

CRYSTAL-FIELD SPLITTING OF ORTHOHYDROGEN IN SOLID PARAHYDROGEN*

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From nmr measurements we have estimated the crystal-field splitting of an ortho-H₂ molecule in solid para-H₂ to be 8.2×10^{-3} °K. This lowers the existing upper limit on the splitting by a factor 5.

There has been considerable speculation in the past decade as to the magnitude of the splitting, ϵ , of the $J=1$ rotational state of ortho-H₂ in solid para-H₂. Nakamura¹ fitted the specific-heat data of Hill and Ricketson² with an expression containing a linear and a quadratic term in the ortho-H₂ concentration. Assuming a two-level model ($m_J=0$ level below the $m_J=\pm 1$ level) he predicted a value $\epsilon/k=2.2$ °K from the coefficient of the linear term. In a theoretical calculation, Nakamura and Uryu³ used the Einstein model of lattice vibrations to estimate ϵ and obtained $\epsilon/k=0.52$ °K. Ortung⁴, using a more refined theory of the anomalous heat capacity and additional specific-heat data, predicted $\epsilon/k=2.6$ °K for a two-level model and $\epsilon/k=1.89$ °K for a three-level model. Sears and Van Kranendonk⁵ were able to estimate ϵ from the difference in the separations of the components of the rotational Raman triplet $S_0(0)$ ($\Delta v=0$, $\Delta J=2$) in pure para-H₂.^{6,7} Under the assumption that the crystalline potential giving rise to the splitting is the same for the $J=1$ and $J=2$ states, they obtained $\epsilon/k=0.03 \pm 0.02$ °K. Recently, Van Kranendonk and Sears⁸ have presented a detailed theory of the interaction between the rotational motion of the molecules and the lattice vibrations, and gave a revised value of 0.29 °K for ϵ/k . These authors have emphasized the large uncertainty in this estimate due to uncertainty in the parameters of the intermolecular potential. Harris and Hunt⁹ have deduced a value of about 0.3 °K for the rms splitting of the $J=1$ level from measurements of the second moment of the nmr line in solid H₂. However, as we shall see later, at the high ortho-H₂ concentrations where their measurements were made, the splitting of the $J=1$ level is completely dominated by the electric quadrupole interactions between ortho molecules. The present note

describes a measurement of ϵ/k by observation of the nmr free-induction decay at low temperatures in samples with low ortho-H₂ concentrations.

Consider an ortho-H₂ molecule in an infinite crystal of para-H₂: Because of the hexagonal close-packed structure of para-H₂, the $J=1$ level of ortho-H₂ is expected to be split into three sublevels labeled by \bar{m}_J , the projection of the angular momentum along the crystal c axis, with the $\bar{m}_J=0$ state separated from the $\bar{m}_J=\pm 1$ levels by the crystal-field energy ϵ .^{5,8} The value of ϵ is dependent on the c/a ratio of the crystal and is zero for an ideal, rigid hcp structure.⁵ Following Reif and Purcell,¹⁰ the nmr spectrum of such a molecule is described by the energy-level scheme

$$h^{-1}E_{m_I, \bar{m}_J} = am_I + c\gamma m_I \bar{m}_J + d(3\gamma^2 - 1)(1 - \frac{3}{2}\bar{m}_J^2)(1 - \frac{3}{2}m_I^2), \quad (1)$$

where a is the Larmor frequency of the protons (20 MHz in our case) and $c=113.8$ kHz and $d=57.7$ kHz express the spin rotation and dipolar interactions within the molecule, respectively. m_I is the projection of the total nuclear angular momentum along the external field, and γ is the cosine of the angle between the crystal c axis and the magnetic field.

The nmr transitions $m_I=-1 \rightarrow m_I=0$ and $m_I=0 \rightarrow m_I=1$ are both split into three components corresponding to the three values of \bar{m}_J . However, if the rate of transitions among the molecular \bar{m}_J levels is fast compared with these splittings,¹¹ then each of the triplets is reduced to a single line having a frequency equal to the average over the \bar{m}_J states, or

$$v_{\text{Larmor}} = a \pm \frac{3}{4}d(3\gamma^2 - 1)[2p_0 - (p_1 + p_{-1})], \quad (2)$$

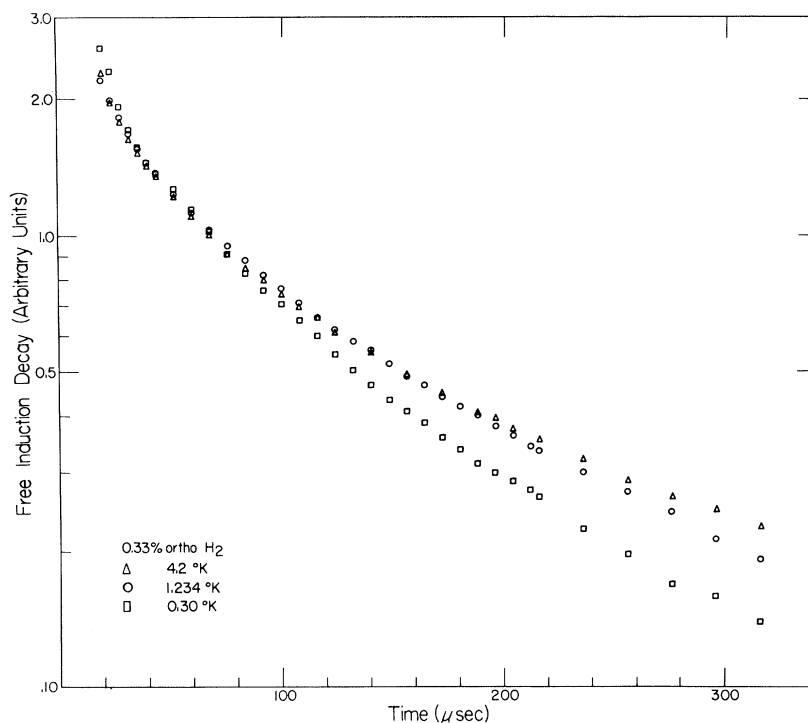


FIG. 1. Detected free-induction decay in solid H₂ with 0.33% ortho.

where p_0 , p_1 , and p_{-1} designate the populations of the states $\bar{m}_J=0, 1$, and -1 , respectively. The angular dependence of the splitting is just that of a normal Pake doublet¹²; the magnitude of the splitting, however, is temperature dependent. The envelope of the Fourier transform of a Pake doublet averaged over a polycrystalline sample is given by¹³

$$G(t) = \left(\frac{\pi}{6}\right)^{1/2} \left\{ \frac{\cos \alpha t}{(\alpha t)^{1/2}} C(3\alpha t) + \frac{\sin \alpha t}{(\alpha t)^{1/2}} S(3\alpha t) \right\}, \quad (3)$$

where $\alpha = \frac{3}{4} d[2p_0 - (p_1 + p_{-1})]$, and C and S are the real and imaginary parts of the Fresnel integral ($3\alpha t = z$ in the tables of Jahnke and Emde¹⁴). Our intent was to lower the temperature of the sample so that by observation of the nmr free-induction decay the zeros of the function $G(t)$ and hence the values of $[2p_0 - (p_1 + p_{-1})]$, and then ϵ , could be determined.

The detected free-induction decay in H₂ with 0.33% ortho at 4.2, 1.23, and 0.3°K is given in Fig. 1, suitably normalized to account for changes in temperature. In order to extract the temperature-dependent part of the free-induction decay we have divided the signal at 0.3°K by the high-temperature signal (i.e., signal at 4.2°K) and the result is shown in Fig.

2 (curve A). It is evident that even at 0.3°K the first zero of $G(t)$ has not been observed. [The small oscillation in the detected signal

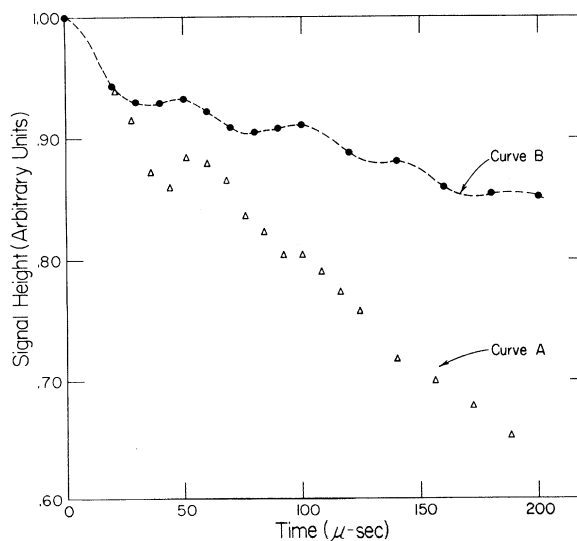


FIG. 2. Temperature-dependent part of free induction decay. Curve A obtained by dividing signal at 0.3°K by signal at 4.2°K. Curve B shows effect of residual quadrupole-quadrupole interactions at 0.3°K (theoretical).

near the origin ($t < 60 \mu\text{sec}$) has been shown to be the result of a residual nmr signal from the empty sample cell.] Provided allowances are made for all other sources of broadening, it is still possible to deduce a value for ϵ .

Broadening from inhomogeneities in the external magnetic field and from residual intermolecular dipolar interactions is only a slight perturbation on the natural linewidth and in addition can be safely assumed to be temperature independent. The main source of broadening is due to the intramolecular dipolar and spin-rotational interactions, i.e., the same interactions giving rise to the splitting of the nmr line. If it is assumed that the molecular transition rates are not intrinsically temperature dependent, then it can be shown that the fractional change in broadening due to depopulation of the upper \bar{m}_J level (or levels) is of the order of $(\epsilon/kT)^2$, and this is negligible a posteriori. In the temperature range 4.2 to 0.3°K one expects the change in molecular re-orientation rates to be small, as the majority of the molecules in a 0.3% ortho sample have nearest-neighbor ortho-ortho interactions smaller than 10^{-3} °K. However, those ortho molecules having a sizable electric quadrupole coupling with a neighboring ortho molecule do exhibit a temperature-dependent line shape and we have devised a simple model to estimate their effect on the free-induction decay.

Consider a single pair of ortho- H_2 molecules in a crystal of para- H_2 and for simplicity assume that the crystal-field splitting due to ortho-para interactions is zero. The energy level scheme of a pair of ortho- H_2 molecules has been calculated by Orttung⁴ and consists of nine levels (several of which are degenerate) having a total energy span of 7.43°K for ortho molecules at the nearest-neighbor distance, a_0 , in solid para- H_2 .¹⁵ Assuming fast transitions among the molecular levels, one again finds a temperature-dependent splitting of the nmr line. For a given temperature it is possible to calculate the free-induction decay characteristic of each ortho-ortho separation. If we confine our attention to times $\leq 200 \mu\text{sec}$, then at 0.3°K only pairs of molecules separated by $r^2 \leq 5a_0^2$ contribute significantly to the damping of the free-induction decay. For an ortho concentration of 0.33%, the majority of nearest-neighbor pairs having $r^2 \leq 5a_0^2$ are isolated pairs, i.e., the two molecules are nearest neighbors of each other. Guided by the fact

that the quadrupole-quadrupole interaction is fairly short ranged ($1/r^5$), we have assumed that to a first approximation the interaction of an isolated pair with other ortho molecules can be neglected. It is then possible to sum up all contributions to the free-induction decay (assuming a random distribution for the ortho molecules) and the result is shown in Fig. 2 (curve B).

If the above source of damping is extracted from the experimental free-induction decay by dividing curve A by curve B, then the resulting curve can be compared with the short-time expansion of $G(t)$: $G(t) = 1 - \frac{2}{3}\alpha^2 t^2$.¹⁰ The result gives $\epsilon/k = (8.2 \pm 2.1) \times 10^{-3}$ °K. It is not possible to say whether the $\bar{m}_J = 0$ state lies above or below the $\bar{m}_J = \pm 1$ states since $T \gg \epsilon/k$. Should there be a distribution of values of ϵ our number will correspond to the rms value of ϵ .

We have also measured T_1 at 0.3°K and find it to be equal to the value¹⁶ at 4.2°K to within experimental error. Unfortunately our resolution was very poor (~25%) because of heating of the sample chamber by rf pulses.

The error of $\pm 2.1 \times 10^{-3}$ °K in ϵ/k is not intended to indicate the absolute accuracy of the measurement but only the accuracy of the fit of the corrected free-induction decay to $G(t)$. The procedure used for extracting the desired contribution to the damping of the free-induction decay is an approximation used in the absence of a detailed theory for solid H_2 . We note, however, that if the correction for residual quadrupolar interactions is neglected entirely, the resulting change in ϵ/k is much smaller than the uncertainty. It is quite possible that strains play a role in splitting the $J=1$ level; if so the value we have deduced is an upper limit on the true value of ϵ .

The large value of ϵ obtained from specific-heat measurements^{3,4} may be due to the fact that the quadrupole coupling constant used in correcting for residual quadrupole interactions was in error¹⁵; a definitive answer cannot be given until the effect of the crystal polarizability on the quadrupole-quadrupole interaction at small separations is known. Clustering of the ortho- H_2 molecules,⁵ which results in a larger number of ortho-ortho pairs than would be predicted for a random distribution, could also affect the value of ϵ deduced: The specific-heat data were corrected for residual quadrupolar interactions under the assumption

of a random distribution.¹⁷ If, however, the clustering is due solely to the electrostatic quadrupole-quadrupole interaction, then it can be shown that the deviation from a random distribution is less than 2%.¹⁸ Another possibility is that the presence of an ortho molecule produces a local distortion of the para crystal. This will disturb the c/a ratio locally and can induce a crystal-field splitting in a nearby ortho- H_2 molecule. If we treat the para- H_2 as an isotropic elastic solid, then it is easy to show that the lowest order term in this additional ortho-ortho interaction (from the isotropic part of the disturbance) has a $1/r^3$ dependence on the ortho-ortho separation. Thus although this interaction is expected to be much smaller than the electric quadrupole coupling at short distances, it has a longer range and may be important in samples with low concentrations of ortho- H_2 .

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¹¹For ortho concentrations down to 0.2%, molecular reorientation in solid H_2 is dominated by the adiabatic process of mutual reorientation of the molecular electric quadrupoles. The correlation times are of the order of 10^{-8} sec for a sample having 0.2% ortho. See W. N. Hardy and J. R. Gaines, *Phys. Rev. Letters* **17**, 1278 (1966).

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¹⁵It has been brought to the authors' attention by Dr. J. Van Kranendonk that the recently calculated and presumably very reliable value of the quadrupole moment of H_2 [G. Karl and J. D. Poll, *J. Chem. Phys.* **46**, 2944 (1967)] differs somewhat from the value used by Nakamura (Ref. 1) and later authors (Refs. 4, 5, and 9), and we have used the new value. In addition, Dr. Donald McQuarrie has pointed out that the dielectric properties of the para- H_2 host must be taken into account when applying the free-space electric quadrupole-quadrupole interaction to the case of two ortho- H_2 molecules in solid para- H_2 . When the ortho- H_2 are separated by many lattice distances, the interaction will simply be reduced by a factor equal to the dielectric constant of the medium. It is doubtful that this approximation holds for molecules at the nearest-neighbor distance but this will have little effect on the result of the present calculation. Taking the dielectric constant to be 1.29 for H_2 at 4.2°K, and $a_0 = 3.78 \text{ \AA}$, the value 7.43°K is obtained for the over-all separation of the energy levels.

¹⁶A unique value of T_1 does not exist but $\langle 1/T_1 \rangle^{-1}$ is of the order of 10^{-3} sec. W. N. Hardy and J. R. Gaines, *Phys. Rev. Letters* **17**, 1278 (1966), and to be published.

¹⁷Dr. J. Van Kranendonk brings our attention to the result of H. P. Gush and J. Van Kranendonk, *Can. J. Phys.* **40**, 1461 (1962), who found the effect of the lattice polarizability to be negligible for a nearest-neighbor pair.

¹⁸The authors are indebted to Dr. Donald McQuarrie and Dr. John McTague for discussions on this point.