Size-Selective Depletion Spectroscopy of Predissociated States of Na₃

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The Na_3 photodissociation spectrum is studied by depletion spectroscopy. For the first time a predissociated state is observed in Na_3 and detected both by depletion and by recording of the related photofragments. The method is promising for photodissociation studies in clusters.

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The electronic properties of small metal clusters are of great importance for the understanding of the fundamental mechanisms of catalysis and surface chemistry. The investigation of the electronic structure is primarily carried out through molecular spectroscopy. In recent years, it has been demonstrated¹⁻³ that the metal clusters can efficiently be produced in molecular beams. Moreover, the resonant two-photon ionization (TPI) spectroscopy is now well known as an efficient method to select transitions belonging to a given species in a molecular beam. However, despite intense activity in the cluster research area, no gas-phase spectra of homonuclear particles larger than dimers has been reported in the literature, with the exception of Na₃^{4,5} and Cu₃.⁶ This situation can be partially explained by radiationless fast relaxation of the excited states. The relaxation can occur either by predissociation if the corresponding channels are open, or by intramolecular vibronic relaxation, possibly leading to dissociation for states having energy larger than the dissociation threshold. Such fast-relaxation channels of alkaline-earth-cluster states located beyond the first ionization limit have been demonstrated,^{7,8} for example, to be predominant compared to autoionizing transitions. In the case of Na₃, Gole et al.⁹ have shown the importance of the photodissociation processes in the visible absorption spectrum. For transition-metal clusters, also, the excited states beyond the dissociation threshold have been found to be mainly dissociative.¹⁰

We report in this Letter the first observation by depletion spectroscopy of a predissociated excited state in Na₃, which was simultaneously recorded with the resulting Na^{*}₂ photofragments. The principle of our method is illustrated in Fig. 1: A tunable laser $h\nu_1$ excites Na₃ to the predissociated state. Then the Na^{*}₃ particles dissociate to Na^{*}₂ + Na. A second photon of the same laser $h\nu_1$ ionizes the resulting Na^{*}₂ fragments, and the Na^{*}₂ -ion signal is recorded as a function of the irradiating laser wavelength. In order to confirm that the parent cluster of the observed transitions is Na₃, the molecular beam is simultaneously irradiated by a probe uv laser $h\nu_2$. This uv laser is capable of ionizing Na^{*}₃ directly from the ground state, and is used to monitor the Na₃ population in the molecular beam. Whenever a predissociating transition is excited by $h\nu_1$, a depletion of the Na₃⁺ intensity is monitored by $h\nu_2$. The effect is most efficient when the predissociating transition is saturated. Smalley and coworkers⁶ have performed a similar experiment on Cu₃, without, however, recording the related photofragments.

The basic part of the experimental setup is the seeded supersonic molecular beam which has been described in detail in the work of Delacrétaz and coworkers.¹¹ The metal vapor is expanded into the vacuum through a small orifice of typically 50 μ m diameter with argon as a carrier gas. The argon pressure, which is mainly limited by the 1500-l/s diffusion pump, can reach up to 10 bars. The metal pressure ranges between 10 and 100 mbar. With this expansion system, very cold Na clusters can be generated. A hotband analysis of the obtained spectra showed vibrational temperatures of Na₃ down to 30 K,¹² and rotational temperatures of Na₂ down to 7 K.⁵ Neutral-



FIG. 1. Principle of the depletion experiment.

beam intensities are permanently monitored with a surface ionization detector. A stilbene-420 dye laser pumped by a pulsed Lambda Physik excimer laser is used to excite the Na₃ clusters $(h\nu_1)$, and to ionize the resulting Na^{*}₂ photofragments. A small portion (2%) of the excimer laser output, which operates at 308 nm, directly ionizes the remaining Na₃ particles. The two laser beams are collinear and irradiate the molecular beam at right angles. The entrance of a quadrupole mass spectrometer is perpendicular to the particle and the laser beams. The measurements are performed by setting the mass spectrometer on the Na₂⁺- or Na₃⁺- mass peak, while the ion signal is recorded as a function of the $h\nu_1$ wavelength.

Figure 2(a) shows the ion signal recorded on the Na_2^+ channel at moderate laser power (typically 0.2 mJ/pulse and 10 kW peak power). The spectrum exhibits a regular progression corresponding to vibrational levels of the excited state. In the same experimental conditions, no ion signal is detected on Na^+ and Na_3^+ .

Despite the harmonic character of this progression, the observed excited state cannot be attributed to Na₂: All the Na₂ states in this region have been calculated by Jeung¹³ to high precision ($\approx 100-200 \text{ cm}^{-1}$) and none of them correspond with the experimental results. In fact, for Na₂, the energy of the laser ($\approx 23\,000 \text{ cm}^{-1}$) would lead to an excitation close to the dissociation limit Na(3p) + Na(3s), where the Franck-Condon factors are negligible¹⁴ and the vibrational spacings very small ($< 30 \text{ cm}^{-1}$). Moreover, the rather symmetric structure of each band [Fig. 2(b)] does not correspond to the usual rotational progression of a dimer.

As a consequence, the observed spectrum must be related to the predissociation of larger clusters. To find the parent cluster, we performed the depletion spectroscopy experiment. The power of the exciting



FIG. 2. Spectrum recorded on the Na_2^+ channel at moderate laser power. (a) The whole spectrum; (b) detailed structure of a vibrational band.

laser (hv_1) (typically 5 mJ/pulse) was set to saturate the predissociating transition. The energy of the 308nm probe laser was sufficient^{1,7} to directly ionize odd clusters like Na₃, Na₅, etc., but not to reach the Na₂ ionization potential.¹⁵ The power of this laser (typically 5 mJ/pulse) was set to avoid undesired multiphoton transitions, but sufficiently high to provide a good signal on Na₃⁺ and Na₅⁺.

Figure 3 (upper curve) shows the depletion signal obtained on Na3⁺ while the relating predissociation spectrum, recorded on the Na_2^+ fragment, is shown below. The perfect coincidence between the Na₃⁺ depletion spectrum and the predissociation spectrum is striking and demonstrates that Na₃ is the parent cluster. The comparison of Figs. 2(a) and 3 (lower curve) shows the influence of the saturation on the relative intensities of various vibrational bands. Figure 4 compares the depletion signal observed for Na₃⁺ (upper curve) and Na5⁺ (lower curve). Both curves show the relative depletions, which, in the case of Na₅, reaches 70%. For Na₅, however, no structure is observed. This confirms the importance of dissociation processes in the excited states of large clusters. The decrease in the dissociation yield beyond 420 nm is partially due to the profile intensity of the dye laser. From the dissociation yield and the laser power, we deduce that the



FIG. 3. Upper: depletion signal recorded on Na_3^+ . Lower: TPI signal recorded on the Na_2^+ channel.



FIG. 4. Depletion signal recorded on Na_3^+ (upper curve) and Na_5^+ (lower curve).

photodissociation cross section can be estimated to be about 10^{-17} cm² which corresponds to a typical value for a strongly allowed transition.¹⁶

In the Na_2^+ spectrum [Fig. 2(a)], a continuous hump is observed in the 415-nm region. This continuous spectrum corresponds to a direct bound-free transition in Na₃, as confirmed by the fact that the continuous signal of the Na₃ depletion spectrum, which is superimposed on the vibrational progression, increases for wavelength shorter than 420 nm (Fig. 3, upper curve). No hump is observed in the Na₂⁺ saturated spectrum of Fig. 3 (lower curve), because of a nonresonant TPI background from unexcited Na₂. The continuous dissociation spectrum probably corresponds to the absorption by the dissociative state which is also responsible for predissociation. With this hypothesis and from Fig. 2, the Franck-Condon pattern of the dissociative state appears slighly shifted as compared to the predissociated one. This would emphasize a location of the dissociative potential surface almost parallel to that of the predissociated state, and slightly shifted toward high energy (as schematically drawn on Fig. 1).

The fact that no resonant TPI signal is detected in the 420-nm range on the Na_3 channel indicates that the predissociation lifetime is significantly less than 1 ns over the whole spectrum.

Let us now focus on the possible dissociation channels of Na₃. We observe in our experiment the predissociation $Na_3^* \rightarrow Na_2^* + Na$. The Na_2 ionization is performed with the same laser as the Na₃ excitation. The laser energy is 23 200 cm⁻¹ < $h\nu_1$ < 24 100 cm⁻¹, which excludes energetically all dissociation channels of the kind $Na^* + Na_2(X \text{ state})$ except for $Na^{*}(3p) + Na_{2}(X)$. The laser $h\nu_{1}$ has insufficient energy to ionize the Na in the 3p state and therefore the last channel cannot be observed in our experiment. It cannot be very dominant, however, because an intense signal is detected on Na₂⁺. With Na₃ dissociation en $ergy^{17,18}$ taken into account (3000 cm⁻¹) and the Na₂ ionization potential (34480 cm^{-1}) , a simple energy balance demonstrates that the Na2 excited states resulting from the predissociation are located between 16 200 and 20 200 cm⁻¹ above the Na₂ ground state. With use of the theoretical calculations of Jeung,¹² and recent spectroscopic results,¹⁹⁻²¹ only four states can be candidates for this dissociation: the $a^3\Pi_u$, $A^1\Sigma_u^+$, $(1)^{3}\Sigma_{g}^{+}$, and $(2)^{1}\Sigma_{g}^{+}$ which are located at about 13 600, 14 700, 18 400, and 19 300 cm⁻¹, respectively, the $B^{1}\Pi_{u}$ state being slightly too high. In the case of the two first states (a and A states) ionization occurs only to fragments with a vibrational excitation exceeding 2600 and 1500 cm^{-1} , respectively. For this reason, the dissociation toward the two other states seems more probable. Let us notice, moreover, that the existence of an excited Na₃ state can be invoked to explain the recently observed very efficient relaxation process²²

$$\operatorname{Na}_{2}(B^{1}\Pi_{u}) + \operatorname{Na}(3s) \rightarrow \operatorname{Na}_{2}(2)^{1}\Sigma_{g}^{+} + \operatorname{Na}(3s).$$

If the assignment of the $(2)^1\Sigma_g^+$ state as the main dissociation channel could be confirmed, the presently observed predissociated Na₃ state would be the intermediate complex of the collision.

We have also recorded the Na₃ predissociation spectrum as a function of the cluster temperature by varying the carrier-gas pressure. The intensities of the small peaks observed between the intense bands of the main vibrational progression have the characteristic behavior of hot bands. The wave numbers of all the measured transitions are reported in Table I. The average distance between the main bands and the hot bands is 45 ± 3 cm⁻¹. This result is an indication for the Na₃ ground-state structure. It is in good agreement with hot bands which we observed in different Na₃ electronic states.²³

In conclusion, we have shown that depletion spectroscopy combined with two-photon ionization techniques is a powerful tool to observe completely and

TABLE I. Wave numbers of all the measured transitions. E' is the energy difference between two consecutive cold bands and E'' the energy difference between hot and cold bands.

Running number	Cold bands $1/\lambda$ (cm ⁻¹)	Hot bands E' $1/2$ (am^{-1}) E''		
		<i>L</i>		L
1	23 161.5	88	23 1 1 5.0	46.5
2	23 249.5	86.5	23 209.5	40
3	23 336.0	85	23 288.0	48
4	23 421.0	85.5	23 374.5	46.5
5	23 506.5	85	23 462.0	44.5
6	23 591.5	84	23 548.0	43.5
7	23 675.5	84.5	23 633.0	42.5
8	23 760.0	81.5	23716.0	44
9	23 841.5	83	23 798.5	43
10	23 924.5		23 875.5	49

partially predissociated excited states of clusters. With inclusion of the present result, four spectroscopic systems are now known for Na₃. Three of them, ranging around 675, 625-550, and 475 nm, have been observed⁵ by means of two-photon ionization spectroscopy. The corresponding excited states are not or only weakly dissociated with lifetimes larger than 1 ns. The fourth system is the presented 420-nm system which is completely predissociated. A systematic investigation of Na₃ by means of depletion spectroscopy, covering the whole spectral range to 330 nm, is in progress. In addition, in contrast to TPI spectroscopy of the trimer and smaller clusters which is limited by fragmentation, the present method may also be applied to larger clusters so long as the predissociative states are not overwhelmed by dissociative states.

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