Superconducting Behavior in Compressed Solid SiH₄ with a Layered Structure

Xiao-Jia Chen,^{1,2,3} Jiang-Long Wang,¹ Viktor V. Struzhkin,² Ho-kwang Mao,² Russell J. Hemley,² and Hai-Qing Lin¹

¹Department of Physics and ITP, Chinese University of Hong Kong, Hong Kong, China

²Geophysical Laboratory, Carnegie Institution of Washington, Washington, D.C. 20015, USA

³School of Physics, South China University of Technology, Guangzhou 510640, China

(Received 19 June 2007; published 12 August 2008)

The electronic and lattice dynamical properties of compressed solid SiH₄ have been calculated in the pressure range up to 300 GPa with density functional theory. We find two energetically preferred insulating phases with $P2_1/c$ and Fdd2 symmetries at low pressures. We demonstrate that the *Cmca* structure having a layered network is the most likely candidate of the metallic phase of SiH₄ over a wide pressure range above 60 GPa. The superconducting transition temperature in this layered metallic phase is found to be in the range of 20–75 K.

DOI: 10.1103/PhysRevLett.101.077002

PACS numbers: 74.10.+v, 74.25.Jb, 74.62.Fj

SiH₄ offers a source for high purity silicon in epitaxial and thin film deposition which is at the base of electronics and microdevices. There is ongoing interest in this material as well due to the suggestion of Ashcroft [1] that SiH_4 would eventually undergo a transition to metallic and then a superconducting state at pressures considerably lower than may be necessary for solid hydrogen. Exploring the possibility of metallic hydrogen has long been a major driving force in high-pressure condensed matter science and remains an important challenge in modern physics and astrophysics [2]. Recent experimental work on SiH₄, using diamond-anvil cell techniques, has revealed an enhanced reflectivity with increasing pressure [3,4]. It was found [4] that solid SiH₄ becomes opaque at 27-30 GPa and exhibits Drude-like behavior at around 60 GPa, signalling the onset of pressure-induced metallization. Structural information is the primary step toward understanding these observed electronic properties.

SiH₄ has a rich phase diagram with at least seven known phases [4–6]. Only one solid phase has been reported in the pressure range between 10 and 25 GPa and at room temperature, with a monoclinic structure (space group $P2_1/c$) [7]. The very low hydrogen scattering cross section of hydrogen-containing materials in most diffraction methods makes structural determination very difficult, specifically in determining the H positions, and the neutron diffraction technique is limited to 30 GPa [8]. The challenge of experimentally or theoretically determining the high-pressure structures of SiH₄ is still enormous.

The sequence of SiH₄ structures provides the basis for understanding whether the material is a favorable candidate of a high-temperature superconductor [9]. Currently, there has been much effort in searching for possible candidate structures of this material on the basis of the firstprinciples density functional theory which has proved an efficient means of calculating quite accurate energies. For a metallic *Pman* SiH₄ phase, Feng *et al.* [10] obtained a superconducting transition temperature T_c of 166 K at 202 GPa by using the electron-phonon coupling strength for lead under ambient pressure as both materials have the same characteristic density of states per volume at relevant pressures. Pickard and Needs [11] also studied the structural properties of SiH₄ and mentioned the possibility of superconductivity in a C2/c phase. All previous work [7,10,11] was done without including the calculations of the phonon spectra and electron-phonon coupling parameters. Later phonon calculations by Yao *et al.* [12] showed that the *Pman* structure is in reality not stable and that a new C2/c structure is dynamically stable from 65 to 150 GPa. This C2/c SiH₄ phase was predicted to exhibit superconductivity close to 50 K at 125 GPa [12]. The identification of high-pressure structures of SiH₄ is still a matter of debate, and no study on superconductivity with other stable structures has been attempted.

In this Letter, we report a theoretical study of structural, electronic, vibrational, and superconducting properties of solid SiH₄ at high pressures. We find two insulating phases $P2_1/c$ and Fdd2 at low pressures and identify the layered *Cmca* structure as the most likely metallic phase of SiH₄, with a superconducting transition temperature in the range of 20–75 K.

To study the structural and electronic behavior of SiH₄ over a wide range of pressure, we used the Perdew-Burke-Ernzerhof generalized gradient approximation (GGA) density functional and projector augmented wave method as implemented in the Vienna ab initio simulation package (VASP) [13]. An energy cutoff of 450 eV is used for the plane wave basis sets, and $16 \times 16 \times 16$ and $8 \times 8 \times 8$ Monkhorst-Pack k-point grids are used for Brillouin zone sampling of two SiH₄ molecular cells and four SiH₄ molecular cells, respectively. The lattice dynamical and superconducting properties were calculated by the QUANTUM-ESPRESSO package [14] using Vanderbilt-type ultrasoft potentials with a cutoff energy of 25 and 200 Ry for the wave functions and the charge density, respectively. $12 \times$ 12×12 Monkhorst-Pack k-point grids with Gaussian smearing of 0.02 Ry were used for the phonon calculations at $4 \times 4 \times 4$ *q*-point mesh, and double *k*-point grids were used for calculation of the electron-phonon interaction matrix element.

0031-9007/08/101(7)/077002(4)

We performed a systematic study of the phase stability of SiH₄ based on *ab initio* first-principles calculations. Out of more than 100 structures that we studied, six new polymorphs of SiH₄ with low enthalpies are found in the pressure range from 0 to 300 GPa. In Fig. 1, we plot the pressure dependence of their enthalpies along with the results for the C2/c and $I4_1/a$ structure reported previously [11]. A monoclinic structure with $P2_1/c$ symmetry has the lowest enthalpy below 27 GPa, in good agreement with the recent experiments [7]. The $P2_1/c$ structure consists of four isolated covalently bonded SiH₄ tetrahedra with the H atom of one molecule pointing away from the H atoms of a neighboring molecule. We find that a facecentered orthorhombic structure with Fdd2 symmetry appears stable from 27 to 60 GPa. These two phases are insulating.

Near 50 GPa, there are three other competitive, lowenthalpy structures with the C2/c, $I4_1/amd$, and $I4_1/a$ symmetry. Our C2/c structure was based on the UI₄ arrangement, which exhibits a layered structure. However, the C2/c structure predicted by Pickard and Needs [11] forms three-dimensional networks at high pressures. For clarification, we name their structure as C2/c (3D) and our layered structure as C2/c (2D). There are subtle differences in band structures between the C2/c predicted by Yao et al. [12] and our C2/c (2D). Our C2/c (2D) structure is composed of sixfold-coordinated SiH₆ octahedra inside a layer, but Si-H bonding between different layers is absent. It is still dynamically stable up to 250 GPa. However, the stability of the C2/c phase of Yao *et al.* [12] is stable only up to 150 GPa. On compression, the sixfoldcoordinated SiH₆ octahedra are transformed into eightfold-coordinated SiH_8 dodecahedra with a S_2H_2 bridgelike bonding arrangement within the layer. Above 110 GPa, the C2/c (2D) structure becomes metallic. The $I4_1/amd$ SiH₄ phase is composed of eightfold-coordinated SiH₈ dodecahedra, and every Si atom shares two H atoms with other Si atoms, as in the $I4_1/a$ structure. The



FIG. 1 (color online). The enthalpy versus pressure for competitive structures of SiH₄. The enthalpy of the $I4_1/a$ phase is taken as the reference point.

 $I4_1/amd$ -type SiH₄ is a semimetal between 40 and 70 GPa.

The $I4_1/a$ structure was found to have the lowest enthalpy over a wide pressure range (60-220 GPa), consistent with previous calculations [11]. In an early powder xray diffraction study [15], $I4_1/a$ was considered as one of most plausible structures for the low-temperature phase II. Raman measurements [4] indicate that this structure may not exist for SiH₄ under high pressure and at room temperature. SiH₄ in the $I4_1/a$ structure is believed to be stable only at low temperature. Between 220 and 270 GPa, a metallic *Cmca* structure has the lowest enthalpy. Upon further compression, the C2/c (3D) phase possesses the lowest enthalpy between 270 and 300 GPa, which confirms the previous calculations [11]. The *Cmca* phase is also a layered structure that consists of eightfold-coordinated SiH₈ dodecahedra. In each layer, the Si-H bonding arrangement is the same as that in the $I4_1/a$ structure. Thus, the *Cmca* phase can be viewed as the 2D analogue of the $I4_1/a$ phase.

In the 60–100 GPa range, we found a metallic triclinic structure with a $P\bar{1}$ space group with eightfold SiH₈ coordination. It has almost the same enthalpy as the Cmca structure between 60 and 100 GPa. Both of their enthalpies are within 0.25 eV of the insulating $I4_1/a$ structure. The enthalpic difference between them and $I4_1/a$ is reduced with increasing pressure. We thus obtain six energetically favorable structures for SiH₄ at high pressures. The atomic arrangement for each of these structures is shown in Fig. 2. The P1, Cmca, and C2/c (2D) phases have layered structures and are metallic. Experimentally, the emergence of a metallic phase of SiH₄ was reported for pressures just above 60 GPa [4]. The theoretically obtained insulating $I4_1/a$ phase as the ground state is in contradiction to experiment. This is probably due to temperature effects (total energy calculations correspond to the zero temperature enthalpy) or a slight failure of the GGA approximation. We also should mention that the enthalpic contributions from electron-phonon coupling were not included when determining structure. The electron-phonon coupling was assumed to have negligible and similar effects on enthalpy for different structure. This may shift the pressure at which structure transform occurs, but the overall picture of the phase diagram is believed to remain unchanged. The enthalpy of the metallic *Cmca* could be lowered if we would include enthalpic contributions from electron-phonon coupling because it plays an important role in superconductivity (discussion below).

As the 2D analogue of $I4_1/a$, the metallic *Cmca* has comparable enthalpy to $I4_1/a$ over the pressure range from 60 to 220 GPa, but it possesses the lowest enthalpy at higher pressures. Although the experimentally observed x-ray diffraction patterns have been refined as the $P6_3$ structure [9], we found that the $P6_3$ has relatively high enthalpy compared to the *Cmca*. Phonon calculations further reveal that the $P6_3$ is a dynamically unstable state.



FIG. 2 (color online). The energetically most favorable structures computed for SiH_4 polymorphs at various pressures.

Consequently, it can be concluded that the most likely candidate for the metallic SiH_4 is the *Cmca* phase.

We now discuss the electronic, phononic, and superconducting properties of the metallic SiH₄ with Cmca symmetry. Figure 3(a) shows the calculated band structure for the Cmca SiH₄ at 250 GPa. It can be seen that the Cmca structure is metallic. The valence bands cross the Fermi level E_F along the $Y\Gamma$ direction, while the conduction bands cross E_F near the Γ point. Upon compression, the conduction band crossing across E_F at the Γ point shifts lower in energy, while the valence band across E_F along the $Y\Gamma$ direction only moves up slightly in energy. The net effect of the pressure-induced band shifts is to increase the volume of the Fermi surface and the phase space for the electron-phonon interaction. Note that there exists a platform near the Fermi level along $Y\Gamma$ direction. This is also a characteristic of layered cuprate superconductors. The flat band at the Fermi level will result in a large density of states, which in turn could lead to high- T_c superconductivity.

The structural stability of the Cmca SiH₄ phase has been examined through lattice dynamics calculations. The typical results of the phonon dispersion and projected phonon density of states at pressure of 250 GPa are displayed in Fig. 3(b). The Cmca stability is confirmed by the absence of imaginary frequency modes. There are weak interactions between the Si framework and H atoms over the whole frequency range. The heavy Si atoms dominate the low-frequency vibrations, and the light H atoms contribute significantly to the high-frequency modes. Three separate regions of bands can be recognized. The modes for the



FIG. 3 (color online). (a) Electronic band structure for the Cmca SiH₄ phase at 250 GPa. The horizontal line shows the location of the Fermi level. (b) The phonon dispersion and phonon density of states (PDOS) projected on Si and H atoms for the Cmca SiH₄ at 250 GPa.

frequencies below 750 cm⁻¹ are mainly due to the motions of Si. The bands around 200 cm⁻¹ are caused by acoustic phonons. The Si-H-Si bending vibrations dominate the intermediate-frequency region between 750 and 1200 cm^{-1} . At high frequencies above 1200 cm^{-1} , the phonon spectrum belongs to the Si-H bond stretching vibrations.

The electron-phonon spectral function $\alpha^2 F(\omega)$ is essential in determining the electron-phonon coupling strength λ and logarithmic average phonon frequency ω_{log} . We have calculated $\alpha^2 F(\omega)$ for the metallic SiH₄ phases in the pressure range of interest. Figure 4 shows the results for the *Cmca* SiH₄. Below 600 cm⁻¹, the major contributions to $\alpha^2 F(\omega)$ come from the phonon modes involving Si-Si vibrations, and the remaining part of the electron-phonon coupling is mainly due to the phonon modes involving H-H vibrations. The $\alpha^2 F(\omega)$ on the wide high-energy side is significantly higher than that on the narrow low-energy side. Thus the high-energy H-H vibrations dominate the total λ value. Among the pressure levels considered, the *Cmca* phase at 70 GPa has a relatively large $\alpha^2 F(\omega)$ over the frequency range from 700 to 1700 cm^{-1} , resulting in a large λ .



FIG. 4 (color online). Electron-phonon spectral function $\alpha^2 F(\omega)$ versus frequency ω of the *Cmca* SiH₄ at pressures.

We next examine the superconductivity in the metallic SiH₄ with Cmca structure, using the T_c equation derived by Allen and Dynes [16]. In the calculations, we took the Coulomb pseudopotential μ^* to be 0.1, which was found to reproduce T_c in MgB₂ [17]. Figure 5 shows the pressure dependence of ω_{\log} , λ , and T_c for the *Cmca* SiH₄. The calculated ω_{\log} and λ do not show a monotonic pressure dependence. The variation of T_c with pressure is found to resemble the λ behavior, decreasing with increasing pressure but increasing upon further compression after passing a minimum value of \sim 20 K at 150 GPa. This behavior is very similar to the experiments [9], though the magnitude of T_c is different. It should be mentioned that the Coulomb pseudopotential may affect the estimation of T_c . In hydrogen-dominant alloys, the phonon energy is usually a larger fraction of the Fermi energy than in most metals, and this tends to enhance μ^* . We have gotten a slightly smaller value of T_c when using a larger μ^* of 0.12, yielding the correct tendency of T_c as experimentally observed. For hydrogen-rich material like SiH₄, we believe that a choice of μ^* in the range of 0.09–0.12 would capture the basic



FIG. 5 (color online). Calculated (a) logarithmic average phonon frequency ω_{\log} , (b) electron-phonon coupling parameter λ , and (c) superconducting transition temperature T_c of the *Cmca* phase of metallic SiH₄ as a function of pressure.

physics. In fact, typical values of μ^* have been confirmed to be in the vicinity of 0.1 at least for high density systems [18].

In summary, we have investigated the structural stability of silane under pressure up to 300 GPa. The $P2_1/c$ phase is confirmed to be a good candidate for the low-pressure insulating phase. Between 27 and 60 GPa, Fdd2 is predicted to be the structure of another insulating phase. At higher pressures, the *Cmca* is proposed to be the most likely candidate of the metallic phase of SiH₄. The superconductivity of the *Cmca* phase with a transition temperature ranging from 20 to 75 K is found to be mainly attributed to the strong electron-phonon coupling due to the phonon modes involving H-H vibrations.

We are grateful to J.S. Tse, R.E. Cohen, and S.A. Gramsch for discussions. This work was supported by Grants No. HKRGC-402205, No. DEFG02-02ER34P5, No. DEFC03-03NA00144, and No. DMR-0205899. X.-J.C. thanks CUHK for kind hospitality during the course of this work.

- [1] N.W. Ashcroft, Phys. Rev. Lett. 92, 187002 (2004).
- [2] V.L. Ginzburg, Rev. Mod. Phys. 76, 981 (2004).
- [3] L. L. Sun, A. L. Ruoff, C. S. Zha, and G. Stupian, J. Phys. Condens. Matter 18, 8573 (2006); 19, 479001 (2007).
- [4] X. J. Chen, V. V. Struzhkin, Y. Song, A. F. Goncharov, M. Ahart, Z. X. Liu, H. K. Mao, and R. J. Hemley, Proc. Natl. Acad. Sci. U.S.A. 105, 20 (2008).
- [5] K. Clusius, Z. Phys. Chem., Abt. B 23, 213 (1933).
- [6] R. E. Wilde and T. K. K. Srinivasan, J. Phys. Chem. Solids 36, 119 (1975).
- [7] O. Degtyareva, M. Martinez-Canales, A. Bergara, X.J. Chen, Y. Song, V.V. Struzhkin, H.K. Mao, and R.J. Hemley, Phys. Rev. B 76, 064123 (2007).
- [8] Y. Ding, J. Xu, C. T. Prewitt, R. J. Hemley, H. K. Mao, J. A. Cowan, J. Z. Zhang, J. Qian, S. C. Vogel, K. Lokshin, and Y. S. Zhao, Appl. Phys. Lett. 86, 052505 (2005).
- [9] After submission of the present work, evidence for superconductivity in metallic silane was reported [M. I. Eremets *et al.*, Science **319**, 1506 (2008)].
- [10] J. Feng, W. Grochala, T. Jaroń, R. Hoffmann, A. Bergara, and N. W. Ashcroft, Phys. Rev. Lett. 96, 017006 (2006).
- [11] C. J. Pickard and R. J. Needs, Phys. Rev. Lett. 97, 045504 (2006).
- [12] Y. Yao, J. S. Tse, Y. Ma, and K. Tanaka, Europhys. Lett. 78, 37003 (2007).
- [13] G. Kresse and J. Furthmuller, Comput. Mater. Sci. 6, 15 (1996).
- [14] See http://www.pwscf.org; also S. Baroni, S. de Gironcoli, A. Dal Corso, and P. Giannozzi, Rev. Mod. Phys. 73, 515 (2001).
- [15] W. M. Sears and J. A. Morrison, J. Chem. Phys. 62, 2736 (1975).
- [16] P.B. Allen and R.C. Dynes, Phys. Rev. B 12, 905 (1975).
- [17] P.P. Singh, Phys. Rev. Lett. 97, 247002 (2006).
- [18] C. F. Richardson and N. W. Ashcroft, Phys. Rev. Lett. 78, 118 (1997).