

Investigations of the synthesis of the superheavy element $Z = 122$

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We have identified the most probable projectile-target combination by studying the fusion cross section, evaporation residue cross section, compound nucleus formation probability (P_{CN}), and survival probability (P_{Surv}) of different projectile target combinations to synthesize the superheavy element $Z = 122$. The selected most probable projectile-target combinations to synthesize the superheavy element $Z = 122$ are Cr + Cf, Fe + Cm, Se + Ra, and As + Ac. Superheavy nuclei may decay through the different decay modes such as spontaneous fission, ternary fission, and cluster decay. We have also studied the half-lives of spontaneous fission, ternary fission, and cluster decay of the predicted nuclei for $Z = 122$ and compared with that of alpha decay. This enables us to study the competition between spontaneous fission, ternary fission, cluster decay, and alpha decay in the superheavy nuclei of $Z = 122$. The comparison of half lives for different decay modes reveals that alpha decay is having smaller half lives than the other studied decay modes. A detail study of branching ratio of alpha decay with respect to other decay modes also confirms that alpha decay is most dominant decay mode for the isotopes superheavy nuclei $^{307-314}122$ and hence these nuclei can be detected through the alpha decay mode only. We hope that our predictions may be guide for the future experiments in the synthesis of more isotopes of superheavy nuclei $Z = 122$.

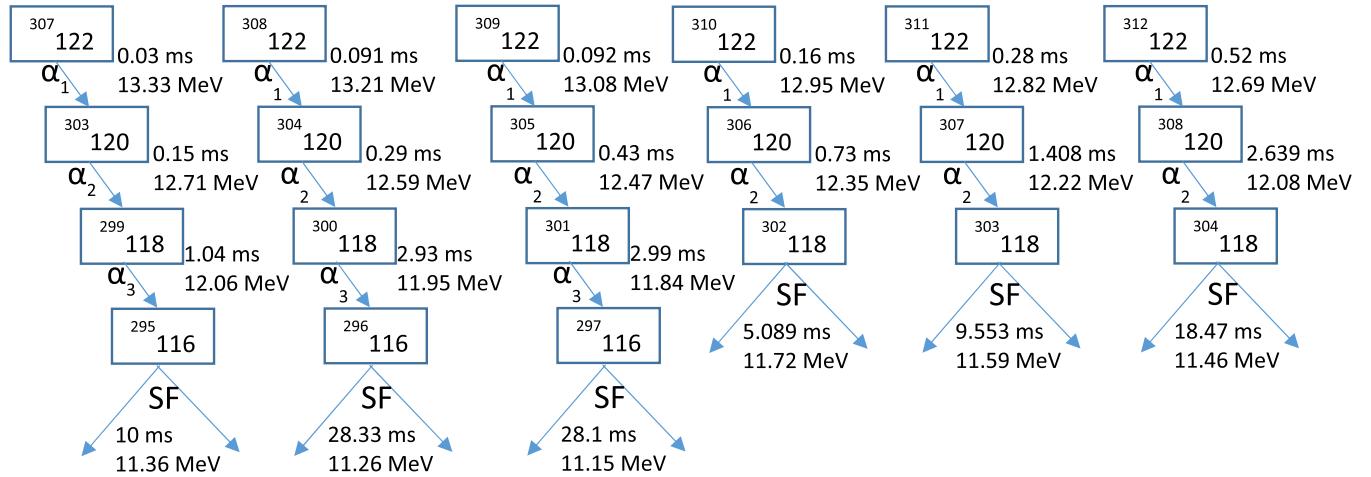
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I. INTRODUCTION

The study of superheavy elements has become the subject of strong interest both for experimental and theoretical research. Superheavy elements are formed by a fusion process in collisions of two heavy ions. The fusion is described in the literature with two different methods: as a melting of two nuclei along their relative coordinate or as a transfer of nucleons or clusters from the lighter nucleus to the heavier one. In this paper we describe the fusion as a melting process along the relative coordinate. The other method is based on the dinuclear system (DNS) concept (see Refs. [1,2]). In this model the formed DNS evolves as a diffusion process in the mass asymmetry coordinate $\eta = (A_1 - A_2)/(A_1 + A_2)$ to the compound nucleus. The DNS model assumes that the motion to smaller relative distances (at fixed η) is hindered in collisions of heavy nuclei [3]. The diffusion process can lead to the decay of the dinuclear system, which is quasifission. Torres *et al.* [4] studied the quasifission process in a transport model for a dinuclear system. The competition between fusion and quasifission reduces the cross section for the compound nucleus formation. The DNS model works well for the description of fusion in symmetric reactions with heavy nuclei and in reactions producing superheavy nuclei [5]. The dependence of the fusion cross section on the isotopic composition of colliding nuclei is studied using the dinuclear system concept for compound nucleus formation [6]. Adamian *et al.* [7]

studied the possibilities of the synthesis of superheavy nuclei in actinide based fusion reactions within the dinuclear system model for compound nucleus formation. Previous workers studied the influence of angular momentum on the competition between complete fusion and quasifission within the dinuclear system model [8]. The use of light- and medium-mass neutron-rich radioactive beams may produce superheavy nuclei. Such a possibility is also provided by the multinucleon transfer processes in low-energy damped collisions of heavy actinide nuclei, if the shell effects really play an important role in such reactions [9]. Multinucleon transfer reactions occurring in low-energy collisions of heavy ions are considered an important method for the production of superheavy elements [10]. The possibilities of using radioactive beams, multinucleon transfer reactions, and neutron capture processes for this purpose are discussed by previous workers [11–13]. Marinov *et al.* [14] obtained possible evidence for the existence of a long-lived superheavy nucleus with mass number $A = 292$ and atomic number $Z = 122$ or 124 in natural thorium. The discovery of many isotopes of new superheavy elements of $Z = 114, 115, 116, 118$ were reported in a review paper [15]. There have been some experimental studies reported on the investigations of superheavy nuclei (SHN) $Z = 122$ [16,17]. Some theoretical studies also observed on the predictions of SHN $Z = 122$ [18–20]. Previous theoretical studies on the evaporation residue (σ_{ER}) cross section for $Z = 122$ refer to a cold, near-symmetric reaction $^{154}\text{Sm}(^{150}\text{Nd}, 1n)^{303}122$ [18,19]. Also observed is the synthesis of the superheavy element $Z = 122$ through a hot fusion reaction $(^{58}\text{Fe} + ^{248}\text{Cm} \rightarrow ^{306-x}122 + xn)$ and $(^{64}\text{Ni} + ^{244}\text{Pu} \rightarrow ^{308-x}122 + xn)$ [20].

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FIG. 1. Decay chain of the predicted probable isotopes for $Z = 122$.

The predicted value of the $1n$ decay cross section for the cold fusion reaction $^{154}\text{Sm}(^{150}\text{Nd}, 1n)^{303}\text{122}$ by previous researchers [16] using the fusion-by-diffusion (FBD) model [21–23] is $\sigma_{1n} \sim 10^{-11}$ pb, whereas other authors [17] had predicted the same to be ~ 1 pb, an incredible 11 orders of magnitude higher, for their use of an old variant of the FBD model [21]. There is a discrepancy in the predictions of synthesis parameters of superheavy nuclei. The hot fusion reactions have been used as the tool to study the SHEs. Superheavy nuclei and their decay properties is one of the important fields in nuclear physics. Previous researchers [14] obtained the half-life $T_{1/2} > 10^8$ yr and an abundance $(1-10) \times 10^{-12}$ for a superheavy nucleus with mass number $A = 292$ and atomic number $Z = 122$. The possibility of such an extremely heavy Z nucleus motivated us to study the most probable projectile-target combinations to synthesize the superheavy nuclei $Z = 122$. Previous researchers also studied the possible projectile target combinations to synthesize the superheavy nuclei [22–28].

Manjunatha [26] studied the theoretical predictions on the possible isotopes of the superheavy element $Z = 122$ and predicted that the nuclei $^{307\text{--}314}\text{122}$ were found to have long half-lives and hence could be sufficient to detect them if synthesized in a laboratory. The decay chains of predicted nuclei for $Z = 122$ are shown in Fig. 1. The present work consists of two parts. In the first part we have identified the most probable projectile-target combination to synthesize these superheavy nuclei. We have identified the most probable projectile-target combination by studying the fusion cross section, evaporation residue cross section, compound nucleus formation probability (P_{CN}), and survival probability (P_{Surv}) of different projectile target combinations to synthesize the superheavy element $Z = 122$. Superheavy nuclei may decay through the different decay modes such as spontaneous fission, ternary fission, and cluster decay. There is a need to study the different decay modes such as spontaneous fission, ternary fission, and cluster decay of the predicted nuclei for $Z = 122$. Hence in the second part of this work, we have studied the half-lives of spontaneous fission, ternary fission, and cluster decay of this predicted nuclei for $Z = 122$ and compared them with that of alpha decay. This enables us to study the competition

between spontaneous fission, ternary fission, cluster decay, and alpha decay in the superheavy nuclei of $Z = 122$.

II. THEORETICAL FRAMEWORK

A. Projectile-target combinations to synthesize SHN $Z = 122$ via fusion

The interacting potential barrier for two spherical nuclei is given by

$$V = V_N(R) + V_C(R) + \frac{\hbar^2 l(l+1)}{2\mu \times r^2}, \quad (1)$$

where l represents the angular momentum and μ is the reduced mass. Coulomb potential $V_C(R)$ is calculated by

$$V_C(R) = Z_1 Z_2 e^2 \begin{cases} \frac{1}{R} & (R > R_C) \\ \frac{1}{2R_c} [3 - (\frac{R}{R_c})^2] & (R < R_C) \end{cases} \quad (2)$$

where $R_C = 1.24 \times (R_1 + R_2)$, R_1 and R_2 are respectively the radii of the emitted alpha and daughter nuclei. Here Z_1 and Z_2 are the atomic numbers of the daughter and emitted cluster.

The nuclear potential is calculated from the proximity potential [29] and it is given as

$$V_p(z) = 4\pi\gamma b \left[\frac{C_1 C_2}{C_1 + C_2} \right] \Phi\left(\frac{z}{b}\right) \quad (3)$$

with the nuclear surface tension coefficient,

$$\gamma = 0.9517[1 - 1.7826(N - Z)^2/A^2] \text{ MeV/fm}^2. \quad (4)$$

Here N , Z , and A represent the neutron, proton, and mass number of the parent and Φ represents the universal proximity potential [30] and the recent proximity function is

$$\Phi(\varepsilon) = \frac{p_1}{1 + \exp\left(\frac{s_0 + p_2}{p_3}\right)}. \quad (5)$$

Here p_1 , p_2 , and p_3 are -7.65 , 1.02 , and 0.89 , respectively [31], and s_0 is calculated by the equation

$$s_0 = (R - R_1 - R_2)/b. \quad (6)$$

The width (diffuseness) of the nuclear surface $b \approx 1$ and the Sümann central radii C_i of the fragments is related to the sharp

radii R_i as

$$C_i = R_i - \left(\frac{b^2}{R_i} \right). \quad (7)$$

For R_i , we use a semiempirical formula in terms of mass number A_i as [32]

$$R_i = 1.28A_i^{1/3} - 0.76 + 0.8A_i^{-1/3}. \quad (8)$$

After calculation of the total potential, fusion barriers are estimated. Since fusion happens at a distance larger than the touching configuration of the colliding pair, the above form of the Coulomb potential is justified. One can extract the barrier height V_B and barrier position R_B using the following conditions:

$$\left. \frac{dV(r)}{dr} \right|_{r=R_B} = 0 \quad \text{and} \quad \left. \frac{d^2V(r)}{dr^2} \right|_{r=R_B} \leq 0. \quad (9)$$

To study the fusion cross sections, we shall use the model given by Wong [33]. In this formalism, the cross section for complete fusion is given by

$$\sigma_{\text{fus}} = \frac{\pi\hbar^2}{2\mu \times E_{\text{cm}}} \sum_{l=0}^{l_{\text{max}}} (2l+1) \times T_l(E_{\text{cm}}) P_{\text{CN}}(E_{\text{cm}}, l), \quad (10)$$

where μ is the reduced mass. The center of mass energy is denoted by E_{cm} . In the above formula, l_{max} corresponds to the largest partial wave for which a pocket still exists in the interaction potential and $T_l(E_{\text{cm}})$ is the energy-dependent barrier penetration factor. P_{CN} is the probability for the compound nucleus (CN) formation by two nuclei coming in contact. The probability of compound nucleus formation P_{CN} suggested by previous workers [34–40] is used in the present calculation. The calculation of P_{CN} requires effective fissility which in turn depends on x_{thr} and c . x_{thr} and c are adjustable parameters [22–24]. These parameters were suggested by Loveland [41]. This form of energy dependence of fusion probability is similar to the one proposed by Zargrebeav and Greiner [41].

After the fusion of two nuclei, the corresponding compound nuclei come to the ground state by emitting neutrons. The evaporation residue cross section of SH element production in a heavy-ion fusion reaction with subsequent emission of x neutrons is given by [41]

$$\sigma_{ER}^{xn} = \frac{\pi}{k^2} \sum_{l=0}^{\infty} (2l+1) T(E, l) P_{\text{CN}}(E, l) P_{\text{sur}}^{xn}(E^*, l). \quad (11)$$

P_{sur} is the survival probability and it is the compound nucleus to decay to the ground state of the final residual nucleus via evaporation of neutrons/light particles. The survival probability is

the probability that the fused system emits several neutrons followed by observing a sequence of α decay from the residue. The survival probability under the evaporation of x neutrons is

$$P_{\text{sur}} = P_{\text{xn}}(E_{\text{CN}}^*) \prod_{i=1}^{i_{\text{max}}=x} \left(\frac{\Gamma_n}{\Gamma_n + \Gamma_f} \right)_{i, E^*}, \quad (12)$$

where the index “ i ” is equal to the number of emitted neutrons. The calculation of P_{sur} requires the probability of evaporation of x neutrons from the compound nucleus (P_{xn}). To calculate the P_{xn} , we have adopted the procedure explained by previous workers [39,40]. The term $[\Gamma_n/(\Gamma_n + \Gamma_f)]$ in Eq. (12) is calculated by the knowledge of the ratio of the emission width of a neutron to the fission width (Γ_n/Γ_f). In the present work, we have used the expression for Γ_n/Γ_f based on the level densities of the Fermi-gas model [40].

B. Competition between binary fission, ternary fission, cluster radioactivity, and alpha decay process in superheavy nuclei

The interacting potential between two nuclei of fission fragments is taken as the sum of the Coulomb potential and proximity potential. To study the ternary and binary fission, we have used Denisov nuclear potential $V_P(r)$ [42] and it is given by

$$\begin{aligned} V_P(r) = & -1.989\,843 \frac{R_1 R_2}{R_1 + R_2} \varphi(r - R_1 - R_2 - 2.65) \\ & \times \left[1 + 0.003\,525\,139 \left(\frac{A_1}{A_2} + \frac{A_2}{A_1} \right)^{3/2} \right. \\ & \left. - 0.411\,326\,3(I_1 + I_2) \right], \end{aligned} \quad (13)$$

where the effective nuclear radius is given by

$$\begin{aligned} R_i = & R_{ip} \left(1 - \frac{11.654\,15}{R_{ip}} \right) \\ & + 1.284\,589 \left(I_i - \frac{0.4A_i}{A_i + 200} \right) (i = 1, 2), \end{aligned} \quad (14)$$

where R_{ip} is given by

$$\begin{aligned} R_{ip} = & 1.24 A_i^{3/2} \left[1 + \frac{1.646}{A_i} - 0.191 \left(\frac{A_i - 2Z_i}{A_i} \right) \right] \quad \text{with} \\ I_i = & \frac{N_i - Z_i}{A_i}. \end{aligned} \quad (15)$$

The universal function $\varphi(s = r - R_1 - R_2 - 2.65)$ is given by

$$\Phi(\xi) = \begin{cases} 1 - s/0.788\,166\,3 + 1.229\,218s^2 - 0.223\,427\,7s^3 - 0.103\,876\,9s^4 \\ - \frac{R_1 R_2}{R_1 + R_2} (0.184\,493\,5s^2 + 0.075\,701\,01s^3) + (I_1 + I_2)(0.044\,706\,45s^2 + 0.033\,468\,70s^3) & \text{for } -5.65 \leq s \leq 0 \\ 1 - S^2 [0.054\,101\,06 \frac{R_1 R_2}{R_1 + R_2} \exp(-\frac{s}{1.760\,580})] \\ - 0.539\,542\,0(I_1 + I_2) \exp(-\frac{s}{2.424\,408}) \times \exp(-\frac{s}{0.788\,166\,3}) & \text{for } s \geq 0 \end{cases}.$$

To study the cluster decay and alpha decay, Coulomb potential $V_c(R)$ is taken as

$$V_c(R) = Z_1 Z_2 e^2 \begin{cases} \frac{1}{R} & (R > R_C) \\ \frac{1}{2R_c} [3 - (\frac{R}{R_c})^2] & (R < R_C), \end{cases} \quad (16)$$

where $R_C = 1.24 \times (R_1 + R_2)$, R_1 and R_2 are respectively the radii of the emitted alpha/cluster and daughter nuclei. Here Z_1 and Z_2 are the atomic numbers of the daughter and emitted cluster. We have used the proximity function defined specially for cluster/alpha decay and it is given as [30]

$$\Phi(\varepsilon) = \frac{p_1}{1 + \exp(\frac{s_0 + p_2}{p_3})} \quad \text{with} \quad s_0 = \frac{R - R_1 - R_2}{b}. \quad (17)$$

Here p_1 , p_2 , and p_3 are -7.65 , 1.02 , and 0.89 , respectively [31].

For all four processes such as spontaneous fission, alpha ternary fission, cluster decay, and alpha decay, the barrier penetrability P is given as

$$P = \exp \left\{ -\frac{2}{\hbar} \int_a^b \sqrt{2\mu(V - Q)} dz \right\}. \quad (18)$$

Here $\mu = mA_1 A_2 / A$, where m is the nucleon mass and A_1 , A_2 are the mass numbers of daughter and emitted clusters, respectively. For cluster/alpha decay, the turning points “ a ” and “ b ” are determined from the equation, $V(a) = V(b) = Q$. For the fission process, the first turning point is determined from the equation $V(a) = Q$ and the second turning point from $b = 0$. The above integral can be evaluated numerically or analytically, and the half-life time is given by

$$T_{1/2} = \frac{\ln 2}{\lambda} = \frac{\ln 2}{v P}, \quad (19)$$

where $v = \frac{\omega}{2\pi} = \frac{2E_v}{\hbar}$ represents the number of assaults on the barrier per second and λ is the decay constant. E_v , the empirical

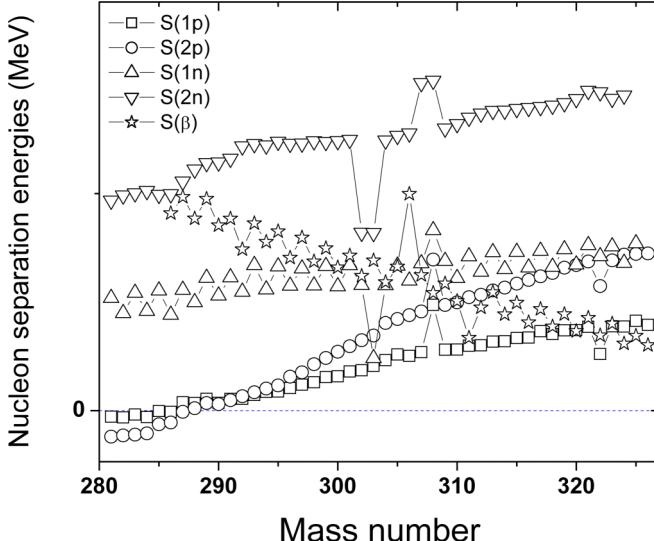


FIG. 2. Nucleon separation energies as a function of mass number for $Z = 122$.

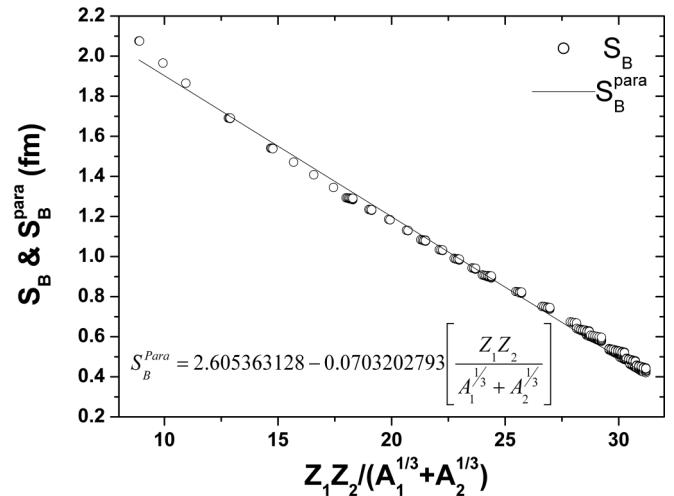


FIG. 3. Reduced fusion barrier positions S_B (fm) as a function of $\frac{Z_1 Z_2}{A_1^{1/3} + A_2^{1/3}}$.

vibration energy, is given as

$$E_v = Q \left\{ 0.056 + 0.039 \exp \left[\frac{4 - A_2}{2.5} \right] \right\} \quad \text{for} \quad A_2 \geq 4. \quad (20)$$

III. RESULTS AND DISCUSSION

In the previous study [26], it is observed that the nuclei $^{307-314}122$ were found to have long half-lives and hence it could be sufficient to detect them if synthesized in a laboratory. To check isotopes for the stability against the proton, neutron, and beta emission, we have calculated the corresponding separation energies. The calculated separation energies for different isotopes of superheavy nuclei $Z = 122$ are shown in Fig. 2. On calculating the separation energies for the isotopes of $Z = 122$, the two-proton separation energy $S(2p)$ is negative

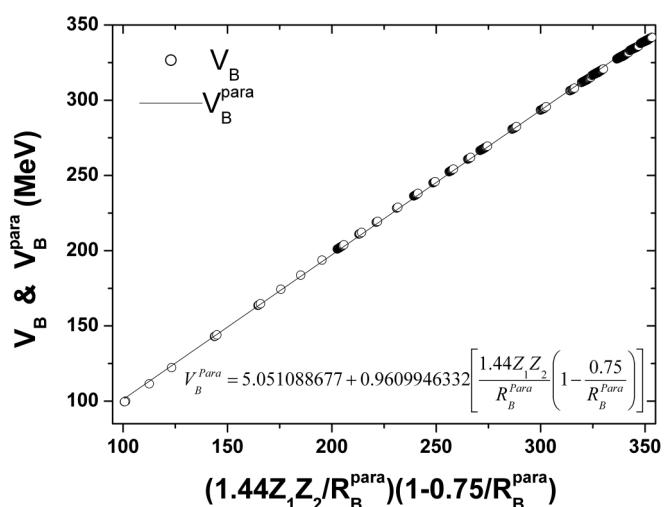
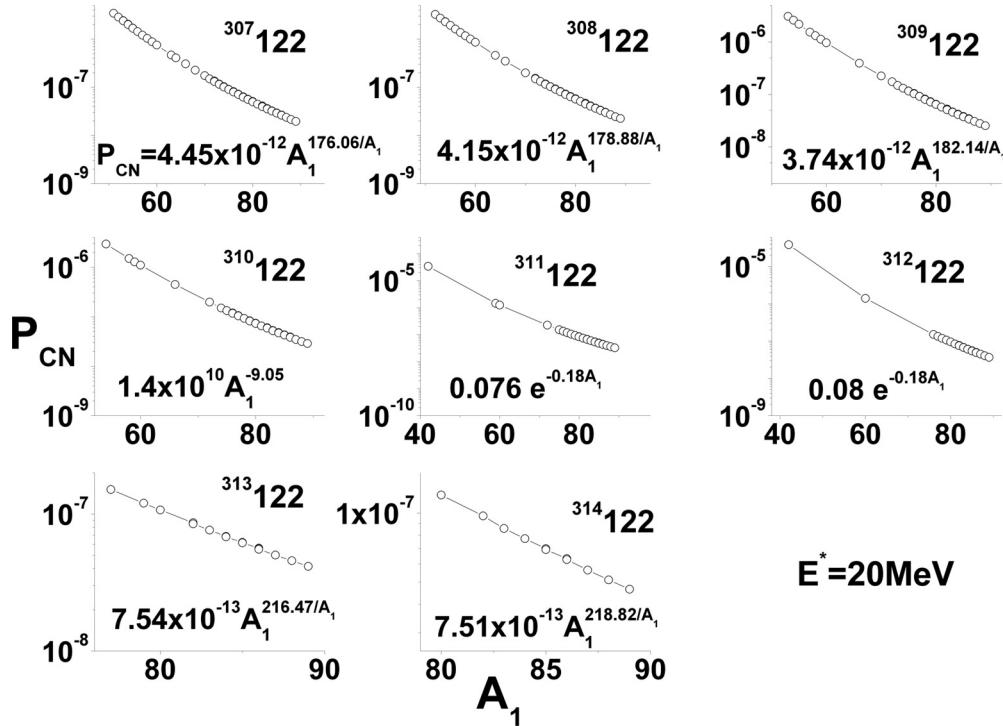
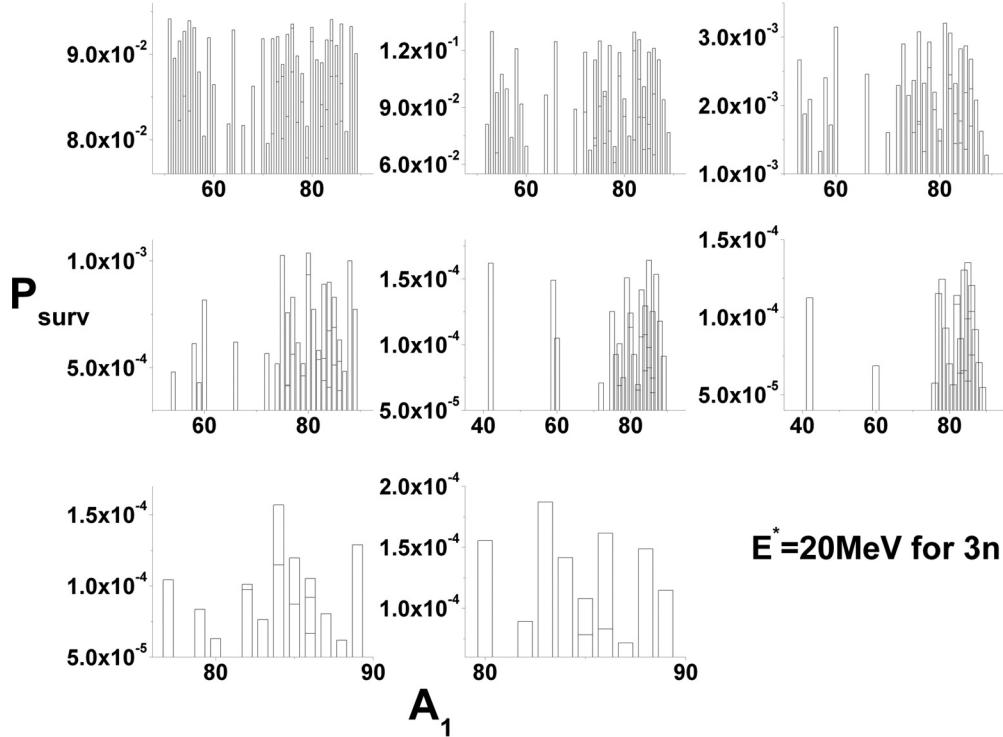


FIG. 4. Fusion barrier heights V_B (MeV) as a function of $\frac{Z_1 Z_2}{R_B^{\text{par}}} (1 - \frac{1}{R_B^{\text{par}}})$.

FIG. 5. Variation of compound nucleus probability (P_{CN}) at 20 MeV with mass number of projectile.

for those isotopes within the range $280 \leq A \leq 289$. These observations make it clear that all those isotopes within the range $280 \leq A \leq 289$ are outside the proton drip line and thus may easily decay through proton emission. Hence the nuclei $^{307-314}122$ may not undergo proton, neutron, and beta decay.

After identifying the most probable isotopes for the superheavy element $Z = 122$, we have studied the possible fusion reactions for their synthesis. We have studied around 900 possible projectile-target combinations to synthesize superheavy nuclei $^{307-314}122$. For all projectile-target combinations, we have

FIG. 6. Variation of survival probability (P_{Surv}) at 20 MeV (for $3n$ evaporation channel) with mass number of projectile.

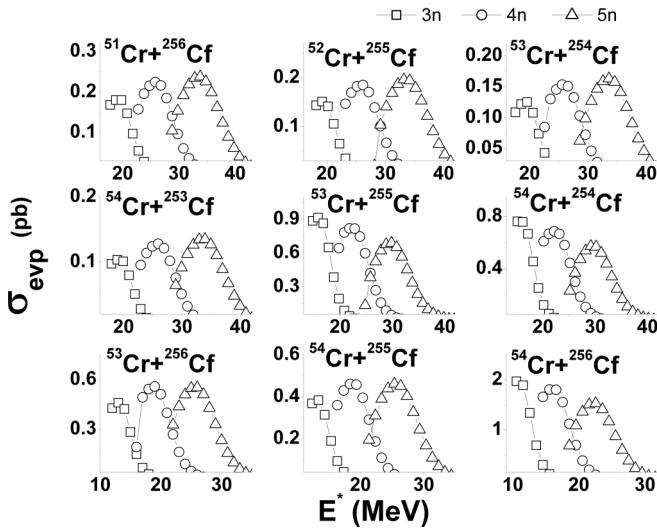


FIG. 7. Evaporation residue cross section as a function of E^* for projectile-target combination Cr + Cf.

calculated the fusion barrier heights (V_B) and positions (R_B). Once fusion barrier heights and positions were calculated, a search was made for their parametrization. We have calculated the reduced fusion barrier $S_B = R_B - C_1 - C_2$ and plotted reduced fusion barrier as a function of $Z_1 Z_2 / (A_1^{1/3} + A_2^{1/3})$ and it is shown in Fig. 3. We have fitted the function for the reduced fusion barrier in terms of $Z_1 Z_2 / (A_1^{1/3} + A_2^{1/3})$ as follows:

$$S_B^{\text{para}} = 2.6054 - 0.0703 \times \left[\frac{Z_1 Z_2}{A_1^{1/3} + A_2^{1/3}} \right], \quad (21)$$

hence the fusion barrier position (R_B) becomes $R_B^{\text{para}} = S_B^{\text{para}} + C_1 + C_2$.

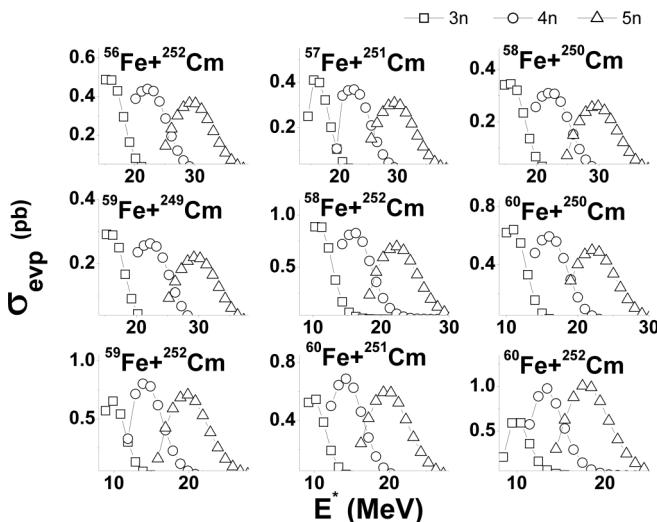


FIG. 8. Evaporation residue cross section as a function of E^* for projectile-target combination Fe + Cm.

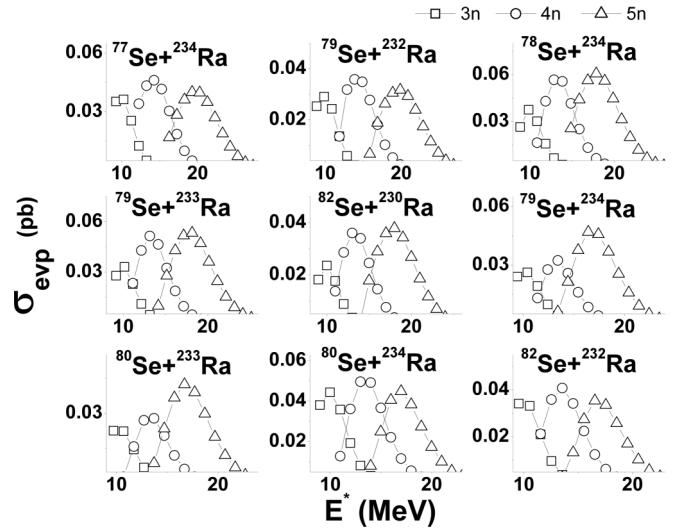


FIG. 9. Evaporation residue cross section as a function of E^* for projectile-target combination Se + Ra.

Finally, the parametrized fusion barrier position can be expressed as

$$R_B^{\text{para}} = 2.6054 - 0.0703 \times \left[\frac{Z_1 Z_2}{A_1^{1/3} + A_2^{1/3}} \right] + C_1 + C_2. \quad (22)$$

The calculated fusion barrier height (V_B) is plotted as a function of $(Z_1 Z_2 / R_B^{\text{para}})(1 - 1/R_B^{\text{para}})$ and it is shown in Fig. 4. We have fitted nonlinear function for fusion barrier height as follows;

$$V_B^{\text{para}} = 5.051 + 0.961 \left[\frac{1.44 Z_1 Z_2}{R_B^{\text{para}}} \left(1 - \frac{0.75}{R_B^{\text{para}}} \right) \right]. \quad (23)$$

The above equations may be used to produce the barrier heights (V_B) and the positions (R_B) of $^{307-314}$ 122 nuclei. After

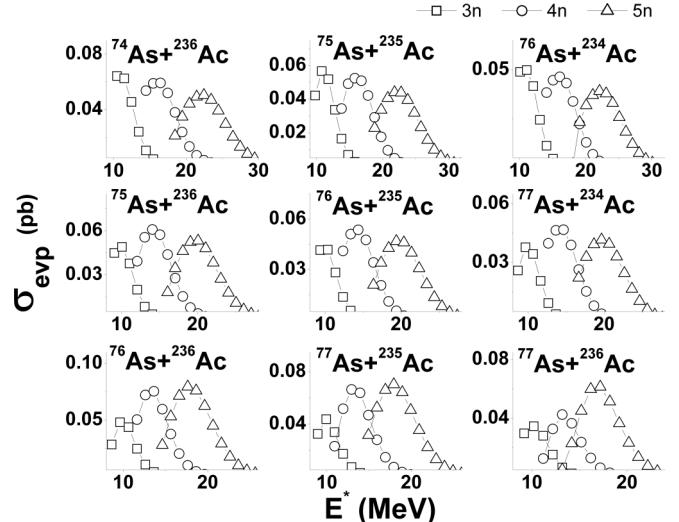


FIG. 10. Evaporation residue cross section as a function of E^* for projectile-target combination As + Ac.

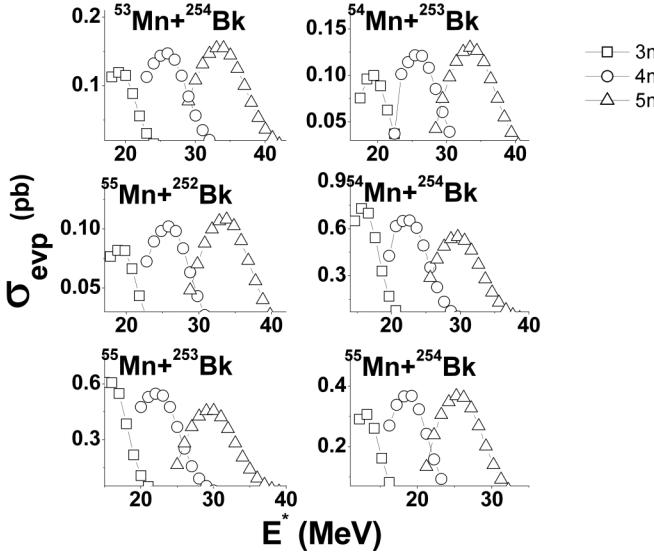


FIG. 11. Evaporation residue cross section as a function of E^* for projectile-target combination Mn + Bk.

studying the fusion barriers, we have calculated the compound nucleus formation probability (P_{CN}) and the survival probability for different projectile-target combinations.

The variation of calculated compound nucleus formation probability (P_{CN}) at 20 MeV with a mass number of projectile for superheavy nuclei $^{307-314}122$ is shown in Fig. 5. From this figure, it is clear that P_{CN} decreases with increasing mass number of projectile. The variation of survival probability (P_{Surv}) at 20 MeV (for 3n) with mass number of projectile for superheavy nuclei $^{307-314}122$ is as shown in Fig. 6. We have calculated the evaporation residue cross sections for all studied possible projectile target combinations to synthesize superheavy nuclei $^{307-314}122$. Figures 7–11 show the evaporation residue cross section as a function of E^* for projectile-target combination systems Cr + Cf, Fe + Cm, Se + Ra, As + Ac, and Mn + Bk respectively.

We have studied the variation of evaporation residue cross section as a function of mass number. Figure 12 shows the variation of evaporation residue cross section for $^{307-314}122$ vs mass number of the projectiles. The evaporation residue cross section decreases with increase in the mass number of the projectiles. We have selected the most probable projectile-target combination to synthesize superheavy nuclei $^{307-314}122$ that have minimum driving potential, maximum fusion, and evaporation residue cross sections. The selected most probable projectile-target combinations such as Cr + Cf, Fe + Cm, Se + Ra, As + Ac are listed in Table I.

The parameters required to decide the synthesis of superheavy nuclei such as compound nucleus fissility (χ_{CN}), charge product in the entrance channel ($Z_p Z_t$), effective entrance channel fissility (χ_{eff}), fusion barrier height (V_B), and fusion barrier width (R_B) for the most probable fusion reactions are also shown in Table I. The compound nucleus fissility (χ_{CN}) is calculated using the following equation:

$$\chi_{CN} = \frac{Z^2}{A} \left/ \left(\frac{Z^2}{A} \right)_{crit} \right., \quad (24)$$

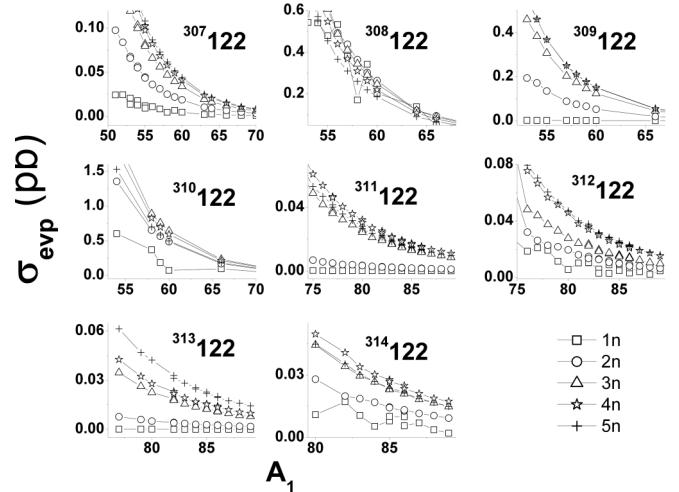


FIG. 12. Evaporation residue cross section as a function of mass number of the projectile.

where Z and A are the atomic and mass numbers of the compound nucleus, respectively. The denominator is taken as

$$\left(\frac{Z^2}{A} \right)_{crit} = 50.883(1 - 1.7826I^2), \quad (25)$$

where $I = (A - 2Z)/A$ is the relative neutron excess of the compound nucleus.

The effective entrance channel fissility (χ_{eff}) is defined as

$$\chi_{eff} = \frac{4Z_1 Z_2}{[A_1^{1/3} A_2^{1/3} (A_1^{1/3} + A_2^{1/3})]} \left/ \left(\frac{Z^2}{A} \right)_{crit} \right., \quad (26)$$

where A_1 and A_2 are the mass number of projectile and target nucleus respectively. Tabulated data of compound nucleus fissility (χ_{CN}), charge product in the entrance channel ($Z_p Z_t$), effective entrance channel fissility (χ_{eff}), fusion barrier height (V_B), and fusion barrier width (R_B) for the most probable fusion reactions are useful in the experiments to synthesize more isotopes of SHN $Z = 122$.

After identifying the most probable projectile-target combinations, we have studied the different decay modes of $Z = 122$. The formed compound nucleus may undergo different decay modes such as alpha decay, cluster decay, and binary and ternary fission processes. It is essential to study the competition between different decay modes of superheavy nuclei. The energy released (Q) during the studied process such as spontaneous fission, alpha ternary fission, cluster decay, and alpha decay is calculated by using the following equation:

$$Q = \Delta M(A, Z) - \sum_i^n \Delta M(A_i, Z_i), \quad (27)$$

where $\Delta M(A, Z)$ and $\Delta M(A_i, Z_i)$ are the mass excess of the parent and emitted nuclei respectively.

For ternary fission n varies from 1 to 3. For spontaneous fission and cluster and alpha decay processes, n varies from 1 to 2. In the present work, we have used this experimental mass excess data [43]. Some of the experimental mass excess values are not available. For those nuclei, where experimental

TABLE I. Presynthesis parameters for SHE $Z = 122$.

CN	Most probable projectile-target combination	$(\sigma_{\text{Evp}})_{\text{max}}$ pb	V_B (MeV)	R_B (fm)	$Z_p Z_t$	χ_{CN}	$\chi_{\text{eff}} (\times 10^3)$	$E_{c.m}$ (MeV)	N/A
³⁰⁷ 122	⁵⁷ Fe (S 2.12%) + ²⁵⁰ Cm(9000 yr)	5.61×10^{-2}	268.731	12.338	2496		1.31	288	
	⁶⁰ Fe (2.6×10^6 yr) + ²⁴⁷ Cm (1.56×10^7 yr)	3.38×10^{-2}	267.65	12.397	2496		1.26	287	
	⁶³ Ni (100 yr) + ²⁴⁴ Pu (8.08×10^7 yr)	1.92×10^{-2}	282.425	12.361	2632	1.0302	1.28	302	0.6026
	⁷² Zn (S 46.5%) + ²³⁵ U (7.04×10^8 yr)	4.80×10^{-3}	294.373	12.423	2760		1.22	314	
³⁰⁸ 122	⁷⁹ Se (3.27×10^5 yr) + ²²⁸ Ra (5.75 yr)	1.80×10^{-3}	319.253	12.357	2992		1.24	339	
	⁵⁸ Fe (S 0.28%) + ²⁵⁰ Cm (9000 yr)	3.45×10^{-1}	268.141	12.37	2496		1.29	284	
	⁶⁰ Fe (2.6×10^6 yr) + ²⁴⁸ Cm (3.4×10^5 yr)	2.47×10^{-1}	267.429	12.409	2496		1.25	283	
	⁶⁴ Ni (S 0.926%) + ²⁴⁴ Pu (8.08×10^7 yr)	1.21×10^{-1}	281.845	12.391	2632		1.26	297	
	⁷⁰ Zn (S 0.65%) + ²³⁸ U (4.468×10^9 yr)	4.67×10^{-2}	294.741	12.405	2760	1.0289	1.24	310	0.6039
	⁷² Zn (S 46.5%) + ²³⁶ U (2.342×10^7 yr)	3.55×10^{-2}	294.121	12.436	2760		1.21	310	
	⁷⁶ Ge (1.78×10^{21} yr) + ²³² Th (1.41×10^{10} yr)	1.98×10^{-2}	306.834	12.412	2880		1.22	322	
	⁸⁰ Se (S 49.85%) + ²²⁸ Ra (5.75 yr)	1.15×10^{-2}	318.685	12.385	2992		1.23	334	
³⁰⁹ 122	⁸² Se (1.08×10^{20} yr) + ²²⁶ Ra (1600 yr)	9.10×10^{-3}	318.129	12.412	2992		1.21	334	
	⁵³ Cr (S 9.5%) + ²⁵⁶ Cf (12.3 min)	4.60×10^{-1}	253.032	12.374	2352		1.29	266	
	⁵⁴ Cr (S 2.365%) + ²⁵⁵ Cf (85 min)	3.83×10^{-1}	252.656	12.395	2352		1.27	266	
	⁵⁷ Fe (S 2.12%) + ²⁵² Cm (<1 d)	2.03×10^{-1}	268.289	12.362	2496		1.30	281	
	⁵⁸ Fe (S 0.28%) + ²⁵¹ Cm (16.2 min)	1.74×10^{-1}	267.921	12.382	2496		1.28	281	
	⁵⁹ Fe (44.6 d) + ²⁵⁰ Cm (9000 yr)	1.46×10^{-1}	267.561	12.402	2496		1.26	281	
	⁶⁰ Fe (2.6×10^6 yr) + ²⁴⁹ Cm (64.15 min)	1.23×10^{-1}	267.209	12.421	2496	1.0277	1.25	280	0.6052
	⁶⁶ Ni (54.6 h) + ²⁴³ Pu (4.956 h)	4.50×10^{-2}	280.945	12.438	2632		1.22	294	
	⁷⁰ Zn (S 0.6%) + ²³⁹ U (23.45 min)	2.33×10^{-2}	294.49	12.418	2760		1.23	308	
	⁷² Zn (S 46.5%) + ²³⁷ U (6.75 d)	1.79×10^{-2}	293.871	12.449	2760		1.21	307	
³¹⁰ 122	⁷⁶ Ge (1.78×10^{21} yr) + ²³³ Th (21.83 min)	9.90×10^{-3}	306.568	12.425	2880		1.21	320	
	⁷⁹ Se (3.27×10^5 yr) + ²³⁰ Ra (92 min)	6.50×10^{-3}	318.692	12.384	2992		1.23	332	
	⁸² Se (1.08×10^{20} yr) + ²²⁷ Ra (42.2 min)	4.60×10^{-3}	317.849	12.426	2992		1.20	331	
	⁵⁴ Cr (S 2.365%) + ²⁵⁶ Cf (12.3 min)	1.97×10^0	252.452	12.407	2352		1.27	263	
	⁵⁸ Fe (S 0.28%) + ²⁵² Cm (<1 d)	8.88×10^{-1}	267.701	12.395	2496		1.27	278	
	⁵⁹ Fe (44.6 d) + ²⁵¹ Cm (16.8 min)	7.56×10^{-1}	267.341	12.413	2496		1.26	278	
	⁶⁰ Fe (2.6×10^6 yr) + ²⁵⁰ Cm (9000 yr)	6.40×10^{-1}	266.99	12.432	2496		1.24	278	
	⁶⁶ Ni (54.6 h) + ²⁴⁴ Pu (8 $\times 10^7$ yr)	2.30×10^{-1}	280.711	12.45	2632		1.22	291	
	⁷² Zn (S 46.5%) + ²³⁸ U (4.468×10^9 yr)	9.19×10^{-2}	293.621	12.462	2760	1.0265	1.20	304	0.6065
	⁷⁶ Ge (1.78×10^{21} yr) + ²³⁴ Th (24.1 d)	5.15×10^{-2}	306.304	12.438	2880		1.21	317	
³¹¹ 122	⁷⁵ As (S100%) + ²³⁵ Ac (40 s)	5.67×10^{-2}	313.219	12.381	2937		1.24	324	
	⁸⁰ Se (S 49.8%) + ²³⁰ Ra (92 min)	2.99×10^{-2}	318.126	12.412	2992		1.21	329	
	⁸² Se (1.08×10^{20} yr) + ²²⁸ Ra (5.75 yr)	2.36×10^{-2}	317.569	12.439	2992		1.19	328	
	⁵⁹ Fe (44.6 d) + ²⁵² Cm (<1 d)	6.53×10^{-1}	267.123	12.425	2496		1.25	277	
	⁶⁰ Fe (2.6×10^6 yr) + ²⁵¹ Cm (16.8 min)	5.45×10^{-1}	266.771	12.444	2496		1.23	277	
	⁷² Zn (S 46.5%) + ²³⁹ U (23.45 min)	7.90×10^{-2}	293.373	12.474	2760		1.19	303	
	⁷⁵ As (S 100%) + ²³⁶ Ac (2 min)	4.89×10^{-2}	312.949	12.394	2937		1.24	323	
	⁷⁶ As (1.1 d) + ²³⁵ Ac (40 s)	4.18×10^{-2}	312.644	12.409	2937		1.22	323	
	⁷⁷ As (38.33 h) + ²³⁴ Ac (44 s)	3.78×10^{-2}	312.345	12.424	2937	1.0254	1.21	322	0.6077
	⁷⁷ Se (S 7.6%) + ²³⁴ Ra (30 s)	3.63×10^{-2}	318.729	12.383	2992		1.24	329	
³¹² 122	⁷⁸ Se (S 23.69%) + ²³³ Ra (30 s)	3.22×10^{-2}	318.429	12.397	2992		1.23	328	
	⁷⁹ Se (3.27×10^5 yr) + ²³² Ra (250 s)	2.91×10^{-2}	318.136	12.412	2992		1.22	328	
	⁸⁰ Se (S 49.8%) + ²³¹ Ra (103 s)	2.56×10^{-2}	317.848	12.426	2992		1.21	328	
	⁸² Se (1.08×10^{20} yr) + ²²⁹ Ra (4 min)	2.04×10^{-2}	317.292	12.453	2992		1.19	327	
	⁶⁰ Fe (2.6×10^6 yr) + ²⁵² Cm (<1 d)	5.91×10^{-1}	266.554	12.456	2497		1.23	277	
	⁷⁶ As (1.1 d) + ²³⁶ Ac (2 min)	4.81×10^{-2}	312.375	12.423	2937		1.22	322	
	⁷⁷ As (38.33 h) + ²³⁵ Ac (40 s)	4.38×10^{-2}	312.076	12.437	2937		1.21	322	
	⁷⁸ Se (S 23.69%) + ²³⁴ Ra (30 s)	3.75×10^{-2}	318.154	12.411	2992	1.0243	1.22	328	0.6090
³¹³ 122	⁷⁹ Se (3.27×10^5 yr) + ²³³ Ra (30 s)	3.30×10^{-2}	317.86	12.425	2992		1.21	328	
	⁸⁰ Se (S 49.8%) + ²³² Ra (250 s)	2.78×10^{-2}	317.572	12.439	2992		1.20	328	
	⁸² Se (1.08×10^{20} yr) + ²³⁰ Ra (92 min)	2.36×10^{-2}	317.015	12.466	2992		1.18	327	

TABLE I. (Continued.)

CN	Most probable projectile-target combination	$(\sigma_{\text{Evp}})_{\text{max}}$ Pb	V_B (MeV)	R_B (fm)	$Z_p Z_t$	χ_{CN}	$\chi_{\text{eff}} (\times 10^3)$	$E_{c.m.}$ (MeV)	N/A
³¹³ 122	⁷⁷ As (38.33 h) + ²³⁶ Ac (2 min)	3.46×10^{-2}	311.809	12.451	2937	1.0232	1.20	322	0.6102
	⁷⁹ Se (3.27×10^5 yr) + ²³⁴ Ra (30 s)	2.62×10^{-2}	317.585	12.438	2992		1.20	328	
	⁸⁰ Se (S 49.8%) + ²³³ Ra (30 s)	2.25×10^{-2}	317.298	12.453	2992		1.19	327	
	⁸² Se (1.08×10^{20} yr) + ²³¹ Ra (103 s)	1.86×10^{-2}	316.74	12.48	2992		1.17	327	
³¹⁴ 122	⁸⁰ Se (S 49.8%) + ²³⁴ Ra (30 s)	4.44×10^{-2}	317.024	12.466	2992	1.0221	1.19	327	0.6115
	⁸² Se (1.08×10^{20} yr) + ²³² Ra (250 s)	3.41×10^{-2}	316.467	12.493	2992		1.17	326	

mass excess was unavailable, we have used theoretical values [44,45]. The total potential for spontaneous and ternary fission of different fission fragments are calculated. The driving potential ($V - Q$) is calculated. The variation of driving potential for spontaneous and ternary fission as a function of the mass number of fragment A_1 is shown in Fig. 13. The calculated spontaneous and ternary fission half-lives for the combination of different fission fragments of superheavy nuclei $Z = 122$ are presented. The variation of logarithmic half-lives of spontaneous and ternary fission as a function of mass number A_1 for different isotopes of SHN $Z = 122$ are shown in Fig. 14.

The energy released (Q) during the emission of different cluster nuclei from the different isotopes of SHN $Z = 126$ is calculated. Figure 15 represents the energy released (Q) during the emission of different clusters (^{12–14}C, ¹⁴N, ^{20–24}Ne, ^{28–34}Si, ^{36–44}Ar, ^{40–48}Ca) from the superheavy nuclei $Z = 122$ as a function of mass number of clusters. From this figure, it is clear that the Q value in the cluster decay increases with an increase in the mass number of the emitted cluster. The half-lives of cluster emission (^{12–14}C, ¹⁴N, ^{20–24}Ne, ^{28–34}Si, ^{36–44}Ar, ^{40–48}Ca) from the different isotopes of superheavy nuclei $Z = 122$ are calculated. Figure 16 shows the variation of logarithmic half-lives for the emission of different clusters (^{12–14}C, ¹⁴N, ^{20–24}Ne, ^{28–34}Si, ^{36–44}Ar, ^{40–48}Ca) from su-

perheavy nuclei $Z = 122$ as a function of mass number of daughter nuclei.

The computed logarithmic cluster decay half-lives, decay constant, barrier penetrability, and other characteristics of cluster decay during the emission of ^{12,14}C, ¹⁴N, ^{20,22,24}Ne, ^{28,30,32,34}Si, ^{36,38,40,42,44}Ar, ^{40,42,44,46,48}Ca from the different isotopes of superheavy nuclei $Z = 122$ are tabulated in the Table II. To identify the dominant decay mode of the most predicted isotopes of superheavy nuclei $Z = 122$, we have calculated the branching ratios. Branching ratios are calculated using their decay constants. The branching ratio of alpha decay with respect to spontaneous fission is defined as

$$\text{BR} = \frac{\lambda_\alpha}{\lambda_{\text{SF}}} \quad (28)$$

The branching ratio of alpha decay with respect to alpha ternary fission is defined as

$$\text{BR} = \frac{\lambda_\alpha}{\lambda_{\text{TF}}} \quad (29)$$

The branching ratio of alpha decay with respect to cluster decay is defined as

$$\text{BR} = \frac{\lambda_\alpha}{\lambda_{\text{CR}}} \quad (30)$$

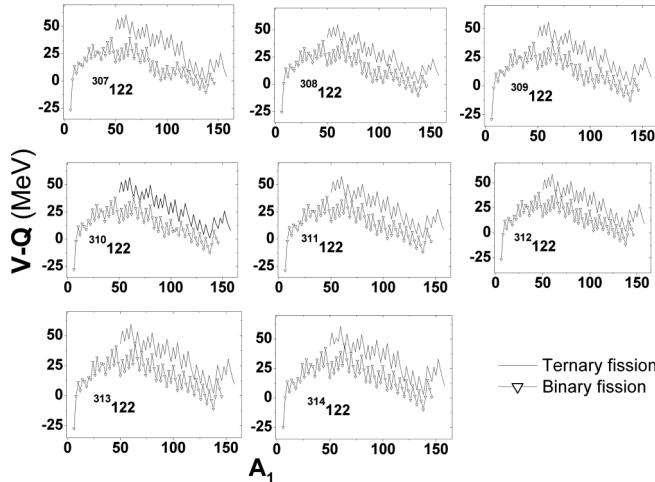


FIG. 13. Variation of driving potential for spontaneous and ternary fission as a function of the mass number of fragment A_1 .

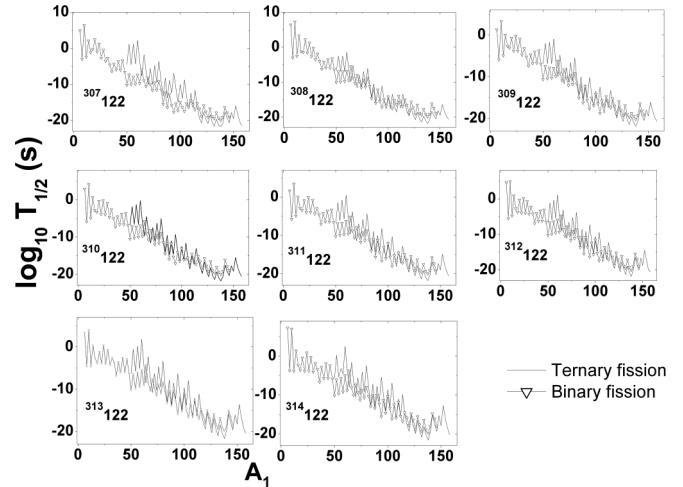


FIG. 14. Variation of logarithmic half-lives of spontaneous and ternary fission as a function of mass number A_1 for different isotopes of SHN $Z = 122$.

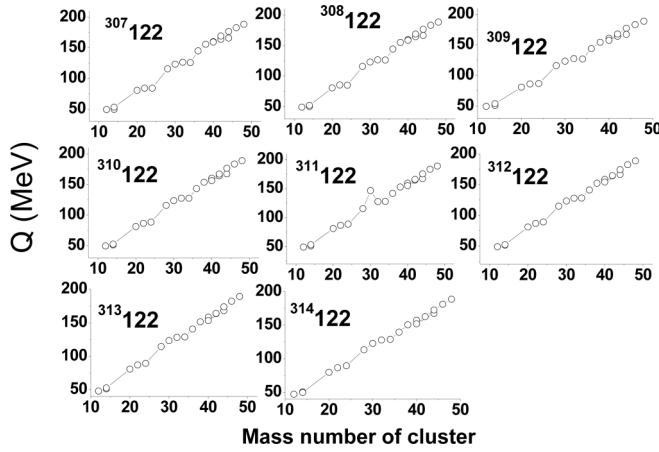


FIG. 15. Energy released (Q) during the emission of different clusters ($^{12-14}\text{C}$, ^{14}N , $^{20-24}\text{Ne}$, $^{28-34}\text{Si}$, $^{36-44}\text{Ar}$, $^{40-48}\text{Ca}$) for superheavy nuclei $Z = 122$ as a function of mass number of clusters.

where λ_α , λ_{SF} , λ_{TF} , and λ_{CR} are decay constants, corresponds to alpha decay, spontaneous fission, alpha ternary fission, and cluster decay processes respectively. The calculated branching ratio of alpha decay with respect to the spontaneous fission, ternary fission, and cluster decay for different isotopes of superheavy nuclei $Z = 122$ is as shown in Table III. The computed logarithmic half-life values for various decay modes of superheavy nuclei $Z = 122$ is as given in Table IV. We have compared the logarithmic half-lives for spontaneous fission, ternary fission, cluster radioactivity, and alpha decay for superheavy nuclei $^{307}\text{122}$, $^{318}\text{126}$, $^{319}\text{126}$, $^{320}\text{126}$, and $^{323-326}\text{126}$ and it is shown in Table IV. From this table, it is clear that alpha decay half-lives are smaller than those of spontaneous fission, ternary fission, and cluster decay for the isotopes of superheavy nuclei $Z = 122$.

TABLE II. The computed logarithmic half-life values, decay constant, barrier penetrability and other characteristics, and cluster decay during the emission of $^{12,14}\text{C}$, ^{14}N , $^{20,22,24}\text{Ne}$, $^{28,30,32,34}\text{Si}$, $^{36,38,40,42,44}\text{Ar}$, $^{40,42,44,46,48}\text{Ca}$ from the different isotopes of superheavy nuclei $Z = 122$.

Parent nuclei	Emitted cluster	Daughter nuclei	Q value(MeV)	Penetrability P	Decay constant λ	$\log_{10} T_{1/2}$	Parent nuclei
$^{307}\text{122}$	^{12}C	$^{295}\text{122}$	49.3300	1.34×10^{21}	1.15×10^{-20}	1.54×10^1	-1.347
	^{14}C	$^{293}\text{122}$	49.5100	1.34×10^{21}	2.54×10^{-29}	3.41×10^{-8}	7.307929
	^{14}N	$^{293}\text{122}$	53.2070	1.44×10^{21}	2.44×10^{-25}	3.51×10^{-4}	3.294449
	^{20}Ne	$^{287}\text{122}$	80.5120	2.18×10^{21}	1.76×10^{-25}	3.83×10^{-4}	3.256649
	^{22}Ne	$^{285}\text{122}$	84.1380	2.28×10^{21}	2.22×10^{-5}	2.81×10^{-4}	-16.8631
	^{24}Ne	$^{283}\text{122}$	84.2190	2.28×10^{21}	1.13×10^{-22}	6.78×10^{-21}	0.428553
	^{28}Si	$^{279}\text{122}$	115.8330	3.14×10^{21}	2.07×10^{-25}	6.50×10^{-4}	3.027158
	^{30}Si	$^{277}\text{122}$	123.1470	3.33×10^{21}	4.83×10^{-22}	0.161×10^1	-0.36646
	^{32}Si	$^{275}\text{122}$	126.4180	3.42×10^{21}	2.52×10^{-20}	8.63×10^1	-2.09612
	^{34}Si	$^{273}\text{122}$	125.9880	3.41×10^{21}	4.24×10^{-20}	3.41×10^{21}	-2.32031
	^{36}Ar	$^{271}\text{122}$	145.1920	3.93×10^{21}	2.55×10^{-26}	1.00×10^{-4}	3.839018
	^{38}Ar	$^{269}\text{122}$	155.7450	4.22×10^{21}	1.86×10^{-22}	7.84×10^{-1}	-0.05432
	^{40}Ar	$^{267}\text{122}$	160.1260	4.34×10^{21}	4.83×10^{-20}	2.09×10^2	-2.48052
	^{42}Ar	$^{265}\text{122}$	164.0040	4.44×10^{21}	7.63×10^{-19}	3.39×10^3	-3.68996
	^{44}Ar	$^{263}\text{122}$	166.1260	4.5×10^{21}	5.07×10^{-18}	2.28×10^4	-4.51753

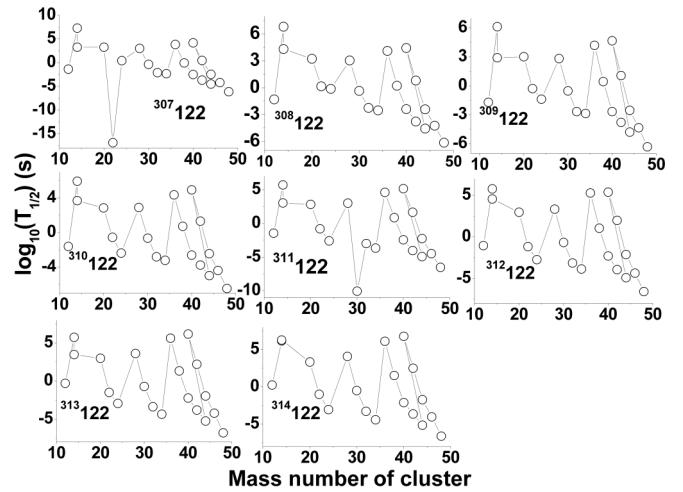


FIG. 16. Variation of logarithmic half-lives for the emission of different clusters ($^{12-14}\text{C}$, ^{14}N , $^{20-24}\text{Ne}$, $^{28-34}\text{Si}$, $^{36-44}\text{Ar}$, $^{40-48}\text{Ca}$) from superheavy nuclei $Z = 122$ as a function of mass number of cluster nuclei.

IV. CONCLUSION

We have also selected the most probable projectile-target combinations, Cr + Cf, Fe + Cm, Se + Ra, As + Ac to synthesize the superheavy element $Z = 122$. We hope that our predictions may be a guide for future experiments in the synthesis of more isotopes of superheavy nuclei $Z = 112$. We have also studied the different decay modes of the most predicted isotopes of superheavy element $Z = 122$. The comparison of half-lives for different decay modes reveals that alpha decay has smaller half-lives than the other studied decay modes. A detailed study of branching ratio of alpha decay with respect to other decay modes also confirms that alpha decay is the most dominant decay mode for the isotopes of superheavy nuclei $^{307-314}\text{122}$ and hence these nuclei can be detected through the alpha decay mode only.

TABLE II. (*Continued.*)

Parent nuclei	Emitted cluster	Daughter nuclei	<i>Q</i> value(MeV)	Penetrability <i>P</i>	Decay constant λ	$\log_{10} T_{1/2}$	Parent nuclei
³⁰⁸ 122	⁴⁰ Ca	²⁶⁷ 122	159.3960	4.32×10^{21}	1.07×10^{-26}	4.61×10^{-5}	4.176363
	⁴² Ca	²⁶⁵ 122	169.3870	4.59×10^{21}	4.95×10^{-23}	2.27×10^{-1}	0.483641
	⁴⁴ Ca	²⁶³ 122	176.7790	4.79×10^{21}	3.91×10^{-20}	1.87×10^2	-2.4324
	⁴⁶ Ca	²⁶¹ 122	182.9210	4.95×10^{21}	2.28×10^{-18}	1.13×10^4	-4.2126
	⁴⁸ Ca	²⁵⁹ 122	188.3950	5.1×10^{21}	1.79×10^{-16}	9.12×10^5	-6.11979
	¹² C	²⁹⁶ 122	49.2600	1.33×10^{21}	1.07×10^{-20}	1.42×10^1	-1.31257
	¹⁴ C	²⁹⁴ 122	50.1300	1.36×10^{21}	7.92×10^{-29}	1.08×10^{-7}	6.808576
	¹⁴ N	²⁹⁴ 122	52.2570	1.41×10^{21}	2.35×10^{-26}	3.33×10^{-5}	4.317715
	²⁰ Ne	²⁸⁸ 122	80.5120	2.18×10^{21}	1.88×10^{-25}	4.10×10^{-4}	3.226809
	²² Ne	²⁸⁶ 122	85.4950	2.31×10^{21}	2.11×10^{-22}	4.88×10^{-1}	0.151704
	²⁴ Ne	²⁸⁴ 122	85.0780	2.3×10^{21}	3.91×10^{-22}	9.01×10^{-1}	-0.11467
	²⁸ Si	²⁸⁰ 122	115.7230	3.13×10^{21}	1.96×10^{-25}	6.15×10^{-4}	3.051528
	³⁰ Si	²⁷⁸ 122	122.9930	3.33×10^{21}	4.61×10^{-22}	0.154×10^1	-0.34604
	³² Si	²⁷⁶ 122	126.6580	3.43×10^{21}	3.58×10^{-20}	1.23×10^2	-2.2496
	³⁴ Si	²⁷⁴ 122	126.3320	3.42×10^{21}	6.75×10^{-20}	2.31×10^2	-2.52302
	³⁶ Ar	²⁷² 122	144.0720	3.9×10^{21}	1.34×10^{-26}	5.23×10^{-5}	4.121359
	³⁸ Ar	²⁷⁰ 122	154.9650	4.2×10^{21}	9.88×10^{-23}	4.14×10^{-1}	0.222669
	⁴⁰ Ar	²⁶⁸ 122	159.5660	4.32×10^{21}	3.88×10^{-20}	1.68×10^2	-2.3848
	⁴² Ar	²⁶⁶ 122	164.2470	4.45×10^{21}	9.45×10^{-19}	4.20×10^3	-3.78361
	⁴⁴ Ar	²⁶⁴ 122	166.1970	4.5×10^{21}	5.73×10^{-18}	2.58×10^4	-4.57162
	⁴⁰ Ca	²⁶⁸ 122	158.3360	4.29×10^{21}	5.9×10^{-27}	2.53×10^{-5}	4.436773
	⁴² Ca	²⁶⁶ 122	168.5070	4.56×10^{21}	2.38×10^{-23}	1.09×10^{-1}	0.804158
	⁴⁴ Ca	²⁶⁴ 122	176.5190	4.78×10^{21}	3.7×10^{-20}	1.77×10^2	-2.40724
	⁴⁶ Ca	²⁶² 122	182.8840	4.95×10^{21}	2.44×10^{-18}	1.21×10^4	-4.24201
	⁴⁸ Ca	²⁶⁰ 122	188.3030	5.1×10^{21}	1.87×10^{-16}	9.52×10^5	-6.13872
³⁰⁹ 122	¹² C	²⁹⁷ 122	49.6300	1.34×10^{21}	2.6×10^{-20}	3.49×10^1	-1.70283
	¹⁴ C	²⁹⁵ 122	50.9400	1.38×10^{21}	3.71×10^{-28}	5.12×10^{-7}	6.130767
	¹⁴ N	²⁹⁵ 122	53.7170	1.45×10^{21}	5.83×10^{-25}	8.48×10^{-4}	2.911618
	²⁰ Ne	²⁸⁹ 122	80.9320	2.19×10^{21}	3.12×10^{-25}	6.84×10^{-4}	3.005138
	²² Ne	²⁸⁷ 122	86.1250	2.33×10^{21}	5.9×10^{-22}	0.138×10^1	-0.29874
	²⁴ Ne	²⁸⁵ 122	86.6950	2.35×10^{21}	7.12×10^{-21}	1.67×10^1	-1.383
	²⁸ Si	²⁸¹ 122	115.9430	3.14×10^{21}	3.37×10^{-25}	1.06×10^{-3}	2.815154
	³⁰ Si	²⁷⁹ 122	123.4030	3.34×10^{21}	7.08×10^{-22}	0.236×10^1	-0.53357
	³² Si	²⁷⁷ 122	127.4250	3.45×10^{21}	9.4×10^{-20}	3.24×10^2	-2.67067
	³⁴ Si	²⁷⁵ 122	126.9240	3.44×10^{21}	1.46×10^{-19}	5.02×10^2	-2.86061
	³⁶ Ar	²⁷³ 122	143.7120	3.89×10^{21}	1.15×10^{-26}	4.48×10^{-5}	4.188599
	³⁸ Ar	²⁷¹ 122	154.3050	4.18×10^{21}	6.03×10^{-23}	2.52×10^{-1}	0.438656
	⁴⁰ Ar	²⁶⁹ 122	160.7000	4.35×10^{21}	7.52×10^{-20}	3.27×10^2	-2.67447
	⁴² Ar	²⁶⁷ 122	164.1390	4.44×10^{21}	9.61×10^{-19}	4.27×10^3	-3.79061
	⁴⁴ Ar	²⁶⁵ 122	166.8840	4.52×10^{21}	9.42×10^{-18}	4.26×10^4	-4.78901
	⁴⁰ Ca	²⁶⁹ 122	157.4160	4.26×10^{21}	3.59×10^{-27}	1.53×10^{-5}	4.654801
	⁴² Ca	²⁶⁷ 122	167.7270	4.54×10^{21}	1.32×10^{-23}	5.97×10^{-2}	1.063913
	⁴⁴ Ca	²⁶⁵ 122	176.9380	4.79×10^{21}	4.97×10^{-20}	2.38×10^2	-2.53637
	⁴⁶ Ca	²⁶³ 122	183.0760	4.96×10^{21}	3.00×10^{-18}	1.49×10^4	-4.33175
	⁴⁸ Ca	²⁶¹ 122	188.6300	5.11×10^{21}	2.72×10^{-16}	1.39×10^6	-6.30323
³¹⁰ 122	¹² C	²⁹⁸ 122	49.4900	1.34×10^{21}	2.03×10^{-20}	2.72×10^1	-1.59428
	¹⁴ C	²⁹⁶ 122	51.1600	1.39×10^{21}	6.06×10^{-28}	8.39×10^{-7}	5.91617
	¹⁴ N	²⁹⁶ 122	52.5770	1.42×10^{21}	1.02×10^{-25}	1.46×10^{-4}	3.677131
	²⁰ Ne	²⁹⁰ 122	81.2220	2.2×10^{21}	4.54×10^{-25}	9.98×10^{-4}	2.840811
	²² Ne	²⁸⁸ 122	86.4150	2.34×10^{21}	1.05×10^{-21}	0.246×10^1	-0.55053
	²⁴ Ne	²⁸⁶ 122	88.3420	2.39×10^{21}	6.84×10^{-20}	1.64×10^2	-2.37369
	²⁸ Si	²⁸² 122	115.7730	3.13×10^{21}	2.81×10^{-25}	8.82×10^{-4}	2.894855
	³⁰ Si	²⁸⁰ 122	123.5830	3.35×10^{21}	8.97×10^{-22}	0.300×10^1	-0.6374
	³² Si	²⁷⁸ 122	127.5610	3.45×10^{21}	1.24×10^{-19}	4.28×10^2	-2.79095
	³⁴ Si	²⁷⁶ 122	127.4540	3.45×10^{21}	3.18×10^{-19}	1.10×10^3	-3.19979

TABLE II. (*Continued.*)

Parent nuclei	Emitted cluster	Daughter nuclei	<i>Q</i> value(MeV)	Penetrability <i>P</i>	Decay constant λ	$\log_{10} T_{1/2}$	Parent nuclei
³¹¹ 122	⁴⁰ Ar	²⁷⁰ 122	160.2100	4.34×10^{21}	6.25×10^{-20}	2.71×10^2	-2.5931
	⁴² Ar	²⁶⁸ 122	163.8690	4.44×10^{21}	8.92×10^{-19}	3.96×10^3	-3.75732
	⁴⁴ Ar	²⁶⁶ 122	167.4170	4.53×10^{21}	1.42×10^{-17}	6.44×10^4	-4.96906
	⁴⁰ Ca	²⁷⁰ 122	156.2660	4.23×10^{21}	1.91×10^{-27}	8.09×10^{-6}	4.932028
	⁴² Ca	²⁶⁸ 122	166.9570	4.52×10^{21}	7.54×10^{-24}	3.41×10^{-2}	1.30761
	⁴⁴ Ca	²⁶⁶ 122	176.3480	4.77×10^{21}	3.95×10^{-20}	1.89×10^2	-2.43543
	⁴⁶ Ca	²⁶⁴ 122	183.1060	4.96×10^{21}	3.34×10^{-18}	1.66×10^4	-4.37892
	⁴⁸ Ca	²⁶² 122	188.8830	5.11×10^{21}	3.79×10^{-16}	1.94×10^6	-6.44697
	¹² C	²⁹⁹ 122	49.3100	1.34×10^{21}	1.46×10^{-20}	1.95×10^1	-1.45061
	¹⁴ C	²⁹⁷ 122	51.3700	1.39×10^{21}	9.81×10^{-28}	1.36×10^{-6}	5.704987
	¹⁴ N	²⁹⁷ 122	53.4570	1.45×10^{21}	4.47×10^{-25}	6.48×10^{-4}	3.028727
	²⁰ Ne	²⁹¹ 122	81.1920	2.2×10^{21}	4.71×10^{-25}	1.04×10^{-3}	2.824779
	²² Ne	²⁸⁹ 122	86.6750	2.35×10^{21}	1.88×10^{-21}	0.441×10^1	-0.80441
	²⁴ Ne	²⁸⁷ 122	88.8120	2.4×10^{21}	1.1×10^{-19}	2.65×10^2	-2.583
	²⁸ Si	²⁸³ 122	115.5630	3.13×10^{21}	2.22×10^{-25}	6.93×10^{-4}	2.999274
	³⁰ Si	²⁸¹ 122	146.8030	3.97×10^{21}	1.92×10^{-12}	7.63×10^9	-10.0422
	³² Si	²⁷⁹ 122	127.8110	3.46×10^{21}	1.91×10^{-19}	6.61×10^2	-2.98039
	³⁴ Si	²⁷⁷ 122	128.0610	3.47×10^{21}	9.08×10^{-19}	3.15×10^3	-3.6578
	³⁶ Ar	²⁷⁵ 122	141.9320	3.84×10^{21}	4.38×10^{-27}	1.68×10^{-5}	4.613658
	³⁸ Ar	²⁷³ 122	152.9550	4.14×10^{21}	2.31×10^{-23}	9.56×10^{-2}	0.859796
	⁴⁰ Ar	²⁷¹ 122	159.3900	4.32×10^{21}	4.38×10^{-20}	1.89×10^2	-2.43691
	⁴² Ar	²⁶⁹ 122	164.8430	4.46×10^{21}	1.68×10^{-18}	7.49×10^3	-4.03434
	⁴⁴ Ar	²⁶⁷ 122	167.1490	4.53×10^{21}	1.31×10^{-17}	5.93×10^4	-4.93269
	⁴⁰ Ca	²⁷¹ 122	155.3460	4.21×10^{21}	1.18×10^{-27}	4.97×10^{-6}	5.144005
	⁴² Ca	²⁶⁹ 122	165.8770	4.49×10^{21}	3.44×10^{-24}	1.54×10^{-2}	1.65157
	⁴⁴ Ca	²⁶⁷ 122	175.4080	4.75×10^{21}	2.63×10^{-20}	1.25×10^2	-2.25625
	⁴⁶ Ca	²⁶⁵ 122	183.3650	4.96×10^{21}	4.28×10^{-18}	2.13×10^4	-4.48764
	⁴⁸ Ca	²⁶³ 122	188.9150	5.12×10^{21}	4.40×10^{-16}	2.25×10^6	-6.51241
³¹² 122	¹² C	³⁰⁰ 122	48.8500	1.32×10^{21}	6×10^{-21}	0.793×10^1	-1.05919
	¹⁴ C	²⁹⁸ 122	51.2500	1.39×10^{21}	8.29×10^{-28}	1.15×10^{-6}	5.779013
	¹⁴ N	²⁹⁸ 122	51.9270	1.41×10^{21}	1.39×10^{-26}	1.96×10^{-5}	4.547803
	²⁰ Ne	²⁹² 122	80.8320	2.19×10^{21}	3.45×10^{-25}	7.55×10^{-4}	2.962102
	²² Ne	²⁹⁰ 122	86.9850	2.36×10^{21}	4.26×10^{-21}	1.00×10^1	-1.16119
	²⁴ Ne	²⁸⁸ 122	89.1220	2.41×10^{21}	1.55×10^{-19}	3.73×10^2	-2.73218
	²⁸ Si	²⁸⁴ 122	114.9930	3.11×10^{21}	1.02×10^{-25}	3.19×10^{-4}	3.336952
	³⁰ Si	²⁸² 122	123.4930	3.34×10^{21}	9.81×10^{-22}	0.328×10^1	-0.67592
	³² Si	²⁸⁰ 122	128.0110	3.47×10^{21}	2.84×10^{-19}	9.86×10^2	-3.15362
	³⁴ Si	²⁷⁸ 122	128.2170	3.47×10^{21}	1.47×10^{-18}	5.09×10^3	-3.86671
	³⁶ Ar	²⁷⁶ 122	141.3020	3.83×10^{21}	9.64×10^{-28}	3.69×10^{-6}	5.273239
	³⁸ Ar	²⁷⁴ 122	152.3050	4.12×10^{21}	1.49×10^{-23}	6.13×10^{-2}	1.052494
	⁴⁰ Ar	²⁷² 122	158.5800	4.29×10^{21}	3.1×10^{-20}	1.33×10^2	-2.28429
	⁴² Ar	²⁷⁰ 122	164.3730	4.45×10^{21}	1.38×10^{-18}	6.16×10^3	-3.94965
	⁴⁴ Ar	²⁶⁸ 122	166.8990	4.52×10^{21}	1.22×10^{-17}	5.51×10^4	-4.90109
	⁴⁰ Ca	²⁷² 122	154.3260	4.18×10^{21}	6.91×10^{-28}	2.89×10^{-6}	5.379844
	⁴² Ca	²⁷⁰ 122	164.7470	4.46×10^{21}	1.56×10^{-24}	2.89×10^{-6}	1.996791
	⁴⁴ Ca	²⁶⁸ 122	174.6580	4.73×10^{21}	1.93×10^{-20}	9.15×10^1	-2.12119
	⁴⁶ Ca	²⁶⁶ 122	182.7950	4.95×10^{21}	3.31×10^{-18}	1.64×10^4	-4.37401
	⁴⁸ Ca	²⁶⁴ 122	188.9650	5.12×10^{21}	5.21×10^{-16}	2.66×10^6	-6.58556
³¹³ 122	¹² C	³⁰¹ 122	47.9200	1.3×10^{21}	1.09×10^{-21}	0.141×10^1	-0.30957
	¹⁴ C	²⁹⁹ 122	51.2500	1.39×10^{21}	8.89×10^{-28}	1.23×10^{-6}	5.749116
	¹⁴ N	²⁹⁹ 122	52.6770	1.43×10^{21}	1.62×10^{-25}	2.31×10^{-4}	3.476317
	²⁰ Ne	²⁹³ 122	80.6920	2.18×10^{21}	3.19×10^{-25}	6.96×10^{-4}	2.997325
	²² Ne	²⁹¹ 122	87.1350	2.36×10^{21}	9.36×10^{-21}	2.21×10^1	-1.50396
	²⁴ Ne	²⁸⁹ 122	89.5620	2.43×10^{21}	2.46×10^{-19}	5.96×10^2	-2.93501
	²⁸ Si	²⁸⁵ 122	114.4530	3.1×10^{21}	5.39×10^{-26}	1.67×10^{-4}	3.616912
³⁰ Si	²⁸³ 122	123.4630	3.34×10^{21}	1.04×10^{-21}	0.347×10^1	-0.70058	
	³² Si	²⁸¹ 122	128.2510	3.47×10^{21}	4.68×10^{-19}	1.63×10^3	-3.37104

TABLE II. (*Continued.*)

Parent nuclei	Emitted cluster	Daughter nuclei	<i>Q</i> value(MeV)	Penetrability <i>P</i>	Decay constant λ	$\log_{10} T_{1/2}$	Parent nuclei
$^{314}\text{122}$	^{34}Si	$^{279}\text{122}$	128.6470	3.48×10^{21}	4.48×10^{-18}	1.56×10^4	-4.35275
	^{36}Ar	$^{277}\text{122}$	140.6620	3.81×10^{21}	4.13×10^{-28}	1.57×10^{-6}	5.643293
	^{38}Ar	$^{275}\text{122}$	151.3750	4.1×10^{21}	7.76×10^{-24}	3.18×10^{-2}	1.33764
	^{40}Ar	$^{273}\text{122}$	158.2400	4.28×10^{21}	2.79×10^{-20}	1.20×10^2	-2.2378
	^{42}Ar	$^{271}\text{122}$	163.7330	4.43×10^{21}	1.04×10^{-18}	4.60×10^3	-3.82299
	^{44}Ar	$^{269}\text{122}$	168.0530	4.55×10^{21}	2.74×10^{-17}	1.25×10^5	-5.25536
	^{40}Ca	$^{273}\text{122}$	153.0760	4.14×10^{21}	1.12×10^{-28}	4.66×10^{-7}	6.172113
	^{42}Ca	$^{271}\text{122}$	164.0070	4.44×10^{21}	9.8×10^{-25}	4.35×10^{-3}	2.201385
	^{44}Ca	$^{269}\text{122}$	173.7580	4.7×10^{21}	1.32×10^{-20}	6.23×10^1	-1.95411
	^{46}Ca	$^{267}\text{122}$	182.0350	4.93×10^{21}	2.29×10^{-18}	1.13×10^4	-4.21167
	^{48}Ca	$^{265}\text{122}$	189.4040	5.13×10^{21}	8.83×10^{-16}	4.53×10^6	-6.81602
	^{12}C	$^{302}\text{122}$	47.1500	1.28×10^{21}	2.97×10^{-22}	3.80×10^{-1}	0.260678
	^{14}C	$^{300}\text{122}$	50.7300	1.37×10^{21}	3.49×10^{-28}	4.80×10^{-7}	6.158867
	^{14}N	$^{300}\text{122}$	49.9370	1.35×10^{21}	2.67×10^{-28}	3.62×10^{-7}	6.281974
	^{20}Ne	$^{294}\text{122}$	79.9020	2.16×10^{21}	1.5×10^{-25}	3.25×10^{-4}	3.327909
	^{22}Ne	$^{292}\text{122}$	86.7150	2.35×10^{21}	2.89×10^{-21}	0.679×10^1	-0.99161
	^{24}Ne	$^{290}\text{122}$	89.8120	2.43×10^{21}	3.3×10^{-19}	8.01×10^2	-3.06371
	^{28}Si	$^{286}\text{122}$	113.5230	3.07×10^{21}	1.85×10^{-26}	5.7×10^{-5}	4.084265
	^{30}Si	$^{284}\text{122}$	122.8330	3.33×10^{21}	6.58×10^{-22}	0.219×10^1	-0.4997
	^{32}Si	$^{282}\text{122}$	128.0410	3.47×10^{21}	3.86×10^{-19}	1.34×10^3	-3.28639
	^{34}Si	$^{280}\text{122}$	128.7870	3.49×10^{21}	5.17×10^{-18}	1.80×10^4	-4.41547
	^{36}Ar	$^{278}\text{122}$	139.7520	3.78×10^{21}	1.45×10^{-28}	0.54×10^{-7}	6.101623
	^{38}Ar	$^{276}\text{122}$	150.6850	4.08×10^{21}	4.96×10^{-24}	2.03×10^{-2}	1.533564
	^{40}Ar	$^{274}\text{122}$	157.5300	4.27×10^{21}	2.08×10^{-20}	8.89×10^1	-2.10881
	^{42}Ar	$^{272}\text{122}$	162.8630	4.41×10^{21}	6.86×10^{-19}	3.03×10^3	-3.64088
	^{44}Ar	$^{270}\text{122}$	167.5230	4.54×10^{21}	2.13×10^{-17}	9.65×10^4	-5.14425
	^{40}Ca	$^{274}\text{122}$	152.0060	4.12×10^{21}	2.78×10^{-29}	1.14×10^{-7}	6.781859
	^{42}Ca	$^{272}\text{122}$	162.9270	4.41×10^{21}	4.84×10^{-25}	2.14×10^{-3}	2.510313
	^{44}Ca	$^{270}\text{122}$	172.5680	4.67×10^{21}	7.88×10^{-21}	3.68×10^1	-1.72571
	^{46}Ca	$^{271}\text{122}$	181.2250	4.91×10^{21}	1.54×10^{-18}	7.57×10^3	-4.03888
	^{48}Ca	$^{266}\text{122}$	188.7740	5.11×10^{21}	5.66×10^{-16}	2.89×10^6	-6.62145

TABLE III. Branching ratio of alpha decay with respect to the spontaneous fission, ternary fission, and cluster decay for different isotopes of superheavy nuclei $Z = 122$.

$\lambda\alpha/(\text{SF})$	$\lambda\alpha/(\text{TF})$	$\lambda_\alpha/\lambda_{\text{CR}}$					
		^{12}C	^{14}N	^{20}Ne	^{30}Si	^{40}Ar	^{40}Ca
1.72×10^7	2.45×10^{11}	4.59×10^3	2.01×10^8	1.84×10^8	4.39×10^4	3.37×10^2	1.53×10^9
2.27×10^3	1.09×10^{12}	2.94×10^3	1.26×10^9	1.02×10^8	2.72×10^4	2.49×10^2	1.65×10^9
3.85×10^3	6.42×10^7	6.90×10^2	2.84×10^7	3.52×10^7	1.02×10^4	7.37×10^1	1.57×10^9
1.30×10^4	3.07×10^8	4.99×10^2	9.32×10^7	9.32×10^7	4.52×10^3	5.00×10^1	1.68×10^9
3.07×10^4	2.89×10^7	3.82×10^2	1.15×10^7	7.21×10^6	9.8×10^{-7}	3.95×10^1	3.32×10^{-3}
3.77×10^{37}	3.79×10^8	5.08×10^2	2.06×10^8	5.34×10^6	1.23×10^3	3.03×10^1	1.40×10^9
1.99×10^5	3.23×10^7	1.51×10^3	9.23×10^6	3.06×10^6	6.14×10^2	1.78×10^1	4.58×10^9
5.01×10^5	2.88×10^{10}	2.92×10^3	3.07×10^9	3.41×10^6	5.07×10^2	1.25×10^1	9.70×10^9

TABLE IV. The computed logarithmic half-life values for various decay modes of superheavy nuclei $Z = 122$.

Isotope	Spontaneous fission half-life (yr)	Ternary fission half-life (yr)	Cluster radioactivity (yr)					α decay (yr)	
			^{12}C	^{14}N	^{20}Ne	^{30}Si	^{40}Ar		
$^{307}122$	1.69×10^2	2.40×10^6	4.50×10^{-2}	1.97×10^3	1.81×10^3	4.30×10^{-1}	3.31×10^{-3}	1.50×10^4	9.80×10^{-6}
$^{308}122$	3.76×10^{-2}	1.80×10^7	4.87×10^{-2}	2.08×10^4	1.69×10^3	4.51×10^{-1}	4.12×10^{-3}	2.73×10^4	1.66×10^{-5}
$^{309}122$	1.11×10^{-1}	1.84×10^3	1.98×10^{-2}	8.16×10^2	1.01×10^3	2.93×10^{-1}	2.12×10^{-3}	4.52×10^4	2.87×10^{-5}
$^{310}122$	6.63×10^{-1}	1.56×10^4	2.55×10^{-2}	4.75×10^3	4.75×10^3	2.30×10^{-1}	2.55×10^{-3}	8.55×10^4	5.10×10^{-5}
$^{311}122$	2.84×10^0	2.68×10^3	3.54×10^{-2}	1.07×10^3	6.68×10^2	9.07×10^{-11}	3.66×10^{-3}	3.07×10^{-7}	9.27×10^{-5}
$^{312}122$	6.48×10^{33}	6.51×10^4	8.73×10^{-2}	3.53×10^4	9.16×10^2	2.11×10^{-1}	5.20×10^{-3}	2.40×10^5	1.72×10^{-4}
$^{313}122$	6.45×10^1	1.05×10^4	4.90×10^{-1}	2.99×10^3	9.94×10^2	1.99×10^{-1}	5.78×10^{-3}	1.49×10^6	3.24×10^{-4}
$^{314}122$	3.12×10^2	1.80×10^7	1.82×10^0	1.91×10^6	2.13×10^3	3.16×10^{-1}	7.78×10^{-3}	6.05×10^6	6.24×10^{-4}

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