High-Pressure Structural Phase Transitions in Na, Mg, and Al

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First-principles total-energy calculations by two independent methods predict new transition-metal-like sequences of stable crystal structures in the third-period simple metals Na, Mg, and Al, under increasing pressure. These structural phase transitions are controlled by d electrons through the lowering and partial filling of the initially empty $3d$ band as the metals are compressed. ^A number of these transitions occur in the megabar range and below and could be investigated by existing experimental techniques.

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There has been a rapidly growing interest in understanding the systematics of crystal phase understanding the systematics of crystal phase
stability in elemental metals^{1–10} and their binar stability in elemental metals¹⁻¹⁰ and their binary compounds.^{11, 12} For nonsimple metals, a recurcompounds.^{11, 12} For nonsimple metals, a recurring theme in most,^{3-9, 11} although not all, ^{10, 12} of this work is the critical role played by the d electrons in explaining many of the observed trends. We present here theoretical evidence that under sufficiently high pressure the controlling influence of d electrons on crystal structure is extended to even simple metals and we predict new, as yet undetected, structural phase transitions in the third-period metals Na, Mg, and Al. Specifically, we have performed a series of first-principles total-energy calculations which suggest the possible sequence of structures $hep \rightarrow bce \rightarrow hep$ in Na, hcp \rightarrow bcc \rightarrow fcc in Mg, and fcc \rightarrow hcp \rightarrow bcc in Al under increasing pressure. These sequences are similar to those observed in transition-series metals with increasing atomic number and, at least for Mg and Al, result from the lowering and partial filling of the initially unoccupied $3d$ band as the metals are compressed. In addition, for Mg and Al we have obtained the same qualitative results from two entirely different techniques: the gener alized pseudopotential theory $\rm (GPT)^6$ and the linear-muffin-tin-orbitals $\rm (LMTO)$ method.¹³ method.¹³

With both the GPT and LMTO methods, our analysis is based on calculation of the zero-temperature total energy of the solid as a function of atomic volume and crystal structure,¹⁴ within the general framework of the Kohn-Sham localthe general framework of the Kohn-Sham loct
density formalism,¹⁵ using the Hedin-Lundqvi
prescription for exchange and correlation.¹⁶ prescription for exchange and correlation. Beyond that common starting point, however, the two methods utilize quite different secondary approximations. In the GPT, all three metals are treated in the simple-metal limit of the theory, which represents a refined version of the conventional nonlocal pseudopotential perturba-

tion theory.^{1,6} A rigorous, optimized pseudo potential of the orthogonalized-plane-wave type is employed. The two important approximations in this case are the small-core treatment of the inner-core electrons (1s, 2s, and $2p$ here) and the neglect of third- and higher-order (structuredependent) terms in the total energy expansion. In the LMTO method, on the other hand, we em-
nloy the atomic-sphere approximation $(ASA)^{13}$ ploy the atomic-sphere approximation (ASA) ,¹³ in which the electron density is spherically averaged within the Wigner-Seitz spheres. The LMTO method effectively retains the higherorder terms discarded in the GPT, but at the same time introduces the structure-dependent ASA which the GPT does not. Except as indicated, the present LMTO calculations were carried out
as described in previous work on silicon,¹⁷ with as described in previous work on silicon, $^{\mathrm{17}}$ with all electrons treated self-consistently and s , p , and d components retained in the angular momentum basis. The LMTO structural energy differences are insensitive, over the volume range considered, to whether the $2s$ and $2p$ states are treated as bands or as atomic levels, indicating that the GPT small-core approximation for these electrons is adequate.

Total-energy calculations with use of the GPT and the LMTO methods have been carried out for fcc, bcc, and (ideal) hcp structures, over a range in atomic volume Ω extending from normal density $(\Omega = \Omega_0)$ down to the vicinity of tenfold compression $(\Omega/\Omega_0=0.1)$. The resultant structural energy differences, relative to the fcc phase, are plotted in Figs. 1 and 2. Figures $1(a)$ and 1(b) show our GPT results for Na and Mg, respectively, while Figs. 2(a) and 2(b) give a direct comparison of our LMTO and GPT results for Al. As can be seen in Fig. 2, the qualitative agreement between GPT and LMTO predictions about phase stability in Al is excellent: Both show large regions of fcc stability near normal volume and bcc stability at high compression

FIG. 1. Relative total energies of bcc, hcp, and fcc structures vs volume, as calculated by the GPT method. (a) Na and (b) Mg.

separated by a small island of hcp stability. The qualitative agreement for Mg is comparable. No LMTO calculations on Na were attempted because of the tiny energy differences (typically < 0.1 mRy) between phases in that metal.

The structural energy differences in Mg and Al are profoundly influenced at high densities by the presence of d electrons. Under compression, the initially empty $3d$ band moves downward towards the Fermi level E_F , narrowing relative to the sp bands in the process. In Al, the LMTO X_s level (approximately the bottom of the $3d$ band) moves below E_F at $\Omega/\Omega_0 = 0.15$, at which point each Wigner-Seitz cell contains about one d electron. We have directly assessed the effect of this d charge by repeating the LMTO calculations with d components removed from the angular momentum basis. The effect is striking, as shown in Fig. 3 for Al. The resultant bcc-fcc and hcpfcc total energy differences now remain positive throughout the whole volume range considered. Analogous behavior is found for Mg. Thus $with$ out d electrons Mg and Al would remain hcp and fcc, respectively, down to at least $\Omega/\Omega_0 = 0.1$. Moreover, the *addition* of f components to the basis changes the results in Figs. $2(a)$ and 3 only slightly.

The increasing quantitative discrepancies between GPT and LMTO results with decreasing volume, as seen in Figs. 2 and 3, are probably due to differing treatments of d states. While the LMTO method is presumably treating the d states accurately regardless of their spatial character (i.e., either nearly-free-electron or tight-binding like), the GPT, as applied here in the simple-metal limit, is treating the d states accurately only to the extent that they remain nearly-free-electron like. Indeed, the purely

FIG. 2. Relative total energies of the bcc, hcp, and fcc structures of Al vs volume, as calculated by (a) the LMTO methods, and (b) the GPT method.

p-like X_4 , and L_2 , levels are calculated to within a few percent of each other by the two techniques at all volumes. The sd hybridized X_1 level, on the other hand, shows close agreement only near normal volume, with the LMTO value moving progressively lower than the GPT value upon compression. This lowering is exactly the type

FIG. 3. The bcc-fcc total energy difference for Al vs volume, as calculated by the GPT method and by the LMTO method with and without d states.

of hybridization effect one expects from a narrowed d band centered above E_F and suggests that the $3d$ band in Al is slowly developing some tight-binding-like character under pressure, which the present simple-metal form of the GPT is not detecting. The natural remedy to this situation is to treat Al as an empty- d -band metal in the manner of the heavy alkali and alkaline-earth metals.⁶ This possibility is currently under study.

At lower densities it is possible that the ASA in the LMTO method is most responsible for quantitative differences between our GPT and LMTO results. At normal density both methods correctly identify the observed stable structure in each case and the calculated hcp-fcc energy differences are in good quantitative agreement. The GPT energy differences involving the bcc structure, however, are always larger and generally in better agreement with the available experimental data than the corresponding LMTO values, as shown in Table I. This is consistent with the expectation that spherical averaging inherent in the ASA should be better for the closepacked structures than for the more open bcc structure.

An interesting common feature of our results is a predicted region of bcc stability in all three metals under pressure. This is reminiscent of

TABLE I. Normal density structural energy differences for Na, Mg, and Al as calculated by the GPT and the LMTO methods, and compared with available experimental data. Energy differences are in millirydbergs.

	bec-fee	hcp-fcc	bcc-hcp
		Na	
GPT Expt.	0.055	-0.010	0.065 $0.055^{\rm a}$
		Mg	
LMTO GPT Expt.	0.2 1.5 2.0 ^b	-0.7 -0.6 -1.5^{b}	0.9 2.1 $3.5^{\rm b}$
		Al	
LMTO GPT Expt.	3.9 7.3 7.7 ^b	1.9 1.9 4.2 ^b	2.0 5.4 $3.5^{\rm b}$

 a Measured heat of transformation (0.032, Ref. 18) less the calculated bce-hcp difference in zero-point vibrational energies $(-0.023, \text{ Ref. } 19)$.

^bThermodynamically based estimates of Ref. 20.

the occurrence of the bcc phase in the center of the transition metals, which has been ascribed the occurrence of the bcc phase in the center of
the transition metals, which has been ascribed
to the half filling of a narrow d band.^{3,4} The true origin of our bcc stability, however, appears to be more subtle. For example, the point of greatest bcc stability in Al $(\Omega/\Omega_0 \sim 0.15$ for our LMTO calculation) corresponds to only a 10% filling of the $3d$ band. Moreover, the effect of d -band occupation is generally opposite in Al and Mg, tending to stabilize the bcc structure in Al, while destabilizing that structure in Mg near Ω/Ω_0 $=0.15$. We have analyzed this situation in detail and believe it can be understood from the oneelectron density of states. As suggested by Andersen's force relation, 4 we find the structural energy differences within the ASA well approximated by just the difference in (occupied) oneelectron eigenvalue sums, provided the same potential is used for each lattice structure. We have calculated LMTO eigenvalue sums for Mg and Al, using the respective convergent fcc potentials for $\Omega/\Omega_0 = 0.15$. For *both* the Mg and Al sums, we find fcc stability for two-electron occupation (i.e., Mg) and bcc stability for three-electron occupation (i.e., Al). The former occurs because the lowering of the $3d$ band causes semimetal behavior near $\Omega/\Omega_0 = 0.15$, creating a large dip in the fcc density of states at the two-electron E_F , in exactly the manner reported for fcc Ca at larger volumes. 21 Neither the bcc nor hcp structures have such dips in their density of states. Relative to these structures, therefore, the fcc structure has state density shifted away from the two-electron E_F down to lower energies, giving it the lowest eigenvalue sum. Since the dip in the fcc density of states also implies state density shifted to higher energies above the two-electron E_F , this same argument would make the fcc structure unfavorable if another valence electron were added. This is precisely what happens for Al.

Lastly, we estimate the initial hcp \rightarrow bcc transitions in Na and Mg, and $fcc + hep$ transition in Al, to occur at pressures of about 10, 570, and 1300 kbar at zero temperature, corresponding to $\Omega/$ kbar at zero temperature, corresponding to Ω / Ω ₀ = 0.86, 0.56, and 0.58, respectively.¹⁴ While Na is complicated by a temperature-induced hcp \rightarrow bcc transition at 36 K and ambient pressure.¹⁸ \rightarrow bcc transition at 36 K and ambient pressure,¹⁸ high-pressure bcc stability has been observed at 77 K up to 600 kbar.²² There is also evidence at room temperature of anomalous behavior in Mg
near 100 kbar,²³ and possibly a partial fcc-to-ho near 100 kbar, $^\mathrm{23}$ and possibly a partial fcc-to-hc $_\mathrm{I}$ near 100 kbar,²³ and possibly a partial fcc-to-h
transition in Al at about 205 kbar.²⁴ Since high pressure technology has improved dramatically in recent years, new experimental investigation

of all three elements would be of considerable interest.

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Flow of 3 He-B through Narrow Channels

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The critical current J_c of superfluid ${}^{3}\text{He}-B$ through 0.8- μ m-diam channels has been measured. For small currents the pressure difference $\Delta P = 0$ along the flow channels within the resolution, implying small or zero dissipation. ΔP grows rapidly with increasing current above J_c ; a clear transition to dissipative flow is thus observed. The temperature dependence of J_c indicates that the superfluid density and the critical temperature are reduced inside the narrow flow channels.

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The most important feature of a simple superfluid is that it can sustain mass flow without friction. At some critical current J_c , however, the superfluid state becomes unstable, which leads

to dissipation. This model was derived from experiments on He II; its validity in the case of 3 He- B is currently of considerable interest. Parpia and Reppy¹ have observed the onset of