Anomalous Behavior of Isotropic Raman Line Shapes near Gas-Liquid Critical Points

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Temperature dependencies of Raman linewidths $FWHM(T)$ of the two polarized modes in N₂O have been determined along the coexistence line with mK resolution. Comparing the neat fluid with diluted states of N₂O in Xe (and CO₂), quite opposite behaviors of FWHM (T) have been observed when the liquids approach their respective critical temperatures T_c : While the mixture $(N_2O)_{0.05}$ -Xe_{0.95} shows the well known λ -shaped effect of critical broadening, a unique and sharp density-correlated line narrowing has been observed in the neat state. The gradual switchoff of intermolecular resonance couplings with decreasing density is responsible for this narrowing phenomenon. [S0031-9007(96)01274-4]

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Isotropic Raman line shapes (IRLS) $J(\omega)_{\text{iso}}$ may show additional (Gaussian) broadening if the liquid-gas critical point $T_c^{(lg)}$ is reached [1,2]. This so-called critical line broadening (CLB) results in a T_c -peaked and nearly λ -shaped temperature dependence of the line width $FWHM(T)$ (full width at half maximum), which is most pronounced if the fluid follows a path along the critical isochore ($\rho = \rho_c$) [3]. The common interpretation assigns CLB, which is also expected to occur near the consolute point $T_c^{(ll)}$ of a liquid mixture [4,5], to those modes whose instantaneous frequencies $\omega(t) = \langle \omega \rangle + \delta \omega(t)$ are modulated by fluctuations of order parameters (density ρ , concentration *x*) [6,7]. Thus, CLB might reflect the universal power law behavior of three-dimensional thermodynamic systems near T_c (see [8] for a review). The most significant example of CLB is given by the width of N_2 , which increases eightfold compared to its low noncritical background value [9,10]. However, it remains an open question whether CLB is universal or not. In addition to N_2 it has also been observed (counting also minor effects) in O_2 [2], HD and H₂ [3], CO [9], CO₂ (mainly ν_1) [11], CH₄(ν_1) [9,12], and H₂S (ν_1) [13], whereas CLB was undetectable in CF₄(ν ₁) [9] and C₂H₆(ν ₃) [14]. Furthermore, the CLB searches around $T_c^{(ll)}$ are contradictory, i.e., either a null effect (3-methylpentanenitroethane) [4] or line broadening by concentration fluctuations (He-N₂, H₂) [5] were reported. Near T_c either line broadening or a total insensitivity to critical peculiarities was observed, while line narrowing has never been presented. To elucidate whether CLB might possibly be overlooked by an insufficient approach to T_c , we studied $J(\omega)_{\text{iso}}$ of the two polarized modes (ν_1, ν_3) in N₂O in closest vicinity to T_c (i.e., $|\Delta T^*| = |T/T_c - 1| \ge 10^{-5}$) along the critical isochore $\rho/\rho_c - 1 \equiv \Delta \rho^* \approx 0$. In a dense state of oscillators intermolecular couplings of both resonant and nonresonant nature contribute to line broadening [15,16]. We therefore compared the neat fluid with

diluted solutions of N₂O in Xe and CO₂ [17] in order to judge the influence of the couplings on CLB. Solution in a monatomic environment (Xe) removes all of them and additionally narrows the line due to the more isotropic environment [18], whereas isotopic dilution suppresses the resonant contribution alone (here the expensive isotopomers of N_2O are surrogated by CO_2 [19]).

The Raman line shapes were measured with multichannel detectors on either a 50 cm focal length triple monochromator Dilor XY (Salzburg, henceforward marked with XY) (v_3^{neat}) or a 100 cm double instrument Jobin Yvon U1000 (Jena, marked with U) (all other experiments). 180° (XY) and 90 $^\circ$ (U) scattering geometries were used. In both setups, a kernel-in-shell principle (e.g., see [14]) was used for temperature stabilization; i.e., a fine tuned and weakly heated copper kernel containing the sample was surrounded by a more coarsely controlled aluminum shell. Maintenance of $T_{ke} - T_{sh} \le 200$ mK, rms fluctuations $|\delta T_{\text{ke}}^{XY}| \le 1 \text{ mK}$ and $|\delta T_{\text{ke}}^{U}| \le 50 \mu\text{K}$ could be achieved over periods of weeks, as was checked with various Pt 100 resistors and thermistors [20]. The improved stability in the U experiments came from $|\delta T_{\text{sh}}^{U}| \lesssim 5 \text{ mK}$ compared to $|\delta T_{\text{sh}}^{XY}| \lesssim 100 \text{ mK}$ and a tenfold greater mass of the kernel ($m_{\text{ke}}^U = 2.8$ kg). To improve heat conduction, the fluid in the U experiment $(V^{U} \approx 2.1 \text{ cm}^{3})$ was in immediate contact with the (gold-coated) copper, whereas in the XY experiment a sealed thin-walled glass tube was used as the primary container ($V^{XY} \approx 2$ cm³), fitted firmly into the copper. In both experiments the trial-and-error filling method of [14] was adapted to match the critical density $\Delta \rho^* = 0$ within 1%. Impurities were <0.3% N₂ [21]. Usually, after reaching the nominal temperature within $|\delta T| \leq 1$ mK, the samples were allowed to equilibrate for $t_{eq}^U \ge 0.5$ h or $t_{eq}^{XY} \ge 4$ h (because of the intermediate glass wall). Longer equilibration $(t_{eq} > 12 \text{ h}$ in both cases) did not change our results. Additionally, they

FIG. 1. Polarized CCD Raman spectra of neat and diluted states of liquid N_2O . Lower part: Temperature dependence of ν_1 (neat state) near below T_c , upper part: Solvent dependence of ν_3 near 218 K. Aquisition times up to 30 min.

were not influenced when, during *T* adjustment, the laser beam ($\lambda_0 = 514$ nm, $P \le 400$ mW) was (horizontally) directed onto the sample or blocked. Since ν_1 and ν_3 are strongly polarized ($\approx 10^3$: 1), it is sufficient to measure the parallel component for evaluating $J(\omega)_{\text{iso}}$. The shapes (see Fig. 1) were fitted with inverse polynomials, i.e., $J(\omega)_{exp} = \left[\sum_{n=0}^{4} a_n (\omega - \omega_0)^n\right]^{-1}$. Except for the critical region in the $(N_2O)_{0.05}$ -Xe_{0.95} mixture, the *J*(ω) were close to ordinary Lorentzians. In the $(N_2O)_{0.05}$ - $(CO_2)_{0.95}$ mixture we could only determine $\nu_3^{\text{N}_2\text{O}}$ due to strong overlap of $v_1^{N_2O}$ with the tenfold more intense $2v_2^{CO_2}$ mode near 1285 cm^{-1} . Figure 2 summarizes the dependencies of FWHM on *T* in the different chemical environments. When approaching T_c from the two-phase region, in the neat liquid the comparatively broad lines (ν_1, ν_3) *narrow.* The v_3 line in the $(N_2O)_{0.05}$ - $(CO_2)_{0.95}$ mixture also narrows near T_c ; however, the overall effect is of smaller magnitude, whereas the well-known CLB effect becomes clearly evident for both lines in the $(N_2O)_{0.05}$ -Xe_{0.95} mixed fluid.

In Fig. 3, the three data sets showing line narrowing are scaled to their individual T_c to examine in greater detail the behavior in close proximity to T_c . In the neat fluid, any indication of CLB is missed. Our highest quality data (i.e., those of ν_1) limit the *T* range where such a broadening might exist to $-10^{-4} \leq \Delta T^* \leq 10^{-5}$ (it is easier to approach T_c from $\Delta T > 0$). The ν_1 line (in Fig. 3) most clearly shows that even minor contributions of CLB are not present (even an effect as low as 0.05 cm⁻¹ would be sufficient to smooth the sharp edge at T_c [22]).

The FWHM (T) resemble the temperature dependence of the orthobaric liquid density $\rho(T)$ near T_c [21], therefore the underlying broadening mechanism of the IRLS is expected to be highly density correlated. The inset in

FIG. 2. Line narrowing vs critical line broadening of $v_1^{N_2O}$ (\approx 1285 cm⁻¹) and $v_3^{N_2O}$ (\approx 2220 cm⁻¹) in the fluids N_2O , $(N_2O)_{0.05}$ - $(CO_2)_{0.95}$, and $(N_2O)_{0.05}$ -Xe_{0.95}. The curves are scaled by appropriate factors to allow an easy comparison. Widths are corrected for finite slits $(\Delta_{\text{slit}}^{XY})$ Δ_{slit}^{U} = 1.0, 0.5 cm⁻¹). The rms fluctuations in *T* are \leq 1 mK. If not stated otherwise the critical density has been matched within 1%. Note, that each data set is obtained over periods of weeks in cycles of lowering and raising temperature.

Fig. 3 proves this nearly linear interdependence between half-width and density. To account for the crossover from critical to noncritical temperature dependence of thermodynamic properties a power law expansion is usually used [8], e.g., for the reduced density $\Delta \rho^*$ along the coexistence line

$$
\Delta \rho_{\rm exc}^* = \pm A |\Delta T^*| \beta \bigg[1 + \sum_{k=1}^n a_k^{(n)} |\Delta T^*|^{-k\Delta} \bigg],
$$

where $\beta = 0.326(2)$ is the order parameter exponent of the three-dimensional Ising model and $\Delta = 0.51(3)$ is the Wegner correction. With FWHM $\propto \rho$ and $n = 2$ this expansion is sufficient to characterize the dependencies in Fig. 3. Using T_c , A, β , and the $a_k^{(2)}$ as adjustable, and $\Delta = 0.51$ as fixed, parameters, we obtained a critical exponent of $\beta_{\text{FWHM}} = 0.30(5)$ from our highest quality data (ν_1^{neat}) . This result depends somewhat on the choice of *n*. T_c could be fixed in this fit within ± 1 mK. The noisier data sets in Fig. 3 produce $\beta_{\text{FWHM}} = 0.35(10)$. Thus, as should be expected, the power law behavior observed for $FWHM(T)$ is closer to the three-dimensional

FIG. 3. Line narrowing in the range $T_c - 1.5$ K $\leq T \leq$ $T_c + 0.9$ K. The full lines are the result of a Wegner expansion (see the expression inside) with fixed $\Delta = 0.51$. The highest quality data (ν_1^{neat}) result in $\beta = 0.30(5)$. The inset shows the direct correlation between width and density.

Ising-like behavior of $\rho(T)$ rather than to the classical mean-field behavior.

What are the reasons for this strong correlation between width and density? Considering the nature of the comparatively broad IRLS of the polarized modes in dense N_2O , it was argued [19] that resonant coupling (RC) between adjacent molecules should be the most effective of the competing broadening mechanisms [15]. However, this conclusion might have been questionable because the authors in [19] were forced to approximate the thermodynamic state dependencies of the individual mechanisms by theoretical expressions which were not well proven experimentally. To avoid any arbitrariness, we studied in detail the *x* dependencies of $J(\omega)^{\nu_1}$ and $J(\omega)^{\nu_3}$ in liquid mixtures of N_2O with its isotopomers ¹⁵NNO and N¹⁵NO as well as Xe and $CO₂$ [23]. Fortunately, the predictions in [19] were completely confirmed; i.e., chemical as well as isotopic dilution lead to a large line narrowing (see Fig. 1) which has never been observed before in any other liquid. The polarized modes in N_2O are among the rare cases where resonance coupling *dominates* the IRLS of a liquid. Thus, we understand the observed anomalous line narrowing as a switch off of RC, caused by the significant decrease of $\rho(T)$ for $T \leq T_c$. In other words, when RC (as a predominantly two-oscillator process [16]) dominates line broadening, it becomes insensitive to critical fluctuations. The line narrowing observed in the mixture $(N_2O)_{0.05}$ - $(CO_2)_{0.95}$ does not contradict this statement because a residual (nonresonant) interaction between N_2O and $CO₂$ (with a similar density dependence as in the neat system) cannot be excluded.

In Fig. 4 the λ -shaped curves of the $(N_2O)_{0.05}$ -Xe_{0.95} mixed fluid are depicted in a log-log plot to show that $\lim_{\Delta T^* \to \pm 0}$ FWHM (T) = const and $\lim_{\Delta T^* \to \pm 0} \partial \text{FWHM}(T)/\partial T = 0$. It must be stressed that we did not find any indication that $FWHM(T)$ diverges near T_c . At least several among the thousands of individual spectra which were measured (with integration times of 30 s) in the vicinity of T_c over a period of 1 week would have matched, by chance, the critical temperature even better than $|\Delta T^*| < 10^{-5}$. If FWHM(*T*) diverges in this very narrow range, a "sudden" and instantaneous *broadening* of individual spectra would have been observed. This was never the case. The early results in [1] gave first hints that CLB may cross over into a constant width for $10^{-3} \leq |\Delta T^*| \leq 10^{-2}$. Mukamel *et al.* [6] have made reference to these experimental observations to justify the results of their hydrodynamic mode-coupling model which predicts $\lim_{\Delta T^* \to \pm 0} \partial FWHM(T)/\partial T = 0$. Our data in Fig. 4 confirm this proposed flattening of

FIG. 4. Critical line broadening of the ν_1 and the ν_3 mode in the mixture $(N_2O)_{0.05}$ -Xe_{0.95}. Note the log-log plot. The arrows show the slope resulting from a fitting procedure for $|\Delta T^*| > 0.01$, i.e., $|\Delta T| \ge 3$ K.

FWHM($|\Delta T|$) as close as $|\Delta T^*| \ge 10^{-5}$, i.e., well inside the region between T_c and the crossover.

In conclusion, our measurements show (1) that the critical dependencies of the purely environmentally induced broadening are completely different from those caused by intermolecular vibrational (resonance) couplings and (2) that any diverging line broadening can be excluded for $|\Delta T^*| \ge 10^{-5}$, or more precisely, approaching T_c , the isotropic linewidth of uncoupled vibrational states increases regressively up to a finite critical width below $|\Delta T^*| \leq 10^{-3}$.

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- [22] Outliers of the $(N_2O)_{0.05}$ - $(CO_2)_{0.95}$ data set near $\Delta T^* \approx$ -5×10^{-4} in Figs. 2 and 3 are inadequate to indicate CLB.
- [23] D. Keutel, F. Seifert, and K.-L. Oehme (to be published): Even a small shift of the vibrational energy of the surrounding molecules [i.e., $\hbar(\omega_i^{N_2O} - \omega_i^{15} N_2^O) < k_B T$] decreases the efficiency of the intermolecular coupling to a large extent. Furthermore, the dilution in $CO₂$ and the two isotopic dilutions are nearly indistinguishable. This justifies our use of $CO₂$ as a surrogate for the expensive isotopomers ${}^{15}N_2O$ and $N{}^{15}NO$. The concentration dependence can be parametrized according to FWHM $(x_{N_2O}) = A_0 + A_1x - A_2x^2$. For ν_3 and 218 K we obtain A_n {Xe, N¹⁵NO, CO₂} in [cm⁻¹] as $A_0 =$ $\{0.36(6), 4.8(1), 3.98(4)\}, \quad A_1 = \{10.5(3), 6.4(6), 9.1(2)\},\$ and $A_2 = \{1.2(3), 1.7(6), 3.5(2)\}.$