Unusual C-O Bond Weakening on a Clean Metal Surface: CO on Cr(110)

Neal D. Shinn and Theodore E. Madey

Surface Science Division, National Bureau of Standards, Gaithersburg, Maryland 20899

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A unique CO chemisorption mode (α_1 CO), with the lowest reported CO stretching frequencies (1150–1330 cm⁻¹) on any clean or promoted metal surface, has been identified on Cr(110) by means of EELS, ESDIAD, LEED, AES, and oxygen coadsorption experiments. A "lying down" binding configuration in twofold-symmetric hollow sites is proposed for α_1 CO molecules. For $\theta_{CO} > 0.25$, a second mode (α_2 CO) bound perpendicular to Cr(110) is populated.

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In this Letter, we present evidence for a previously unreported molecular CO chemisorption mode $(\alpha_1 CO)$ on a clean transition-metal surface, Cr(110), at 120 K. This new binding mode consists of molecules adsorbed with the intramolecular axis approximately *parallel* to the chromium surface and is the only CO binding mode at low coverage. In contrast, a second, sequentially populated, CO chemisorption mode (α_2 CO) is identified as terminally bonded CO with the intramolecular axis perpendicular to the surface. The observed $\alpha_2 CO$ stretching frequencies $(1865-1975 \text{ cm}^{-1})$ are within the expected range for chemisorbed CO,¹ but the comparable frequencies for $\alpha_1 CO$ $(1150-1330 \text{ cm}^{-1})$ are substantially shifted from the gas-phase value (2143 cm^{-1}) and are the lowest reported to date for any clean or "promoted" single-crystal metal surface.²

High-resolution electron-energy-loss spectroscopy (EELS), electron-stimulated desorption ion angular distributions (ESDIAD), low-energy electron diffraction (LEED), and Auger-electron spectroscopy (AES) experiments were performed in a stainless-steel UHV system with a base pressure of 4×10^{-11} Torr. Gas exposures were made with a microcapillary array doser, which allowed spectroscopic measurements to be made at or below 7×10^{-11} Torr, and thereby minimized contamination problems. The sputtered and annealed Cr(110) surface exhibited a good-quality (1×1) LEED pattern against a low-intensity background; faceting or periodic step formation was not observed. The only surface impurity detectable via AES was carbon, which, on the basis of Cr(110)/CO AES studies,³ was estimated to be less than 5% of a monolayer. Because the Cr LMM Auger transitions obscure the oxygen KLL lines, EELS was used to estimate the residual oxygen coverage by calibration against Cr(110)/O data.³ Although a "clean" surface has, typically, 3% of a monolayer of atomic oxygen, coadsorption effects

have been explicitly measured⁴ and do not alter the conclusions of this work.

At 120 K, CO adsorbs on Cr(110) with an initial sticking probability of ~ 0.9 , which drops smoothly to ~ 0.4 after a 0.7-L (1 L = 10⁻⁶ Torr sec) CO exposure.³ A fully developed $c(4 \times 2)$ LEED pattern is then observed, corresponding to an α_1 CO coverage of 0.25 monolayer. Corresponding EELS data are shown in Fig. 1. The clean-surface spectrum (Fig. 1, curve a) contains one discrete loss at 595 cm^{-1} due to the residual atomic oxygen.³ A 0.3-L CO exposure results in three new losses at ~ 450 (weak), 1150, and 1330 cm⁻¹ (Fig. 1, curve b), which increase in intensity with CO exposure, and shift to ~ 475 , 1200, and 1330 cm⁻¹ when the $c(4 \times 2) \alpha_1 CO$ overlayer is complete. Figure 1, curve c (0.7-L CO) marks the onset of α_2 CO by the appearance of a new vibrational triplet at 495, 1865, and 1975 cm^{-1} ; this indicates that (to within our EELS sensitivity) the α_1 CO mode is the only CO binding mode for $\theta_{CO} \leq 0.25$ and that it saturates with the formation of the $c(4 \times 2)$ overlayer.

For CO exposures beyond ~ 0.7 L, the following are observed: (1) The measured CO sticking probability declines to 0.1; (2) the α_2 CO vibrational intensity increases (Fig. 1, curve d) while the 1200and 1330-cm⁻¹ EELS peaks merge and are slightly attenuated; and (3) the $c (4 \times 2)$ LEED pattern reverts to a diffuse (1×1) pattern with a highintensity background at $\theta_{CO} \approx 0.3$. After 2–3-L CO exposure ($\theta_{CO} \sim 0.35$), broad α_2 CO vibrational losses dominate the EEL spectrum and the (1×1) diffraction beams are barely visible above the intense background, which indicates extensive adlayer disorder. Additional experiments at higher CO coverage have shown that some CO dissociation and surface reconstruction can occur even below 300 K.

Coverage-dependent ESDIAD measurements also reflect the sequential population of distinctly different α_1 CO and α_2 CO modes. When one starts with the clean surface and continues through an



FIG. 1. High-resolution (90 cm⁻¹ full width at half maximum) EELS data for CO on Cr(110) at 120 K: Curve *a*, clean; curve *b*, 0.3-L CO; curve *c*, 0.7-L CO; curve *d*, 0.9-L CO. All spectra are normalized to the elastic peak height for the clean surface (shown).

~ 0.7-L CO exposure, only vanishingly weak O⁺ion emission normal to the surface is observed. This extremely low signal is consistent with ESD (electron-stimulated desorption) of the residual atomic oxygen but is remarkably weak for ~ 0.25 monolayer of α_1 CO, *if* it were bound in a conventional manner.⁵ With the onset of the α_2 CO EELS vibrational features, a single O⁺ beam along the [110] normal (±20°) is observed, increasing in intensity along with the normalized EELS α_2 CO losses. Additional tests for α_1 CO ESD O⁺ ions with use of incident electron energies up to 1 keV failed to produce detectable ion desorption along any azimuth.

The absence of O⁺ emission from α_1 CO could be attributed to a dramatic cross-section difference between α_1 CO and α_2 CO for electron-induced dissociation and subsequent ion desorption. However, EELS or LEED measurements following ESDIAD revealed severe electron-beam damage effects for both $\alpha_1 CO$ and $\alpha_2 CO$. Even low electron-beam doses (10⁻⁸ C) caused detuning of the EELS spectrometer (attributed to work-function changes due to CO dissociation) and more diffuse $\alpha_1 CO c (4 \times 2)$ LEED beams. The significant distinction between the α_1 and α_2 binding modes lies not in their cross sections for electron-stimulated dissociation but rather in their probabilities for O⁺ ion desorption; chemisorption models for $\alpha_1 CO$ and $\alpha_2 CO$ must account for this difference.

The α_2 CO ESDIAD results imply that these CO molecules are oriented nearly perpendicular to the (110) plane, with the carbon end bound to the Cr surface. The simplest interpretation of the two α_2 CO EELS losses at 1865 and 1975 cm⁻¹ is an assignment to excitations of CO stretching modes for normally bound CO at both bridge-bonded and atop-bonded chromium sites, respectively. The 495-cm⁻¹ vibration is then assigned to the unresolved Cr-CO stretching modes for the two normally bound species. Other possible interpretations are discussed elsewhere.³

A suitable α_1 CO chemisorption model is not immediately obvious. In fact, the intense vibrations at 1150–1330 cm⁻¹ are more suggestive of an adsorbed carbonate, acetate, or formate species⁶ (as the product of a surface reaction) than chemisorbed CO. Compelling evidence that these reactions, which require atomic oxygen or some CO dissociation, are *not* occurring is provided by CO/O₂ titration studies. If the α_1 CO mode were a reaction product, then deliberate oxygen predosing should facilitate product formation. In fact, the *opposite* effect is experimentally observed. Oxygen predosing serves to *inhibit* α_1 CO formation with $\theta_0 \approx 0.2$ monolayer sufficient to block the α_1 CO mode completely; only α_2 CO adsorbs under these conditions.

A more dramatic demonstration that the $\alpha_1 CO$ mode is indeed chemisorbed CO is its conversion to the α_2 CO mode, also at 120 K. Figure 2 highlights the results of one such EELS experiment. The clean surface (Fig. 2, curve a) is first dosed with CO to form the saturated $c(4 \times 2) \alpha_1$ CO layer (Fig. 2, curve b), which is then stepwise dosed with oxygen. Two major changes in the vibrational spectrum with oxygen exposure are observed: First, losses appear at 605 and 1030 cm^{-1} corresponding to adsorbed atomic and molecular oxygen⁴; second, and more importantly, the intensity at ~ 1300 cm^{-1} from $\alpha_1 CO$ smoothly decreases and losses at both 1865 and 1975 cm⁻¹, due to α_2 CO, monotonically *increase* in intensity (Fig. 2, curve c and inset). AES measurements have demonstrated the conservation of surface carbon, in accord with the rear-



FIG. 2. Sample EELS data for (curve *a*) clean Cr(110); (curve *b*), 0.6-L CO; and (curve *c*), after an additional 0.4-L oxygen exposure. Inset shows normalized peak areas for the α_1 (1150–1330 cm⁻¹) and α_2 (1865–1975 cm⁻¹) CO vibrational bands.

rangement of adsorbed CO. These experiments confirm that the α_1 CO mode is indeed a molecularly adsorbed CO binding state. Additional experiments demonstrate that the α_1 CO mode undergoes dissociation upon heating to $T \ge 200$ K, indicating that the α_1 CO mode serves as a precursor state to CO dissociation.³

A model for the α_1 CO adsorption mode must account for the anomously low vibrational bands at 1150–1330 cm⁻¹. Simplistic normal-mode analyses of Cr_nCO clusters, configured to simulate CO adsorbed on Cr(110), have been used to investigate the frequency dependence of the *highest*-frequency normal mode (ν^*) as a function of CO bond strength and CO binding configuration. The expected finding was that, given reasonable choices for the intracluster force constants and atomic distances, the force constant chosen for the C–O bond largely determines ν^* . Hence, the observed 1150–1330-cm⁻¹ vibrations imply substantial CO bond weakening, which within a molecular-orbital description requires electron donation into the anti-

bonding CO $2\pi^*$ orbital. An alternate explanation is that the CO-chromium bonding, in the α_1CO mode only, involves molecular-orbital rehybridization. Either of these possibilities is difficult to rationalize if the CO is constrained to be perpendicular to the Cr(110) surface. Greater overlap of the CO $2\pi^*$ orbitals and the chromium valence electron density could be achieved if the molecular axis were nearly parallel to the surface, thus facilitating CO bond weakening. Such a binding configuration has already been postulated for a transition state during CO dissociation on transition metals⁷ and a side-on bonded carbonyl group ($\nu_{CO} = 1330 \text{ cm}^{-1}$) has been reported in the $(\eta^5-C_5H_5)_3Nb_3(CO)_7$ cluster.⁸

The ESDIAD results further support a "lying down" α_1 CO binding configuration. The absence of measurable O⁺ ion emission implies that O⁺ ions ejected along the C-O bond axis are recaptured by the attractive image field outside the metal surface. A classical-trajectory approach⁹ predicts that O⁺ will be recaptured if CO is tilted at least 60° away from the surface normal; however, the Cr-CO libration should then be visible via EELS. But an α_1 CO molecule "lying flat," with direct Cr-O interaction, would be consistent with both a higher ESD O⁺ ion neutralization probability for α_1 CO and no observable CO libration.

We propose that both the electronic structure and morphology of the Cr(110) surface are significant factors in determining the bonding characteristics of α_1 CO. In particular, unlike the low-index fcc surfaces, Cr(110) exhibits twofold-symmetric hollow sites of appropriate geometry to align the majority of the $\alpha_1 \text{CO}$ molecules roughly along the [110] direction ($\nu_{\text{CO}} = 1330 \text{ cm}^{-1}$). Prior to full c (4×2) α_1 CO overlayer development, other CO molecules adsorbed at defects, near residual surface oxygen atoms, or in small islands could lead to the minority $\nu_{\rm CO}$ band which shifts with $\theta_{\rm CO}$ (1150 \rightarrow 1200 cm^{-1}). The weak 450-475- cm^{-1} loss may be associated with $\alpha_1 CO$ molecules or due to a very low coverage of normally bound CO, since the EELS metal-CO stretch is frequently more intense than $\nu_{\rm CO}$ at very low $\theta_{\rm CO}$.¹

Similar "low CO stretching frequencies" have previously been reported for CO adsorbed on potassium-promoted Ru(001).² In this case, it has been argued that the CO molecule is "side-on bonded" with the C-O axis nearly parallel to the surface. However, because of the complicated $\theta_{\rm K}$ dependent behavior and direct interactions between coadsorbates, the local CO adsorption geometry corresponding to the anomolously low $\nu_{\rm CO}$ is uncertain. Very recent ESDIAD studies of CO/K/ Ru(001) in our laboratory¹⁰ support the model of CO molecules parallel to the surface; the possible correlation between the electronic structures of CO/K/Ru(001) and the α_1 CO mode is particularly enticing.

In summary, the sequential population of two, generically different CO binding modes on Cr(110)at 120 K has been demonstrated. The low-coverage $(\theta_{CO} \leq 0.25) \alpha_1 CO$ mode is without analog on any other clean metal surface and may be comparable to certain CO binding states on K-promoted Ru(001). EELS/ESDIAD α_1 CO results are modeled by molecules having weakened C-O bonds oriented parallel to the Cr(110) surface in an ordered $c(4 \times 2)$ overlayer at saturation. Twofold-symmetric hollow Cr sites are suggested as local adsorption sites for the new α_1 CO mode. The second, α_2 CO mode is modeled as a disordered combination of both bridge- and atop-bonded, terminally bound CO, consistent with the conventional picture of carbon monoxide chemisorption on single-crystal metal surfaces. Finally, CO/O coadsorption results have demonstrated the blocking of $\alpha_1 CO$ by adsorbed atomic oxygen and the low-temperature conversion of $\alpha_1 CO$ to $\alpha_2 CO$.

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¹J. T. Yates, Jr., T. E. Madey, and J. C. Campuzano, in "The Chemical Physics of Solid Surfaces and Heterogeneous Catalysis," edited by D. A. King and D. P. Woodruff (Elsevier, North-Holland, Amsterdam, to be published).

 2 F. M. Hoffmann and R. A. de Paola, Phys. Rev. Lett. **52**, 1697 (1984).

³N. D. Shinn and T. E. Madey, to be published.

⁴N. D. Shinn and T. E. Madey, to be published.

⁵T. E. Madey, F. P. Netzer, J. E. Houston, D. M. Hanson, and R. Stockbauer, in *Desorption Induced by Electronic Transitions, DIET I*, edited by N. H. Tolk, M. M. Traum, J. C. Tully, and T. E. Madey, Springer Series in Chemical Physics, Vol. 24 (Springer, Berlin, 1983).

⁶N. R. Avery, W. H. Weinberg, A. B. Anton, and B. H. Toby, Phys. Rev. Lett. **51**, 682 (1983); B. A. Sexton, Surf. Sci. **88**, 319 (1979), and Chem. Phys. Lett. **65**, 469 (1979); E. M. Stuve, R. J. Madix, and B. A. Sexton, Chem. Phys. Lett. **89**, 48 (1982).

 7 V. Ponec, Catal. Rev. Sci. Eng. **18**, 151 (1978), and references cited therein.

⁸W. A. Herrmann, H. Biersack, M. L. Ziegler, K. Weidenhammer, R. Siegel, and D. Rehder, J. Am. Chem. Soc. **103**, 1692 (1981).

⁹Z. Miskovic, J. Vukanic, and T. E. Madey, Surf. Sci. **141**, 285 (1984).

¹⁰C. Benndorf and T. E. Madey, to be published.