sub> and orthorhombic LaMnO<sub>3</sub> in comparison with hexagonal and orthorhombic DyMnO<sub>3</sub> epitaxial films grown on Pt(111)/Al<sub>2</sub>O<sub>3</sub>(0001) and SrTiO<sub>3</sub>(100) substrates, respectively [24]. For both polarizations, the XAS spectra of YMnO<sub>3</sub> and LaMnO<sub>3</sub> show very similar line shapes to those of the hexagonal and orthorhombic DyMnO<sub>3</sub>, respectively. The spectra of the hexagonal manganites exhibit strong polarization depen-



FIG. 3. Mn *L* edge polarization dependent XAS spectra and the difference spectrum,  $(E \parallel c) - (E \parallel ab)$ , of YMnO<sub>3</sub> in comparison with the CI model calculations.



FIG. 4. Polarization dependent XAS spectra of hexagonal  $RMnO_3$  (R = Y, Dy) and orthorhombic  $RMnO_3$  (R = La, Dy) at (a) Mn  $L_{2,3}$ - and (b) O K edge collected in TEY mode. The  $E \parallel c$  spectra for DyMnO<sub>3</sub> films were approximated by the spectra at  $\angle (E, c) = 20^{\circ}$ .

dence at both Mn  $L_{2,3}$ - and O K edges, while the orthorhombic manganites barely show polarization dependence. It demonstrates that the ground state symmetry and the Mn 3*d*-orbital energetics are mostly determined by the crystal structure. Further one can find that Dy 5d region in the O K edge XAS of the hexagonal DyMnO<sub>3</sub> is rather similar to the Y 4d region not only for the polarization dependence but also in the absorption intensity. It indicates that the hybridization strength is similar for Y 4d-O 2p and Dy 5d-O 2p bonds in spite of the difference of the 4d and 5dorbitals. The O K edge XAS spectrum of another hexagonal HoMnO<sub>3</sub> also shows very similar absorption intensity at the Ho 5d region, indicating similar hybridization in Ho 5d-O 2p [25]. The identical XAS spectrum was also found at Mn  $L_{2,3}$  edge for a hexagonal ErMnO<sub>3</sub> [26]. Further, the similarity in the hybridization can be also seen in other 4dand 5d oxides [27]. These results suggest that the  $d^0$ -ness scenario can be extended to other hexagonal manganites with rare-earth  $5d^0$ -ness.

In conclusion, we investigated the details of electronic and bond anisotropies of hexagonal YMnO<sub>3</sub>. The studies show strong anisotropic hybridization not only for Mn 3d-O 2p but also for Y 4d-O 2p bonds. As the results, the off-center movements contribute large anomalies in Born effective charges through rehybridization. These results manifest that the ferroelectric instability is originated from the Y  $d^0$ -ness with rehybridization together with the structural phonon instability. Further we also found that the  $d^0$ -ness scenario is applicable for the ferroelectricity of other hexagonal manganites.

This work is supported by the KOSEF through the CSCMR at SNU and eSSC at POSTECH, Korean Ministry of Science and Technology through Creative Research Program at SNU, the National Science Foundation-MRSEC, POSTECH research fund, and BK21 program. Experiments at PLS were supported in

part by MOST and POSTECH.

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