

Curvature affects electrolyte relaxation: Studies of spherical and cylindrical electrodes

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With two minimal models, I study how electrode curvature affects the response of electrolytes to applied electrostatic potentials. For flat electrodes, Bazant *et al.* [*Phys. Rev. E* **70**, 021506 (2004)] popularized the RC timescale $\lambda_D L/D$, with λ_D being the Debye length, $2L$ the electrode separation, and D the ionic diffusivity. For thin electric double layers near concentric spherical and coaxial cylindrical electrodes, I show here that equivalent circuit models again predict the correct ionic relaxation timescales. Importantly, these timescales explicitly depend on both electrode radii, not simply on their difference.

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I. INTRODUCTION

Many functionalities in nature and technology rely on the out-of-equilibrium behavior of electrolytes. Transport of ions through nerve membranes, for example, underlies the firing of neurons [1]. Similarly, ionic fluxes in nanoporous carbon electrodes determine the power of supercapacitors [2] and the operation speed of capacitive deionization devices [3]. To optimize the performance of both capacitive devices through rational design, one needs a fundamental understanding of what sets the characteristic timescale τ of ionic response to electrode potentials. Since ions forming the electric double layer (EDL) must be partially drawn from a reservoir, τ could depend on “long” length scales like the electrode separation. This makes predicting τ with molecular simulations difficult, as typical simulation domains only capture a small portion of the nanoporous electrode structure [4] or rely on simplified geometries [5].

Theoretical predictions for τ typically concern either nontrivial electrode morphologies, treated approximately [6,7], or concern the simplest of geometries, i.e., electrolytes between parallel planar blocking electrodes. In the latter case, the ionic charge density reacts to small suddenly imposed (DC) electrostatic potential differences on the timescale $\lambda_D L/D$, which was derived with both microscopic and equivalent circuit model calculations [8,9]. The bulk diffusion timescale L^2/D can also appear, for instance, when large potentials are applied [8] or when the ionic diffusivities are unequal [10,11]. Different timescales even (with fractional powers of λ_D and L) appear when nonblocking electrodes are driven with an AC voltage [12]. The analytical parallel-plate results of Refs. [8–12] suggest that, to find τ for capacitive devices with complex nanoporous electrodes, one should identify relevant length scales and their relative importance. However, there are no general principles yet on how one should go about this task. Hence, it is timely to diminish the gap between analytical and molecular simulation predictions of electrolyte relaxation.

As a step towards an analytical understanding of the influence of nontrivial electrode morphology on ionic relaxation, in this article I discuss EDL capacitors with blocking concentric spherical or coaxial cylindrical electrodes (see Fig. 1). I use superscripts s and c throughout this article to specify observables to either geometry. For both systems, the electrodes have radii R_1 and R_2 ($\Delta R = R_2 - R_1 > 0$), respectively. The length ℓ of the cylinders is sufficiently large that I can ignore edge effects. While both systems then contain one relevant geometric length scale more than the parallel plate geometry (depending only on L), I will show that they allow for similar analytical insight. I assume spherical or axial symmetry in either case. Hence, all observables only depend on the radial distance r , with $R_1 \leq r \leq R_2$. In between the electrodes is a dilute 1:1 electrolyte of dielectric constant ϵ . The ionic charge density, the difference between cationic and anionic densities, vanishes throughout the cell initially. Application of a small dimensionless potential difference $\Delta\Phi \ll 1$ (with electrostatic potentials measured in units of the thermal voltage $k_B T/e$, with $k_B T$ being the thermal energy and e the proton charge) then drives the formation of EDLs at both electrode surfaces. Their equilibrium width is set by the Debye length $\lambda_D = \kappa^{-1}$.

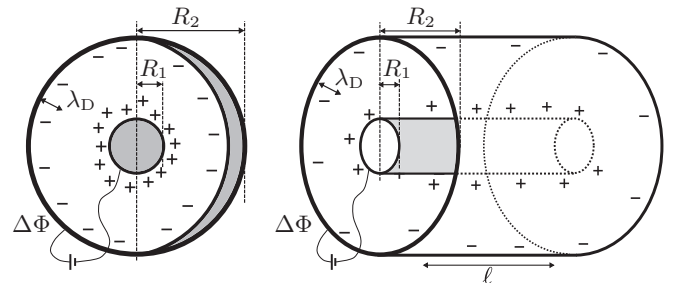


FIG. 1. Two model EDL capacitors consisting of 1:1 electrolytes with Debye length λ_D (solvent and surface charge not shown) between two concentric spherical or coaxial cylindrical electrodes of radii R_1 and R_2 . At time $t = 0$, a dimensionless potential difference $\Delta\Phi$ is applied between the electrodes.

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II. RC REASONING

Equivalent circuit representations of both setups in Fig. 1 contain two capacitors representing the EDLs at both electrode surfaces, a resistor for the electrolytic resistance, and a voltage source, all connected in series. In the spirit of Helmholtz, I treat the EDLs as dielectric capacitors of width λ_D and permittivity ε . Then, using that the capacitance of a dielectric capacitor of two conducting concentric spheres at r_1 and r_2 is $C^s = 4\pi\varepsilon/(1/r_1 - 1/r_2)$, I find the capacitance of the EDL at the inner electrode ($r_1 = R_1$, $r_2 = R_1 + \lambda_D$) as $C_{R_1}^s \approx 4\pi\varepsilon R_1^2/\lambda_D$, where I assumed $\lambda_D/R_1 \ll 1$. Likewise, the EDL at the outer electrode ($r_1 = R_2 - \lambda_D$, $r_2 = R_2$) has a capacitance $C_{R_2}^s \approx 4\pi\varepsilon R_2^2/\lambda_D$ if $\lambda_D/R_2 \ll 1$. The two in-series EDLs have a total capacitance

$$C^s = \frac{4\pi\varepsilon}{\lambda_D} \frac{1}{1/R_1^2 + 1/R_2^2}. \quad (1)$$

The resistance of the electrolyte is $R^s = \rho/4\pi(1/R_1 - 1/R_2)$, with $\rho = \lambda_D^2/\varepsilon D$ being its resistivity [8]. Multiplying $R^s = \lambda_D^2 \Delta R / 4\pi\varepsilon D R_1 R_2$ by C^s then yields the RC time

$$\tau_{RC}^s = \frac{\lambda_D R_2}{D} \frac{1 - R_1/R_2}{R_1/R_2 + R_2/R_1}. \quad (2)$$

For the cylindrical electrode system, starting from $C^c = 2\pi\epsilon\ell/\ln(r_2/r_1)$ and applying the same steps gives

$$C^c = \frac{2\pi\epsilon\ell}{\lambda_D} \frac{1}{1/R_1 + 1/R_2}. \quad (3)$$

With the resistance $R^c = \rho \ln(R_2/R_1)/2\pi\ell$, I now find

$$\tau_{RC}^c = \frac{\lambda_D R_2}{D} \frac{\ln(R_2/R_1)}{1 + R_2/R_1}. \quad (4)$$

When $R_1 \rightarrow R_2$, the electrodes locally resemble parallel plates and the relaxation times reduce to the familiar $\tau_{RC}^c \approx \tau_{RC}^s \approx \lambda_D \Delta R / 2D$. Conversely, for $R_1 \ll R_2$, $\tau_{RC}^s \approx R_1 \lambda_D / D$: The relaxation then only depends on the shortest geometric length scale. However, for general cases, τ_{RC}^s and τ_{RC}^c explicitly depend on both R_2 and R_1 .

III. MICROSCOPIC MODEL

A. Governing equations

The dimensionless electrostatic potential $\phi(r, t)$ is related to the dimensionless ionic charge density $q(r, t)$ via the Poisson equation

$$\frac{2}{r^d} \partial_r [r^d \partial_r \phi] = -\kappa^2 q, \quad (5)$$

where $d = 0$ for rectangular, $d = 1$ for cylindrical, and $d = 2$ for spherical coordinates. Moreover, $q(r, t)$ satisfies a continuity equation $\partial_t q = -\nabla \cdot \mathbf{J}_q$. At small potentials $\phi(r) \ll 1$, $\mathbf{J}_q = \hat{e}_r J_q$ with $J_q = -D(\partial_r q + 2\partial_r \phi)$ [8,9]. Inserting \mathbf{J}_q into the continuity equation and using Eq. (5) yields the Debye-Falkenhagen equation [13]

$$\frac{\partial_t q}{D} = -\frac{1}{r^d} \partial_r [r^d \partial_r q] + \kappa^2 q, \quad (6)$$

subject to

$$q(r, t = 0) = 0, \quad (7a)$$

$$\phi(R_2, t > 0) - \phi(R_1, t > 0) = \Delta\Phi, \quad (7b)$$

$$J_q(R_1, t) = J_q(R_2, t) = 0, \quad (7c)$$

which account for initial charge neutrality, the suddenly imposed potential difference, and the no-flux (blocking) boundary conditions.

B. Solution to Laplace-transformed Debye-Falkenhagen equation

I determine $q(r, t)$ as follows. With Laplace transformations, the partial differential equation for $q(r, t)$ [Eq. (6)] turns into a solvable ordinary differential equation [Eq. (9b)] for its Laplace transformed counterpart $\hat{q}(r, s) = \mathcal{L}\{q(r, t)\}$ [likewise $\hat{\phi}(r, s) = \mathcal{L}\{\phi(r, t)\}$]. Then $q(r, t)$ is determined through

$$q(r, t) = \sum_j \text{Res}(\hat{q}(r, s) \exp(st), s_j), \quad (8)$$

where Res stands for residue and with s_j being the poles of $\hat{q}(r, s)$, labeled with j .

Applying $\mathcal{L}\{\}$ on both sides of Eqs. (5)–(7), I find

$$\frac{2}{r^d} \partial_r [r^d \partial_r \hat{\phi}] = -\kappa^2 \hat{q}, \quad (9a)$$

$$\frac{1}{r^d} \partial_r [r^d \partial_r \hat{q}] = k^2 \hat{q}, \quad (9b)$$

with $k^2 = \kappa^2 + s/D$, subject to

$$\hat{\phi}(R_2, s) - \hat{\phi}(R_1, s) = \frac{\Delta\Phi}{s}, \quad (10a)$$

$$-\partial_r \hat{q} - 2\partial_r \hat{\phi}|_{r=\{R_1, R_2\}} = 0, \quad (10b)$$

where I used Eq. (7a) for Eq. (9b). The solution to Eq. (9b) reads $\hat{q}^s(r) = a_2 \exp[-kr]/r + b_2 \exp[kr]/r$ for $d = 2$ and $\hat{q}^c(r) = a_1 I_0(kr) + b_1 K_0(kr)$ for $d = 1$, with I_0 and K_0 being modified Bessel functions of the first and second kind, respectively. The constants a_1 , a_2 , b_1 , and b_2 could be fixed with the boundary conditions (10), which, however, inconveniently contain both \hat{q} and $\hat{\phi}$. Aiming at two constraints on \hat{q} only, I integrate Eq. (9a) over $\int_{R_1}^r dr$ and use Eq. (10b) to find

$$-2r^d \partial_r \hat{\phi} = R_1^d \partial_r \hat{q}(R_1) + \int_{R_1}^r dr r^d \kappa^2 \hat{q}, \quad (11)$$

with $\partial_r \hat{q}(R_1)$ shorthand for $\partial_r \hat{q}(r)|_{r=R_1}$.

Repeating the same calculation for $\int_{R_2}^r dr$ gives

$$-2r^d \partial_r \hat{\phi} = R_2^d \partial_r \hat{q}(R_2) + \int_{R_2}^r dr r^d \kappa^2 \hat{q}. \quad (12)$$

The difference and the sum (integrated over $\int_{R_1}^{R_2} dr$) of Eqs. (11) and (12) read

$$0 = R_1^d \partial_r \hat{q}(R_1) - R_2^d \partial_r \hat{q}(R_2) + \int_{R_1}^{R_2} dr r^d \kappa^2 \hat{q}, \quad (13a)$$

$$-\frac{4\Delta\Phi}{s} = \int_{R_1}^{R_2} \frac{dr}{r^d} \left[\int_{R_1}^r dr r^d \kappa^2 \hat{q} + \int_{R_2}^r dr r^d \kappa^2 \hat{q} \right] + [R_1^d \partial_r \hat{q}(R_1) + R_2^d \partial_r \hat{q}(R_2)] \int_{R_1}^{R_2} \frac{dr}{r^d}, \quad (13b)$$

which are two constraints on $\hat{q}^s(r, s)$ and $\hat{q}^c(r, s)$ each, which fix the constants a_1 , a_2 , b_1 , and b_2 therein.¹ For spherical electrodes, I find

$$\hat{q}^s \equiv \frac{2\Delta\Phi}{s} \frac{\Gamma^s}{\Upsilon^s}, \quad (14a)$$

$$\Gamma^s = \{m\xi \cosh[m(\xi - \bar{r})] - m \cosh[m(1 - \bar{r})] - \sinh[m(\xi - \bar{r})] + \sinh[m(1 - \bar{r})]\}/\bar{r}, \quad (14b)$$

$$\Upsilon^s = \frac{2n^2}{m} + m \left(2 - \frac{2n^2}{m^2} - \xi - \frac{1}{\xi} \right) \cosh[m(1 - \xi)] - \left(m^2 - n^2 - \frac{1}{\xi} \right) (1 - \xi) \sinh[m(1 - \xi)], \quad (14c)$$

while for cylindrical electrodes I find

$$\hat{q}^c \equiv \frac{2\Delta\Phi}{s} \frac{\Gamma^c}{\Upsilon^c}, \quad (15a)$$

$$\Gamma^c = m^2 \{ [\xi K_1(m\xi) - K_1(m)] I_0(m\bar{r}) + [\xi I_1(m\xi) - I_1(m)] K_0(m\bar{r}) \}, \quad (15b)$$

$$\Upsilon^c = m(m^2 - n^2) \xi \ln \xi [I_1(m) K_1(m\xi) - I_1(m\xi) K_1(m)] + \frac{2n^2}{m} - n^2 [\xi I_1(m\xi) K_0(m) + I_1(m) K_0(m\xi)] + \xi I_0(m) K_1(m\xi) + I_0(m\xi) K_1(m), \quad (15c)$$

where $m \equiv kR_2$, $n \equiv \kappa R_2$, $\xi \equiv R_1/R_2$, and $\bar{r} = r/R_2$. Here n measures the thickness of the EDLs relative to the system size. For most practical devices, the nanometer-sized EDLs are well separated, i.e., $n \gg 1$.

C. Equilibrium

The pole $s_0 \equiv 0$ in Eqs. (14a) and (15a) sets the equilibrium charge density through $q_{\text{eq}}(r) \equiv \text{Res}(\hat{q}(r, s), 0)$. This amounts to $q_{\text{eq}}^s(r) \equiv 2\Delta\Phi \Gamma_{\text{eq}}^s / \Upsilon_{\text{eq}}^s$, with

$$\Gamma_{\text{eq}}^s = \{n\xi \cosh[n(\xi - \bar{r})] - n \cosh[n(1 - \bar{r})] - \sinh[n(\xi - \bar{r})] + \sinh[n(1 - \bar{r})]\}/\bar{r}, \quad (16a)$$

$$\Upsilon_{\text{eq}}^s = \frac{1 - \xi}{\xi} \sinh[n(1 - \xi)] - n \frac{\xi^2 + 1}{\xi} \cosh[n(1 - \xi)] + 2n, \quad (16b)$$

and $q_{\text{eq}}^c(r) \equiv 2\Delta\Phi \Gamma_{\text{eq}}^c / \Upsilon_{\text{eq}}^c$, with

$$\Gamma_{\text{eq}}^c = [\xi K_1(n\xi) - K_1(n)] I_0(n\bar{r}) + [\xi I_1(n\xi) - I_1(n)] K_0(n\bar{r}), \quad (17a)$$

$$\Upsilon_{\text{eq}}^c = -\xi I_1(n\xi) K_0(n) - I_1(n) K_0(n\xi) - \xi I_0(n) K_1(n\xi) - I_0(n\xi) K_1(n) + \frac{2}{n}, \quad (17b)$$

respectively [i.e., $\Gamma_{\text{eq}} \equiv \Gamma(m = n)$ and $\Upsilon_{\text{eq}} \equiv \Upsilon(m = n)$]. Note that $q_{\text{eq}}^s(r)$ and $q_{\text{eq}}^c(r)$ can be derived more easily: At equilibrium, $J_q(r) = 0$ gives $q(r) = -2\phi(r)$, and thus $\partial_r[r^2 \partial_r \phi] = r^2 \kappa^2 \phi$ [Eq. (5)]. The solution to this equation,

$\phi^s(r) = A_2 \exp[-\kappa r]/r + B_2 \exp[\kappa r]/r$, contains two constants (A_2 and B_2), which are fixed with Eq. (7b) and particle conservation, $\int_{R_1}^{R_2} dr r^2 q = 0 \Rightarrow r^2 \partial_r \phi|_{R_1}^{R_2} = 0$. With $q(r) = -2\phi(r)$, $q_{\text{eq}}^s(r)$ then trivially follows. The same steps for $d = 1$ lead to $q_{\text{eq}}^c(r)$. Further details on the equilibrium EDL near curved electrodes can be found in Refs. [14–17] and in the Appendix, where I confirm Eqs. (1) and (3) with q_{eq}^s and q_{eq}^c , respectively.

D. Relaxation time

For the relaxation of $q(r, t)$, I need to determine the locations of the poles $s_j \in \mathbb{C}$ of $\hat{q}(r, s)$. However, instead of immediately focusing on s_j , Ref. [9] showed for the case of planar electrodes that it is easier to determine the corresponding poles $m_j \in \mathbb{C}$ of $\hat{q}(r, m)$ first. As in Ref. [9], each m_j that I could find was either purely real or purely imaginary (discussed below). Thus, all corresponding $s_j = (m_j^2 - n^2)D/R_2^2$ are real and, as it turns out, $s_{j \neq 0} < 0$. As I am interested in the late-time response of $q(r, t)$, I focus here on s_1 , the pole closest to s_0 , as this pole sets the late-time relaxation time $\tau_1 = -1/s_1$.

To find s_1 , I first note that neither $\Gamma^s(m)$ nor $\Gamma^c(m)$ has poles in $m \in \mathbb{C}$. Thus, all $s_{j \neq 0}$ come from the zeros of $\Upsilon^s(m)$ and $\Upsilon^c(m)$, respectively. Both $\Upsilon^s(m)$ and $\Upsilon^c(m)$ oscillate around zero on the imaginary m axis [Fig. 2(a)] and hence contribute to Eq. (8) with infinitely many poles. However, only the zero at the smallest m value has the potential of leading to s_1 ; all zeros further along the imaginary m axis

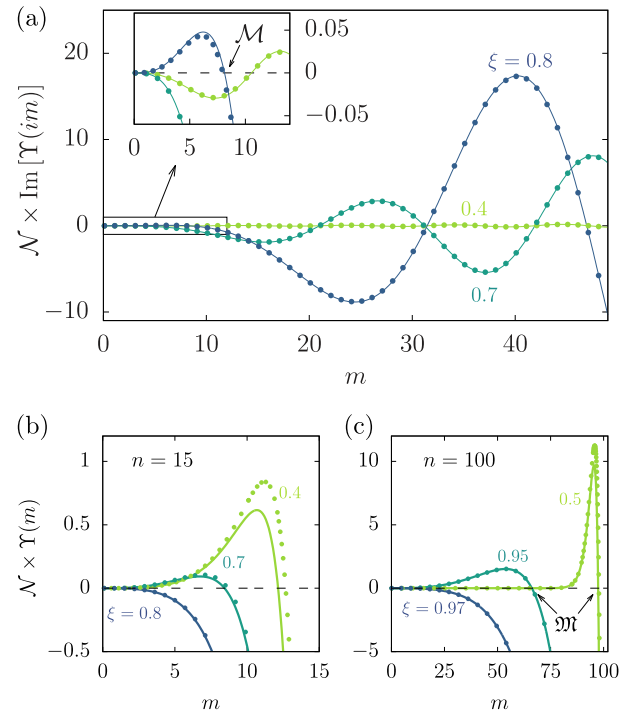


FIG. 2. Functions $\Upsilon^s(m)$ [Eq. (14c)] (lines) and $\Upsilon^c(m)$ [Eq. (15c)] (circles), both scaled with $\mathcal{N} \equiv \exp[n(\xi - 1)]$, for several ξ (see labels). (a) Plot of $\text{Im}(\Upsilon)$ along the imaginary m axis at $n = 15$. The inset zooms in near $m = 0$. Also shown is Υ along the real m axis for (b) $n = 15$ and (c) $n = 100$.

¹For $d = 0$, inserting $\hat{q}(r) = a_0 \exp[-\kappa r] + b_0 \exp[\kappa r]$ gives $\hat{q}(r)$, as reported in Refs. [8,9].

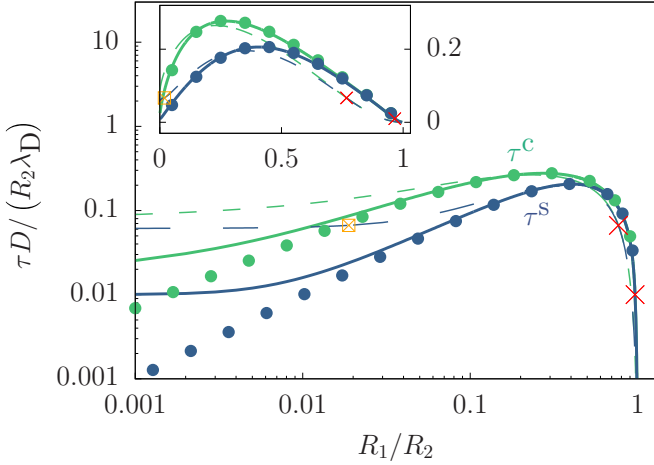


FIG. 3. Timescales τ_1^s (blue) and τ_1^c (green) at $\kappa R_2 = 15$ (dashed lines) and $\kappa R_2 = 100$ (solid lines) as obtained from numerically solving $\Upsilon^s(m) = 0$ [Eq. (14c)] and $\Upsilon^c(m) = 0$ [Eq. (15c)]. Also shown with symbols are τ_{RC}^s [Eq. (2)] and τ_{RC}^c [Eq. (4)]. The inset plots the same data against linear axes. Red crosses and orange boxed crosses indicate where m_1 changes from \mathfrak{M} to \mathcal{M} and vice versa, respectively.

give smaller s_j , hence faster decaying modes. Figures 2(b) and 2(c) show $\Upsilon(m)$ for $m \in \mathbb{R}$ at $n = 15$ and $n = 100$, respectively. For certain values of ξ , solutions $m_1 = \mathfrak{M}$ to $\Upsilon^s(m) = 0$ and $\Upsilon^c(m) = 0$ are visible there. For $n = 15$, \mathfrak{M} disappears at $\xi = 0.767$ (Υ^s) and $\xi = 0.768$ (Υ^c), respectively. Conversely, at $n = 100$, \mathfrak{M} persists until higher ξ , disappearing only at $\xi = 0.965$ (both Υ^s and Υ^c). In both cases, m_1 disappears through the origin (not shown), after which it reappears as a zero $m_1 = i\mathcal{M}$ (with $\mathcal{M} \in \mathbb{R}$) on the imaginary m axis [see the inset of Fig. 2(b)] that moves away from the origin with increasing ξ .² (This transition from \mathfrak{M} to \mathcal{M} occurs also at small ξ and n with decreasing ξ .) Associated with these zeros, s_1 is either $s_1 = (\mathfrak{M}^2 - n^2)D/R_2^2$ or $s_1 = -(\mathcal{M}^2 + n^2)D/R_2^2$.

Figure 3 shows $\tau_1 = -1/s_1$ for both setups at $n = \kappa R_2 = 15$ (dashed lines) and $n = 100$ (solid lines). At red crosses, m_1 transitions from \mathfrak{M} to $i\mathcal{M}$. The opposite transition (from \mathcal{M} to \mathfrak{M}), indicated with orange boxed crosses, does not occur at $n = 100$ and occurs for Υ^c only around $\xi = 10^{-12}$. The plateaus at small ξ are understood as follows. For $\xi \ll 1$, $\Upsilon^s = 0$ reduces to $\tanh m = m$, whose only solution $m_1 = 0$ gives $\tau D / R_2 \lambda_D \approx 1/n$. Also shown in Fig. 3 with circles are the equivalent circuit model predictions τ_{RC}^s [Eq. (2)] and τ_{RC}^c [Eq. (4)]. For $\xi > 0.1$ and $n = 15$, τ_1 and τ_{RC} are qualitatively similar. For larger n , τ_1 and τ_{RC} become identical. This is understood analytically as follows. For $m \in \mathbb{R}$ and $n \gg 1$,

$\Upsilon^s(m) = 0$ reduces to

$$2\left(m - \frac{n^2}{m}\right) = \left(m^2 - n^2 - \frac{1}{\xi}\right)(1 - \xi) + m\left(\xi + \frac{1}{\xi}\right) + O\left\{\frac{n^2}{m} \exp[m(\xi - 1)]\right\}. \quad (18)$$

From Fig. 2(c) I see that $\mathfrak{M} \approx n$ if $n \gg 1$. Inserting the approximation $\mathfrak{M}_{\text{ap}} = n - \delta$ into Eq. (18) and keeping terms up to $O(\delta)$, I find $\delta = (\xi + 1/\xi)/2(1 - \xi)$. This reproduces Eq. (2): $\tau_1^s \approx R_2 \lambda_D (1 - \xi)/D(\xi + 1/\xi)$.

Similarly, for $n \gg 1$, $\Upsilon^c(m) = 0$ amounts to $m(m^2 - n^2)\xi \ln \xi = n^2(1 + \xi) + O(\exp[2n(\xi - 1)])$. [I used Hankel's large argument expansions $I_\alpha(z) \sim \exp(z)/\sqrt{2\pi z}$ and $K_\alpha(z) \sim \sqrt{\pi} \exp(-z)/\sqrt{2z}$ here, which imply $I_\alpha(n) \gg I_\alpha(n\xi)$, $K_\alpha(\xi n) \gg K_\alpha(n)$, $I_1(n)/I_0(n) \rightarrow 1$, and $K_1(n)/K_0(n) \rightarrow 1$.] Inserting $\mathfrak{M}_{\text{ap}} = n - \epsilon$ and keeping terms up to $O(\epsilon)$ yields $\epsilon = -(1 + \xi)/(2\xi \ln \xi)$. This gives $\tau_1^c \approx -R_2 \lambda_D \xi \ln \xi / D(1 + \xi)$, i.e., Eq. (4).

E. Relaxation of $q(r, t)$

I use that, close to s_1 , $\Upsilon(s) \stackrel{s \rightarrow s_1}{\approx} \partial_m \Upsilon(m_1)[(s - s_1)R_2^2/2mD]$. The slowest relaxation mode $q_1(r, t) \equiv \text{Res}(\hat{q}(r, s) \exp(st), s_1)$ now amounts to

$$q_1(r, t) = 4\Delta\Phi \frac{m_1}{m_1^2 - n^2} \frac{\Gamma(m_1)}{\partial_m \Upsilon(m_1)} \exp[-t/\tau_1]. \quad (19)$$

Explicit expressions for $\partial_m \Upsilon(m)$ for the respective geometries follow with Eqs. (14c) and (15c). I find

$$\begin{aligned} \partial_m \Upsilon^s(m) = m(\xi - 1) & \left[\frac{1}{\xi}(\xi^2 + 1) + \frac{2n^2}{m^2} \right] \sinh[m(1 - \xi)] \\ & + \left[\frac{2n^2}{m^2} - (\xi - 1)^2(m^2 - n^2) \right] \cosh[m(1 - \xi)] \\ & - \frac{2n^2}{m^2} \end{aligned} \quad (20)$$

and

$$\begin{aligned} \partial_m \Upsilon^c(m) = (3m^2 - n^2)\xi \ln(\xi) & [I_1(m)K_1(m\xi) - K_1(m)I_1(m\xi)] \\ & + m(m^2 - n^2)\xi \ln(\xi) [I_0(m)K_1(m\xi) \\ & + K_2(m)I_1(m\xi) - \xi K_1(m)I_0(m\xi) - \xi I_1(m)K_2(m\xi)] \\ & + \frac{n^2 \xi^2}{2} \{ [K_0(m\xi) + K_2(m\xi)]I_0(m) \\ & - [I_0(m\xi) + I_2(m\xi)]K_0(m) \} \\ & + \frac{n^2}{2} \{ [K_0(m) + K_2(m)]I_0(m\xi) \\ & - [I_0(m) + I_2(m)]K_0(m\xi) \} - \frac{2n^2}{m^2}. \end{aligned} \quad (21)$$

Truncating the sum in Eq. (8) after $j = 1$, I approximate the relaxation of $q(r, t)$ by

$$q_{\text{ap}}^s(r, t) = q_{\text{eq}}^s(r) + q_1^s(r, t), \quad (22a)$$

$$q_{\text{ap}}^c(r, t) = q_{\text{eq}}^c(r) + q_1^c(r, t). \quad (22b)$$

In Fig. 4, I compare $q_{\text{ap}}(r, t)$ to numerical inversions of Eqs. (14) and (15) with the 't Hoog algorithm, respectively.

²A similar transition from a solution \mathfrak{M} to \mathcal{M} occurs for planar electrodes [9]. In that simpler geometry (without ξ), the transition happens at $n = \sqrt{3}$ always.

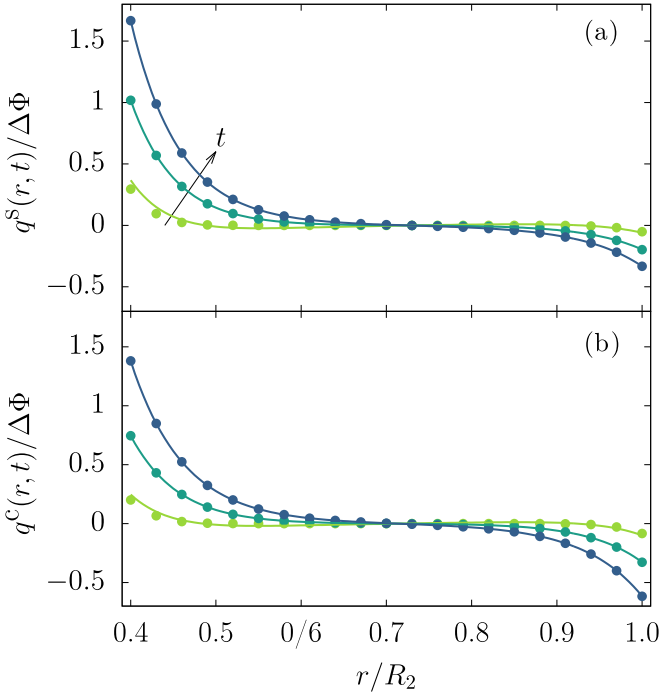


FIG. 4. Ionic charge densities (a) $q^s(r, t)$ and (b) $q^c(r, t)$, for $\kappa R_2 = 15$ and $R_1/R_2 = 0.4$ and at times $t D/R_2^2 = \{0.001, 0.01, 0.1\}$. Shown are analytical approximations $q_{\text{ap}}(r, t)$ [Eq. (22)] (lines) and numerical inversions of Eqs. (14) and (15) (circles).

I observe a stronger asymmetry in q^s than in q^c , which must stem from the difference in their ratios of inner to outer electrode surface areas (ξ^2 and ξ in either case, respectively). Note that, at $n = 15$ as used here, the numerically determined $\mathfrak{M}^s = 12.15$ and $\mathfrak{M}^c = 12.67$ deviate substantially from their analytical approximations $\mathfrak{M}_{\text{ap}}^s = 12.58$ and $\mathfrak{M}_{\text{ap}}^c = 13.10$, respectively; hence, I use the former. Clearly, $q_{\text{ap}}^s(r, t)$ describes $q^s(r, t)$ well for $t \approx \tau_1^s = 0.0129 R_2^2/D$, while deviations are visible at early times. The same is true for $q_{\text{ap}}^c(r, t)$ and $q^c(r, t)$ for times around $\tau_1^c D/R_2^2 = 0.0155$. Better capturing $q(r, t)$ at early times requires truncating the sum in Eq. (8) at higher j , which I leave for future work.

Finally, as I focused on electrolytes with equal ionic diffusivities $D_+ = D_-$ in this article, the results derived here can be expected to be accurate for, for instance, KCl, RbBr, and CsBr (which have $D_+/D_- = 0.97, 1.00$, and 1.00 , respectively [18]), but less so for NaCl ($D_+/D_- = 0.66$). Electrolytes with different ionic diffusivities will probably relax on two timescales: A fast RC timescale as described here and a slower diffusive timescale that becomes more important the more D_+ and D_- differ [10,11,19]. Finding the precise functional form of this diffusive timescale can be done with calculations along the lines of the ones presented in Refs. [11,19].

IV. CONCLUSION

I have studied the influence of electrode morphology on the relaxation of EDL capacitors, both with equivalent circuit models and with the (microscopic) Debye-Falkenhagen equation. For two different curved-electrode geometries, I have shown that the timescale of ionic response to an applied elec-

trostatic potential explicitly depends on all geometric length scales present. The uplifting message is that, for thin EDLs (a case of high practical relevance), easily obtainable RC times capture the ionic relaxation times decently. Conversely, for thick EDLs, corrections must be taken into account. These results form a small step towards an analytical understanding of the relaxation of supercapacitors and deionization devices; however, I expect complications at each further step of the way from planar to nanoporous electrodes.

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APPENDIX: EQUILIBRIUM SURFACE CHARGE AND CAPACITANCE

The total surface charge Q_1^s of the smaller electrode is related to its unit surface charge density σ_1^s by $Q_1^s = 4\pi R_1^2 e \sigma_1^s$. With Gauss's law $e^2 \sigma_1^s = -\epsilon k_B T \partial_r \phi(R_1)$ and $\phi(r) = -q(r)/2$ I find $Q_1 e = 2\pi R_1^2 \epsilon k_B T \partial_r q_1^s(R_1)$. Likewise, the charge on the larger electrode reads $Q_2^s e = -2\pi R_2^2 \epsilon k_B T \partial_r q^s(R_2)$. With Eq. (16), it is now straightforward to verify that the electrodes carry opposite surface charge $Q_1^s = -Q_2^s$. Likewise, the capacitance $C = e Q_2 / k_B T \Delta \Phi$ reads

$$\begin{aligned} \frac{C^s}{4\pi \epsilon R_2} &= -\frac{1}{\Upsilon_{\text{eq}}^s} \partial_r \Gamma_{\text{eq}}^s \Big|_{r=R_2} \\ &= \frac{1}{\Upsilon_{\text{eq}}^s} \{ (1 - n^2 \xi) \sinh[n(1 - \xi)] \\ &\quad + n(\xi - 1) \cosh[n(1 - \xi)] \} \\ &\stackrel{n \gg 1}{\approx} \frac{n\xi}{\xi + 1/\xi} + O(n^0). \end{aligned} \quad (\text{A1})$$

For the cylindrical electrode system, the surface charge $Q_1^c = 2\pi R_1 \ell e \sigma_1^c$ of the inner electrode amounts to $Q_1^c e = \pi R_1 \ell \epsilon k_B T \partial_r q(R_1)$. Similarly, $Q_2^c e = -\pi R_2 \ell \epsilon k_B T \partial_r q(R_2)$. Again, with Eq. (17), $Q_1^c = -Q_2^c$ can be shown to hold. The capacitance reads

$$\begin{aligned} \frac{C^c}{2\pi \epsilon \ell} &= -\frac{1}{\Upsilon_{\text{eq}}^c} \partial_r \Gamma_{\text{eq}}^c \Big|_{r=R_2} \\ &= \frac{n\xi}{\Upsilon_{\text{eq}}^c} [K_1(n) I_1(n\xi) - I_1(n) K_1(n\xi)] \\ &\stackrel{n \gg 1}{\approx} \frac{n\xi}{1 + \xi} + O(\exp[2n(\xi - 1)]), \end{aligned} \quad (\text{A2})$$

where, going to the third line, I again used Hankel's large argument expansion, stated below Eq. (18). Note that Eqs. (A1) and (A2) are equivalent to Eqs. (1) and (3), respectively.

- [1] A. L. Hodgkin and A. F. Huxley, *J. Physiol.* **117**, 500 (1952).
- [2] F. Béguin, V. Presser, A. Balducci, and E. Frackowiak, *Adv. Mater.* **26**, 2219 (2014).
- [3] M. E. Suss, S. Porada, X. Sun, P. M. Biesheuvel, J. Yoon, and V. Presser, *Energy Environ. Sci.* **8**, 2296 (2015).
- [4] C. Péan, C. Merlet, B. Rotenberg, P. A. Madden, P.-L. Taberna, B. Daffos, M. Salanne, and P. Simon, *ACS Nano* **8**, 1576 (2014).
- [5] K. Breitsprecher, C. Holm, and S. Kondrat, *ACS Nano* **12**, 9733 (2018).
- [6] R. de Levie, *Electrochim. Acta* **8**, 751 (1963).
- [7] P. M. Biesheuvel and M. Z. Bazant, *Phys. Rev. E* **81**, 031502 (2010).
- [8] M. Z. Bazant, K. Thornton, and A. Ajdari, *Phys. Rev. E* **70**, 021506 (2004).
- [9] M. Janssen and M. Bier, *Phys. Rev. E* **97**, 052616 (2018).
- [10] A. L. Alexe-Ionescu, G. Barbero, I. Lelidis, and M. Scalerandi, *J. Phys. Chem. B* **111**, 13287 (2007).
- [11] B. Balu and A. S. Khair, *Soft Matter* **14**, 8267 (2018).
- [12] I. Rubinstein, B. Zaltzman, A. Futerman, V. Gitis, and V. Nikonenko, *Phys. Rev. E* **79**, 021506 (2009).
- [13] P. Debye and H. Falkenhagen, *Phys. Z.* **29**, 121 (1928).
- [14] C. Lian, D.-e. Jiang, H. Liu, and J. Wu, *J. Phys. Chem. C* **120**, 8704 (2016).
- [15] A. Reindl, M. Bier, and S. Dietrich, *J. Chem. Phys.* **146**, 154703 (2017); **146**, 154704 (2017).
- [16] G. V. Bossa, R. Downing, J. Abrams, B. K. Berntson, and S. May, *J. Phys. Chem. C* **123**, 1127 (2019).
- [17] A. J. Asta, I. Palaia, E. Trizac, M. Levesque, and B. Rotenberg, *J. Chem. Phys.* **151**, 114104 (2019).
- [18] J. N. Agar, C. Y. Mou, and J. L. Lin, *J. Phys. Chem.* **93**, 2079 (1989).
- [19] M. Janssen and M. Bier, *Phys. Rev. E* **99**, 042136 (2019).
- [20] A. C. L. de Lichtervelde, J. P. de Souza, and M. Z. Bazant, [arXiv:1908.03223](https://arxiv.org/abs/1908.03223).