PHYSICAL REVIEW **LETTERS**

VOLUME 19 NUMBER 4

HIGH-ACCURACY UPPER AND LOWER BOUNDS FOR EIGENVALUES ILLUSTRATED WITH H_2^+ \dagger

C. M. Rosenthal* and E. Bright Wilson, Jr. Department of Chemistry, Harvard University, Cambridge, Massachusetts {H.eceived 19 June 1967)

We have calculated, for the ground state of the hydrogen molecule ion H_2^+ at an internuclear distance of $R = 2$ a.u., upper and lower bound to the nonrelativistic Schrödinger energy E which differ in the eleventh significant figure. This calculation was carried out to illustrate a method which we are developing with the following advantages: Upper and lower bounds of comparable accuracy are obtained, and this accuracy may be carried as far as desired; no integrations of any kind are used; the algebra required and computer programming are elementary; no secular equations need be solved excited states as well as ground states can be treated; and the wave equation need not be separable. In this method the wave function was expressed as an infinite series

$$
\Psi = e^{-\delta \xi} (\xi + 1)^{\gamma} \sum a_{kl} [(\xi - 1) / (\xi + 1)]^{k} (1 - \eta^{2})^{l}
$$

in which ξ and η are the usual confocal elliptic coordinates. Insertion of this expansion into the wave equation leads to the following recursion relation for the coefficients:

$$
[-2k(k-1) + 2k(\gamma - 2\delta - 1) - 4l^2 - 2l + \gamma(1 + 2\delta)]a_{k,l}
$$

+ $(k+1)^2 a_{k+1,l} + [\gamma^2 - 2(k-1)\gamma$
+ $(k-1)^2]a_{k-1,l} + 4(l+1)^2 a_{k,l+1}$
+ $\alpha a_{k,l-1} = 0$,

where $\delta = (-\frac{1}{2}R^2E)^{1/2}$, $\alpha = \frac{1}{2}R^2E$, and $\gamma = R/\delta$ Asymptotically, we expect the lowest eigenfunc- $1.$ tion at a large distance to approach spherical symmetry and to die off exponentially in the variable ξ . If the wrong value of the energy is inserted in the recursion relations and the above limiting spherical symmetry is imposed, we know that the solution will go to either plus or minus infinity at infinite radius depending upon whether E is larger than or smaller than a nearby eigenvalue. We impose this limiting spherical symmetry by truncating the equations at some order N (say, $N=10$), where we have set the coefficients $a_{k,l}$ to zero for $k+l=N_l$ except for a_{N0} . With this truncation the recursion formula gives a set of simultaneous equations which can be solved for a_{N0} , and for all the coefficients of order less than N , in terms of $a_{00}=1$. To solve the simultaneous equations, we used an efficient procedure based on their special form. The solutions of these simultaneous equations will change as we change N. From their sign, their size, and the way in which they change by increasing the order of truncation, one. can judge whether the value of E used in the computation is less than or greater than the true eigenvalue.

Wind' gives the ground-state energy as -1.¹⁰² ⁶³⁴ ² a.u. at this interatomic distance. Table I shows that for N large enough and the energy too low, the coefficients all increase or stay constant as the order of truncation in-

 $=$

 $=$

creases. They first decrease and then increase as the index r increases for a given order of truncation. Consequently, as the order of truncation is extended to infinity, we see that the solution of the differential equation would the solution of the differential equation would
ultimately go to +∞ for η =0 and ξ → ∞. Similar ly for an energy too high, the coefficients eventually all become algebraically smaller or stay constant as the order of truncation is increased, and first diminish in absolute value and then increase in absolute value as the index is increased. Consequently, as the order of truncation is extended to infinity, the solution goes to $-\infty$ for $\eta = 0$ and $\xi \rightarrow \infty$. This behavior of the solution shows that we have bracketed the true eigenvalue. Greater accuracy could be obtained if needed by subdividing the grid for E and repeating the calculation. The result agrees with Wind's value to the accuracy he claimed. In

addition to the eigenvalue, this method provides a rapidly convergent expansion for the eigenfunction and hence the capability of providing expectation values of known accuracy. Less than a minute and a half of IBM-7094 time was required.

We have also tried this method on a number of simple one-dimensional problems, where it works extremely efficiently to give rigorous upper and lower bounds of almost any desired accuracy, and we are exploring its possibilities for application to more involved problems.

)This research was supported in part by a Grant extended Harvard University by the National Science Foundation, Grant No. GP-6413.

*National Science Foundation Predoctoral Fellow. ¹H. Wind, J. Chem. Phys. 42, 2372 (1965).

SPIN-ORBIT COUPLING PARAMETER FOR Mn^{2+} IN RbMnF₃

Anjani Mehra and Putcha Venkateswarlu Department of Physics, Indian Institute of Technology, Kanpur, India (Received 26 June 1967)

Fine structure observed in the ${}^6A_{1g}(S) \rightarrow {}^4T_{2g}(D)$ band in the absorption spectrum of RbMnF₃ at 77°K has been assigned to the spin-orbit splitting of the ${}^4T_{2g}(\mathcal{D})$ level, and the analysis yields a value of 320 ± 10 cm⁻¹ for the parameter ξ_{3d} .

The experimental value of the spin-orbit coupling parameter ξ for Mn²⁺ having d^5 configuration is known neither for the free ion nor for any crystal. In the case of the free ion, the spin-orbit splitting is of second order (halffilled shell), and it is not possible to obtain a reliable value of ξ by a least-squares fit to the spectrum of free Mn^{2+} ion. Manganous salts that have usually been studied have a lower than cubic symmetry, and because of the uncertainties in finding the contribution from the lower symmetry components of the crystalline field, it has not been possible to obtain the value of ξ from the fine structure observe in the absorption bands.¹ In order to avoid this difficulty, RbMnF_s crystal, which has a cubic structure, 2 was chosen for the present study.

The absorption band having ${}^4T_{2g}(D)$ as upper state is found to be particularly suitable since the spin-orbit splitting of ${}^4T_{2\alpha}(D)$ level is the-The spin-orbit splitting of $\frac{1}{2}$ and $\frac{1}{2}$ or even is the over, as this level has a small slope in the

energy-level diagram, it is expected that these small splittings will be resolved in the spectra taken at low temperatures.

Under the spin-orbit interaction a ${}^4T_{2g}$ level splits as

$$
T_{2g} \otimes \Gamma_8 = \Gamma_6 + \Gamma_7 + 2\Gamma_8.
$$

First-order energies of the split levels using fictitious $L' = 1$ and $S = \frac{3}{2}$ are

$$
E = 5a(\Gamma_7), \ 2a(\Gamma_8), \text{ and } -3a(\Gamma_6 + \Gamma_8),
$$

where a is a proportionality constant.³ The exact calculations of the spin-orbit splitting involve parameters B , C , Dq , α (Trees correction parameter), and ξ . Details of these calculations are described elsewhere.⁴ The dimensions of the matrices are the following'. Γ_6 , 20×20; Γ_7 , 22×22; and Γ_8 , 42×42.⁵ Values of B, C, Dq, and α used in these calculations were obtained by fitting to the observed band energies in the spectrum taken at $77^\circ K$, and the values are 840, 3080, 780, and 76 cm^{-1} ,