## Comment on "Effect of Hydrogen Adsorption on the X-Ray Absorption Spectra of Small Pt Clusters"

In a recent Letter, Ankudinov et al. [1] attributed the origin of the changes observed in our experimental Pt  $L_2$  and  $L_3$  x-ray-absorption near-edge structure caused by H adsorption on supported Pt clusters [2] largely to a H-induced change in the local Pt potential. Since the absorption  $\mu$  equals  $\mu_o(1+\chi)$ , the total change,  $\Delta\mu=$  $\mu(H/Pt) - \mu(Pt)$ , can be expressed as  $\Delta \mu = \Delta \mu_o +$  $\Delta[\mu_o\chi_{\text{Pt-Pt}}] + \mu'_o\chi_{\text{Pt-H}}$ . The three terms include changes in the existent atomic x-ray-absorption fine structure  $(\Delta \mu_o)$  and Pt-Pt scattering  $(\Delta [\mu_o \chi_{\text{Pt-Pt}}])$ , and the additional Pt-H scattering ( $\mu'_o \chi_{\text{Pt-H}}$ ). Ankudinov et al. [1] showed with FEFF8 calculations on octahedral Pt<sub>6</sub> and Pt<sub>6</sub>H<sub>8</sub> clusters (H in threefold sites) that the first two terms dominate in  $\Delta \mu = \mu(Pt_6H_8) - \mu(Pt_6)$ . We interpreted our experimental  $\Delta \mu$  as resulting entirely from the last term.

We do not agree with their conclusions for two reasons: (1) both the  $Pt_6$  and  $Pt_6H_8$  clusters used in their calculations are not representative of our active supported Pt catalysts, and (2) the Pt-H scattering does dominate in FEFF8 results for more representative clusters. The  $Pt_6$  cluster has a large narrow (0.2 eV) surface state falling right at the Fermi level,  $E_f$ , which is not representative of our small (nonmetallic) supported Pt particles, nor of other clusters as confirmed by Pt calculations. Their  $Pt_6H_8$  cluster has the Pt in the threefold sites; however, Pt Kua et al. [3] show that Pt on the model Pt cluster, and on most small clusters at low coverage, will preferentially adsorb in the atop positions.

Figure 1 compares FEFF8 [4] results for their Pt<sub>6</sub>/Pt<sub>6</sub>H<sub>8</sub> clusters (H in threefold hollow sites) [Fig. 1(a)] with those for our more representative Pt<sub>6</sub>/Pt<sub>6</sub>H<sub>6</sub> clusters [Fig. 1(d)]. We give the three contributions and the total difference  $\Delta \mu$ . Figure 1(a) is consistent with their conclusions (i.e.,  $\Delta[\mu_o \chi_{\text{Pt-Pt}}]$  dominates and  $\Delta \mu_o$  is comparable to  $\mu_o'\chi_{\text{Pt-H}}$ ). Figure 1(d) shows results when the H atoms are moved to the atop position and the "artificially" large density of states (DOS) at  $E_f$  has been removed by shifting  $E_f$  upward by 0.2 eV to better approximate the (non*metallic*) small Pt particles in our experiments. The  $\Delta \mu_{o}$ and  $\Delta[\mu_o\chi_{\text{Pt-Pt}}]$  contributions are now drastically reduced. Figure 1 also shows intermediate results when just the H is moved to the atop positions [Fig. 1(b)] or just the Fermi level is shifted [Fig. 1(c)]. The total  $\Delta \mu$ line shape in Fig. 1(d) reproduces most of our adsorbed H [2], S, and O [5] experimental results.

The Ankudinov  $Pt_6/Pt_6H_8$  results are nearly appropriate for H adsorption on bulk Pt (threefold H adsorption and large DOS at  $E_f$ ) and for large metallic clusters. Even in these latter instances, we attributed the entire  $\Delta\mu$  to the  $\mu'_o\chi_{Pt-H}$  contribution [2]. Ankudinov et al. [1] correctly point out that the changing Fano-like profile in

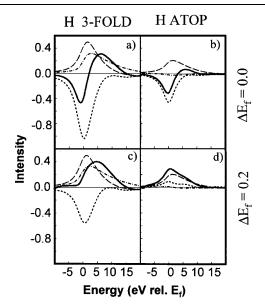


FIG. 1. FEFF8 results for (a) threefold,  $Pt_6H_8$  and (b) atop,  $Pt_6H_6$  adsorbed H. Calculations in (c) and (d) have the Fermi level shifted up by 0.2 eV.  $\Delta\mu = \mu(Pt_6H_x) - \mu(Pt_6)$  (solid lines),  $\mu'_o\chi_{Pt-H}$  (dashed lines),  $\Delta[\mu_o\chi_{Pt-Pt}]$  (dotted lines), and  $\Delta\mu_o$  (dot-dashed lines).

these latter instances arises from the changing relative size of the three contributions. We show here that the final profile depends strongly on the adsorption site, and therefore the claim that the first two terms in  $\Delta\mu$  generally dominate [1] is not justified for supported small Pt particles such as those found in most active catalysts.

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