Stoichiometric and Structural Origin of Electronic States at the Pd₂Si-Si Interface

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Extra Si density of states has been observed within about 3–4 Å of the Pd₂Si-Si(111) interface. Spectral analysis indicated that most of these states exist near the Si band-gap region and originate from an atomic environment more Si rich than $Pd₂Si$. Transmissionelectron-microscopy lattice images showed a structurally sharp Pd₂Si-Si interface with misfit dislocations and atomic-step imperfections. It is suggested that the interfacial bonding in such a structure can account for the observed interface states.

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Knowledge on the nature of the electronic properties at the metal-semiconductor interface is important for the basic understanding of Schottkybarrier formation. Recent experiments show that the effects of interfacial reactions, such as atomic diffusion¹ and defect generation,² are important for determining the interfacial electronic properties. The interface between a transition metal and Si forms a distinct class by itself because of the strong reactivity of these systems, which causes the formation of silicide compounds. During silicide formation, the reaction front proceeds into the Si, resulting in a silicide-Si interface with bonding and structural characteristics basically different from those of nonreacting metal-Si interfaces. Recent spectroscopy studies, $3 - 7$ which use ultraviolet photoelectron spectroscopy (UPS), Auger-electron spectroscopy (AES), and x-ray photoemission spectroscopy (XPS) on the interfaces of Pd, Pt, and Ni on Si, revealed that silicide formation in the first few layers of metal coverage indeed dominates the electronic properties of such interfaces.

Among the silicides, the Pd-Si system has attracted the most interest. In bulk Pd,Si, the electronic structure has been investigated with use of UPS and AES,⁸ where an analysis of the spectroscopic data based on the calculated partial densities of states showed that the filled Pd 4d band shifted 2.75 eV below the Fermi level $(E_{\text{F}} \equiv 0)$ and the bonding in the Pd, Si lattice is primarily due to the hybridization of the Pd $4d$ and Si $3p$ electrons which form separate groups of bonding (at about -5 eV) and antibonding states (near $E_{\rm F}$). At the Pd₂Si-Si interface, UPS observation during the initial formation stage revealed a valenceband structure similar to the bulk silicide but with the major peak in the d band starting from —3.⁵ eV and then shifting to the bulk value of -2.75 eV. This peak shift and the corresponding

surface work-function change were attributed to a stoichiometry gradient with Si enrichment near the interface.³ Similar conclusions have been suggested for the Pt-Si and Ni-Si interfaces. Indeed, it has been proposed that a Si-rich stoichiometry gradient with the $MSi₄$ composition at the interface can be used as a basis to confirm the Schottky model for silicide-Si interface.

Several relevant questions concerning the interfacial stoichiometry gradient have to be resolved. First, since transmission electron microscopy (TEM) revealed only the $Pd₂Si$ phase at the initial (TEM) reveated only the Pu₂SI phase at the initial (~5 Å) Pd/Si interface,⁵ it is difficult to justif thermodynamically a large stoichiometry variation within a single phase layer of a few angstroms; so far there has been no direct stoichiometry measurement within, e.g., 10 Å of the interface. Second, since surface spectroscopy techniques sense only the local atomic environment, the data alone do not reveal the overall atomic structure of the interface. And third, there remains an important, yet unanswered, question on the nature of the electronic states associated with the stoichiometry variation. In an attempt to resolve these questions, we have carried out a direct composition and structure determination for the $Pd_2Si-Si(111)$ interface using Auger-intensity calibration and cross-sectional TEM. For these studies, the Pd-Si(111) interface is well suited. First, the difference in the line shapes of the Si $L_{2,3}VV$ Auger transition is sufficient for separating the silicide intensity from that of the Si substrate, a condition required for composition calibration. And second, the Pd,Si formed is epitaxial to the substrate, making it possible to observe the interfacial structure with μ se of lattice-imaging techniques. Experimental variable of lattice-imaging techniques. Experimental variable $\frac{3}{2}$, $\frac{8}{2}$ details have been previously published.^{3,5,8}

The composition calibration is based on the variation of the integrated intensities for Si $L_{2,3}VV$,

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FIG. 1. (a) Variations of the integrated intensities of Pd MVV and Si KLL and LVV Auger lines for increasing coverages of Pd_2Si ; (b) intensity variations of the substrate Si, silicide, and interface portions of the Si \it{LVV} spectrum for increasing coverages of Pd_2Si ($\lambda = 8.4$ \AA).

 KLL , and Pd $M_{4,5}VV$ as a function of silicide coverage and the depth sensitivity comes from the different escape distances (λ) of these Auger transitions. The intensity variations are first formulated for various types of surface coverformulated for various types of surface cover
age,¹⁰ e.g., uniform or island coverage of the substrate by a silicide of a certain composition. Then a match between the predicted and the observed intensity variations determines the composition and the escape distances.

Figure 1(a) shows the variations of the Auger intensities observed in one experiment as a function of thickness. The MVV and KLL intensity variations (normalized to bulk silicide and Si, respectively) indicate an overall composition of Pd₂Si with characteristic λ of 11.7 and 28 Å, respectively. The analysis of the LVV intensity variation is carried out by first subtracting the

ELECTRON ENERGY (eV)

FIG. 2. (a) The silicide portion of the Si $L_{2,3}VV$ Auger spectrum for increasing Pd_2Si coverage; (b) the interface portion of the Si LVV spectrum as a function of coverage.

Si-substrate contribution. In Fig. 2(a), the variation of the LVV spectrum after subtraction is displayed for increasing coverages. At 20 \AA , the spectrum is essentially that of bulk Pd,Si, but for low coverages the overall feature is different, particularly for the portion in 91-96 eV. In addition, the integrated intensities consistently exceed those predicted for a Pd,Si layer. Further decomposition¹⁰ by subtracting the Pd₂Si contribution in these spectra reveals extra features [Fig. 2(b)] which seem to be invariant for all coverages but characteristically different from those of the substrate and Pd,Si. Moreover, the intensity of these spectra decreases with coverage in a manner similar to that of the substrate $[Fig. 1(b)],$ and therefore we attribute this to an interface contribution. The coverage dependence of the extra intensity (note the maximum at about 3 Å) is consistent with a depth distribution of the addiconsistent with a depth distribution of the addi-
tional states within about 3 Å from the interface.¹¹

There are two other pieces of experimental evidence indicating an interface layer within a few idence indicating an interface layer within a few
angstroms in the silicide.¹⁰ First, measuremen of the surface work function by a Kelvin-probe technique indicated that the increase of the work function from that of the Si surface to the $Pd₂Si$ value was saturated after about 3 Å of coverage. And second, the energy shift observed for the MVV peak was consistent with the presence of Pd atoms in a Si-rich interface layer of about $4-5$ Å wide.

The interface spectrum $[Fig. 2(b)]$ is dominated by a peak at 94 eV which has been shown to originate from the antibonding states. 8 Since the antibonding states locate mostly within the band gap, and so a major portion of the observed interface states are most probably band-gap states derived from Si atoms. Furthermore, the shift of the 94 eV interface peak toward higher energy for increasing coverage can be related to a decreasing occupation of the antibonding states caused by shifts in E_F . Based on the calculated stoichiometry dependence of the density of states, this re-
flects a Si-rich environment at the interface.¹² flects a Si-rich environment at the interface. We note that the extra states occur with relatively high density of about 0.1 of the interface Si atoms, which is insufficient, however, to account for an interfacial silicide layer of high silicon concentration, e.g., $PdSi₄$.

In Fig. 3(a) we show the TEM lattice image of the $Pd₂Si-Si(111)$ interface as obtained from a sample where Pd₂Si was formed by evaporating \sim 100 Å of Pd on a chemically cleaned surface. In Si the image shows the $\{111\}$ and $\{220\}$ planes

which are interwoven to yield the $\langle 110 \rangle$ channels. while in Pd₂Si the $\{22\overline{4}0\}$ planes are observed. These lattice images clearly extend from both sides to the interface, indicating that the lattice structures of Pd, Si and Si reach the interface with no evidence for a different interface compound. However, even for a structurally sharp Pd,Si-Si (111) interface, there is a transition region in general consisting of approximately one Si and

 (b)

FIG. 3. (a) TEM lattice images observed in a vertical cross section of the $Pd_2Si-Si(111)$ interface; (b) schematic illustration of the lattice structure in a $Pd_2Si Si(111)$ interface.

two $Pd₂Si layers because of the *AB*-type stacking$ of the hcp Pd, Si lattice as shown schematically in Fig. 3(b). Within this interface region of $3-4 \text{ Å}$, the atomic environment is Si rich compared to bulk Pd,Si and the chemical bonding is different as well. Thus, it appears that the chemical bonding contained in such an interface structure should have the necessary characteristics to account for the electronic states observed at the interface. In addition, one can distinguish two types of structural imperfections at the interface: the misfit dislocation and the atom steps [shown at sites a and b in Fig. 3(b), respectively]. Both types of defects can change the local bonding configuration without varying the overall lattice structure, thereby inducing interface states different from those associated with the flat interface. While the electronic properties of the different interfacial bonds are interesting for future studies, qualitatively the range of the interface layer and its Si-rich environment as indicated in Fig. 3(b) provide a satisfactory explanation for the extra Si density of states observed at the Pd, $Si-Si(111)$ interface.

In summary, our results show directly the existence of Si interface states derived from a Si-rich environment within a range of about two to three atomic layers at the Pd,Si-Si interface. The origin of these states can be qualitatively accounted for by the atomic structure of the silicide-Si interface. The fact that a major portion of the interface states lies in the band-gap region show that they are important for an understanding of Schottky-barrier formation at the silicide-Si interface. This report appears to be the first correlation between stoichiometry, microstructure, and electronic states at a silicide-Si interface.

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 11 Complications due to nonuniform surface coverages have been considered. While a partial coverage with islands of uniform thickness can be ruled out readily, it is difficult to rule out a complete coverage by islands with random size and thickness. For this case, the analysis based on a uniform coverage is still valid, provided that the layer thickness and λ are replaced by the proper statistical averages. We have attempted to model the stoichiometry distribution within the interface layer but found it difficult to ascertain the uniqueness of such calculations; so the results are not reported.

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FIG. 3. (a) TEM lattice images observed in a vertical cross section of the Pd₂Si-Si(111) interface; (b) schematic illustration of the lattice structure in a $Pd_2Si Si(111)$ interface.

 (b)