## Metal-Insulator Transition and Orbital Order in PbRuO<sub>3</sub>

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Anomalous low temperature electronic and structural behavior has been discovered in PbRuO<sub>3</sub>. The structure [space group Pnma, a=5.563 14(1), b=7.864 68(1), c=5.614 30(1) Å] and metallic conductivity at 290 K are similar to those of SrRuO<sub>3</sub> and other ruthenate perovskites, but a sharp metalinsulator transition at which the resistivity increases by 4 orders of magnitude is discovered at 90 K. This is accompanied by a first-order structural transition to an Imma phase [a=5.569 62(1), b=7.745 50(1), c=5.662 08(1) Å at 25 K] that shows a coupling of Ru<sup>4+</sup> 4d orbital order to distortions from Pb<sup>2+</sup> 6s6p orbital hybridization. The Pnma to Imma transition is an unconventional reversal of the group-subgroup symmetry relationship. No long range magnetic order is evident down to 1.5 K. Calculations show that Pb 6s6p and Ru 4d orbital hybridization and strong spin-orbit coupling are significant.

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Transition metal oxide perovskites display a remarkable range of electronic phenomena such as superconductivity, colossal magnetoresistances, and coupled charge, orbital, and spin orderings [1]. Perovskite-related ruthenates based on Ru<sup>4+</sup> have proved interesting as broad Ru: 4d bands lead to metallicity without chemical doping, so very clean correlated itinerant electron physics may be observed in single crystals. The layered perovskite Sr<sub>2</sub>RuO<sub>4</sub> is a p-wave superconductor [2], while bilayered Sr<sub>3</sub>Ru<sub>2</sub>O<sub>7</sub> is a metallic metamagnet [3]. All of the cubic-type ARuO<sub>3</sub> perovskites (A = Ca, Sr, Ba) are metallic to the lowest measured temperatures. SrRuO3 and BaRuO3 are ferromagnets with Curie temperatures of  $T_C = 160$  and 60 K, respectively [4–7], but CaRuO<sub>3</sub> does not show a magnetic transition [8]. Another ruthenate perovskite, PbRuO<sub>3</sub>, was synthesized at high pressures in 1970 [9], but the electronic properties of this material have not been reported. Pb<sup>2+</sup> is intermediate in size between Sr<sup>2+</sup> and Ba<sup>2+</sup>, so PbRuO<sub>3</sub> is expected to be a ferromagnetic metal. However, in this Letter, we report that PbRuO<sub>3</sub> behaves very differently from the other ARuO<sub>3</sub> perovskites and instead shows a sharp metal-insulator transition at 90 K but without apparent long range spin order. Electron localization is strongly coupled to the lattice through Ru-orbital ordering, resulting in an anomalous structural change to higher lattice symmetry at low temperatures. We propose that strong hybridization of Ru 4d with Pb 6s and 6p states and spin-orbit coupling are responsible for inducing this alternative ground state for ruthenate perovskites.

Small (ca. 10 mg) polycrystalline pellets of PbRuO<sub>3</sub> were synthesized by heating the oxygen-deficient pyrochlore Pb<sub>2</sub>Ru<sub>2</sub>O<sub>6.5</sub> at 11 GPa and 1100 °C using a

Walker-type multianvil press. Synchrotron x-ray diffraction profiles in the temperature range 10 < T < 300 K were collected from instrument ID31 at the ESRF, France, with wavelength  $\lambda = 0.45\,621$  Å, and time-of-flight neutron powder data were recorded using the GEM spectrometer at ISIS, United Kingdom, over the range 1.5 < T < 300 K. Rietveld fits to diffraction data were performed using the GSAS package [10]. Magnetic susceptibility and electronic resistivity measurements were made using Quantum Design MPMS and PPMS (Magnetic and Physical Properties Measurement System) instruments.

The (hkl) Bragg reflections observed in the room temperature x-ray and neutron powder diffraction profiles of PbRuO<sub>3</sub> are all consistent with the orthorhombic Pnma space group of SrRuO<sub>3</sub> [11], and refinement of this model gave good fits to the data. We also attempted refinements of noncentrosymmetric variants of this structure, in case of a steric ("lone pair") effect from the Pb 6s<sup>2</sup> state, but no improvements were obtained and the fits diverged. All of the site occupancies refined to within error (ca. 1%) of full occupancy, showing that the sample is stoichiometric [12]. The Pnma superstructure is often observed in oxide perovskites and results from a generic, oxygen-centered, tilting instability rather than specific, transition metal-centered electronic instabilities.

To compare the contributions of Pb<sup>2+</sup> and Sr<sup>2+</sup> to the ruthenate band structures, we performed electronic structure calculations for room temperature *Pnma*-type PbRuO<sub>3</sub> and SrRuO<sub>3</sub> [13] in the local density approximation (LDA) using the full-potential augmented plane wave plus local orbital method [14]. The results (Fig. 1) for SrRuO<sub>3</sub> are very similar to those previously reported [5].

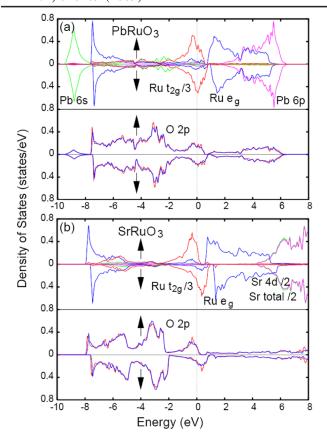


FIG. 1 (color online). Density of states (DOS) of (a) PbRuO<sub>3</sub> and (b) SrRuO<sub>3</sub> in the *Pnma* structure calculated by the LDA in the ferromagnetic state. Up (down) arrows refers to spin up (down). The Ru  $t_{2g}$  DOS is scaled by  $\frac{1}{3}$ , and the Sr DOS is scaled by  $\frac{1}{2}$ . The exchange splitting of the Ru  $t_{2g}$ -O 2p bands is significantly suppressed in PbRuO<sub>3</sub>.

The  $Sr^{2+}$  states lie far from the Ru 4d-O 2p bands near the Fermi level [6], and a large exchange splitting is found. The itinerant ferromagnetic state is stabilized by 25 meV/f.u., in keeping with the 160 K Curie transition. By contrast, the Ru 4d and O 2p states in PbRuO<sub>3</sub> lie just between the occupied Pb 6s and unoccupied Pb 6p states. The Ru  $t_{2g}$ -Pb 6s and Ru  $e_g$ -Pb 6p hybridizations, both aided by the O 2p state, and  $t_{2g}$ - $e_g$  mixing due to the lattice distortion, significantly suppress the exchange splitting of the Ru  $t_{2g}$ -O 2p conduction bands, reducing the magnetic stabilization energy to near zero (<4 meV/f.u.). To confirm that the slight lattice differences between the two materials are unimportant, we also calculated the electronic structure of SrRuO<sub>3</sub> using the PbRuO<sub>3</sub> parameters and found that the electronic and magnetic properties were virtually identical.

An unexpected difference between PbRuO<sub>3</sub> and the other perovskite ruthenates was discovered by resistivity measurements (Fig. 2). At ambient temperatures, the PbRuO<sub>3</sub> has a resistivity of  $\sim 10^{-1}~\Omega$  cm with little temperature dependence, characteristic of metallic conduction with a resistive grain boundary contribution. However, on cooling, the resistivity increases sharply by 4 orders of

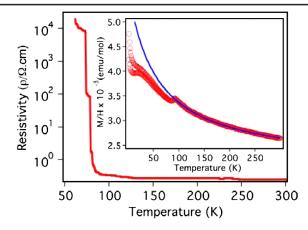


FIG. 2 (color online). Resistivity variation for PbRuO<sub>3</sub>; the inset shows magnetic susceptibility measured in a 1 T field with a modified Curie-Weiss fit as described in the text.

magnitude (and was immeasurably large below 60 K), signifying a metal-insulator transition at  $T_{\rm MI} = 90$  K. This transition is also evident in magnetization data. The high temperature susceptibility is fitted as  $\chi = C/(T - \theta) + \chi_p$ , a sum of Curie-Weiss and Pauli paramagnetic terms, respectively, with  $\chi_p = 2.09(1) \times 10^{-3}$  emu/mol, Curie constant C = 0.195 emu K/mol (corresponding to a paramagnetic moment of  $1.25\mu_B$ ), and a Weiss temperature  $\theta =$ -54 K. Similar large temperature-independent contributions have been reported for nonperovskite ruthenates such as the pyrochlore  $Tl_2Ru_2O_7$  ( $\chi_p \approx 2 \times 10^{-3}$  emu/mol) [15] close to metal-insulator instabilities. A small dip is observed in the magnetic susceptibility on cooling through  $T_{\rm MI}$ , showing that antiferromagnetic correlations are present in the insulating state. However, no long range magnetic transition is evident down to 4 K, although a broad hump with divergence between field- and zero-fieldcooled susceptibilities is observed below 50 K.

A strong coupling of structure to the metal-insulator transition is observed in both x-ray and neutron powder diffraction measurements (Fig. 3). PbRuO<sub>3</sub> remains orthorhombic down to 1.5 K; however, the lattice parameters and volume change discontinuously on cooling through the transition (Fig. 4), and, surprisingly, the *Pnma* superstructure reflections with odd (h + k + l) values disappear (see Fig. 3 inset). No new superstructure reflections, peak broadenings, or splittings were observed in the low temperature diffraction patterns of PbRuO<sub>3</sub>, which are indexed by the body-centered space group Imma. This describes another common tilting superstructure of perovskites, and a refined *Imma* model gives excellent fits to both x-ray and neutron data [12]. Fits of lower-symmetry, acentric bodycentered structures were unsuccessful. No magnetic diffraction peaks were observed down to 1.5 K in the GEM time-of-flight neutron diffraction data or in subsequent constant wavelength profiles collected from instrument E6 at the HZB reactor. We estimate the upper limit for any ordered Ru moment to be  $\sim 0.5 \mu_B$ .

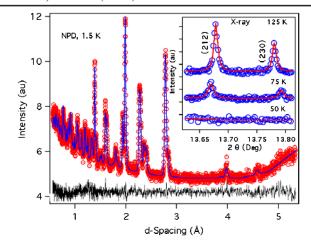


FIG. 3 (color online). Fit of the Imma model to the time-of-flight neutron diffraction profile of PbRuO<sub>3</sub> at 1.5 K. The inset shows the disappearance of Pnma x-ray superstructure reflections as the Imma phase is formed on cooling.

The (high temperature) *Pnma* to (low temperature) *Imma* transition in PbRuO<sub>3</sub> is remarkable as *Pnma* is a subgroup of *Imma*, so a continuous group-subgroup transition from *Imma* to *Pnma* is allowed in Landau theory and is observed in many simple perovskites such as SrSnO<sub>3</sub> [16]. The *Pnma-Imma* "subgroup-group" transition in PbRuO<sub>3</sub> is clearly first-order, with a small volume anomaly typical of metal-insulator transitions and a substantial hysteresis of 20 K in the cell parameters between warming and cooling experiments (Fig. 4). The subgroup-group structural contribution to the transition entropy is negative, but this is evidently outweighed by the large positive electronic contribution from the delocalization of Ru 4*d* electrons.

The evolution of the Ru-O bond distances (Fig. 4) reveals an important aspect of the metal-insulator transition. At room temperature, the RuO<sub>6</sub> octahedra are almost regular with Ru-O bond lengths of 2.00–2.01 Å, but, below  $T_{\rm MI}$ , a Jahn-Teller distortion is apparent in the *Imma* structure, with two short Ru-O1 bonds (1.97 Å) aligned approximately along z and four long Ru-O2 bonds (2.02 Å) in the xy plane. To a first approximation, this corresponds to a  $d_{xy}^2 d_{xz}^1 d_{yz}^1$  orbital ordering of the Ru<sup>4+</sup> $t_{2g}^4$  configuration in the insulating Imma phase, creating planes of minorityspin-occupied  $d_{xy}$  orbitals, as shown in Fig. 4. Pb<sup>2+</sup> shows an unusual A-site distortion, having a near-regular square pyramidal coordination with five short Pb-O bonds (Pb-O1, 2.51 Å  $\times$  1; Pb-O2, 2.50 Å  $\times$  4), while other Pb-O distances are >2.82 Å. The O1-Ru-O2 angle of 122.5° shows that this is not a lone pair effect, for which an angle of <90° is expected. The Pb and Ru distortions are cooperative as O1 forms short bonds to Ru and only one short bond to Pb, whereas O2 has long bonds to Ru and four short bonds to Pb.

To clarify the orbitally ordered state, we have carried out LSDA + U calculations for the Imma phase with an effective Hubbard U = 3.5 eV [17]. Spin-orbit coupling (SOC) was also included since this  $\sim 160$  meV interaction

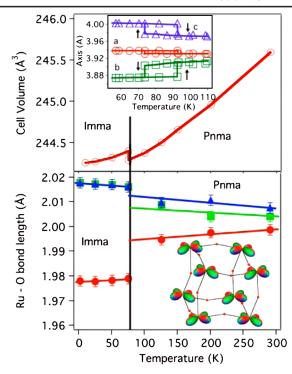


FIG. 4 (color online). Cell volume (top) and Ru-O distance (bottom) variations for PbRuO<sub>3</sub> from fits to neutron diffraction data. The upper inset shows hysteresis in the lattice parameters, normalized to the cubic perovskite cell, from x-ray measurements while warming or cooling at 10 K/min. Idealized  $d_{xy}$  orbital order in the Imma phase is shown in the lower panel.

is large relative to the calculated crystal field splitting between the  $d_{xy}$  and  $d_{xz}/d_{yz}$  levels of ~50 meV. Our LSDA + U + SOC calculations gave an insulating ground state with a small gap of ~0.1 eV, verifying that PbRuO<sub>3</sub> is in the vicinity of a metal-insulator transition. The minority spin electron has a  $0.46(1+i)d_{xy} - 0.38(1-i)(d_{xz}+d_{yz})$  orbital state which consists of 42%  $d_{xy}$ , 29%  $d_{xz}$ , and 29%  $d_{yz}$ , with  $d_{xy}$  being dominant as expected from the above structural results. Different magnetic solutions lie close in energy; e.g., the lowest G-type antiferromagnetic state is only 6.8 meV/f.u. below the ferromagnetic state, in keeping with the observed lack of magnetic order discussed below.

The insulating, orbitally ordered ground state of PbRuO<sub>3</sub> is anomalous in comparison to the other  $ARuO_3$  perovskites (A = Ca, Sr, Ba), which remain metallic to the lowest temperature. This is not due to a size effect as  $CaRuO_3$  has the most tilted Pnma superstructure but is stable to orbital order. (The combination of small  $Ca^{2+}$  and an imposed tetragonal symmetry in layered  $Ca_2RuO_4$  is sufficient to induce a weak orbital order [18], but this phase is a Mott insulator.) The electronic nature of Pb is a key factor, and Ru 4d-Pb 6s6p-O 2p hybridizations are evident in the above band structure calculations. This Pb<sup>2+</sup> covalency is sometimes manifest as a lone pair distortion resulting in ferroelectricity, e.g., in PbTiO<sub>3</sub>, but lone pair distortions are not observed in either the Pnma or the

Imma phases of PbRuO<sub>3</sub>. Another consequence of covalency is the stabilization of lower Pb<sup>2+</sup> coordination numbers than expected from cation size arguments [19] as shown by the change from three short (2.47 Å) Pb-O bonds in the Pnma structure of PbRuO<sub>3</sub> to five, described above, in the Imma phase. The metal-insulator transition in PbRuO<sub>3</sub> is thus driven by electronic instabilities of both cations as the orbital order of  $t_{2g}^4$ Ru<sup>4+</sup> is coupled to an order of  $t_{2g}^2$ Pb<sup>2+</sup> hybrid states. By contrast, Ru-orbital order is suppressed, and the metallic state remains stable in the other  $t_{2g}^2$ RuO<sub>3</sub> perovskites that lack  $t_{2g}^2$ RuO<sub>3</sub> instabilities.

Orbital order lowers magnetic dimensionality relative to the structural dimensionality, and this can open a spin gap in some nonperovskite ruthenates, e.g., a singlet dimerized phase in La<sub>4</sub>Ru<sub>2</sub>O<sub>10</sub> [20,21] and possible Haldane chains in Tl<sub>2</sub>Ru<sub>2</sub>O<sub>7</sub> [15]. This seems not to be the case here; however, the observation of a broad susceptibility maximum at 25 K and the lack of a long range magnetic transition down to 1.5 K, which in conjunction with a Weiss temperature of -54 K corresponds to a frustration factor  $|\theta/T_c| > 36$ , shows that PbRuO<sub>3</sub> does not have a conventional ordered magnetic ground state. The divergence of field- and zerofield-cooled susceptibilities evidences some glassy character to the ground state, but there is no obvious source for structural disorder. One possibility is that the combination of orbital order and octahedral tilting (which gives a Ru-O-Ru angle of 159.8°) weakens nearest neighbor antiferromagnetic superexchange interactions in the xy plane so that they become comparable to the next nearest neighbor couplings. This frustrates spin order in the xy plane, leading to one-dimensional (z direction) magnetic behavior in the three-dimensional perovskite lattice imposed by orbital

In summary, the low temperature properties of PbRuO<sub>3</sub> show that normally hidden orbitally ordered states such as that of degenerate  $t_{2g}^4 \mathrm{Ru}^{4+}$  ions in ruthenate perovskites may be stabilized by coupling to electronic instabilities of other cations. This may provide a strategy for accessing orbitally ordered states of other 4d and 5d transition metal oxide networks. The combined order of Pb s and p hybridized orbitals, Ru d orbitals, and O-centered octahedral tilting instabilities results in an anomalously high symmetry ground state structure that inverts the usual groupsubgroup symmetry descent. These distortions also suppress long range spin order in PbRuO<sub>3</sub>, and further experiments and theoretical work will be needed to elucidate the magnetic ground state.

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