## Solid Molecular Hydrogen: The Broken Symmetry Phase

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(Received 20 December 1996)

By performing constant-pressure variable-cell *ab initio* molecular dynamics simulations we find a quadrupolar orthorhombic structure of  $Pca2_1$  symmetry for the broken symmetry phase (phase II) of solid H<sub>2</sub> at T = 0 and P = 110-150 GPa. We present results for the equation of state, lattice parameters, and vibronic frequencies, in very good agreement with experimental observations. Anharmonic quantum corrections to the vibrational frequencies are estimated using available data on H<sub>2</sub> and D<sub>2</sub>. We assign the observed modes to specific symmetry representations. [S0031-9007(97)02938-4]

PACS numbers: 62.50.+p, 64.30.+t, 78.30.-j

The quest for the structure of the high-pressure phases of hydrogen is a long standing one. Early predictions of an insulator-metal transition [1] led to a large body of experimental and theoretical work during the past 60 years. Metallization was not found as promptly as initially expected [2] but a rich phase diagram emerged. The current picture of the low and room-temperature phase diagram up to ~230 GPa is essentially based on the optical studies performed in diamond anvil cell (DAC) devices during the past decade [3,4].

There is a consensus that hydrogen exhibits at least three different solid molecular phases. (1) At low pressures (<110 GPa for para-H<sub>2</sub>) the centers of the H<sub>2</sub> molecules crystallize into an hcp lattice, but zero-point motion overcomes rotational energy barriers, leading to a free-rotator phase (phase I). (2) Between 110 and 150 GPa intermolecular interactions freeze the molecular rotations into an ordered broken-symmetry phase (BSP, or phase II). (3) Above  $\sim 150$  GPa, a third phase (H-A, or phase III) is attained, whose structure is unknown. Here we focus specifically on phase II. Optical measurements in phase II indicate the presence of two, possibly three, infrared (IR) active modes, in constrast to phase I, where only one mode is observed. Raman spectra show a single peak, at a frequency  $\sim 10\%$  lower than that of the IR modes. A consistent picture of the structural and dynamical properties of phase II is still lacking [5].

On the theoretical side, most of the existing work consists of static total energy calculations within the local density approximation (LDA). Zero-point energy (ZPE) of the protons has been in a few cases included *a posteriori* based on frozen phonon calculations [6,7]. Hexagonal close packed structures with two and four molecules per unit cell appear to be the strongest candidates for the ground structure of phase II, but the relative orientation of the molecules is uncertain [8–10]. Cubic structures were also suggested [9]. This uncertainty persists, due to the incomplete optimization of the lattice parameters and the *a priori* selection of a particular space group symmetry in static calculations. *Ab initio* molecular dynamics simulations, which do not rely on the choice of a specific space group, have also been reported [11,12]. However, these earlier attempts were hampered by a fixed simulation cell and a poor Brillouin zone (BZ) sampling.

We report extensive *ab initio* molecular dynamics simulations of solid H<sub>2</sub> in the 110–150 GPa range. Additional crucial ingredients are (i) a variable cell, constant-pressure approach that largely eliminates prejudice on the space group symmetry, molecular orientations, and lattice constants [13]; (ii) fully converged BZ sampling, achieved by a freshly implemented  $k \cdot p$  technique [14]. Electronic correlations are treated within the LDA supplemented with Becke-Perdew gradient corrections [15], while the proton-electron interaction is described through a local pseudopotential [16] requiring an energy cutoff of 60 Ry [17]. We stress that electrostatic and band energies are fully included in this approach [10].

Preliminary runs using large simulation cells (up to 128 atoms) with  $\Gamma$ -point sampling only produced layered ground state structures with in-plane triangular ordering of molecules. However, we found a strong dependence of the results on cell shape and atom number, confirming that an accurate BZ sampling is crucial [10,18]. Convergence on BZ sums for insulating molecular H<sub>2</sub> turned out to require at least  $8 \times 8 \times 8 k$  points in the full BZ of a four-molecule cell [10]. This is particularly demanding in our approach because of the lack of symmetry constraints. Thus, at variance with the standard approach where the electronic orbitals for each k point in the BZ are expanded on a generic basis set (e.g., plane waves), we adopt a  $k \cdot p$  expansion in terms of the (occupied plus empty) self-consistently generated  $\Gamma$ -point orbitals,

$$u_i^{\mathbf{k}}(\mathbf{r}) = \sum_{i=1}^N a_{ij}^{\mathbf{k}} u_j^{\mathbf{0}}(\mathbf{r}), \qquad (1)$$

where  $u_i^{\mathbf{k}}(\mathbf{r})$  is the periodic part of the *i*th Bloch function at wave number  $\mathbf{k}$ , and  $a_{ij}^{\mathbf{k}}$  are unitary matrices obtained by direct diagonalization the  $k \cdot p$  Hamiltonian at  $\mathbf{k}$  [19]. This expansion is strictly equivalent to the standard approach when the number of states N equals the size of the basis set [20]. In practice the sum can be truncated at a significantly smaller number of  $\Gamma$ -point orbitals with excellent accuracy [14]. With this improved functional we studied a system of 8 atoms with 512 k points in the full BZ [21], and with N = 64. We also performed simulations with 32 atoms, 128 k points, and N = 96, and with 64 atoms, 64 k points, and N = 128.

The 8-atom simulation started from the  $Pmc2_1$  structure proposed by Kaxiras and Broughton [8] (see Fig. 1) that is the most favorable among a class of structures containing also the P2/m. Pressure was set to 140 GPa. When constant-pressure dynamics was switched on, the system spontaneously transformed into Kitaigorodskii and Mirskaya's quadrupolar structure  $Pca2_1$  [22], proposed for H<sub>2</sub> by Nagara and Nakamura [9] (see Fig. 1). We then performed additional runs starting with different initial structures, namely, the cubic Pa3 and the orthorhombic *Cmca* (see Fig. 1), previously proposed for solid  $H_2$ at lower and higher pressures, respectively [4,18]. Although these structures can be ruled out for phase II on the basis of the number of vibronic modes [5], they represent reasonable starting points for our dynamical structural search. In the range of existence of phase II, the *Cmca* structure turned out to be unstable against  $Pca2_1$ , while Pa3 was found to be locally stable, but much higher in enthalpy than  $Pca2_1$ . The quadrupolar structure of



FIG. 1. Candidates for the ground state structure of solid  $H_2$  in the broken symmetry phase (phase II), projected along the *c* axis. Black (gray) arrows represent molecules centered on the *c* (*c*/2) plane and pointing towards the positive-*z* hemisphere.

symmetry  $P2_1/c$  [9] was also considered and found to be locally stable, with an enthalpy 0.015 eV/molecule higher than that of the  $Pca2_1$  (at 140 GPa). The simulations were repeated with supercells of 32 atoms (double size along the two in-plane directions), and 64 atoms (double size in all directions). However, we failed to observe any signature of ground state structures with unit cells larger than 8 atoms, and  $Pca2_1$  prevailed in all cases.

The above results have been obtained by treating the protons as classical particles, since quantum effects have been proven irrelevant to the relative hierarchy of molecular phases [23]. However, given the high level of accuracy of the present calculations, it seems wise to include the quantum effects. By rotating a single molecule on the  $Pca2_1$  structure [24] we do not find multiple energy wells, suggesting that tunneling (of high order in  $\hbar$ ) should not be important, at least near 150 GPa. Quantum corrections were therefore evaluated a posteriori within the harmonic approximation (i.e., to first order in  $\hbar$ ). To this aim, we generated molecular dynamics trajectories at very low temperature ( $T \approx 2$  K), from which we extracted [25] the frequencies of the 21 modes (4 vibrons, 9 phonons, and 8 librons) corresponding to the  $\Gamma$  point of the 4-molecule BZ. The ZPE, computed by summing over the above modes, was evaluated only for structures  $Pca2_1$  and  $P2_1/c$ , the closest competitors for phase II. In fact, a Pa3 space group is incompatible with experimental data [5], while all remaining structures are unstable in the clamped-nuclei approximation. The instability implies the existence of a distortion with negative clamped-nuclei energy curvature, and excludes stabilization through quantum corrections, to first order in  $\hbar$ . The ZPE turned out to be lower in  $P2_1/c$  by only 0.003 eV/molecule with respect to  $Pca2_1$ , yielding a quantum-corrected enthalpy difference of 0.012 eV/molecule, still in favor of the classically determined  $Pca2_1$  structure.

It is worth stressing that in  $Pca2_1$  the centers of the  $H_2$  molecules occupy the sites of a slightly distorted *hcp* lattice. The temperature-induced transformation of phase II into phase I should then be accompanied by orientational disordering and by full recovery of the *hcp* symmetry. Thus, the equation of state (EOS) of phases I and II are expected not to be significantly different. In Fig. 2 we report our EOS and the c/a ratio calculated for the  $Pca2_1$  structure in the range from 100 to 150 GPa. The b/a ratio turns out to be essentially pressure independent, although slightly smaller ( $b/a \approx 1.715$ ) than the *hcp* value of  $\sqrt{3}$ . Comparison with the experimental results for the EOS and the c/a ratio in phase I [26] (full line) is also rather satisfactory.

A structure with  $Pca2_1$  symmetry is compatible with optical measurements on phase II, where a strong Raman peak, and two IR modes are observed [3,5]. A symmetry analysis of the four vibronic modes of the  $Pca2_1$  structure, usually labeled as  $A_1$ ,  $A_2$ ,  $B_1$ , and  $B_2$  [5,27], shows that all of them should be Raman active, and three of



FIG. 2. Calculated equation of state of  $H_2$  in phase II: 8atom simulation cell (solid circles), and 64-atom simulation cell (open squares). The experimental EOS of  $H_2$  in phase I [26] is also reported (solid line). The inset shows the pressure dependence of the c/a ratio (triangles are data from [26], other symbols as above).

them  $(A_1, B_1, \text{ and } B_2)$  IR active. The frequencies of these four modes were evaluated by decomposing molecular dynamics trajectories at 100 and 150 GPa into the symmetry representations of space group  $Pca2_1$  [28]. The resulting frequencies turned out to be substantially higher than experiment. Moreover, they *increased* as a function of pressure, in agreement with the calculated reduction of the molecular bond length (see Table I), but in flagrant disagreement with the observed trends, particularly for the Raman mode.

A closer glance at the experimental data, however, indicates that quantum effects are far from negligible (see Fig. 7 of Ref. [3]). In fact, the D<sub>2</sub> Raman frequency (scaled by  $\sqrt{2}$ ) is roughly 200 cm<sup>-1</sup> higher than that of H<sub>2</sub>. To first order in  $\hbar$ , quantum corrections to the frequency are known to scale as the inverse of the particle's mass, no matter what the potential is [29]. Thus, using the experimental H<sub>2</sub> and D<sub>2</sub> frequencies, we obtain empirical but accurate quantum corrections to the classical values following Ref. [30]. In the case of the H<sub>2</sub> Raman mode the correction is as large as 500 cm<sup>-1</sup>, while

TABLE I. Bond length, volume, and band gap in the two quadrupolar structures  $Pca2_1$  and  $P2_1/c$ , as a function of pressure.

	P (GPa)	$d_{HH}$ (a.u.)	$V (cm^3/mole)$	$E_g$ (eV)
$Pca2_1$	100	1.377	2.75	4.12
$Pca2_1$	150	1.373	2.35	2.53
$P2_{1}/c$	150	1.374	2.35	2.61

for the IR modes the effect is more modest, amounting to about 150 cm<sup>-1</sup>. The resulting quantum-corrected vibron frequencies are reported in Fig. 3, and compared with experimental data on H<sub>2</sub> [3]. The agreement with experiment is now very good, and shows that the Raman vibron turnover is a purely quantum effect. Moreover, the agreement allows us to interpret the experimental Raman mode as our  $A_1$  mode: a totally symmetric, in-phase vibration of the four molecules in the unit cell. According to its symmetry classification, this mode should also be IR active. It does not appear in the experimental IR spectra, possibly due to a small oscillator strength. This is reasonable for a mode where all molecules oscillate in phase, since IR activity can only come from a small intramolecular asymmetry.

We attribute the two IR vibrons to the  $B_1$  and  $B_2$ modes. The predicted splitting of the two IR vibrons turns out to be in fairly good agreement with the data. Finally, we predict a Raman vibron, of symmetry  $A_2$ , with a "classical" frequency similar to that of the IR modes, being similarly an out-of-phase oscillation of the four molecules. Applying the same type of quantum correction as for the IR modes, the  $A_2$  should in principle be observed in the region around 4500  $\text{cm}^{-1}$ . Failure to observe modes  $A_2$ ,  $B_1$ , and  $B_2$  in Raman spectroscopy could be attributed to their small Raman strength, contributions from out of phase vibrations canceling out. The observed Raman mode could alternatively be attributed to mode  $A_2$ , but this would imply a quantum-corrected frequency of 4370  $\text{cm}^{-1}$  (at 150 GPa), in clear disagreement with experiment. Symmetry lowering arguments and larger unit cells have been very recently invoked to explain a larger



FIG. 3. IR and Raman vibron frequencies of  $H_2$  as a function of pressure. The lower and the couple of upper solid lines are the experimental Raman and IR data [3,5], respectively. Circles are calculated Raman vibrons (mode  $A_1$ ). Squares and hexagons are calculated IR active vibrons (modes  $B_1$  and  $B_2$ , respectively). Open triangles are estimates for the new  $A_2$ , Raman-only vibron.



FIG. 4. Electronic band structure (left panel) and density of states (right panel) of solid  $H_2$  in the  $Pca2_1$  structure, at P = 100 GPa. The energy gap is 4.12 eV. The density of states at 150 GPa is also reported (dashed line, right panel).

number of optical modes by band folding in ortho-para  $D_2$  mixtures [31]. The ortho-para distinction is, however, not included in our calculation.

Finally, our calculation provides the full electronic structure and pressure coefficients, and in principle the optical properties. In Fig. 4 we report the band structure and the electronic density of states of the  $Pca2_1$  structure, at P = 100 GPa (the density of states at 150 GPa is also reported, for comparison). A gap of about 4 eV opens up the otherwise fairly free-electron-like bands. As shown (see also Table I), the gap decreases monotonically with pressure but is always finite and large in the range of stability of phase II.

In summary, by means of extensive and accurate constant-pressure *ab initio* molecular-dynamics simulations we have obtained what appears to be a definitive description of the broken symmetry phase of  $H_2$ . The *Pca2*<sub>1</sub> structure, first described by Kitaigorodskii and Mirskaya [22] for the ground state of a classical quadrupolar system, and later proposed for  $H_2$  [9], is favored. Equation of state and vibronic frequencies turn out to be in very good agreement with available experimental data.

This work was partly sponsored by INFM project LOTUS, and by the European Commission under Contract No. ERBCHRXCT 940438.

- E. Wigner and H. B. Huntington, J. Chem. Phys. 3, 764 (1935).
- [2] S.T. Weir, A.C. Mitchell, and W.J. Nellis, Phys. Rev. Lett. 76, 1860 (1996).
- [3] H.K. Mao and R.J. Hemley, Rev. Mod. Phys. **66**, 671 (1994), and references therein.
- [4] I.F. Silvera, in *Simple Molecular Systems at Very High Pressure*, edited by A. Polian, P. Loubeyre, and N. Boccara (Plenum, New York, 1989).

- [5] L. Cui, N.H. Chen, and I.F. Silvera, Phys. Rev. B 51, 14987 (1995).
- [6] T. W. Barbee III, A. Garcia, M. L. Cohen, and J. L. Martins, Phys. Rev. Lett. 62, 1150 (1989); T. W. Barbee III and M. L. Cohen, Phys. Rev. B 44, 11563 (1991).
- [7] M. P. Surh, T. W. Barbee III, and C. Mailhiot, Phys. Rev. Lett. 70, 4090 (1993).
- [8] E. Kaxiras, J. Broughton, and R.J. Hemley, Phys. Rev. Lett. 67, 1138 (1991); E. Kaxiras and J. Broughton, Europhys. Lett. 17, 151 (1992).
- [9] H. Nagara and T. Nakamura, Phys. Rev. Lett. 68, 2468 (1992).
- [10] I.I. Mazin and R.E. Cohen, Phys. Rev. B 52, R8597 (1995).
- [11] D. Hohl et al., Phys. Rev. Lett. 71, 541 (1993).
- [12] J.S. Tse and D.D. Klug, Nature (London) 378, 595 (1995).
- [13] P. Focher et al., Europhys. Lett. 36, 345 (1994).
- [14] S. Scandolo and J. Kohanoff (unpublished).
- [15] A. D. Becke, Phys. Rev. A 38, 3098 (1988); J. P. Perdew and Y. Wang, Phys. Rev. B 33, 8822 (1986).
- [16] P. Giannozzi (unpublished). See also F. Gygi, Phys. Rev. B 48, 11 692 (1993).
- [17] The calculated clamped-nuclei  $H_2$  bond length and frequency are 0.752 Å and 4290 cm<sup>-1</sup>, respectively. The exact value obtained within the Born-Oppenheimer approximation, i.e. corrected for anharmonic quantum effects, is 4050 cm<sup>-1</sup>, to be compared with the experimental gasphase value of 4160 cm<sup>-1</sup>.
- [18] B. Edwards, N.W. Ashcroft, and T. Lenosky, Europhys. Lett. 34, 519 (1996).
- [19] W. A. Harrison, Solid State Theory (McGraw-Hill, New York, 1970).
- [20] I. J. Robertson and M. C. Payne, J. Phys. Condens. Matter 2, 9837 (1990).
- [21] H.J. Monkhorst and J.D. Pack, Phys. Rev. B 13, 5188 (1976).
- [22] A. I. Kitaigorodskii and K. V. Mirskaya, Sov. Phys. Crystallogr. 10, 121 (1965).
- [23] V. Natoli, R. M. Martin, and D. M. Ceperley, Phys. Rev. Lett. 70, 1952 (1993).
- [24] This was done using a supercell containing eight molecules such that periodic images of the rotating molecule are not nearest neighbors.
- [25] J. Kohanoff, Comput. Mater. Sci. 2, 221 (1994).
- [26] P. Loubeyre et al., Nature (London) 383, 702 (1996).
- [27] G. Turrell, *Infrared and Raman Spectra of Crystals* (Academic Press, London, 1972).
- [28] A constant upward shift of 110 cm<sup>-1</sup> has been added to all frequencies, so as to reproduce exactly the zero-pressure value [17].
- [29] M. I. M. Scheerboom and J. A. Schouten, Phys. Rev. B 53, R14 705 (1996).
- [30] See expression (8) in N.W. Ashcroft, Phys. Rev. B 41, 10 963 (1990). For the isolated molecule, we find that this type of correction accounts for 90% of the exact one.
- [31] A.F. Goncharov et al., Phys. Rev. B 54, R15 590 (1996).