Coalescence in Two-Dimensional Foams: A Purely Statistical Process Dependent on Film Area

Emilie Forel,[1] Benjamin Dollet,[2] Dominique Langevin,[1] and Emmanuelle Rio[1,*]

[1]*Laboratoire de Physique des Solides, CNRS, Univ. Paris-Sud, Université Paris-Saclay, Orsay 91405, France*
[2]*Université Grenoble Alpes, CNRS, LIPhy, 38000 Grenoble, France*

While coalescence is ultimately the most drastic destabilization process in foams, its underlying processes are still unclear. To better understand them, we track individual coalescence events in two-dimensional foams at controlled capillary pressure. We obtain statistical information revealing the influence of the different parameters which have been previously proposed to explain coalescence. Our main conclusion is that coalescence probability is simply proportional to the area of the thin film separating two bubbles, suggesting that coalescence is mostly stochastic.

Liquid foams are essential to provide a pleasant texture to food and cosmetic products, to ensure thermal insulation, or to serve as precursors for structural materials such as concrete foams [1]. They are usually ephemeral, which can be very useful in many situations such as metallurgy, the glass industry, or water treatment, where the destruction of large volumes of foam after use is needed to extract the liquid for further processing. Among the different mechanisms leading to foam disappearance, coalescence, which is due to the rupture of the thin liquid film separating two neighboring bubbles, is the most efficient one.

However, coalescence remains uncontrolled, impeding a full exploitation of the potential of foams [2–4]. Several studies highlight the importance of the liquid content (see references in Ref. [2]), which decreases over time because of gravity, capillary drainage, and evaporation. Foam then eventually reaches a dry enough state and collapses. Nevertheless, the exact control parameters of coalescence, together with the appropriate mechanisms, remain elusive and are still debated.

A first scenario states that it is induced by the capillary suction from the channels, called Plateau borders, formed at the intersection of three foam films. Their radius of curvature R_{PB} in a foam stabilized by a solution of surface tension γ sets the capillary pressure: $P_c = \gamma/R_{PB}$. At equilibrium the film thickness is fixed by a balance between the disjoining pressure [5] and the capillary pressure. When the latter exceeds the equilibrium disjoining pressure, the thin films are prone to burst [6,7]. Hence, this scenario predicts a drastic upsurge of coalescence events below a critical R_{PB}.

In a second family of studies, the important parameter is the liquid fraction Φ, which is the ratio between the volume of liquid within the foam and the volume of foam. As discussed later, below a critical liquid fraction Φ_c, avalanches of coalescence occur [8] because the shortage of liquid prevents any topological rearrangements necessary along foam aging [9].

A third scenario stems from the observation that foam film bursting seems stochastic [10]. In such a scenario, the rupture probability of a film separating two bubbles increases with the film area, which scales as the square of the bubble radius R_B^2 [11].

To discriminate between these three mechanisms, it would be necessary to vary independently R_B, R_{PB}, and Φ. This is intrinsically difficult, since the radius of the Plateau borders is set by the bubble size and the liquid fraction as

$$R_{PB}^2 \propto \Phi R_B^2, \tag{1}$$

where Φ is a combination of R_B and P_c [12]. Therefore, a foam can be considered as wet either because the capillary pressure is small or because it contains small bubbles. Moreover, these parameters are not uniform in a given foam, since it can be polydisperse and exhibit gradients of liquid fraction leading to inhomogeneous values of Φ, R_B, and P_c. A last difficulty is that coalescence events are actually both rare and unpredictable [2,10], so that tackling the question requires an excellent statistics on such events. No former study varying one of these parameters independently has yet been performed. Moreover, our study permits an unprecedented reliable statistical analysis and proposes an original focus on the difference between average and local parameters.

In this Letter, we present experiments on 2D foams to record each coalescence event independently. A careful control of the capillary pressure together with the natural bubble growth due to coarsening allows us to vary the bubble size and the capillary pressure independently, leading to variable liquid fractions. This allowed us to

identify a regime in which the best candidate among the three aforementioned scenarios is the stochastic one.

The experimental setup, set on a heavy table to avoid vibrations, is inspired by the one proposed by Roth et al. [13], as shown in Fig. 1. A 3D foam is created by bubbling air through a needle at the bottom of a reservoir filled with a solution of tetradecyltrimethylammonium bromide (TTAB) at a concentration of 5 g/L. TTAB is a very common surfactant allowing us to obtain long-lived foams. In addition, it presents the advantage of being chemically stable along time. The top of the reservoir is in contact with a Hele-Shaw cell of horizontal dimensions 10×14 cm^2, whose thickness is set by a spacer. A seal of width 1 cm prevents any leak or evaporation of the liquid during the experiment. As the foam rises, a 2D foam invades
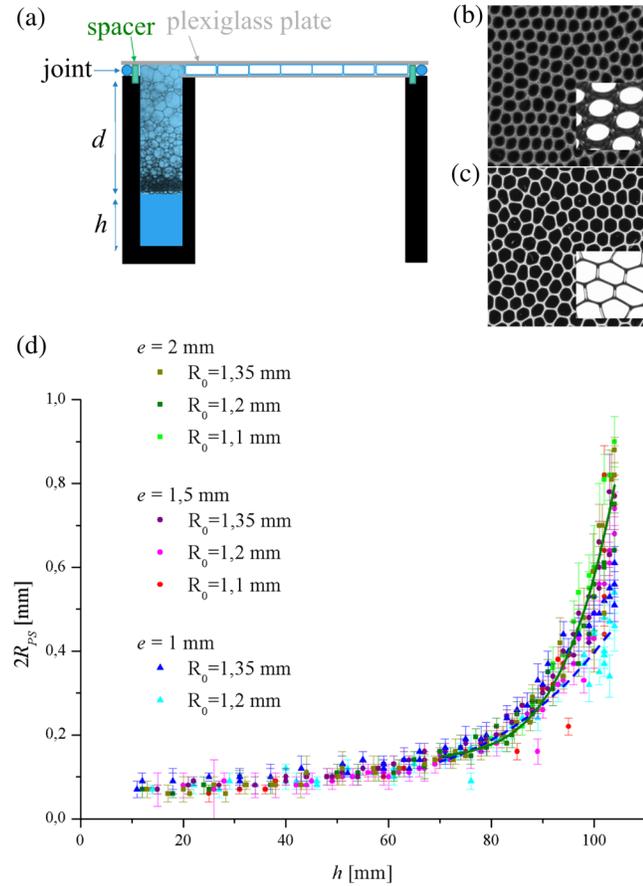
FIG. 1. (a) Experimental setup. The liquid height h allows us to fix the capillary pressure in the 2D foam confined by the Plexiglas plates in a Hele-Shaw cell. (b),(c) Top view of the bubbles for two different capillary pressures with an enlargement at the scale of a few bubbles in the insets. (d) Evolution of the radius R_{PS} of the pseudo-Plateau borders with the liquid height. The lines are the fits used to extract R_{PS} for each liquid height. For a spacer of 1 mm (dashed blue line), the equation is $3.3335 \times 10^{-3} h - 8.0883 \times 10^{-5} h^2 + 6.739 \times 10^{-7} h^3$, whereas for a spacer of 2 mm (solid green line), we used $7.09 \times 10^{-3} h - 4.7568 \times 10^{-4} h^2 + 1.3484 \times 10^{-5} h^3 - 1.6367 \times 10^{-7} h^4 + 7.2203 \times 10^{-10} h^5$

the Hele-Shaw cell. At equilibrium, a fixed liquid fraction profile is established in the 3D foam [14]. Thus, a fixed height h of liquid in the reservoir imposes a given value of R_{PB} in the 2D foam [Fig. 1(d)], which in turns sets the capillary pressure in the 2D foam [13]. We characterize the foam liquid content by measuring the width R_{PS} of the pseudo-Plateau borders, which are the channels at the junction between the top plate and two bubbles (Fig. 2). An enlarged view of a few bubbles [15] allows us to measure accurately this parameter as a function of h [Figs. 1(b) and 1(c)]. Figure 1(d) shows that R_{PS} increases continuously with h and depends only a little on the bubble radius and on the thickness e separating the plates. For the 36 experiments presented in this study, we fixed $e = 1$ mm or $e = 2$ mm. The millimetric bubbles are close packed so that more than 500 bubbles are present at the beginning of the experiment. A high-resolution camera (u-eye 1490, resolution 3840×2748 pixel^2) records the time evolution of the foam at a frequency of 1 frame per second.

The foam evolution is automatically treated, which is necessary because of the rarity and unpredictability of the coalescence events. We adapted an existing homemade software [16], which tracks each bubble between two successive images to detect individual coalescence events (Supplemental Material, Sec. A [17]). For each of our 36 experiments, each individual event can be identified (Fig. 3) and our software also allows us to record the area A_B of each bubble before and after coalescence as well as the length ℓ of each bubble side. Thus, we recorded 2.9×10^7 bubbles (accounting for their multiple occurrence in different images of a given movie), among which 5668 did coalesce. This highlights the rarity of the events and
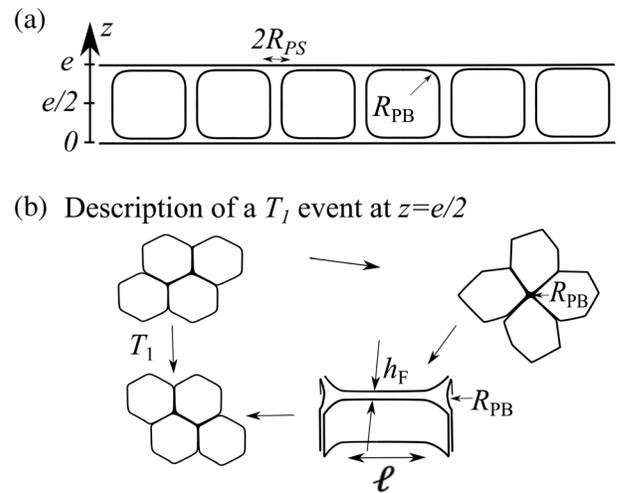
FIG. 2. (a) Section of a 2D foam confined in a Hele-Shaw cell of thickness e. (b) Sketch of the different steps along a T_1 event with the extraction of a film of length ℓ from a Plateau border of radius of curvature R_{PB}.
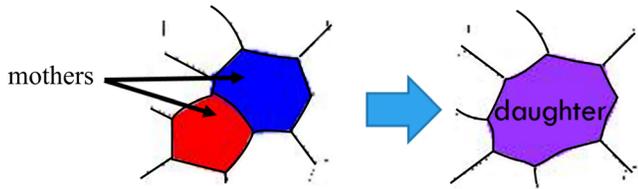
FIG. 3. Detection of two mother bubbles and of their common daughter.

the importance of automatic treatment to get sufficient statistics.

To discriminate between the possible coalescence mechanisms, we measured the parameters relevant to the different scenarios. In Figs. 4(a), 5(a), and 6(a), we plot, for each value of the tested parameter, the total number of bubbles that we observed N_o (in black) and the number of these bubbles which coalesce N_c (in gray). The ratio N_c/N_o is the coalescence probability between two images. It is plotted in Figs. 4(b), 5(b), and 6(b).

To test the scenario of the critical capillary pressure, we studied the influence of the pseudo-Plateau border radius R_{PS}. For each experiment, we measure the value of h by direct visualization through a lateral window. We fitted the curve observed in Fig. 1(d) for each value of the thickness to obtain a direct relationship between h and R_{PS} (see

caption of Fig. 1). In Fig. 4, each value of R_{PS} corresponds to a given experiment. We observe that, whatever the size of the Plateau borders, coalescence events occur. Moreover, the probability to coalesce exhibits no clear tendency. This shows that, in our configuration, the capillary pressure is not the right parameter to describe the coalescence.

We now study the scenario of a critical liquid fraction proposed by Biance *et al.* [9]. According to these authors, below a critical liquid fraction, the films break during topological rearrangements called T_1 events. Indeed, for bubbles to swap neighbors, a new soap film is created. Owing to dynamic effects, this film is thicker during its extension following its creation than at equilibrium. In a too dry foam, the quantity of liquid available in the Plateau borders is too small to feed this extending film, and it pops. In 2D foams, the same mechanism can be invoked. During a T_1 event, the film is generated from a channel formed by the transient intersection of four bubbles [Fig. 2(c)]. This transient Plateau border has a radius $R_{PB} = R_{PS}/\sqrt{3}$ [18], so that its section scales as R_{PB}^2. It thus contains a volume of liquid scaling as $R_{PS}^2 e/3$, where e is the vertical height of the film. Moreover, a new film contains a volume $eh_F\ell$ of liquid, where h_F is the transient thickness of the extending film. As suggested by Biance *et al.* [9], we propose that the film thickness is fixed by the velocity of the rearrangements
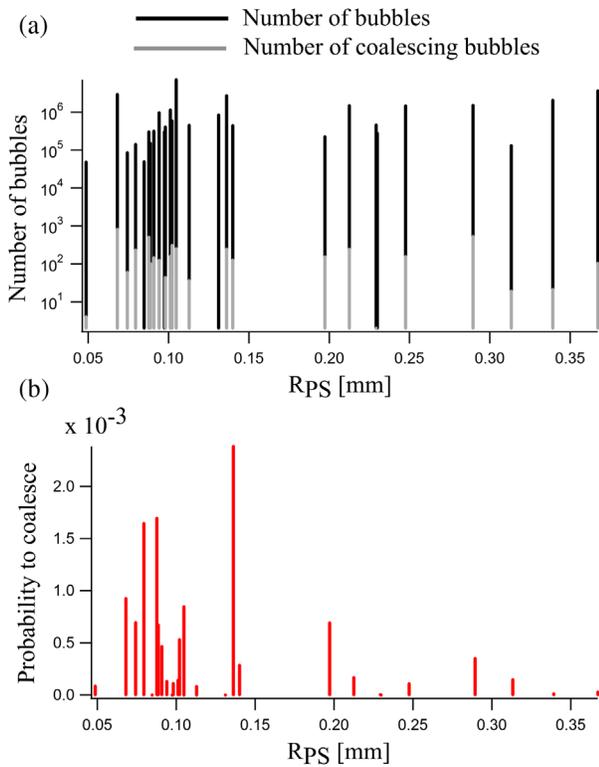
FIG. 4. For each experiment, at fixed R_{PS}, plot of (a) the number of observed bubbles and the number of coalescence events, and (b) the probability of a bubble to coalesce.
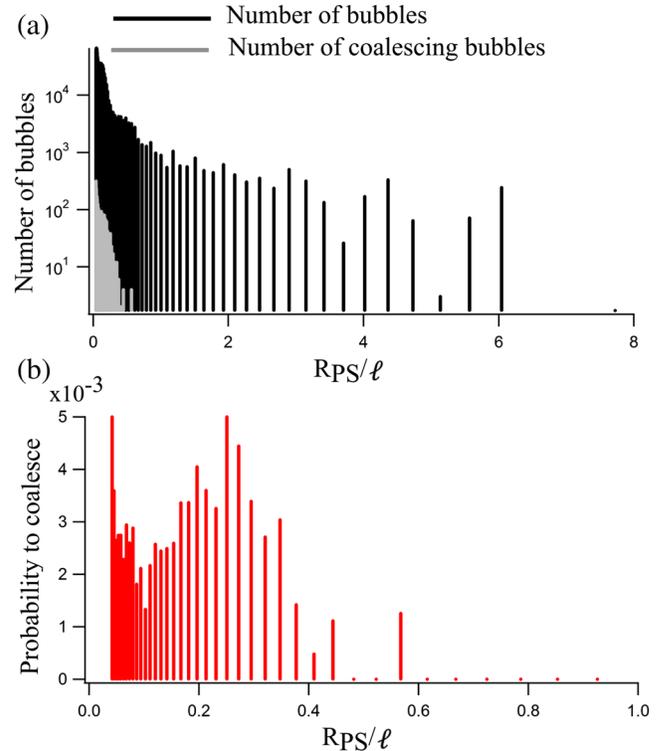
FIG. 5. Plot of (a) the number of observed bubbles and of coalescence events and (b) the probability of a bubble to coalesce, as functions of R_{PS}/ℓ. Note that the x axis is different in both plots.
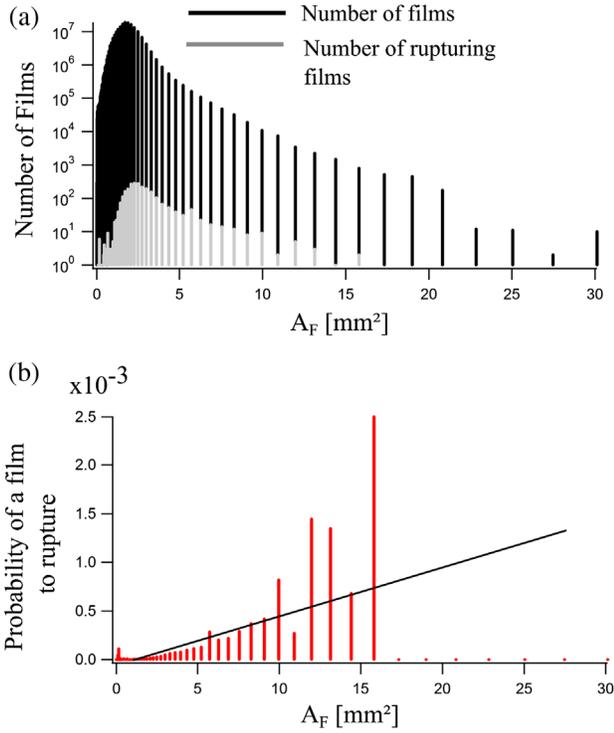
FIG. 6. Plot of (a) the number of observed films and of coalescence events and (b) the probability of a film to break, as functions of the film area. The black solid line is a guide for the eye.

V, i.e., the typical extension velocity of a film after its creation in a T_1 event, through Frankel's law,

$$h_F = 1.84 R_{\mathrm{PB}} \mathrm{Ca}^{2/3}, \qquad (2)$$

with the capillary number $\mathrm{Ca} = \eta V / \gamma$, with η the liquid viscosity. The film is expected to burst during the rearrangements as soon as the Plateau border volume is comparable to the film volume, i.e., $e h_F \ell \approx e R_{\mathrm{PS}}^2$, which, combined with Eq. (2), yields to a critical value of R_{PS}/ℓ scaling as $R_{\mathrm{PS}}/\ell \approx \mathrm{Ca}^{2/3}$. The right parameter to account for this mechanism is finally R_{PS}/ℓ.

We thus plot the probability of a bubble side to break versus the value of R_{PS}/ℓ. Two families of bubbles can exist. For very small bubbles in a rather wet foam, the length ℓ of their frontier with a neighboring bubble is smaller than $2 R_{\mathrm{PS}}$ and the bubbles are spherical. For larger bubbles and/or drier foams, a flat film appear because $R_{\mathrm{PS}}/\ell < 1/2$. Given the excellent sampling of bubble size, the second family of bubbles is well described with a continuous variation of R_{PS}/ℓ between 0 and 0.5. The sampling is not as good for the first family, but this is unimportant since these wet bubbles were never observed to coalesce. Finally, the number of observed coalescence events observed versus R_{PS}/ℓ gives the very smooth curve in Fig. 5(a), which lends confidence to the

probability shown in Fig. 5(b). In particular, we mention that, for each value of this parameter, more than 10^3 bubbles have been observed. The first observation is that no coalescence events are observed for R_{PS}/ℓ larger than 0.5 whereas we observe a nonzero probability for the bubble to coalesce for all values of R_{PS}/ℓ smaller than 0.5 [Fig. 5(a)]. Nevertheless, we observe no systematic variation of the probability for a bubble side to rupture in this parameter range [Fig. 5(b)]. This naturally excludes the T_1 scenario from the mechanisms triggering to coalescence.

The last scenario is the stochastic one, in which the relevant parameter is the area A_F of the film separating the two mother bubbles. This area is given by $A_F = e\ell$, where e is the thickness separating the glass plates and ℓ the length of the film. Figure 6 shows the influence of A_F on the coalescence events. The film area grows continuously over time, because the mean bubble area increases due to both coarsening and coalescence. This explains the excellent sampling observed in Fig. 6, at least for the bubbles of area under 20 mm^2. If coalescence were purely stochastic along the film area, we would observe a linear increase of the rupture probability with the film area. As shown in Fig. 6(b), such a linear increase is actually observed, except for very small and very big films.

The scatter may be due to lack of statistics: there was less than ten coalescence events (for small films) or bubbles (for large films) per bin. No coalescence was detected for the largest films ($A_F > 16$ mm^2), but only a few of them could be detected reliably. With this word of caution, these results are then compatible with a stochastic rupture of the foam films, except for very small bubbles.

In conclusion, we developed an experiment allowing us to detect single coalescence events in a 2D foam by image analysis. We were able to control the capillary pressure and to measure the foam film size, as well as the ratio R_{PS}/ℓ between the size of the pseudo-Plateau borders and the length of the ruptured film. This allowed us to fully characterize almost 6000 coalescence events and to obtain unprecedented statistics concerning the impact of the capillary pressure, the foam film size, and the ratio R_{PS}/ℓ on the probability of a bubble to coalesce. Our conclusions are that neither the capillary pressure nor the liquid fraction trigger coalescence in our explored range of parameters. The probability for a film to rupture is rather linear with the film area, in accordance with a purely statistical scenario. The reason why a critical capillary pressure is sometimes observed in the literature could be the explored range of R_{PS}. The capillary pressure indeed varies between 100 and 700 Pa in our experiment, which is of the order of, but smaller than, the critical capillary pressure observed by Khristov et al. [6]. It is likely that above a critical capillary pressure, the films may rapidly rupture, while below this pressure, film rupture is stochastic and less frequent.

To generalize the impact of the foam film area in 3D foams, a study similar to the one described here, i.e., at the scale of each bubble, would be necessary to overcome the inhomogeneities in size and liquid fraction inherent to aging bubbles. Such experiments are of course very challenging; nevertheless, our results suggest that the statistical rupture of a film should remain important for coalescence in 3D foams as well.

---

[*]emmanuelle.rio@u-psud.fr

[1] P. Stevenson, *Foam Engineering: Fundamentals and Applications* (Wiley, New York, 2012).

[2] E. Rio and A. L. Biance, Chem. Phys. Chem. **15**, 3692 (2014).

[3] A. Colin, in *Foam Engineering: Fundamentals and Applications*, edited by P. Stevenson (Wiley, New York, 2012), Chap. 4, pp. 75–90.

[4] D. Langevin and E. Rio, Coalescence in foams and emulsions, in *Encyclopedia of Surface and Colloid Sciences* (Taylor & Francis, London, 2012).

[5] J. N. Israelachvili, *Intermolecular and Surface Forces* (Academic Press, London, 2011).

[6] K. Khristov, D. Exerowa, and G. Minkov, Colloids Surf. A **210**, 159 (2002).

[7] S. Tcholakova, N. D. Denkov, and A. Lips, Phys. Chem. Chem. Phys. **10**, 1608 (2007).

[8] V. Carrier and A. Colin, Langmuir **19**, 4535 (2003).

[9] A. L. Biance, A. Delbos, and O. Pitois, Phys. Rev. Lett. **106**, 068301 (2011).

[10] S. T. Tobin, A. J. Meagher, B. Bolfin, M. Möbius, and S. Hutzler, Am. J. Phys. **79**, 819 (2011).

[11] D. Georgieva, A. Cagna, and D. Langevin, Soft Matter **5**, 2063 (2009).

[12] I. Cantat, S. Cohen-Addad, F. Elias, F. Graner, R. Höhler, O. Pitois, F. Rouyer, and A. Saint-Jalmes, *Foams: Structure and Dynamics* (Oxford University Press, Oxford, 2013).

[13] A. E. Roth, C. D. Jones, and D. J. Durian, Phys. Rev. E **87**, 042304 (2013).

[14] A. Maestro, W. Drenckhan, E. Rio, and R. Höhler, Soft Matter **9**, 2531 (2013).

[15] E. Forel, E. Rio, M. Schneider, S. Beguin, D. Weaire, S. Hutzler, and W. Drenckhan, Soft Matter **12**, 8025 (2016).

[16] B. Géraud, Y. Méheust, I. Cantat, and B. Dollet, Phys. Rev. Lett. **118**, 098003 (2017).

[17] See Supplemental Material at http://link.aps.org/supplemental/10.1103/PhysRevLett.122.088002 for detailed description of the procedure used to track the bubbles along time by image analysis.

[18] C. D. Schimming and D. J. Durian, Phys. Rev. E **96**, 032805 (2017).