Trapping of Molecular Oxygen together with Lithium Atoms

Nitzan Akerman, Michael Karpov, Yair Segev, Natan Bibelnik, Julia Narevicius, and Edvardas Narevicius

Department of Chemical Physics, Weizmann Institute of Science, Rehovot 7610001, Israel

(Received 30 October 2016; published 16 August 2017)

We demonstrate simultaneous deceleration and trapping of a cold atomic and molecular mixture. This is the first step towards studies of cold atom-molecule collisions at low temperatures as well as application of sympathetic cooling. Both atoms and molecules are cooled in a supersonic expansion and are loaded into a moving magnetic trap that brings them to rest via the Zeeman interaction from an initial velocity of 375 m/s. We use a beam seeded with molecular oxygen, and entrain it with lithium atoms by laser ablation prior to deceleration. The deceleration ends with loading of the mixture into a static quadrupole trap, which is generated by two permanent magnets. We estimate 10^9 trapped O₂ molecules and 10^5 Li atoms with background pressure limited lifetime on the order of 1 sec. With further improvements to lithium entrainment we expect that sympathetic cooling of molecules is within reach.

DOI: 10.1103/PhysRevLett.119.073204

During the last decades trapping of atoms has become a workhorse of atomic physics with countless experiments where atoms have been routinely cooled down to nano-Kelvin temperatures. On the other hand, cooling and trapping of molecules remains challenging. Several methods demonstrated molecular trapping and cooling including assembly of ultracold diatomic molecules from lasercooled alkali-metal atoms [1-4], direct laser cooling, and magneto-optical trapping of molecules with favorable vibrational transitions that allow scattering of a large number of photons [5,6]. Sisyphus molecular cooling has been demonstrated in the case of electrostatically trapped molecules [7]. It is also possible to cool molecules without relying on laser transitions via collisions with cold buffer gas [8]. Cold molecules can be extracted from a cryogenic buffer gas cell in hydrodynamically enhanced flow, producing an intense and versatile source that can be used as a starting point in other experiments [9]. A similar direct and general molecular cooling method is based on collisions that occur during adiabatic expansion of high pressure gas into vacuum. Atomic and molecular beams that are formed by such expansion have been successfully decelerated using inhomogeneous magnetic and electric fields [10]. Subsequent molecular trapping of Stark decelerated beams has been demonstrated using electric and magnetic traps [11–13]. In a similar fashion paramagnetic atomic or molecular beams have been trapped following Zeeman deceleration [14–16].

Far less progress has been made in the generation of cold mixtures of atoms and molecules, even though it opens many possibilities in both physics and chemistry. An immediate advantage that atom-molecule cotrapping offers is the orders of magnitude longer interrogation times compared to molecular crossed beam methods. This enables the study of cold chemistry for especially slow processes. Particularly, inelastic collisions in such a setup have already been successfully studied by Parazzoli et al. [17] and the upper limit on reactive collisions between N and NH has been placed by Campbell et al. [18]. In addition, as in the production of cold alkali molecules, photoassociation or Feschbach resonances can be used to construct polyatomic cold molecules. One of the most exciting opportunities that cold atom-molecule mixtures enables is the application of sympathetic cooling, where cold atoms that are amenable to laser cooling can be used to collisionally cool molecules. Prospects of such a cooling scheme strongly depend on the collisional properties of atoms and molecules with several candidates identified [19] and the cooling mechanism studied in detail [20]. A general guide for a successful application of sympathetic cooling is weak anisotropy in the interaction potential and low reduced mass of interacting particles that helps to suppress inelastic scattering channels by increasing the centrifugal barrier height in the exit channel.

In order to achieve the highest possible trapped molecular density, we choose molecular oxygen that can be easily seeded in a supersonic expansion and decelerated via Zeeman interaction, with mass to magnetic moment ratio of ~16 amu/Bohr magneton. Deceleration of O_2 molecules by pulsed magnetic fields has been demonstrated [21,22]. Recently, Liu *et al.* have also reported short confinement, on the order of 600 μ s, of the molecular packet in an electromagnetic trap [23].

Since most of the atoms are paramagnetic in the ground or long lived metastable state many are well suited for Zeeman deceleration. In a previous work, we have demonstrated codeceleration of metastable argon atoms together with molecular oxygen [24]. In that experiment oxygen was cooled in a molecular expansion and metastable argon was generated by electric discharge from the carrier argon gas. Here, we go beyond deceleration and demonstrate how the decelerated molecular ensemble can be transferred into a permanent trap in order to open the possibility of sympathetic cooling. Furthermore, we have cotrapped lithium atoms, by entraining them into the beam prior to deceleration. Importantly, metastable argon is unstable with the Penning ionization process taking place with both molecular oxygen and Ar*. In contrast, the ground state Li and O₂ reaction is endothermic and does not occur at our trapping temperatures. Moreover, Li has the advantages of lower reduced mass and is highly suitable for laser cooling. Throughout the deceleration and subsequent trapping the atoms and molecules are confined in a three-dimensional trap, leading to small losses during both the deceleration and transfer into the permanent trap. We estimate around 10^9 trapped O₂ molecules at a temperature of 300 mK together with 10⁵ lithium atoms in a permanent magnetic quadrupole trap. Our results here provide a pathway to further implementation of molecular cooling by forced evaporation and sympathetic cooling.

Our experimental apparatus is presented in Fig. 1. A pulsed beam of O_2 is produced by expanding a mixture of O_2 and Kr with stagnation pressure of about 10 bar into vacuum using an Even-Lavie valve [25]. The valve is cooled to a temperature of 165 K in order to reduce the mean initial velocity of the beam to below 400 m/s, from which it can be decelerated to a stop. The velocity spread of the beam is ± 25 m/s, which corresponds to translational temperature of ≈ 3 K. In other works [22] it was found that the rotational temperature of O_2 seeded in Kr was around 5 K, which means that only the N = 1 manifold (the lowest state for O_2) is occupied.

At a distance of about 15 cm the cold supersonic beam enters a moving trap Zeeman decelerator. The working principles of our decelerator are given elsewhere [24,26].

Briefly, it consists of 480 spatially overlapping quadrupole traps and spans over 2.4 meters long. The traps are activated sequentially in a temporally overlapping manner, where each pulse follows a half sine shape and each trap is activated at the peak current of the preceding trap. The instantaneous velocity of the decelerated beam is controlled by the width of each current pulse. All the pulses are generated by ten driver modules, which are real-time configurable LC circuits capable of delivering up to 600 A with variable pulse duration ranging form 20 to 500 μ s. These currents can generate magnetic fields as high as 0.8 Tesla along the trap symmetry axis, which corresponds to a trap depth of about 300 mK in the transverse direction and 400 mK in the longitudinal for O2 at a deceleration value of 35 000 m/s². Using a computercontrolled pulsed sequence we can set the final beam velocity to anywhere between 450 and 20 m/s, which is slow enough for loading into a static magnetic trap.

Once the molecules and the atoms are slowed to low enough velocities, they can be trapped in a static trap. Generating a magnetic field on the order of 1 Tesla by running a current in a coil for millisecond time scales is fairly straightforward. Extending it to seconds becomes a much more difficult problem, as one needs to deal with a significant amount of heat dissipation. To circumvent this problem we use a static magnetic trap, which is formed by two neodymium iron boron permanent magnets. Each magnet has an outer diameter of 69 mm, a central bore with a diameter of 12 mm, and a width of 6 mm. The onaxis magnetic field peak of a single magnet has a magnitude of about 0.5 Tesla. The two magnets are aligned in opposite directions and separated by 1 cm (center to center) to form a quadrupole trap with a longitudinal depth of 0.5 K and a



FIG. 1. Schematic diagram of the experiment. A pulsed supersonic beam of O_2 seeded in Kr is produced by an Even-Lavie valve. Li atoms can be entrained into the beam by laser ablation of a solid target placed near the valve. Low field seeking states are decelerated by a 2.4 meter long moving trap decelerator and then trapped in a permanent magnet quadrupole trap. Both O_2 molecules (REMPI) and Li atoms (single photon) are ionized by a 225 nm laser pulse. The ions are then extracted by an electrostatic lens towards an MCP detector. The right inset shows a TOF trace with two prominent peaks that correspond to the two products. The left inset shows the calculated magnetic field magnitude of the permanent trap.





FIG. 2. Loading sequence of the static magnetic trap, showing the magnetic field magnitude along the Z axis as a function of time. Additional coils are used to cancel the front permanent magnet's field at the right time to allow the molecules to enter the static trap. The minimum of the magnetic field closely follows a trajectory of constant deceleration (dashed white line) that reaches zero velocity at the static trap's center.

radial depth of 0.3 K for O_2 molecules. Although particles in such a trap are subject to nonadiabatic spin flip (Majorana) losses near the trap center, this part of the phase space is negligible for our experimental parameters and therefore does not result in significant loss.

During the loading process into the permanent magnets trap, the front barrier needs to be momentarily eliminated, in order to let the slow molecular beam enter into the center of the trapping region. This is achieved by using three extra coils, which adiabatically guide the molecules from the last decelerator trap into the static trap. The first coil generates a quadrupole trap together with the front magnet, bringing the molecules as close to it as possible. Then a second coil is used to cancel the magnetic field of the front magnet, letting the molecular beam enter the trapping region. A third coil is used to increase the back magnet's field in order to bring the molecules to a stop close to the trap center. Figure 2 shows the calculated magnetic field magnitude along the symmetry axis as a function of time during the loading process. The dashed white line is a parabolic curve, which illustrates a trajectory of constant deceleration to zero velocity at the static trap center, which coincides well with the generated magnetic field minimum.

Unfortunately O_2 does not possess a dipole transition from the ground state in the accessible optical spectrum that can be used for detection and manipulation. Moreover, those excited states that do exist are in the extreme UV and undergo fast predissociation. Therefore, for detection of the O_2 molecules we use a resonance-enhanced multiphoton ionization (REMPI) process, where the produced ions are then measured with a microchannel plate (MCP) detector. Here we use a 2 + 1 REMPI process at 225 nm [27]. The two-photon transition excites the $3d\pi^3\Sigma_1^-(\nu'=2) \leftarrow$ $X^3\Sigma_q^-(\nu''=0)$ intermediate level and the third photon

FIG. 3. Measured and simulated loading dynamics of the static magnetic trap. The solid blue is the measured REMPI signal from the static trap center as a function of time. The dashed black line is the result of a Monte Carlo simulation of 4000 molecules showing the density at the trap center (average over a volume of 10 mm³).

ionizes it into the continuum. In order to minimize any free flight of the molecules, the static trap magnets are mounted on the same 1 cm diameter vacuum tube of the decelerator. As a result, the detection region has a single optical axis in the counterdirection of the beam. We use an electrostatic lens to extract the ions from the grounded vacuum tube and direct them towards the MCP. The lens consists of three metal tubes that are inserted into the vacuum tube and their voltages can be controlled externally. The right inset of Fig. 1 shows a typical time of flight (TOF) trace measured by the MCP, where the two dominant peaks correspond to the different ionic outcomes of the REMPI pulse O_2^+ and Li⁺ (the small peak in the middle is an O^+).

Figure 3 presents the measured O₂ REMPI signal during the first few milliseconds of trapping, along with results from a simulation of the trap dynamics. The highest initial peak comes from the decelerated O_2 packet crossing the center of the trap, followed by a few additional density oscillations with a period of around 1 ms, which settle to about 50% of the peak density after a few oscillations. These oscillations indicate a nonperfect adiabatic loading with a small residual mean velocity. This is expected, because reaching zero velocity in a perfect adiabatic sequence would take infinite time. In our case, a nonperfect adiabatic loading compromises the resulting phase space through the time it takes to enter the trap (canceling the field of the front magnet). This is further evident in the 50%loading efficiency. We simulate the loading process with 4000 molecules and get a good quantitative agreement with the experiment.

The Li atoms are entrained into the beam by ablation of a solid lithium target, placed near the nozzle on a wobble stick. This is achieved by using a pulsed laser (Quanta Ray Indi) with pulses of a few tens of mJ at 355 nm, focused on the target. In this way we have entrained Li atoms into the



FIG. 4. Measured lifetime of the trapped O₂ molecules and Li atoms in the permanent trap. By fitting the data to a decaying exponent (solid line) we infer a 1/e lifetime of $\tau_{O_2} = 670 \pm 60$ ms for the oxygen and $\tau_{Li} = 380 \pm 35$ for the lithium.

supersonic beam and trapped them together with the O_2 molecules in the permanent magnetic trap. The 225 nm laser pulses have sufficient energy to ionize the Li atoms together with the O_2 molecules, such that the extraction and detection procedures for the two species are the same and are performed simultaneously.

Figure 4 shows the O_2 and Li REMPI signals as a function of time from the moment the trap was loaded. Each data point is an average of 40 repetitions. From fitting the measured results to an exponential decay we infer a trap lifetime of 670 ± 60 msec for the O₂ molecules and 380 ± 35 msec for the Li atoms. We assume our lifetime is limited by the background pressure, which in this case implies a few 10^{-8} Torr [28]. Although our vacuum gauge reading is 10 times smaller, such a pressure gradient between the gauge and the trap is reasonable due to geometry and vacuum conductance. The difference in lifetime is consistent with the ratio of 1.8 obtained by calculating the collision rates of Li-H₂ and O₂-H₂ systems based on van der Waals dispersion coefficients [29,30]. In order to verify that the lithium lifetime is limited by the background and not by inelastic collisions with the oxygen molecules, we have performed the same measurement using a beam of pure Kr, in which the Li atoms have been entrained. The lithium lifetime obtained from this measurement is almost the same, which supports our assumption.

Estimating the absolute number of trapped molecules from the REMPI signal is a difficult task, as the process efficiency is not well known and strongly depends on the laser beam intensity and shape. Therefore, our best estimation relies on a residual gas analyzer (SRS model RGA100) measurement. To calibrate the RGA sensitivity to O_2 we flooded the vacuum chamber with pure O_2 to significantly raise the pressure above the nominal background and used an independent vacuum gauge to measure the absolute pressure. Since the RGA is mounted about 7 cm downstream from the trap we could measure only molecular beams that were not decelerated to very low velocities. Using a beam at 350 m/s and taking into account the experimental uncertainties (ionization volume of the RGA, gauge calibration, and pressure gradient) we estimate a lower limit to the peak beam density of 10^{10} molecules/cm³. We were then able to infer the trapped molecules' density from the relative REMPI signal of the guided 350 m/s beam and trapped molecules. The number of Li atoms here was too small to be detected with the RGA; hence, we estimated it from the single-photon ionization cross section $\sigma_{\rm Li} = 1.6 \times 10^{-18} \text{ m}^2$ [31], which is much more reliable than REMPI. The beam shape here was also known to better accuracy, as in this case the atoms were not at the focus, but rather where the beam diverges. We estimate about 10^5 trapped Li atoms, which indicates an inefficient entrainment.

In conclusion, by combining the high intensity of a supersonic source with the efficiency of a moving trap decelerator we have demonstrated the largest ensemble of trapped cold molecules achieved to date. We believe that a significant improvement will be gained by performing the ablation during the expansion stage, which requires modification of the nozzle geometry. Our method is applicable to other species, for example NH₂ and NH, where the latter has favorable Frank-Condon factors and, in principle, can be directly laser cooled. In addition, magnetic deceleration is not limited to supersonic sources, and can be used together with other setups, such as buffer gas cooling systems. One of the exciting uses is in the direct cooling of molecules, such as SrF, which was recently trapped and cooled to temperatures as low as 2.5 mK, after being loaded to a magneto-optical trap from a cryogenic buffer gas source and slowed using radiation pressure [5]. Our current work paves the way for studying cold collisions of atoms and molecules in a magnetic trap, and for using sympathetic cooling in order to reach the ultracold regime for molecules at high densities.

This work was financially supported by European Research Council (Grant No. EU-FP7-ERC-CoG 1485).

- J. D. Miller, R. A. Cline, and D. J. Heinzen, Phys. Rev. Lett. 71, 2204 (1993).
- [2] K.-K. Ni, S. Ospelkaus, M. De Miranda, A. Pe'er, B. Neyenhuis, J. Zirbel, S. Kotochigova, P. Julienne, D. Jin, and J. Ye, Science **322**, 231 (2008).
- [3] J. W. Park, S. A. Will, and M. W. Zwierlein, Phys. Rev. Lett. 114, 205302 (2015).
- [4] T. Takekoshi, L. Reichsöllner, A. Schindewolf, J. M. Hutson, C. R. Le Sueur, O. Dulieu, F. Ferlaino, R. Grimm, and H.-C. Nägerl, Phys. Rev. Lett. 113, 205301 (2014).
- [5] E. B. Norrgard, D. J. McCarron, M. H. Steinecker, M. R. Tarbutt, and D. DeMille, Phys. Rev. Lett. 116, 063004 (2016).

- [6] V. Zhelyazkova, A. Cournol, T. E. Wall, A. Matsushima, J. J. Hudson, E. A. Hinds, M. R. Tarbutt, and B. E. Sauer, Phys. Rev. A 89, 053416 (2014).
- [7] A. Prehn, M. Ibrügger, R. Glöckner, G. Rempe, and M. Zeppenfeld, Phys. Rev. Lett. 116, 063005 (2016).
- [8] N. R. Hutzler, H.-I. Lu, and J. M. Doyle, Chem. Rev. 112, 4803 (2012).
- [9] D. Patterson and J. M. Doyle, J. Chem. Phys. 126, 154307 (2007).
- [10] M. Lemeshko, R. V. Krems, J. M. Doyle, and S. Kais, Mol. Phys. **111**, 1648 (2013).
- [11] H. L. Bethlem, G. Berden, F. M. Crompvoets, R. T. Jongma, A. J. Van Roij, and G. Meijer, Nature (London) 406, 491 (2000).
- [12] S. Y. T. van de Meerakker, P. H. M. Smeets, N. Vanhaecke, R. T. Jongma, and G. Meijer, Phys. Rev. Lett. 94, 023004 (2005).
- [13] B. K. Stuhl, M. T. Hummon, M. Yeo, G. Quéméner, J. L. Bohn, and J. Ye, Nature (London) 492, 396 (2012).
- [14] N. Vanhaecke, U. Meier, M. Andrist, B. H. Meier, and F. Merkt, Phys. Rev. A 75, 031402 (2007).
- [15] J. D. Weinstein, R. deCarvalho, T. Guillet, B. Friedrich, and J. M. Doyle, Nature (London) **395**, 148 (1998).
- [16] Y. Liu, M. Vashishta, P. Djuricanin, S. Zhou, W. Zhong, T. Mittertreiner, D. Carty, and T. Momose, Phys. Rev. Lett. 118, 093201 (2017).
- [17] L. P. Parazzoli, N. J. Fitch, P. S. Żuchowski, J. M. Hutson, and H. J. Lewandowski, Phys. Rev. Lett. **106**, 193201 (2011).

- [18] W. C. Campbell, E. Tsikata, Hsin -I. Lu, L. D. van Buuren, and J. M. Doyle, Phys. Rev. Lett. 98, 213001 (2007).
- [19] A. O. G. Wallis and J. M. Hutson, Phys. Rev. Lett. 103, 183201 (2009).
- [20] J. Lim, M. D. Frye, J. M. Hutson, and M. R. Tarbutt, Phys. Rev. A 92, 053419 (2015).
- [21] E. Narevicius, A. Libson, C. G. Parthey, I. Chavez, J. Narevicius, U. Even, and M. G. Raizen, Phys. Rev. A 77, 051401 (2008).
- [22] A. W. Wiederkehr, H. Schmutz, M. Motsch, and F. Merkt, Mol. Phys. **110**, 1807 (2012).
- [23] Y. Liu, S. Zhou, W. Zhong, P. Djuricanin, and T. Momose, Phys. Rev. A 91, 021403 (2015).
- [24] N. Akerman, M. Karpov, L. David, E. Lavert-Ofir, J. Narevicius, and E. Narevicius, New J. Phys. 17, 065015 (2015).
- [25] U. Even, Adv. Chem. 2014, 636042 (2014).
- [26] E. Lavert-Ofir, S. Gersten, A. B. Henson, I. Shani, L. David, J. Narevicius, and E. Narevicius, New J. Phys. 13, 103030 (2011).
- [27] R. J. Yokelson, R. Lipert, and W. Chupka, J. Chem. Phys. 97, 6153 (1992).
- [28] T. Arpornthip, C. A. Sackett, and K. J. Hughes, Phys. Rev. A 85, 033420 (2012).
- [29] S. Bali, K. M. O'Hara, M. E. Gehm, S. R. Granade, and J. E. Thomas, Phys. Rev. A 60, R29 (1999).
- [30] W. Rijks, M. Van Heeringen, and P. Wormer, J. Chem. Phys. 90, 6501 (1989).
- [31] R. D. Hudson and V. L. Carter, J. Opt. Soc. Am. 57, 651 (1967).