

Effectiveness of the statistical potential in the description of fermions in a worm-algorithm path-integral Monte Carlo simulation of ^3He atoms placed on a ^4He layer adsorbed on graphite

Humam B. Ghassib,¹ Asaad R. Sakhel,² Omar Obeidat,² Amer Al-Oqali,³ and Roger R. Sakhel⁴

¹*Department of Physics, The University of Jordan, Amman, Jordan*

²*Applied Sciences Department, Faculty of Engineering Technology, Al-Balqa Applied University, Amman 11134, Jordan*

³*Department of Physics, Mu'tah University, Al-Karak, Amman, Jordan*

⁴*Department of Basic Sciences, Faculty of Information Technology, Isra University, Amman 11622, Jordan*

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We demonstrate the effectiveness of a statistical potential (SP) in the description of fermions in a worm-algorithm path-integral Monte Carlo simulation of a few ^3He atoms floating on a ^4He layer adsorbed on graphite. The SP in this work yields successful results, as manifested by the clusterization of ^3He , and by the observation that the ^3He atoms float on the surface of ^4He . We display the positions of the particles in 3D coordinate space, which reveal clusterization of the ^3He component. The correlation functions are also presented, which give further evidence for the clusterization.

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I. INTRODUCTION

The ^3He - ^4He II sandwich system [1] and a system of ^3He atoms placed on ^4He layers are among the most fascinating He systems, thanks to their unusual properties. The investigation of this type of system on graphite is quite rare [2], and only a few similar studies along the same lines have been presented, for one component, either ^3He or ^4He , or their mixture [2–20]. An investigation involving ^3He atoms on a ^4He film adsorbed instead on Nuclepore [17,21] was also presented, as well as ^3He atoms on a superfluid ^4He film [14,17]. Several methods have been applied to explore the latter systems. For example, path-integral Monte Carlo (PIMC) has been applied for simulating helium films on a graphite surface [10,22], variational path-integral (VPI) Monte Carlo to investigate ^4He films adsorbed on alkali metal substrates [16], and diffusion Monte Carlo (DMC) to compute the ground-state properties of two-dimensional ^4He [8]. Further, variational Monte Carlo (VMC) was used to explore small mixed clusters of ^3He and ^4He atoms [23]. Nevertheless, because of the presence of the fermionic component, the simulation of systems containing ^3He using MC methods is not an easy task, although Ceperley [24] introduced earlier a PIMC method for the simulation of continuum fermions at nonzero temperature.

In this work, the worm-algorithm path-integral Monte Carlo (WAPIMC) method [25] is used to compute the thermal properties of a ^3He - ^4He system where the ^3He floats on a ^4He film adsorbed on a graphite substrate. WAPIMC is a powerful technique, where one can simulate finite systems at finite temperatures in the grand canonical ensemble. That is, the exchange of particles with the surrounding particle reservoir is allowed. The number of particles is therefore allowed to change during a WAPIMC simulation and is controlled by the chemical potential. We are thus able to simulate a realistic situation, where particles can enter and leave the system. The latter feature is not found in other MC methods and is a big advantage for WAPIMC. In addition, WAPIMC enables us to investigate the microscopic properties, such as clusterization of helium atoms and the structural order. WAPIMC works also very well at high densities, and it can simulate a large number

of particles $N \sim O(10^3)$. In contrast, the standard PIMC method [26] simulates systems only in the canonical ensemble at finite temperature and to $N \sim O(10^2)$. Whereas VPI and DMC work only at zero temperature, WAPIMC works at finite temperatures. One disadvantage of WAPIMC (and therefore PIMC) is that a simulation in the millikelvin temperature regime requires substantial computational time, of the order of a month, on a (for example) Intel Xeon workstation. Therefore WAPIMC simulations with T approaching zero would take years of CPU time.

In analytical approaches, previous work calculated the binding energy of ^3He - ^4He [5] using an analytical variational procedure, following an ansatz due to Feynman (Ref. [6] in Gasparini *et al.* [5]); linear response theory was used to investigate the excitation spectrum of ^3He films adsorbed on substrates [27]; second order perturbation theory was applied to evaluate the ^3He - ^4He effective interaction resulting from the exchange of third-sound quanta [6]; and Fermi-hypernetted-chain techniques were used to obtain the ground-state properties of a ^3He - ^4He mixture using a wave function composed of a product of pair correlation functions [28]. A statistical-mechanical approach was further used to compute the thermodynamic properties of ^3He in thin ^3He - ^4He films [29], and Krotscheck *et al.* [12] applied a variational theory with correlated basis functions to compute the phase diagram of ^3He - ^4He mixtures in 2D. Although analytical methods are useful in obtaining macroscopic quantities like the thermodynamic properties (heat capacity, pressure, entropy, etc.), they lack the ability to display microscopic many-body features of the system, such as the structural properties, particle promotion and demotion between the layers, and the exchange of particles with a surrounding particle reservoir.

The interactions between helium atoms can usually be described by the HFDHE2 potential of Aziz *et al.* [30]. However, in the systems considered here, the ^3He component is initially quite dilute as we begin the simulation with a small number of ^3He atoms. The ^3He component can therefore be approximated by an ideal gas. Such an ideal system can in turn be described by a Slater determinant composed of noninteracting single-particle wave functions, and consequently the

exchange of two fermions leads to a change in the sign of the total wave function. In the case of an interacting system, the wave function is a many-body wave function combined of a product of single-particle wave functions and a train of Jastrow functions $\prod_{i<j} f(r_{ij})$. The Jastrow function $f(r_{ij})$, where r_{ij} is the distance between atoms i and j , describes the interactions between the particles in the wave function. Therefore, the Slater determinant includes correlations between the particles, and no sign change arises when two fermions are swapped. Although in the current simulations we do not use a Slater determinant, the presence of fermions causes the well-known sign problem [31] to arise when two fermions are swapped in a dilute system. Our purpose in this work is to present a method to circumvent the sign problem in a worm-algorithm path-integral Monte Carlo (WAPIMC) [25] simulation of a ^3He - ^4He system, where the MC statistics blow up to infinity if the partition function goes to zero [32]. We propose to disallow the swap of fermions and to use instead a so-called statistical potential (SP) [33]. The SP describes “statistical interactions” between fermions. Earlier, Anderson and Miller [6] included the effects of a hard core in the ^3He - ^3He interaction potential to describe their repulsion.

In this paper, our chief goal is to demonstrate the effectiveness of the SP in a WAPIMC simulation of a *few* ^3He atoms placed on a layered ^4He structure adsorbed on graphite. We focus mainly on the clusterization of the ^3He component [2,34,35] driven by an effective interaction, resulting from a combination of interaction potentials in the system. We therefore present figures displaying the positions of the particles, which reveal the clusterization of the ^3He component. The correlations between the helium atoms are also explored, which provide further evidence for clusterization. This paper reports a sample of results from the current project of a few ^3He atoms on a ^4He layer adsorbed on graphite. This project is part of a set of investigations involving a previous examination of the ^3He - ^4He sandwich system [36] and a future one of ^4He in Vycor.

Our key results are as follows. First, the SP is effective in describing the fermions in a WAPIMC simulation because (i) it allows the ^3He atoms to “float” on the outer surface of ^4He in agreement with experimental observations [2,14]; (ii) it does not prohibit the clusterization of ^3He atoms which display strong correlations; and (iii) the energies of the ^3He atoms are in line with values reported earlier [5]. Second, the properties of the system are found to be largely sensitive to the chemical potential, particularly the final number of ^3He atoms that can be achieved at the end of a simulation. However, it was found that the error bars of the WAPIMC simulation are somewhat large at temperatures $T < 80$ mK, whereas at $T \geq 80$ mK they are small. Further, we would like to draw attention to the fact that, in the temperature range considered, it was very hard to obtain a number of ^3He atoms of the order $\sim O(1)$. The simulation always ended up with $\sim O(10)$.

II. METHOD

In what follows, we briefly outline the method. We modified a previously written WAPIMC code [25,32], originally developed for the ^4He on graphite (used in Ref. [19]) to include the ^3He component, as explained in our previous paper [36].

However, the positioning of the ^3He atoms is slightly different from the sandwich structure [36], as we place the ^3He atoms only on the outer surface of the ^4He layer. In addition, we did not apply the mass-update procedure of a fictitious particle as in Refs. [36,37]. This is because the ^3He atoms floating on the surface of ^4He already have a high mobility. The ^3He - ^4He as well as the ^4He - ^4He interactions are, as usual, described by the HFDHE2 potential of Aziz *et al.* [30]. On the other hand, the ^3He component is dilute and can therefore be considered an ideal gas. However, there are important ^3He interactions which are statistical in nature. That is, the fermionic nature of ^3He enters into play through the SP [33]:

$$v_{\text{SP}}(r) = -k_B T \ln [1 - \exp(-2\pi r^2 / \lambda_{dB}^2)], \quad (1)$$

r being the distance between a pair of particles, and λ_{dB} the thermal de Broglie wavelength. Yet, the code is very much the same as in Ref. [36]. The simulations were conducted in the grand canonical ensemble (GCE) in the millikelvin temperature regime, with carefully chosen chemical potentials for which the number of ^3He atoms would not “disappear” later into the surrounding particle reservoir. It was difficult to find chemical potentials for which the simulations would simultaneously evolve effectively and stabilize at a reasonable number of ^3He atoms. We found that it was very difficult to stabilize the number of ^3He atoms to $O(\sim 1)$.

The dimensions of the simulation box were fixed at $19.693 \times 17.054 \times 26.798 \text{ \AA}^3$ for all systems, set by a pre-defined density. Each simulation was initialized with a few (from 2 to 15) ^3He atoms placed on the surface of 72 ^4He atoms comprising two layers, adsorbed on the surface of a graphite substrate. For 2–15 ^3He atoms, the ^3He coverage lies between 0.5 and 4 % and is quite dilute (cf. Ref. [12]). Then it was run for several weeks of CPU time until thermalization stabilized the energy within a certain error bar. The time step chosen was $\tau = (1/400) \text{ K}^{-1}$, and the temperatures and chemical potentials used are tabulated in Tables I and II. The values of μ listed are renormalized values, by the self-energy of the interacting particles Σ , i.e., $\mu = \mu_0 + \Sigma$, where μ_0 is an initial chemical potential of our choice.

In this work we essentially let the chemical potential decide the number of particles in our system by running it in the GCE. Consequently, the numbers of ^3He and ^4He particles, $\langle N_3 \rangle$ and $\langle N_4 \rangle$ respectively, are updated from their initial values depending on the magnitude of the chemical potential. In passing, it is interesting to note that the gradual dynamic update of $\langle N_3 \rangle$ during thermal evolution represents the well-known adiabatic switching-on process of the ^3He interaction [38]. Interestingly enough, this might be an experimental analog to the Gell-Mann-Low theorem [39].

III. RESULTS AND DISCUSSION

A. Structural properties

We begin with a visualization of the spatial positions of the ^3He and ^4He atoms for two of our systems in 3D coordinates, so as to check for clusterization of the ^3He component. Figures 1 and 2 display these positions for systems at $T = 60, 70$, and 100 mK with $\mu = -14.392, -14.139$, and -11.216 K , respectively. More details about these systems are found in

TABLE I. WAPIMC thermal properties of a system of a number of ^3He atoms placed on a ^4He layer adsorbed on a graphite substrate at temperatures ranging from $T = 60$ to 100 mK, labeled $a-e$, respectively. From left to right: System label, temperature T in mK, chemical potential μ , average total number of particles $\langle N \rangle$, total energy per particle $\langle E \rangle / \langle N \rangle$, pressure per particle $\langle P \rangle / \langle N \rangle$, entropy per particle $\langle S \rangle / \langle N \rangle$ (unitless), internal energy per particle $\langle U \rangle / \langle N \rangle$, and superfluid fraction ρ_s / ρ . Each thermal property indicated is a thermodynamic average in units of K, except for the entropy which is unitless.

System	T (mK)	μ (K)	$\langle N \rangle$	$\langle E \rangle / \langle N \rangle$ (K)	$\langle P \rangle / \langle N \rangle$ (K)	$\langle S \rangle / \langle N \rangle$	$\langle U \rangle / \langle N \rangle$ (K)	ρ_s / ρ (%)
a	60	-14.392	127.24	-47.988	5.507×10^{-2}	2473.57	-74.662	1.936
			± 7.48	± 4.425	$\pm 3.261 \times 10^{-3}$	± 3.071	± 0.398	
b	70	-14.139	125.55	-53.027	5.489×10^{-2}	2304.26	-78.679	1.755
			± 7.07	± 4.948	$\pm 1.609 \times 10^{-2}$	± 1.886	± 0.544	
c	80	-15.397	91.987	-56.575	8.129×10^{-2}	1959.80	-92.741	8.426
			$\pm 2.632 \times 10^{-3}$	± 0.681	$\pm 9.338 \times 10^{-3}$	± 0.450	± 0.217	
d	90	-9.003	162.27	-44.249	3.775×10^{-2}	1398.28	-64.348	3.051
			± 0.906	± 0.296	$\pm 2.450 \times 10^{-4}$	± 0.157	$\pm 2.929 \times 10^{-2}$	
e	100	-11.216	133.25	-45.294	5.710×10^{-2}	1463.56	-70.411	0.210
			± 0.614	± 0.809	$\pm 1.696 \times 10^{-3}$	± 0.491	± 0.105	

Tables I and II, which we shall elaborate below. The ^3He atoms are the solid green circles, whereas the ^4He atoms are the solid red triangles. The vertical positions (z) are with respect to the graphite surface in the xy plane. One can see that there are two ^4He layers, above which a ^3He component resides. In Figs. 1 and 2, a partial clusterization of ^3He is observed. The clusterization of the ^3He atoms in a ^4He environment was predicted earlier by Ghassib and Chatterjee [34], as well as by Ghassib [35]. Nanoclusterization of the ^3He fluid in the same system as ours was observed earlier [2]. In addition, Krotscheck *et al.* [12] found that the effective interaction between ^3He impurities in ^4He is sufficiently attractive to generate ^3He dimers. One could argue that ^3He clusterization is rather surprising, since the interactions between the ^3He atoms are repulsive, as dictated by the fermionic SP, and because the ^3He atoms have a large zero-point motion. Further, the ^3He atoms at $z \sim 9$ Å in Figs. 1 and 2 are located at positions where they only feel the weak tail of the graphite potential (Ref. [25] in Corboz *et al.* [19]), and hence this effect can largely be excluded for ^3He . In contrast, the ^3He atoms strongly feel the ^3He - ^4He interaction at a vertical distance of ~ 3 Å from the second ^4He layer. Therefore at low T , ^3He - ^4He mixing overcomes the repulsive statistics of the ^3He atoms and does not allow them to aggregate into

a separate phase. The ^3He - ^4He interaction further localizes the ^3He atoms on the ^4He surface [14,21,40], as it enhances the effective mass of ^3He atoms [3,14,17]. The increase in the effective mass therefore overcomes the effects of the large zero-point motion of the ^3He atoms. Some ^4He atoms have been promoted (i.e., “evaporated”) to the ^3He layers in the course of the thermalization process. This is an inevitable consequence of any vacancies introduced into the ^3He cloud. As the temperature is increased, clusterization of ^3He ceases to exist as displayed in Fig. 3 at 100 mK. This is because as one increases T , the repulsive statistics becomes stronger in Eq. (1) and begins to overcome the ^3He - ^4He mixing effects. One can then argue that the SP can account for a larger solubility of ^3He into the ^4He at low T . One further observes that the ^3He component has separated into a floating phase above the ^4He component, which is also driven by a larger ^3He density. Not one single ^3He atom has evaporated into the ^3He phase.

It must be mentioned that the finiteness of the system and the boundary conditions play a decisive role in the clusterization process of the ^3He atoms. As the system becomes more confined, the ^3He atoms approach the ^4He atoms which increases the binding, and therefore ^3He clusterization is enhanced. Next to this, the box-wall boundaries of our simulation cell cause the matter waves of the ^3He atoms to

TABLE II. As in Table I, except for the ^3He component of the same systems with the kinetic energy per particle at the end instead of the superfluid fraction.

System	T (mK)	μ (K)	$\langle N_3 \rangle$	$\langle E_3 \rangle / \langle N \rangle$ (K)	$\langle P_3 \rangle / \langle N \rangle$ (K)	$\langle S_3 \rangle / \langle N \rangle$	$\langle U_3 \rangle / \langle N \rangle$ (K)	$\langle K_3 \rangle / \langle N \rangle$
a	60	-14.392	40.576	-2.095	4.326×10^{-5}	165.37	-4.633	3.595
			± 5.421	± 1.108	$\pm 8.347 \times 10^{-3}$	± 0.534	± 0.455	
b	70	-14.139	34.308	-1.717	1.674×10^{-4}	150.91	-4.263	3.394
			± 5.766	± 1.224	$\pm 5.463 \times 10^{-3}$	± 0.638	± 0.447	
c	80	-15.397	12.412	-0.888	1.055×10^{-2}	539.01	-2.648	2.251
			$\pm 6.908 \times 10^{-2}$	$\pm 4.923 \times 10^{-3}$	$\pm 7.803 \times 10^{-6}$	$\pm 3.677 \times 10^{-3}$	$\pm 1.180 \times 10^{-3}$	
d	90	-9.003	76.961	-1.874	2.031×10^{-2}	1122.80	-5.451	4.626
			± 0.881	± 0.083	$\pm 1.312 \times 10^{-4}$	$\pm 6.661 \times 10^{-2}$	$\pm 5.769 \times 10^{-3}$	
e	100	-11.216	60.698	-2.344	2.753×10^{-2}	1056.12	-4.713	3.634
			± 0.338	± 0.139	$\pm 7.230 \times 10^{-5}$	$\pm 6.280 \times 10^{-2}$	$\pm 6.031 \times 10^{-2}$	

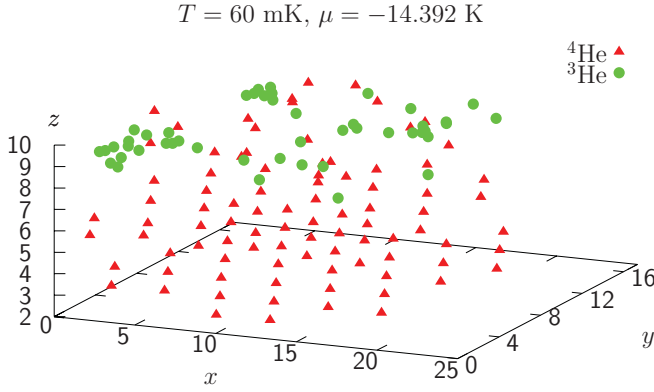


FIG. 1. (Color online) WAPIMC spatial positions of the atoms for a system of ${}^3\text{He}$ placed on the surface of a ${}^4\text{He}$ layer adsorbed on a graphite substrate at $T = 60$ mK. The total number of particles is $\langle N \rangle = 127.24 \pm 7.48$ and the number of ${}^3\text{He}$ atoms is $\langle N_3 \rangle = 40.576 \pm 5.421$. The chemical potential is fixed at $\mu = -14.392$ K. Solid red triangles, ${}^4\text{He}$ atoms; solid green circles, ${}^3\text{He}$ atoms. x , y , and z are in units of Å .

be reflected from the hard walls, and therefore may result in standing matter waves inside the ${}^3\text{He}$ - ${}^4\text{He}$ system.

Importantly, the number of bosons alone plays also an important role in the clusterization of the fermions. According to Guardiola and Navarro [23], the clusterization of ${}^3\text{He}$ atoms in an environment of ${}^4\text{He}$ atoms requires a certain number of bosons, N_B , in order to overcome the effects of zero-point motion and fermionic repulsion. It was expected [23] that a small number of bosons N_B and a large number of fermions N_F will cause instabilities in the mixture. Further, for only one ${}^4\text{He}$ atom a ${}^3\text{He}$ cluster could only be achieved if N_F was above 20. Next to this, bound states of ${}^3\text{He}$ - ${}^4\text{He}$ were anticipated for $20 < N_F < 30$. Peculiarly, the above authors found that two ${}^4\text{He}$ atoms cannot achieve a ${}^3\text{He}$ cluster if $3 < N_F < 17$.

B. Adequacy of statistical potential

Evidently, the SP used to describe the fermions seems to be effective, as dictated by the results in Sec. III A, particularly because it allows the ${}^3\text{He}$ atoms to float on the ${}^4\text{He}$ surface. The ${}^3\text{He}$ pair distances in Figs. 1–3 are $r \sim O(1 \text{ Å})$ and therefore much smaller than the de Broglie wavelength λ_{dB} for ${}^3\text{He}$

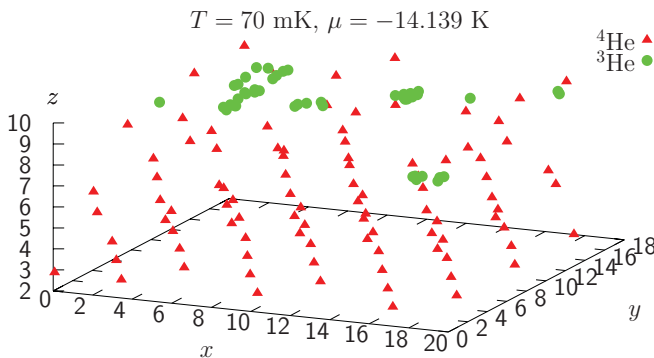


FIG. 2. (Color online) As in Fig. 1, but for $T = 70$ mK with $\mu = -14.139$ K, $\langle N \rangle = 125.55 \pm 7.07$, and $\langle N_3 \rangle = 34.308 \pm 5.766$. x , y , and z are in units of Å .

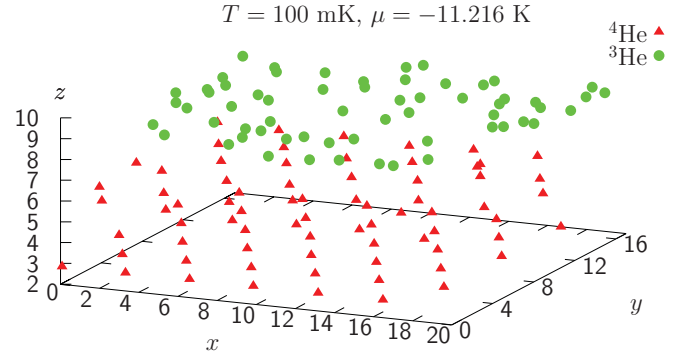


FIG. 3. (Color online) As in Fig. 1, but for $T = 100$ mK with $\mu = -11.216$ K, $\langle N \rangle = 133.25 \pm 0.614$, and $\langle N_3 \rangle = 60.698 \pm 0.338$. x , y , and z are in units of Å .

atoms, where λ_{dB} lies between 41.51 Å at $T = 60$ mK and 31.87 Å at $T = 100$ mK. The Fermi-Dirac statistical potential (1) goes to zero as r/λ_{dB} becomes larger than 1, and in our case $r/\lambda_{dB} \ll 1$ for ${}^3\text{He}$ pairs. Thus the SP is adequate for the description of the ${}^3\text{He}$ atoms in our systems. Further, the “range” of the SP in the above temperature range is significant, approximately equal to λ_{dB} for the latter temperatures. As the temperature is increased, the range of the SP decreases, and its repulsion becomes stronger. Essentially, then, spatial pair correlations between the ${}^3\text{He}$ atoms are significant.

C. Thermal properties

Next, we present some numerical WAPIMC results for the thermal properties of this type of system. As such, Table I displays results for the whole system, i.e., ${}^3\text{He}$ and ${}^4\text{He}$ together on graphite. From left to right the table lists a system label, the temperature T in mK, renormalized chemical potential μ , total number of particles $\langle N \rangle$, total energy per particle $\langle E \rangle / \langle N \rangle$, pressure per particle $\langle P \rangle / \langle N \rangle$, entropy per particle $\langle S \rangle / \langle N \rangle$ (unitless), internal energy per particle $\langle U \rangle / \langle N \rangle$, and superfluid fraction ρ_s / ρ (which is for ${}^4\text{He}$ only). The superfluid fraction was computed using the winding number approach according to formula (3.31) in the excellent review of Ceperley [26]:

$$\frac{\rho_s}{\rho} = \frac{\langle \mathbf{W}^2 \rangle}{2\lambda\beta N}, \quad (2)$$

ρ_s and ρ being the superfluid and total density, respectively, $\lambda = \hbar^2 / (2m)$, $\beta = M\tau$, and N the number of particles. \mathbf{W} is the winding number given by

$$\mathbf{W} = \sum_{i=1}^N \sum_{j=1}^M (\mathbf{r}_{i,j-1} - \mathbf{r}_{i,j}), \quad (3)$$

where $\mathbf{r}_{i,j}$ is the position of the i th particle at the j th time slice. In essence, we exclude the ${}^3\text{He}$ atoms in the evaluation of \mathbf{W} . Thus, the superfluid density would correspond to the fraction of ${}^4\text{He}$ atoms which are able to flow without friction in the higher layers away from the graphite surface. The ${}^4\text{He}$ layers close to the graphite surface are in the solid phase at low temperatures. Each thermal property $\langle \dots \rangle$ indicated is the WAPIMC thermodynamic average in units of K, except for the entropy and number of particles which are unitless. Table II is the same as Table I, except that it displays the ${}^3\text{He}$ statistics

separately, indicated by a subscript “3” in each thermal property, with the kinetic energy per particle $\langle K_3 \rangle / \langle N \rangle$ in the last column instead of ρ_s / ρ . We emphasize that the properties in Table II include essentially the effects of the ^4He as well as the graphite potential environment. There are five systems at temperatures ranging from $T = 60$ to 100 mK, labeled from a to e , respectively. For 60 and 70 mK, the same systems of Figs. 1 and 2 are considered, labeled a and b , respectively. The energies $\langle E \rangle / \langle N \rangle$ of the systems are negative, signaling that they have reached stable configurations. The energy of the ^3He component $\langle E_3 \rangle / \langle N \rangle$ (for systems a – e) is also negative, revealing that it may have formed bound states with the ^4He [5]. Further, our ^3He energies are in line with those obtained by Gasparini *et al.* [5] for ^3He on a ^4He film of 13 \AA thickness. Being far away from the graphite surface, our ^3He atoms have a much smaller energy per particle $|\langle E_3 \rangle / \langle N \rangle|$ than the ^4He component ($|\langle E \rangle - \langle E_3 \rangle| / \langle N \rangle$). The ^3He energy has a large kinetic component $\langle K_3 \rangle / \langle N \rangle$, or quantum pressure, as compared to its internal energy $\langle U_3 \rangle / \langle N \rangle$, but it is small compared to the total energy of the system. By going to higher temperatures, such as $T \geq 80$ mK, $\langle S_3 \rangle / \langle N \rangle$ rises substantially as compared to $\langle S \rangle / \langle N \rangle$. For example, it may be easily verified that $\langle S_3 \rangle / \langle S \rangle$ is ~ 80 and 70% for d and e ($T > 80$ mK), respectively, whereas it is less than 10% for a and b ($T < 80$ mK). Therefore, there is a lower order associated with ^3He at $T > 80$ mK. The small values of ρ_s / ρ reveal that the ^3He atoms deplete the superfluid substantially. The pressure in Table I remains of order 10^{-2} K for all systems, and is largest for system c which has the largest chemical potential $|\mu|$. In Table II it is of orders 10^{-5} and 10^{-4} K for a and b , respectively, and for c – e it is order 10^{-2} K. However, for a and b the error bar is very large. An interesting WAPIMC calculation to pursue in the future would be for a system of ^3He atoms placed on a ^4He surface without a graphite substrate.

From the tables, one can compute the same quantities separately for ^4He and ^3He , such as the energy of ^4He per ^4He particle, $\langle E_4 \rangle / \langle N_4 \rangle$, or ^3He per ^3He particle, $\langle E_3 \rangle / \langle N_3 \rangle$. The values have been listed in the tables found in the Supplementary Material [41]. However, the error bars need to be carefully computed using common rules for the different mathematical operations on these error bars. Typical values obtained are $\langle E_4 \rangle / \langle N_4 \rangle = (-67.381 \pm 10.589) \text{ K}$ and $\langle E_3 \rangle / \langle N_3 \rangle = (-6.570 \pm 3.605) \text{ K}$ at $T = 60$ mK. Particularly, the binding energy of ^3He , $\langle U_3 \rangle / \langle N_3 \rangle$, is $\sim O(-10) \text{ K}$.

D. Pair correlations

The pair correlation function $g(r)$ provides further information about the structure of the systems, where r is the distance between a pair of particles. Figure 4 displays $g(r)$ for the systems a (red crosses), b (open green squares), and d (solid blue squares) listed in Table I. For each temperature, one observes four peaks: a major one at $r \sim 1 \text{ \AA}$, and consecutively at ~ 3 , ~ 5.9 , and $\sim 8.5 \text{ \AA}$, respectively. By referring to the structure in Figs. 1 and 2, one can understand that the first peak in Fig. 4 is for the clusterized ^3He atoms, since these display the smallest interparticle distances. The rest of the peaks are largely associated with the correlations between the ^4He atoms in the layers $z \sim 3 \text{ \AA}$ and $z \sim 6 \text{ \AA}$. That is, the peak at $r \sim 3 \text{ \AA}$ is for ^4He particles surrounded by their

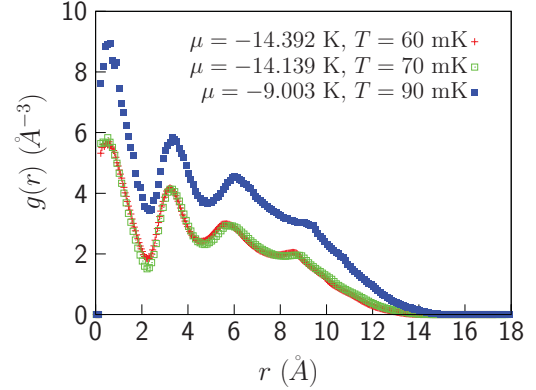


FIG. 4. (Color online) Correlation function $g(r)$ for systems a (red crosses), b (open green squares), and d (solid blue squares) listed in Table I.

first nearest neighbors, the one at $r \sim 5.9 \text{ \AA}$ for ^4He atoms and their second nearest neighbors, etc. At 90 mK, the correlations are stronger than at the other temperatures, because system d has the largest number of helium atoms among the systems displayed in the tables. The $g(r)$ for $T = 60$ and 70 mK are almost identical with little differences.

IV. CONCLUSION

In summary, we have demonstrated the effectiveness of a repulsive statistical potential (SP) in the description of fermions in a WAPIMC simulation of a few ^3He atoms floating on a ^4He layer adsorbed on a graphite substrate. The use of this SP yields results in agreement with earlier experimental observations, where the ^3He component preferentially floats on the outer surface of the ^4He layer [2, 14], and occupies surface states [6]. Further, the SP allows the clusterization of the ^3He component [2, 34, 35]. As one increases the temperature, the clusterization of ^3He ceases to exist as the ^3He – ^4He statistical repulsion becomes stronger, thereby overcoming ^3He – ^4He mixing. Eventually, the ^3He component separates into a floating phase above the ^4He surface. We additionally computed some thermal properties of these systems in the millikelvin temperature regime. The total energies of these systems turned out to be negative, indicating that these systems have reached their thermal equilibrium. The quantum pressure of the ^3He component turns out to be $\sim 10\%$ of the total energy, and the disorder in ^3He rises sharply beyond $T = 80$ mK. The superfluid component in ^4He is substantially depleted as a result of adding ^3He . Finally, our ^3He energies are in line with those reported earlier by Gasparini *et al.* [5].

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