Stability of Membrane Bound Reactions

R. Thul and M. Falcke

Hahn-Meitner Institut, Abteilung Theorie, Glienicker Strasse 100, D-14109 Berlin, Germany (Received 30 April 2004; published 29 October 2004)

We present a novel approach to the dynamics of reactions of diffusing chemical species with species fixed in space, e.g., by binding to a membrane. The nondiffusing reaction partners are clustered in areas with a diameter smaller than the diffusion length of the diffusing partner. The activated fraction of the fixed species determines the size of an active subregion of the cluster. Linear stability analysis reveals that diffusion is one of the major determinants of the stability of the dynamics. We illustrate the model concept with Ca²⁺ dynamics in living cells, which has release channels as fixed reaction partners. Our results suggest that spatial and temporal structures in intracellular Ca²⁺ dynamics are caused by fluctuations due to the small number of channels per cluster.

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Huge chemical gradients occur during dynamic changes of concentrations in living cells. Ions may enter the cytosol through tiny channel pores or chemical species in solution may react with membrane bound partners, causing gradients by spatially inhomogeneous production or consumption. An example is the reaction of the membrane bound adenylate cyclase with adenosine triphosphate (ATP) to form cyclic adenosine monophosphate. Usually, reactions are modeled as occurring between species with spatially uniform concentrations neglecting spatial gradients. Here, we show that these gradients have a substantial impact on the dynamics of the reaction. This applies, in particular, when the membrane bound reaction partners are packed into small clusters with a few tens of individual molecules, which is the case we consider. The small number of elements per cluster may necessitate stochastic approaches which have been carried out for intracellular Ca²⁺ dynamics [1–3]. Here, we present an investigation of the deterministic limit of a single cluster.

We consider the cluster to be a membrane patch reacting with the dissolved reaction partner. The small number of molecules per cluster entails a cluster size of the same order of magnitude as the single molecules. Therefore when a membrane bound molecule enters the reaction (e.g., by activation) or leaves it (e.g., by deactivation) the area of the reacting patch changes rather than the concentration of active molecules in a fixed area. Consequently, we model the dynamics of the number of membrane bound molecules taking part in the reaction as a size change of the reacting area. This modeling concept is supported by the results of recent investigations [4,5]. Simulations in which the reaction was released from an intracellular store through ion channels showed that reactions saturate at much lower maximal rates when the fraction of reacting membrane bound molecules is described by a changing concentration in a fixed area, compared to the rate values obtained with the approach we have chosen [4]. Since cluster diameters are smaller than the diffusion length of the partner in solution, the size of the cluster limits the transport of reaction partners to the reaction and therefore the rate.

We demonstrate our modeling ideas and results with the example of intracellular Ca²⁺ dynamics. From the very beginning of the life of an organism to its end, calcium is involved in signaling and control of many processes in almost all of its cells. As a second messenger, it communicates the fertilization across an egg cell and controls apoptosis; it plays an active role in muscle contraction and secretion and has many other functions [6]. Ca²⁺ fulfills its signaling task by a transient rise of the Ca²⁺ concentration in the cytosol. This is accomplished by release and uptake of Ca²⁺ by storage compartments like the endoplasmic reticulum (ER) or mitochondria. These stores are embedded in the cytosol. The flux of Ca²⁺ between the cytosol and the ER is controlled by inositol-1,4,5-trisphosphate receptor channels (IP₃R) for release and the sarco-endoplasmic reticulum Ca²⁺ ATPase pumps for uptake. Although the receptor channels can be found isolated on the membrane of the ER, they usually build clusters with a diameter of 60–100 nm comprising between 5 and 40 IP₃R's. Swillens et al. estimated that channels are densely packed inside a cluster, whereas the distance between clusters may range from 3 to 7 μ m [5,7]. Thus cluster distances are 2 orders of magnitude larger than cluster diameters.

The open probability of IP_3R 's depends on the cytosolic Ca^{2^+} concentration. A moderate increase raises the opening probability, whereas a large concentration rise inhibits and closes the release channel. The channel cannot open as long as it is inhibited. Hence, channels are coupled by Ca^{2^+} diffusion in the cytosol since they release what controls their states. The channels within a cluster are strongly coupled since the concentration does not decay on the length scale of a cluster diameter, but the coupling of clusters is only weak.

The small number of channels per cluster and the weak coupling of clusters suggest that fluctuations caused by the random opening and closing of channels are important. This was confirmed by stochastic modeling in the last three years [1–3]. Stochastic models showed spatial and temporal structures even with parameters providing a nonoscillatory or nonexcitable deterministic regime. The transition from deterministic to stochastic models was accompanied by a transition from continuous to spatially discrete models. The loss of the oscillatory regime in going from the stochastic to the deterministic approach is explained by the results of this study.

Discrete models require information on the concentration gradients. Simulations of release of Ca²⁺ through IP₃R's close to the experimental situation showed that the concentration values at an open channel reach 25–170 μ M; i.e., they are 3–4 orders of magnitude larger than the resting level while the concentration increase at the neighboring cluster reaches only 1.1-2.0 times the resting level [4]. Experimental observations like propagating waves and theoretical considerations indicate that the sensitivity of the activating Ca²⁺ dependence of the opening probability on Ca²⁺ is tuned to concentration levels close to the resting level and that the inhibitory process has half maximum values of several μ M. Hence, realistic concentration values at an open channel are likely to saturate inhibition as well as activation by Ca²⁺. These restrictions on the dynamics have to be taken into account in modeling. Spatially averaged concentration values cannot be used since they change by 1 order of magnitude only.

Our approach to the modeling of intracellular Ca²⁺ dynamics incorporates the extreme localization of release. As mentioned above, channel clusters have a typical diameter below 100 nm. The size of a channel molecule with all its four subunits is about 18 nm in diameter. When a single channel opens or closes, the fraction of cluster area taking part in conduction of Ca²⁺ changes to a nonnegligible degree. Open channels are randomly scattered across the cluster area. However, Swillens et al. [5] have shown that it is possible to merge the individual areas of all open IP₃R's to a single area of the same size without affecting the concentration dynamics around a cluster. Thus the dynamic change of the number of open channels in a cluster can be described as the dynamic change of the radius a of a concentric conducting area. The value of a is 0 when all channels are closed and equal to the cluster radius a_0 when all channels are open. Merely for the purpose of simplifying the calculations we consider the cluster to be a sphere rather than a membrane area.

The dynamics of the cytosolic Ca^{2+} concentration c obeys the partial differential equation:

$$\dot{c} = D\nabla_r^2 c + k_I (E - c) - k_D c + k_C (E - c)\theta(a - r). \tag{1}$$

Here, ∇_r^2 denotes the radial part of the Laplace operator in three dimensions. The second term on the right-hand side of Eq. (1) refers to a leak flux; the term $k_p c$ describes uptake of Ca^{2+} by the ER and $k_c(E-c)$ release through open channels. E is the Ca^{2+} concentration in the ER. Release is restricted to the volume $r \leq a$. $\theta(x)$ is the Heaviside step function.

The radius a of the conducting area is determined by the fraction of channels in the open state. The open state depends on the state of the channel subunits. Each channel consists of four identical subunits. Subunits have binding sites for Ca²⁺ and IP₃. The state of a subunit is determined by the occupation of the binding sites. We use the state scheme of the De Young-Keizer (DK) model for the subunit state dynamics. Each subunit has an IP₃-binding site, as well as an activating and an inhibiting Ca²⁺-binding site. If IP₃ is bound to its binding site and a Ca²⁺ ion to the activating Ca²⁺-binding site, the subunit is activated. The channel is open if a minimum number of subunits is activated. Binding of Ca²⁺ to the inhibiting binding site inhibits the subunit. A subunit can be activated again only upon recovery from inhibition. Eight states of a subunit arise from the three binding sites. We denote the fraction of subunits in a certain state with p_{ijk} where an index is 1, if a binding site is occupied, and 0 otherwise. The indices i, j, and k represent the IP₃-binding site, the activating Ca²⁺-binding site, and the inhibiting Ca²⁺-binding site, respectively. Therefore the fraction of open subunits is p_{110} . We refer to the set $\{p\}$ of the p_{ijk} as gating variables. Their dynamics is of the general form

$$\dot{p}_{ijk} = g_{ijk}(c(a, t), \{p\}).$$
 (2)

Together with the concentration field c(r, t) the gating variables $\{p\}$ determine the radius a by an algebraic relation. The fraction of open channels is given by the probability that at least three out of four subunits of a channel are activated. The radius a of the volume occupied by this fraction is

$$a = a_0 p_{110} \sqrt[3]{4 - 3p_{110}}. (3)$$

Stationary solutions of Eqs. (1) and (2) are determined by

$$0 = D\nabla_r^2 c + k_l(E - c) - k_p c + k_c(E - c)\theta(a - r),$$
 (4)

$$0 = g_{ijk}(c(a), \{p\}). \tag{5}$$

Because of the Heaviside function the equation can be treated separately for r < a and r > a. Solving Eq. (4) for the stationary calcium concentration c_{st} yields

$$c_{st}(r) = B(a) \frac{\exp[k_2(r-2b)] - \exp(-k_2r)}{r} \Theta(r-a)$$

$$+A(a) \frac{\sinh(k_1r)}{r} \Theta(a-r), \qquad (6)$$

$$k_1^2 = \frac{k_l + k_p + k_c}{D}, \qquad k_2^2 = \frac{k_l + k_p}{D}.$$

We applied the boundary conditions $c(b) = k_l E/(k_l + k_p)$ at the outer radius b of the cytosol, which corresponds to the base level concentration of the system. The requirement of $c_{\rm st}(r)$ to be ${\rm C}^1$ at r=a fixes the constants A(a) and B(a). Using the state scheme of De Young and Keizer for a single subunit [8], the stationary state of p_{110} dependent on c and the IP₃ concentration I is

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$$p_{110}^{\text{st}} = \frac{d_2 c_{\text{st}}(a) I}{[c_{\text{st}}(a) + d_5][d_1 d_2 + c_{\text{st}}(a) d_3 + c_{\text{st}}(a) I + d_2 I]}.$$
(7)

Here d_1 and d_3 denote the dissociation constants for IP₃ when no Ca²⁺ and when Ca²⁺, respectively, is bound to the inhibiting site. The parameters d_2 and d_4 represent the dissociation constants for the inhibiting Ca²⁺ processes, depending on IP₃ binding, and d_5 is the dissociation constant for the activating Ca²⁺ site.

The value of the Ca^{2+} concentration does not vary significantly within the cluster. Hence, we can pick a typical value to enter the dynamics of the gating variables. The value of c at the boundary of the conducting area was chosen [see Eq. (7)]. The dependence of $p_{110}^{\rm st}$ on c(a) turns Eq. (3) into an implicit function for the value of a at the stationary state: a = f(a). The solution set of this equation determines the stationary states of the system. Saddle node bifurcations can be easily obtained by the condition that the bisection line touches f(a): 1 = f'(a). Similarly, the stability analysis can be simplified to the solution of an algebraic equation, too. This will be explained in detail in [9].

The DK model assumes that the dynamics of IP₃ binding and dissociation is much faster than the dynamics of Ca²⁺ binding and dissociation. Therefore we eliminate the IP₃ dynamics in the following by assuming that it is always in its stationary state. This does not alter the expression for p_{110}^{st} . The original DK model is based on a spatially continuous IP₃R density and exhibits two Hopf bifurcations bounding an oscillatory regime. We can transform a spatially continuous model into a discrete model by concentrating the flux in an area determined by a typical cluster distance R into an area of typical cluster size (a_0) while conserving the total flux. Consequently, we rescale the original parameter k_c^{DK} in [8] by $k_c = k_c^{\rm DK} R^3/a_0^3$. With the resulting large value of k_c , which is close to realistic values of $3 \times 10^5~\mu{\rm m\,s^{-1}}$ [4], the oscillatory regime is lost and the model exhibits only a single stationary state which is stable for all values of the IP₃ concentration. Lowering k_c by several orders of magnitude and thus approaching the original value does not restore the linear stability properties of the continuous model since gradients still occur. This is similar to findings in [10].

However, adaptation of a model designed as spatially continuous to spatially discrete source terms requires more than rescaling of the source strength. The DK model was set up for spatially averaged ${\rm Ca^{2+}}$ concentrations which are much lower than concentrations at the channel mouth. This would suggest to adapt dissociation constants for ${\rm Ca^{2+}}$ -dependent processes. We increased the value of both ${\rm Ca^{2+}}$ dissociation constants (dissociation constant = dissociation rate/binding rate constant) for inhibition and activation, d_2 and d_5 , respectively. Experimental results for different IP₃-receptor subtypes provide dissociation

constants of the activation process from 77 nM (type 3 receptor [11]) to 309 nM (type 1 receptor [12]). We used $d_5 = 0.823 \ \mu\text{M}$ (see next paragraph). An estimate of the value of the rate constant for the transition from the state with Ca²⁺ not bound to the activating site to the state with Ca^{2+} bound to this site, a_5 , can be obtained from puff frequencies as $1 (\mu M s)^{-1}$ [13]. The inhibition process showed an effective dissociation constant between 50 nM at low IP₃ concentrations and 45 μ M at high concentrations in experiments by Mak et al. [14]. We adopted the data by Taylor et al. suggesting an effective dissociation constant of approximately 3 µM for the inhibition process and effective binding rates, a_2 and a_4 , for the binding of Ca^{2+} to the inhibitory site of about $0.2 (\mu \text{M s})^{-1}$ [15]. Finally, the diffusion coefficient D needs a brief consideration. Diffusion of free Ca2+ in the cytosol is limited by the binding of free Ca²⁺ to Ca²⁺-binding proteins, especially buffer proteins. Buffering leads to an effective diffusion coefficient in the range of 40 μ m² s⁻¹. However, buffers saturate at the values of the concentration of free Ca²⁺ in the vicinity of an open channel, and therefore Ca²⁺ diffuses in this area with its own diffusion coefficient of 220–300 μ m² s⁻¹. Since buffers are not included in our model we present results for both D = 40 and 220 $\mu \text{m}^2 \text{ s}^{-1}$.

Dynamic regimes of the model dependent on the IP₃ concentration and d_5 are shown in Fig. 1. There are two saddle node bifurcation lines terminating in a cusp. A Hopf bifurcation occurs above the two saddle node bifurcations. Oscillations can be found at IP₃ concentrations bounded by the Hopf bifurcation and a bifurcation occurring between the Hopf bifurcation and the lower saddle node bifurcation. The type of this bifurcation is still under investigation but is probably homoclinic. Figure 1 demonstrates that oscillations do not occur at values of d_5 suggested by measurements, since the activation process completely saturates at the concentration values occurring at an open channel. Hence, changing dissociation constants of the original DK model to larger experimentally supported values did not restore oscillations. However, even if the cluster dynamics oscillated, these

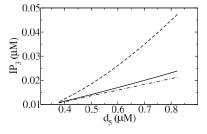


FIG. 1. IP₃ concentration of the saddle node bifurcations (solid and chain dotted lines) and the Hopf bifurcation (dashed line) dependent on d_5 . Parameters read $d_1 = 0.13 \,\mu\text{M}$, $d_2 = 3 \,\mu\text{M}$, $d_3 = 0.9434 \,\mu\text{M}$, $d_4 = 0.4133 \,\mu\text{M}$, $k_p = 80 \,\text{s}^{-1}$, $k_c = 34\,500 \,\text{s}^{-1}$, $E = 750 \,\mu\text{M}$, $a_0 = 0.03 \,\mu\text{m}$, and $D = 40 \,\mu\text{m}^2 \,\text{s}^{-1}$.

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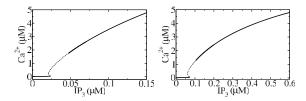


FIG. 2. Stationary values of the Ca²⁺ concentration for $D=40~\mu\mathrm{m}^2\,\mathrm{s}^{-1}$ (left) and $D=220~\mu\mathrm{m}^2\,\mathrm{s}^{-1}$ (right). Solid lines denote linearly stable fixed points and dotted lines linearly unstable points. Parameters as in Fig. 1 and $d_5=0.8234~\mu\mathrm{M}$, $a_2=a_4=0.2~(\mu\mathrm{M}\,\mathrm{s})^{-1}$, $a_5=1~(\mu\mathrm{M}\,\mathrm{s})^{-1}$.

oscillations would not be the experimentally observed ones. To demonstrate this we need to choose parameter values allowing for oscillations. Therefore we use a value of k_c large enough to provide realistic concentration values at the releasing cluster (i.e., larger than 25 μ M [4]), if a large fraction of channels is open, but small enough to still see a variety of dynamic regimes, and a value of d_5 allowing for oscillations (see caption for Fig. 1). Stationary states with these parameter values are presented in Fig. 2. There is just a single stationary state for almost all IP₃ concentrations. An oscillatory regime exists close to the bistable area (see Fig. 2). We did not find limit cycles where the stationary state is stable. Hence, the discrete DK model does not have an oscillatory regime of experimentally relevant extension. Besides the size of the oscillatory regime, there is another observation suggesting that these oscillations are not the experimentally observed global oscillations in cells. Figure 3 shows oscillations of the Ca²⁺ concentration. The initial transient illustrates that realistic concentration values at the cluster are reached for a large fraction of open channels. The amplitude of the oscillations at the releasing cluster is much smaller than the initial peak. It is even more damped down to less than 1 nM in a distance 1.6 μ m from the cluster. This bulk amplitude is too small to represent the observed global oscillations. The oscillation amplitude at the releasing cluster is in the order of magnitude of the dissociation constant of the inhibitory process as is to be expected for such a sinusoidal oscil-

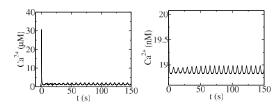


FIG. 3. Oscillation of the Ca^{2+} concentration of $r=0~\mu m$ (left) and $r=1.588~\mu m$ (right). Note the difference in the order of magnitude for the amplitude and the mean. Parameters as in Fig. 2 and $D=40~\mu m^2 s^{-1}$.

lation. This amplitude in the range of dissociation constants—and the small bulk amplitudes—applies to other models than the DK model, too. Hence, not only is the oscillatory regime too small in parameter space to be the experimentally observed regime, but also it is found at unphysiological values. Moreover, the bulk amplitudes are too small.

In summary, the above results strongly suggest that discrete deterministic models of intracellular Ca²⁺ dynamics including no other control of the channel state but activation by IP₃, activation by Ca²⁺, and inhibition by Ca²⁺, do not show the experimentally observed oscillatory regime as already suggested in [2]. Vice versa, the comparison of the bifurcation scheme of the deterministic Li-Rinzel model with simulations of its stochastic counterpart demonstrated that the deterministic regime cannot be concluded from the stochastic behavior [16]. Oscillationlike behavior is reintroduced in intracellular Ca²⁺ dynamics by the fluctuations resulting from the randomness of binding and dissociation of Ca²⁺ and IP₃ at the regulatory binding sites [1-3]. This is also illustrated by array enhanced coherence resonance where stochasticity can induce global oscillations in nonoscillatory systems [17]. Hence, fluctuations drive the spatiotemporal structure formation and render intracellular Ca²⁺ dynamics a truly stochastic medium.

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