## Competition between Superconductivity and Antiferromagnetism in Cu(2) Layers of $Y_{1-x}Pr_xBa_2Cu_{3-y}M_yO_z$ , M = Fe, Co, Zn

I. Felner, I. Nowik, E. R. Bauminger, D. Hechel, and U. Yaron

The Racah Institute of Physics, The Hebrew University of Jerusalem, Jerusalem 91904, Israel

(Received 9 April 1990)

The antiferromagnetic (AFM) and superconducting (SC) phase transition temperatures of  $Y_{1-x}Pr_xBa_2Cu_{3-y}M_yO_z$ , M = Fe,Co,Zn, were studied by magnetometry and <sup>57</sup>Fe Mossbauer spectroscopy. For  $z \sim 7$ , substitutions at all sites except those in the CuO<sub>2</sub> planes (Pr,Fe,Co) cause SC to disappear at  $x_c = 0.5$  or  $y_c = 0.4$  and AFM to appear above  $x_c$  and  $y_c$ . For  $z \sim 6$ , AFM persists for all x and y values. On the other hand, substitution in the Cu(2) planes (Zn) suppresses both SC at  $y_c = 0.24$  for  $z \sim 7$  and AFM for  $z \sim 6$ . No AFM is seen for  $z \sim 7$ .

PACS numbers: 74.10.+v, 74.60.Mj, 75.50.Ee, 76.80.+y

A central issue regarding high- $T_c$  superconductivity is whether or not magnetic interactions are important in stabilizing the superconducting electronic ground state as proposed in many theories.<sup>1</sup> Evidence for magnetic ordering of Cu(2) layers in YBa<sub>2</sub>Cu<sub>3</sub>O<sub>2</sub> (YBCO) was provided by muon-spin-relaxation  $(\mu SR)$ ,<sup>2</sup> neutronscattering experiments,<sup>3</sup> and Mössbauer spectroscopy,<sup>4</sup> and the oxygen-content phase diagram is now well established. This phase diagram<sup>5</sup> shows that for z > 6.45 the system is orthorhombic, metallic, and superconducting with  $T_c$  decreasing as z decreases. For z < 6.45 the system becomes tetragonal, semiconducting, and antiferromagnetic with a Néel temperature of  $T_N = 420$  K for  $z \sim 6$ . The magnetic structure is composed of strong nearest-neighbor antiferromagnetic coupling of the spins within the CuO<sub>2</sub> planes, with antiferromagnetic alignment of the nearest-neighbor spins in the adjacent planes. The magnetic moments are constrained to lie in the planes. The phase diagram of  $(La, M)_2CuO_4$ , which resembles that of YBCO, also exhibits dramatic behavior such as structural changes, 3D antiferromagnetism, disordered magnetism, and superconductivity as a function of hole doping.<sup>6</sup> In the third class of high- $T_c$  superconductors such as Bi-Sr-Cu-Cu-Ca-O, substition of Y for Ca changes the electronic and magnetic properties and a quite similar phase diagram involving superconductivity and magnetism is obtained.<sup>7</sup> There is no conclusive evidence for coexistence of both phenomena in the same composition.<sup>8</sup> From an experimental viewpoint, it is thus of great interest to study and understand the delicate interplay between these two phenomena.

The question we tried to answer is whether the superconductive-magnetic phase diagram for YBCO is unique and depends only on oxygen concentration, or whether substitution of other elements into the system would lead to similar phase diagrams. The present Letter deals with this interrelation between superconductivity and magnetism in the wide family of compounds  $Y_{1-x}Pr_xBa_2Cu_{3-y}M_yO_z$ , where M stands for Fe, Co, and Zn.

The YBCO phase contains both  $CuO_2$  planes [denoted Cu(2)] and CuO chains [denoted Cu(1)]. We show here

that in oxygen-rich samples, when Y is substituted by Pr and Cu(1) is substituted by Fe or Co, with x or y close to the value of the disappearance of superconductivity, a static antiferromagnetic ordering is induced in the Cu(2) site, in a way which is very similar to that observed in YBa<sub>2</sub>Cu<sub>3</sub>O<sub>z</sub> with the removal of oxygen.<sup>5</sup> For oxygendeficient samples ( $z \sim 6$ ) all the compounds are antiferromagnetically ordered and  $T_N$  changes very little with x or y. On the other hand, in oxygen-rich samples, substitution of Cu(2) by Zn in the CuO<sub>2</sub> planes suppresses superconductivity, but magnetic order is never found. For the  $z \sim 6$  materials,  $T_N$  depends on Zn concentration and antiferromagnetism disappears approximately at the same concentration where superconductivity disappears for the oxygen-rich samples.

The ceramic system  $Y_{1-x}Pr_xBa_2Cu_3O_7$  and  $YBa_2$ - $Cu_{3-\nu}M_{\nu}O_{7-\delta}$  were prepared by conventional methods.<sup>9</sup> X-ray-diffraction studies were performed to ensure the purity of the compounds. The oxygen-deficient samples were obtained by quenching the materials from 900°C to liquid nitrogen. Using the published data on the temperature and oxygen dependence of the oxygen concentration in the YBCO system<sup>10</sup> we estimate the oxygen concentration in the quenched sample as 6.1-6.2. dc susceptibility measurements in low fields were carried out in a PAR vibrating magnetometer to determine  $T_c$  of the samples. In order to determine the magnetic nature of the Cu sublattices, all samples were doped with at least 1% <sup>57</sup>Fe for Mössbauer studies. Mössbauer spectroscopy studies were performed using a conventional constant-acceleration spectrometer and a 100-mCi Rh-Co<sup>57</sup> source. The spectra at various temperatures were least-squares fitted with several subspectra corresponding to the various inequivalent iron sites. Mössbauer studies<sup>4</sup> and differential anomalous x-ray scattering<sup>11</sup> clearly indicate that the Fe atoms occupy predominantly the Cu(1) site, with an increasing fraction of Fe residing in the Cu(2) sites as the total amount of dopant increased. When the Cu(2) sublattice becomes magnetically ordered, it produces an exchange field at the iron located in the Cu(2) sites. The iron nuclei experience a magnetic hyperfine field leading to a six-line pattern in the observed Mössbauer spectra.<sup>4</sup> In a recent paper we have shown that in YBa<sub>2</sub>Cu<sub>3</sub>O<sub>6+ $\delta$ </sub>,  $T_N$  is not affected by the presence of Fe, and that Fe is thus a reliable probe of the magnetic behavior of the Cu(2) sites.<sup>4</sup>

Phase diagram of  $YBa_2Cu_{3-y}Fe_yO_{7+\delta}$ . — Substitution for Cu by Fe results in an orthorhombic-tetragonal structure transition at y = 0.06 - 0.09, and produces drastic changes in  $T_c$ . The solubility limit of Fe in YBCO is y = 0.8. Since Fe ions have a larger valence than Cu, they attract oxygen to maintain charge neutrality. Therefore we assign  $z = 7 + \delta$  and  $z = 6 + \delta$  for oxygenrich and quenched samples, respectively. The value of  $\delta$ is proportional to y (one oxygen atom for two Fe atoms<sup>12</sup>) and is not necessarily the same for z = 7 and z = 6 systems. Measurements by us<sup>9</sup> and others<sup>13</sup> established that  $T_c$  remains constant and equal to 90 K up to the orthorhombic-tetragonal transition and then decreases continuously to less than 4.2 K at y > 0.40. Several groups have measured the variation of  $T_c$  with y and the change of  $T_c$  with y obtained by different groups differs somewhat from group to group. The mean values for the concentration dependence of  $T_c$  with y for Fe are shown in Fig. 1. Upon increasing y further, the compounds become semiconductors and our Mössbauer measurements show that simultaneously antiferromagnetism associated with the Cu(2) ions develops. Some Mössbauer spectra of  $YBa_2Cu_{2.25}Fe_{0.75}O_{7+\delta}$  are shown in Fig. 2. The central part of the Mössbauer spectra in Fig. 2 corresponds to Fe which replaces copper in the Cu(1) site.<sup>4</sup> The well-defined magnetic sextet is attributed to iron which replaces copper in the Cu(2) sites and orders antiferromagnetically. The magnetic hyperfine field at 90 K is 495 kOe. As the temperature is raised, the magnetic splitting decreases and the Néel temperature can be obtained. In Fig. 1 the change of  $T_N$  as a function of y is displayed.  $T_N$  decreases gradually from 390 K for Y=0.75 to 330 K for y=0.50 and then drops sharply to 29 K for y=0.45. No magnetic structures are observed in the superconducting range where y < 0.40. This superconducting-magnetic phase diagram is very similar to that obtained for oxygen-content dependence (z) in YBa<sub>2</sub>Cu<sub>3</sub>O<sub>2</sub>.<sup>5</sup> A phase diagram similar to that shown for Fe in Fig. 1 is obtained for Co replacing Cu.

Figure 1 shows that for the quenched oxygen-poor  $YBa_2Cu_{3-y}Fe_yO_{6+\delta}$  samples,  $T_N$  of the Cu(2) sites depends weakly on y and increases from 410 K for y=0 to 430 K for y=0.75. In contrast to the Mössbauer spectra shown in Fig. 2, in the quenched samples we observe a broadening of the central part of the spectra in addition to the sextet corresponding to iron in the Cu(2) sites.<sup>14</sup> This broadening is attributed to a second independent magnetic ordering of Fe in the Cu(1) site with  $T_N = 210$  K (Fig. 1). Our results differ slightly from those obtained in Ref. 15. Since most of the magnetic Fe ions reside in the Cu(1) site, <sup>4,11</sup> they encourage magnetic ordering on the chain site and dictate its nature. The situ-





FIG. 1. Phase diagram for oxygen-rich YBa<sub>2</sub>Cu<sub>3-y</sub>Fe<sub>y</sub>O<sub>7+ $\delta$ </sub> as a function of Fe concentration.  $T_c$  for the superconducting (SC) phases and  $T_N$  for the antiferromagnetic (AFM) phases are shown. The dashed lines show  $T_N$  values for the quenched samples in both Cu(1) and Cu(2) sites.

FIG. 2. Mössbauer spectra of YBa<sub>2</sub>Cu<sub>2.25</sub>Fe<sub>0.75</sub>O<sub>7+ $\delta$ </sub> at several temperatures. The magnetic splitting, which is attributed to Fe in the Cu(2) sites, decreases with increasing temperature and disappears at  $T_N$ .

ation is different in the Cu(2) sites where the concentration of Fe is only a small fraction of that in the Cu(1) site.

Phase diagram of  $Y_{1-x}Pr_xBa_2Cu_3O_z$ .—It is well known that in the orthorhombic  $Y_{1-x}R_xBa_2Cu_3O_7$  system (R denotes rare earth)  $T_c$  does not depend on x when R is a trivalent rare-earth element.<sup>16</sup> An exception to this behavior is PrBa<sub>2</sub>Cu<sub>3</sub>O<sub>7</sub> which is orthorhombic, yet not superconducting. The valency of Pr is still controversial.<sup>17</sup> Neutron-diffraction measurements show that its oxygen concentration is 7.<sup>18</sup> In the orthorhombic  $Y_{1-x}Pr_xBa_2Cu_3O_7$  system it is assumed therefore that the oxygen concentration remains 7 for the entire system,  $T_c$  decreases sharply with increasing x (Refs. 18-21), and the variation of  $T_c$  (and  $T_N$  for the Pr sublattice) in  $Y_{1-x}Pr_xBa_2Cu_3O_7$  was reported by ourselves<sup>19</sup> and others.<sup>18,20</sup> Clearly,  $T_c$  decreases with increasing x, and for x > 0.5 the compounds are not superconducting. Moreover, using the Mössbauer technique on iron-doped samples, we showed that for x > 0.5 the Cu(2) sites are magnetically ordered.<sup>19</sup>  $T_N$  obtained for x = 0.6 is 230 K and it increases gradually to 300 K for x = 0.8 and to 325 K for x = 1. A similar behavior was also obtained in zero-field muon-spin-relaxation measurements.<sup>21</sup> The disappearance of superconductivity and the appearance of antiferromagnetism in both Cu(2) and Pr sites is probably associated with strong Pr 4f electron hybridization.<sup>19,20</sup> It was also shown<sup>19</sup> that  $T_N$  values for the quenched samples  $(z \sim 6)$  depend weakly on x



FIG. 3. Superconducting-magnetic phase diagram for YBa<sub>2</sub>Cu<sub>3-y</sub>Zn<sub>y</sub>O<sub>2</sub>. Note the absence of magnetic ordering at y = 0.3 in both fully oxygenated (square) and quenched (circle) samples.

and range from 350 K for x = 1 to 420 K for x = 0.

We have shown above the similarity between the phase diagrams of YBa<sub>2</sub>Cu<sub>3</sub>O<sub>2</sub>, <sup>5</sup> YBa<sub>2</sub>Cu<sub>3</sub>- $_{y}M_{y}O_{7+\delta}$  (*M* =Fe,Co) (Fig. 1), and Y<sub>1-x</sub>Pr<sub>x</sub>Ba<sub>2</sub>Cu<sub>3</sub>O<sub>7</sub>, <sup>19</sup> in which the competition between superconductivity and magnetism leads to the appearance of antiferromagnetism in the vicinity of the disappearance of superconductivity.

Phase diagram of  $YBa_2Cu_{3-y}Zn_yO_z$ .—A unique exception to the common behavior discussed above is observed in the YBa<sub>2</sub>Cu<sub>3-v</sub>Zn<sub>v</sub>O<sub>7</sub> system. X-ray<sup>11</sup> and neutron-diffraction<sup>22</sup> measurements indicate that in the Zn-doped system, the crystal structure remains orthorhombic up to y=0.3. Several measurements have shown that divalent Zn ions preferentially substitute Cu in the Cu(2) sites,<sup>22</sup> or distribute roughly equally among Cu(2) and Cu(1) sites.<sup>11</sup> [An equal distribution means that about 70% of Zn enters Cu(2) sites, since the number of Cu(2) atoms in the unit cell is twice that of Cu(1)atoms.] The oxygen concentration up to y = 0.3 remains almost unchanged on introduction of divalent Zn.<sup>12,23</sup> The general result is that substituting small amounts of Zn for Cu strongly inhibits superconductivity and for y = 0.24,  $T_c$  vanishes (Fig. 3). This implies that  $T_c$  is more affected when the doping occurs in the CuO<sub>2</sub> planes than in the CuO chains. Our Mössbauer measurements on 1%-57Fe-doped samples definitely show that also in this system, most of the Fe ions occupy the Cu(1) site and only about 20% of Fe resides in the Cu(2) sites. The Mössbauer measurements show that the fully oxygenated samples, which are not superconducting (y > 0.24), also do *not* display magnetic order in the  $CuO_2$  planes. Moreover, unlike in the systems mentioned above (Fig. 2), the magnetic ordering temperatures  $T_N$  of the quenched samples in this system depend on Zn concentration (Figs. 4 and 3), and antiferromagnetism disappears approximately at the same concentration (y = 0.30) where superconductivity disappears for the oxygen-rich samples, y = 0.24. Figure 4(a) displays some Mössbauer spectra of <sup>57</sup>Fe in YBa<sub>2</sub>- $Cu_{2.88}Zn_{0.12}O_{6+\delta}$  at several temperatures. As described above, the central part of the Mössbauer spectra corresponds mainly to Fe which occupies the Cu(1) site. The well-defined magnetic sextet, which is attributed to iron residing in the Cu(2) sites, disappears at  $T_N = 230$  K, much lower than  $T_N = 420$  K found in  $YBa_2Cu_3O_{6+\delta}$ . The magnetic hyperfine field observed at 90 K is 370 kOe. In all samples we identify above  $T_N$ , the typical paramagnetic quadrupole doublet is due to iron in the Cu(2) site.<sup>4,14</sup> [Isomer shift = 0.25(1) mm/s and eqQ/2=0.67(1) mm/s.] This doublet is also observed in samples with y = 0.30 in both fully oxygenated and quenched samples at all temperatures, thus proving that Fe also enters into the Cu(2) site. For all Zn concentrations measured its relative intensity accounts for about 20% of the overall spectral area and Fig. 4(b) exhibits (on an expanded scale) this doublet obtained in quenched YBa<sub>2</sub>Cu<sub>2.7</sub>Zn<sub>0.3</sub>O<sub>6+ $\delta$ </sub> (doublet A). The ab-



FIG. 4. (a) Mössbauer spectra of 1%-<sup>57</sup>Fe-doped YBa<sub>2</sub>-Cu<sub>288</sub>Zn<sub>012</sub>O<sub>6+ $\delta$ </sub> at several temperatures. The magnetic splitting disappears at  $T_N$  = 230 K. (b) Mössbauer spectrum of <sup>57</sup>Fe-doped YBa<sub>2</sub>Cu<sub>2.7</sub>Zn<sub>0.3</sub>O<sub>6+ $\delta$ </sub> in which the doublet corresponding to Fe in the Cu(2) sites (doublet A) is observed. Note the expanded velocity scale for this spectrum.

sence of magnetic order in Zn-rich samples, independent of the amount of oxygen, demonstrates that divalent Zn in the Cu(2) sites inhibits long-range magnetic correlations as well as reducing superconductivity.

In summary, the superconducting-magnetic phase diagrams with various cationic substitutions in oxygen-rich YBCO with different oxygen concentrations show that whenever the doping is *not* in the CuO<sub>2</sub> planes a similar behavior is observed: In the vicinity of the disappearance of superconductivity, the Cu(2) moments order antiferromagnetically. Substitution of Zn *in* the CuO<sub>2</sub> planes destroys both superconductivity and magnetism at approximately the same Zn concentration.

The experimental observations reported in this paper indicate that in oxygen-rich samples, superconductivity and magnetism are confined to the  $CuO_2$  planes. Substitutions affecting the number of mobile charge carriers (holes) in these planes, either by substituting Cu(1) by Fe or Co, oxygen by vacancies, or Y by Pr, all suppress superconductivity and enhance magnetic order. On the other hand, destruction of the periodicity and regularity in the  $Cu^{2+}O_2$  planes, even by diamagnetic  $Zn^{2+}$  is sufficient to suppress both superconductivity and magnetism. This observation indicates the delicate interplay between the two phenomena and might serve as a guideline for theories that attempt to describe the importance of magnetic interaction in the high- $T_c$  superconductors.

This research was supported by a grant from the German-Israel Foundation (GIF) for Scientific Research and Development, No. 1-40-100.10/87, and by a grant from the U.S.-Israel Binational Science Foundation (BSF), Jerusalem, Israel.

<sup>1</sup>P. W. Anderson, Science **235**, 1196 (1987); V. J. Emery, Phys. Rev. Lett. **58**, 2794 (1987).

<sup>2</sup>N. Nishida et al., Jpn. J. Appl. Phys. 26, L1856 (1987).

<sup>3</sup>J. M. Tranquada *et al.*, Phys. Rev. Lett. **60**, 156 (1988); W-H. Li, J. W. Lynn, and Z. Fisk, Phys. Rev. B **41**, 4098 (1990).

<sup>4</sup>I. Nowik, M. Kowitt, I. Felner, and E. R. Bauminger, Phys. Rev. B **38**, 6677 (1988).

<sup>5</sup>D. C. Johnston, S. K. Sinha, A. J. Jacobson, and J. W. Neusan, Physica (Amsterdam) **153-155C**, 572 (1988).

<sup>6</sup>D. Vaknin *et al.*, Phys. Rev. Lett. **59**, 1045 (1987); P. Imbert *et al.*, Hyperfine Interact. **50**, 599 (1989).

<sup>7</sup>T. Oashi, K. Kumagai, Y. Nakajima, T. Tomita, and T. Fujita, Physica (Amsterdam) **157**C, 315 (1989).

<sup>8</sup>R. F. Kiefl et al., Phys. Rev. Lett. 63, 2136 (1989).

<sup>9</sup>I. Felner, I. Nowik, and Y. Yeshurun, Phys. Rev. B 36, 3823 (1987).

<sup>10</sup>T. B. Lindener and A. L. Sutton ORNL Report No. ORNL/TM 10827, 1990 (unpublished); P. K. Gallagher, Adv. Ceram. Matter **2**, 632 (1987); K. Kishio *et al.*, Jpn. J. Appl. Phys. **26**, L1228 (1987).

<sup>11</sup>R. W. Howland, T. H. Geball, S. S. Laderman, A. Fischer-Colbrie, M. Scott, J. M. Tarascon, and P. Barboux, Phys. Rev. B **39**, 9017 (1989).

<sup>12</sup>K. Westerholt *et al.*, Phys. Rev. B **39**, 11680 (1989).

<sup>13</sup>T. J. Kistenmacher, Phys. Rev. B 38, 8862 (1988).

<sup>14</sup>D. Hechel, I. Nowik, E. R. Bauminger, and I. Felner, Phys. Rev. B 42, 2166 (1990).

<sup>15</sup>P. F. Miceli et al., Phys. Rev. B 39, 12375 (1989).

<sup>16</sup>P. H. Hor et al., Phys. Rev. Lett. 58, 1891 (1987).

<sup>17</sup>F. W. Lytle, G. van der Laan, R. B. Greegor, E. M. Larson,

C. E. Violet, and Joe Wong, Phys. Rev. B **41**, 8955 (1990). <sup>18</sup>A. Kebede *et al.*, Phys. Rev. B **40**, 4453 (1989).

<sup>19</sup>I. Felner, U. Yaron, I. Nowik, E. R. Bauminger, Y. Wolfus, E. R. Yacoby, G. Hilscher, and N. Pillmayr, Phys. Rev. B 40, 6739 (1989).

<sup>20</sup>U. Neukirch, C. T. Simmons, P. Sladeczek, C. Laubschat, O. Strebel, G. Kaindl, and D. D. Sarma, Europhys. Lett. **5**, 567 (1988).

<sup>21</sup>D. W. Cooke, R. S. Kwok, R. L. Lichti, T. R. Adams, C. Boekema, W. K. Dawson, A. Kebede, J. Schwegler, J. F. Crow, and T. Mihalisin, Phys. Rev. B **41**, 480 (1990).

<sup>22</sup>G. Xiao, M. Z. Cieplak, A. Gravin, F. H. Streitz, A. Bakhsahai, and C. L. Chein, Phys. Rev. Lett. **60**, 1446 (1988);
M. Mehbod *et al.*, Phys. Rev. B **38**, 11813 (1989); V. Radhakishnan *et al.*, *ibid.* **40**, 6850 (1989).

<sup>23</sup>H. Meada et al., Physica (Amsterdam) 157C, 483 (1989).