

## Waxing and waning of dynamical heterogeneity in the superionic state

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Using molecular dynamics simulations of  $\text{UO}_2$ —a type II superionic conductor—we identify a well-defined onset of dynamic disorder ( $T_\alpha$ ), which is remarkably correlated to a nontrivial advance of dynamical heterogeneity (DH). Quantified by the correlations in the dynamic propensity and van Hove self-correlation function, the DH is shown to grow with increasing temperature from  $T_\alpha$ , peak at an intermediate temperature between  $T_\alpha$  and  $T_\lambda$ —the superionic transition temperature—and then recede. Surprisingly, the DH attributes are not uniform across the temperatures—our investigation shows a low temperature ( $\alpha_T$ ) stage DH, which is characterized by weak correlations and a plateau-like period in the correlations of the propensity, and a high temperature ( $\lambda_T$ ) stage DH with strong correlations that are analogous to those in typical supercooled liquids. Our work, which has rigorously identified the onset of superionicity, gives a different direction for interpreting scattering experiments on the basis of statistical, correlated dynamics.

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Amorphous and disordered materials are particularly challenging to characterize—neither is their structure readily amenable to geometric or topological classifications as in defected crystals, nor can their dynamics be described by hydrodynamic theories as in simple liquids. Solid state superionic or fast ion conductors, which typically fall in the class of disordered solids [1], are principally composed of two or more elements that exhibit concomitant crystalline and dynamically disordered structures with an exceptional liquidlike ionic conductivity across a range of thermodynamic states. The copious conduction pathways may be facilitated by a thermodynamic phase change (type I conductors), or may emerge gradually without an attendant phase transition (type II conductors) [2]. The remarkable increase in the ionic conductivity, often by several orders of magnitude, is the principal impetus for the recent investigations, both experimentally and theoretically, of crystalline, polymeric, glassy, and nano/heterostructured materials for a variety of applications that include fuel cells [3], batteries [4–6], sensors, switching [7], hydrogen storage [8], and thermoelectrics [9].

A key signature of the type II conductors is the sharp rise and fall of the specific heat, known as the superionic or the  $\lambda$  transition, at a certain thermodynamic state [2]; changes to phonon spectra, as well as to the elastic and transport properties, are also reported in the vicinity of this transition. At the microscopic level, investigations with molecular dynamics (MD) simulations show that the ions are primarily transported through correlated hopping processes, where one ion jump coerces the motion of several neighboring ions [1]. Often aided by anharmonic and asymmetric thermal vibrations, the cooperative, but peripatetic ionic motion is a distinguishing dynamic feature of type II superionic conductors [2,10–12].

Given the discrete nature of the jumps, the ionic transport, on one hand, may be cast in the framework of dynamic Frenkel clusters [2,13], which presumes precise defect types and positions that are most likely to be encountered during the hopping processes. On the other hand, ionic hopping takes

place over a time scale that is  $O(1-10)$  ps [13] indicating liquidlike characteristics that have been likened to those in the supercooled liquids [14,15] with the superionic transition temperature ( $T_\lambda$ ) corresponding to the glass transition temperature ( $T_g$ ). Such a comparison elicits an intriguing but unresolved question: Do type II superionic conductors portray dynamical heterogeneity—a definitive hallmark of supercooled and glassy states? Although the similarities between superionic and supercooled states have been brought out before [14–17], a concrete understanding of the evolution of dynamical heterogeneity is somewhat nebulous. A related question arises on the thermodynamic range of the superionic state, which is bounded by equilibrium crystalline and liquid phases. It is tempting to inquire whether a crossover temperature or an onset of superionicity exists, where the dynamic disorder is first manifested in type II conductors. Unlike the superionic transition temperature ( $T_\lambda$ ), the concept of a superionic onset temperature is not well formed even though experimental evidence does attest to such an onset. And lastly, if a well-defined superionic onset exists at  $T_\alpha$ , will the likelihood of dynamical heterogeneity occurring be more pronounced near  $T_\alpha$ , or near  $T_\lambda$  as the loose correspondence to  $T_g$  suggests? The aim of this Rapid Communication is to throw more light on the extent or range of the superionic state, and elucidate the hybrid nature of the dynamically disordered species in type II superionic conductors.

First, we emphasize that several sets of experimental data on type II superionic conductors (Sec. A of [18]) reveal the existence of a definitive transition temperature ( $T_\alpha$ ) that marks the onset of disorder (primarily evaluated from neutron scattering and diffraction experiments), which is distinctly different from the superionic transition temperature ( $T_\lambda$ ). In this work we present evidence for a clear transition at  $T_\alpha$  using MD simulations and associate this temperature to the transition from a crystalline state to a superionic state, which is manifested by the emergence of dynamical heterogeneity (DH). We have chosen  $\text{UO}_2$  as our model fluorite material ( $T_\lambda \approx 2650$  K,  $T_\alpha \approx 1950/2000$  K) as the disordered state spans a wide temperature range that allows an unambiguous examination and interpretation of superionic and disordered transitions. A rigid-ion interionic potential [19], which has

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been benchmarked to a spectrum of properties [12,19,20] with reasonable accuracy, is used in the MD simulations (Sec. B of [18]). Polarization effects are important in the simulation of superionic materials; however, rigid-ion potentials can be made to reproduce the properties and dynamics reasonably well by tuning the dispersion term in the interionic potential [21]. MD simulations are performed with 1728 oxygen ions using periodic boundary conditions at zero pressure for different temperatures ranging from 1500 to 3100 K. At a temperature of  $\sim 1900$  K, we observe the onset of disorder by detecting subdiffusive dynamics from the mean square displacements and discernible changes in the properties such as lattice parameter (Sec. B of [18]).

Detecting DH in a superionic state is challenging for two reasons: First, the superionic states are completely in equilibrium enveloped by two well-defined thermodynamic states—liquid and crystal (solid). A supercooled state, in stark contrast, is in a frustrated, nonequilibrium condition, and is bounded by an equilibrium liquid and a nonequilibrium glassy state. Typically, DH is quantified by the generalized four point susceptibility function ( $\chi_4$ ) or through similar metrics [22–26]. The theoretical study by Toninelli *et al.* [23] shows that  $\chi_4(t)$  can be divided into several time regimes such as ballistic, elastic, early beta relaxation, and so forth. For supercooled liquids,  $\chi_4(t)$  shows a maximum at a time that is  $O(\tau_\alpha)$ , where  $\tau_\alpha$  is the alpha relaxation time. Further, the magnitude of  $\chi_4(t)$ -max shows a monotonic increase with decreasing temperature [27], illustrating the growing spatial correlations. Conspicuously,  $\chi_4(t)$  has an elastic contribution from phonons that increases with time. For perfect crystals,  $\chi_4(t)$  grows unbounded without decay, which indicates a general difficulty in differentiating deeply supercooled states from the crystalline states, especially with MD simulations that have limited time windows. In our study, we have used the correlation in the propensity—a metric defined as the mean square displacement of an ion in an isoconfigurational ensemble [28,29]—and van Hove self-correlation function [30] to uncover the onset of DH. The spatial correlation of the propensity in a system of  $N$  oxygen ions is defined as [31]

$$C_d(r,t) = \frac{\langle \delta w(i,t) \delta w(j,t) \rangle}{\langle [\delta w(t)]^2 \rangle}, \quad (1)$$

where  $w(i,t)$  is the propensity  $\langle r^2 \rangle_{ic}$  of the  $i$ th ion at time  $t$  and  $\delta w(t) \equiv w(i,t) - \langle w(i,t) \rangle$ . The normalization is given by  $\langle [\delta w(t)]^2 \rangle = \frac{1}{N} \sum_{i=1}^N [\delta w(t)]^2$ .

In Eq. (1),  $i$  and  $j$  are two oxygen ions such that the magnitude of the distance between them falls within an interval of  $0.25 \text{ \AA}$ , centered at  $r$ . Analysis by Razul *et al.* [31] shows that  $C_d$  is a reliable metric for detecting DH in the supercooled liquids: It depicts a peak amplitude, which (i) manifests at times that scale with  $\tau_\alpha$ , and (ii) increases monotonically with decreasing temperature (Sec. C of [18]). Most importantly, the mean value of  $C_d$  is nearly zero at all times for perfectly crystalline states, especially at low temperatures. We have verified this with MD simulations for several materials at temperatures close to 0 K. While thermal vibrations enhance the magnitude of  $C_d$ , it remains nearly constant with time, unlike  $\chi_4$ , which increases unbounded for perfectly crystalline

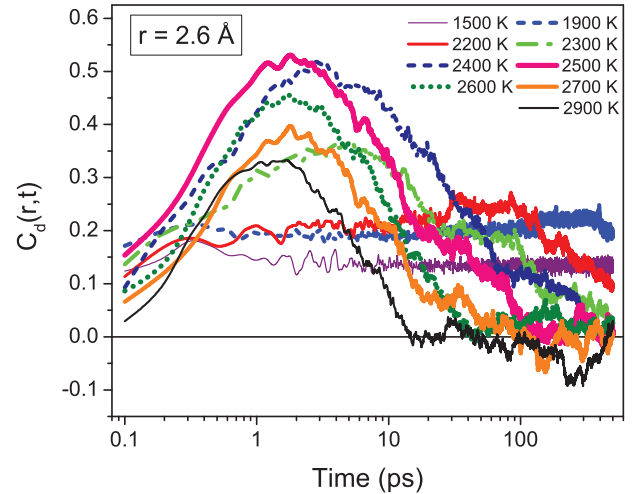


FIG. 1. (Color online) Spatial correlation ( $C_d$ ) of the dynamic propensity of oxygen ions evaluated in a restricted, isoconfigurational ensemble for different temperatures.

states. The difference between  $\chi_4$  and  $C_d$  comes from the pair (two-ion) correlation in the latter that decays very quickly in the crystalline state.

Figure 1 delineates the main result of our investigation: It shows the progression of  $C_d$  based on 100 independent isoconfigurational runs evaluated at  $r = 2.6 \text{ \AA}$ , which is close to the shortest O-O interionic separation distance. At 1500 K,  $C_d$  shows an oscillatory behavior that emanates from the vibratory motion of the oxygen ions; the mean value, however, remains constant for the whole duration of the simulation (500 ps). On increasing the temperature,  $C_d$  exhibits two discernible changes: (i) the magnitude of  $C_d$ , which represents the strength of the correlation increases, and (ii) the magnitude increases gently and then decreases after passing through a maximum. Unlike in the supercooled states,  $C_d$ , however, exhibits a plateau-like period before decay. Thus the correlation in the propensity brings out the transition from a crystalline state into a superionic state through an increased correlation strength, and a peaking and decay behavior in  $C_d$  that signifies the emergence of dynamical heterogeneity. With increasing temperature, the magnitude of  $C_d$  increases with the peaks being exhibited at shorter times—the maximum amplitude can be observed at 2500 K. And on further increase in the temperature, the amplitude of the peak magnitude of  $C_d$  decreases. Figure 2, which shows the variation of the peak magnitude of  $C_d$ , depicts the waxing and waning of DH among the oxygen ions. For temperatures 2200 and 1900 K, the peak amplitude of  $C_d$  is determined as the mean value during the “plateau” period: 30–80 ps and 200–400 ps for 2200 and 1900 K, respectively. Note that  $C_d$  averaged over  $1 \text{ \AA}$  is somewhat smoother with the peak magnitude observed at approximately 2400 K; similar trend is observed at other radial locations as well. Based on the peak magnitudes of  $C_d$ , the maximum DH intensity can be regarded to fall in the temperature range of 2400–2500 K.

The temporal behavior of  $C_d$  reveals a low temperature ( $\alpha_T$ ) and a high temperature ( $\lambda_T$ ) DH behavior with the former corresponding to the lower temperatures near  $T_\alpha$  (1900–2300 K), while the latter corresponds to the higher temperatures near  $T_\lambda$

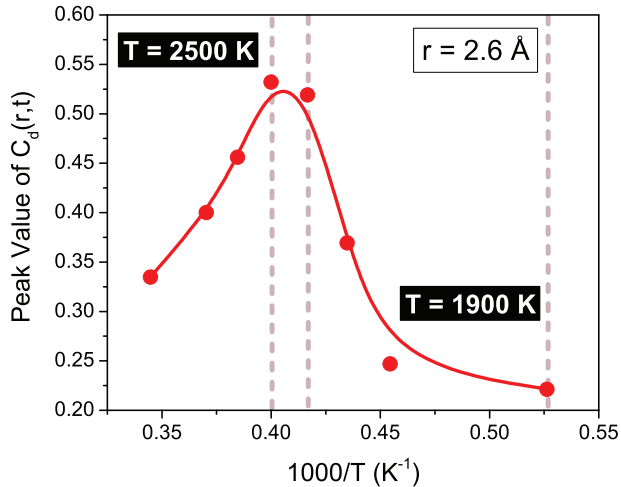


FIG. 2. (Color online) Waxing and waning of DH among the oxygen ions, illustrated by the variation of the peak amplitude of  $C_d$  with temperature.

(2400–2900 K). The  $\alpha_T$  stage is characterized by deviations from a constant  $C_d$ , low correlation strength and a relatively long plateau-like period, signifying correlated but hindered ionic motion. The low correlation strength also indicates that only a relatively small number of ions participate in the DH process.

The  $\alpha_T$  DH stage is distinctively different from that observed in typical supercooled liquids—given the preceding crystalline state and the weak growth, it can be characterized as solidlike. While only 500 ps have been simulated for the lower temperatures, we have verified that the system remains in a vibratory state at temperatures lower than  $T_\alpha$ . Further, we have used the temperature accelerated dynamics (TAD) of Voter [32] to gauge the long-time dynamics—no transitions were accepted for a perfect  $\text{UO}_2$  crystal at temperatures lower than  $T_\alpha$ , indicating the dominant vibratory motion of the ions without crossing an energy barrier to another state (diffusive motion). In contrast, the high temperature  $\lambda_T$  DH stage is characterized by a significant strength in the correlations and a peaking behavior that is analogous to supercooled liquids.

The crossover from the solidlike DH behavior to a more liquidlike response (at  $\sim 2300$  K) has a structural origin, which is described in Sec. D of [18].

We will now provide more evidence for the evolution of DH as well as for the crossover from solidlike to liquidlike behavior of the oxygen ions. The van Hove self-correlation function for oxygen ions—shown in the top panels of Fig. 3 at different temperatures—delineates two striking features: At temperatures above 1900 K,  $G_s(r,t)$  deviates from the Gaussian shape and develops a tail, which (i) depicts discernible peaks, particularly prominent at longer times, and (ii) varies exponentially with distance at a time when  $C_d$  exhibits a peak amplitude ( $\tau_\alpha$ ). The peaks are characteristic of hopping processes wherein ions undergo discrete jumps [33] (Sec. E of [18]); we have verified that the peak locations agree closely with the nearest neighbor positions in the O-O radial distribution function. The exponential tail in  $G_s(r,t)$ , on the other hand, connotes the presence of faster moving ions—the characteristic signature of DH—which is also observed in supercooled liquids, colloids, and granular materials [30]. With increasing temperature, the exponential tail at  $\tau_\alpha$  strengthens, which confirms the presence of DH. The  $\alpha_T$  and  $\lambda_T$  stages of DH can also be distinguished from the progression of the exponential tail. At 2200 K, the exponential tail in  $G_s$  at  $\tau_\alpha$  (80 ps), which is somewhat underdeveloped with corrugations, sustains for longer periods of time suffering only a modest change, signifying a solidlike ( $\alpha_T$ ) DH response. In contrast, at a temperature of 2400 K (and above), the  $G_s$  tail at  $\tau_\alpha$  is smoother, which is analogous to that in typical supercooled liquids, signifying a liquidlike ( $\lambda_T$ ) DH response; for the higher temperatures, the exponential tail of  $G_s$  also deteriorates faster.

In the bottom panels of Fig. 3, we highlight the crossover from a solidlike state to a more liquidlike state by alerting one to the appearance of a hydrodynamic Rayleigh peak ( $\nu \approx 0$ ) in the partial dynamic structure function of the oxygen ions for a wave vector,  $k = 0.28 \text{ \AA}^{-1}$ . For 1900 K  $> T < 2200$  K, a partially formed hydrodynamic peak is discernible, indicating a somewhat lethargic disordering mechanism. The Rayleigh peak, however, becomes unmistakable with increasing temperature and its magnitude relative to that of the elastic peak again confirms a liquidlike state beyond 2400 K.

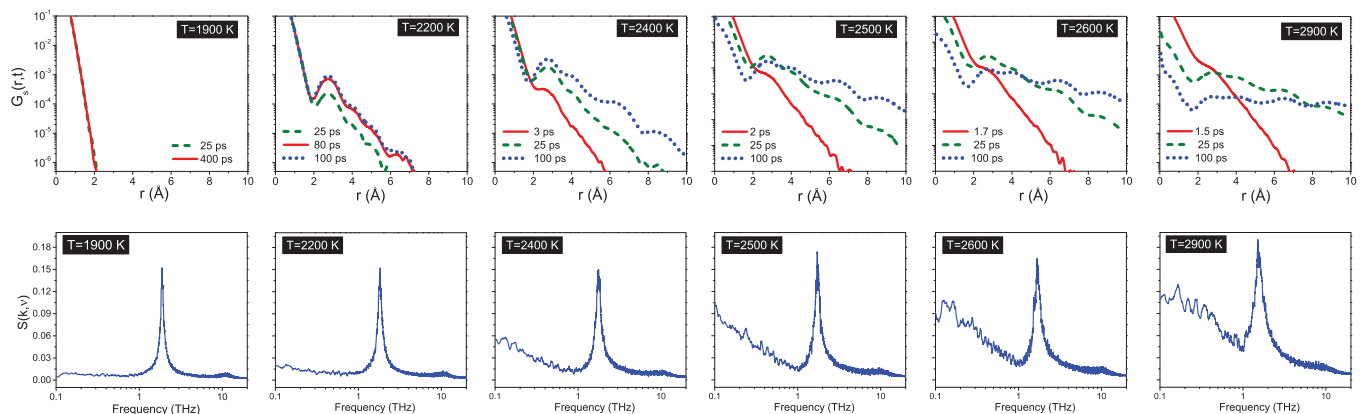


FIG. 3. (Color online) Top panels: The van Hove self-correlation function  $G_s(r,t)$  for the oxygen ions at different temperatures—the solid line corresponds to the time when  $C_d$  exhibits a peak amplitude. Bottom panels: Partial dynamic structure function for oxygen ions (arb. units) for a wave vector of  $k = 0.28 \text{ \AA}^{-1}$ , which is representative of the hydrodynamic limit.

*Outlook.* Illustrated by the correlations in the propensity, DH is demonstrated to advance from a well-marked onset temperature ( $T_\alpha$ )—shown unambiguously—peak at an intermediate temperature between  $T_\alpha$  and  $T_\lambda$ , and then recede. Currently, the superionic state is largely analyzed from a static picture of defect positions [2]; indeed, most neutron and light scattering experiments to date have been interpreted on such a basis. Our investigations have uncovered the remarkable

significance of correlated dynamics, in particular, the nontrivial manifestation of DH that grows against the destabilizing effects of entropy. Our investigations thus give a different direction for interpreting scattering experiments on the basis of statistical, correlated dynamics. Further, the nuanced attributes of the  $\alpha_T$  and  $\lambda_T$  stages of DH in the superionic state, which have become transparent through our inquiry, will be of vivid interest from a theoretical point of view.

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- [1] J. C. Dyre, P. Maass, B. Roling, and D. L. Sidebottom, *Rep. Prog. Phys.* **72**, 046501 (2009).
- [2] S. Hull, *Rep. Prog. Phys.* **67**, 1233 (2004).
- [3] T. Norby, *Nature (London)* **410**, 877 (2001).
- [4] A. Hayashi, K. Noi, A. Sakuda, and M. Tatsumisago, *Nat. Commun.* **3**, 856 (2012).
- [5] N. Kamaya, K. Homma, Y. Yamakawa, M. Hirayama, R. Kanno, M. Yonemura, T. Kamiyama, Y. Kato, S. Hama, K. Kawamoto, and A. Mitsui, *Nat. Mater.* **10**, 682 (2011).
- [6] R. Bouchet, S. Maria, R. Meziane, A. Aboulaich, L. Lienafa, J. P. Bonnet, T. N. T. Phan, D. Bertin, D. Gimes, D. Devaux, R. Denoyel, and M. Armand, *Nat. Mater.* **12**, 452 (2013).
- [7] T. A. Miller, J. S. Wittenberg, H. Wen, S. Connor, Y. Cui, and A. M. Lindenberg, *Nat. Commun.* **4**, 1369 (2013).
- [8] P. Chen, Z. Xiong, J. Luo, J. Lin, and K. L. Tan, *Nature (London)* **420**, 302 (2002).
- [9] H. Liu, X. Shi, F. Xu, L. Zhang, W. Zhang, L. Chen, Q. Li, C. Uher, T. Day, and G. J. Snyder, *Nat. Mater.* **11**, 422 (2012).
- [10] O. Hochrein and D. Zahn, *Solid State Ionics* **180**, 116 (2009).
- [11] M. J. Castiglione and P. A. Madden, *J. Phys.: Condens. Matter* **13**, 9963 (2001).
- [12] S. I. Potashnikov, A. S. Boyarchenkov, K. A. Nekrasov, and A. Ya. Kupryazhkin, *J. Nucl. Mater.* **433**, 215 (2013).
- [13] K. Clausen, W. Hayes, J. E. Macdonald, R. Osborn, and M. T. Hutchings, *Phys. Rev. Lett.* **52**, 1238 (1984).
- [14] A. Gray-Weale and P. A. Madden, *J. Phys. Chem. B* **108**, 6624 (2004).
- [15] A. Gray-Weale and P. A. Madden, *J. Phys. Chem. B* **108**, 6634 (2004).
- [16] K. L. Ngai, J. Habasaki, C. Leon, and A. Rivera, *Z. Phys. Chem.* **219**, 47 (2005).
- [17] K. L. Ngai, *J. Phys.: Condens. Matter* **15**, S1107 (2003).
- [18] See Supplemental Material at <http://link.aps.org/supplemental/10.1103/PhysRevE.89.010301> for more details.
- [19] E. Yakub, C. Ronchi, and D. Staicu, *J. Chem. Phys.* **127**, 094508 (2007).
- [20] A. V. Lunev and B. A. Tarasov, *J. Nucl. Mater.* **415**, 217 (2011).
- [21] A. Dent, P. A. Madden, and M. Wilson, *Solid State Ionics* **167**, 73 (2004).
- [22] S. C. Glotzer, V. N. Novikov, and T. B. Schröder, *J. Chem. Phys.* **112**, 509 (2000).
- [23] C. Toninelli, M. Wyart, L. Berthier, G. Biroli, and J.-P. Bouchaud, *Phys. Rev. E* **71**, 041505 (2005).
- [24] C. Dalle-Ferrier, C. Thibierge, C. Alba-Simionesco, L. Berthier, G. Biroli, J.-P. Bouchaud, F. Ladieu, D. L'Hôte, and G. Tarjus, *Phys. Rev. E* **76**, 041510 (2007).
- [25] S. Karmakar, C. Dasgupta, and S. Sastry, *Phys. Rev. Lett.* **105**, 015701 (2010).
- [26] S. Karmakar, C. Dasgupta, and S. Sastry, *Proc. Natl. Acad. Sci. USA* **106**, 3675 (2009).
- [27] L. Berthier and W. Kob, *J. Phys.: Condens. Matter* **19**, 205130 (2007).
- [28] A. Widmer-Cooper, P. Harrowell, and H. Fynewever, *Phys. Rev. Lett.* **93**, 135701 (2004).
- [29] A. Widmer-Cooper, H. Perry, P. Harrowell, and D. R. Reichman, *Nat. Phys.* **4**, 711 (2008).
- [30] P. Chaudhuri, L. Berthier, and W. Kob, *Phys. Rev. Lett.* **99**, 060604 (2007).
- [31] M. S. G. Razul, G. S. Matharoo, and P. H. Poole, *J. Phys. Condens. Matter* **23**, 235103 (2011).
- [32] A. F. Voter, F. Montalenti, and T. C. Germann, *Annu. Rev. Mater. Res.* **32**, 321 (2002).
- [33] W. Kob and H. C. Andersen, *Phys. Rev. E* **51**, 4626 (1995).