Final-State Effects in the Eu L_{III} -Edge Spectrum of EuPd₂P₂

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The Eu L_{III} -edge x-ray-absorption spectrum of EuPd₂P₂ exhibits a double-peaked white-line structure typical for mixed-valent materials, while Mössbauer isomer-shift and other magnetic measurements prove the divalent ground state of this compound. This discrepancy clearly shows the possible presence of final-state configuration-interaction effects in the L_{III} -edge spectra of heavy-rare-earth systems, as well as a significant 4*f* covalent bonding, contrary to previous assumptions.

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One of the major issues in the field of valence fluctuation phenomena in rare-earth (RE) compounds is the precise determination of the mean valence, \overline{v} , of the RE ion. In this respect, L_{III} -edge x-ray-absorption (XA) spectroscopy—due to its universal applicabil-ity—is widely used.^{1,2} The L_{III} -edge spectra of RE systems are characterized by intense "white line" at the edges, caused by E1 excitation of a 2p-core electron into an empty 5d state at the Fermi level $(E_{\rm F})^3$. In mixed-valent systems, two white-line peaks, separated by 7-8 eV, are commonly observed, which are due to final states with different 4f occupation numbers.⁴ A major assumption in deriving mean valences from such double-peaked $L_{\rm III}$ -edge profiles is the absence of final-state configuration-interaction effects in systems of the heavy RE's, despite the creation of a 2p-core hole. Though this aspect has been discussed in connection with Ce and its compounds,^{5,10} there is no consensus on this assumption.

In this paper, we report the first observation of such final-state effects in the L_{III} -edge spectrum of a heavy RE system, namely divalent $EuPd_2P_2$ (with ThCr₂Si₂) structure). While low-energy probes (Mössbauer isomer-shift, magnetic susceptibility, electron-spin resonance) as well as valence-band photoemission spectra clearly prove the divalent- $4f^7$ ground state of the Eu ion, the high-energy L_{III} -edge probe results in a hitherto unexpected double-peaked L_{III} -edge profile, with both $4f^7$ and $4f^6$ signatures in the final state. A comparative study of trivalent GdPd₂P₂, on the other hand, leads to the expected single-peaked L_{III} -edge XA spectrum. It is suggested that the L_{III} -edge anomaly of $EuPd_2P_2$ is the consequence of a considerable 4f participation in covalent bonding in this compound. The present results bear a strong significance for L_{III} -edge studies of mixed-valent RE compounds.

The studied samples were synthesized as described in the literature.¹¹ To avoid oxidation, the material was finely ground in a purified argon atmosphere and sandwiched in the form of a homogeneous absorber layer between Be sheets in a vacuum-tight, In-sealed metal capsule. The $L_{\rm III}$ -edge XA measurements were performed at the EXAFS-II beam line of HASYLAB/DESY, Hamburg, with a resolution of $\cong 2 \text{ eV}$ full width at half maximum at 7 keV, as well as with a laboratory XA spectrometer.¹² To check for Eu³⁺ contamination, ¹⁵¹Eu-Mössbauer measurements were done after the L_{III} -edge studies on the same absorbers at various temperatures, with a ¹⁵¹SmF₃ single-line source. An opposed-anvil device (sintered B₄C anvils) was employed for high-pressure Mössbauer studies at room-temperature (40 kbar) and at liquid-He temperature (80 kbar).¹³

Typical L_{III} -edge XA spectra obtained for EuPd₂P₂ and GdPd₂P₂ are displayed in Fig. 1. A single "whiteline" peak, due to the optical excitation of a 2*p*-core electron into an empty 5*d* state with a high density of states at $E_{\rm F}$, is observed for trivalent GdPd₂P₂.³ In the



FIG. 1. RE $L_{\rm III}$ -edge XA spectra of EuPd₂P₂ and GdPd₂P₂, with the solid lines representing least-squares fit results. The subspectra due to the two different final states in the case of EuPd₂P₂ are indicated by dashed lines. Zero on the energy scale represents the $L_{\rm III}$ -edge positions (6958.5 eV for EuPd₂P₂ and 7227.4 eV for GdPd₂P₂).

case of EuPd₂P₂, a system undoubtedly divalent in the ground state (see below),¹⁴ however, a double-peaked $L_{\rm III}$ -edge structure, typical for mixed-valent RE systems, is clearly observed.⁵ We have carefully investigated three different batches of EuPd₂P₂ samples, as well as absorbers with largely different thickness, with practically identical results. In addition, each absorber was checked by ¹⁵¹Eu-Mössbauer measurements, performed *after* the $L_{\rm III}$ -edge studies, for Eu³⁺ contamination; in no case was the Eu³⁺ fraction found to exceed 1% (see below).

We have least-squares fitted the sum of a Lorentzian line and an arctanlike edge feature, convoluted by a Gaussian spectrometer function, to the experimental spectrum of $GdPd_2P_2$. In the case of $EuPd_2P_2$, a superposition of two such subspectra, shifted by 7.7 eV against each other (dashed lines in Fig. 1), describes the data well. The low-energy peak, as well as the single peak in GdPd₂P₂, is assigned to a $2p^{5}4f^{7}$ final-core state, while the peak at higher excitation energy is assumed to be due to a $2p^54f^6$ final-core state. Its fractional intensity is found to be $(15 \pm 5)\%$. Even though the exact lineshapes of L_{III} -edge spectra are not well known, the fractional intensity of the highenergy peak in $EuPd_2P_2$ is rather independent of different spectral shapes used in the analysis (e.g., Fano line shape), and possible systematic errors are included in the quoted error bar. The given assignment of the high-energy peak as due to a $2p^54f^6$ final-core state is strongly supported by the observed energy separation, typical for Eu systems representing¹⁵ the difference in Coulomb energies of the 2p-core hole in the two final states.⁴ An anomalous near-edge structure can also be excluded as the possible origin of this peak, since such a structure seems to be totally absent in isostructural trivalent $GdPd_2P_2$ [see Fig. 1(b)].

We will now briefly describe our careful ¹⁵¹Eu-Mössbauer spectroscopic work to characterize the divalent ground state of $EuPd_2P_2$, and check each L_{III} -edge absorber for possible Eu^{3+} contamination. The 21.6-keV Mössbauer resonance of ¹⁵¹Eu is uniquely suitable for such an undertaking, since the isomer shift S is known to be a very sensitive measure of the Eu valence.¹⁶ Typical results of such Mössbauer experiments are shown in Fig. 2 for different absorber temperatures, along with a spectrum taken at 4.2 K and an external pressure of 80 kbar. A heterogeneous mixed-valent situation can be clearly excluded as the origin of the high-energy $L_{\rm III}$ -edge peak in EuPd₂P₂, since no trivalent-Eu signal, expected around zero Doppler velocity, is observed at all temperatures studied. An anomalously low Debye-Waller factor of such an Eu^{3+} phase can also be excluded on the basis of the low-temperature spectra. We conclude that a possible Eu^{3+} content in the studied absorbers is definitely below 1% in all cases.



FIG. 2. (a)–(c) ¹⁵¹Eu-Mössbauer spectra at various temperatures at ambient pressure, taken after the L_{III} -edge measurements on the identical absorber. (d) The spectrum at 4.2 K at a high pressure (80 kbar).

We can also rule out a dynamic or homogeneously mixed-valent state for the Eu ion in EuPd₂P₂ on the basis of the observed temperature and pressure dependence of S (see Fig. 2). The observed isomer shift, $S = -9.73 \pm 0.05$ mm/s (at 300 K; relative to the SmF₃ source), falls within the range known for divalent Eu compounds, and is only weakly temperature dependent due to thermal red shift.¹⁶ In accordance with this observation, the pressure-induced variation of S is found to be weak, dS/dP = +0.015 (mm/s)/kbar, and in agreement with dS/dP values known for other divalent Eu compounds.¹⁷ It should be recalled here that S is generally observed to vary appreciably with pressure and temperature in homogeneously mixed-valent Eu compounds.^{13, 18, 19} Therefore, magnitude as well as temperature and pressure dependence of S prove the divalent ground state of the Eu ion in $EuPd_2P_2$. This is also borne out by the observed magnetic ordering of $EuPd_2P_2$ below $\cong 30$ K, leading to a magnetically split spectrum [see Fig. 2(c)], with a magnetic hyperfine field of $B_{\rm hf} = 35.5 \pm 0.5$ T. Again, the observed pressure dependence, $d \ln B/dP = 1.10$ $\times 10^{-3}$ kbar⁻¹, is characteristic for divalent Eu compounds.14

Independent support for the above conclusion is given by the following experiments: (1) The effective

magnetic moment derived from susceptibility measurements is $\mu_{eff} = (8.0 \pm 0.1)\mu_B$, which is in good agreement with the free-ion value for a $4f^{78}S_{7/2}$ ion.¹⁴ (2) The electron spin resonance (ESR) signal from the Eu ions, with a g value of 2.00 ± 0.05 , and a linewidth of 800 Oe at 77 K, typical for divalent Eu systems,²⁰ could be observed. In an intermediate valent (IV) compound, the fast relaxation between two valence states completely wipes out the ESR signal.²⁰ (3) We have performed²¹ valence-band photoemission measurements on EuPd₂P₂, using synchrotron radiation. We do not see any emission from the trivalent Eu ions in the region 5–10 eV below E_F , even at the giant $4d \rightarrow 4f$ resonance. This observation is strictly against any kind of mixing between $4f^7$ and $4f^6$.

With $EuPd_2P_2$ we are therefore faced with a situation that the $L_{\rm III}$ -edges show mixed-valent features, while Mössbauer spectroscopy, magnetic susceptibility, ESR, and valence-band photoemission clearly prove the divalent ground state of the Eu ions. This finding represents the first conclusive evidence for possible discrepancies in valence determinations by L_{III} -edge XA spectroscopy, even in compounds of the heavy RE's. We assign the observed discrepancy to corehole induced final-state configuration-interaction effects. In the present case, this obviously leads to a 15% probability for a shakeup of a 4f electron to one of the ligand orbitals (Pd, P). In this connection, we point out that shakeup effects have also been reported in core-level x-ray photoemission spectra for (XPS) of a few integral-valent Eu compounds.²² Very recently, the connection between such effects in core-level XPS and *L*-edge XA spectra has also been pointed out.²³

The question one would like to answer is why such strong final-state effects are seen in the L_{III} -edge spectra of $EuPd_2P_2$, whereas in other compounds they seem to be small or absent.¹⁵ The answer must lie in the degree of covalent mixing of the 4f electrons with ligand states, since the magnitude of final-state configuration-interaction effects is known to depend generally on this hybridization strength.²⁴ In this way, we come to the conclusion that the 4f electrons in $EuPd_2P_2$ participate rather strongly in chemical bonding with the ligands, contrary to the present assumption for heavy RE systems. Our conclusion is supported by the unusual lattice-parameter anomaly reported previously for $EuPd_2P_2$.^{11,14} The crystal structure of EuPd₂P₂ contains layers of identical atoms in the sequence Eu-P-Pd-P, stacked perpendicular to the c axis. The lattice parameter c of EuPd₂P₂, when compared with those of the series of trivalent REPd₂P₂ compounds, deviates considerably to a smaller value even though a constant or larger value would be expected from ionic-size arguments.¹⁴ This c deviation of $EuPd_2P_2$ is considerably larger than the observed c variation in the series of trivalent compounds from Ce

to Tm. It indicates strong covalent bonding between Eu and the ligands in this compound, presumably involving the 4f electrons. In this connection, we would like to recall that the covalent bonding should not be confused with the mixed valence due to charge fluctuations $(4f^{n} \leftrightarrow 4f^{n-1})$, as pointed out by Wachter²⁵ for the case of CeO₂.

Finally, we would like to point out that a related discrepancy between lattice properties and effective magnetic moments from susceptibility measurements has recently been reported for the semiconducting $TmSe_{1-x}Te_x$ system.²⁶ It was interpreted as an indication of a strong hybridization which leads to a partial extension of 4f orbital radius, at the same time preserving the localized character of 4f electrons. These views could be applicable to the present case also.

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