# Ab initio phase diagram of PbSe crystals calculated with the random phase approximation

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Understanding the phase behavior of semiconductor materials is important for applications in solid state physics and nanoscience. Accurate experimental data is often difficult to obtain due to strong kinetic effects. In this work, we calculate the temperature-pressure phase diagram of lead selenide (PbSe) using the random phase approximation (RPA), an accurate wave-function-based many-body technique. We consider three crystalline phases, the low pressure B1 phase (NaCl type), the intermediate B33 phase (CrB type), and the high pressure B2 phase (CsCl type). The electronic contributions to the free energy (at  $T=0\,\mathrm{K}$ ) are calculated in the Born-Oppenheimer approximation using the RPA, whereas phononic contributions are computed in the quasiharmonic approximation using density functional theory and the PBEsol functional. At room temperature, we find transition pressures of  $4.6\pm0.3\,\mathrm{GPa}$  for the B1  $\leftrightarrow$  B33 transition and  $18.7\pm0.3\,\mathrm{GPa}$  for the B33  $\leftrightarrow$  B2 transition, in good agreement with experiments. In contrast to the interpretation of recent experiments, we observe a negative Clapeyron slope for both transitions. Gibbs free energy differences between competing structures have small gradients close to coexistence, consistent with pronounced hysteresis observed in experiments. The phase diagram presented in this work can serve as a reference for future studies of PbSe and should prove useful in the development of accurate and efficient force fields.

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### I. INTRODUCTION

Lead chalcogenides, PbX (X = S, Se, and Te), are widely studied semiconductor materials with applications in optoelectronics, sensors, and thermoelectrics [1]. PbX quantum dots are used as versatile building blocks for nanomaterials due to their narrow bandgap and strong interparticle interactions in self-assembled superlattices [2,3]. The bandgap of these materials can be tuned over a wide range by applying external pressure [4–8]. A particularly sudden and dramatic change of the electronical and mechanical materials properties can be induced via structural transformations [9]. By controlling nanoscale morphology, semiconductors can even be trapped in high-pressure crystal structures that are unstable in the bulk [10–12]. Controlling material properties of PbX via pressure processing, however, requires knowledge of the structural stability and phase behavior of these materials.

The pressure-induced structural phase transitions of lead selenide (PbSe), which are the focus of this work, have been studied experimentally [13–16] and computationally [6,8,15–19] for several decades. At ambient conditions, PbSe crystallizes in the semiconducting B1 phase (NaCl-type structure,  $Fm\overline{3}m$ , No. 225). At pressures of a few GPa, an intermediate orthorhombic semiconducting phase is observed; the metallic high pressure phase is B2 (CsCl-type,  $Pm\overline{3}m$ , No. 221). Several crystal structures have been suggested for the intermediate phase, including B16 (GeS-type, Pnma, No. 62), B27 (FeB-type, Pnma, No. 62), and B33 (CrB-type, Cmcm, No. 63) [14–16,18,19]. In this work, we report results for the B33 structure, since it is the structure with the highest symmetry and the free energy-volume curves are assumed to be very similar for all three possible intermediate structures.

One of the first accurate measurements of the transition pressures of PbSe was reported by Chattopadhyay et al. [13] using high pressure x-ray diffraction with synchrotron radiation. At room temperature, they report transition pressures for  $B1 \rightarrow B33$  of approximately 4.5 GPa and for  $B33 \rightarrow B2$  of approximately 16 GPa. A more recent room temperature x-ray diffraction study by Streltsov et al. [14] reports a measurable persistence of the B1 structure to at least 7.28 GPa, indicating that coexistence of the B1 and B33 phases can be observed over a broad pressure range. Strong hysteresis was also reported by Li et al. [15] who observed the onset of the B1  $\rightarrow$ B33 transition at 4.8 GPa and 2.9 GPa in the forward and back directions, respectively, at room temperature. The onset of the  $B33 \rightarrow B2$  transition was observed at 19.5 GPa in that study. Wang et al. [16] recently reported a temperature-pressure phase diagram for the B1 and B33 structures, displaying temperature-dependent hysteresis and a positive Clapeyron slope: At room temperature the onset of the B1  $\rightarrow$  B33 transition was found at 3.48 GPa, whereas at 1000 K the B33 → B1 back-transition commenced at 6.12 GPa.

To our knowledge, available computational *ab initio* studies of the transition pressures of PbSe are restricted to density functional theory (DFT) calculations at zero temperature, which ignore finite temperature effects. Furthermore, the published DFT results do not present a consistent picture. Depending on the choice of DFT functional, reported transition pressures range from 6.2 to 10.0 GPa for the B1  $\leftrightarrow$  B33 transition and from 16.39 to 22.6 GPa for the B33  $\leftrightarrow$  B2 transition [8,15,17,18]. These different results directly reflect the problem of choosing the best DFT functional for a given material. Moreover, as pointed out by Skelton *et al.* [20], temperature plays an important role in lead chalcogenides and

FIG. 1. Supercells of the three considered phases B1, B33, and B2 of PbSe, as used for phonon calculations. Pb atoms are shown in gray, Se atoms in green color. Primitive cells are indicated by dark red lines.

needs to be considered in *ab initio* investigations to obtain accurate results.

In this work, we present an ab initio study of the phase diagram of PbSe by employing the random phase approximation (RPA) [21,22] for the electronic contributions (i.e., at  $T = 0 \,\mathrm{K}$ ) in order to reduce the variability associated with different DFT functionals. Applications of the RPA to solids are becoming increasingly popular, as RPA outperforms DFT in systematic benchmark studies [23-25]. In addition, low complexity implementations [26] of the RPA lend themselves to the study of large systems, including the B33 phase of PbSe studied here, which involves 56 valence electrons per primitive cell at a volume of approximately 200 Å<sup>3</sup>. Nevertheless, the RPA is not exact and some shortcomings have been identified, motivating the research of systematic corrections to the RPA [25,27–32]. Still, as described in [33,34] the corrected RPA methods provide only small quantitative changes to transition pressures, hence we expect no significant qualitative changes.

To treat finite temperature effects, we include phonon contributions to the free energy in the quasiharmonic approximation using the PBEsol functional [35]. Following the strategy proposed in Ref. [36], the choice of PBEsol was made after comparing the pressure-volume curves calculated with the PBE [37], PBEsol [35], and SCAN [38] functionals with those obtained by RPA.

### II. THEORY AND METHODS

At given pressure p and temperature T, the phase with the lowest molar Gibbs free energy g is thermodynamically stable. To obtain the relations g(p,T) for the three crystal structures, we first calculate molar Helmholtz free energies f(v,T) at several temperatures and molar volumes v. Interpolations were performed using the Birch-Murnaghan equation

of state [39]. Gibbs free energies are then obtained via the Legendre transform,

$$g = f + pv$$
,  $p = -\left(\frac{\partial f}{\partial v}\right)_T$ . (1)

Helmholtz free energies are calculated as the sum of electronic and phononic contributions,  $f = f_{el} + f_{phon}$ , neglecting electron phonon interactions. More precisely, the Born-Oppenheimer approximation was assumed for electronic degrees of freedom and the quasiharmonic approximation was used for lattice vibrations. The electronic contributions to the free energy were calculated within the RPA, as implemented in the Vienna *ab initio* simulation package (VASP) [26,40,41]. For the phonon contributions, we computed harmonic force constants using large supercells of the B1, B33, and B2 phases, as illustrated in Fig. 1, with DFT and the PBEsol functional in VASP. Phonon free energies were then calculated with the phonopy program [42] in the quasiharmonic approximation, which is a reasonable approach for lead chalcogenides [20]. For the intermediate B33 phase, relaxations of the unit cell shape and atom positions were performed for each considered cell volume. Due to its superior computational speed, the PBEsol functional was also used for these calculations. (Note, however, that RPA force calculations are already implemented in VASP [43]). All (free) energies and volumes are reported in units of meV and Å<sup>3</sup> per PbSe, respectively.

# III. COMPUTATIONAL DETAILS

#### A. The crystal structures of B1, B33, and B2

The construction of primitive cells for the B1 and B2 crystal lattices is straightforward, since the volume is the only free parameter for these structures. For the B33 phase, the unit cell geometry and atom positions depend on the volume of the cell. Therefore, we performed lattice relaxations using

TABLE I. Relaxed unit cells of the B33 phase (CrB-type, *Cmcm*, No. 63) for different volumes, containing 4 Pb atoms and 4 Se atoms. For the relaxation, the PBEsol functional was used with a  $12 \times 4 \times 12$  *k*-point mesh and a plane wave cutoff of 700 eV.

Volume (Å <sup>3</sup> )	lattice parameters (Å)	Pb Wyckoff site	Se Wyckoff site	
168.37	a = 3.8441, b = 10.4987, c = 4.1719	4c, (0, 0.3826, <sup>1</sup> / <sub>4</sub> )	4c, (0, 0.1270, 1/4)	
195.68	a = 4.0622, b = 11.1456, c = 4.3218	4c, (0, 0.3792, 1/4)	$4c, (0, 0.1300, \frac{1}{4})$	
222.98	a = 4.2415, b = 11.8782, c = 4.4258	4c, (0, 0.3717, <sup>1</sup> / <sub>4</sub> )	4c, (0, 0.1342, 1/4)	

TABLE II. List of explicitly treated valence electrons, core radii  $r_c$ , and energy cutoffs ENMAX for the PAW potentials.

Element	Valence	$r_c$ (Å)	ENMAX (eV)			
	For free energy calculation of electrons					
Pb	$5s^2 5p^6 5d^{10} 6s^2 6p^2$	2.3	317			
Se	$4s^2 4p^4$	212				
	For free energy calcula	ation of phonon	IS			
Pb	$5s^2 5d^{10} 6s^2 6p^2$	2.5	238			
Se	$4s^2 4p^4$	2.1	212			

the PBEsol functional at several volumes. As a starting point we took the experimentally measured lattice parameters given in Table 1 in Ref. [16]. Structural parameters of relaxed structures at three selected volumes can be found in Table I.

### B. Pseudopotentials, basis set, and k-point meshes

All VASP calculations are based on the frozen core approximation and the projector augmented wave [44] method (PAW), using the potentials specified in Table II.

The size of the plane wave basis is controlled by the kinetic energy cutoff (ENCUT flag in VASP). The k-point mesh is specified by three numbers  $(k_1 \times k_2 \times k_3)$ , corresponding to a uniform sampling of the Brillouin zone in each direction of the reciprocal lattice. We applied the following criteria to determine the energy cutoff and the density of the k-point mesh. For all electronic free energy calculations using DFT, we required the total free energy to be converged to within 1 meV per PbSe. For all electronic free energy calculations based on the RPA, we only required the free energy differences to be converged within 1 meV per PbSe. The resulting energy cutoffs and k-point meshes can be found in Table III. For phonon calculations, we built supercells containing 128 atoms for B1 and B2, and 144 atoms for B33, as illustrated in Fig. 1. To determine energy cutoffs and k-point meshes, we required a convergence of the zero point energy to within 1 meV per PbSe. Parameters used in phonon calculations can be found in Table IV.

### C. Error estimates

The transition pressure  $p_{XY}$  of a transition from phase X to phase Y (at fixed temperature) obeys the equation,

$$p_{XY} = -\frac{f_X(v_X) - f_Y(v_Y)}{v_X - v_Y},$$
 (2)

TABLE III. Computational parameters for the electronic contribution to the free energy (at  $T=0\,\mathrm{K}$ ).

Phase	$k_1 \times k_2 \times k_3$	ENCUT
	PBEsol	
B1	$8 \times 8 \times 8$	550
B33	$12 \times 4 \times 12$	550
B2	$17 \times 17 \times 17$	550
	RPA	
B1	$9 \times 9 \times 9$	450
B33	$9 \times 3 \times 9$	450
B2	$9 \times 9 \times 9$	450

TABLE IV. Computational parameters for the phonon contributions to the free energy. Supercells were constructed by replicating primitive cells  $n_i$  times along the respective lattice directions.

Phase	$k_1 \times k_2 \times k_3$	ENCUT	$n_1 \times n_2 \times n_3$	No. of atoms	
B1	$2 \times 2 \times 2$	400	$4 \times 4 \times 4$	128	
B33	$2 \times 1 \times 2$	500	$3 \times 2 \times 3$	144	
B2	$3 \times 3 \times 3$	400	$4 \times 4 \times 4$	128	

where  $v_X$  and  $v_Y$  are the volumes of the phases X and Y at the transition pressure, respectively. Hence, we can estimate the error  $\delta p_{XY}$  of the transition pressure as

$$\delta p_{\rm XY} \leqslant \frac{2\delta f}{|v_{\rm X} - v_{\rm Y}|},\tag{3}$$

where  $\delta f$  is the error of the free energy due to unconverged basis sets and k-point grids. According to the convergence criteria described in Sec. III B, we can safely assume that  $\delta f \leq 2$  meV. Furthermore, we obtain a latent volume larger than  $2 \text{ Å}^3$  (see Fig. 4 and Table V) for both transitions. We thus estimate a computational error of

$$\delta p_{\rm XY} \leqslant \frac{4 \,\mathrm{meV}}{2 \,\mathring{\mathrm{A}}^3} \approx 0.3 \,\mathrm{GPa}$$
 (4)

for both transitions.

## IV. RESULTS

### A. Comparison of DFT functionals with RPA

To select a DFT functional for phonon calculations, we first calculated electronic free energy-volume curves  $f_{\rm el}(v)$  using all three functionals and the RPA, as illustrated in Fig. 2. Pressure-volume relations (i.e., equations of state) calculated from these data via  $p=-\partial f_{\rm el}/\partial v$  are shown in Fig. 3. Note that the calculated pressures originate only from the electronic contribution to the free energy (at T=0 K). As evident from Fig. 3, the PBEsol functional provides the overall best performance compared to RPA results. Accordingly, we chose the PBEsol functional for all DFT calculations.

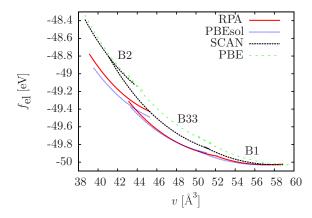


FIG. 2. Electronic free energy-volume curves (at  $T=0\,\mathrm{K}$ ) for each phase calculated using three different DFT functionals and the RPA. The curves are shifted such that the minima of the free energy coincide for the B1 phase.

TABLE V. Free energy and volume data of the B1 and B33 phases at coexistence. Here  $p_{\text{B1}\leftrightarrow\text{B33}} = -\Delta f/\Delta v$  is the transition pressure, v is the volume,  $\Delta v = v_{\text{B33}} - v_{\text{B1}}$  is the volume difference, and  $\Delta f = f_{\text{B33}} - f_{\text{B1}}$  is the free energy difference, calculated as the sum of the electronic free energy difference  $\Delta f_{\text{el}}$  and the phonon free energy difference  $\Delta f_{\text{phon}}$ .

	No zero point vibrations 5.0		T = 0  K $5.0$		T = 300  K $4.6$		$T = 800 \mathrm{K}$ $3.9$	
$p_{\mathrm{B1}  o \mathrm{B33}}$ (GPa)								
	B1	B33	B1	B33	B1	B33	B1	B33
υ	53.16	50.92	53.30	51.06	53.99	51.90	55.66	53.51
$\Delta v$	-2.24		-2.24		-2.09		-2.16	
$\Delta f_{ m el}$	70		70		66		65	
$\Delta f_{ m phon}$	0		0		-6		-12	
$\Delta f$	70		70		60		53	

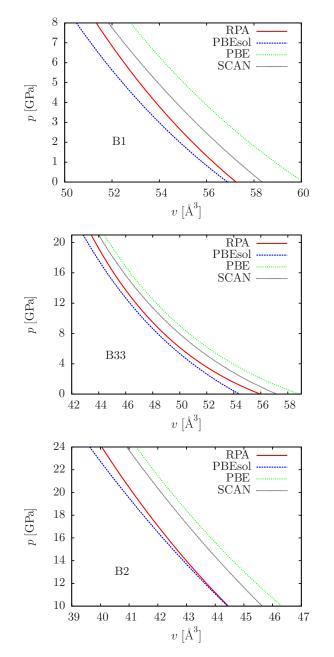


FIG. 3. Pressure-volume relations calculated from the electronic free energy-volume curves ( $T=0\,\mathrm{K}$ ) shown in Fig. 2.

### B. The temperature-pressure phase diagram

Total free energies  $f = f_{el} + f_{phon}$  as a function of volume and temperature are shown in Fig. 4. The three crystal structures display conventional thermodynamic properties, including positive thermal expansion coefficients and increasing pressure with increasing temperature (at fixed volume). (Note that the latter implies that the entropy of each phase decreases with decreasing volume.) Gibbs free energies as a function of T and p are obtained from these data via the Legendre transform Eq. (1). The resulting temperature-pressure phase diagram of PbSe crystals is shown in Fig. 5. At room temperature, we find transition pressures of  $4.6 \pm 0.3$  GPa for the B1  $\leftrightarrow$  B33 transition and 18.7  $\pm$  0.3 GPa for the B33  $\leftrightarrow$  B2 transition. We provide free energy and volume differences of the B1 and B33 structures at three points along the coexistence curve in Table V. As evident from Fig. 5, the RPA significantly stabilizes the intermediate B33 phase compared to PBEsol. The Clapeyron slope is clearly negative for both phase transitions. Interestingly, this result is in contrast to the experimentally measured phase diagram reported in Ref. [16], which shows broad hysteresis but suggests a positive Clapeyron slope.

The negative slope can be explained by softer phonon modes in the high pressure phases. In Fig. 6 we show the phonon density of states for all phases at volumes close to the

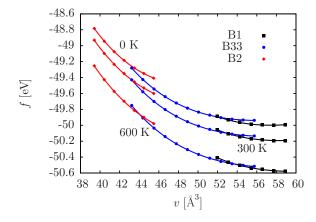


FIG. 4. Free energy-volume curves of the B1, B33, and B2 structures, including electronic (RPA) and phononic (DFT/PBEsol) contributions, at three selected temperatures. For the free energy-volume data see Supplemental Material [45].

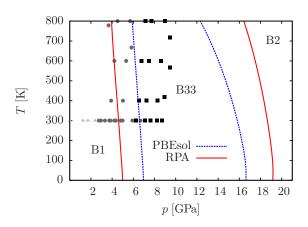


FIG. 5. The temperature-pressure phase diagram of PbSe, obtained with pure DFT (blue curves) and the RPA (red curves). Experimental results from Ref. [16] are included: B1 (triangles), B33 (squares), and B1/B33 coexistence (circles). In contrast to Ref. [16] we find a negative Clapeyron slope for both phase transitions.

phase transitions at  $T=300\,\mathrm{K}$ . For the first transition (B1  $\to$  B33, top panel in Fig. 6) the average phonon frequencies of both phases are equal within 0.1%, explaining why the transition pressure is largely independent of the temperature. The slightly negative Clapeyron slope  $\frac{dT}{dp} = \frac{\Delta v}{\Delta s} < 0$  signifies

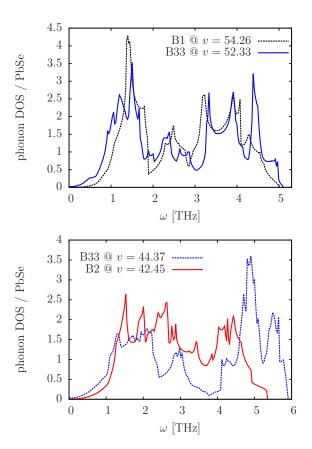


FIG. 6. Phonon density of states (DOS) of the B1, B33, and B2 phases. The volumes are close to the transition volumes for the B1  $\leftrightarrow$  B33 transition (top) and for the B33  $\leftrightarrow$  B2 transition (bottom) at T=300 K.

a larger entropy s for the B33 phase at higher temperature. This is related to a larger density of states at low frequencies in the B33 phase, caused by the presence of rather soft acoustic modes in the layered B33 structure. The transition from the B33 to the B2 phase is accompanied by a strong reduction of the average phonon frequencies, which implies a larger entropy s of the B2 phase at higher temperatures (bottom panel in Fig. 6). For this transition, the Clapeyron slope is clearly negative. The softer phonon modes in the B2 phase are mainly related to an *increase* of the nearest neighbor distances from 2.72 Å (B33 at  $v = 44.37 \,\text{Å}^3$ ) to 3.02 Å (B2 at  $v = 42.45 \,\text{Å}^3$ ), as the B2 phase is more densely packed than the B33 phase.

#### V. DISCUSSION AND CONCLUSION

We have calculated the temperature-pressure phase diagram of three crystal structures (B1, B33, B2) of PbSe using the *ab initio* method RPA. The RPA yields accurate free energy differences and thus provides a more accurate phase diagram of PbSe, compared to DFT results. By employing the RPA for the electronic free energy calculations at zero temperature, the computational error associated with the choice of DFT functional is reduced. The results presented in this work can serve as an accurate benchmark for the development of classical force fields that enable dynamic studies of transition mechanisms.

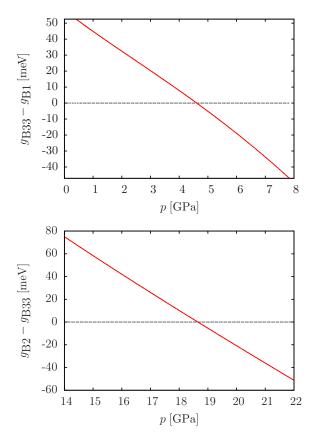


FIG. 7. Gibbs free energy differences  $\Delta g$  per PbSe at  $T=300 \, \mathrm{K}$  for the B1  $\leftrightarrow$  B33 and B33  $\leftrightarrow$  B2 transitions. The small gradients of  $\Delta g$  ( $\approx$ 15 meV/GPa) are consistent with the experimentally observed hysteresis and broad coexistence ranges.

We observe pronounced temperature effects for both transitions, with transition pressures changing by approximately 20% when the temperature is increased from zero to 800 K. Our calculated transition pressures at room temperature ( $4.6\pm0.3\,\mathrm{GPa}$  and  $18.7\pm0.3\,\mathrm{GPa}$ ) lie well within the experimentally reported ranges of 2.9–7.28 GPa for the B1  $\leftrightarrow$  B33 transition and 16–19.5 GPa for the B33  $\leftrightarrow$  B2 transition. We find that the transition pressures *decrease* with increasing temperature, in apparent contradiction with experiments by Wang *et al.* [16]. However, we note that the scatter in the experimental data is substantial, with a fairly broad region of coexistence, so that a careful reassessment of the experimental data might be required.

While we have focused on PbSe in this work, it is instructive to consider the phase behavior of the other lead chalcogenides, PbS and PbTe. At ambient pressure, these materials display similar physical and electronic properties [46]; all three materials crystallize in the B1 structure, with typical trends of increasing unit cell dimension and decreasing bulk modulus from PbS to PbTe [20]. The phase behavior at elevated pressure, however, depends on the chalcogen. Recent measurements of electrical resistance suggest a positive Clapeyron slope for the B1 to B33 transition of PbS [47]. PbTe, on the other hand, does not transform

to the B33 structure but adopts the B27 structure as an intermediate [48].

Experimental studies of structural phase transitions are often plagued by strong hysteresis. If transitions are initiated by nucleation followed by growth, classical nucleation theory (CNT) provides a means for estimating hysteresis widths. According to CNT, the free energy barrier to nucleation  $\Delta G_{
m nuc}$ depends strongly on the molar free energy difference  $\Delta g$ between the two phases,  $\Delta G_{\rm nuc} \propto |\Delta g|^{-2}$ . Small gradients of  $\Delta g$  in the vicinity of the transition point, as observed in this work (Fig. 7), thus result in large nucleation barriers and pronounced hysteresis. Furthermore, coexistence of crystal structures is often observed experimentally over a broad range of conditions, due to the powder-crystalline nature of experimental samples and kinetic effects associated with grain boundaries and other crystal defects. These experimental realities can complicate an accurate determination of coexistence curves and might well explain the discrepancy between this work and Ref. [16]. We expect that our calculations will provide useful guidance for future experiments. For instance, the observation of a negative Clapeyron slope  $\frac{dT}{dp} = \frac{\Delta v}{\Delta s} < 0$  signifies a negative latent heat of transformation and a larger entropy s of the high-pressure phase. More experiments are likely needed to clarify the thermodynamic details of structural transformations in PbSe.

- [1] D. Khokhlov, *Lead Chalcogenides: Physics and Applications* (CRC Press, Boca Raton, 2002).
- [2] T. Hanrath, J. J. Choi, and D.-M. Smilgies, ACS Nano 3, 2975 (2009).
- [3] W. J. Baumgardner, K. Whitham, and T. Hanrath, Nano Lett. 13, 3225 (2013).
- [4] J. M. Besson, W. Paul, and A. R. Calawa, Phys. Rev. 173, 699 (1968).
- [5] S. H. Wei and A. Zunger, Phys. Rev. B 55, 13605 (1997).
- [6] Z. Nabi, B. Abbar, S. Méçabih, A. Khalfi, and N. Amrane, Comput. Mater. Sci. 18, 127 (2000).
- [7] A. Svane, N. E. Christensen, M. Cardona, A. N. Chantis, M. van Schilfgaarde, and T. Kotani, Phys. Rev. B **81**, 245120 (2010).
- [8] P. Bhambhani, K. Kabra, B. K. Sharma, and G. Sharma, J. Solid State Phys. 2014, 921092 (2014).
- [9] K. Bian, Z. Wang, and T. Hanrath, J. Am. Chem. Soc. 134, 10787 (2012).
- [10] T. Wang, R. Li, Z. Quan, W. S. Loc, W. A. Bassett, H. Xu, Y. C. Cao, J. Fang, and Z. Wang, Adv. Mater. 27, 4544 (2015).
- [11] M. Grünwald, K. Lutker, A. P. Alivisatos, E. Rabani, and P. L. Geissler, Nano Letters 13, 1367 (2013).
- [12] K. Jacobs, J. Wickham, and A. P. Alivisatos, J. Phys. Chem. B 106, 3759 (2002).
- [13] T. Chattopadhyay, H. G. von Schnering, W. A. Grosshans, and W. B. Holzapfel, Phys. B+C 139-140, 356 (1986).
- [14] S. V. Streltsov, A. Y. Manakov, A. P. Vokhmyanin, S. V. Ovsyannikov, and V. V. Shchennikov, J. Phys. Condens. Matter 21, 385501 (2009).
- [15] Y. Li, G. Li, C. Lin, X. Li, and J. Liu, J. Appl. Phys. **116**, 053502 (2014).

- [16] S. Wang, C. Zang, Y. Wang, L. Wang, J. Zhang, C. Childs, H. Ge, H. Xu, H. Chen, D. He, and Y. Zhao, Inorg. Chem. 54, 4981 (2015).
- [17] R. Ahuja, Phys. Status Solidi 235, 341 (2003).
- [18] Y. Bencherif, A. Boukra, A. Zaoui, and M. Ferhat, Mater. Chem. Phys. 126, 707 (2011).
- [19] F. Demiray and S. Berber, Phys. Scr. 88, 015603 (2013).
- [20] J. M. Skelton, S. C. Parker, A. Togo, I. Tanaka, and A. Walsh, Phys. Rev. B 89, 205203 (2014).
- [21] P. Nozières and D. Pines, Phys. Rev. 111, 442 (1958).
- [22] J. Harl and G. Kresse, Phys. Rev. Lett. 103, 056401 (2009).
- [23] J. Harl, L. Schimka, and G. Kresse, Phys. Rev. B **81**, 115126 (2010).
- [24] J. Paier, X. Ren, P. Rinke, G. E. Scuseria, A. Grüneis, G. Kresse, and M. Scheffler, New J. Phys. 14, 043002 (2012).
- [25] X. Ren, P. Rinke, C. Joas, and M. Scheffler, J. Mater. Sci. 47, 7447 (2012).
- [26] M. Kaltak, J. Klimeš, and G. Kresse, Phys. Rev. B 90, 054115 (2014).
- [27] A. Grüneis, M. Marsman, J. Harl, L. Schimka, and G. Kresse, J. Chem. Phys. 131, 154115 (2009).
- [28] T. Olsen and K. S. Thygesen, Phys. Rev. B **86**, 081103(R) (2012).
- [29] T. Olsen and K. S. Thygesen, Phys. Rev. Lett. 112, 203001 (2014).
- [30] P. Bleiziffer, M. Krug, and A. Görling, J. Chem. Phys. (2015) 142.
- [31] B. Mussard, D. Rocca, G. Jansen, and J. G. Ángyán, J. Chem. Theory Comput. 12, 2191 (2016).

- [32] M. Hellgren, N. Colonna, and S. de Gironcoli, Phys. Rev. B 98, 045117 (2018).
- [33] N. Sengupta, J. E. Bates, and A. Ruzsinszky, Phys. Rev. B **97**, 235136 (2018).
- [34] N. K. Nepal, A. Ruzsinszky, and J. E. Bates, Phys. Rev. B 97, 115140 (2018).
- [35] G. I. Csonka, J. P. Perdew, A. Ruzsinszky, P. H. T. Philipsen, S. Lebègue, J. Paier, O. A. Vydrov, and J. G. Ángyán, Phys. Rev. B 79, 155107 (2009).
- [36] M. Bokdam, J. Lahnsteiner, B. Ramberger, T. Schäfer, and G. Kresse, Phys. Rev. Lett. 119, 145501 (2017).
- [37] J. P. Perdew, K. Burke, and M. Ernzerhof, Phys. Rev. Lett. 77, 3865 (1996).
- [38] J. Sun, A. Ruzsinszky, and J. P. Perdew, Phys. Rev. Lett. 115, 036402 (2015).
- [39] F. Birch, Phys. Rev. 71, 809 (1947).

- [40] G. Kresse and J. Hafner, Phys. Rev. B 47, 558 (1993).
- [41] G. Kresse and D. Joubert, Phys. Rev. B 59, 1758 (1999).
- [42] A. Togo and I. Tanaka, Scr. Mater. 108, 1 (2015).
- [43] B. Ramberger, T. Schäfer, and G. Kresse, Phys. Rev. Lett. 118, 106403 (2017).
- [44] P. E. Blöchl, Phys. Rev. B 50, 17953 (1994).
- [45] See Supplemental Material at http://link.aps.org/supplemental/ 10.1103/PhysRevB.98.144103 for free energy volume data.
- [46] Y. Zhang, X. Ke, C. Chen, J. Yang, and P. R. C. Kent, Phys. Rev. B 80, 024304 (2009).
- [47] S. Wang, J. Zhang, Y. Zhang, A. Alvarado, J. Attapattu, D. He, L. Wang, C. Chen, and Y. Zhao, Inorg. Chem. 52, 8638 (2013).
- [48] G. Rousse, S. Klotz, A. M. Saitta, J. Rodriguez-Carvajal, M. I. McMahon, B. Couzinet, and M. Mezouar, Phys. Rev. B 71, 224116 (2005).