

Generalized Energy Equipartition in Harmonic Oscillators Driven by Active Baths

Claudio Maggi,^{1,*} Matteo Paoluzzi,² Nicola Pellicciotta,¹ Alessia Lepore,¹ Luca Angelani,² and Roberto Di Leonardo^{1,2}

¹Dipartimento di Fisica, Università di Roma "Sapienza," I-00185 Roma, Italy

²CNR-IPCF, UOS Roma, Dipartimento di Fisica Università Sapienza, I-00185 Roma, Italy

(Received 18 July 2014; revised manuscript received 10 October 2014; published 3 December 2014)

We study experimentally and numerically the dynamics of colloidal beads confined by a harmonic potential in a bath of swimming *E. coli* bacteria. The resulting dynamics is well approximated by a Langevin equation for an overdamped oscillator driven by the combination of a white thermal noise and an exponentially correlated active noise. This scenario leads to a simple generalization of the equipartition theorem resulting in the coexistence of two different effective temperatures that govern dynamics along the flat and the curved directions in the potential landscape.

DOI: 10.1103/PhysRevLett.113.238303

PACS numbers: 82.70.Dd, 05.40.-a, 47.63.Gd, 87.17.Jj

Introduction.—A remarkable result of equilibrium statistical mechanics is the theorem of energy equipartition. In its simplest form, the theorem states that each quadratic term in the Hamiltonian contributes with the same amount of energy $k_B T/2$ to the average energy of the system [1]. In the case of a harmonic oscillator this applies to both kinetic and potential energies. In the colloidal realm, particle motions are strongly overdamped and velocity fluctuates on a time scale that is often hardly accessible [2]. However, the value of kinetic energy imposed by the equipartition theorem is reflected in a diffusion coefficient that is proportional to the mean squared velocity [3]: $D_T = \mu k_B T$. Therefore, for a colloidal harmonic oscillator the equipartition theorem establishes a link between the thermal diffusion constant D_T and the average potential energy $U = D_T/2\mu$. Out-of-equilibrium systems are frequently found in nature and the search for generalized equipartition laws constitutes a very interesting and hot topic [4,5]. In particular, there is a growing family of off-equilibrium, active colloidal particles that are able to harness some form of locally stored energy to self-propel in persistent random walks [6,7]. An interesting example is provided by passive colloidal tracers suspended in active baths of swimming bacteria. Over time scales that are larger than the persistence time of active forces, those particles display a diffusive behavior with a diffusivity D^* that can be orders of magnitude larger than the thermal counterpart D_T [8]. It is found that, whenever the external potential changes smoothly on the characteristic length scale of the persistent motion, the system is well described by a quasi-Boltzmann distribution with an effective temperature given by $k_B T_{\text{eff}} = D^*/\mu$ [9,10]. In this limiting case the equipartition theorem is recovered in its original form, being a straightforward consequence of Boltzmann statistics. However, when the external potential does not meet these requirements, Boltzmann statistics breaks down [11,12] and an equilibriumlike picture with one single effective temperature fails. This is particularly evident in the case of

rectification effects, as those investigated in Refs. [13–15]. In these works it has been shown that a bacterial bath can spontaneously induce the unidirectional motion of nanofabricated asymmetric objects. Similarly, microfabricated structures can rectify the motion of motile bacteria and accumulate them in specific spatial regions [16]. Moreover, passive colloidal tracers can be delivered onto target sites by the rectification of fluctuating forces from a bacterial bath [17,18]. Failure of Boltzmann statistics also leads to novel nonequilibrium effects such as the emergence of effective attraction in the presence of purely repulsive potentials [19,20]. In this context, a simple generalization of equipartition could seem unlikely.

In this Letter we demonstrate that, in the case of active harmonic oscillators, the average value of potential energy is still linked to the diffusivity by a simple generalization of the equipartition theorem. As a consequence, the effective temperature associated with the potential energy is always lower than the one obtained from the free diffusion coefficient. We investigate experimentally and numerically the dynamics of colloidal beads, subject to a harmonic potential, suspended in a bath of swimming *E. coli* cells. The elastic force field is obtained experimentally by placing the microspheres in a cylindrical microcapillary. Sedimenting colloids fluctuate near the bottom of the capillary where they experience a near-perfect harmonic potential.

Experiment.—Motile *E. coli* cells are prepared following the protocol described in [21]. Silica beads of radius $a = 3.5 \mu\text{m}$ are first diluted in deionized water and then mixed with bacteria directly on a glass slide. The final bacteria density is estimated to be $\sim 10^{10}$ cells/ml. The bacteria-colloids solution is loaded in a microcapillary glass tube (Vitrocom) of internal radius $R = 25 \mu\text{m}$ by capillarity. The sample is left open for a few minutes before sealing with index matching oil. This procedure results in the formation of two air bubbles at the edges of the capillary tube as shown in Fig. 1(a). Residual distortions

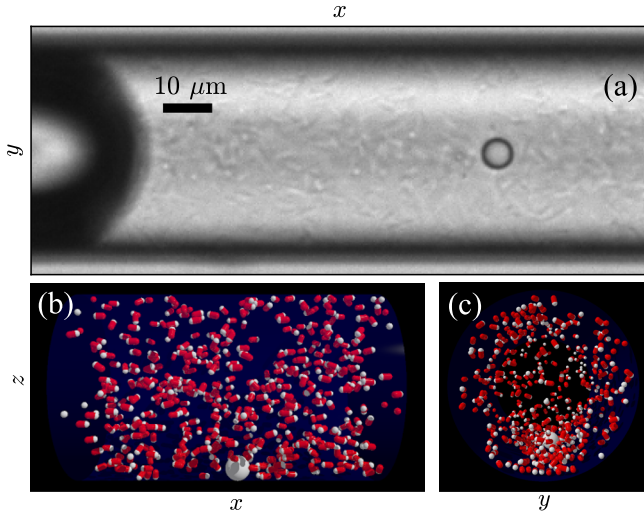


FIG. 1 (color online). (a) A $3.5 \mu\text{m}$ radius silica bead is suspended in a bath of motile *E. coli* bacteria filling a $25 \mu\text{m}$ radius capillary glass tube. (b),(c) Snapshots from the numerical simulation.

due to the internal glass-water interface have a negligible effect as shown by the absence of anisotropies in both particle shape and diffusion in the absence of bacteria (see [21]).

Colloidal beads sediment at the bottom of the capillary and align along the tube axis with an average distance of

about $20 \mu\text{m}$. We collect bright field images using a $20\times$, NA 0.25 microscope objective. After background subtraction and thresholding we obtain particle trajectories by center-of-mass tracking. We report data for 10 beads that were simultaneously tracked for 100 s at a rate of 100 frames/s. The beads span a capillary length of approximately $250 \mu\text{m}$ probing a local environment characterized by a bacterial activity that decreases as the distance from the trapped air bubble increases.

Simulations.—The numerical simulations are performed by considering spherical colloidal particles of radius a immersed in a bath of self-propelling dumbbells following a “run and tumble” dynamics. Both particles and bacteria are confined in a cylindrical volume as shown in Figs. 1(b) and 1(c). All interactions are modeled by repulsive steric forces. In addition, particles experience a gravitational force $f_z = -mg$ due to gravity where m is the buoyant mass of the colloidal particle and g is the acceleration due to gravity. We include Brownian motion only for particles dynamics and neglect hydrodynamic interactions [13,19]. A detailed description of the simulation can be found in [21].

Results.—The mean-squared displacement $\langle \Delta x^2(t) \rangle$ (MSD) along the capillary axis is shown in Fig. 2(a) for two beads located at about 30 and $230 \mu\text{m}$ from the edge of the air bubble. Both particles show an MSD characterized by a superdiffusive regime at short times followed by a diffusive dynamics at longer time scales. This is in

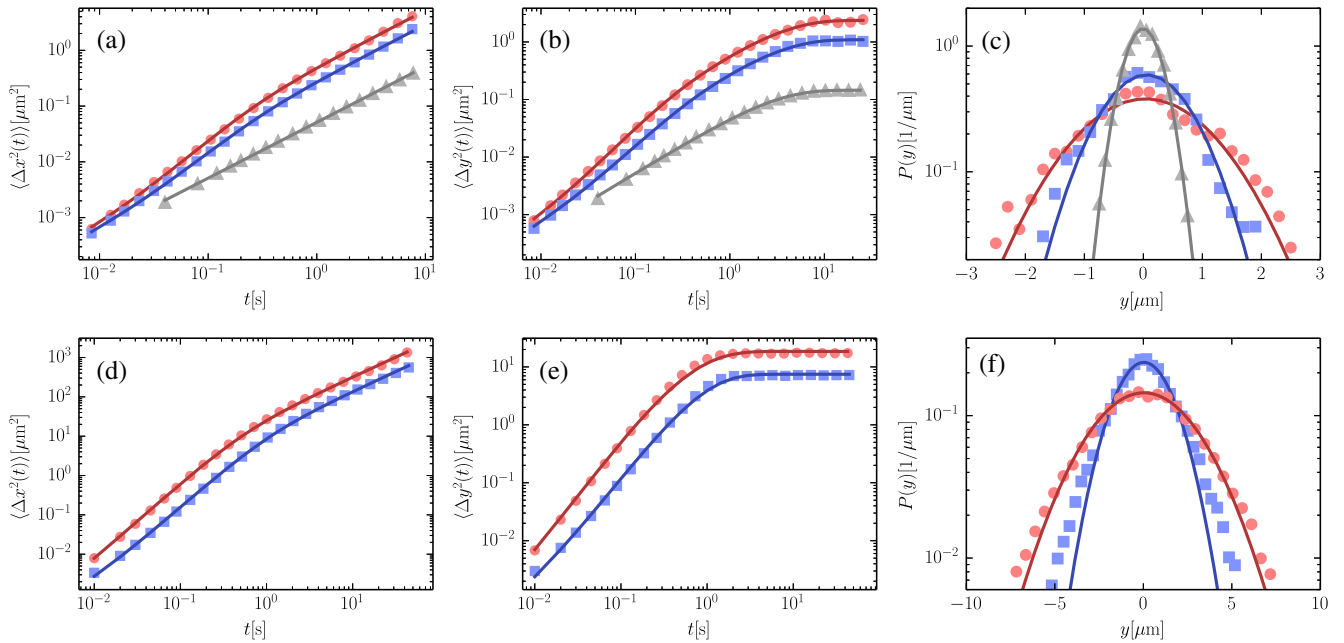


FIG. 2 (color online). (Top panels—experimental data) (a) MSD along the capillary axis of two colloids located at 30 (circle) and $230 \mu\text{m}$ (square) from the air bubble (oxygen reservoir), (triangle) MSD of the colloid in the absence of bacteria [21]. The solid lines are fits with Eq. (2). (b) MSD of the same particles in (a) along an orthogonal direction to the capillary axis. Solid lines are fits with Eq. (3). (c) Probability distributions of y of the same two colloids in (a) and (b), the solid lines are fits with a Gaussian distribution. (Bottom panels—simulation data) (d), (e), and (f) are the same as (a), (b), and (c) for two simulated colloids in a bacterial bath with varying average swimming speed $30 \mu\text{m/s}$ (circle) and $15 \mu\text{m/s}$ (square).

qualitative agreement with the results of previous experiments performed on flat surfaces [8,17,29]. Along the y axis the particle motion is constrained by the curved capillary surface and the MSD saturates at a constant value [Fig. 2(b)]. The MSD along both axes decreases when moving away from the air bubble, due to a decrease of bacterial activity, as discussed in more detail in the following. Figure 2(c) shows the probability distribution $P(y)$ together with the best Gaussian fits.

A very similar behavior is observed in simulations where the diffusivity of colloids is tuned by varying the average swimming speed of bacteria [see Figs. 2(d), 2(e), and 2(f)]. In particular, when bacteria are faster the diffusivity of the particles increases, but also the transition between the ballistic and the diffusive regime shifts to a shorter time scale.

Modeling and discussion.—The motion of the particles is mostly constrained on the capillary surface. In this case the z coordinate of the center of mass of the colloidal particle is directly given by $z = -\sqrt{\rho^2 - y^2} \sim -\rho + y^2/2\rho$ with $\rho = R - a$. The force acting along the y axis is therefore computed as $f_y = -mg\partial z/\partial y \sim -mgy/\rho$. The resulting force field is then well approximated by an elastic force acting along the y axis with a spring constant k defined by $k = mg/\rho$. The cylindrical geometry of the capillary results in $f_x = 0$. In addition to the deterministic force \mathbf{f} the colloids are subject to thermal fluctuations and to interactions with swimming bacteria. To account for these we model the dynamics of the beads with the following stochastic differential equation:

$$\dot{\mathbf{r}} = \mu\mathbf{f}(\mathbf{r}) + \boldsymbol{\eta}^T + \boldsymbol{\eta}^A, \quad (1)$$

where $\mathbf{r}(t) = (x(t), y(t))$, $\mathbf{f} = (0, -ky)$, and μ is the mobility of the colloidal particle. We assume that the noise term can be split into two independent components: the standard Langevin thermal noise $\boldsymbol{\eta}^T$ with $\langle \eta_\alpha^T(t)\eta_\beta^T(t') \rangle = 2D_T\delta_{\alpha\beta}\delta(t-t')$ and $D_T = \mu k_B T$, an active noise $\boldsymbol{\eta}^A$ that is exponentially time correlated [18] $\langle \eta_\alpha^A(t)\eta_\beta^A(t') \rangle = D_A\delta_{\alpha\beta}\exp(-|t-t'|/\tau)/\tau$, where α, β represent individual Cartesian components. From Eq. (1) we can compute the MSD along x :

$$\langle \Delta x^2(t) \rangle = 2D_T t + 2D_A[t - \tau(1 - e^{-t/\tau})], \quad (2)$$

and the MSD of $y(t)$:

$$\langle \Delta y^2(t) \rangle = \frac{2D_T}{\mu k}(1 - e^{-\mu k t}) + \frac{2D_A}{\mu k} \frac{1 - e^{-\mu k t} - \mu k \tau(1 - e^{-t/\tau})}{1 - (\mu k \tau)^2}. \quad (3)$$

Equations (2) and (3) provide an excellent fit to the MSD along both x and y [Figs. 2(a) and 2(b)]. Along both axes

the parameter D_A shows a clear dependence on the average position of the particle $\langle x \rangle$ with respect to the edge of the air bubble. D_A is found to decrease from 0.31 to 0.14 $\mu\text{m}^2/\text{s}$ upon increasing $\langle x \rangle$ by a few hundreds of microns. This suggests that bacterial motility depends on the concentration of oxygen that is progressively consumed by bacteria along the capillary [30]. Differently, the fitting parameters D_T , τ , and μk do not show any clear dependence on the distance from the air bubble. The obtained averages over all particles are $\tau = 0.093(0.015)$ s, $D_T = 0.030(0.002)$ $\mu\text{m}^2/\text{s}$ along x and $\tau = 0.097(0.023)$ s, $D_T = 0.046(0.013)$ $\mu\text{m}^2/\text{s}$, $\mu k = 0.289(0.067)$ s^{-1} along y , where standard deviations are shown in brackets. MSD in the absence of bacteria are reported in Fig. 2. The corresponding fitting parameters are $\mu k = 0.36$ s^{-1} and $D_T = 0.025$ and 0.026 $\mu\text{m}^2/\text{s}$ along x and y , respectively. Those values are compatible with those found in the presence of bacteria, although correlations between fitting parameters result in larger uncertainties in D_T along y . The values of D_T , with or without bacteria, are about a factor of 2 smaller than the bulk value which can be attributed to increased drag due to wall effects [21,31]. We can now estimate the sedimentation length of our colloids in the bacterial bath as $(D_T + D_A)/\mu mg \sim 40$ nm, which validates our initial assumption of a colloidal motion that is mostly restricted at the capillary surface. Previous studies of bacterial swimming in the presence of confining walls have evidenced the possibility of complex swimming patterns that could possibly result in anisotropies in the bacterial bath [32,33]. Such an anisotropy, if present, should be reflected in a corresponding asymmetry in the motion of the colloidal tracers along x and y . However, we do not observe any systematic deviation between D_A and τ when fitted independently to the x and y components of MSD.

The MSD along x and y of the beads from simulations can be fitted with the same Eqs. (2) and (3) where this time D_A and τ are the only free parameters. As seen in Figs. 2(d) and 2(e), these functions fit very well the simulation data. By fitting the MSD along x with the free parameters D_A and τ we find that D_A grows continuously from 6.8 to 16.0 $\mu\text{m}^2/\text{s}$ as we increase the average speed from 15 to 30 $\mu\text{m}/\text{s}$. The parameter τ shows also a marked change upon changing the average speed going from $\tau = 0.44$ to 0.23 s.

The $\langle \Delta y^2(t) \rangle$ from simulations is fitted with Eq. (3) with the same free parameters D_A and τ giving D_A growing from 6.7 to 14.0 $\mu\text{m}^2/\text{s}$, that is almost identical to the one found from the fitting of $\langle \Delta x^2(t) \rangle$. Also the τ found from the fit of $\langle \Delta y^2(t) \rangle$ is very close to the one found from the MSD decreasing from ~ 0.4 to 0.2 s upon increasing the average swimming speed of bacteria. It has to be noted that simulations are in qualitative agreement with the experiments, although a quantitative comparison shows that both τ and D_A result in being considerably larger in simulations than in experiments.

It is, however, clear that the model of Eq. (1) can be used to fit both numerical and experimental curves and that this allows us to make a precise statement on how to generalize equipartition of energy for active particle systems in harmonic potentials. At equilibrium, when only thermal noise is present, the average potential energy of the particle $U = k\langle y^2 \rangle / 2$ is simply given by the equipartition theorem $U = D_T / 2\mu$. When active forces are introduced, they will add an extra contribution to U that can be obtained taking the limit of (3):

$$U = \frac{1}{2} k \lim_{t \rightarrow \infty} \langle \Delta y^2(t) \rangle / 2 = \frac{D_T}{2\mu} + \frac{D_A}{2\mu} \frac{1}{1 + \mu k \tau}. \quad (4)$$

It is worth noting that even within the very general premises of exponentially correlated noise, the expression for the average potential energy retains a form that is very close to the equilibrium equipartition result, with the only difference being that the contribution from the active noise is reduced by a factor $1 + \mu k \tau$. When the persistence time τ is much shorter than the relaxation time in the potential well $1/\mu k$, we recover the equilibrium form. This result is a direct consequence of the fact that when τ is much shorter than any other time scale in the problem, the active noise is practically white and Boltzmann statistics holds with the (unique) effective temperature $k_B T_{\text{eff}} = (D_A + D_T)/\mu$. However, if $\mu k \tau \sim 1$, even when the stationary distribution deviates strongly from the Boltzmann (as in run and tumble dynamics [11]), the average potential energy will be given by the simple formula (4).

We now discuss this generalized equipartition formula in both experiments and simulation. We plot the variance $\langle y^2 \rangle$ in Fig. 3. Figure 3 shows that the experimental $\langle y^2 \rangle$ is close to the straight line when plotted as a function of $(D_A + D_T)/\mu k$ indicating only weak deviation from the unique T_{eff} regime. This is consistent with the experimental $\mu k \tau \approx 0.028 (\ll 1)$. On the other hand, in simulations $\mu k \tau$ ranges approximately from 0.47 to 0.23. As shown in Fig. 3 we observe $\langle y^2 \rangle$ deviating considerably from the straight line when plotted as a function of $(D_A + D_T)/\mu k$, while when plotted as a function of Eq. (4) we see a substantial agreement with the straight line.

We remark that all these considerations are restricted to the second moment of fluctuations (i.e., $\langle y^2 \rangle$), and remain valid as long as the active noise is exponentially correlated, whatever is the static noise distribution. In our specific case, we have empirically found that the probability distribution $P(y)$ is well approximated by a Gaussian. This implies that $P(y) \sim \exp[-ky^2/2k_B T_{\text{eff}}]$ and all the static properties of the colloids in the active bath under the influence of the harmonic potential can be predicted by setting $k_B T_{\text{eff}} = D_A/[\mu(1 + \mu k \tau)] + D_T/\mu$. This effective temperature is, however, different from the effective temperature governing long time diffusion along the flat direction x which is given by $(D_A + D_T)/\mu$. In a way, the system

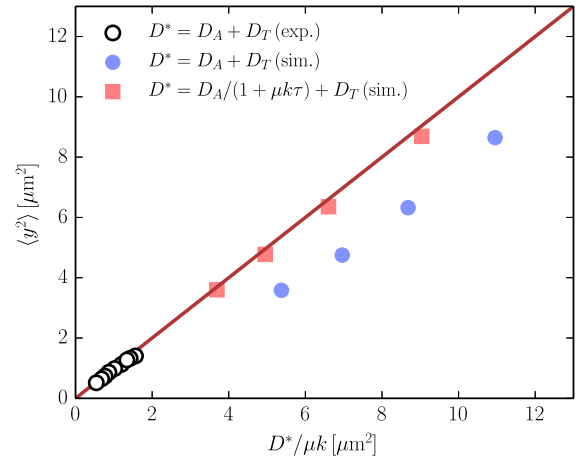


FIG. 3 (color online). Generalized equipartition plot. The variance of particle fluctuations along the y -coordinate is proportional to a weighted sum of thermal and active diffusivities, Eq. (4).

behaves like an equilibrium system whose free diffusivity and potential energy are governed by two different effective temperatures, the latter being a function of the curvature k of the external potential.

Conclusions.—We have investigated, experimentally and numerically, the possibility of generalizing energy equipartition to out-of-equilibrium systems consisting of colloidal particles that are subject to both a harmonic potential and the interactions with a bath of swimming bacteria. We found that the system obeys a modified energy equipartition law. A harmonic degree of freedom contributes an average potential energy that takes the equilibrium form for small curvatures and decreases when the relaxation time in the harmonic well starts to be comparable to the persistence time of active forces. Based on these observations we expect that using a different type of self-propelled colloids, i.e., Janus particles, one could have direct experimental access to higher τ values and observe the predicted strong deviations from equilibrium equipartition.

The research leading to these results has received funding from the European Research Council under the European Union's Seventh Framework Programme (FP7/2007-2013)/ERC Grant Agreement No. 307940. We also acknowledge funding from the Italian Ministry of Education, University and Research (MIUR) under the Basic Research Investment Fund program (FIRB Grant No. RBFR08WDBE).

*claudio.maggi@roma1.infn.it

- [1] K. Huang, *Statistical Mechanics* (John Wiley, New York, 1987).
- [2] B. Lukić, S. Jeney, C. Tischer, A. Kulik, L. Forró, and E.-L. Florin, *Phys. Rev. Lett.* **95**, 160601 (2005).

- [3] H. Risken, *The Fokker-Planck Equation: Methods of Solution and Applications* (Springer, Berlin, 1984).
- [4] K. To, *Phys. Rev. E* **89**, 062111 (2014).
- [5] L. Conti, P. De Gregorio, and G. Karapetyan, *J. Stat. Mech.* (2013) P12003.
- [6] W. K. Poon, in *Physics of Complex Colloids, Proceedings of the International School of Physics "Enrico Fermi," Course CLXXXIV*, edited by C. Bechinger, F. Sciortino, and P. Zihlerl (IOS, Amsterdam, 2012).
- [7] M. E. Cates, *Rep. Prog. Phys.* **75**, 042601 (2012).
- [8] X. L. Wu and A. Libchaber, *Phys. Rev. Lett.* **84**, 3017 (2000).
- [9] J. Palacci, C. Cottin-Bizonne, C. Ybert, and L. Bocquet, *Phys. Rev. Lett.* **105**, 088304 (2010).
- [10] C. Maggi, A. Lepore, and J. Solari, *Soft Matter* **9**, 10 885 (2013).
- [11] J. Tailleur and M. E. Cates, *Europhys. Lett.* **86**, 60002 (2009).
- [12] M. E. Cates, *Rep. Prog. Phys.* **75**, 042601 (2012).
- [13] L. Angelani, R. Di Leonardo, and G. Ruocco, *Phys. Rev. Lett.* **102**, 048104 (2009).
- [14] R. Di Leonardo, L. Angelani, D. Dell'Arciprete, G. Ruocco, V. Iebba, S. Schippa, M. P. Conte, F. Mecarini, F. De Angelis, and E. Di Fabrizio, *Proc. Natl. Acad. Sci. U.S.A.* **107**, 9541 (2010); A. Sokolov, M. M. Apodaca, B. A. Grzybowski, and I. S. Aranson, *Proc. Natl. Acad. Sci. U.S.A.* **107**, 969 (2010).
- [15] L. Angelani and R. Di Leonardo, *New J. Phys.* **12**, 113017 (2010).
- [16] P. Galajda, J. Keymer, P. Chaikin, and R. Austin, *J. Bacteriol.* **189**, 8704 (2007).
- [17] N. Koumakis, A. Lepore, C. Maggi, and R. Di Leonardo, *Nat. Commun.* **4**, 2588 (2013).
- [18] N. Koumakis, C. Maggi, and R. Di Leonardo, *Soft Matter* **10**, 5695 (2014).
- [19] L. Angelani, C. Maggi, M. L. Bernardini, A. Rizzo, and R. Di Leonardo, *Phys. Rev. Lett.* **107**, 138302 (2011).
- [20] I. Buttinoni, J. Bialké, F. Kümmel, H. Löwen, C. Bechinger, and T. Speck, *Phys. Rev. Lett.* **110**, 238301 (2013).
- [21] See Supplemental Material at <http://link.aps.org/supplemental/10.1103/PhysRevLett.113.238303> for details about bacteria preparation, numerical simulations and control measurements without bacteria, which includes Refs. [22–28].
- [22] M. T. Madigan *et al.*, *Brock Biology of Microorganisms*, 13th ed. (Benjamin-Cummings Publishing Company, San Francisco, 2010).
- [23] H. C. Berg, *E. Coli in Motion* (Springer-Verlag, New York, 2004).
- [24] S. Chattopadhyay, R. Moldovan, C. Yeung, and X. L. Wu, *Proc. Natl. Acad. Sci. U.S.A.* **103**, 13 712 (2006).
- [25] N. C. Darnton, L. Turner, S. Rojevsky, and H. C. Berg, *J. Bacteriol.* **189**, 1756 (2007).
- [26] V. A. Martinez, R. Besseling, O. A. Croze, J. Tailleur, M. Reufer, J. Schwartz-Linek, L. G. Wilson, M. A. Bees, and W. C. K. Poon, *Biophys. J.* **103**, 1637 (2012).
- [27] S. Kim and S. Karrila, *Microhydrodynamics* (Dover, New York, 2005).
- [28] W. H. Press, W. T. Vetterling, S. A. Teukolsky, and B. P. Flannery, *Numerical Recipes in C*, 2nd ed. (Cambridge University Press, Cambridge, England, 1992).
- [29] C. Valeriani, M. Li, J. Novosel, J. Arlt, and D. Marenduzzo, *Soft Matter* **7**, 5228 (2011).
- [30] C. Douarche, A. Buguin, H. Salman, and A. Libchaber, *Phys. Rev. Lett.* **102**, 198101 (2009).
- [31] E. Schaffer, S. F. Norrelykke, and J. Howard, *Langmuir* **23**, 3654 (2007); J. Leach, H. Mushfique, S. Keen, R. Di Leonardo, G. Ruocco, J. M. Cooper, M. J. Padgett, *Phys. Rev. E* **79**, 026301 (2009).
- [32] S. van Teeffelen, U. Zimmermann, and H. Löwen, *Soft Matter* **5**, 4510 (2009).
- [33] P. K. Radtke and L. Schimansky-Geier, *Phys. Rev. E* **85**, 051110 (2012).