Counterion Release and Electrostatic Adsorption

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The effective charge of a rigid polyelectrolyte (PE) approaching an oppositely charged surface is studied. The cases of a weak (annealed) and strongly charged PE with condensed counterions (such as DNA) are discussed. In the most interesting case of the adsorption onto a substrate of low dielectric constant (such as a lipid membrane or a mica sheet) the condensed counterions *are not always* released as the PE approaches the substrate, because of the major importance of the image-charge effect. For the adsorption onto a surface with freely moving charges, the image-charge effect becomes less important and full release is often expected.

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A deep understanding of the adsorption of DNA or other charged biomolecules onto oppositely charged membranes is of fundamental importance to understand many key physiological processes, and many experimental studies have approached this problem from very different viewpoints [1], strongly motivated by applications to gene therapy [2]. On the theoretical side, the electrostatic adsorption of rigid polyelectrolytes (PE) has been studied within the Debye-Huckel approximation in different situations [3]. Here, we focus on the effective charge of the adsorbed macromolecule. We calculate the attraction energy between an infinitely long, charged cylinder (rigid PE) parallel to an oppositely charged plane, as a function of their distance *h*. The energy variation leads to the determination of the equilibrium charge density of the rod as a function of *h*. This (effective) charge density can be interpreted in terms of the release of condensed counterions for highly charged PE such as DNA, which are beyond the Manning condensation threshold, or in terms of the recombination of ionized charges on the rod for weak (annealed) PE. In contrast to what would be naively expected, the full release of the counterions condensed onto a highly charged rod is not always observed in the vicinity of an oppositely charged surface.

The adsorption energy is derived by perturbating the Gouy-Chapmann solution of the Poisson-Boltzmann equation in a planar geometry. The perturbative treatment is, strictly speaking, valid only for low linear charge densities τ of the rod $l_B \tau \ll 1$, where l_B is the Bjerrum length $l_B = e^2/(4\pi\epsilon)$ (all energies are in $k_B T$ units). However, the physical picture which emerges from this calculation leads to qualitative statements concerning highly charged PE as well. We discuss the adsorption free energy on a substrate of low dielectric constant with respect to water (ϵ_w = 80), which is of most practical importance in biology related problems (adsorption of DNA on a lipid membrane for which $\epsilon_{lp} \approx 2$ [4]) and other situations (adsorption onto the mica surfaces of an SFA: $\epsilon_{mc} \approx 6$ [5]). We have checked that the case of a membrane of thickness $l = 50$ Å and dielectric constant $\epsilon_{lp} = 2$ does not show quantitative differences with the present situation ($l = \infty$) and $\epsilon_{lp} = 0$). Finally, we also study the case where the charges on the plane are free to adjust to the field created by the rod, a situation of great interest for fluid interfaces such as biological lipid membranes.

The electrostatic potential $\phi^{(0)}$ near a charged wall of density $\sigma > 0$ [or equivalently with a Gouy-Chapmann length $\lambda \equiv 1/(2\pi l_B \sigma)$], in a salt solution of average concentration n_0 (or Debye length κ^{-1} with $\kappa^2 \equiv 8\pi l_B n_0$) satisfies the Poisson-Boltzmann (PB) equation [6]. In the low salt limit $\kappa \lambda \ll 1$, which is the case discussed in this paper, the potential near the wall $(\kappa z \ll 1)$ follows the Gouy-Chapmann (GC) law [7]

$$
\phi^{(0)} = -2\log\frac{\kappa(\lambda + z)}{2}, \qquad n^{(0)}(z) = \frac{1}{2\pi l_B(\lambda + z)^2},
$$
\n(1)

where *z* is the coordinate normal to the wall ($z = 0$ at the interface), and $n^{(0)}$ is the density of (negative) counterions (or c-i) near the wall. The Gouy-Chapmann solution predicts a dense counterion layer (the GC layer) of thickness λ containing half the c-i, followed by a diffuse ci region. The local screening length in the GC layer is small, $L_k \approx \lambda/\sqrt{2}$, while it is self-similar in the diffuse region, $L_{\kappa} \simeq z/\sqrt{2}$. The electrostatic potential decreases exponentially for $\kappa z > 1$.

The potential variation $\delta \phi$ due to a negatively charged rod (of linear charge density $-\tau$) located at an altitude $z = h$ above the plane is calculated by a linear expansion of the GC theory, provided that the perturbation is small. A small perturbation of Eq. (1), $\delta n \ll n^{(0)}$, supposes that the perturbation potential $\delta \phi$ is smaller than unity. The calculation of the potential is carried out in Fourier space for the coordinate *x* parallel to the wall and perpendicular to the rod: $\tilde{f}_q = \int dx \, e^{iqx} f(x)$. The linearized PB equation and the boundary conditions are

$$
\partial_z^2 \delta \tilde{\phi} = \left(q^2 + \frac{2}{(z + \lambda)^2} \right) \delta \tilde{\phi}, \qquad \partial_z \delta \tilde{\phi} \big|_{z=0} = 0,
$$

and
$$
\partial_z \delta \tilde{\phi} \big|_{z=h_+} - \partial_z \delta \tilde{\phi} \big|_{z=h_-} = 4\pi l_B \tau
$$
 (2)

with natural boundary conditions for $z \rightarrow \infty$.

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From the perturbed potential $\delta\phi(x, z)$, the free energy due to the presence of the rod can be calculated by a charging process of the rod: $\delta \mathcal{F} = \int_0^{\pi} (\phi_{z=h}^{(0)} + \delta \phi_{z=h}) d\tau$. The first part of the integral gives the interaction energy between the rod and the charged plane and the unperturbed GC counterion layer, while the second part represents the self-energy of the rod and its interaction with the perturbed c-i cloud. The direct interaction energy is

$$
\delta \mathcal{F}_{\rm int} = 2\tau \log \frac{\kappa(\lambda + h)}{2}.
$$
 (3)

The solution of the perturbed PB equation, Eq. (2), leads to the "self-energy": $\delta \mathcal{F}_{\text{self}} = 1/2l_B\tau^2 I_p$, with

$$
I_p = \int_0^{p_M} dp \frac{[1 + p(1 + \bar{h})]^2}{p^3 (1 + \bar{h})^2}
$$

$$
\times \left(e^{-2p\bar{h}} \frac{1 - p + p^2}{1 + p + p^2} - \frac{1 - p(1 + \bar{h})}{1 + p(1 + \bar{h})} \right)
$$
(4)

with $p \equiv q\lambda$ and $\bar{h} \equiv h/\lambda$. The integral cutoff is $p_M \equiv$ $2\pi\lambda/a$ where *a* is a "microscopic" size of order the rod radius ($a = 20$ Å for DNA [8]). This complicated expression can be approximated in the two important limits: for $h \gg \lambda$

$$
\delta \mathcal{F}_{\text{self}} \simeq \frac{1}{2} l_B \tau^2 \left[\frac{2}{3} + \log \left(\frac{4\pi}{3} \frac{h}{a} \right) \right] \tag{5}
$$

and for $h \ll \lambda$

$$
\delta \mathcal{F}_{\text{self}} \simeq \frac{1}{2} l_B \tau^2 \bigg(-\Gamma + \frac{2\pi}{3\sqrt{3}} + \log \frac{\pi \lambda^2}{ah} \bigg). \quad (6)
$$

The self-energy behaves very differently far from and close to the wall. It shows an attraction between the rod and the plane at large distances, which superimposes to the bare attraction between the two oppositely charged macroions. At short distances, there is a strong (logarithmically divergent) repulsion between the rod and the plane [Eq. (6)].

The self-energy shows a deep minimum for a position of the rod: $h_{\text{min}} \approx 0.8\lambda$. It results from a balance between the repulsive image-charge effect and the attraction due to the screening of the ci in the GC layer. The selfenergy of a cylinder in a bath of mobile charges is of order $l_B\tau^2 \log L_K/a$, where L_K is the characteristic ("screening") length of the bath—the range of the electrostatic interactions. Note that although the free ions are mostly of the same sign as the rod, one can speak of a screening effect, as the interaction between the rod and the unperturbed ion cloud is taken into account in $\delta \mathcal{F}_{\text{int}}$ [Eq. (3)]. The self-interaction includes also the interaction with the image charge, which, for $\epsilon_{z<0} \ll \epsilon_{z>0}$, is a virtual rod of the same charge located at $z = -h$ [9]. This gives an extra contribution $\sim l_B\tau^2 \log L_K/h$ per unit length. The large and short distance behaviors Eqs. (5) and (6) can now be explained from the counterion profile Eq. (1), since $L_k \sim h$ for $h > \lambda$ and $L_k \sim \lambda$ for $h < \lambda$.

The adsorption of a polyelectrolyte onto a biomembrane or other fluid membranes, which are generally a mixture

of charged and neutral lipid molecules, involves the movement of surface charges as a response to the field created by the rod. We address this case, disregarding the fact that the lipid bilayer is a flexible object which would, to a certain extent, wrap around the PE [10]. We also assume that the charge reorganization at the interface is not limited by the availability of moving charges on the plane. The charge on the plane follows a Boltzmann law, $\sigma = \sigma_0 e^{-\delta \phi}$. As a result, the boundary condition for the perturbed field on the plane [corresponding to Eq. (2)] has to be modified: $\partial_z \delta \tilde{\phi} |_{z=0} = 4\pi l_B \sigma_0 \delta \tilde{\phi} |_{z=0}$. This affects the self-energy term only, which can be expressed similarly to Eq. (4): $\delta \mathcal{F}_{\text{self}} = 1/2 l_B \tau^2 I_p^{\text{bis}}$ with

$$
I_p^{\text{bis}} = \int_0^{p_M} dp \frac{[1 + p(1 + \bar{h})]^2}{p^3 (1 + \bar{h})^2} \times \left(e^{-2p\bar{h}} \frac{3 - 3p + p^2}{3 + 3p + p^2} - \frac{1 - p(1 + \bar{h})}{1 + p(1 + \bar{h})} \right). \tag{7}
$$

For $\bar{h} > 1$ this self-energy is equivalent to the constant surface charge case [Eq. (5)], while for \bar{h} < 1

$$
\delta \mathcal{F}_{\text{self}} \simeq \frac{1}{2} l_B \tau^2 \bigg(-\Gamma - \frac{\pi}{3\sqrt{3}} + \log \frac{\pi \lambda^2}{6ah} \bigg). \quad (8)
$$

The mobile (positive) charges of the surface are attracted toward the rod $(x = 0)$; this effectively decreases the "Gouy-Chapman" screening length around the PE and reduces the image-charge effect. The case of an annealed surface charge is qualitatively similar to, but quantitatively different from, the case of a quenched surface charge. Since the repulsion of the wall is weakened, the minimum of the self-energy is much deeper, and closer to the wall in the case of moving surface charges.

We now discuss the variation of the line charge of an annealed, or weak, polyelectrolyte near a charged wall. The charges on the chain result from a partial ionization of specific chemical groups. The ionization occurs at chemical equilibrium with the free ions in solution, and is governed by quantities such as the *p*H of the solution [11]. Formally, the PE charge density can be determined by equating the chemical potential of the charges on the chain to a (given) chemical potential μ_0 for the free charges. For an infinitely long rigid PE in a salt solution, the free energy (per unit length) of the charges on the chain is the sum of the translational entropy of the charges along the rod, the electrostatic energy, and the chemical potential: $\mathcal{F} = \tau \log \tau a/e + l_B \tau^2 \log \kappa a$ $\mu_0\tau$. The equilibrium charge density of the rod for a given chemical potential is obtained by differentiation of the free energy: $\mu_0 = \log \tau a + 2l_B\tau \log \kappa a$. The equivalent expression for a charged rod near an oppositely charged plane can be computed from the electrostatic free energy Eqs. (3) and (4): $\mu_0 = \log \tau a + l_B \tau I_p + 2 \log \kappa (\lambda + h)/2$. Because of the minimum in the electrostatic energy, the equilibrium charge density τ is maximum for a finite height of order the GC length, and decreases sharply near the wall.

This nontrivial behavior of the charge density near the wall is mostly due to the importance of the image charge effect in the vicinity of the wall. It is extended further below to the case of highly charged PE with Manning condensation. Note that the optimum charge of the weak PE can reach higher values if the charges on the surfaces are mobile, but still decreases at shorter distances.

The perturbative treatment is expected to fail in the important case of the release of the counterions condensed onto a highly charged rod, as the rod approaches an oppositely charged plane. However, the qualitative argument which translates the concentration of free charges into a local screening length should still hold in this case.

A charged cylinder surrounded by its counterions undergoes the so-called Manning condensation [12]. Solutions of the Poisson-Boltzmann equation in this geometry [13] predict that if the rod is highly charged (namely, $l_B\tau > 1$), a finite fraction $1 - \beta$ of the counterions are confined in the close vicinity of the rod. The electrostatic properties far from the rod are the same as those of a rod with an effective charge $l_B\tau^* \approx 1$. The (over)simplified Oosawa picture of counterion condensation [14] gives a qualitative account of this phenomenon [15]. It is based on a chemical equilibrium between two types of counterions: condensed c-i with a reduced entropy and subjected to a large electrostatic attraction from the rod on the one hand, and free c-i in solution far from the rod on the other hand. This picture can be adapted to the case of a cylinder of charge τ in a salt solution of screening length κ^{-1} . The effective charge of the cylinder, $\tau^* = \beta \tau$, is obtained by balancing the chemical potentials of the condensed c-i $\mu_{\text{cond}} = \log \left[\frac{(1-\beta)\tau v}{\pi a^2}\right] + 2l_B\tau\beta \log \left[\kappa a\right]$ and of the c-i dispersed among the salt molecules $\mu_{\text{free}} = \log[n_0 v]$ (*v* is the volume of a c-i molecule). The resulting fraction of free c-i β is $\log[8l_B\tau(1-\beta)] = (1 - l_B\tau\beta)\log[(\kappa a)^2]$. This result is very similar to the Oosawa relationship for a rod which would occupy a (small) volume fraction f_v = $(\kappa a)^2$ in a salt-free solution [16], namely, $\beta \simeq 1$ for $l_B\tau <$ 1 and $\beta l_B \tau \simeq 1$ for $l_B \tau > 1$.

The counterion condensation on a rod near a charged plane can be derived in the same way. The chemical potential of the condensed counterions depends upon the electrostatic potential on the rod, which can be determined (at the level of the scaling laws) using the calculation of the previous sections. Equating the chemical potentials of the free and condensed ci, we obtain the fraction of free ci β ,

$$
\log[8l_B\tau(1-\beta)] = \log[(\kappa a)^2] - 2l_B\tau\beta\log\left[\frac{ah}{L_{\kappa}(h)^2}\right] + 2\log\left[\frac{\kappa(h+\lambda)}{2}\right],
$$
 (9)

where the local screening length follows the asymptotic behaviors, $L_k = \lambda$ for $0 \le h \le \lambda$ and $L_k = h$ for $\lambda < h < \kappa^{-1}$. Examples of the counterion release as a function of the distance to the wall is shown in Fig. 1 (for which proper screening due to the salt has 4864

been taken into account for $\kappa h > 1$). The Manning parameter $l_B\tau = 4$ and the radius $a = 20 \text{ Å}$ are of order those for DNA, three surface charge densities ranging from $\lambda = 200$ to 50 Å have been considered, corresponding to one charge every $(100 \text{ Å})^2$ to $(50 \text{ Å})^2$. The salt concentration $\kappa^{-1} = 1000 \text{ Å}$ corresponds to a concentration of 10^{-5} mol/liter. While the c-i are released as the rod penetrates the Gouy-Chapmann layer, full release ($\beta = 1$) is reached only for $\lambda = 50$ Å—the maximum being 60% for $\lambda = 200$ Å (it is of the order of 30% for the free rod). Furthermore, the free charges recondense at short distance if λ is large enough; the effective charge reaches 50% of the bare charge for $\lambda = 200$ Å. It can be shown that for large salt concentration or weak surface charge of the wall, $k^2\lambda^3$ < 2*a*, the PE in contact with the wall can have a lower effective charge than the free PE.

It should be noted that the short range repulsion due to the image charge could, in principle, prevent real adsorption between the rod and the plane. In many cases, however, a short range attraction of nonelectrostatic origin (such as a hydrophobic force) dominates near the wall, and must be added to the free energy calculated here in order to study the equilibrium adsorption.

Different conclusions are reached in the case of moving surface charges. A highly charged rod has a strong effect on the surface charge distribution, for the potential it creates is likely to dominate over the Gouy-Chapman potential, even near the wall. A quantitative description of the phenomenon would require the solution of the full nonlinear Poisson-Boltzmann equation, with complex (nonlinear as well) boundary conditions. In the following, we merely try to give a feeling of the way moving charges can influence the counterion release of an adsorbed polyelectrolyte. We assume that the surface charge distribution obeys Boltzmann statistics, $\sigma = \sigma_0 e^{-\phi_h}$, where the potential ϕ_h created by the rod at the surface reflects the screening due to the Gouy-Chapmann layer, the image charge, and the fraction $(1 - \beta)$ of condensed counterions: $\phi_h \sim -2l_B\tau\beta \log(L_K/h)^2$. Since the local

FIG. 1. Fraction β of free counterions as a function of the distance to the wall in λ units [dashed: $\beta(h \to \infty)$] for $a =$ 20 Å, $l_B\tau = 4$, $\kappa^{-1} = 1000$ Å, for decreasing values of the Gouy-Chapmann length $\lambda = 200$, 100, and 50 Å.

screening length $L_k \approx (\lambda + h)/\sqrt{2}$ is influenced by the surface charge σ , which in turn depends upon the screening length, we obtain a self-consistent relationship for the local Gouy-Chapman length near the rod,

$$
\lambda(h) \left(\frac{h + \lambda(h)}{\sqrt{2}h} \right)^{4l_B \tau \beta} = \lambda_0.
$$
 (10)

This expression shows that moving surface charges strongly reduce the screening length near the rod, which becomes of order *h* when the rod is close to the wall (recall that $l_B \tau \beta \simeq 1$). As a consequence, the interaction with the image charge and the self-interaction along the rod are both strongly screened. The dominant interaction between the cylinder and the wall is the attractive part given by Eq. (3), and we are likely to observe a full release of the condensed counterions in this case.

To summarize, we have studied the evolution of the effective charge of a polyelectrolyte near an oppositely charged plane. The case of a weak PE is studied fairly rigorously, via a perturbative treatment of the nonlinear Poisson-Boltzmann equation. We show that at large distance, the charge of the PE increases as the distance to the wall *h* decreases, as expected. However, the charge decreases "strongly" as the PE enters the Gouy-Chapman layer ($h \approx \lambda$) because of the combined effect of image charge (for the most common case of a wall with a low dielectric constant) and self-interaction along the PE. These two effects are very much influenced by the value of the Gouy-Chapman length λ , and are partly suppressed in the case of a fluid interface with moving surface charges (for a fluid lipid bilayer, for instance), where the charge of the adsorbed PE can reach higher values.

The most interesting case of a strongly charged PE beyond the Manning condensation threshold (such as DNA) is discussed qualitatively, using "scaling" arguments inferred from the perturbation theory. We predict that in the case of fixed surface charges, and in contrary to a widely spread idea, most of the condensed counterions are not released if the Gouy-Chapman length is larger than the radius of the rod, $\lambda \gg a$. In the case of freely moving surface charges, a full release of the condensed counterions is expected, as the effective Gouy-Chapman length near the rod is of order the rod radius.

In all the discussion, we have assumed that the Gouy-Chapman length is larger than a molecular size. In many real cases, the two lengths are of the same order of magnitude and the finite size of the ions must be taken into account in order to obtain quantitative results. In this case our results can at best be considered as qualitative.

To conclude, we emphasize that our theory suffers the limitations inherent to the Poisson-Boltzmann theory, the most serious of which is that it does not include lateral correlations between condensed counterions. These correlations are expected to be important in the case of multivalent counterions, and are known to modify the mean field

picture [17]. Studies are under way to address this important issue.

Some predictions of our work could be tested experimentally. Although a direct determination of counterion release is not an easy task, comparisons between the theory we describe and experiments are possible for weak polyelectrolytes, for the ionization of these polymers can be measured by means of Fourier transform infrared spectroscopy. Experiments using this technique are currently being performed on macromolecules near charged surfaces [18].

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