COVALENCY AND TEMPERATURE DEPENDENCE OF HYPERFINE COUPLING Mn^{2+} IN MgO[†]

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A covalent model is suggested for the calculation of the temperature dependence of hyperfine coupling. The results of calculations on Mn^{2+} in MgO show that covalent admixtures of inner shells form an important contribution to the strength of the temperature dependence.

In a previous paper, one of the present authors and Orbach¹ have proposed a mechanism for the explicit temperature dependence of the hyperfine interaction of Mn^{2+} in cubic crystals. A near-neighbor point-charge model has been adopted to evaluate the phonon-induced admixture of the excited $3d^4ns$ states into the ground $3d^5$ configuration of Mn²⁺ in MgO, and it has been shown that the calculated strength of the variation of the hyperfine constant A is an order of magnitude smaller than that deduced from measurements by Walsh, Jeener, and Bloembergen.² The purpose of the present note is to show that this discrepancy can be reduced by taking into account covalency effects in calculating the phonon-induced hyperfine field. The importance of covalent delocalization of the d electrons for the temperature dependence has been pointed out by Walsh, Jeener, and Bloembergen, who have observed that the smaller the absolute value of A the greater the temperature dependence. We consider a Mn^{2+} ion in a cubic crystal, especially MgO. The σ antibonding orbitals of the $(MnO_{6})^{10}$ octahedral complex are

$$\psi_u = N(\varphi_u - \lambda_\sigma \chi_u)$$
, transforming like $(3z^2 - r^2)$,

$$\psi_{\upsilon} = N(\varphi_{\upsilon} - \lambda_{\sigma} \chi_{\upsilon}), \text{ transforming like } (x^2 - y^2), (1)$$

where φ_u , φ_v are atomic *d* functions and χ_u , χ_v are linear combinations of $2p_\sigma$ oxygen orbitals.³ It has been shown recently^{4,5} that spin density can be transmitted from p_σ ligand orbitals to central ion *s* shells by overlap and transfer effects. In a perfect octahedron the overlap integrals of the orbitals ψ_u and ψ_v with the central ion *s* functions φ_{ns} are zero, but a phonon excitation in the crystal distorts ψ_u and ψ_v so that nonzero overlaps result. These phonon-induced overlaps can be calculated by expanding the atomic $\langle 2p_\sigma | \varphi_{ns} \rangle$ overlaps and λ_σ in terms of nuclear displacements. Using the strain tensor components ϵ_{ij} and retaining only terms linear in ϵ_{ij} , we have

$$\langle \psi_{u}' | \varphi_{ns} \rangle \simeq -\frac{R}{\sqrt{3}} \left[\lambda_{\sigma} \frac{\partial S}{\partial R} + S_{ns} \frac{\partial \lambda_{\sigma}}{\partial R} \right] (3\epsilon_{zz} - \epsilon_{xx} - \epsilon_{yy})$$

$$\equiv a_{ns} (2\epsilon_{zz} - \epsilon_{xx} - \epsilon_{yy}),$$

$$\langle \psi_{v}' | \varphi_{ns} \rangle \simeq -R \left[\lambda_{\sigma} \frac{\partial S}{\partial R} + S_{ns} \frac{\partial \lambda_{\sigma}}{\partial R} \right] (\epsilon_{xx} - \epsilon_{yy})$$

$$\equiv a_{ns} \sqrt{3} (\epsilon_{xx} - \epsilon_{yy}),$$

$$(2)$$

where *R* is the anion-cation distance, and $S_{ns} = \langle 2p_{\sigma} | \varphi_{ns} \rangle$. The overlap integrals with the occupied 1s, 2s, and 3s shells are accounted for by orthogonalizing the distorted orbitals ψ_{u}' and ψ_{v}' by the Schmidt process. The 4s function becomes admixed as a result of $2p_{\sigma} \rightarrow 4s$ transfer. Thus, the orthogonal antibonding orbitals perturbed by phonon excitation have the form

$$\psi_{u}'' = N(\Psi_{u}' - \sum_{n=1}^{3} \langle \psi_{u}' | \varphi_{ns} \rangle \varphi_{ns} + a_{u} \varphi_{4s});$$

$$\psi_{v}'' = N(\Psi_{v}' - \sum_{n=1}^{3} \langle \psi_{v}' | \varphi_{ns} \rangle \varphi_{ns} + a_{v} \varphi_{4s}), \quad (3)$$

where a_u , a_v are the 4s admixture coefficients, which are assumed to be the sum of two terms: the point-charge crystal-field admixture,¹ and a covalent transfer contribution. They have the same strain dependence as (2), so that we can define

$$a_{u} = -a_{4s} (2\epsilon_{zz} - \epsilon_{xx} - \epsilon_{yy});$$

$$a_{v} = -a_{4s} \sqrt{3} (\epsilon_{xx} - \epsilon_{yy}). \qquad (4)$$

Then the resulting phonon-induced hyperfine

field is

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$$hf^{=} (8\pi/3)\mu_{B} \left[\langle \psi_{u}^{"} | \delta(r) | \psi_{u}^{"} \rangle + \langle \psi_{v}^{"} | \delta(r) | \psi_{v}^{"} \rangle \right]$$

$$\approx \frac{8\pi}{3}\mu_{B} \sum_{n=1}^{2} \sum_{n'=1}^{4} a_{n's} \cdot a_{n's} \langle \varphi_{ns} | \delta(r) | \varphi_{n's} \rangle 4 \langle \epsilon_{xx}^{2} + \epsilon_{yy}^{2} + \epsilon_{zz}^{2} - \epsilon_{xx} \epsilon_{yy} - \epsilon_{xx} \epsilon_{zz} - \epsilon_{yy} \epsilon_{zz} \rangle.$$
(5)

This field is parallel to the spin of the $3d^5$ ground state, and we obtain a net decrease of the total Mn^{2^+} hyperfine field. By introducing the mean-square strains, $\langle \epsilon_{ij}^2 \rangle$, we can express the temperature-dependent part of the hyperfine constant A as follows¹:

$$A(T) = A(T=0)(1-CT^{4}) \int_{0}^{\theta/T} x^{3} dx/(e^{x}-1), \quad (6)$$

where $\boldsymbol{\theta}$ is the averaged Debye temperature and

$$C = \frac{\frac{(8\pi/3)\mu_{\rm B}\sum_{n}\sum_{n'}a_{ns}a_{n's}\langle\varphi_{ns}|\delta(r)|\varphi_{n's}\rangle}{H_c} \times \frac{6\hbar}{5\rho\pi^2} \left(\frac{1}{v_t^{5}} + \frac{2}{3vl^5}\right) \left(\frac{k_{\rm B}}{\hbar}\right)^4, \tag{7}$$

where ρ is the mass density of the crystal, v_t and v_l are the transverse and longitudinal sound velocities, respectively, and H_c is the static hyperfine field coming from exchange polarization of the cores. We estimate the coupling parameters a_{ns} for Mn²⁺ in MgO as follows. Using Watson's Mn²⁺ free-ion wave functions for φ_{ns} and φ_u and O^{2-} wave functions⁷ for $2p_{\sigma}$, we calculate the overlap integrals S_{ns} and $\langle 2p_{\sigma}|\varphi_{u}\rangle$ for various bond lengths. From these we deduce the values for $\partial S_{nS}/\partial R$ and $\partial \langle 2p_{\alpha} | \varphi_{\mu} \rangle / \partial R$ evaluated at the equilibrium anion-cation distance R = 4.08 a.u. In order to estimate $\partial \lambda_{\sigma} / \partial R$ we assume that λ_{σ} is proportional to $\langle 2p_{\sigma} | \varphi_{\mu} \rangle$. The proportionality constant can be deduced from the equilibrium value of λ_{α} . A reliable experimental value of λ_{α} is not available since the σ -ligand spin density contains an undetermined contribution from $2p_{\sigma}$ \rightarrow 4s transfer. The a priori calculations of Hubbard, Rimmer, and Hopgood⁸ give $\lambda_{\alpha} = 0.28$ for KMnF₃. The degree of covalency in oxides is expected to be stronger than in fluorides so that we estimate $\lambda_{\sigma} = 0.3$ for Mn²⁺:MgO. Since the $2p_{\alpha} - 4s$ transfer matrix element cannot be reliably estimated we limit ourselves to calculating a_{4s} from the near-neighbor pointcharge model. Using Rimmer's⁹ wave function for $Mn^+(4s)$ to evaluate $\varphi_{4s}(0)$ and inserting the values of ρ , v_t , and v_l appropriate to MgO crystal,¹ we find from Eq. (7) a strength constant $C = 1.24 \times 10^{-13} (^{\circ}K)^{-4}$ to be compared with $C_{exp} = 2.58 \times 10^{-13} (^{\circ}\text{K})^{-4}$ obtained by comparing Eq. (6) with the experimental values of Walsh, Jeener, and Bloembergen. If we neglect the covalency contributions of the 1s, 2s, and 3s shells, we obtain, for the near-neighbor point-charge 4s contribution, $C_{4s} = 0.38$ $\times 10^{-13}$ (°K)⁻⁴ which is more than three times smaller than C estimated from Eq. (7). Therefore, a significant enhancement for the strength of the temperature dependence of A is obtained by taking into account the covalent admixture of inner shells. We can expect a further enhancement of a_{4s} caused by $2p_{\sigma} - 4s$ transfer. Hence, a large part of the discrepancy between point-charge calculation and the experimental strength can be removed by the covalency effects.

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⁴J. Owen and D. R. Taylor, Phys. Rev. Letters <u>16</u>, 1164 (1966).

⁵Nai Li Huang, R. Orbach, and E. Šimánek, Phys. Rev. Letters <u>17</u>, 134, 343(E) (1966).

⁶R. E. Watson, Massachusetts Institute of Technology Solid State Molecular Theory Group Technical Report No. 12, 1959 (unpublished).

⁷R. E. Watson, Phys. Rev. <u>111</u>, 1108 (1958).

⁸J. Hubbard, D. E. Rimmer, and F. R. A. Hopgood, Proc. Phys. Soc. (London) <u>88</u>, 13 (1966).

⁹D. E. Rimmer, private communication.

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¹E. Šimánek and R. Orbach, Phys. Rev. <u>145</u>, 191 (1966).

²W. M. Walsh, Jr., J. Jeener, and N. Bloembergen, Phys. Rev. 139, A1338 (1965).

³R. G. Shulman and S. Sugano, Phys. Rev. <u>130</u>, 506 (1963).