Hydrogen-Acceptor Pairs in Silicon

In a recent Letter, Assali and Leite $(AL)^1$ propose a model for the structure of a hydrogen-boron pair in silicon and the concomitant loss of shallow-acceptor activity. Using the $MS-X\alpha$ cluster method, they explore single-particle and total energies as functions of H displacement from the T_d interstitial position towards the substitutional B.

In this Comment, we would like to point out recent theoretical and experimental studies of the H-B pair which were not addressed in the Letter of AL. These results and further calculations that we report here strongly suggest that the AL model for the H-B pair (B_sH_i) is not the stable, minimum-energy configuration.

We have previously reported^{2, 3} both $MS \cdot X \alpha^4$ and modified neglect of diatomic overlap (MNDO)⁵ computational results for the H-B pair. Using the MNDO-cluster method, we found that, with B on center, H did not remain along the line explored by AL.^{2, 3} Furthermore, allowing both the H and B to relax, we found that the interstitial H moved into a position between the B and one of its bonding Si neighbors producing a Si—H bond, while the B moved off center in the opposite direction to attain threefold coordination with the remaining silicons.

This acceptor-deactivating configuration was originally suggested by Pankove and co-workers.⁶ Here, the B and H interact with each other only weakly; we have effectively a SiH--BSi₃ system. This can be contrasted with the AL model, which has an interstitial H interacting primarily with on-center substitutional B (a H—BSi₄ system).

Further support for the Pankove model comes from recent infrared studies^{7,8} and supporting theoretical work.³ In the experimental studies, an 1870-cm⁻¹ absorption was correlated with the H-B pair defect. The Pankove (SiH--BSi₃) model was favored since the observed H vibrational frequency was consistent with a Si—H bond. This conclusion was subsequently supported by our MNDO computations, where we calculated a value of 1880 cm^{-1.3} Any model which would involve H vibrations in a B—H bond was rejected by Pankove since it would be expected to produce an absorption at ~ 2560 cm⁻¹.

We have done follow-up MNDO calculations using a hydrogen-terminated version of the AL cluster (BHSi₁₀H₁₈ with frozen silicons). Exploring first the AL configuration, we find a minimum when H is 1.19 Å along the interstitial direction from an on-center B (compare with 1.8 Å from Fig. 2 of AL). For this configuration, we calculate a 2590-cm⁻¹ H vibration,⁹ which is consistent with the argument by Pankove and co-workers.

We also calculate the total energies for the Pankove and AL orientations of the B-to-H pair axis [[111] vs [111], respectively, in a (110) plane}. For each orientation, the total energy is minimized with respect to independent H and B motion along the line. We find that the [111] Pankove orientation is lower in energy by 1.3 eV. Therefore, the AL minimum is a saddle point according to our calculations.

As a final, related comment, we note that the acceptor-deactivating configuration suggested by AL has been proposed and well studied for the Li-B pair in silicon.¹⁰ In fact, we find in using the MNDO method that it is the [111] AL configuration of the Li-B pair which is favored over the [111] Pankove orientation by 1.9 eV. This affinity of Li to the interstitial site is consistent with our earlier studies.¹¹

In summary, we suggest that the potential-energy surface explored by AL was too limited. This we conclude on the basis of our MNDO calculations which indicate that (i) the AL configuration is a saddle point, with the Pankove configuration producing the true minimum, and (ii) the computed H vibrational frequency is in agreement with recent infrared studies for the Pankove model, but not for the AL model.

The authors would like to thank Professor G. D. Watkins, Professor F. S. Ham, and Professor F. J. Feigl for helpful discussions. This work was supported by U.S. Naval Research Electronics and Solid State Science Program Contract No. N00014-85-K-0460.

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PACS numbers: 71.55.Fr

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Received 23 September 1985