Nanometer Scale Electronic Reconstruction at the Interface between LaVO₃ and LaVO₄

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Electrons at interfaces, driven to minimize their free energy, are distributed differently than in bulk. This can be dramatic at interfaces involving heterovalent compounds. Here we profile an abrupt interface between V $3d^2$ LaVO₃ and V $3d^0$ LaVO₄ using electron energy loss spectroscopy. Although no bulk phase of LaVO_x with a V $3d^1$ configuration exists, we find a nanometer-wide region of V $3d^1$ at the LaVO₃/LaVO₄ interface, rather than a mixture of V $3d^0$ and V $3d^2$. The two-dimensional sheet of $3d^1$ electrons is a prototypical electronic reconstruction at an interface between competing ground states.

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Recent advances in oxide thin film growth and probe techniques have revealed a host of important considerations for the electronic structure at interfaces. For example, carrier depletion has been found to limit the superconducting critical current at cuprate grain boundaries [1]. Reduced surface magnetism limits the magnetoresistance of manganite tunnel junctions [2]. In these examples, atomic scale doping of the interface was found to significantly improve device performance. The role of this doping is to compensate for deviations of the surface electronic structure from the bulk, which can often occur in response to the electrostatic boundary conditions at the interface [3]. These deviations can be large in multivalent transition metal oxides, for which significant charge shifts are energetically accessible.

In addition to being a mechanism for degrading bulklike properties in device geometries, electronic interface reconstructions also present a unique synthetic opportunity to create new artificial states between two materials. Recent examples include interface ferromagnetism induced between two paramagnets [4] and conducting interfaces induced between two insulators [5,6]. These phenomena have stimulated theoretical efforts to examine interface phases created by charge transfer in artificial structures [7,8]. The examples of interface phases given above utilize valence states that are accessible in bulk form. Here we demonstrate experimentally that interface reconstructions can also be used to induce and stabilize nonbulklike valence states via a similar mechanism.

The multivalent $LaVO_r$ shows a variety of electronic and magnetic properties as the vanadium valence changes from 3+ to 5+ [9-12]. LaV³⁺O₃, with a $3d^2$ shell configuration, is a Mott insulator, whereas $LaV^{5+}O_4$ is a band insulator with a $3d^0$ configuration. Intriguingly, bulk LaVO_x lacks a structural phase corresponding to V^{4+} [13]. However, at the interface between $LaV^{+3}O_3$ and $LaV^{+5}O_5$, continuity of the charge distribution implies that the vanadium valence should transition through 4 +, at least in an average sense. Microscopically, the question is whether this is achieved through a mixture of the ground state $3d^0$ and $3d^2$ configurations, likely to be favored by a Maxwell construction from the bulk phases, or whether a local region of $3d^1$ -configured vanadium ions can be stabilized by the interface.

In this Letter, we present a study of the microscopic electronic structure of LaVO_x films by spatially resolved electron energy loss spectroscopy (EELS) in a scanning transmission electron microscope (STEM). Vanadium L edge fingerprints for V^{3+} , V^{4+} , and V^{5+} were established from bulk reference samples and used for the characterization of the vanadate films and the interface between $LaV^{3+}O_3$ and $LaV^{5+}O_4$. We find that $LaVO_3$ grows epitaxially on the SrTiO₃ substrate, resulting in atomically flat surface regions, whereas LaVO₄ nucleates on a thin LaVO₃ layer and forms polycrystalline 3D islands. At the $LaV^{3+}O_3/LaV^{5+}O_4$ interface, our EELS measurements reveal that a two-dimensional layer of $3d^1$ -configured V^{4+} has been stabilized, thus achieving an interface charge configuration that was not accessible in bulk LaVO_x.

LaVO_x films were grown on TiO_2 terminated (001)oriented SrTiO₃ substrates by pulsed laser deposition using single-phase LaVO₄ polycrystalline targets. The LaVO_x films studied here were grown at $T_g = 800 \,^{\circ}\text{C}$ and P_{O2} of 10^{-5} – 10^{-4} Torr, conditions for which LaVO₃ and LaVO₄ are stabilized simultaneously, resulting in mixed phase growth [14].

The microstructures of the LaVO_x films are studied by annular dark field (ADF) imaging in a 200 kV FEI Tecnai F20-ST STEM with a minimum probe size of ~ 1.6 Å [15] and a convergence semiangle of (9 ± 1) mrad. The chemical and electronic structures are probed on an atomic scale using spatially resolved EELS, performed both on a vacuum generator (VG) HB-501 100 kV dedicated STEM $(\sim 10 \text{ mrad})$, which is equipped with a parallel electron energy loss spectrometer, and on the Tecnai F20-ST fitted with a monochromator and a Gatan imaging filter (GIF) 865-ER. The energy resolution, as measured from the FWHM of the zero loss peak, was 0.5 and 0.6 eV in these two respective cases.

Figure 1 shows V- $L_{2,3}$ and O-K EELS spectra of bulk LaVO₃, SrVO₃, and LaVO₄. The first two main peaks of the spectra are the V- L_3 and L_2 edges. At higher energies, the peaks are dominated by the O-K fine structure; however, extraction of the O-K edge is hampered by the proximity of the V- $L_{2,3}$ edge. Large changes of the O-K and the V- $L_{2,3}$ edge fine structure are observed, in which the changes of the V- $L_{2,3}$ edge peak position are dominated by the change of the vanadium valence from 3+ to 4+ to 5 + [16,17]. With increasing vanadium valence, the V- $L_{2,3}$ peak position shifts by ~0.9 (V⁴⁺) and ~1.7 eV (V⁵⁺) towards higher energies. This spectral signature can be used to determine the vanadium valence in materials with predominantly ionic character, as the V- $L_{2,3}$ edge can be reasonably described by a superposition of the spectra corresponding to the different valence states [5,18].

Figure 2(a) shows a cross-sectional ADF-STEM image of a LaVO_x film grown on SrTiO₃ viewed along a substrate [001] zone axis. As expected for these growth conditions, we find two regions in the film: The brighter regions show a uniform, well-crystallized, smooth film, whereas the darker regions appear to be 3D islands that nucleated during the growth. The composition was probed using spatially resolved EELS. Figure 2(b) shows V- $L_{2,3}$ /O-Kspectra for the smooth regions of the film and the 3D islands, respectively. For better comparison of the fine

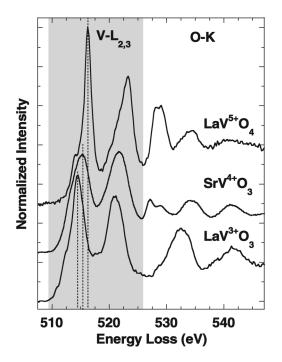


FIG. 1. $V-L_{2,3}$ and O-K electron energy loss spectra taken from bulk LaVO₃, SrVO₃, and LaVO₄. The dashed lines indicate the L_3 peak position, which shifts to higher energies as the vanadium valence increases. The shaded region of the spectra is due to $V-L_{2,3}$ excitations only, whereas at higher energies the EELS signal is composed of V and O-K edge contributions. The measurements were performed on the FEI Tecnai F20-ST equipped with a GIF 865-ER.

structure, the spectra were normalized to the integrated intensity under the V- $L_{2,3}$ edge. On the 3D islands, the V- L_3 peak shifts by $\Delta E \sim 1.7$ eV towards higher energies. This energy shift and the direct comparison of the spectra with the V fingerprints (Fig. 1) are consistent with a vanadium valence change from V³⁺ in the smooth film to V⁵⁺ in the 3D islands. Hence, under these growth conditions there is two-phase growth of uniform, smooth LaVO₃ and 3D LaVO₄ islands. Figure 2(c) shows an atomic force microscopy (AFM) image representative of the film, confirming the formation of 3D islands and regions of atomically flat surface, which correspond to LaVO₄ and LaVO₃, respectively.

The reduced contrast of the LaVO₄ islands compared to the LaVO₃ layer is due to the relative orientation of the crystals compared to the electron propagation direction. ADF-STEM images of crystals are sensitive to crystal orientation [19], because channeling of the probe on the atom column for crystals oriented on the zone axis leads to contrast enhancement in the ADF-STEM image. As the crystal is tilted off axis, the channeling effect is reduced and the contrast diminishes. In Fig. 2, the LaVO₃ and the SrTiO₃ substrate are aligned on the zone axis, whereas the LaVO₄ polycrystalline islands are mostly oriented off the axis. Figure 3(c) shows a high resolution ADF-STEM image of a region in a LaVO₄ island that is oriented on the zone axis. The difference between the $LaVO_4$ [Fig. 3(c)] and the SrTiO₃ crystal structure [Fig. 3(d)] is apparent. SrTiO₃ shows the typical cubic perovskite structure, whereas LaVO₄ is monoclinic with monazite-type structure $(P2_1/c, a = 7.047 \text{ Å}, b = 7.286 \text{ Å}, c = 6.725 \text{ Å}, c$

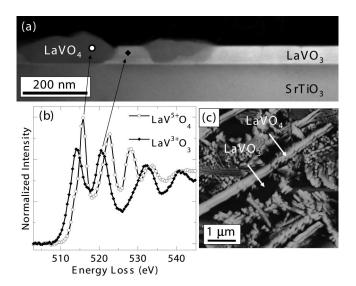


FIG. 2. (a) Cross-sectional ADF image of a LaVO_x film grown at $T_g = 800$ °C and $P_{O2} = 10^{-4}$ Torr showing mixed phase growth of 3D LaVO₄ islands and flat LaVO₃ on a SrTiO₃ substrate. (b) V- $L_{2,3}$ /O-K electron energy loss spectra taken from an island and from the smooth film, respectively. (c) AFM image of the same film showing both flat regions of LaVO₃ and 3D islands of LaVO₄.

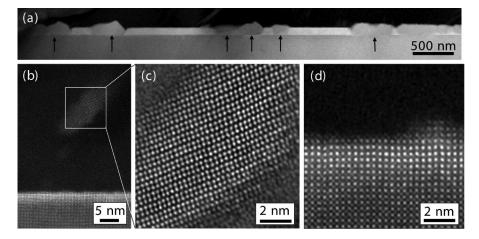


FIG. 3. (a) Low magnification ADF image of a LaVO₃ film grown at $T_g = 800$ °C and $P_{O2} = 10^{-4}$ Torr, where LaVO₄ 3D islands are marked with arrows. (b) Higher magnification ADF image of a LaVO₃ film grown at $T_g = 800$ °C and $P_{O2} = 10^{-5}$ Torr showing a bright band at the interface between the LaVO₄ and the SrTiO₃ substrate. (c) The crystal structure of the monoclinic LaVO₄ recorded along the [010] crystallographic direction for the area marked by the box in (b). (d) The sharp interface between the LaVO₄ and the bright band.

 $\beta = 104.85^{\circ}$) [20], viewed in Fig. 3(c) along the LaVO₄ [010] direction.

Figures 3(b) and 3(d) show a bright band between the LaVO₄ and the SrTiO₃ substrate, which suggests a chemical change for zone-oriented crystals. EELS on the bright layer confirms the presence of LaVO₃; i.e., the 3D LaVO₄ islands do not nucleate directly on the SrTiO₃ substrate but on a thin (~2 nm) LaVO₃ layer. This raises an interesting question of how the microscopic transition from V^{3+} to V^{5+} is made at this abrupt interface. To explore this, we have measured V- $L_{2,3}$ edge spectra across the interface between the thin LaVO₃ layer and the LaVO₄ island. Figure 4 shows the result of a scan across the interface. The probing path is indicated by the dotted line in the ADF image [Fig. 4(c)]. By decomposing the V- $L_{2,3}$ near edge structure into a linear combination of V^{3+} , V^{4+} , and V^{5+} components [Fig. 4(b)], we can monitor the change of V valence at the interface. The results of this threecomponent fit are shown in Figs. 5(a) and 5(c). As expected, we find the transition from V^{3+} in the thin bright layer to V⁵⁺ in the LaVO₄ islands. However, at the crossover between these valence states, we find a significant contribution of V^{4+} (i.e., V in a $3d^1$ configuration), with an area accounting for close to one electron at the LaVO₃/LaVO₄ interface. The residuals, i.e., the rootmean square deviation of the experimental data from the fit, are constantly low across the interface in the threecomponent case. A fit of this series of spectra with only two components (V^{3+} and V^{5+}) as shown in Fig. 5(b) results in large residuals at the interface, indicating the presence of a third component. Hence, at the $LaVO_3/LaVO_4$ interface the vanadium valence varies smoothly from V^{3+} to V^{5+} , passing through a region of V^{4+} at the interface.

Microscopically, the local electronic distribution is not phase separated into only the bulk-accessible $3d^0$ and $3d^2$ -configured V d states but also includes a thin layer

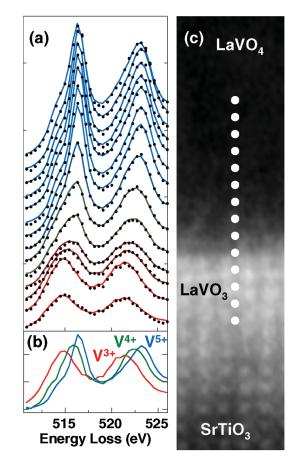


FIG. 4 (color). (a) $V-L_{2,3}$ electron energy loss spectra recorded across the LaVO₄/LaVO₃ interface (dotted lines). The solid lines show the fits to the experimental spectra using reference spectra shown in (b). (c) ADF image of the interface where the electron beam position is denoted by the white dotted line. The measurements (a),(c) were performed on the VG HB-501 100 kV dedicated STEM equipped with a parallel electron energy loss spectrometer.

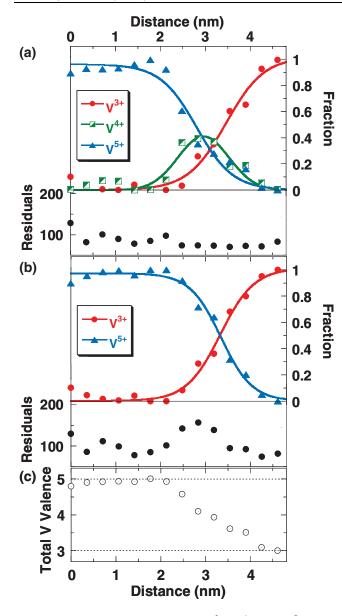


FIG. 5 (color). (a) The fractions of V³⁺, V⁴⁺, and V⁵⁺ across the interface obtained by a three-component fit to the experimental data shown in Fig. 4(a) using the reference spectra in Fig. 4(b). These fractions were used to color code the fits to the experimental data in Fig. 4(a). The root-mean square residuals of the fit are shown below. (b) A two-component fit of the data using V³⁺ and V⁵⁺ reference spectra results in large residuals at the interface. (c) Total vanadium valence across the LaVO₄/LaVO₃ interface as determined from (a).

of $3d^1$ V atoms at the interface. The $3d^1$ state would not be expected from the bulk available phases, nor is it required from continuity of the wave function in an average sense, highlighting the very different electronic response at the interface.

In summary, we have shown that the nonbulklike charge state $3d^1$ V can be stabilized at the LaVO₃/LaVO₄ interface. Here an interface between two competing ground states (the $3d^2$ Mott insulator and $3d^0$ band insulator) has

led to a two-dimensional sheet of intermediate valence $3d^1$ electrons. As there is a large family of closely latticematched transition metal oxides with different and exotic ground states, other electronic reconstructions and intermediate valence states may also be stabilized by this approach.

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- G. Hammerl, A. Schmehl, R.R. Schulz, B. Goetz, H. Bielefeldt, C. W. Schneider, H. Hilgenkamp, and J. Mannhart, Nature (London) 407, 162 (2000).
- [2] H. Yamada, Y. Ogawa, Y. Ishii, H. Sato, M. Kawasaki, H. Akoh, and Y. Tokura, Science **305**, 646 (2004).
- [3] N. Nakagawa, H. Y. Hwang, and D. A. Muller, Nat. Mater. 5, 204 (2006).
- [4] K. S. Takahashi, M. Kawasaki, and Y. Tokura, Appl. Phys. Lett. 79, 1324 (2001).
- [5] A. Ohtomo, D. A. Muller, J. L. Grazul, and H. Y. Hwang, Nature (London) 419, 378 (2002).
- [6] A. Ohtomo and H. Y. Hwang, Nature (London) **427**, 423 (2004).
- [7] S. Okamoto and A. J. Millis, Nature (London) 428, 630 (2004).
- [8] Z. S. Popovic and S. Satpathy, Phys. Rev. Lett. 94, 176805 (2005).
- [9] V.G. Zubkov, G.V. Bazuev, V.A. Perelyaev, and G.P. Shveikin, Sov. Phys. Solid State 15, 1079 (1973).
- [10] G. Khaliullin, P. Horsch, and A. M. Oles, Phys. Rev. Lett. 86, 3879 (2001).
- [11] S. Miyasaka, Y. Okimoto, and Y. Tokura, J. Phys. Soc. Jpn. 71, 2086 (2002).
- [12] H. Sawada, N. Hamada, K. Terakura, and T. Asada, Phys. Rev. B 53, 12742 (1996).
- [13] M. A. Subramanian, G. Aravamudan, and G. V. SubbaRao, Prog. Solid State Chem. 15, 55 (1983).
- [14] Y. Hotta, Y. Mukunoki, T. Susaki, H. Y. Hwang, L. Fitting, and D. A. Muller, Appl. Phys. Lett. 89, 031918 (2006).
- [15] M. Weyland and D. A. Muller, Nano Sol. 1, 24 (2005).
- [16] G. van der Laan and I.W. Kirkman, J. Phys. Condens. Matter 4, 4189 (1992).
- [17] P.F. Schofield, C.M.B. Henderson, G. Cressey, and G. van der Laan, J. Synchrotron Radiat. 2, 93 (1995).
- [18] A. Fujimori, I. Hase, M. Nakamura, H. Namatame, Y. Fujishima, Y. Tokura, M. Abbate, F. M. F. de Groot, M. T. Czyzyk, J.C. Fuggle, O. Strebel, F. Lopez, M. Domke, and G. Kaindl, Phys. Rev. B 46, 9841 (1992).
- [19] R. F. Loane, E. J. Kirkland, and J. Silcox, Acta Crystallogr. Sect. A 44, 912 (1988).
- [20] C. E. Rice and W. R. Robinson, Acta Crystallogr. Sect. B 32, 2232 (1976).