Checkerboard Self-Patterning of an Ionic Liquid Film on Mercury

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Å-resolution studies of room temperature ionic liquid (RTIL) interfaces are scarce, in spite of their long-recognized importance for the science and many applications of RTILs. We present an \AA -resolution x-ray study of a Langmuir film of an RTIL on mercury. At low (high) coverage [90 (50) \AA^2 /molecule] a mono-(bi)layer of surface-parallel molecules is found. The molecules self-assemble in a lateral ionic checkerboard pattern, unlike the uniform-charge, alternate-ion layers of this RTIL at its bulk-solid interface. A 2D-smectic order is found, with molecules packed in parallel stripes, forming long-range order normal to, but none along, the stripes.

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Room temperature ionic liquids (RTILs) are a novel class of liquids, consisting solely of ions without a solvent. The bulky and irregular shape of the ions, often comprising hydro- and fluorocarbon chains and rings, inhibits solidification even at room temperature in spite of the strong Coulomb interaction [[1\]](#page-3-1). Their favorable chemical properties, such as [[1](#page-3-1)[,2\]](#page-3-2) nonvolatility, wide electrochemical window, high ion mobility, etc., renders them outstanding "green" replacements for volatile, corrosive and polluting solvents, reaction media, and working fluids in the chemical industries [[1](#page-3-1)], and in energy- and electronics-related applications in batteries, solar and fuel cells [\[3\]](#page-3-3), and microelectromechanical systems [\[4](#page-3-4)].

The intense scientific interest in RTILs stems from their complex combination of interactions, van der Waals, Coulomb, dipole, and hydrogen bonding, rarely occurring simultaneously in other materials [[5](#page-3-5),[6\]](#page-3-6). Being infiniteconcentration ionic ''solutions,'' RTILs are strongly correlated systems, going beyond the validity regime of the Poisson-Boltzman equation, and the Gouy-Chapman and Stern models of the electric double layer [[7\]](#page-3-7), and thus allowing insight into this scarcely studied regime. Theoretical progress in understanding the interplay of interactions in, and of the strong-correlation aspects of, RTILs is hampered by the scarcity of molecular-resolution structural data, particularly at interfaces, where such measurements are highly challenging. While a few such studies of liquid-vapor (l-v) RTIL interfaces are available [[8](#page-3-8),[9\]](#page-3-9), only two address liquid-solid (l-s) interfaces [[10](#page-3-10),[11](#page-3-11)]. No bulk liquid-liquid (l-l) interface studies were published. Even these handful of studies address only interfaces between bulk phases. No A-resolution structure of 2D Langmuir films (LFs) has been published to date for any RTIL, even though numerous studies of LFs on water [\[12–](#page-3-12)[14\]](#page-3-13) and mercury [[15](#page-3-14)[–17\]](#page-3-15) of non-RTIL compounds demonstrate the deep insights obtainable from such measurements. In particular, LF x-ray measurements can determine the surface-parallel structure and its coverage dependence. These are masked in l-s and l-l interface studies by the strong bulk scattering.

We present here an A-resolution x-ray study of the structure of a liquid-mercury supported RTIL LF, 1-butyl-1-methylpyrrolidinium tris(pentafluoroethyl) trifluorophosphate $([BMPL]^+[FAP]^-)$. The surface-parallel molecules self-organize into stripes, which exhibit onedimensional, smecticlike order, implying a checkerboardlike ionic surface tiling. This is consistent with the expected domination of the lateral order by the strong ionic Coulomb interaction, but differs from the ordering found for the bulk of the same RTIL at a charged solid interface, where alternately charged ion layering was observed [[10\]](#page-3-10).

The energy-optimized calculated $[BMPL]^+$ and $[FAP]^+$ ion configurations (Fig. [1\)](#page-0-0) yield areas of 36 and 51 \AA^2 [[18\]](#page-3-16). The measured surface pressure (π) /molecular area(A) isotherm $[15,16]$ $[15,16]$ (for method see $[19]$) is shown in Fig. [2.](#page-1-0) The steep rise at \sim 90 Å²/molecule marks the point where the molecules of the $A \gg 90 \text{ Å}^2/\text{molecule } 2D$ gas start touching, forming a densely packed monolayer [\[13,](#page-3-19)[15\]](#page-3-14). Fits (dashed lines) by the Volmer [\[20\]](#page-3-20) and van der Waals (vdW) [\[21\]](#page-3-21) state equations of an ideal 2D gas of finite-area (A_0) molecules yield exclusion areas $A_0^V = 93 \text{ Å}^2/\text{molecule}$ and $A_0^{vdW} = 88 \text{ Å}^2/\text{molecule}$,

FIG. 1 (color online). Configurations and dimensions of the ions.

FIG. 2 (color online). Measured (symbols and eye-guiding solid line) and fitted (dashed lines) π -A isotherm of the mercury-supported RTIL LF, spread from an acetonitrile solution. Inset: Schematic of the ions' arrangement on the surface.

respectively. The close coincide of the measured A_0 with the sum of areas of the two ions, 87 \AA^2 /molecule, suggests that the ions comprising the molecule lie flat, side by side, on the mercury surface (Fig. [2,](#page-1-0) inset), not on top of each other, thus maximizing the vdW contacts of the molecule with the surface. Surface-parallel ion orientations are also found in simulations of similar RTILs, albeit on graphite [\[22\]](#page-3-22). This ionic arrangement has a zero net local and global charge, and a lower free energy than that of the alternating uniformly charged layers found at the same RTIL's bulk interface with a charged sapphire substrate [\[10\]](#page-3-10). The average of the fitted exclusion areas, $\bar{A}_0 = 90.5 \text{ Å}^2/\text{molecule}$, and the molar volume, 0.370 l/mol, yields a monolayer thickness $\bar{d}_0 = 6.8$ Å, supported by the x-ray results below.

Two additional isotherm features should be noted. First, the linear sloping plateau at $A < A_0$ resembles that observed at the same A range for a mercury-supported stearic acid LF [\[15\]](#page-3-14), which was shown to correspond to a continuous conversion with decreasing A of a monolayer to a bilayer of surface-parallel molecules. Second, no sharp rise, the sign of a transition from lying-down to standing-up molecules, is observed near film collapse at $A \le 50 \text{ Å}^2/\text{molecule}$, in contrast with mercury-supported LFs of fatty acids, thiols, and alcohols [\[15,](#page-3-14)[16](#page-3-17)], which show such transitions. These isotherm-derived inferences are confirmed by our x-ray measurements, discussed next.

Since RTILs resist ordering, as their low melting temperatures demonstrate, our x-ray measurements (for meth-ods see [\[19\]](#page-3-18)) were done at the highest π below film collapse, at $A \approx \bar{A}_0/2 = 50 \text{ Å}^2/\text{molecule}$, where order is most likely to emerge. The x-ray reflectivity (XR) , $R(q_z)$, was measured at beamline X22B, NSLS, Brookhaven National Laboratory, USA, as a function of the incoming beam's grazing angle of incidence onto the surface, α , where $q_z = (4\pi/\lambda)\sin\alpha$, and $\lambda = 1.5127 \text{ Å}$ is the x rays' wavelength [\[23](#page-3-23)[,24\]](#page-3-24). The interface's structural information resides in the deviations of $R(q_z)$ from $R_F(q_z)$, the theoretical Fresnel XR of an ideally smooth and abrupt interface. Figure [3\(a\)](#page-1-1) shows the measured (symbols) and box model fitted (line) $R(q_z)/R_F(q_z)$. The model, successfully used for numerous LFs of organic molecules on mercury [[15](#page-3-14),[16](#page-3-17),[25](#page-3-25)], uses 6 slabs to mimic the decaying layering of mercury at the interface [[26](#page-3-26)–[28](#page-3-27)]. As the measurements were done at 50 \AA^2 /molecule, half the area of a flat-lying molecule, a bilayer was expected, as discussed above. To reproduce well [Fig. $3(a)$, line] the complex $R(q_z)/R_F(q_z)$ measured [Fig. [3\(a\)](#page-1-1), symbols] three slabs were necessary to describe the RTIL layer covering the mercury surface: two identical slabs of fixed nominal density $\rho = 0.485 e/\text{\AA}^3$ (calculated from the molecular volume and number of electrons) and thickness, 6.8 Å, separated by a low-density layer \sim 1 Å thick and $\rho = 0.27 \frac{e}{\text{A}^3}$ dense. The best-fit surface-normal electron density profile, $\rho(z)$, is shown in Fig. [3\(b\)](#page-1-1) with (solid line) and without (dashed line) the smearing by roughness due to

FIG. 3 (color online). (a) Fresnel-normalized measured (symbols) and model fitted (line) XR curve. (b) Fit-derived surfacenormal electron density profile, $\rho(z)$, with (solid line) and without (dashed line) capillary wave roughness smearing. (c) π -A isotherm. The arrow marks the measurements' $A = 50 \text{ Å}^2/\text{molecule.}$

the thermally excited capillary waves decorating all liquid interfaces [[29](#page-3-28),[30](#page-3-29)]. The good fit in Fig. [3\(a\)](#page-1-1) supports, therefore, the two-layer structure of flat-lying molecules inferred above from the isotherm.

To elucidate the surface-parallel order within the layer, grazing incidence diffraction (GID) measurements [\[23,](#page-3-23)[24\]](#page-3-24) were done at 50 \AA^2 /molecule. Here a small incidence angle α is used and the detector is scanned out of the reflection plane by an angle 2θ , to yield a finite surfaceparallel diffraction vector $q_{\parallel} \approx (4\pi/\lambda) \sin\theta$. An extensive search revealed only a single GID peak [Fig. [4\(a\)\]](#page-2-0) at $q_{\parallel}^0 = 0.742 \text{ Å}^{-1}$, which corresponds to a repeat distance of $d = 2\pi/q_{\parallel}^0 = 8.47 \text{ Å}$, suggestively close to the 8.5–9 Å lengths of both $[BMPL]^+$ and $[FAP]^-(Fig. 1)$ $[FAP]^-(Fig. 1)$. A single GID peak implies either a hexagonal [\[13](#page-3-19)[,24,](#page-3-24)[31\]](#page-3-30) or a onedimensional [[15](#page-3-14)] lateral order. The former occurs only rarely in 2D packing of molecules of noncircular surfaceparallel cross sections, such as our RTIL. Moreover, a hexagonal packing with our q_{\parallel}^0 implies an area of mexagonar packing with our q_{\parallel} implies an area of $4\sqrt{3}\pi^2/(q_{\parallel}^0)^2 = 124 \text{ Å}^2/\text{molecule}, 40\%$ larger than the

FIG. 4 (color online). (a) Measured (symbols) and Lorentzianplus-linear background fitted (line) GID peak at $A =$ 50 \AA^2 /molecule. Inset: Schematic top view of the ionic order, discussed in the text. (b) Measured BR (symbols) and theoretical fit (line) at the GID peak. (c) Contour plot of the measured BR.

 $A_0 \approx 90 \text{ Å}^2/\text{molecule observed in the isotherm, a value}$ supported strongly also by the density needed to fit the measured $R(q_z)/R_F(q_z)$. Thus, a lateral hexagonal order is highly unlikely in our LF.

The second possibility, a one-dimensional order, is consistent with the molecular area of \sim 90 Å²/molecule inferred from the isotherm and the XR results. This packing motif is shown schematically in the inset of Fig. [4\(a\)](#page-2-0): alternately charged ions assemble (without positional order) into long stripes. The extended stripes (longitudinally shifted relative to each other, see below) lie side by side, forming long-range order in the stripe-normal direction. q_{\parallel}^0 corresponds to the positional order of the stripes, with a repeat distance equal to the stripe's width, \sim 8.5 Å, close to the ions' length. The absence of long-range positional order along the strip, is explainable by small random variations in the azimuthal molecular orientations along the stripe, due to the irregular molecular shape. A very similar ordering was found for LFs of intermediate-length $($ \sim 18 carbons) fatty acid molecules on mercury, in their lying-down phase [[15](#page-3-14)]. There, linear carboxyl-bound dimers aggregate to form stripes, with dimers lying stripe normal at random positions along the stripe. The parallellying stripes exhibit stripe-normal positional order with a repeat distance equal to the dimers' length, 52.3 Å for octadecanoic acid [[15](#page-3-14)]. Note however that for fatty acids the structure-dominating interdimer interaction is purely dispersive, unlike here, where the interionic Coulomb interaction dominates the structure. Also, the requirement of minimizing the overall Coulomb energy induces a roughly half-molecule longitudinal shift between adjacent stripes to create the lowest-Coulomb-energy checkerboardlike motif, albeit with long-range order in one direction only. A recent study of another RTIL, $[TOMA]^+[C_4C_4N]^-,$ reports also a checkerboardlike ion arrangement at the bulk l-v interface, without, however, any indication for a lateral long-range order in any direction [\[9](#page-3-9)].

The crystalline coherence length, ξ , is obtained from the Debye-Scherrer formula, $\xi = 0.9 \times 2\pi/(\Delta^2 q_{\parallel} - \Delta^2_{\text{res}})^{1/2}$, where $\Delta q_{\parallel} = 0.0274 \text{ Å}^{-1}$ is the GID peak's full width at half maximum, and $\Delta_{\text{res}} = 0.008 \text{ Å}^{-1}$ is the diffractometer's resolution. The poor crystallinity obtained, $\xi \approx$ 200 A only, is in line with the RTIL's dominant property of resisting crystallization. The poor crystallinity is also reflected in the low intensity of the GID peak and of the Bragg rod (BR), discussed next.

The BR [\[24\]](#page-3-24), the surface-normal q_z distribution of the diffracted intensity at q_{\parallel}^0 , provides further information on the ordered molecules. It is shown in Figs. $4(b)$ and $4(c)$. The intensity contour plot in Fig. [4\(c\)](#page-2-0) is typical of a surface layer comprising laterally ordered untilted molecules. The measured BR [Fig. [4\(b\)](#page-2-0), symbols] exhibits a surface enhancement peak at 0.055 \AA^{-1} , indicating that the GID peak originates in a surface, rather than a bulk, structure.

A model (solid line) assuming a *monolayer* of 6.8 A high molecules untilted from the surface normal is consistent with the measured curve. As both XR and the isotherm show the layer at this coverage to be a bilayer, the BR result indicates either that while both layers of the film are ordered, the crystallites in one monolayer are out of registration with those of the other, or that one of the monolayers is disordered and the GID peak originates in a single monolayer only. The present data do not allow distinguishing between these two possibilities. We note however, that simulations, albeit for a different RTIL $([BMIM]^+[PF_6]^-)$ and a different substrate (ordered solid graphite) [\[22](#page-3-22)[,32\]](#page-3-31) indicate that for an RTIL multilayer on graphite the interface-adjacent RTIL layer is laterally more ordered than (the two) subsequent ones observed, which somewhat supports the first possibility. On the other hand, the Coulomb repulsion between like charges places a high energetic penalty on the registration of crystallites in the upper and lower layer. The full elimination of this penalty is achievable by a lateral shift of the upper and lower layers to yield a perfect oppositely charged vertical ion registration. This, however, is achievable only with fully ordered layers, rather than the 1D order found here. Thus, in our case, the frustration due to the interlayer Coulomb interaction would destroy the (shifted) registration and the BR would reflect a single-layer order only, even when both layers are ordered. This frustration may also account, at least partially, for the small ξ and low intensity of the GID peak and BR. In the corresponding bilayer phase of medium-length fatty acids, discussed above, where the strong Coulomb interaction is absent, a registration of the layers does occur, and is reflected in a twice shorter BR [\[15\]](#page-3-14).

In conclusion, we have demonstrated here that x rays can resolve the \AA -scale structure of an RTIL LF in both the lateral and longitudinal directions. The structure found here, a checkerboard pattern ordered in one direction only, has similarities with, but also differences from, previously studied structures of mercury-supported fatty acid LFs. In those LFs full 2D lateral order eventually emerges upon increasing the length of the alkyl moiety. Whether a similar order emerges in our RTIL upon increasing its alkyl tail length, in spite of the irregular ionic shapes and the strong Coulomb interaction, remains to be investigated. The dichotomy between the electrically neutral monoand bilayers observed here and the alternating, uniformly charged, (multi)layers formed at the charged sapphire interface in contact with a bulk of the same RTIL [[10](#page-3-10)] also deserves further investigation, to elucidate the role of the overlying RTIL bulk and the level of substrate charging in the structure formation and characteristics. Above all, a greater body of experiments on LFs of other RTILs, over a broad size and molecular architecture range, is required to broaden and deepen the insights, gleaned from this first study, on the interactions dominating the structure of RTILs at interfaces.

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- [1] Ionic Liquids in Synthesis, edited by H. Wasserscheid and T. Welton (Wiley-VCH, Weinheim, Germany, 2007), 2nd ed..
- [2] R.D. Rogers and K.R. Seddon, Science 302[, 792 \(2003\).](http://dx.doi.org/10.1126/science.1090313)
- [3] S. Y. Lee et al., [J. Am. Chem. Soc.](http://dx.doi.org/10.1021/ja102367x) 132, 9764 (2010).
- [4] I.B. Malham et al., Soft Matter 6[, 4062 \(2010\).](http://dx.doi.org/10.1039/c0sm00377h)
- [5] L. Crowhurst et al., [J. Am. Chem. Soc.](http://dx.doi.org/10.1021/ja046757y) **126**, 11 549 (2004).
- [6] H. Weingärtner, [Angew. Chem., Int. Ed.](http://dx.doi.org/10.1002/anie.200604951) 47, 654 (2008).
- [7] M. Z. Bazant et al., Phys. Rev. Lett. **106**[, 046102 \(2011\).](http://dx.doi.org/10.1103/PhysRevLett.106.046102)
- [8] E. Sloutskin et al., [J. Am. Chem. Soc.](http://dx.doi.org/10.1021/ja0509679) 127, 7796 (2005).
- [9] N. Nishi et al., J. Chem. Phys. **132**[, 164705 \(2010\).](http://dx.doi.org/10.1063/1.3398029)
- [10] M. Mezger et al., Science 322[, 424 \(2008\)](http://dx.doi.org/10.1126/science.1164502).
- [11] M. Mezger et al., J. Chem. Phys. **131**[, 094701 \(2009\).](http://dx.doi.org/10.1063/1.3212613)
- [12] S. G. Wolf et al., [Nature \(London\)](http://dx.doi.org/10.1038/328063a0) 328, 63 (1987).
- [13] V. M. Kaganer et al., [Rev. Mod. Phys.](http://dx.doi.org/10.1103/RevModPhys.71.779) **71**, 779 (1999).
- [14] E. Sloutskin et al., Phys. Rev. Lett. 99[, 136102 \(2007\).](http://dx.doi.org/10.1103/PhysRevLett.99.136102)
- [15] H. Kraack et al., Science 298[, 1404 \(2002\).](http://dx.doi.org/10.1126/science.1078372)
- [16] B. M. Ocko et al., Phys. Rev. Lett. 94[, 017802 \(2005\)](http://dx.doi.org/10.1103/PhysRevLett.94.017802).
- [17] L. Tamam et al., Langmuir 25[, 5111 \(2009\)](http://dx.doi.org/10.1021/la804109h).
- [18] Files Pyr14.pdb & (C2F5)3PF3.pdb, [www.il-eco.uft](www.il-eco.uft.unibremen.de) [.unibremen.de](www.il-eco.uft.unibremen.de).
- [19] See supplemental material at [http://link.aps.org/](http://link.aps.org/supplemental/10.1103/PhysRevLett.106.197801) [supplemental/10.1103/PhysRevLett.106.197801](http://link.aps.org/supplemental/10.1103/PhysRevLett.106.197801) for experimental details.
- [20] M. Volmer, Z. Phys. Chem. (Leipzig) 115, 253 (1925).
- [21] J. Israelachvili, Intermolecular and Surface Forces (Academic Press, London, UK, 1992), 2nd ed..
- [22] S. A. Kislenko et al., [Phys. Chem. Chem. Phys.](http://dx.doi.org/10.1039/b823189c) 11, 5584 [\(2009\)](http://dx.doi.org/10.1039/b823189c).
- [23] J. Als-Nielsen and D. McMorrow, Elements of Modern X-ray Physics (Wiely, New York, USA, 2001).
- [24] B. M. Ocko et al., Phys. Rev. E 55[, 3164 \(1997\)](http://dx.doi.org/10.1103/PhysRevE.55.3164).
- [25] L. Tamam et al., Soft Matter 6[, 526 \(2010\)](http://dx.doi.org/10.1039/b914714d).
- [26] O.M. Magnussen et al., [Phys. Rev. Lett.](http://dx.doi.org/10.1103/PhysRevLett.74.4444) 74, 4444 [\(1995\)](http://dx.doi.org/10.1103/PhysRevLett.74.4444).
- [27] A. Elsen et al., Phys. Rev. Lett. **104**[, 105501 \(2010\).](http://dx.doi.org/10.1103/PhysRevLett.104.105501)
- [28] L. Tamam et al., [J. Phys. Chem. Lett.](http://dx.doi.org/10.1021/jz1000209) 1, 1041 (2010).
- [29] B. M. Ocko et al., [Phys. Rev. Lett.](http://dx.doi.org/10.1103/PhysRevLett.72.242) **72**, 242 (1994).
- [30] P. S. Pershan, [J. Phys. Chem. B](http://dx.doi.org/10.1021/jp806113n) 113, 3639 (2009).
- [31] X.Z. Wu et al., [Phys. Rev. Lett.](http://dx.doi.org/10.1103/PhysRevLett.70.958) **70**, 958 (1993).
- [32] M. Sha et al., J. Chem. Phys. **128**[, 134504 \(2008\)](http://dx.doi.org/10.1063/1.2898497).