Electron Localization in Dissociating H₂⁺ by Retroaction of a Photoelectron onto Its Source

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We investigate the dissociation of H_2^+ into a proton and a H^0 after single ionization with photons of an energy close to the threshold. We find that the p^+ and the H^0 do not emerge symmetrically in the case of the H_2^+ dissociating along the $1s\sigma_g$ ground state. Instead, a preference for the ejection of the p^+ in the direction of the escaping photoelectron can be observed. This symmetry breaking is strongest for very small electron energies. Our experiment is consistent with a recent prediction by Serov and Kheifets [Phys. Rev. A 89, 031402 (2014)]. In their model, which treats the photoelectron classically, the symmetry breaking is induced by the retroaction of the long-range Coulomb potential onto the dissociating H_2^+ .

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Introduction.—Symmetry is one of the most fundamental concepts for the quantum mechanical description of molecules. Because of their symmetry, homonuclear molecules have electronic eigenfunctions of either gerade or ungerade parity. This is commonly assumed to hold true while a molecule is dissociating, even though after dissociation, any measurement will detect the hole at one of the fragments (i.e., a symmetry-broken system). The welldefined symmetry of the electron wave function will, however, create a hole with equal probability at each of the two fragments. External fields which are present during the dissociation can break this symmetry. A variety of scenarios have been reported in which strong laser fields have been utilized to induce such symmetry breaking. A pioneering experiment used a carrier envelope phase locked few cycle pulse [1]. Later experiments used two color pulses of neighboring harmonics [2,3], an attosecond pulse synchronized to the driving pulse [4–6], or broke the laser field symmetry by measuring the field direction at the instant of creating a H₂⁺ ion employing the attoclock technique [7]. In all of these scenarios, the laser field mixed gerade and ungerade states with a well-defined phase during the dissociation leading to a localization of the bound electron. Such a coherent mixture of states of two symmetries can also occur already in the ionization step if the ionization energy is in the range of doubly excited resonances [4,8,9].

Here, we show experimental evidence for a noninvasive and much more fundamental way to break the symmetry of ${\rm H_2}^+$, avoiding any external fields and occurring in the absence of doubly excited states. We demonstrate experimentally that the transient field of the photoelectron which is ejected when a ${\rm H_2}^+$ ion is created by photoionization is

sufficient to preferentially localize the bound electron at one side of the molecule. This retroaction of the photoelectron onto its parent molecule was recently suggested in pioneering theoretical work by Serov and Kheifets [10] but has never been recognized in an experiment [9,11–13].

In a broader context the influence of a photoelectron onto its emission source has been discussed for the photoelectric effect in solids, in particular for conducting surfaces. There, it is obvious that the photoelectron will induce a positive mirror charge in the conductor. For this to happen a reservoir of highly movable conduction band electrons is necessary. The time scale on which such mirror charges are formed is still under dispute. For electrons in molecules, such mobility and the ability of the bound electrons to react is hindered by the absence of bands. The essence of the effect is still captured by the concept of polarizability. An escaping electron will transiently polarize the molecular ion left behind. If the emitted electron is slow, the tail of its Coulomb potential is still significantly present while the dissociation of the molecule occurs. In such a case, one might envision that such polarization can freeze out and the charge can become unequally distributed on the fragments, even for homonuclear diatomic molecules. While this may sound obvious from a general perspective, no such observation of broken symmetry has been reported so far [9].

We have used photoabsorption of linearly polarized photons in the range of $E_{\gamma}=19.1$ to 21.1 eV to photoionize H_2 . In this energy range two reaction channels are energetically open:

$$\gamma + H_2 \to e^- + H_2^+(\nu),$$
 (1)

$$\gamma + H_2 \to e^- + H + p^+.$$
 (2)

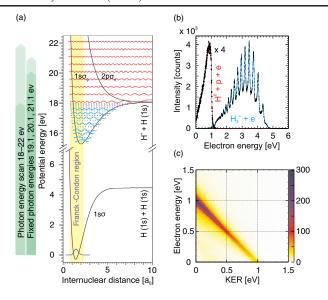


FIG. 1. (a) Relevant potential energy surfaces for the investigated reaction channels: bound states (blue) and continuum states (red). The green arrows indicate the applied photon energies. (b) Measured electron energy distribution for $E_{\gamma}=19.1$ eV: the red line corresponds to reaction channel (2) and is multiplied by 4, while the blue line shows the electron energy for the breakup according to reaction (1). The ion mass enables the separation of both channels. (c) Correlation between electron energy and the KER in channel (2) for $E_{\gamma}=19.1$ eV.

The relevant potential energy surfaces and the measured electron energy distributions for a photon energy $E_{\gamma}=19.1~{\rm eV}$ are shown in Fig. 1(a). The dominant channel is ionization, leaving a bound but vibrationally excited ${\rm H_2}^+$ behind [see Eq. (1)]. The electron energy distribution reflects the distribution of vibrational states [see Fig. 1(b)]. There is less than 5% of Franck-Condon overlap of the ${\rm H_2}$ ground state with the continuum states of ${\rm H_2}^+$ at small internuclear distances [14]. Here, the ${\rm H_2}^+$ will dissociate [see Eq. (2)] and it is this small fraction of events which we will investigate further.

Measurements were carried out at beam line UE112-PGM-1 of the synchrotron radiation scource at the Helmholtz-Zentrum Berlin in single bunch operation using the cold target recoil ion momentum spectroscopy technique [15,16]. The photon beam was crossed with a supersonic H₂ gas jet. The molecules in this jet are in the vibrational ground state. Electrons and ions formed in the overlap region of the photon and the molecular beam were guided by a 6 V/cm electric field onto two microchannel plate detectors (4π collection solid angle) with hexagonal delay line position sensitive readout [17]. All three components of the electron and ion momentum vectors are obtained from the times of flight, the positions of impact on the detector, and the ion mass. For channel (2) the neutral fragment is not detected. Its momentum vector can be determined from the proton and the electron momentum using momentum conservation. We have performed experiments at fixed photon energies of 19.1, 20.1, and 21.1 eV and by scanning the photon energy from 18 to 22 eV.

Figure 1(c) shows the correlation between the electron energy and the kinetic energy release (KER) for channel (2), which is the sum of the proton and H kinetic energy. The diagonal structure indicating a constant sum of all kinetic energies at KER $+E_e=E_\gamma-E_{\rm diss}$ results from energy conservation where $E_{\rm diss}=18.075$ eV is the ionization potential of H₂ plus the dissociation energy of H₂⁺ [18]. The distribution peaks at KER = 0 with a smooth decrease towards a higher KER. The width of the diagonal line is mainly given by the momentum resolution of our spectrometer, which for both particles is best at zero. To make best use of this high resolution at low energy we use energy conservation and calculate KER = $[(KER_m - E_{e,m}/KER_m + E_{e,m}) + 1](E_{\gamma} - E_{diss}/2)$. Here, KER_m corresponds to the KER calculated from the center of mass motion of the system and $E_{e,m}$ is the measured electron energy.

The widely used two step model of molecular photoionization assumes that the process can be split in an ionization step in which the photoelectron escapes from the molecule, leaving it in a superposition of states given by the Franck-Condon principle. In a second step the molecular ion then evolves according to its potential energy surface and the composition of the nuclear wave packet created by the preceding ionization step. The H_2^+ [see Fig. 1(a)] on the ground state $1s\sigma_q$ potential curve is the only one which can lead to a low energy KER. At photon energies close to threshold, this restriction to the $1s\sigma_q$ ionic state is further corroborated by the vanishing Franck-Condon overlap of the H₂ ground state wave function with the energetically accessible part of the $2p\sigma_u$ nuclear wave function. Thus for a KER smaller than 2 eV, according to the two step model, the photoelectron is described by a wave function of pure ungerade parity and the ion by a wave function of pure gerade parity. It has therefore been implicitly assumed or even concluded in several experimental [9,11-13] and theoretical [19,20] studies that the electron angular distribution in the molecular frame should be symmetric with respect to the p and H side of the fragmentation. This consensus has only recently been challenged theoretically by Serov and Kheifets [10]. In order to study such possible asymmetries, we plot the angular distribution of the p-H breakup in a coordinate frame where the x axis is given by the molecular axis (see Fig. 2). The momentum of the electron \vec{k}_e^{lab} is small compared to the momenta of the heavy fragments, but there is still a difference between the proton momentum with respect to the laboratory frame \vec{k}_p^{lab} and the proton momentum in the center of mass of the p-H system $\vec{k}_p^{\text{c.m.}} = \vec{k}_p^{\text{lab}} - 0.5 \cdot \vec{k}_e^{\text{lab}}$ as noted in Ref. [11]. We follow Ref. [11] and plot the angle between the photoelectron momentum and the molecular axis, given by $\vec{k}_p^{\text{c.m.}}$. We

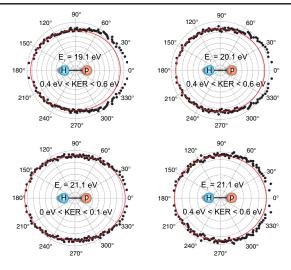


FIG. 2. Angular distribution of the ejected photoelectron. Shown is the angle between the electron momentum vector \vec{k}_e and the molecular axis for photon energies $E_\gamma=19.1,\,20.1,\,$ and 21.1 eV. The KER is restricted to intervals from 0 to 0.1 eV and from 0.4 to 0.6 eV, respectively. The red line is a quadratic function of the form $a+b(\cos(\theta))^2$ fitted to the data in the interval from 90° to 270° as a guide for the eye. The molecular orientation is fixed as shown in the middle of the picture. The statistical error bars which are not visible are smaller than the symbol size. For each histogram, the data points and the fit are mirrored at the horizontal axis for better visual inspection.

integrate over all directions of the polarization and the photon propagation. Our data show a significant asymmetry of the p-H breakup for very slow photoelectrons. The asymmetry decreases with increasing electron energy. Note that energy conservation couples the electron energy and the KER, as these data are taken at a fixed photon energy [see Fig. 1(c)].

Our data show a clear preference for the bound electron to localize during the dissociation at the proton opposite to the direction of the photoelectron. We emphasize that our setup has a collection solid angle of 4π for electrons and ions. Therefore, we can cross-check our data for any possible instrumental source of asymmetry. We have confirmed that the asymmetry flips sides when we select electron emission to the left (right) or up (down) in the laboratory frame (not shown).

To elucidate the origin of the observed symmetry breaking we study its dependence on the KER and the electron energy. For a more qualitative assessment of its strength, we define an asymmetry parameter

$$\delta = \frac{n_p - n_H}{n_p + n_H},\tag{3}$$

where n_p and n_H are the count rate for the break of the p-H bond with the proton towards and opposite to the electron, respectively. Thus, $\delta > 0$ corresponds to the case of the proton emerging in the same hemisphere as the electron

(see Fig. 2) and corresponds to $\beta = \tilde{\beta}/E_e$ in Ref. [10]. For channel (2), we show the asymmetry parameter as a function of the KER at three photon energies: 1, 2, and 3 eV above threshold. As the KER and the electron energy are related by energy conservation, the electron energy corresponding to each photon energy is plotted on an additional axis. The amount of asymmetry rises consistently with decreasing electron energy and an increasing KER. The full lines show the prediction from Ref. [10]. The validity range of this calculation which treats the electron classically is restricted to $E_e \gg$ KER. We have, therefore, cut down the lines showing the theory at $E_e = \frac{1}{3}$ KER. The general trend of the data and the sign and overall size of the effect is well predicted by the very approximate calculation in [10]. In this figure, both the KER and E_e vary as the photon energy is fixed. To unravel if the change of the asymmetry is caused by the electron energy as expected for a polarization effect and from Ref. [10], we have performed an additional experiment in which we scanned the photon energy. This allows us to plot the asymmetry as a function of electron energy for a fixed value of the KER in Fig. 4. The modeling of a retroaction of the photoelectron onto the dissociation in Ref. [10] predicts that the asymmetry is inversely proportional to the electron energy for all KERs. Our data nicely confirm that prediction, as shown by the hyperbolic fit to our data in this figure.

As argued above, ionization and dissociation in two independent steps would lead to symmetric angular distributions. Thus, the validity of the two step model for H₂ at threshold is clearly disproven by our data. We suggest that the observed symmetry breaking is induced by a retroaction of the photoelectron onto the dissociating H_2^+ . This is supported by the qualitative agreement of our data with the predictions in from Ref. [10]. There, the effect of the retroaction is calculated in the simple approximation of a classical electron creating a time dependent field which acts on the molecular wave packet as it dissociates on the potential energy surfaces shown in Fig. 1(a). This model assumes that initial conditions of the wave packet are given by the Franck-Condon overlap with the H₂ ground state. The wave packet evolves initially on the bound and the continuum states of $1s\sigma_g$. The electron creates a time dependent field which decreases as the electron moves away. Once the nuclear wave packet has moved out to a region where the $2p\sigma_{\mu}$ potential energy curve approaches that of the ground state, the field of the electron couples the bound vibrational state and the continuum states of $1s\sigma_a$ to the $2p\sigma_u$ continuum. Therefore, the nuclear wave packet is in a superposition of gerade and ungerade states which describes the localization of the bound electrons. In this scenario the amount of coupling to the ungerade state and thus the asymmetry will increase with the KER. This is because the high energy part of the wave packet reaches the distance at which the coupling to the $2p\sigma_{\mu}$ occurs earlier when the electron is still closer. This trend of an increase of

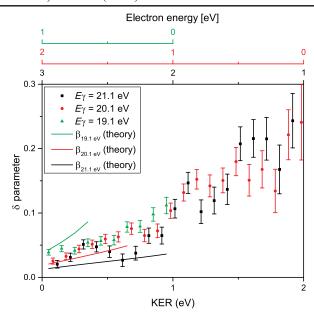


FIG. 3. The asymmetry parameter δ [see Eq. (3)] as a function of the kinetic energy release (KER) for three different photon energies. Experiment: $E_{\gamma}=19.1~{\rm eV}$ (the green triangles), 20.1 eV (the red circles), and 21.1 eV (the black squares). Lines: corresponding predictions from Ref. [10], color coding as for the experiment. The theoretical curves are shown for $E_{e}>2*{\rm KER}$, as the validity range of the calculation is $E_{e}\gg{\rm KER}$.

 δ with the KER is confirmed by the data in Fig. 3. The second scaling one can expect from this model is a decrease of the asymmetry with increasing electron energy. According to Ref. [10], this decrease is inversely proportional to the electron energy and is again in agreement with our observation in Fig. 4.

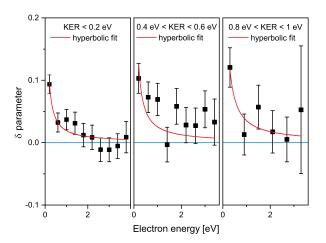


FIG. 4. Asymmetry parameter δ [see Eq. (3)] as a function of electron energy for KER < 0.2 eV (left panel), 0.4 eV < KER < 0.6 eV (center panel), and 0.8 eV < KER < 1 eV (right panel). The asymmetry decreases with higher electron energies, and the trend is in good agreement with the hyperbolic decline $\delta \propto (\text{const}/E_e)$ predicted by Ref. [10] (the red curve).

If one would describe the electron quantum mechanically, one can expect a time dependence which mirrors that of the ionic bound part. Initially, the bound electron is part of the entangled two-electron wave function of H₂. At all times the parity of the two-electron wave function after photon absorption is ungerade. After some time the twoelectron wave function will factorize in a bound gerade and a continuum part at larger distances which is ungerade. If one would perform a measurement at this time, one would find $\delta = 0$. When the nuclei have separated to the region where the $1s\sigma_a$ and $2p\sigma_u$ come close, the electron-electron interaction will entangle the wave function of the bound and the free electron. The two-electron wave function is that of a Bell state of total ungerade symmetry [21]. This entanglement will survive the dissociation and will lead to the measured angular correlations.

In conclusion, we have demonstrated the retroaction of an escaping photoelectron onto its source. In molecules which dissociate after single photoionization, the effect leads to a preferential localization of the remaining bound electron on a site opposite to the continuum electron. For very low energetic photoelectrons, the escaping electron and the fragmentation of the molecule cannot be treated separately and the process can no longer be classified as a Franck-Condon transition. While we have observed this effect in H_2 , the simplest system where it can occur, we speculate that the effect is general for all symmetric molecules and for all processes ejecting an electron.

We expect that ionization by a strong laser field or by electron or ion impact as well as the dissociative ionization of heavier molecules will show similar effects if the escaping electron is slow enough. For larger molecules than diatomics, we expect that the retroaction effect shown here will also influence which of several energetically degenerate bonds will break. An example would be the question of which proton is ejected in a deprotonation of symmetrical hydrocarbons, which might be determined by a slow escaping photoelectron inducing a polarization in the molecular ion.

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