Nearly Space-Filling Fractal Networks of Carbon Nanopores

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Small-angle x-ray scattering, nitrogen adsorption, and scanning tunneling microscopy show that a series of activated carbons host an extended fractal network of channels with dimension $D_p = 2.8-3.0$ (pore fractal), channel width 15–20 Å (lower end of scaling), network diameter 3000–3400 Å (upper end of scaling), and porosity of 0.3–0.6. We interpret the network as a stack of quasiplanar invasion percolation clusters, formed by oxidative removal of walls between closed voids of diameter of ~ 10 Å and held in registry by fibrils of the biological precursor, and point out unique applications.

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Since the first experimental studies of fractal surfaces of disordered solids [1], it has been conjectured that situations may exist in which the pore space—as opposed to the solid, or the surface alone—is fractal. Such fractal networks of channels crisscrossing the solid, termed pore fractals or "negative image" of mass fractals, have attracted interest as a laboratory for unusual dynamics of confined processes, induced by the long-range correlation of the pore space. The scaling laws predicted relate the dynamic exponents to the fractal and spectral dimensions (space-filling and branching properties) of the network and include anomalous diffusion, reaction, free motion [2], phase transitions [3], electric conduction of pore fluid [4], and hydrodynamic flow [5]. Here we report the first well-documented case of a pore fractal.

The network is a new member in the family of nanostructured carbons. Its channel width is 15-20 Å, comparable to the width of single-wall carbon nanotubes, but instead of forming an assembly of freestanding tubes or bundle of tubes, the channels are embedded in a solid and connected. The network is of multiple interest: (i) Its synthesis differs vastly from that of isolated nanotubes. Created by controlled oxidation, a mainstay of mass production of porous carbons, it promises to be a low-cost competitor of isolated nanotubes for gas storage. (ii) For gas storage, it has outstanding mechanical stability, nanofluidic properties (rapid transport through branched channels), and capacity (high porosity, condensation in high dimensions) compared to nanotube bundles [6]. (iii) It offers a stage for "chemistry in confined spaces" and control of pathways similar to zeolites and other microscopic vessels [7]. (iv) The extended scaling regime, created by what we believe is invasion percolation, provides a unique platform to compare predicted dynamic exponents with experiment.

Sandstones [4] and Vycor glass [8] have been put forth as pore fractals, but these proposals have been controversial or withdrawn [9]. Pore fractals are more challenging to ascertain than mass or surface fractals because they do not reveal their fractality when probed with material yardsticks: the pore-size distribution of a pore fractal is a delta function (mass and surface fractals give a power law), so an intruding nonwetting liquid, capillary condensate, or adsorbed layer will either fill the entire pore space or leave it entirely empty, depending on whether the radius of curvature of the liquid, or adsorbate size, is smaller or larger than the channel. Such an experiment shows the presence of channels, but gives no information about their spatial arrangement.

The porous carbons in this paper are activated carbons [10–13], prepared as follows [12]. Olive stones were carbonized in N₂ at 850 °C, yielding a char denoted by H0. The char was oxidized by H₂O at 750 °C (C + H₂O \rightarrow CO + H₂) at 1% carbon per hour, yielding carbons with mass loss of 8%–74%, denoted by H8–H74. The surface geometry of the samples (ground, dried) was investigated by small-angle x-ray scattering (SAXS). The scattered intensity at momentum transfer q, I(q), was measured on a Bonse-Hart spectrometer for $q < 0.1 \text{ Å}^{-1}$ and a pinhole spectrometer for $q > 0.01 \text{ Å}^{-1}$ [14]. This gives data over an unusually wide range, $0.0003 \le q \le 0.7 \text{ Å}^{-1}$, with one decade of overlap of the two instruments.

To understand the scattering curves and their evolution with progressive oxidation (Fig. 1), we need to consider all three types of fractals. The intensity from a surface/mass/pore fractal obeys the power law $I(q) \propto q^{D_s-2(D_m+D_p)+6}$ [15,16], where D_s , D_m , and D_p is the fractal dimension of the surface, mass, and pore space, respectively. For a surface fractal, $D_m = D_p = 3$, it reduces to the Bale-Schmidt scattering law, $I(q) \propto q^{D_s-6}$

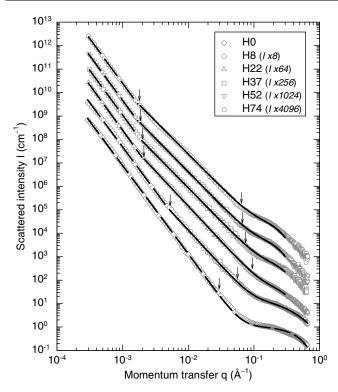


FIG. 1. SAXS curves. The full curves show the fit for the pore-fractal regime, Eq. (2) $[I_0=0$ for H22–H74]. The dashed curves show the fit for the surface-fractal regime [Eq. (1) for H0; $I(q) \propto q^{D_s-6}$ for H8–H74]. Arrows indicate the nominal end of the respective power-law regimes.

[17]; for a mass fractal, $D_{\rm s}=D_{\rm m}$ and $D_{\rm p}=3$, it reduces to $I(q) \propto q^{-D_{\rm m}}$ [18]; and for a pore fractal, $D_{\rm s}=D_{\rm p}$ and $D_{\rm m}=3$, it reduces to $I(q) \propto q^{-D_{\rm p}}$ [15]. Sample H0 exhibits an extended power law, corresponding to a surface fractal with $D_{\rm s}=2.23$ at length scales 2.0×10^2 Å (= $L_{\rm s}$) $\leq 2\pi/q \leq 2.1\times 10^4$ Å [13,19]. At shorter lengths, the fractal structure is masked by scattering from randomly distributed voids in the char (knee in the scattering curve). The two contributions are independent (additive intensities) and fitted with

$$I(q) = \text{const} \times q^{D_s - 6} + I_0 e^{-q^2 L_0^2 / 20}.$$
 (1)

The exponential is the Guinier approximation for scattering from a sphere of diameter L_0 , yielding $L_0 = 10$ Å. The activated samples, H8–H74, show the same surface-fractal structure above 4000 Å, but power laws $q^{-\alpha}$ with $\alpha \leq 3$, characteristic of mass or pore fractals, at shorter lengths. Since the oxidation process "drills" channels into the char [10–13] (confirmed by the adsorption data below), mass fractals are ruled out. The expression for the pore-fractal structure is

$$I(q) = I_{p} \left[1 + (qL_{\min}/2)^{-D_{p}} \right] e^{-q^{2}L_{\min}^{2}/20} + I_{0}e^{-q^{2}L_{0}^{2}/20},$$
(2)

valid for $q \ge 2\pi/L_{\text{max}}$, where L_{min} is the channel width and L_{max} is the network diameter (inner/outer cutoff of the fractal regime, respectively). Equation (2) represents

the network as an "aggregate of spherical pores" (dimension D_p , primary pore diameter L_{min}), with structure factor given by the term in brackets [16,18] and form factor $e^{-q^2L_{\min}^2/20}$, using Babinet's principle. When networks touch, as they may, the overall network is fractal at scales below L_{max} and uniform at scales above L_{max} . The random-void term exists in all samples, but is rapidly overtaken by the network term and is dropped, $I_0 = 0$, in H22-H74 (it can no longer be fitted with significance). The fits yield $D_p = 2.81-3.03$, $L_{min} = 15-20$ Å, $L_{\text{max}} = 1.2 \times 10^3 - 3.4 \times 10^3 \text{ Å (Table I) [20]}$. The external surface undergoes no discernible change (constant D_s [$\alpha \sim 3.8$], masked at large q by scattering from the internal surface [$\alpha \sim 2.9$]). In H8, the network is still small and D_p may be on the high side by an admixture from the power law $I(q) \propto q^{-3.77}$ (likewise in H22). The asymptotic fractal network is reached in H37, with $D_{\rm p}=2.85,~L_{\rm min}=16~{\rm \AA},~L_{\rm max}=3.2\times10^3~{\rm \AA},~{\rm and}$ little change upon further oxidation. The high $D_{\rm p}$ values mark the networks as nearly space filling. It is remarkable that H22-H74 can be fitted with just three parameters at $q > 0.002 \text{ Å}^{-1}$ and span a scaling regime of $L_{\rm max}/L_{\rm min} \sim 200$, comparable to the most extended scaling regimes known [17(a),21].

From these results we can estimate the porosity, $\phi_{\rm SAXS}$, of a block of solid invaded by the network. The ratio of $(L_{\rm max}/L_{\rm min})^{D_{\rm p}}L_{\rm min}^3$ (network volume) to $L_{\rm max}^3$ (network-plus-solid volume) gives

$$\phi_{\text{SAXS}} = (L_{\text{max}}/L_{\text{min}})^{D_{\text{p}}-3}.$$
 (3)

For a network that fills space, $D_{\rm p}=3$, but not uniformly [17(b),17(c)], (3) does not apply. In this case we consider V(L), the volume of the set of all points with distance $\leq L/2$ from the network, taken as a branched curve. The set equals the pore space if $L=L_{\rm min}$, and the space occupied by pores and solid if $L=L_{\rm max}$. Fractal scaling and a correction term for small L lead to $V(L)=V_1(L/L_{\rm min})^{3-D_{\rm p}}+V_2$, where the constants V_1 and V_2 are fixed by the conditions $V(L_{\rm min})=V_{\rm pore}$ (pore volume) and $V'(L_{\rm min})=V_{\rm pore}/L_{\rm min}$ (pore area) [22]. This yields $V(L)/V_{\rm pore}=[(L/L_{\rm min})^{3-D_{\rm p}}-1]/(3-D_{\rm p})+1$ and

TABLE I. Fractal structure from SAXS and N_2 adsorption. The $\phi_{\rm SAXS}$ values for H8 and H22 are calculated from Eq. (4), those for H37–H74 from (3). The $V_{\rm N_2}$ and $\phi_{\rm N_2}$ values are for mass $m_{\rm solid}=1.0$ g and density $\rho_{\rm solid}=2.0$ g/cm³ (amorphous carbon). Error bars include variations in different batches.

	H0	Н8	H22	H37	H52	H74
$D_{\rm s}~(\pm 0.05)$	2.23	2.23	2.12	2.19	2.18	2.18
$L_{\rm s}~(10^3~{\rm \AA})$	0.20	1.4	3.2	4.0	3.7	3.7
$D_{\rm p}~(\pm 0.05)$		3.03	2.99	2.85	2.85	2.81
\dot{L}_{\min} (Å)		17	15	16	20	20
$L_{\rm max} \ (10^3 \ {\rm \AA})$		1.2	3.0	3.2	3.4	3.4
L_0 (Å)	10	8				
$\phi_{\rm SAXS}~(\pm 0.05)$		0.23	0.19	0.45	0.46	0.38
$V_{\rm N_2}~({\rm cm}^3)$	0	0.25	0.39	0.56	0.61	0.84
ϕ_{N_2} (±0.05)	0	0.33	0.44	0.53	0.55	0.63

115502-2 115502-2

$$\phi_{\text{SAXS}} = \frac{V(L_{\text{min}})}{V(L_{\text{max}})} = \left[\frac{(L_{\text{max}}/L_{\text{min}})^{3-D_{p}} - 1}{3 - D_{p}} + 1\right]^{-1}$$
$$\sim \left[\ln(L_{\text{max}}/L_{\text{min}})\right]^{-1}, \tag{4}$$

for $D_{\rm p}=3$ and $L_{\rm max}\gg L_{\rm min}$. The results are $\phi_{\rm SAXS}=0.19$ –0.46 (Table I) [23].

We turn to the nitrogen adsorption isotherms, Fig. 2. The nearly flat high-pressure uptake and weak hysteresis document the narrow range of pore sizes requisite for a pore fractal. By assigning a monolayer of thickness $n^{-1/3}$ at $P/P_0=0.05$ (n: number density of liquid N_2) [24] and a film of thickness $L_{\rm min}/2$ at $P/P_0=0.99$ (full channel) to each of two opposite walls of a channel, we can relate the uptake to the channel width:

$$N(0.99)/N(0.05) = (1/2)L_{\min}n^{1/3}.$$
 (5)

Were the samples mass fractals, the volume uptake at the two pressures would be $\phi_{\rm SAXS}L_{\rm max}^3L_{\rm min}^{-1}n^{-1/3}$ (monolayer on an aggregate with volume fraction $\phi_{\rm SAXS}$ occupied by mass) and $(1-\phi_{\rm SAXS})L_{\rm max}^3$ (pore volume of the aggregate filled with capillary condensate), giving

$$N(0.99)/N(0.05) = (\phi_{\text{SAXS}}^{-1} - 1)L_{\min}n^{1/3}.$$
 (6)

Estimates (5) and (6), evaluated with the SAXS data, predict uptake ratios of 1.9-2.6 and 5-17, respectively. Thus the experimental ratios, N(0.99)/N(0.05) = 1.2-1.9 (Fig. 2), agree with the pore-fractal scenario within a factor of order 1, and disagree with the mass-fractal scenario by a factor of order 10. The widths from (5), $L_{\min} = 9-15$ Å, agree with those from SAXS; they are on the low side because in narrow pores N(0.05) overestimates the monolayer value [24]. The quantities $V_{\rm N_2} =$

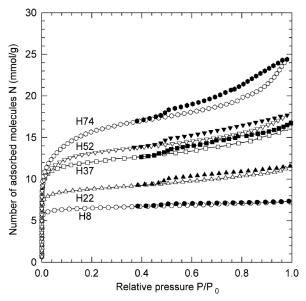


FIG. 2. N_2 adsorption (open symbols) and desorption (full symbols) isotherms at 77 K, with P the vapor pressure and P_0 the coexistence pressure of bulk liquid and vapor. H0 adsorbs N_2 negligibly on the scale of this graph.

N(0.99)/n and $\phi_{\rm N_2} = V_{\rm N_2}/(V_{\rm N_2} + m_{\rm solid}/\rho_{\rm solid})$ estimate the network's volume and porosity (Table I). As they include condensate in external pores [12] (isotherm rise in H37–H74), they are upper bounds. The agreement of $\phi_{\rm SAXS}$ and $\phi_{\rm N_2}$ (mostly within error bars) and compliance with $\phi_{\rm SAXS} \le \phi_{\rm N_2}$ shows that the high porosity, $\phi_{\rm N_2} = 0.33-0.63$, resides mostly in pores of width 15–20 Å.

Since most channels of a pore fractal are deep inside the solid, invisible from the outside, and fill the solid nonuniformly, one expects sparse entrances at the external surface. Scanning tunneling microscopy (STM, Fig. 3) confirms this. The channel width of 13 ± 7 Å for H37, sampled over 42 images, agrees with $L_{\rm min} = 16$ Å from SAXS, sampled over the whole network. So the width at the entrance and in the solid is the same. Selected as an STM-imageable region, Fig. 3 shows little external roughness. Roughness consistent with $D_{\rm s} = 2.19$ at large scales is seen in scanning electron micrographs [11].

What drives the formation of pore fractals in our samples? To answer this question, we return to H0, which has an easily accessible external surface ($D_{\rm s}=2.23$), no open pores in which N₂ would condense, and closed pores with mean diameter $L_0=10$ Å. From $V_{\rm pore}({\rm H0})=V_{\rm N_2}({\rm H8})-0.08 m_{\rm solid}/\rho_{\rm solid}=0.21$ cm³ we find the volume fraction of closed pores in H0 as 0.30, which gives a mean distance of 12 Å between centers of neighboring voids. Oxidation by H₂O removes the weakest wall between the solid's exterior and a nearby closed pore, creating a channel through which new H₂O removes the next weak wall, and so on. This carves out an invasion percolation cluster. A planar invasion percolation cluster has fractal dimension 1.90 (= 1 + $\frac{43}{48}$) [25], and a stack of such clusters in registry has dimension 2.90. This is the model we propose for H8–H74, consistent with

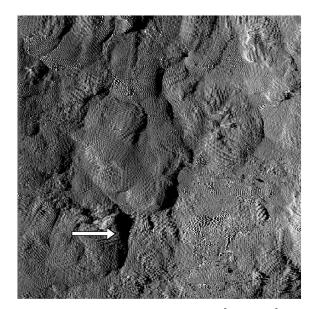


FIG. 3. STM image of sample H37 (320 $\text{Å} \times 320$ Å). Solid black areas are channel entrances. The width of the entrance marked with an arrow is 14 Å, at that location.

115502-3 115502-3

 $D_{\rm p}=2.9\pm0.1$, that $I_{\rm p}$ and N(0.05) grow and I_0 drops with increasing burnoff, that pore mouths lie on parallel fault lines (running vertically in Fig. 3, left half), and that noninvaded regions, presumably pseudomorphs of fibrils of the biological precursor [26], are in registry transverse to the fault planes (Fig. 3, lower right quadrant). Without registry, the stack would be uniform above a few planes and scatter only at $q \sim 2\pi/(\text{spacing of planes})$. model predicts that L_{\min} remains constant and the width of the invaded zone, L_{max} , grows with increasing burnoff, as observed. The growth stops when the accumulation of H₂ deep inside the network slows the removal rate of weak walls down to the removal rate of strong walls at the external surface: the invasion front then advances in unison with the receding external surface, and D_p remains constant because strong walls stay inert until they become part of the external surface (H37-H74).

We expect pore fractals similar to the ones established here to form under a variety of conditions. However, for oxidation of H0 with CO₂, CO₂/H₂O mixtures and at higher temperatures, we find broad pore-size distributions [12] and surface fractals [13]. Most likely, the conditions for "pore aggregation" to yield pore fractals will be as complex as the conditions are for particle aggregation to yield mass fractals.

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115502-4 115502-4