

## Plasma Luminescence from Femtosecond Filaments in Air: Evidence for Impact Excitation with Circularly Polarized Light Pulses

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Filaments produced in air by intense femtosecond laser pulses emit UV luminescence from excited  $N_2$  and  $N_2^+$  molecules. We report on a strong dependence at high intensities ( $I \geq 1.4 \times 10^{14}$  W/cm<sup>2</sup>) of this luminescence with the polarization state of the incident laser pulses. We attribute this effect to the onset of new impact excitation channels from energetic electrons produced with circularly polarized laser pulses above a threshold laser intensity.

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Intense femtosecond laser pulses launched in transparent media experience filamentary propagation, a generic phenomenon in solids, liquids, and gases [1,2]. A distinctive signature of filamentation in air is the formation of a long, bright channel of underdense plasma in the wake of the propagating laser pulses. Much interest has been devoted to the study of the luminescence of this plasma in air. As shown first by the group at Laval University, it differs from that obtained with laser pulses of longer duration [3]. Instead of nitrogen and oxygen atomic lines superimposed on a broad continuum, it consists of discrete lines due to excited nitrogen molecules. These lines correspond to transitions between excited triplet states  $C^3\Pi_u^+$  and  $B^3\Pi_g^+$  of neutral nitrogen molecules (2nd positive system of  $N_2$ ) and between second excited state  $B^2\Sigma_u^+$  and ground state  $X^2\Sigma_g^+$  of nitrogen molecular ions (1st negative system of  $N_2^+$ ) [3,4]. A detailed understanding of the process leading to this filament luminescence is important for several reasons. Filament luminescence has been used as a tool to characterize the length and width of the plasma column and to extract the electron temperature and laser intensity inside filaments [5–8]. Because of its dependence on an external electric field, it has found application in the remote measurement of dc electric field or intense terahertz radiation [9,10]. A good understanding of its excitation process is also central to an interpretation of the recently discovered laser action from either neutral or ionized nitrogen molecules inside filament plasmas [11–17].

In this Letter, we measure and discuss the dependence of the luminescence from both  $N_2$  and  $N_2^+$  molecules on polarization of the incident femtosecond laser pulse. It reveals new nonradiative routes to populate excited neutral and ionic molecular levels through electron collisions. A minimum electron kinetic energy is necessary in order to achieve impact excitation of neutral or ionized molecules. In the range of laser intensities of this study, numerical simulations predict that this threshold kinetic energy is only

obtained with circular polarization, in good agreement with our experimental observations.

In the experiment, femtosecond laser pulses with pulse energy up to 10 mJ (45 fs, 800 nm) from a commercial laser system (Thales Laser, Alpha 100) were focused by an  $f = 1000$  mm convex lens in ambient air. A quarter-wave plate ( $\lambda/4$ ) was installed before the focal lens to change the laser polarization from linear to circular. Plasma channels with length varying from a few millimeters to several centimeters were created, depending on the incident pulse energy. In order to detect the spontaneous luminescence, a lens of focal distance  $f = 25$  mm was employed to collimate the luminescence emitted in the direction orthogonal to the laser propagation axis and a second lens with  $f = 100$  mm was used to focus the collected emission onto the entrance slit of a monochromator (Jobin-Yvon H-20 UV). The luminescence signal from the monochromator was measured by a photomultiplier tube (Hamamatsu, model: H10722). In order to resolve longitudinally the luminescence signal along the filament axis, a rectangular diaphragm (1 mm  $\times$  6 mm) was placed 1 mm away from the filament axis.

We first measured the luminescence spectrum for linearly and circularly polarized laser pulses at two representative incident laser pulse energies  $E_{in}$ . Detection was performed around the middle position of the plasma in each case, where the maximum fluorescence signals along the filament axis in the case of linear laser polarization were observed. Figure 1 shows the results for  $E_{in} = 250 \mu\text{J}$ , just below the threshold for filamentation [Fig. 1(a)] and at 8.3 mJ, where the peak power is 30 times higher than the critical power for filamentation [Fig. 1(b)]. The corresponding peak laser intensities were determined by measuring the laser flux and pulse duration transmitted through circular diaphragms of 80  $\mu\text{m}$  diameter placed in the middle of the plasma string. It yields  $I = 3 \times 10^{13}$  W/cm<sup>2</sup> in case (a) and a value  $I \geq 1.4 \times 10^{14}$  W/cm<sup>2</sup> for incident pulse energies above 1 mJ, when filaments are formed [17]. No significant

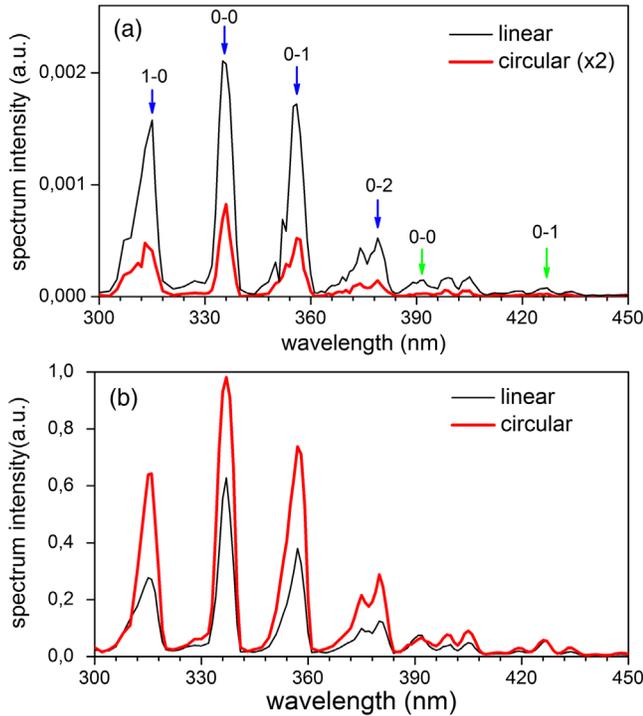


FIG. 1 (color online). Luminescence spectrum of the air plasma filaments for linearly and circularly polarized laser pulses of 250  $\mu\text{J}$  (a) and 8.3 mJ (b). In (a), the signal obtained with circular laser polarization is multiplied by a factor of 2 for visibility. The vibration quantum numbers related to the second positive system of the  $\text{N}_2$  and the first negative system of  $\text{N}_2^+$  are denoted.

difference of laser intensity was observed for linear and circular laser polarization. The luminescence lines visible in Fig. 1 belong to two categories, as already mentioned in the introduction: transitions between excited triplet states of the neutral nitrogen molecules or transitions between different ionic states  $\text{N}_2^+$  of the nitrogen molecule [3]. Their precise identification in terms of vibrational levels is given in Fig. 1. At lower intensities, linearly polarized pump light is more efficient. Upon increase of laser intensity, there is a reversal in the relative intensity of luminescence between linear and circular pump laser polarization.

The same trend is better observed in the dependence of the 337 nm (representative of luminescence from neutral molecules) and 391 nm signals (due to molecular ion emission) as a function of pump laser ellipticity. In Fig. 2, results are presented for 4 different pulse energies as a function of rotation angle of the  $\lambda/4$  wave plate. The angles  $\varphi = 90^\circ \times m$  correspond to linearly polarized laser, with  $m = 0, 1, 2, 3$ . The angles  $\varphi = 45^\circ + 90^\circ \times m$  correspond to circular laser polarization. With a laser pulse energy of 250  $\mu\text{J}$ , the dominance of linear laser polarization over elliptical and circular is observed for both 337 and 391 nm emission lines [Fig. 2(a) and 2(a')]. With a pulse energy of 600  $\mu\text{J}$ , the intensity of the triplet luminescence line at

337 nm becomes independent of the pump polarization [Fig. 2(b)]. Upon a further increase of the pump pulse energy, the signal at 337 nm becomes more intense with a circularly polarized pump [Fig. 2(c) and 2(d)]. Concerning the emission of  $\text{N}_2^+$  at 391 nm, its relative intensity increases with circularly polarized laser pulses [see Fig. 2(b') and 2(c')], until almost no dependence on ellipticity is observed for pulses of 10 mJ [Fig. 2(d')].

We have studied the evolution of luminescence lines at 337 nm along the filament axis  $z$  for both linear and circular laser polarization, for three different incident laser pulse energies. Below the threshold for filamentation, the length of the plasma corresponds to the Rayleigh distance calculated by assuming linear laser pulse propagation. As expected for such a case, the plasma luminescence peaks around the geometric focus. The line is more intense along the  $\sim 25$  mm plasma string with linearly polarized pump. At higher laser energies, the plasma string moves towards the laser, a signature of filamentation. The luminescence obtained with circularly polarized light becomes equal to that obtained with linearly polarized laser pulse of energy  $E_{\text{in}} = 1.1$  mJ and predominant at still higher laser energies ( $E_{\text{in}} = 10$  mJ). We verified that other emission lines at 357, 391, and 428 nm exhibit a similar behavior.

How should we understand the dependence of the luminescence with pump laser polarization? Populating the excited ionic state  $\text{N}_2^+(B^2\Sigma_u^+)$  during filamentation in air is generally accepted as being due to direct high-field photon ionization of inner-valence electrons of the nitrogen molecules [18,19]. Linear laser polarization is known to be more efficient [18,19]. Direct high-field photonic excitation of the triplet state  $\text{N}_2(C^3\Pi_u^+)$  is a spin forbidden process and therefore unlikely. Two indirect excitation processes have been proposed. A first scheme consists in a dissociative recombination through the processes:  $\text{N}_2^+ + \text{N}_2 + \text{N}_2 \rightarrow \text{N}_4^+ + \text{N}_2$  followed by  $\text{N}_4^+ + e \rightarrow \text{N}_2(C^3\Pi_u^+) + \text{N}_2$  [20]. Another more recent scenario proposes that collision-assisted intersystem crossing from excited singlet states is the dominant path to produce the triplet state, while the dissociative recombination would be a minor contributor [21]. In any case, both processes should generate a larger signal with a linearly polarized pump. In the case of dissociative recombination, the final density of  $\text{N}_2(C^3\Pi_u^+)$  molecules depends on the density of  $\text{N}_2^+$ , which is more effectively produced by linearly polarized laser in the laser intensity regime of  $10^{13}$ – $10^{15}$   $\text{W}/\text{cm}^2$  [18,19]. With the intersystem crossing mechanism, one expects a similar dependence on laser polarization, because the transition from the fundamental singlet state of  $\text{N}_2$  to an intermediate singlet state is more effective with linearly polarized laser pulses. So the question is why circularly polarized laser pulses become more efficient at higher laser intensity.

An important difference between linear and circular laser polarization in gas plasma generation is the kinetic energy

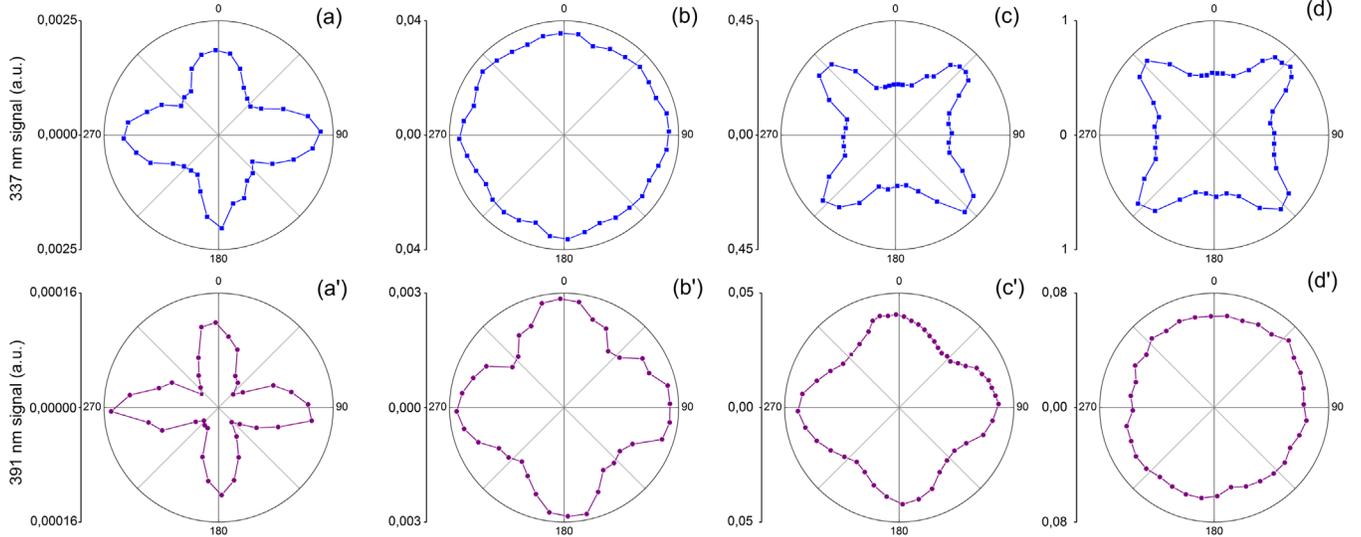


FIG. 2 (color online). Luminescence at 337 nm [(a)–(d)] and 391 nm [(a')–(d')] as a function of the rotation angle of the quarter-wave plate. The incident pulse energy was 250  $\mu\text{J}$ , 600  $\mu\text{J}$ , 2.7 mJ, and 10 mJ for (a) and (a'), (b) and (b'), (c) and (c'), (d) and (d'), respectively. All the measurements were performed around the center of the filament. Angle  $0^\circ$  corresponds to linearly polarized light.

of free electrons remaining after the passage of the intense laser field [22,23]. With linearly polarized laser pulses, free electrons are left with low kinetic energy because they experience alternative acceleration and deceleration by the laser field during each optical cycle of the pulse. By contrast, with a circularly polarized laser, electrons are always accelerated away from the molecular ion. As discussed below, free electrons acquire an average energy of  $\sim 2U_p$  at the end of the laser pulse, where  $U_p = e^2/c\epsilon_0 m_e \times I/2\omega_0^2$  is the ponderomotive potential of the electron in a linearly polarized laser field with  $\epsilon_0$ ,  $m_e$ ,  $I$ ,  $\omega_0$  being the vacuum permittivity, the mass of the electron, the intensity, and frequency of the laser field. Therefore, a large number of electrons with a kinetic energy around  $2U_p$  are produced inside a filament.

The distribution of transverse electron kinetic energies can be predicted by semianalytical laws: integration of Newton's equations for electron motion leads to a transverse momentum  $\vec{p}(t) = -e[\vec{A}(t) - \vec{A}_0]$ , where  $\vec{A}(t)$  denotes the vector potential at instant  $t$  and  $\vec{A}_0$  denotes its counterpart when the electron is liberated, at rest. After the passage of the pulse,  $\vec{A}(t)$  vanishes. The transverse momentum becomes  $p(\infty) = eA_0$ , and the transverse kinetic energy reads  $E_{\text{kin}}(t_0) = e^2 A_0^2 / 2m$ , where  $t_0$  indicates that the electron was liberated at instant  $t_0$  within the pulse. We can infer the vector potential by integration of  $\vec{E} = -\partial\vec{A}/\partial t$  by using an analytical form for the electric field with a cosine envelope

$$E = \begin{cases} E_0 \cos(\frac{\pi t}{T}) [\cos(\omega_0 t + \theta) \vec{u}_x + \varepsilon \sin(\omega_0 t + \theta) \vec{u}_y] & \text{for } -T/2 < t < T/2 \\ 0 & \text{for } t < -T/2 \quad \text{and} \quad t > T/2 \end{cases} \quad (1)$$

where  $\theta$  denotes an arbitrary carrier envelope phase. The kinetic energy of an electron, born at time  $t_0$ , after acceleration by the pulse reads

$$E_{\text{kin}} = 2U_p \cos^2(\pi t_0/T) [1 - (1 - \varepsilon^2) \cos^2(\omega_0 t_0 + \theta)]. \quad (2)$$

Therefore, all electrons generated between  $t_0$  and  $t_0 + \Delta t$ , with probability  $(\partial n_e / \partial t)_{t_0} \times (\Delta t / n_{e,\infty})$ , where  $n_e(t)$  denotes the electron density calculated from the rate equations and  $n_{e,\infty}$  denotes the total electron density generated by the pulse, will have a kinetic energy between  $E_{\text{kin}}(t_0)$  and  $E_{\text{kin}}(t_0 + \Delta t)$ . We retrieve that the maximum

kinetic energy of the electrons is  $2U_p$  for a circularly polarized pulse ( $\varepsilon = 1$ ) when it is born at the peak of the field envelope ( $t = 0$ ). A parametric representation of the distribution of kinetic energies is presented as a continuous curve in Fig. 3. In the case of linear laser polarization most electrons are left with energy below 1 eV, as shown in Fig. 3(a). By contrast, an almost monoenergetic distribution around 14.6 eV is achieved for circular laser polarization [Fig. 3(c)]. An intermediate distribution is obtained with an elliptically polarized pulse, as shown, for instance, for  $\varepsilon = 1/2$  in Fig. 3(b). We also calculated the electron energy distribution after the passage of the laser pulses by

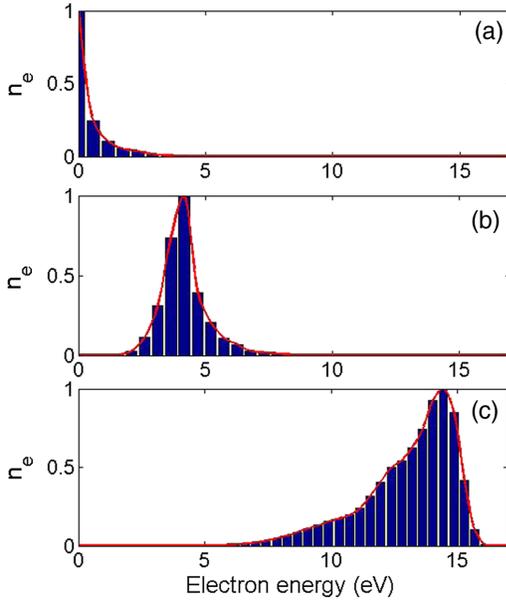


FIG. 3 (color online). Calculated electron energy distribution in the case of linearly (a), elliptically with  $\epsilon = 1/2$  (b), and circularly (c) polarized laser pulses. The laser intensity used in the calculation was  $1.4 \times 10^{14}$  W/cm<sup>2</sup>. The red lines represent analytic results and the bars numerical simulation.

numerical simulations. In the simulations, we assume that electrons are generated by optical field ionization of oxygen and nitrogen molecules, described by a set of coupled rate equations. Liberated electrons are assumed at rest and are accelerated by the electromagnetic (Lorentz) force mainly in the polarization plane. We calculated the classical motion of a set of electrons under the action of the Lorentz force and then performed statistics by weighting each electron by its probability to be liberated at a given instant during the pulse. The results of these numerical simulations are presented by the bars in Fig. 3 and agree very well with that of the semianalytical analysis.

Electrons with high kinetic energy can populate excited states of N<sub>2</sub> and N<sub>2</sub><sup>+</sup> via a collision process. Process N<sub>2</sub>(X<sup>1</sup>Σ<sub>g</sub><sup>+</sup>) + e → N<sub>2</sub>(C<sup>3</sup>Π<sub>u</sub><sup>+</sup>) + e populating the triplet molecular state opens up if the electron energy exceeds a rather sharp threshold energy of ~11 eV, with a cross section peaking at 14.5 eV with a value of 0.58 Å<sup>2</sup> [24]. In a traditional nitrogen laser pumped by electric discharge, it is actually this inelastic collision that gives rise to population inversion between the C<sup>3</sup>Π<sub>u</sub><sup>+</sup> and B<sup>3</sup>Π<sub>g</sub><sup>+</sup> states [25]. In our case, the energetic electrons produced inside the filament plasma with a circularly polarized laser pulse lead to efficient population buildup in the C<sup>3</sup>Π<sub>u</sub><sup>+</sup> state, and thus stronger luminescence at 337 nm, as observed in Fig. 2(d). In the presence of abundant energetic free electrons, electron collision excitation of the excited ionic state B<sup>2</sup>Σ<sub>u</sub><sup>+</sup> is also possible through the process N<sub>2</sub>(X<sup>1</sup>Σ<sub>g</sub><sup>+</sup>) + e → N<sub>2</sub><sup>+</sup>(B<sup>2</sup>Σ<sub>u</sub><sup>+</sup>) + 2e. The corresponding threshold electron energy is 18.75 eV with a maximum effective collision

section of ~0.01 Å<sup>2</sup> [24,26,27]. With circularly polarized laser pulses of intensity more than  $1.4 \times 10^{14}$  W/cm<sup>2</sup>, energetic electrons can reach the required energy.

We have verified the interpretation of new collision-induced excitation routes by repeating at low air pressure the ellipticity measurements shown in Fig. 2. Low gas pressure effectively suppresses collisions and therefore is also expected to suppress the corresponding excitation routes. The result is shown in Fig. 4 for the signal at 337 nm with a gas pressure of 10 mbar and a laser pulse energy of 350 μJ. Since the propagation of the pulse is linear at such a low pressure, the intensity at the focus can be readily estimated from diffraction to be  $1.45 \times 10^{14}$  W/cm<sup>2</sup>, of the same order of magnitude as in a filament at normal pressure. As can be seen, the response to ellipticity is the same as that observed at lower intensities and normal pressure [Fig. 2(a)] although the reason is different. At normal pressure and low laser intensity, collisions are present, but the electron energy is insufficient to excite the molecules [Fig. 2(a)]; at low pressure and high laser intensity (Fig. 4), electrons have the required kinetic energy but collisions are suppressed. A similar result is obtained for the line at 391 nm when the laser pulse energy is varied between 250 and 500 μJ.

In conclusion, we have demonstrated that the luminescence emitted by neutral and singly ionized N<sub>2</sub> molecules inside femtosecond laser filaments in air depends strongly on the polarization state of the incident laser pulses. At lower laser pump energies, the predominance of the luminescence from both species with a linearly polarized pump is simply explained by the higher optical field ionization rate that constitutes the prime excitation route. At higher laser intensity, new excitation routes become available, due to the presence of electrons with high kinetic energy left after the laser pulse. A collision-assisted mechanism populates the excited triplet state of N<sub>2</sub> and

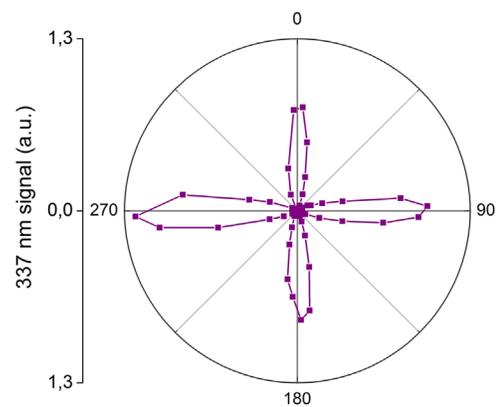


FIG. 4 (color online). Luminescence at 337 nm as a function of the rotation angle of the quarter-wave plate for air pressure of 10 mbar. The incident pulse energy was 350 μJ and the calculated laser intensity is  $1.45 \times 10^{14}$  W/cm<sup>2</sup>.

the excited ionic molecular state of nitrogen directly from the ground state of neutral molecules. These findings are important for the understanding of the stimulated radiation from filaments and may find applications in remote sensing of electric field and THz radiation. We believe that these findings are not just restricted to the laser filamentation process and they should intervene in other laser-gas interaction phenomena, such as laser-induced gas breakdown and its relevant applications.

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- [1] A. Couairon and A. Mysyrowicz, *Phys. Rep.* **441**, 47 (2007).
- [2] S. L. Chin, S. A. Hosseini, W. Liu, Q. Luo, F. Théberge, N. Aközbek, A. Becker, V. P. Kandidov, O. G. Kosareva, and H. Schroeder, *Can. J. Phys.* **83**, 863 (2005).
- [3] A. Talebpour, S. Petit, and S. L. Chin, *Opt. Commun.* **171**, 285 (1999).
- [4] F. Martin, R. Mawassi, F. Vidal, I. Gallimberti, D. Comtois, H. Pepin, J. C. Kieffer, and H. P. Mercure, *Appl. Spectrosc.* **56**, 1444 (2002).
- [5] S. A. Hosseini, Q. Luo, B. Ferland, W. Liu, N. Aközbek, G. Roy, and S. L. Chin, *Appl. Phys. B* **77**, 697 (2003).
- [6] A. Filin, R. Compton, D. A. Romanov, and R. J. Levis, *Phys. Rev. Lett.* **102**, 155004 (2009).
- [7] L. Shi, W. Li, Y. Wang, X. Lu, L. Ding, and H. Zeng, *Phys. Rev. Lett.* **107**, 095004 (2011).
- [8] S. Xu, X. Sun, B. Zeng, W. Chu, J. Zhao, W. Liu, Y. Cheng, Z. Xu, and S. L. Chin, *Opt. Express* **20**, 299 (2012).
- [9] J. Liu, J. Dai, S. L. Chin, and X.-C. Zhang, *Nat. Photonics* **4**, 627 (2010).
- [10] K. Sugiyama, T. Fujii, M. Miki, M. Yamaguchi, A. Zhidkov, E. Hotta, and K. Nemoto, *Opt. Lett.* **34**, 2964 (2009).
- [11] A. Dogariu, J. B. Michael, M. O. Scully, and R. B. Miles, *Science* **331**, 442 (2011).
- [12] J. Yao, B. Zeng, H. Xu, G. Li, W. Chu, J. Ni, H. Zhang, S. L. Chin, Y. Cheng, and Z. Xu, *Phys. Rev. A* **84**, 051802(R) (2011).
- [13] Q. Luo, W. Liu, and S. L. Chin, *Appl. Phys. B* **76**, 337 (2003).
- [14] D. Kartashov, S. Ališauskas, G. Andriukaitis, A. Pugžlys, M. Shneider, A. Zheltikov, S. L. Chin, and A. Baltuška, *Phys. Rev. A* **86**, 033831 (2012).
- [15] G. Point, Y. Liu, Y. Brelet, S. Mitryukovskiy, P. Ding, A. Houard, and A. Mysyrowicz, *Opt. Lett.* **39**, 1725 (2014).
- [16] Y. Liu, Y. Brelet, G. Point, A. Houard, and A. Mysyrowicz, *Opt. Express* **21**, 22791 (2013).
- [17] S. Mitryukovskiy, Y. Liu, P. Ding, A. Houard, and A. Mysyrowicz, *Opt. Express* **22**, 12750 (2014).
- [18] A. Talebpour, A. Bandrauk, and S. L. Chin, in *Multiphoton Processes*, edited by L. F. Dimauuro, R. R. Freeman, and K. C. Kulander (AIP, New York, 2000), p. 508.
- [19] A. Becker, A. D. Bandrauk, and S. L. Chin, *Chem. Phys. Lett.* **343**, 345 (2001).
- [20] H. L. Xu, A. Azarm, J. Bernhardt, Y. Kamali, and S. L. Chin, *Chem. Phys.* **360**, 171 (2009).
- [21] B. R. Arnold, S. Roberson, and P. M. Pellegrino, *Chem. Phys.* **405**, 9 (2012).
- [22] P. H. Bucksbaum, M. Bashkansky, R. R. Freeman, and T. J. McIlrath, and L. F. DiMauro, *Phys. Rev. Lett.* **56**, 2590 (1986).
- [23] P. B. Corkum, N. H. Burnett, and F. Brunel, *Phys. Rev. Lett.* **62**, 1259 (1989).
- [24] Y. Itikawa, *J. Phys. Chem. Ref. Data* **35**, 31 (2006).
- [25] R. S. Kunabenchi, M. R. Gorbali, and M. I. Savadatt, *Prog. Quantum Electron.* **9**, 259 (1984).
- [26] D. H. Crandall, W. E. Kauppila, R. A. Phaneuf, P. O. Taylor, and G. H. Dunn, *Phys. Rev. A* **9**, 2545 (1974).
- [27] O. Nagy, *Chem. Phys.* **286**, 109 (2003).