Tight-Binding Molecular-Dynamics Study of Amorphous Carbon

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The structural and electronic properties of amorphous carbon are studied with tight-binding molecular-dynamics simulations. An amorphous carbon structure with 216 atoms obtained from our simulation gives a structure factor $S(Q)$ in very good agreement with the results of neutron scattering from a sputtered a-C sample. We found that the amorphous structure consists of graphitelike fragments embedded in a matrix of both twofold and fourfold coordinated atoms. We also found a small pseudogap in the electronic density of states at the Fermi level.

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Amorphous carbon (a-C) attracts continuing attention because of its importance in technological applications as well as its fascination from a scientific point of view. The ability of carbon to form strong chemical bonds under a variety of coordination numbers (from 2 to 4) endows the noncrystalline phase of carbon with many unique properties, and at the same time makes the system quite complex. Experimentally, most amorphous carbon samples are prepared by evaporation or sputtering of graphite or by mass-selected ion beam deposition (MSIB) techniques [1]. While the a-C samples produced by the evaporation or sputtering techniques are sp^2 rich, the carbon films obtained by the MSIB are found to be dominated by diamondlike sp^3 bonding [1]. Despite the considerable theoretical and experimental efforts [1—11] that have been devoted to the study of this fascinating material, the understanding of its microscopic structure and physical properties is still far from complete. Even the most basic information such as the ratios of twofold, threefold, and fourfold sites are still subject to controversy.

In this paper, we report a study of the structural and electronic properties of a-C using a novel molecular-dynamics scheme which incorporates the electronic structure into molecular dynamics (MD) through an empirical tight-binding Hamiltonian. We focus here on the lower density regime; therefore, the a-C structures presented in this paper should be comparable to those obtained experimentally using evaporation or sputtering preparation techniques (e.g., Ref. [7]). Under a density of 2.20 $g/cm³$, we obtained an a-C network which gives a structure factor $S(Q)$ in very good agreement with neutron scattering measurements recently obtained by Li and Lannin $[7]$ on a sputtered a -C sample. The electronic density of states of the a-C network also exhibits a clear deep minimum at the Fermi level, similar to that observed in experiments [1]. We found that amorphous structure in this density regime consists of graphitelike fragments embedded in a matrix of both twofold and fourfold coordinated atoms. Our observation of twofold atoms in the a-C network provides new insights into the properties of *a*-C.

In our tight-binding molecular-dynamics (TBMD)

scheme, the system is described by a Hamiltonian of the form

$$
H(\lbrace \mathbf{r}_i \rbrace) = \sum_{i} \frac{\mathbf{P}_i^2}{2m} + \sum_{n}^{\text{occupied}} \langle \psi_n | H_{\text{TB}}(\lbrace \mathbf{r}_i \rbrace) | \psi_n \rangle
$$

+ $E_{\text{rep}}(\lbrace \mathbf{r}_i \rbrace)$, (1)

where $\{r_i\}$ denote the positions of the atoms (i) $=1,2,\ldots,N$, and P_i denotes the momentum of the *i*th atom. The first term in (1) is the kinetic energy of the ions, the second term is the electronic band-structure energy calculated from a parametrized tight-binding Hamiltonian $H_{\text{TB}}(\{r_i\})$, and the third term is a short-ranged repulsive energy. In this scheme, the quantum mechanical nature of the covalent bonding among the carbon atoms is taken into account explicitly in the electronic structure which is calculated for each of the atomic configurations in the MD simulations. Details of the tight-binding model are described in Ref. [12]. The accuracy and transferability of this model have been well tested: It describes accurately the properties of crystalline carbon structures with various coordination numbers including the linear carbon chain, graphite, and diamond structures, as well as more complex systems such as liquid carbon and the structures of carbon clusters with sizes ranging from ⁵ to 100 atoms [12-14].

The amorphous structures are generated in our simulations by quenching from the liquid phase. The simulations are performed with 216 atoms in a cubic cell with periodic boundary conditions. The Γ point is used for the electronic calculations. The time step used in our simulaions is 1.05×10^{-15} s. In order to understand the effects of density on the microscopic structure of the amorphous phase, we perform simulations at three different densities of 2.20, 2.44, and 2.69 $g/cm³$. The initial configurations are liquid structures at 5000 K for each given density [15]. The volumes and shapes of the MD cells are fixed during the quenching process. The stochastic temperature control method [16] is used to control the temperature, with a quench rate of about 500 K/ps. The structural and electronic properties of the amorphous structure are analyzed at $T = 700$ K. This temperature is

comparable to the experimental condition of room temperature if one includes the effect of zero point motion of the carbon atoms. After 2000 MD steps (corresponding to 2 ps) of thermalization at $T = 700$ K, another 2000 MD steps without temperature control are used to perform the statistical average for the system.

Figure ¹ shows the simulation results for the structure factor $S(Q)$ and pair correlation function $g(r)$ of our a-C structure with a density of 2.20 $g/cm³$. Our results agree well with the neutron scattering data (dashed line) recently measured by Li and Lannin [7]. Detailed structural analysis shows that this a -C sample contains 80.6% threefold sites, 7.4% fourfold sites, and 12% of twofold sites. The average coordination number is 2.96. The percentage of fourfold sites found in our simulation is in good agreement with the value of $(6.5 \pm 1.5\%)$ estimated by a nuclear magnetic resonance (NMR) study [10] on the same a-C sample used in the neutron scattering measurement [7]. However, the presence of twofold coordinated atoms was not considered in the data analysis of Ref. [7]. In view of the good agreement between experiment and simulation for the structure factor $S(Q)$ and the percentage of fourfold sites, it would be interesting to

FIG. 1. (a) TBMD result of structure factor $S(O)$ (solid line) is compared to that of Ref. [7] (dashed line). The density of the TBMD a -C sample is 2.2 g/cm^3 , while that of the a -C film used in Ref. [71 was estimated to be between 2.0 and 2.44 g/cm³. (b) TBMD result of pair correlation function $g(r)$ for the a -C sample of 2.2 g/cm³.

reanalyze the experimental data [7,10], including the effects of twofold coordination. In fact, an extended xray absorption fine-structure spectroscopy (EXAFS) experiment by Comelli et al. [11] suggested that twofold configurations are necessary to explain the EXAFS data from evaporated and sputtered a-C samples.

In Fig. 2, the atomic arrangement of the amorphous carbon obtained by the present simulation with a density equal to 2.20 $g/cm³$ is plotted. We found that the threefold coordinated atoms in this amorphous structure form several clusters embedded in the matrix of both twofold and fourfold atoms. Side views of Fig. 2 (not shown in this paper) and a quantitative ring statistics analysis [17] show that these clusters are graphitelike sheets of sizes about 10 $\AA \times 10$ Å, containing mostly six- and some fiveand seven-membered rings. Most of the twofold as well as the fourfold coordinated atoms serve as bridges to hold together the graphitelike clusters. This unique feature is also observed in the other two higher-density samples. However, as the density increases (from 2.20 to 2.69 $g/cm³$, the number of fourfold sites increases from 7.4% to 20%, and twofold sites decrease from 12% to 6.5% (see Table I). At the same time, the graphitelike clusters hrink in size while the "bridge" regions expand and are increasingly dominated by fourfold atoms which also tend to form clusters. This picture of graphitelike clusters embedded in a random matrix of disordered twofold and fourfold configurations is similar to the models proposed

FIG. 2. Microscopic structure of the TBMD-generated a-C network. The density of the sample is 2.2 g/cm³. Various coordinated atoms are colored differently (twofold: green; threefold: black and white; fourfold: red). Smaller balls represent the carbon atoms outside the central cell (periodic boundary conditions are used).

TABLE I. The ratios of various coordinated atoms of amorphous carbon obtained by TBMD with 216 atoms. n_c is the average coordination number.

Density (g/cm^3)	Twofold $(\%)$	Threefold $(\%)$	Fourfold $(\%)$	n_c
2.20	12.0	80.6	7.4	2.96
2.44	5.8	80.0	14.2	3.08
2.69	6.5	71.0	22.5	3.16

several years ago by Robertson and O'Reilly based on electronic structure analysis [18] and by Comelli, Stohr, Robinson, and Jark based on EXAFS results [11]. Nevertheless, these models considered either only fourfold coordinated [18] configurations or only twofold coordinated [11] configurations for the random matrix while our simulation showed that both configurations are present with ratios sensitive to the density of the sample.

Although many features of the a-C structure obtained from our simulation agree well with that obtained by the ab initio molecular-dynamics simulation of Galli et al. [2], there are some subtle difterences. Our simulation for a density of 2.20 $g/cm³$ produces 80.6% threefold sites, 7.4% fourfold sites, and 12% twofold sites, whereas the ab initio molecular-dynamics simulation of Galli et al. [2] for a density of 2.0 g/cm³ gives 85% threefold sites, 15% fourfold sites, and no twofold sites at all. The existence of twofold atoms in our simulation should be contrasted to the absence of twofold atoms in the ab initio MD result of Ref. [2]. Moreover, the percentage of fourfold atoms from the ab initio MD seems a bit too high compared with our result as well as the experimental data of Ref. [10]. We note that the energy of linear carbon chain (twofold) with respect to that of diamond (fourfold) and graphite (threefold) from our tight-binding model is 1.20 and 1.22 eV/atom, respectively, which are in excellent agreement with the values of 1.22 and 1.24 eV/atom obtained from converged first-principles local-densityapproximation (LDA) calculations [19,20]. Therefore, the twofold atoms in the a -C network obtained from our simulation are not an artifact of our potential model. Zhu, Chan, and Ho [20] have performed a LDA planewave calculation [using the same Bachelet-Hamann-Schliiter (BHS) potential as quoted in Ref. [2]] and found that the relative energies of the linear chain, diamond, and graphite are highly sensitive to the number of plane waves used in the calculation. They found that the energy difference between the linear chain and the diamond structure is 1.72 eV/atom for using the basis set with a kinetic energy cutoff of 20 Ry, but is 1.26 eV /atom with a cutoff of 35 Ry. They also found that at a 20 Ry cutoff the energy of the graphite structure is higher than that of the diamond structure by about 0.16 eV/atom. The simulation in Ref. [2] was initially performed with a 20 Ry plane-wave cutoff, which is grossly inadequate for

the BHS pseudopotential, and, in particular, it overestimates the linear chain energy by about 0.50 eV/atom (-5000 K) with respect to diamond. Under such conditions, it is natural that the concentration of twofold carbon atoms is seriously underestimated and that of fourfold carbon atoms is overestimated. The authors of Ref. [2] mentioned that they reannealed the a-C structure obtained with the small basis set by using a basis set with a 32 Ry cutoff, and obtained very similar distributions of twofold, threefold, and fourfold atoms. With such a great difference in the relative energies of various coordinated structures, it is quite unlikely that the results for 20 and 32 Ry cutoff simulations can be unchanged. We take this as an indication that the reannealing at 32 Ry in Ref. [2] does not have enough time and/or a high enough temperature for the a -C network to get out of the metastable configurations (which overemphasizes fourfold C atoms and underestimates the number of twofold carbon atoms) "frozen in" by the quenching performed at small cutofI.

Finally, we wish to discuss another interesting aspect: the electronic structure of a -C. Experimentally, it has been shown that most evaporated a -C films exhibit a small optical gap (-0.5 eV) . Whether a theoretical structural model of a -C can produce this optical gap is usually considered as an important criterion for judging the reality of the model. Robertson and O'Reilly [18] performed tight-binding calculations for the structural models proposed by Beeman *et al.* [21] and found no optical gaps in these models. Robertson and O'Reilly [181 argued that in order to open a gap in the π band, it is necessary to have medium-ranged order in the form of graphitelike clusters of dimensions of about 15 A, in the amorphous structure. Figure 3 is a plot of the electronic density of states (DOS) obtained from our simulation at 700 K. The data have been averaged over 2000 configurations. We found that the valence band DOS obtained from our simulation is very similar to the photoemission spectra of Wesner *et al.* [22]. We found a deep

FIG. 3. Electronic density of states of the TBMD-generated a -C sample with a density of 2.2 g/cm³. Note a small pseudogap at the Fermi level $(E=5)$.

minimum in the DOS near the Fermi level, similar to the experimental measurement [1] and *ab initio* MD result [2]. It is interesting to note that the electronic DOS obtained from our three samples of different densities are similar, despite the fact that the dimensions of the graphitelike clusters have been reduced by a factor of 2 when the density increases from 2.20 to 2.69 $g/cm³$. The width of the pseudogap decreases slightly as the density of the a-C sample is increased.

In summary, we have shown that tight-binding molecular dynamics can be used for realistic studies of the structural and electronic properties of amorphous carbon. Our results support the proposals of Comelli et al. [11] and Robertson and O'Reilly [18] that sputtered or evaporated a-C consists of graphitelike regions embedded in a random matrix of disordered twofold [11] and fourfold [18] coordinated atoms. The existence of twofold atoms in the a-C network in this density regime has been overlooked for many years. We hope that our new results will stimulate further experimental as well as theoretical studies.

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