Reentrant magnetism at the borderline between long-range antiferromagnetic order and spin-glass behavior in the *B*-site disordered perovskite system Ca_{2-x}Sr_xFeRuO₆

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We report on the coexistence of magnetic order and disorder in the atomically disordered double perovskites Ca₂FeRuO₆ and CaSrFeRuO₆. Powder x-ray and neutron diffraction were used to investigate the crystal structure and magnetic ordering of these oxides. Both compounds are described by the orthorhombic space group Pbnm down to 3 K, where the B site is found to be statistically occupied by Fe^{3+} and Ru^{5+} ions. The compound Ca_2FeRuO_6 shows a G-type antiferromagnetic ordering at $T_N \approx 220$ K, where the moments are aligned parallel to the c axis. The exchange of Ca by Sr suppresses long-range ordering in this system with the consequence that CaSrFeRuO₆ shows a diffuse scattering pattern, indicating only the presence of a short-range order of the magnetic moments. Mössbauer measurements additionally reveal the coexistence of a long-range ordered and paramagnetic phase in Ca₂FeRuO₆, and spin-glass behavior in CaSrFeRuO₆. The random occupancy of iron and ruthenium atoms at the B site gives rise to locally varying competing magnetic exchange interactions, which favors the emergence of reentrant magnetism with a spin-glass-like transition at $T_f \approx 87 \,\mathrm{K}$ for Ca₂FeRuO₆ and a spin-glass transition at $\sim 65 \text{ K}$ for CaSrFeRuO₆, as evidenced by frequency dependent ac susceptibility measurements. Our results are an interesting example for crossing the borderline between antiferromagnetism and spin-glass behavior in a 3d-4d hybrid perovskite system by modifying the structural distortion associated to the tolerance factor of the perovskite structure rather than changing the concentration of magnetic ions.

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I. INTRODUCTION

Perovskite related compounds having the general formula ABO₃, with diverse combinations of A (alkaline earth or rareearth metal) and B (transition metal or smaller lanthanide) cations, continue to attract long-term research attention due to their large variety of structural and functional properties [1-10]. Unpaired d electrons of transition metals introduce useful properties such as ferromagnetism, multiferroicity, colossal magnetoresistance, and magnetoimpedance [5-18]. The discovery of colossal tunneling-type magnetoresistance in the Sr₂FeMoO₆ double perovskite (general formula of double perovskites: $A_2BB'O_6$) also has attracted great attention in solid-state research as this material can be useful for spintronic applications [19]. The combination of two different transition-metal atoms (labeled as B and B') in perovskite related oxides often leads to competing exchange interactions within and between the B and B' sublattices [20-32]. Unusual properties are particularly found due to interactions between strongly correlated 3d electrons and less correlated 4d or 5d

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electrons, where additionally spin-orbit coupling is considered as essential.

The crystallographic arrangement of different atoms B and B' in the octahedral site strongly influences the structural and physical properties of double perovskites. The degree of atomic order depends on the charge and radius of the Band B' as well as on the A atoms. From earlier reports it is well established that with comparable sizes of B and B' a compound can adopt structure types with disordered cation arrangements at the B site while for a charge difference of more than 2 between them an ordered arrangement of Bcations is more likely to be formed [20,29-40]. In order to investigate new double perovskites, the combination of 3dtransition metals with 4d and 5d transition metals can be a preferential choice as it combines narrow spin-polarized bands with wider partially filled bands [41]. Previous efforts produced a series of osmium-based double perovskites with unique structural and electronic properties [34-40]. The choice of ruthenium in place of osmium can be fruitful for generating new properties as, while retaining the same number of unpaired d electrons, ruthenium ions have a smaller ionic radius than the corresponding osmium ions and a spin-orbit coupling strength which is intermediate between that of 3dand 5d ions.

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Due to the substantial itinerant character of the 4d electrons, pure ruthenium-based oxides reveal unusual electronic properties at the verge between metallic and insulating behavior [42,43]. A well-studied example presents the Ruddlesden-Popper series $Sr_{n+1}Ru_nO_{3n+1}$. Whereas the $n = \infty$ member SrRuO₃ was the first ferromagnetic 4d oxide having an ordering temperature of 160 K [44], the n = 1 member Sr₂RuO₄ has attracted a lot of interest as an unconventional superconductor [45,46]. The n = 2 member $Sr_3Ru_2O_7$ is a paramagnetic metal with quantum critical behavior [47]. Within the Ruddlesden-Popper series the ruthenium ions are in the +4-oxidation state and the electronic properties evolve from the $4d^4(t_{2g}^4)$ low-spin configuration. The electronic state of ruthenium-based double perovskites, which often contain Ru^{5+} ions ($t_{2\rho}^3$ configuration), is more localized and the magnetic states frequently reveal magnetic frustration and glassy characteristics [48–54]. The observation of two-dimensional magnetic correlations and partial long-range magnetic ordering in geometrically frustrated Sr₂YRuO₆ [55] has motivated us to synthesize further magnetically frustrated rutheniumbased compounds and to study their magnetic properties.

Not only the dimensionality of the transition-metal oxide lattice but also cationic substitution at the A site of the perovskite structure is a convenient way for tuning electronic and magnetic properties. If, for instance, the Sr²⁺ ion (ionic radius 1.44 Å for 12-fold coordination) is replaced by the smaller Ca²⁺ (1.34 Å) ion, the tolerance factor t {where $t = (r_A + r_O)/[\sqrt{2}(r_A + r_B)]$, where r_A , r_B , and r_O are the ionic radii of A, B and O, respectively} decreases, which causes large structural distortion with different ground states, strange magnetic ordering, and electronic transitions [48,55– 58]. In double perovskites r_B corresponds to the average ionic radius of the B and B' ions. Here, we focus on the system A_2 FeRuO₆. The Sr member, Sr₂FeRuO₆ [52], as well as phases with minor percentage of Ba and Ca substitution [59], were shown to be atomically disordered spin-glass (SG) systems lacking long-range magnetic order. By contrast, an artificial ordered superlattice of composition Sr₂FeRuO₆ was reported to be ferro- or ferrimagnetic with $T_{\rm C}$ as high as 390 K [48]. The behavior of the Ca analog Ca₂FeRuO₆ and of Ca rich phases in the system $Sr_{2-x}Ca_xFeRuO_6$ is still unknown [20]. The crucial importance of the alkaline-earth ion for tuning the magnetostructural properties of double perovskites was clearly established for the corresponding much more ordered system A_2 FeOsO₆, where a monoclinic crystal structure and ferrimagnetism ($T_{\rm C} \sim 320 \,{\rm K}$) [21], a tetragonal crystal structure and antiferromagnetism ($T_N =$ 140 K) [35], and a hexagonal crystal structure with ferrimagnetism ($T_{\rm C} \sim 370 \,{\rm K}$) [60] were found for $A = {\rm Ca}$, Sr, and Ba, respectively. In the present paper, we report our systematic study of the crystal structure and magnetic properties of the two B-site disordered perovskites Ca₂FeRuO₆ (I) and CaSrFeRuO₆ (II). Here, the double perovskite notations were chosen as for the earlier reported compound Sr_2FeRuO_6 [52] although the transition-metal sites in contrast to the Os analogs are disordered [actually, CaFe_{0.5}Ru_{0.5}O₃ (I) and Ca_{0.5}Sr_{0.5}Fe_{0.5}Ru_{0.5}O₃ (II) in a simple perovskite notation]. Powder x-ray, high-resolution synchrotron, and neutron-diffraction studies were carried out with variation of temperature to investigate the crystal structure and cation disorder. Neutron-diffraction, Mössbauer, and ac and dc susceptibility measurements were performed to establish the magnetic structures and ordering behavior which reflect competing magnetic exchange interactions. It will be shown that due to the increase in tolerance factor on going from compound I to compound II a change from a long-range ordered antiferromagnetic state with reentrant magnetic behavior to pure spin-glass behavior is induced. The magnetism of the system $Ca_{2-x}Sr_xFeRuO_6$ shows features, which can be compared with those of diluted magnetic systems [61,62] where reentrant magnetism with glassy properties below the magnetic ordering temperature was observed near the percolation threshold. Unusual coexistence of long-range order with spin-glass-like freezing was reported also for several other systems with atomic disorder [63–67].

II. EXPERIMENTAL DETAILS

Synthesis of polycrystalline powders of Ca_2FeRuO_6 (I) and $CaSrFeRuO_6$ (II) was carried out by a solid-state method from the stoichiometric ratio of binary oxides. The reagents $CaCO_3/SrCO_3$ (Sigma Aldrich, 99%), RuO_2 (Sigma Aldrich, 99.9%), and Fe_2O_3 (Sigma Aldrich, 99%) were used without any further purification. A typical bunch of reagents consisting of a stoichiometric mixture (2:1:0.5) of $CaCO_3$, RuO_2 , and Fe_2O_3 for compound I and (1:1:1:0.5) of $CaCO_3$, $SrCO_3$, RuO_2 , and Fe_2O_3 for compound II was ground for 2 h. After grinding, the powders were pressed into pellets under a pressure of 10 bars. Pressed pellets were first fired at 980 °C for 15 h and then again at 1280 °C for 15 h after intermediate grinding.

At room temperature, powder x-ray-diffraction (PXRD) studies were carried out using a Rigaku diffractometer with a rotating anode Cu-K α source. The PXRD patterns were collected from 15 to 120° in steps of 0.02° with a counting time of 4 s per step. In order to investigate the structural properties with better accuracy, we have performed highresolution synchrotron powder-diffraction studies at the beam line BM1A (Swiss-Norwegian) at the ESRF in Grenoble using the x-ray wavelengths $\lambda = 0.6973$ and 0.7152 Å. For this experiment, a powder sample was mounted in a glass capillary of diameter 0.3 mm. A full powder pattern of this sample was collected at 295 K in the 2θ range between 1 and 50°. In the powder patterns, no impurities were found. In order to find possible structural changes in the magnetically ordered state we have collected another powder pattern at 100 K in the 2θ range between 1 and 42° .

dc magnetization (*M*) was measured as a function of temperature ($2 \le T \le 800 \text{ K}$) and magnetic field ($-9 \le H \le +9 \text{ T}$) using the vibrating sample magnetometer (VSM) attachment to the Physical Property Measurement System (PPMS, Quantum Design). For the high-temperature ($T \ge 400 \text{ K}$) measurements, a high-*T* oven was used. ac susceptibility was measured as a function of *T* and frequency (ν) using the ACMS option of the PPMS. The dc resistivity (ρ) was measured as a function of temperature on a small rectangular pellet in the PPMS using the standard four-probe technique.

Neutron powder-diffraction experiments of both the compounds were carried out on the instruments E6 and E9 at the BER II reactor of the Helmholtz-Zentrum Berlin. The instrument E9 uses a Ge monochromator selecting the neutron wavelength $\lambda = 1.3087$ Å, while the instrument E6 uses a pyrolytic graphite monochromator selecting the neutron wavelength $\lambda = 2.43$ Å. In order to investigate the crystal structures in detail, low-temperature powder patterns were recorded on the instrument E9 between the diffraction angles 7.5 and 136.5°. The variations of the crystal and magnetic structures of both compounds as a function of temperature were investigated on the instrument E6, where powder patterns were collected between the diffraction angles 5 and 141.8°. Rietveld refinements of the powder-diffraction data were carried out with the program FULLPROF [68]. For the x-ray data we used the atomic scattering factors provided by this program. For the refinement of the neutron powder data, the nuclear scattering lengths b(O) = 5.805 fm, b(Ca) =4.70 fm, b(Fe) = 9.54 fm, and b(Ru) = 7.03 fm were used [69]. The magnetic form factors of the Fe and Ru atoms were taken from elsewhere [70].

⁵⁷Fe-Mössbauer spectra were collected between 5.5 and 291 K using a standard WissEl spectrometer operated in the constant acceleration mode (⁵⁷Co/Rh source) and a Janis SHI 850–5 closed cycle refrigerator. The sample powders were mixed with boron nitride and homogeneously distributed in an acrylic glass sample container (~10 mg Fe/cm²). All isomer shifts are given relative to α-iron. The data were evaluated with the MOSSWINN program [71] using the thin absorber approximation.

III. RESULTS AND DISCUSSION

A. Crystal structure

The crystal structures of Ca₂FeRuO₆ (I) and CaSrFeRuO₆ (II) were investigated by room-temperature laboratory PXRD and high-resolution synchrotron powder diffraction. From the analysis of the powder-diffraction data, it was found that both compounds crystallize in the orthorhombic space group Pbnm (no. 62). The absence of additional Bragg reflections in the diffraction patterns confirms the phase purity of the samples. The shifting of peaks towards lower angles for CaSrFeRuO₆ is due to the incorporation of the larger Sr²⁺ ion at the position of the smaller Ca^{2+} ion in Ca_2FeRuO_6 . An orthorhombic cell is typically observed when the A-O bond length is less than $\sqrt{2}$ times the (B-O) bond length, which results in rotations of the BO₆ octahedra (inset of Fig. 1). It is known that the counterclockwise rotation of equivalent magnitude about the [0,1,0] and [0,0,1] cubic direction and clockwise rotation about [1,0,0] leads to the *Pbnm* space group. In this space group the Ca atoms in Ca₂FeRuO₆ occupy the Wyckoff $4c(x, y, \frac{1}{4})$ positions, whereas Ca and Sr atoms are statistically disordered on $4c(x, y, \frac{1}{4})$ in CaSrFeRuO₆ as are Fe and Ru on the position $4b(\frac{1}{2},0,0)$ in both compounds. The O1 and O2 atoms in both compounds I and II fully occupy the Wyckoff positions $4c(x,y,\frac{1}{4})$ and 8d(x,y,z), respectively. The results of the Rietveld analysis of the synchrotron powder data of Ca₂FeRuO₆ are shown in Fig. 1. The refinement of the positional and isotropic thermal parameters resulted in a residual $R_F = 0.0340$ (defined as $R_F = \sum ||F_{obs}| - |F_{calc}|| / \sum |F_{obs}|$ from the synchrotron data of compound I. In the synchrotron powder-diffraction



FIG. 1. Results of the Rietveld refinements of the synchrotron powder-diffraction data of Ca₂FeRuO₆ collected at 295 K with $\lambda =$ 0.6973 Å and refined in the orthorhombic space group *Pbnm*. The calculated patterns (red) are compared with the observed ones (black circles). The difference patterns (blue) as well as the positions (green bars) of the nuclear Bragg reflections are also shown. Inset: Crystal structure of the compound (light gray sphere for Ca, red sphere for O, green octahedra for Fe and Ru).

pattern collected at 100 K, well below the magnetic ordering temperature $T_N \approx 220$ K, we could find no evidence of additional peak splitting in the case of Ca₂FeRuO₆. Further, we could not observe additional Bragg reflections, thus excluding any symmetry lowering. Additionally, laboratory x-ray powder-diffraction data were collected at room temperature for compounds I and II (Figs. S1 and S2 in Supplemental Material [72]), where the refinements resulted in the residuals $R_F = 0.0711$ and 0.0214, respectively. The results of the refinements are given in Table S1 in Supplemental Material [72]. The resulting bond lengths from powder x-ray data for Ca_2FeRuO_6 are d(Fe/Ru-O1) = 1.9541(2)Å; two O2 distances are d(Fe/Ru-O2) = 1.9955(10) and 1.9831(10) Å, for $CaSrFeRuO_6 d(Fe/Ru-O1) = 1.9611(2) \text{ Å}; two O2 distances$ are d(Fe/Ru-O2) = 1.9428(2) and 2.0010(1) Å. These bond lengths are close to those of earlier reported Sr₂FeRuO₆ [52]. In the present compounds, Fe^{3+} and Ru^{5+} ions exist in $3d^5$ and $4d^3$ configurations with equal ionic radius. An additional confirmation for the oxidation states of Fe and Ru was obtained by bond valence sum (BVS) calculations which were performed by using the room-temperature diffraction data and the SPUDS software. Applying all the possible valance states for Fe and Ru, the best match was found for Fe as Fe^{3+} and Ru as Ru⁵⁺ oxidation states. Calculated BVS values from powder x-ray data are listed in Table I. The other selected bond lengths and bond angles are also given in Table I.

The crystal structures of compounds I and II were also refined from neutron powder-diffraction data collected on the instrument E9 at 3.2 and 3.0 K, respectively (Fig. 2). Again, the crystal structure could be successfully refined in the space group *Pbnm* resulting in a residual $R_F = 0.0256$ and 0.0366 for I and II, respectively. In the case of Ca₂FeRuO₆ some Bragg reflections contained magnetic intensities. The two TABLE I. Some selected bond lengths, bond angles, and bond valence sums from laboratory PXRD data refinement for (a) Ca_2FeRuO_6 and (b) $CaSrFeRuO_6$, respectively.

(a)					
Bond length	nd length (Å)		Bond angle	(°)	
Ca-O1	2.4176(9)		Fe/Ru-O1-Fe/Ru	157.89 (1)	
			Fe/Ru-O2-Fe/Ru	150.22 (5)	
Ca-O'2	2.6200 (10)		O1-Fe/Ru-O1	180	
Ca-O"2	2.6823 (9)		O1-Fe/Ru-O'2	93.49 (2)	
Fe/Ru-O1	1.9541 (2)		O1-Fe/Ru-O"2	88.37 (12)	
Fe/Ru-O'2	1.9955 (10)		O'2-Fe/Ru-O"2	90.99(4)	
Fe/Ru-O"2	1.9831(10)		Fe/Ru-O1-Ca	91.66 (3)	
O1-O'2	2.7065 (10)		Fe/Ru-O'2-Ca	85.01 (1)	
O1-O''2	2.8234 (10)		Fe/Ru-O"2-Ca	83.58 (2)	
O'2-O''2	2.8377 (9)				
BVS calculations					
Fe-O $(1) \times 2$	1.18		Ru-O (1) \times 2	1.72	
		BVS of $Fe^{3+} = 3.32$			BVS of $Ru^{5+} = 4.86$
Fe-O' (2) \times 2	1.05		Ru-O' (2) \times 2	1.54	
Fe-O'' (2) \times 2	1.09		Ru-O'' (2) \times 2	1.59	
(b)					
Bond length	(Å)		Bond angle	(°)	
Sr/Ca-O1	2.5873 (1)		Fe/Ru-O1-Fe/Ru	164.40(1)	
			Fe/Ru-O'2-Fe/Ru	160.24 (2)	
Sr/Ca-O'2	2.7095 (1)		O1-Fe/Ru-O1	180	
Sr/Ca-O"2	2.7198 (2)		O1-Fe/Ru-O'2	90.55 (1)	
Fe/Ru-O1	1.9611 (2)		O1-Fe/Ru-O"2	88.27 (1)	
Fe/Ru-O'2	1.9428 (2)		O'2-Fe/Ru-O"2	89.58 (2)	
Fe/Ru-O"2	2.0010(1)		Fe/Ru-O1-Ca/Sr	91.15 (1)	
O1-O''2	2.8437 (1)		Fe/Ru-O'2-Ca/Sr	87.98 (1)	
O1-O'2	2.7470(1)		Fe/Ru-O"2- Ca/Sr	86.55 (1)	
O'2-O''2	2.7992 (2)		Fe/Ru-O1-Fe/Ru	164.40(1)	
			Fe/Ru-O'2-Fe/Ru	160.24 (2)	
BVS calculations					
Fe-O $(1) \times 2$	1.16		Ru-O (1) \times 2	1.70	
		BVS of $Fe^{3+} = 3.40$			BVS of $Ru^{5+} = 5.00$
Fe-O' (2) \times 2	1.21		Ru-O' (2) \times 2	1.78	
Fe-O''(2) \times 2	1.03		$\text{Ru-O}''(2) \times 2$	1.52	

strongest reflections are marked in Fig. 2 and the results are summarized in Table II.

B. dc magnetization

The temperature (*T*) dependent dc susceptibility χ of Ca₂FeRuO₆ and CaSrFeRuO₆ measured in an applied field of 0.5 T is shown in Figs. 3 and 4, respectively. At higher temperatures, the inverse susceptibility χ^{-1} increases linearly for both compounds with *T* obeying the Curie-Weiss (CW) law, as expected in the paramagnetic regime. In order to extract the magnetic parameters, $\chi(T)$ in the high-*T* regime was fitted by the CW law:

$$\chi_{\rm CW} = \chi_0 + C/(T - \theta). \tag{1}$$

Here χ_0 is the temperature-independent susceptibility, which includes core diamagnetism and Van-Vleck paramagnetism; *C* is the Curie constant; and θ is the CW temperature. The extracted parameters are tabulated in Table III. The strongly negative value of θ indicates dominant antiferromagnetic (AFM) interactions in both the compounds, while the strength of the AFM interactions in Ca₂FeRuO₆ is much stronger than that in CaSrFeRuO₆. In both compounds, the expected ionic states of Fe and Ru are Fe³⁺ (d^5 ; S = 5/2) and Ru⁵⁺ (d^3 ; S = 3/2), which are having 50% occupancies each in the simple $AFe_{0.5}Ru_{0.5}O_3$ perovskite formula unit. Therefore, the expected effective moment assuming spinonly contributions for both compounds can be calculated as $\mu_{\text{eff}} = \sqrt{0.5 \left[\mu_{\text{eff}}(\text{Fe}^{3+})\right]^2 + 0.5 \left[\mu_{\text{eff}}(\text{Ru}^{5+})\right]^2} = 5.0 \,\mu_{\text{B}}$ [using $\mu_{\text{eff}}(\text{Fe}^{3+}) = 5.92 \,\mu_{\text{B}}$ and $\mu_{\text{eff}}(\text{Ru}^{5+}) = 3.87 \,\mu_{\text{B}}$]. Thus, our obtained value of μ_{eff} for Ca₂FeRuO₆ (see Table III) is only slightly smaller than the expected value, which reflects a somewhat reduced Ru⁵⁺ moment due to spin-orbit coupling and covalency effects. In contrast, the obtained value for CaSrFeRuO₆ is more reduced, which may reflect an insufficient temperature range for the Curie-Weiss analysis (here only measurements up to 400 K were performed).

At lower temperatures, Ca_2FeRuO_6 shows spontaneous changes in χ near 220 K and below about 90 K, indicating that

TABLE II. Results of the crystal structure refinements of Ca_2FeRuO_6 and $CaSrFeRuO_6$ as obtained from synchrotron and neutron powder diffraction. The crystal structure of both compounds was refined in the orthorhombic space group *Pbnm*. The Ca/Sr and O1 atoms occupy Wyckoff 4c(x,y,y) positions, while the O2 atoms are located at the site 8d(x, y, z), respectively. The Fe and Ru atoms are statistically distributed at the site $4b(y_2,0,0)$. The thermal parameters of the oxygen atoms O1 and O2 determined from the neutron data were constrained to be equal during the refinements; for the synchrotron data an overall thermal parameter was used for all atoms.

	Ca ₂ FeRuO ₆	Ca ₂ FeRuO ₆	CaSrFeRuO ₆ Neutron
	Synchrotron	Neutron	
Т (К)	295	3.2	3.0
Space group	Pbnm	Pbnm	Pbnm
<i>a</i> (Å)	5.40277 (13)	5.3849(4)	5.4790(3)
<i>b</i> (Å)	5.47029 (14)	5.4703(5)	5.4957(4)
<i>c</i> (Å)	7.67262 (19)	7.6520(6)	7.7515(5)
$V(\text{\AA}^3)$	226.76 (4)	225.40(3)	233.41(3)
x(Ca)	0.9990 (5)	0.9876(13)	0.9981(15)
y(Ca)	0.0417 (2)	0.0424(5)	0.0297(5)
<i>x</i> (O1)	0.0829 (11)	0.0845(7)	0.0669(10)
y(O1)	0.4795 (8)	0.4768(6)	0.4881(9)
<i>x</i> (O2)	0.7070 (7)	0.7080(4)	0.7171(6)
y(O2)	0.2894 (6)	0.2953(4)	0.2841(6)
z(O2)	0.0395 (6)	0.0432(3)	0.0285(4)
$B(\text{Ca/Sr}) (\text{\AA}^2)$	0.50 (2)	0.51(5)	0.86(10)
$B(\text{Fe/Ru}) (\text{\AA}^2)$	0.50(2)	0.26(2)	0.50(2)
B(O) (Å ²)	0.50 (2)	0.45(2)	1.03(2)
R_F	0.0340	0.0256	0.0366
d_{eq} (Fe/Ru-O2) (Å)	1.9811(36)	1.9566(24)	1.9631(34)
d_{eq} (Fe/Ru-O2) (Å)	1.9619(35)	1.9913(23)	1.9741(33)
d_{ap} (Fe/Ru-O1) (Å)	1.9729(14)	1.9684(9)	1.9719(11)

the compound undergoes two successive magnetic transitions. In order to confirm these transitions, zero-field cooled (ZFC)



FIG. 2. Results of the Rietveld refinements of the neutron powder-diffraction data of Ca₂FeRuO₆ and CaSrFeRuO₆ collected at 3.2 and 3.0 K, respectively, on instrument E9 with $\lambda = 1.3087$ Å. The crystal structure of both compounds was refined in the orthorhombic space group *Pbnm*. The calculated patterns (red) are compared with the observed ones (black circles). The difference patterns (blue) as well as the positions (black bars) of the nuclear Bragg reflections are also shown. In the case of Ca₂FeRuO₆ some Bragg reflections are marked (further details are shown in Fig. 6).

and field cooled (FC) susceptibilities were measured at a very low field of 0.05 T. As shown in the inset of Fig. 3, a clear splitting was observed at 220 K and the ZFC data exhibit two peaks at ~220 and ~83 K. This further confirms two magnetic transitions. On the other hand, the $\chi(T)$ plot of CaSrFeRuO₆ only shows a single peak at ~61 K and the ZFC and FC susceptibilities show a weak splitting at this



FIG. 3. Magnetic susceptibility $\chi(T)$ of Ca₂FeRuO₆ measured at H = 0.5 T. $1/\chi$ vs T is plotted in the right y axis. The solid line represents the fit using CW law. Inset: ZFC and FC susceptibilities vs T measured at H = 0.05 T.



FIG. 4. Magnetic susceptibility $\chi(T)$ of CaSrFeRuO₆ measured at H = 0.5 T. $1/\chi$ vs T is plotted in the right y axis. The solid line represents the fit using CW law. Inset: ZFC and FC susceptibilities vs T measured at H = 0.05 T.

temperature, which is a possible indication of the occurrence of a SG transition.

Further information on the nature of the magnetic state is obtained from isothermal magnetization data M(H) which were measured at three different temperatures (5, 120, and 320 K) and are shown in Fig. 5. At 320 K, it shows a perfect linear behavior without any hysteresis, as expected in the paramagnetic regime for both the compounds. For Ca_2FeRuO_6 , a hysteresis was observed at T = 120 K which is more pronounced at 5 K. On the other hand, for CaSrFeRuO₆, no hysteresis was observed at 120 K but the data at 5 K show a small hysteresis. The remanent magnetization values of 0.027 and 0.004 $\mu_{\rm B}$ for Ca₂FeRuO₆ and CaSrFeRuO₆, respectively, are small and there is no indication of magnetic saturation up to 9 T. These features rule out any ferromagnetic ordering. There is the possibility of a ferrimagnetic ordering since Fe³⁺ and Ru⁵⁺ carry different moments and an AFM interaction between them can give rise to ferrimagnetism. Such type of peculiar magnetic behavior was reported earlier for osmium-based double perovskites [21,37]. However, our neutron-diffraction experiments (discussed later) could not evidence ferrimagnetic ordering, which is also not expected for the atomically disordered compounds. Thus, the most reasonable explanation of the rise in χ and the hysteresis in the M(H) curve is that Ca₂FeRuO₆ undergoes an AFM transition at 220 K with slight spin canting giving rise to a ferromagnetic component followed by a SG-like transition near 83 K

TABLE III. Parameters obtained by fitting the high-temperature part of the $\chi^{-1}(T)$ data (380 to 800 K for Ca₂FeRuO₆ and 290 to 380 K for CaSrFeRuO₆) using the Curie-Weiss law. In our calculation one mol represents the formula units CaFe_{0.5}Ru_{0.5}O₃ and Ca_{0.5}Sr_{0.5}Fe_{0.5}Ru_{0.5}O₃, respectively.

Compounds	$\chi_0 (\mathrm{cm}^3/\mathrm{mol})$	θ (K)	$C (\mathrm{cm}^3 \mathrm{K/mol})$	$\mu_{\rm eff}~(\mu_{\rm B})$
Ca ₂ FeRuO ₆	0.0014	-370.3	2.87	4.79
CaSrFeRuO ₆	0.0020	-166.2	1.75	3.74



FIG. 5. Isothermal magnetization (M vs H) measured at three different temperatures for Ca₂FeRuO₆ (upper panel) and CaSrFeRuO₆ (lower panel).

whereas the transition at 61 K observed for $CaSrFeRuO_6$ is likely to be SG type.

C. Neutron diffraction

In order to investigate the magnetic order of Ca₂FeRuO₆, we have collected neutron powder patterns on the instrument E6 from 1.6 up to 230 K. In comparison to the data collected at room temperature the strongest magnetic intensity could be observed at the position of the reflection pair 101/011 (Fig. 6). For the Fe and Ru atoms in the position $4b(\frac{1}{2},0,0)$; 1/2,0,1/2; 0,1/2,0; 0,1/2,1/2) magnetic intensity could be generated with a G-type model, where the spin sequence is + - - +. In order to determine the moment direction, we carried out Rietveld refinements for different models of spin alignments, where the moments were aligned either parallel to the axis a, or parallel to b or c. Depending on the moment direction one obtains different intensity ratios for the reflections 101 and 011. For μ_x one obtains a ratio of about 1:3, for μ_y one obtains a ratio of 3:1, and for μ_z one obtains a ratio of 1:1. Due to the orthorhombic symmetry the reflections 101 and 011 show a weak splitting of 0.34° , which finally allowed us to find out the correct moment direction. The best fit is obtained when the moment is aligned parallel to the c axis (G_z mode), where the intensity ratio of the reflections 101 and 011 is 1:1.

Possible magnetic structures can be theoretically deduced from Bertaut's representation analysis [73]. Here the four possible spin sequences for the metal atoms in the position 4b of the space group Pbnm are labeled F(+ + + +), G(+ --+), C(+ + --), and A(+ - + -). The analysis shows for the propagation vector $\mathbf{k} = 0$ that a G_z mode can only be admixed with F_x and C_y [representation $\Gamma_2(F_x, C_y, G_z)$]



FIG. 6. Neutron powder patterns of Ca_2FeRuO_6 and $CaSrFeRuO_6$ taken on instrument E6 with $\lambda = 2.43$ Å. At 1.6 K the strongest magnetic intensity of Ca_2FeRuO_6 occurs at the positions of the reflection pair 101 and 011 indicating the presence of a *G*-type spin ordering as shown in the inset of the upper diagram. The calculated pattern of the pure nuclear contribution (blue) as well as the sum of the nuclear and the magnetic contribution (red) is compared with the observed one (black circles). For CaSrFeRuO₆ only a broad diffuse magnetic signal appears below about 150 K close to the position of the reflection pair 101 and 011 indicating a short-range order of the magnetic moments.

or Shubnikov group Pbn'm' given in Table 5 of [Ref. [73]]. Since no magnetic intensity at the position of the reflections at 100, 010, and 001 (forbidden in *Pbnm*) can be detected, the existence of an *A*- or *C*-type ordering along the *x* and *y* directions is excluded. Interestingly, the magnetization measurements revealed a weak ferromagnetic component. In accordance with the representation analysis a ferromagnetic component should be found along the *x* direction. Here it has to be mentioned that a weak magnetic component is hardly observable from neutron data, since the weak magnetic signal is superimposed on stronger nuclear reflections. However, a pure G_z -type ordering could be found, for example, for the vanadium moments of YVO₃ [74].

The iron and ruthenium atoms in Ca₂FeRuO₆ are both located at the Wyckoff position 4*b*, hence there is the difficulty to determine their magnetic moments separately. The refinements of the magnetic structure of the Fe/Ru sublattice from the data set collected on E6 resulted in a satisfactory residual $R_{\rm M} = 0.0359$ (defined as $R_{\rm M} = \sum ||I_{\rm obs}| - |I_{\rm calc}|| / \sum |I_{\rm obs}|$), where the averaged moment value at 1.6 K is found to



FIG. 7. Temperature dependence of the lattice parameters of Ca_2FeRuO_6 and the averaged magnetic moment of the transitionmetal atoms Fe and Ru at the Wyckoff position 4*b*, which is statistically occupied with Fe and Ru atoms.

be $\mu_{exp}(Fe/Ru) = 2.63(2) \mu_B$. For comparison, we have obtained from the data set collected on the instrument E9 the moment $\mu_{exp}(Fe/Ru) = 2.72(3) \mu_B$ resulting in a residual $R_M = 0.0479$. These values are in good agreement with an average $\mu_{exp} = 2.75 \mu_B$ which is calculated if we assume a typical value of $\mu_{exp} = 3.5 \mu_B$ as was obtained for Fe^{3+} ions from neutron data of ternary oxides [35,75,76] and $\mu_{exp} = 2.0 \mu_B$ for Ru^{5+} ions [77]. This suggests that both Fe^{3+} and Ru^{5+} ions constitute the AFM structure. In contrast to Ca₂FeRuO₆ we could not find a long-range order in CaSrFeRuO₆ due to the absence of magnetic Bragg reflections. For CaSrFeRuO₆ only a broad diffuse magnetic signal appears below about 150 K (Fig. 6) close to the position of the reflection pair 101/011, where Ca₂FeRuO₆ showed the strongest magnetic intensity. The appearance of a diffuse peak indicates a short-range order of the magnetic moments.

In the next step, we have investigated the temperature dependence of the lattice parameters for both title compounds and the magnetic moments for Ca_2FeRuO_6 , where the moments of the Fe and Ru atoms are aligned parallel to the *c* axis. Figure 7 shows that the magnetic moments disappear at about 220 K. Accordingly, the steep rise in the magnetic susceptibility below 220 K seems to correspond to the onset of magnetic ordering. In Fig. 7, we also show the temperature dependence of the lattice parameters. Here one finds no anomaly in the magnetically ordered range. But it can be seen that the lattice



FIG. 8. ⁵⁷Fe Mössbauer spectra of (a) Ca_2FeRuO_6 and (b) $CaSrFeRuO_6$ measured at the indicated temperatures. Dots correspond to the experimental data. Black lines correspond to the calculated spectra. Blue and red lines correspond to the subspectra of the magnetically ordered and paramagnetic component, respectively. (c) Temperature dependence of the average hyperfine field for the two compounds. The vertical red and blue dashed lines correspond to the spin freezing temperatures inferred from the ac susceptibility data of Ca_2FeRuO_6 and $CaSrFeRuO_6$, respectively. Filled symbols correspond to data evaluation using the Hesse-Rübartsch method. Open symbols correspond to data evaluation with a single hyperfine component having a Gaussian B_{hf} distribution. Dash-dotted lines are guides to the eye.

parameters a and c show a slight decrease during the cooling process down to 1.6 K, while the parameter b seems to remain unchanged for Ca₂FeRuO₆.

D. Mössbauer study

While the neutron-diffraction data reflect the spatially averaged magnetic structure, Mössbauer spectroscopy gives insights into local variations of the magnetic behavior. A set of Mössbauer spectra of Ca₂FeRuO₆ is shown in Fig. 8(a). For temperatures above 220 K the spectra consist of a broadened quadrupole doublet with an isomer shift (IS) of 0.35 mm/s at 291 K, which is typical for six-coordinated Fe³⁺ in an oxide environment. The line broadening reflects the atomic disorder which leads to a variation in the nearest- and next-nearestneighbor environment. In order to account for the broadening, the spectra were evaluated assuming a Gaussian distribution of quadrupole splittings (QS), which leads to average QS of 0.67 mm/s. Both IS and QS are similar as for Sr_2FeRuO_6 where Fe and Ru are atomically disordered too [52].

Below 220 K magnetic ordering effects are apparent in the spectra. Only at the lowest temperature of 5.5 K all the iron atoms are magnetically ordered (IS = 0.49 mm/s), however due to the atomic disorder the hyperfine pattern is considerably broadened. The spectrum could be best fitted by assuming a Gaussian distribution of hyperfine fields $B_{\rm hf}$ and in addition allowing for a distribution in the quadrupole splitting parameter. The large average $B_{\rm hf}$ of 48 T is typical for Fe³⁺/4*d* or Fe³⁺/5*d* mixed perovskites (cf. $B_{\rm hf}$ = 49 T for Sr₂FeRuO₆ at 4.2 K [52] and Sr₂FeOsO₆ at 5 K [35]). With increasing temperature, a paramagnetic contribution occurs in the spectra, the area fraction of which increases continuously with temperature. Even at 50 K a small paramagnetic fraction is discernible. Furthermore, the spectra of the magnetic component broaden tremendously with increasing temperature and a fit of the spectra with a single Gaussian $B_{\rm hf}$ distribution becomes successively insufficient. In order to reproduce the spectra over the whole temperature range we used the model-independent Hesse-Rübartsch method for extracting the hyperfine field distribution, whereas the paramagnetic component was described by a doublet with a distribution of quadrupole splittings. The average QS and the width of the QS distribution were kept fixed to the values at 240 K, which is above the magnetic ordering temperature. Within this fitting model, any distribution in quadrupole splitting parameters of the magnetic phase is neglected. The isomer shifts of the paramagnetic and magnetically ordered component were constrained to be the same. A magnetic component is seen in the spectra up to 210 K, whereas a spectrum at 230 K consists of a pure quadrupole doublet. Thus, the onset temperature $T_{\rm m}$ of magnetic order should be between 210 and 230 K, which is in good agreement with the neutrondiffraction data, where $T_{\rm m} = 220 \,\rm K$ was obtained and with the rise in the susceptibility below 220 K. With increasing temperature, the average hyperfine field decreases gradually from its maximum value [Fig. 8(c)]; a similar behavior is also apparent in the temperature dependence of the magnetic moments (Fig. 7). The broad distributions in $B_{\rm hf}$ [Fig. S3 (top) in Supplemental Material [72]] as well as the coexistence of magnetically ordered and paramagnetic signals over a large temperature range are evidence that the magnetic state is very inhomogeneous. Actually, there is no well-defined magnetic ordering temperature but rather a broad distribution of ordering or freezing temperatures. This can be explained by the variations in the local environment of the Fe^{3+} ions which have a varying number of Fe^{3+} and Ru^{5+} neighbors and thus differing exchange interactions. There are no obvious anomalies in the spectra and parameters near 83 K, where the magnetic susceptibility measurements indicated a further phase transition.

It is instructive to compare the properties of Ca_2FeRuO_6 with those of the Sr-analog Sr₂FeRuO₆ [52]. The latter was reported to be a SG system without any evidence of longrange magnetic order. While the low-temperature Mössbauer spectra of Sr₂FeRuO₆ and Ca₂FeRuO₆ show similar shapes, a complete collapse of the hyperfine pattern was found for Sr₂FeRuO₆ near 60 K. By contrast, a magnetic component persists up to 210 K in the spectra of Ca₂FeRuO₆ and the neutron-diffraction patterns of Ca₂FeRuO₆ show magnetic Bragg scattering which is supportive of long-range magnetic order. A scenario explaining the broad Mössbauer spectra of Ca₂FeRuO₆ as well as the quite sluggish magnetic phase transition is to assume the coexistence of long-range magnetic ordered regions with spin or cluster-glass-like regions having smaller magnetic blocking temperatures.

In order to study the evolution of magnetic properties from Sr_2FeRuO_6 to Ca_2FeRuO_6 we measured Mössbauer spectra of the intermediate composition CaSrFeRuO₆ [see Fig. 8(b)]. Spectra featuring magnetic hyperfine patterns were evaluated assuming a distribution of magnetic hyperfine fields [Fig. S3 (bottom) in Supplemental Material [72]], while those in the paramagnetic phase were evaluated using a distribution of quadrupole splittings. The distributions were extracted according to the Hesse-Rübartsch method; additionally for the two lowest temperatures an alternate fit with a single Gaussian distribution in $B_{\rm hf}$ was performed. The temperature dependence of the average $B_{\rm hf}$ is included in Fig. 8(c). The Mössbauer parameters are very similar to those of Ca₂FeRuO₆: IS = 0.37 mm/s and $\langle QS \rangle = 0.65 \text{ mm/s}$ at 290 K and IS =0.49 mm/s and $\langle B_{\rm hf} \rangle = 48$ T at 5.3 K. It is apparent that the temperature variation of the spectra compares much more with that of Sr_2FeRuO_6 [52] than with that of Ca_2FeRuO_6 . In particular, there is an inward collapse of the magnetic hyperfine pattern near 60 K, whereas no magnetic components are discernible in spectra for $T \ge 70$ K. These observations are fully consistent with the cusp near 61 K in the $\chi(T)$ data and with the absence of magnetic Bragg reflections in the powder neutron patterns of CaSrFeRuO₆, which supports the clue that CaSrFeRuO₆, similar to Sr₂FeRuO₆, adopts a SG state. Long-range magnetic ordering only occurs for larger Ca contents.

E. ac susceptibility

In order to confirm the SG behavior of the compounds under investigation, ac susceptibility measurements were carried out at different frequencies. The real part of the ac susceptibility χ' as a function of temperature is shown in Fig. 9. As shown in the inset of the upper panel of Fig. 9, $\chi'(T)$ of Ca₂FeRuO₆ shows a broad peak at around 200 K which is frequency independent, consistent with a canted antiferromagnetic transition observed in the dc magnetization measurements. Another sharp peak was observed at $\sim 87 \,\mathrm{K}$ which is found to be shifted towards higher temperatures and the magnitude of the peak decreases with increasing frequency. These are characteristic features of a SG transition. On the other hand, for CaSrFeRuO₆, only one transition was observed at \sim 65 K. With increasing frequency, the peak position was found to be shifted towards high temperatures and also the magnitude of the peak decreases, suggesting that the transition is SG type. Thus, Ca₂FeRuO₆ and CaSrFeRuO₆ undergo SG-type transitions with freezing temperatures $T_f \approx$ 87 and 65 K, respectively. To study the spin dynamics, the frequency dependence of the freezing temperature obtained from $\chi'(T)$ was fitted by the conventional power-law divergence of the critical slowing down model [78,79]:

$$\tau = \tau_0 [(T_f - T_g)/T_g]^{-z\nu'}.$$
 (2)

Here τ is the relaxation time corresponding to the measured frequency ($\tau = 1/\nu$), τ_0 is the characteristic relaxation time of a single spin flip, T_g is the SG transition temperature as ν tends to zero, and zv' is the dynamic critical exponent. The values of T_g were determined to be ~86.6 and ~62 K for Ca₂FeRuO₆ and CaSrFeRuO₆, respectively, by extrapolating the T_f versus ν plot to $\nu = 0$. The obtained fitting parameters are ($\tau_0 \approx$ 3.81×10^{-22} s and $z\nu' \approx 6.65$) and $(\tau_0 \approx 7.57 \times 10^{-12}$ s and $z\nu' \approx 5.4$) for Ca₂FeRuO₆ and CaSrFeRuO₆, respectively. For SG systems, the value of $z\nu'$ typically lies between ~ 4 and 12 while the value of τ_0 ranges from 10^{-10} to 10^{-13} s. Similarly, for the canonical SG and cluster SG, the characteristic range of τ_0 varies from $\sim 10^{-12}$ to $\sim 10^{-13}$ s and $\sim 10^{-7}$ to $\sim 10^{-10}$ s, respectively [80,81]. Our experimental values of zv' are consistent with the SG behavior in both the compounds. Similarly, the value of τ_0 for CaSrFeRuO₆



FIG. 9. Upper panel: Real part $\chi'(T)$ of the ac susceptibility of Ca₂FeRuO₆ measured in an ac field of 8 Oe and at different frequencies. Lower inset: $\chi'(T)$ measured at 100 Hz and for the whole temperature range. Upper inset: Frequency dependence of freezing temperature plotted as $\ln(\nu)$ vs $1/(T_f - T_0)$ together with the fit using the Vogel-Fulcher law. Lower panel: $\chi'(T)$ of CaSrFeRuO₆ measured in an ac field of 8 Oe and at different frequencies. Inset: $\ln(\nu)$ vs $1/(T_f - T_0)$ together with the fit using Vogel-Fulcher law.

falls in the range corresponding to the canonical SG. On the other hand, for Ca₂FeRuO₆ the value of τ_0 is found to be unphysically small.

In order to estimate the activation energy, we tried to fit the frequency dependence of T_f by the simple Arrhenius law, which does not give a good fit, suggesting that the dynamics is not simply associated with single spin flips but rather reflects a cooperative character of the freezing-in process in both the compounds. However, the frequency dependence of T_f could be fitted well using the Vogel-Fulcher law [79]:

$$\nu = \nu_0 \exp[-E_a/k_B(T_f - T_0)],$$
(3)

where ν_0 is the characteristic attempt frequency, E_a is the activation energy, and T_0 is the Vogel-Fulcher temperature, which is often interpreted as a measure of the effective interaction between spins or clusters. The fits are shown in the insets of Fig. 9. The obtained best fit parameters are ($\nu_0 \approx 4.06 \times 10^{11} \text{ s}^{-1}$, $T_0 \approx 86.3 \text{ K}$, and $E_a/k_B \approx 8.6 \text{ K}$) for Ca₂FeRuO₆ and ($\nu_0 \approx 5.4 \times 10^{11} \text{ s}^{-1}$, $T_0 \approx 58 \text{ K}$, and $E_a/k_B \approx 119 \text{ K}$) for Ca₂FeRuO₆. The estimated values of ν_0 in both the



FIG. 10. The electrical resistivity $\rho(T)$ of Ca₂FeRuO₆ and CaSrFeRuO₆ measured at zero field. Inset: $\ln(\sigma)$ vs 1/T. The solid lines are the fits to the data using Eq. (4).

compounds are in the intermediate range expected for canonical and cluster SG systems [82]. Moreover, the development of a glassy phase below a magnetically ordered phase is commonly referred to as "reentrant SG [61]." In Ca_2FeRuO_6 , the canted antiferromagnetic transition at high temperatures is followed by a low-temperature SG transition, thus reflecting reentrant-SG behavior.

F. Resistivity

The temperature-dependent electrical resistivity $\rho(T)$ of Ca₂FeRuO₆ and CaSrFeRuO₆ was measured at zero field as shown in Fig. 10. For both the compounds, $\rho(T)$ increases rapidly with decreasing *T*, suggesting insulating ground states. At low temperatures the resistances could not be measured since the resistance values exceeded the instrument's measurement limit. The conductivity data ($\sigma = \frac{1}{\rho}$) in the high-temperature regime were fitted using a law for activated transport:

$$\sigma(T) = A \exp\left(\frac{-\Delta}{k_{\rm B}T}\right),\tag{4}$$

where Δ is the activation energy, $k_{\rm B}$ is the Boltzmann constant, and A is the proportionality constant. The inset of Fig. 10 shows the plot of $\ln(\sigma)$ versus 1/T to highlight the approximately linear behavior in the high-temperature regime. Our fit in the temperature range $217 \leq T \leq 300$ K for Ca₂FeRuO₆ and $188 \leq T \leq 300$ K for Ca₂FeRuO₆ and $288 \leq T \leq 300$ K for Ca₂FeRuO₆ and $288 \leq T \leq 300$ K for Ca₂FeRuO₆ and $288 \leq T \leq 300$ K for Ca₂FeRuO₆ and $288 \leq T \leq 300$ K for Ca₂FeRuO₆ yields $\Delta \approx 0.25$ eV and ≈ 0.12 eV, respectively. Thus, Sr²⁺ substitution at the Ca²⁺ site leads to a reduction of band gap and increase in conductivity.

It is interesting to note that the $\ln(\sigma)$ versus 1/T plots show an approximately linear behavior only in the hightemperature regime. However, a linearity over an extended temperature range, in particular for CaSrFeRuO₆, was obtained when $\ln(\sigma)$ was plotted against $(1/T)^{1/4}$ (Fig. S4 in Supplemental Material [72]) which possibly suggests that a three-dimensional variable range hopping (VRH) transport mechanism [83,84] proposed by Mott and Davis is operative in these two systems. The VRH mechanism normally occurs in the low-temperature region (below room temperature) where the energy is insufficient to excite the charge carriers across the Coulomb gap. This type of behavior was reported in various double perovskites and related materials [20,85–87].

G. Magnetism in Ca_{2-x}Sr_xFeRuO₆: Summarizing discussion

Atomically ordered as well as disordered $A_2BB'O_6$ perovskites are a versatile playground for magnetism and a large variety of magnetic properties has been realized in this class of compounds [20]. This is a consequence of the chemical flexibility of perovskites and of the various exchange pathways in $A_2BB'O_6$ systems with magnetic B and B' ions which often lead to competing exchange interactions and signatures of magnetic frustration. The balance of interactions and in some cases even the type of magnetic order depends on the size of the A cation. A large degree of magnetic frustration particularly may occur in atomically disordered compounds where the atomic disorder together with competing exchange pathways in several cases leads to spin-glass behavior. Examples related to the present $Ca_{2-x}Sr_xFeRuO_6$ system are the compounds $Sr_2FeB'O_6$ with B' = Nb and Ta [20]. Also the previously reported Sr₂FeRuO₆ [52] is such a spin-glass system. Here, we replaced the larger Sr^{2+} by smaller Ca^{2+} ions, which decreases the tolerance factor and in turn enhances the degree of tilting of the octahedra in the perovskite structure. This leads to a drastic change in the magnetic properties: In contrast to Sr₂FeRuO₆ we observed long-range antiferromagnetic order below $T_N = 220 \text{ K}$ with the peculiarity of reentrant spin-glass behavior at lower temperatures (87 K) for Ca_2FeRuO_6 (I). Slight enhancement of the tolerance factor from 0.969 (I) to 0.987 in CaSrFeRuO₆ (II) is sufficient to suppress the long-range magnetic order and similar to Sr₂FeRuO₆ spin-glass behavior was observed. Both compounds have insulating ground states, but the activation energy for the charge transport in compound II is smaller than in compound I, which is in line with the smaller structural distortion in compound II.

The magnetic features of the present system are reminiscent of comparable phenomena in other systems featuring atomic and magnetic disorder, e.g., metallic ferromagnetic alloy systems like AuFe [61] or the ferrimagnetic spinel system $Mg_{1+x}Fe_{2-2x}Ti_xO_4$ [62]. In these cases, the concentration c of magnetic ions was changed systematically and it was shown that a reentrant magnetic state occurs near the percolation threshold c_n , where the system changes from long-range magnetic order to spin-glass behavior. By contrast, in the present system $Ca_{2-x}Sr_xFeRuO_6$ the concentration of magnetic ions remains unchanged. However, due to the change in tolerance factor the strength of exchange interactions and thus the temperature at which cooperative magnetic ordering occurs is changed because the exchange interactions strongly vary with bond angles. For instance, in compound I, the Fe/Ru-O1-Fe/Ru bond angle along the z axis is 157.9° , which is lower (164.4°) than the observed Fe/Ru-O1-Fe/Ru bond angle of compound II. The Fe/Ru-O2-Fe/Ru bond angle along the xy plane is also lower in compound I (150.2°) than in compound II (160.2°). In the case of the related atomically ordered Os^{5+} double perovskite system $Ca_{2-x}Sr_xFeOsO_6$, the different magnetic structures of Ca₂FeOsO₆ and Sr₂FeOsO₆

(ferrimagnetic and antiferromagnetic, respectively) were attributed to the strengthening of the ferromagnetic σ -exchange pathway between Fe^{3+} and Os^{5+} ions with respect to the antiferromagnetic π -exchange pathway as the Sr content increases [88,89]. This reflects the pronounced magnetostructural correlations where the ferromagnetic pathway is favored for a smaller deviation of the Fe-O-Os bond angle from 180°. Since Ru^{5+} and Os^{5+} ions both have d^3 electron configurations similar arguments apply for the present system. It is noted that according to the Goodenough-Kanamori rule ferromagnetic coupling is expected in the case of d^3 - d^5 high-spin interactions, but due to the higher energy of the unoccupied e_g levels of $4d^3$ and $5d^3$ systems the ferromagnetic exchange path is weakened compared to $3d^3$ systems. Furthermore, due to the atomic disorder in compounds I and II the balance of nearest-neighbor and next-nearest-neighbor exchange interactions is even more complicated. For instance, not only Fe³⁺-Ru⁵⁺ but also antiferromagnetic Fe³⁺-Fe³⁺ and Ru⁵⁺-Ru⁵⁺ nearest-neighbor interactions have to be taken into account. In such a situation of competing interactions it is easily conceivable that subtle structural modifications which change the balance of interactions lead to frustration and spinglass behavior. The sensitive dependence of the coherence of the spin system on the detailed balance of exchange interactions has been demonstrated for Sr₂FeRuO₆ by Monte Carlo simulations [90]. In any case, from the Curie-Weiss temperatures (Table III) and the observation of long-range magnetic order it can be concluded that the average AFM interactions are dominating in the more distorted Ca₂FeRuO₆ and their strength decreases with increasing Sr content. This corroborates the trend already seen in lightly Ca-substituted samples of Sr_2FeRuO_6 [59]. The occurrence of long-range ordering with a quite large T_N of 220 K is reflected in a small frustration factor $f = |\theta|/T_N = 1.7$, whereas strong frustration effects are typically indicated by f > 10 [20]. Nevertheless, in the calcium rich regime the $Ca_{2-x}Sr_xFeRuO_6$ system appears to be at the borderline between long-range antiferromagnetic order and spin-glass behavior. In this situation, reentrant magnetism with long-range ordering below T_N and additional spin freezing below $T_f < T_N$ is observed, while for increased Sr contents atomic disorder and competing exchange interactions drive the system towards a pure spin glass. A comparable reentrant state as in Ca₂FeRuO₆ was reported for the disordered perovskites PbFe_{0.5}Nb_{0.5}O₃ [63] and PbFe_{0.5}Ta_{0.5}O₃ [64] (Pb₂FeNbO₆ and Pb₂FeTaO₆ in our notation).

For explaining the reentrant magnetism two scenarios may be considered. The first one corresponds to microscopic phase separation where large ("infinite") clusters revealing longrange order coexist with smaller clusters remaining paramagnetic above a freezing temperature $T_f < T_N$ [65,91]. In this case, the 87-K transition would correspond to the freezing of such smaller clusters. In fact, the Mössbauer spectra of Ca₂FeRuO₆ verify the coexistence of paramagnetic and magnetically ordered regions over a large temperature range and thus suggest the presence of clusters with smaller ordering temperatures, in contrast to the disordered perovskite systems in Refs. [63,64], where no such clusters were observed. The origin of this difference may be a more complicated balance of exchange interactions as the Ru⁵⁺ ions are magnetic whereas the Nb⁵⁺ and Ta⁵⁺ ions are nonmagnetic. Clustering may be a signature of partial cation ordering as, for instance, in Sr_2FeSbO_6 [20] but there is no experimental evidence for this in the present paper. Obviously, there is no correlation between the appearance of the paramagnetic signal in the Mössbauer spectra and the T_f obtained from the dc and ac susceptibility measurements. The paramagnetic signal increases gradually with temperature and the broad hyperfine patterns rather suggest a distribution of freezing temperatures and thus of cluster sizes which is difficult to reconcile with the sharp freezing temperature of 87 K. Another scenario which was favored to explain the reentrant magnetic properties in the above-mentioned systems [61–63,92] is that below T_N longrange magnetic order occurs along the z direction, whereas the transverse moments remain magnetically disordered and freeze below $T_f < T_N$. In such a model the anomaly in the ac and dc susceptibilities of Ca2FeRuO6 would correspond to the freezing of the transverse magnetic moments. This would naturally explain the well-defined transition temperature and the coexistence of the long-range magnetic order with spinglass-like features. In this scenario the spin freezing is an intrinsic property of the material giving rise to spin canting, and not a consequence of phase separation. In several magnetically diluted reentrant systems the additional spin freezing is reflected in an anomaly in the temperature dependence of the Mössbauer hyperfine field [61–63,92], but such an anomaly is not apparent in the $B_{\rm hf}$ data of Ca₂FeRuO₆ [Fig. 8(c)]. The present system may involve both cluster formation and transverse spin freezing but this issue cannot be resolved unambiguously from the present data.

IV. CONCLUSION

In summary, two 3d-4d hybrid double perovskites Ca_2FeRuO_6 and $CaSrFeRuO_6$ have been synthesized at ambient pressure and characterized using a combination of different techniques in order to study the nature of glassy

features in their magnetic properties. Both compounds crystallize in the orthorhombic centrosymmetric space group *Pbnm*, where the *B* site is randomly occupied by Fe^{3+} and Ru^{5+} ions in the 4b position. Although Ca_2FeRuO_6 shows G-type antiferromagnetic ordering below 220 K, competition between the various exchange interactions due to atomic disorder causes reentrant magnetic behavior with glassy spin freezing at 87 K as evidenced by the frequency dependence of ac susceptibility. Substitution of Ca by Sr in A position leads to the formation of a conventional spin-glass system by destroying long-range ordering. By varying the composition at the A sites, the balance of competing superexchange pathways between the transition metals in B and B' position is modified which induces a change from long-range ordering to spin-glass behavior. In this crossover region reentrant magnetism is observed, similar as in diluted magnetic systems where crossover between spin-glass behavior and long-range order occurs near the percolation threshold. Here, rather a critical tolerance factor t_c separating the magnetic regimes may be defined. The system $Ca_{2-r}Sr_rFeRuO_6$ is an interesting model case for studying the transition region between a spin glass and magnetic order in undiluted mixed perovskites with competing exchange interactions. Further investigations in the Ca-rich region are required to establish more details of the magnetic phase diagram.

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