Absence of spin-ice state in the disordered fluorite Dy₂Zr₂O₇

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Neutron scattering, ac magnetic susceptibility, and specific-heat studies have been carried out on polycrystalline Dy₂Zr₂O₇. Unlike the pyrochlore spin ice Dy₂Ti₂O₇, Dy₂Zr₂O₇ crystallizes into the fluorite structure and the magnetic Dy3+ moments randomly reside on the corner-sharing tetrahedral sublattice with nonmagnetic Zr ions. Antiferromagnetic spin correlations develop below 10 K but remain dynamic down to 40 mK, with a significant amount of magnetic susceptibility. These correlations extend over the length of two tetrahedra edges and grow to six nearest neighbors with the application of a 20-kOe magnetic field. Magnetic heat capacity revealed a correlation peak at 2 K, but no Pauling's residual entropy was observed. We propose that the disorder precludes the development of spin-ice correlations seen in the chemically ordered Dy₂Ti₂O₇ compound, with fluctuating spins in a disordered, liquidlike state (albeit slow) which do not freeze into a canonical spin-glass state that one might intuitively expect.

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I. INTRODUCTION

In a crystalline material containing magnetic ions, the combination of the lattice structure, the nature of the spin, and the couplings between spins can thwart the satisfaction of all nearest-neighbor magnetic interactions simultaneously. This phenomenon is known broadly as geometrically frustrated magnetism [1-3]. Over the past two decades significant advances in our understanding of this magnetism have come from studies on the rare-earth series of pyrochlore titanates, $R_2\text{Ti}_2\text{O}_7$. These model magnets, where the magnetic rareearth ions (R) are distributed on a sublattice of corner-linked tetrahedra, have exhibited a remarkable variety of behavior including spin ice (R = Ho, Dy) [4,5], spin liquid (R = Tb)[6,7], partial order (R = Gd) [8,9], and fluctuation-induced magnetic order (R = Er) [10]. Frustration is also relevant in the Spin-1 chain system CsNiCl3 and the formation of stripelike phases in perovskites [11,12] and plays a critical role outside of magnetism in areas like protein folding and determining the structure of solid nitrogen and water [1,2].

The availability of large single crystals over most of the lanthanide series of pyrochlore titanates resulted in several transformative works over the past 20 y [7,13–16]; however, the open lattice of the oxide pyrochlores, with the chemical formula $R_2M_2O_7$, allows for a large number of combinations of R and M ions. Here the R site is occupied by a trivalent rare-earth ion with eightfold oxygen coordination and the M site by a tetravalent transition-metal ion with sixfold oxygen coordination. Both cations are located at the vertices of two distinct networks of corner-sharing tetrahedra and if either R or M is magnetic, frustration can develop. When considering the chemical bonding, the pyrochlore structure can be described as an ordered defect fluorite [17,18], and recent work on several disordered pyrochlores have revealed exotic magnetism [19,20]. Literature on the zirconate lanthanide series $R_2 Zr_2 O_7$ has reported a structural transition from pyrochlore to defect fluorite, which involves disorder over the R and M sites [21–23]. In disordered fluorites, each of the seven oxygen anions possesses a fourfold coordination of four random cations. The stability field maps proposed in Refs. [2,17,21] evidence the regions in which we could distinguish pyrochlores from the defect fluorite. At the atomic level, Shamblin et al. [24] reported that the disordered fluorite characterizing some zirconates presents local order of weberite type valid for at least two unit cells.

Theoretically, it was predicted that bond disorder on the pyrochlore lattice induces spin-glass behavior at very small concentrations [25]. Sen and Moessner [26] also predicted frozen spin states in disordered spin glasses. More recently a disorder-induced, quantum-entangled liquid phase was predicted in non-Kramers ions based oxide pyrochlores [27]. Technological applications of pyrochlore materials are extensive, ranging from the immobilization of active nuclear waste to high-temperature thermal barrier coatings and from luminescence to solid-oxide fuel [1,2,17].

A central topic in geometrically frustrated magnetism embraces large, Ising spins on the pyrochlore lattice with ferromagnetic interactions. Dy₂Ti₂O₇ is one such pyrochlore oxide, which possesses strong Ising anisotropy along the local (111) directions and ferromagnetic nearest-neighbor interactions [5]. Here, the spins on each tetrahedron randomly satisfy the two-in two-out ice rule; this arrangement of magnetic

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moments can be directly compared to the bonding of hydrogen in hexagonal water ice (I_h) and led to the classification of the spin-ice ground state almost two decades ago [4,28]. This spin state also possesses Pauling's residual entropy equal to $(R/2) \ln(3/2)$ again similar to I_h [4,29]. Recently, experiments have suggested the amount of residual entropy is significantly reduced from the Pauling entropy in Dy₂Ti₂O₇, suggesting that the spin system may order if equilibrated properly [30] and consistent with the spin-ice model which predicts an ordered ground state at the lowest temperatures [31]. In the seminal works by Snyder et al. [32,33], the real and imaginary part of the dynamic susceptibility suggested an exotic crossover from classical to quantum and back to classical relaxation phenomena as spin ice freezes. In their study, χ'_{ac} approaches zero as the temperature is lowered towards absolute zero [33]. Further studies [14,32–36] of the dynamical processes in spin ice have confirmed the existence of low-temperature spin dynamics and identified some of the processes. Confirming the basic near-neighbor and the all-encompassing dipolar spin-ice theories [5], these data observe spin flipping through the first excited crystal-field level. Reducing the temperature below 2 K, neutron diffraction observes the diffuse scattering that can be described well by a disordered spin system with local ice rules [14]. At these low temperatures, a crossover into the spin-ice state occurs with the appearance of a plateau in the [111] isothermal magnetization [37,38] and the reemergence of a thermally activated relaxation process [33,39] that persists down to 20 mK [32-35,40]. These spin dynamics below 5 K can be explained in the framework of the creation and propagation of pointlike defects or emergent magnetic monopoles (a pair of monopoles is produced by the flip of a single spin, giving rise to two neighbor tetrahedra with threein one-out and three-out one-in configurations) interacting via Coulomb potential [15,36,41,42].

To understand the spin-ice state further and in an attempt to influence the magnetic monopoles, researchers have begun to manipulate the chemistry, creating quantum, dilute, and stuffed spin ices [16,27,28,43,44]. For example, researchers are looking outside the structure-field map to make new spin-ice materials [45], are placing smaller moments in the lattice [46], or looking beyond the typical nonmagnetic M-site ions where the structure and availability of crystals is more problematic [20,47,48]. Recently, several groups have been working on geometrically frustrated hafnates and zirconates [20,47–49]. The experimental work is sparse and the lowtemperature magnetism for some rare-earth cations is absent. For example; in the non-Kramers pyrochlore Pr₂Zr₂O₇, spinice-like correlations and strong quantum fluctuations were reported [47,49], but the nature of its ground state is still not fully understood. Initial studies in the pyrochlore Nd₂Zr₂O₇ revealed absence of magnetic order down to 0.5 K [48] and a Curie-Weiss temperature of +0.15 K. Later neutronscattering results indicated a coexistence of an ordered and a fluctuating Coulomb phase at low temperatures [50,51]. Additionally, Tb₂Hf₂O₇ and Pr₂Hf₂O₇ were found not to show long-range magnetic order down to 100 mK [20,52] although Pr₂Hf₂O₇ may freeze into a glass state at 90 mK [53]. Nd₂Hf₂O₇ was found to exhibit a long-range antiferromagnetic order below $T_{\rm N} \approx 0.55$ K with an all-in/all-out spin arrangement [54].

Here, we report low-temperature thermodynamic studies and neutron diffraction on dysprosium zirconate, Dy₂Zr₂O₇, and discuss its magnetic and structural properties. The trivalent Dy ion is the only magnetic species in the compound, similar to that in the spin ice, Dy₂Ti₂O₇. However, it is known that the small lanthanide elements (including Dy³⁺) favor a disordered fluorite structure, rather than the ordered pyrochlore oxide of spin ice [17,21,22,55]. Comparing the two dysprosium compounds, we observe a few thermodynamic similarities; however, the spin-ice character is completely absent in Dy₂Zr₂O₇ and is replaced by spin-liquid characteristics. Very dynamic, short-range, antiferromagnetic correlations are observed below 10 K, reminiscent of those found in terbium pyrochlores and described well by the Gardner-Berlinsky (GB) model [20,56].

II. EXPERIMENTAL

Polycrystalline powder sample of Dy₂Zr₂O₇ was prepared by the sol-gel method. As an alternative to the usual solidstate or "shake and bake" reaction, the wet-chemistry, lowtemperature technique is known to produce excellent atomiclevel mixing, and greater control over particle morphology and size [57]. Dysprosium oxide, Dy₂O₃, and tetrabutyl zirconate, C₁₆H₃₆O₄Zr, were employed as precursors of the dissolutions of the cations Dy^{3+} and Zr^{4+} , respectively. The solution was stirred until it changed into a gel, and finally the resulting gel was calcinated at 1100 °C for 24 h, as detailed in Refs. [57,58]. The same method was employed for preparing Dy₂Ti₂O₇, and the nonmagnetic Lu₂Zr₂O₇ used for specific-heat analysis. X-ray powder-diffraction experiments were performed using a Shimadzu XRD-7000 diffractometer. The Bragg-Brentano geometry and Cu K α_1 radiation (1.5406 Å) were used. The crystal structure was characterized by performing a least-squares Rietveld refinement of the powder x-ray-diffraction data, using the FULLPROF software suite [59] and the graphical interface WINPLOTR [60].

Magnetic measurements were carried out using the superconducting quantum interference device (Quantum Design) down to 1.8 K. The ac magnetic susceptibility (χ_{ac}) data were collected on an adiabatic demagnetization refrigerator (Cambridge Cryogenics) with an amplitude and phase compensator circuit in the frequency range from 10 Hz to 10 kHz in a modulation field of 0.5 Oe. Specific-heat experiments were conducted using the calorimeter insert of a Physical Property Measurement System (Quantum Design) operating with a dilution refrigerator to reach 50 mK. To reduce the effect of the high neutron absorption cross section of $^{161}Dy(600\times10^{-28}~m^2)$ and $^{163}Dy(2840\times10^{-28}~m^2),~^{162}Dy_2Zr_2O_7$ was also synthesized by the sol-gel method using the 98% enriched ¹⁶²Dy₂O₃ for neutron studies. Data were collected at the high-intensity neutron diffractometer, WOMBAT [61] at the Australian Nuclear Science and Technology Organization (ANSTO, Sydney) using 4.744-Å neutrons. The 300-mg sample was placed in an oxygen-free copper can and mounted to the end of a dilution cryostat to reach a temperature of 40 mK.

III. RESULTS AND DISCUSSION

X-ray powder diffraction found all samples to be single phase without any detectable impurity. Modeling the data with

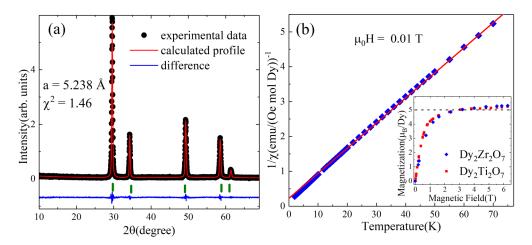


FIG. 1. (a) X-ray powder diffraction and calculated profile data for $Dy_2Zr_2O_7$ at room temperature. Peak positions of the fluorite structure are marked with small vertical lines. (b) Temperature dependence of the inverse susceptibility and Curie-Weiss fit to the lowest temperature. Inset: saturation magnetization as a function of the applied field shows a saturation moment of approximately 5 μ_B/Dy ion (dashed line).

the cubic space group, $Fm\bar{3}m$, consistent with the defectfluorite structure resulted in an excellent fit as shown in Fig. 1(a). The data were not described as well with the oxide pyrochlore structure and space group, $Fd\bar{3}m$. Neutron diffraction on ¹⁶²Dy₂Zr₂O₇ confirmed the x-ray Rietveld refinement and found a cubic lattice constant of value a = 5.238 (2) Å. This structure has one cation site and one oxygen site that is 7/8 occupied. In the defect-fluorite structure, 8 O atoms locally form a perfect cube around both metal ions. This is significantly different from the local coordination of the rare-earth site in the oxide pyrochlore structure, where six oxygen atoms form a puckered hexagon with the other two oxygen atoms forming one of the shortest rare-earth oxygen bonds known, along the local (111) direction. This change in the local environment should result in a strong modification to the crystal electric field states.

Fitting the Curie-Weiss law to the inverse susceptibility of $Dy_2Zr_2O_7$ in the paramagnetic regime above 100 K yields an antiferromagnetic Curie-Weiss temperature $\Theta_{CW}=-10.8$ K. However, if we fit in the linear regime between 10 and 100 K, as shown in Fig. 1(b), this changes to $\Theta_{CW}=-3.2$ K. A ferromagnetic $\Theta_{CW}=+0.5$ K was reported for the spin ice $Dy_2Ti_2O_7$ [5]. The inset of Fig. 1(b) shows the saturation magnetization as a function of the applied field of $Dy_2Zr_2O_7$. The saturation moment at 7 T is close to 5 μ_B/Dy ion similar to $Dy_2Ti_2O_7$, which is the half the free-ion value and due to its crystal-field anisotropy [62]. These data indicate that $Dy_2Zr_2O_7$ has dominant antiferromagnetic interactions and easy-axis anisotropy.

Measurements of the real part of the ac magnetic susceptibility, χ'_{ac} , reveal a frequency-dependent maximum at $T'\approx 1$ K, as shown in Fig. 2. The shape and height of these maxima are similar to those observed in the pyrochlore spin ice $\mathrm{Dy_2Ti_2O_7}$ [32,34] but at a slightly lower temperature. The drop in χ'_{ac} after the maximum indicates that the spins response is slow and they are not able to keep up with the time-varying magnetic field. Contrary to what is observed in $\mathrm{Dy_2Ti_2O_7}$, the values of χ'_{ac} do not vanish below 0.5 K, revealing an incomplete spin freezing of the system and the

ubiquitous presence of persistent spin dynamics seen in many geometrically frustrated magnets. Low-temperature spin dynamics is often recorded by muon spin relaxation; a sensitive, local probe of magnetism, but here we are not implanting an energetic particle and are measuring the bulk. This large susceptibility strongly suggests the system remains dynamic in a liquidlike state down to the lowest measured temperatures. Indeed, muon studies on spin ice are contentious, with Bramwell et al. [63] suggesting exotic monopole dynamics and magnetricity, while others [64,65] have disagreed and suggested the signal was the true signature of slow persistent spin dynamics or lattice ringing. The maximum close to 1 K shifts toward higher temperatures and becomes broader as the frequency of the ac measurement increases. We can characterize the dynamics of Dy₂Zr₂O₇ by fitting the frequency, f, versus the temperature, T', of the maximum in χ'_{ac} to an Arrhenius law $f = f_0 \exp(-E_b/k_BT')$, where E_b is the energy barrier. The entire temperature and frequency range

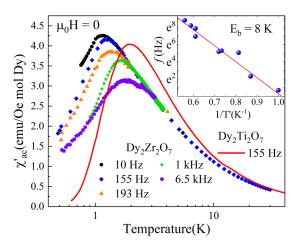


FIG. 2. Real part of the ac magnetic susceptibility χ'_{ac} versus temperature for different frequencies in zero magnetic field for $Dy_2Zr_2O_7$ and $Dy_2Ti_2O_7$ (red solid line). Inset shows the frequency versus the inverse of the temperature of the maximum in χ'_{ac} with a fit to the Arrhenius law.

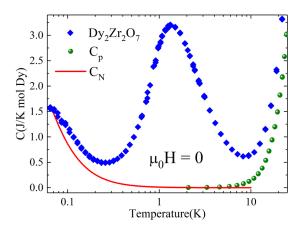


FIG. 3. Temperature dependence of the total specific heat of $Dy_2Zr_2O_7$ at zero field. The lattice term (C_P) was obtained from measurements on nonmagnetic $Lu_2Zr_2O_7$, and the nuclear specific heat contribution (C_N) computed as explained in the text.

measured (see inset of Fig. 2) is described well by the Arrhenius law revealing a single characteristic relaxation time of $\tau_0 = /2\pi\,f_0 = 4.5\times 10^{-5}$ s and an energy barrier $E_b = 8$ K. These values are close to parameters obtained in Dy₂Ti₂O₇ below 1 K in the spin-ice phase, which are an energy barrier $E_b \approx 10$ K and a characteristic relaxation time on the scale of 10^{-7} s [32,34,42,63]. The frequency shift of the χ'_{ac} maximum per decade frequency $\lambda = \Delta T'/(T'\Delta \log f) \approx 0.28$ is two orders of magnitude higher than the expected for a canonical spin glass ($\lambda = 0.005-0.01$) [66], indicating a different kind of sluggish dynamics in our sample.

Specific-heat experiments were performed down to 70 mK for both $Dy_2Zr_2O_7$ and $Dy_2Ti_2O_7$ (data not shown for the latter). Figure 3 shows the temperature dependence of the total specific heat (C), the lattice term C_p accounted for by measuring the nonmagnetic fluorite Lu₂Zr₂O₇, and the nonnegligible nuclear contribution [67,68] for Dy₂Zr₂O₇ at the lowest temperatures. This nuclear term arises from the nuclear magnetic moment, I = 5/2 of two isotopes, ¹⁶¹Dy and ¹⁶³Dy, with quadrupole and hyperfine interactions as discussed by Henelius et al. [67]. The specific-heat data of Dy₂Zr₂O₇ show a broad peak at about 2 K, and shares some resemblance to the quintessential spin ice Dy₂Ti₂O₇ [4], and is associated with short-range magnetic correlations. After isolating the electronic specific heat C_e , The recovered entropy $\Delta S_e(T)$ was obtained by integrating $C_{\rm e}(T)/T$ from the minimum measured to the temperature T. Figure 4 shows that the values of $\Delta S_{\rm e}(T)$ for the Dy₂Zr₂O₇ is close to the expected $R \ln(2)$, for a system with only two discrete orientations, at approximately 8 K. This indicates that the residual entropy left as T goes to zero is small, if not negligible, and significantly less than the $(R/2)\ln(3/2)$ found in spin ice $(Dy_2Ti_2O_7)$ and water [69], and understood by Pauling for water ice [29].

A well-characterized isotopically enriched, powder sample was used for neutron-diffraction investigations. The bulk thermodynamic properties and the x-ray powder-diffraction pattern of ¹⁶²Dy₂Zr₂O₇ were consistent with our other samples. Neutron diffraction confirms that there is no long-range order down to 39 mK, but revealed broad diffuse scattering below 10 K, associated with the correlated spins. Data measured

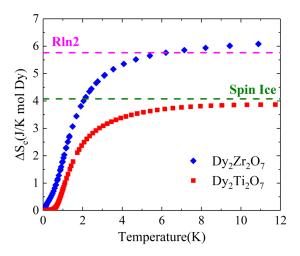


FIG. 4. Recovered electronic entropy $\Delta S_{\rm e}$ as a function of the temperature for Dy₂Zr₂O₇ and Dy₂Ti₂O₇. The dashed lines denote the expected values for Ising spins ($R \ln 2$) and spin ices.

at 10 K revealed a relatively flat background and sharp, resolution-limited Bragg peaks associated with the crystalline structure (not shown). These data were used as a background dataset and subtracted from the lower-temperature data to enhance the magnetic scattering.

Magnetic diffraction datasets at 40 mK are shown in Fig. 5. The net intensity has been corrected for the $|\mathbf{Q}|$ dependence due to the $\mathrm{Dy^{3+}}$ magnetic form factor so that models of possible spin structures can be compared. Although the $|\mathbf{Q}|$ range and the statistical significance of the data are not of the quality necessary to model the diffuse scattering, several observations can immediately be drawn from these data. First, the broad distribution of excess magnetic scattering centered at roughly 1.2 Å⁻¹ is characteristic of antiferromagnetically coupled Ising spins on the corner-sharing tetrahedral lattice [6,7,9,20,56,70-76]. Second, the lack of forward scattering at

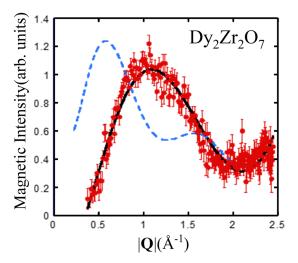


FIG. 5. Magnetic neutron scattering for $Dy_2Zr_2O_7$ at 40 mK, after a dataset at 10 K was subtracted to remove the nonmagnetic background, including that from the crystalline structure. Data (red circles) are plotted against the powder-averaged dipolar spin-ice model (dashed line) and the Gardner-Berlinsky model (solid line).

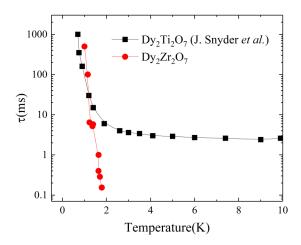


FIG. 6. Characteristic spin-relaxation time as a function of temperature in $Dy_2Zr_2O_7$ and $Dy_2Ti_2O_7$ (taken from Snyder *et al.* [66]). The spin-relaxation time within $Dy_2Zr_2O_7$, closed circles, increases rapidly below 2 K, and match those in the spin-ice phase of $Dy_2Ti_2O_7$.

low $|\mathbf{Q}|$ indicates the absence of ferromagnetic correlations and is consistent with our negative Curie-Weiss temperature. Finally, the data appear to reach a second maximum at $\sim 2.3 \text{ Å}^{-1}$. All these observations are reminiscent of those from the pyrochlore Tb₂Mo₂O₇ [72] where an antiparallel alignment of adjacent spins is observed on both the R and M sublattice and the disordered pyrochlore CsNiCrF₆ [76]. The width of the peak at the antiferromagnetic correlations wave vector of 1.2 Å^{-1} suggests the mean correlation length is that of the two M-M bonds in the fluorite structure (5.24 Å). To understand the correlations in Dy₂Zr₂O₇ further, we also plot the calculated powder-averaged structure factor for spin ice [5,28] and that for near-neighbor antiferromagnetic correlations on the pyrochlore lattice [56]. The metal-ion tetrahedra still exist, albeit disordered, in this fluorite lattice and with nearest-neighbor correlations only it is not unreasonable to begin with these models. The near-neighbor antiferromagnetic correlations on the pyrochlore lattice, so-called Gardner-Berlinsky model, was recently used to describe the magnetic correlations in the disordered pyrochlore Tb₂Hf₂O₇ [20] and originally used to model the spin liquid, Tb₂Ti₂O₇ [56]. Clearly, spin-ice correlations are not present in this Dy compound. At first glance, the data are described well by the GB model; however, the poor statistics makes it difficult to precisely model the high- $|\mathbf{Q}|$ data, where we see a peak at 2.3 Å^{-1} and the model predicts the second peak closer to $3.0 \,\text{Å}^{-1}$. Considering the extensive disorder in the fluorite structure, the powder-averaged data, and the limited amount of momentum space covered, no further attempts were made to model these data and add to the number of variables.

To stress the surprising similarity in the low-temperature relaxation processes within the ferromagnetic, pyrochlore, spin ice, and this antiferromagnetic coupled fluorite, we show in Fig. 6 the temperature dependence of the characteristic relaxation time τ . For the fluorite the relaxation time was obtained from the dynamic susceptibility measured below 2 K, taken as the reciprocal of the measuring frequency at the peak

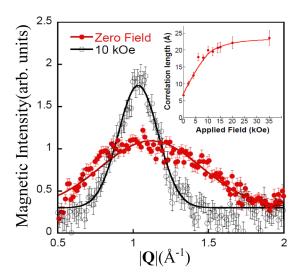


FIG. 7. Spatial spin-spin correlations as a function of applied magnetic field at 40 mK. Main panel: Magnetic diffraction at 0 and 10-kOe Gaussian fits with a common instrumental background are used to describe the data. Inset: Field dependence of the correlation length determined from the Gaussian width of the broad maxima seen around $|\mathbf{O}| \approx 1.15 \text{ Å}^{-1}$.

temperature. In this temperature range the spatial spin-spin correlations are known to exist from our neutron-diffraction studies (Fig. 5); the strong temperature dependence and the characteristic timescale match that of $Dy_2Ti_2O_7$ [33,77]. This similarity extends deep into the spin-ice phase of $Dy_2Ti_2O_7$ where monopole excitations are created and move throughout the lattice. This behavior of the time evolution of our system is also analogous to that exhibited in the antiferromagnetic Er^{3+} -based spinels, $CdEr_2X_4(X = Se, S)$, which was also interpreted as the existence of monopole dynamics [78,79]. At higher temperatures, no quantum tunneling regime was seen in $Dy_2Zr_2O_7$, as reported for $Dy_2Ti_2O_7$ [40,41].

In the presence of magnetic field the shape of the low-angle scattering visibly changed. When 10 kOe are applied, as shown in Fig. 7, the broad scattering sharpens up, but remains centered at 1.15(7) Å⁻¹. Fitting these, and other data, results in the curve shown in the inset to Fig. 7. Here we plot the correlation length calculated from the full width at half maximum of the broad diffuse scattering. The observed short-range, spin-spin correlations lengthen in a field, but appear to saturate above 15 kOe. This may be a plateau that extends to 35 kOe but more studies at higher fields are necessary. Within this field region the correlations extend to 23.6(8) Å, the length of approximately six nearest neighbors or five unit cells.

IV. CONCLUSION

The oxide pyrochlore structure can be considered an ordered $2 \times 2 \times 2$ superstructure of the disordered defect fluorite. We have used wet chemistry and low-temperature annealing to produce a well-mixed, disordered fluorite, $Dy_2Zr_2O_7$. Here we expect the metal ions to decorate two interpenetrating sublattices of corner-sharing tetrahedra, with a statistical distribution of Dy-Dy, Zr-Zr, and Dy-Zr bonds all with the same bond length, but only Dy^{3+} is magnetic. Trivalent dysprosium

pyrochlore oxides ($Dy_2Ti_2O_7$ [4,5] and $Dy_2Sn_2O_7$ [80]) have both shown spin-ice character. Here we have shown that Dy₂Zr₂O₇, with its disordered lattice, but similar moment size, near-neighbor distance, and spin anisotropy is a spin -liquid candidate; however, the presence of a different glassy state cannot be ruled out. Neutron diffraction and specific heat revealed short-range spin-spin correlations below 2 K or $|\Theta_{CW}|$, and no long-ranged order down to 40 mK. These spin correlations are antiferromagnetic, very short range, extending over next-nearest neighbors only. Unlike the classical spin ices, susceptibility and specific heat indicate the presence of significant spin dynamics at the lowest temperatures and absence of residual entropy. Below 2 K, neutron diffraction reveals significant spin-spin spatial correlations, or entanglement, while the time correlations measured by χ'_{ac} indicate that the system is very dynamic. These two ingredients are essential for the identification of a quantum spin liquid and indicate that this fluorite could possess such a ground state.

These studies have shown that a significant amount of disorder can lead to a dynamic ground state when combined with frustration at low temperatures. Other geometrically frustrated magnets also show that disorder plays a similar role. With the availability of zirconate single crystals [81] and the possibility of solid solution between the spin-ice and spin-liquid candidate end members, this family seems to be an excellent platform for further investigations, probing the role of extensive disorder on the spin dynamics of pyrochlores and the propagation of monopoles as disorder is induced to the Coulomb phase.

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- [1] H. T. Diep, *Frustrated Spin Systems* (World Scientific, Singapore, 2013).
- [2] J. S. Gardner, M. J. P. Gingras, and J. E. Greedan, Rev. Mod. Phys. 82, 53 (2010).
- [3] J. S. Gardner, J. Phys.: Condens. Matter 23, 160301 (2011).
- [4] A. P. Ramirez, A. Hayashi, R. J. Cava, R. Siddharthan, and B. S. Shastry, Nature (London) 399, 333 (1999).
- [5] B. C. den Hertog and M. J. P. Gingras, Phys. Rev. Lett. 84, 3430 (2000).
- [6] J. S. Gardner, S. R. Dunsiger, B. D. Gaulin, M. J. P. Gingras, J. E. Greedan, R. F. Kiefl, M. D. Lumsden, W. A. MacFarlane, N. P. Raju, J. E. Sonier *et al.*, Phys. Rev. Lett. **82**, 1012 (1999).
- [7] J. S. Gardner, A. Keren, G. Ehlers, C. Stock, E. Segal, J. M. Roper, B. Fåk, M. B. Stone, P. R. Hammar, D. H. Reich *et al.*, Phys. Rev. B 68, 180401 (2003).
- [8] J. D. M. Champion, A. S. Wills, T. Fennell, S. T. Bramwell, J. S. Gardner, and M. A. Green, Phys. Rev. B 64, 140407 (2001).
- [9] J. R. Stewart, G. Ehlers, A. S. Wills, S. T. Bramwell, and J. S. Gardner, J. Phys.: Condens. Matter 16, L321 (2004).
- [10] L. Savary, K. A. Ross, B. D. Gaulin, J. P. C. Ruff, and L. Balents, Phys. Rev. Lett. 109, 167201 (2012).
- [11] M. Kenzelmann, R. A. Cowley, W. J. L. Buyers, R. Coldea, J. S. Gardner, M. Enderle, D. F. McMorrow, and S. M. Bennington, Phys. Rev. Lett. 87, 017201 (2001).
- [12] V. J. Emery, S. A. Kivelson, and J. M. Tranquada, Proc. Natl. Acad. Sci. U.S.A. 96, 8814 (1999).
- [13] J. S. Gardner, B. D. Gaulin, and D. McK. Paul, J. Cryst. Growth 191, 740 (1998).
- [14] T. Fennell, O. A. Petrenko, B. Fåk, S. T. Bramwell, M. Enjalran, T. Yavors'kii, M. J. P. Gingras, R. G. Melko, and G. Balakrishnan, Phys. Rev. B 70, 134408 (2004).
- [15] C. Castelnovo, R. Moessner, and S. L. Sondhi, Nat. Lett. 451, 42 (2008).
- [16] S. R. Giblin, S. T. Bramwell, P. C. W. Holdsworth, D. Prabhakaran, and I. Terry, Nat. Phys. **7**, 252 (2011).

- [17] M. A. Subramanian, G. Aravamudan, and G. V. Subba Rao, Prog. Solid State Chem. 15, 55 (1983).
- [18] C. Karthik, T. J. Anderson, D. Gout, and R. Ubic, J. Solid State Chem. 194, 168 (2012).
- [19] G. C. Lau, R. S. Freitas, B. G. Ueland, B. D. Muegge, E. L. Duncan, P. Schiffer, and R. J. Cava, Nat. Phys. 2, 249 (2006).
- [20] R. Sibille, E. Lhotel, M. C. Hatnean, G. J. Nilsen, G. Ehlers, A. Cervellino, E. Ressouche, M. Frontzek, O. Zaharko, V. Pomjakushin *et al.*, Nat. Commun. 8, 892 (2017).
- [21] P. E. R. Blanchard, R. Clements, B. J. Kennedy, C. D. Ling, E. Reynolds, M. Avdeev, A. P. J. Stampfl, Z. Zhang, and L. Y. Jang, Inorg. Chem. 51, 13237 (2012).
- [22] E. Reynolds, P. E. R. Blanchard, B. J. Kennedy, C. D. Ling, S. Liu, M. Avdeev, Z. Zhang, G. J. Cuello, A. Tadich, and L. Y. Jang, Inorg. Chem. 52, 8409 (2013).
- [23] M. Glerup, O. F. Nielsen, and F. W. Poulsen, J. Solid State Chem. 160, 25 (2001).
- [24] J. Shamblin, M. Feygenson, J. Neuefeind, C. L. Tracy, F. Zhang, S. Finkeldei, D. Bosbach, H. Zhou, R. C. Ewing, and M. Lang, Nat. Mater. 15, 507 (2016).
- [25] T. E. Saunders and J. T. Chalker, Phys. Rev. Lett. 98, 157201 (2007).
- [26] A. Sen and R. Moessner, Phys. Rev. Lett. 114, 247207 (2015).
- [27] L. Savary and L. Balents, Phys. Rev. Lett. 118, 087203 (2017).
- [28] S. T. Bramwell and M. J. P. Gingras, Science 294, 1495 (2001).
- [29] L. Pauling, J. Am. Chem. Soc. 57, 2680 (1935).
- [30] D. Pomaranski, L. R. Yaraskavitch, S. Meng, K. A. Ross, H. M. L. Noad, H. A. Dabkowska, B. D. Gaulin, and J. B. Kycia, Nat. Phys. 9, 353 (2013).
- [31] R. G. Melko, B. C. den Hertog, and M. J. P. Gingras, Phys. Rev. Lett. **87**, 067203 (2001).
- [32] J. Snyder, J. S. Slusky, R. J. Cava, and P. Schiffer, Nature (London) 413, 48 (2001).
- [33] J. Snyder, B. G. Ueland, J. S. Slusky, H. Karunadasa, R. J. Cava, and P. Schiffer, Phys. Rev. B 69, 064414 (2004).

- [34] K. Matsuhira, Y. Hinatsu, and T. Sakakibara, J. Phys.: Condens. Matter 13, L737 (2001).
- [35] J. Lago, S. J. Blundell, and C. Baines, J. Phys.: Condens. Matter 19, 326210 (2007).
- [36] D. J. P. Morris, D. A. Tennant, S. A. Grigera, B. Klemke, C. Castelnovo, R. Moessner, C. Czternasty, M. Meissner, K. C. Rule, J.-U. Hoffmann *et al.*, Science 326, 411 (2009).
- [37] A. L. Cornelius and J. S. Gardner, Phys. Rev. B 64, 060406(R) (2001).
- [38] O. A. Petrenko, M. R. Lees, and G. Balakrishnan, J. Phys.: Condens. Matter 23, 164218 (2011).
- [39] J. P. Clancy, J. P. C. Ruff, S. R. Dunsiger, Y. Zhao, H. A. Dabkowska, J. S. Gardner, Y. Qiu, J. R. D. Copley, T. Jenkins, and B. D. Gaulin, Phys. Rev. B 79, 014408 (2009).
- [40] G. Ehlers, A. L. Cornelius, M. Orendác, M. Kajnaková, T. Fennell, S. T. Bramwell, and J. S. Gardner, J. Phys.: Condens. Matter 15, L9 (2003).
- [41] L. D. C. Jaubert and P. C. W. Holdsworth, Nat. Phys. 5, 258 (2009).
- [42] L. D. C. Jaubert and P. C. W. Holdsworth, J. Phys.: Condens. Matter 23, 164222 (2011).
- [43] C. R. Wiebe, J. S. Gardner, S.-J. Kim, G. M. Luke, A. S. Wills, B. D. Gaulin, J. E. Greedan, I. Swainson, Y. Qiu, and C.Y. Jones, Phys. Rev. Lett. 93, 076403 (2004).
- [44] J. S. Gardner, A. L. Cornelius, L. J. Chang, M. Prager, Th. Brückel, and G. Ehlers, J. Phys.: Condens. Matter 17, 7089 (2005).
- [45] H. D. Zhou, S. T. Bramwell, J. G. Cheng, C. R. Wiebe, G. Li, L. Balicas, J. A. Bloxsom, H. J. Silverstein, J. S. Zhou, J. B. Goodenough *et al.*, Nat. Commun. 2, 478 (2011).
- [46] P. M. Sarte, A. A. Aczel, G. Ehlers, C. Stock, B. D. Gaulin, C. Mauws, M. B. Stone, S. Calder, S. E. Nagler, J. W. Hollett et al., J. Phys.: Condens. Matter 29, 45LT01 (2017).
- [47] K. Kimura, S. Nakatsuji, J.-J. Wen, C. Broholm, M. B. Stone, E. Nishibori, and H. Sawa, Nat. Commun. 4, 1934 (2013).
- [48] M. C. Hatnean, M. R. Lees, O. A. Petrenko, D. S. Keeble, G. Balakrishnan, M. J. Gutmann, V. V. Klekovkina, and B. Z. Malkin, Phys. Rev. B 91, 174416 (2015).
- [49] J.-J. Wen, S. M. Koohpayeh, K. A. Ross, B. A. Trump, T. M. McQueen, K. Kimura, S. Nakatsuji, Y. Qiu, D. M. Pajerowski, J. R. D. Copley *et al.*, Phys. Rev. Lett. 118, 107206 (2017).
- [50] E. Lhotel, S. Petit, S. Guitteny, O. Florea, M. C. Hatnean, C. Colin, E. Ressouche, M. R. Lees, and G. Balakrishnan, Phys. Rev. Lett. 115, 197202 (2015).
- [51] S. Petit, E. Lhotel, B. Canals, M. C. Hatnean, J. Ollivier, H. Mutka, E. Ressouche, A. R. Wildes, M. R. Lees, and G. Balakrishnan, Nat. Phys. 12, 746 (2016).
- [52] V. K. Anand, A. T. M. N. Islam, A. Samartzis, J. Xu, N. Casati, and B. Lake, J. Cryst. Growth 498, 124 (2018).
- [53] V. K. Anand, L. Opherden, J. Xu, D. T. Adroja, A. T. M. N. Islam, T. Herrmannsdörfer, J. Hornung, R. Schönemann, M. Uhlarz, H. C. Walker *et al.*, Phys. Rev. B **94**, 144415 (2016).
- [54] V. K. Anand, A. K. Bera, J. Xu, T. Herrmannsdörfer, C. Ritter, and B. Lake, Phys. Rev. B 92, 184418 (2015).
- [55] M. J. D. Rushton, R. W. Grimes, C. R. Stanek, and S. Owens, J. Mater. Res. 19, 1603 (2004).

- [56] J. S. Gardner, B. D. Gaulin, A. J. Berlinsky, P. Waldron, S. R. Dunsiger, N. P. Raju, and J. E. Greedan, Phys. Rev. B 64, 224416 (2001).
- [57] A. Garbout, S. Bouattour, and A. W. Kolsi. J. Alloys Compd. 469, 229 (2009).
- [58] J. G. A. Ramon, M.Sc. dissertation, Universidade de São Paulo, Brazil, 2015.
- [59] J. Rodríguez-Carvajal, Physica B 192, 55 (1993).
- [60] T. Roisnel and J. Rodríguez-Carvajal, Mater. Sci. Forum 378-381, 118 (2001).
- [61] A. J. Studer, M. E. Hagen, and T. J. Noakes, Physica B 385–386, 1013 (2006).
- [62] H. Fukazawa, R. G. Melko, R. Higashinaka, Y. Maeno, and M. J. P. Gingras, Phys. Rev. B 65, 054410 (2002).
- [63] S. T. Bramwell, S. R. Giblin, S. Calder, R. Aldus, D. Prabhakaran, and T. Fennell, Nature (London) 461, 956 (2009).
- [64] S. R. Dunsiger, A. A. Aczel, C. Arguello, H. Dabkowska, A. Dabkowski, M.-H. Du, T. Goko, B. Javanparast, T. Lin, F. L. Ning et al., Phys. Rev. Lett. 107, 207207 (2011).
- [65] S. J. Blundell, Phys. Rev. Lett. 108, 147601 (2012).
- [66] J. A. Mydosh, Spin Glasses: An Experimental Introduction (Taylor & Francis, London, 1993).
- [67] P. Henelius, T. Lin, M. Enjalran, Z. Hao, J. G. Rau, J. Altosaar, F. Flicker, T. Yavors'kii, and M. J. P. Gingras, Phys. Rev. B 93, 024402 (2016).
- [68] S. R. Giblin, M. Twengström, L. Bovo, M. Ruminy, M. Bartkowiak, P. Manuel, J. C. Andresen, D. Prabhakaran, G. Balakrishnan, E. Pomjakushina *et al.*, Phys. Rev. Lett. 121, 067202 (2018).
- [69] W. F. Giauque and M. F. Ashley, Phys. Rev. 43, 81 (1933).
- [70] S. Guitteny, S. Petit, E. Lhotel, J. Robert, P. Bonville, A. Forget, and I. Mirebeau, Phys. Rev. B 88, 134408 (2013).
- [71] I. Mirebeau, A. Apetrei, J. Rodríguez-Carvajal, P. Bonville, A. Forget, D. Colson, V. Glazkov, J. P. Sanchez, O. Isnard, and E. Suard, Phys. Rev. Lett. 94, 246402 (2005).
- [72] G. Ehlers, J. E. Greedan, J. R. Stewart, K. C. Rule, P. Fouquet, A. L. Cornelius, C. Adriano, P. G. Pagliuso, Y. Qiu, and J. S. Gardner, Phys. Rev. B 81, 224405 (2010).
- [73] J. E. Greedan, J. N. Reimers, C. V. Stager, and S. L. Penny, Phys. Rev. B 43, 5682 (1991).
- [74] A. Apetrei, I. Mirebeau, I. Goncharenko, D. Andreica, and P. Bonville, J. Phys.: Condens. Matter 19, 145214 (2007).
- [75] H. D. Zhou, C. R. Wiebe, A. Harter, N. S. Dalal, and J. S. Gardner, J. Phys.: Condens. Matter 20, 325201 (2008).
- [76] M. J. Harris, M. P. Zinkin, Z. Tun, B. M. Wanklyn, and I. P. Swainson, Phys. Rev. Lett. 73, 189 (1994).
- [77] L. R. Yaraskavitch, H. M. Revell, S. Meng, K. A. Ross, H. M. L. Noad, H. A. Dabkowska, B. D. Gaulin, and J. B. Kycia, Phys. Rev. B 85, 020410(R) (2012).
- [78] S. Gao, O. Zaharko, V. Tsurkan, L. Prodan, E. Riordan, J. Lago, B. Fåk, A. R. Wildes, M. M. Koza, C. Ritter *et al.*, Phys. Rev. Lett. 120, 137201 (2018).
- [79] G. C. Lau, R. S. Freitas, B. G. Ueland, P. Schiffer, and R. J. Cava, Phys. Rev. B 72, 054411 (2005).
- [80] K. Matsuhira, Y. Hinatsu, K. Tenya, H. Amitsuka, and T. Sakakibara, J. Phys. Soc. Jpn. 71, 1576 (2002).
- [81] M. C. Hatnean, M. R. Lees, and G. Balakrishnan, J. Cryst. Growth 418, 1 (2015).