Multiferroicity and magnetoelastic coupling in α-Mn₂O₃: A binary perovskite

Mohit Chandra,¹ Satish Yadav,¹ R. J. Choudhary,¹ R. Rawat,¹ A. K. Sinha,² Marie-Bernadette Lepetit,³ and Kiran Singh^{1,*}

¹UGC-DAE Consortium for Scientific Research, University Campus, Khandwa Road, Indore 452001, India

³Institut Néel, CNRS UPR 2940, 25 avenue des Martyrs, 38042 Grenoble, France

and Institut Laue Langevin, 72 avenue des Martyrs, 38042 Grenoble, France

(Received 29 June 2018; revised manuscript received 24 August 2018; published 25 September 2018)

Multiferroics, where at least two primary ferroic orders are present and coupled in a single system, constitute an important class of materials. They have attracted special consideration as they present both intriguing fundamental physics problems and technological importance for potential multifunctional devices. Here, we present the evidence of multiferroicity and magnetoelectric (ME) coupling in α -Mn₂O₃; a unique binary perovskite. Corresponding to the antiferromagnetic (AFM) ordering around 80 K, a clear frequency-independent transition is observed in the dielectric permittivity. We showed that electric polarization emerges near the AFM regime that can be modulated with a magnetic field. The detailed structural analysis using synchrotron radiation x-ray diffraction demonstrates the increase in structural distortion with decreasing temperature, as well as changes in the unit cell parameters and bond lengths across the ferroelectric and magnetic ordering temperatures. This observation of multiferroicity and magnetoelastic coupling in α -Mn₂O₃ provides insights for the exploration of ME coupling in related materials.

DOI: 10.1103/PhysRevB.98.104427

I. INTRODUCTION

The search for new multiferroics materials has increased enormously in recent years due to their fascinating interplay between magnetic and electric order parameters [1-4]. The magnetoelectric (ME) coupling was first observed in Cr₂O₃ in 1960 [5–8]. It gained a strong impetus in the recent years after the discovery of very large ME coupling in the TbMnO₃ and TbMn₂O₅ compounds [9,10]. The exploration of ME materials has been mainly sustained by the conjunction of the physical challenge in the understanding of their unique properties and by the prospect of technological applications in spintronics. It has resulted in a remarkable expansion of the field [2,11–21]; notice that the transition metal oxides are the most suitable family where other ME materials can be found.

Manganese presents various stable oxidation states in its oxides that exhibit many fascinating properties, as for instance, colossal magnetoresistance with a record over 14 orders of magnitude in resistivity change under magnetic field [22], and giant magnetocapacitance [23]. Among these oxides, Mn_2O_3 has attracted a lot of attention due to its various applications in energy and environmental fields [24–26]. The chemical formula of the Mn_2O_3 compound can be rewritten as $MnMnO_3$ which is analogous to an ABO_3 perovskite, with Mn ions at both the A and B sites. In perovskites, the magnetic and electronic properties are mainly ascribed by B-O-B interactions. The replacement of the A site cation by a transition metal leads to a new class of compounds, where magnetic and transport properties can be controlled both by the A-O-A and A-O-B interactions, in addition to the B-O-B

ones. Mn_2O_3 is a unique material, not only as it is the only binary oxide which crystallizes in a perovskite structure, but also because the Mn cations occupying A and B sites have the same charge states [27]. Among trivalent ion sesquioxides of Ti, V, Cr, Fe, and Ga, Mn_2O_3 is the only one that does not crystallize in a corundum-type structure. Initially, α -Mn_2O_3 was considered to be crystallized in cubic symmetry with space group *Ia*-3 (*bixbyite*) [28,29]. Subsequently, Geller *et al.* reported that it rather presents an orthorhombic (*Pcab*) structure at room temperature (RT) [28] and undergoes cubic transition around 308 K [30,31]. We will see in this work that this space group should be further probed, at least at low temperature.

Magnetically α -Mn₂O₃ is antiferromagnetic (AFM) with an ordering temperature around ~80 K [32,33]. The magnetic structure of this material is very complex and has not been solved unequivocally. Initially only one transition was observed around 80 K, but later a second transition was detected around 25 K [30] and suggested to be *first order*. The powder neutron diffraction (PND) results show that the magnetic unit cell is equivalent to the crystal unit cell [34–37]. Regulski *et al.* [36] suggested a collinear AFM structure; however, recently, Cockayne *et al.* [37] suggested the possibility of collinear and noncollinear magnetic arrangements.

At RT, the cubic to orthorhombic distortion is very small and it is difficult to distinguish the two symmetry groups using laboratory based diffraction experiments. It is thus essential to have the resolution of synchrotron x-ray diffraction (SXRD) to see the distortion or possible additional peaks in the diffraction pattern. Stoichiometrically Mn_2O_3 resembles the *first* known ME material, namely, Cr_2O_3 , but has not been explored for its dielectric and ME properties. Here, we report the evidence of electric polar-

²HXAL, Synchrotrons Utilization Section, RRCAT, Indore 452013, India

^{*}kpatyal@gmail.com

ization and ME coupling in α -Mn₂O₃ using temperaturedependent magnetic, dielectric, polarization, and SXRD measurements.

II. EXPERIMENT

The polycrystalline samples of α -Mn₂O₃ were prepared through a standard solid-state reaction method using MnO₂ (99.99%) as reported earlier [38]. The temperature-dependent SXRD measurements were performed at angle dispersive xray diffraction beamline (BL-12) [39] at Indus-2, RRCAT Indore ($\lambda = 0.8002$ Å). The data conversion is made using the FIT2D software [40]. The dc magnetization measurements were done on a 7 T Quantum Design SQUID-VSM. The temperature and magnetic-field dependent complex dielectric measurements were performed using a home-made insert coupled with an 8 T Oxford magnet and a Keysight E4980A LCR meter. A parallel plate capacitor geometry with silver paste as electrodes was used to perform these measurements. The temperature-dependent dielectric measurements were performed at different frequencies (1-100 kHz) and different magnetic fields (0, 5, 10, and 80 kOe). Isothermal magnetodielectric measurements were performed at selected temperatures up to 70 kOe, with a field sweep rate of 5 kOe/min. The remanent electric polarization (P) was measured using a Keithley 6517B electrometer in Coulombic mode. A poling electric field of ±310 kV/m was applied during cooling at 90 K to align the electric dipoles and removed at 8 K; then the sample was short circuited for 30 min to remove the extrinsic charges. After that charge vs time was recorded for 1 h to remove the stray charge (if any) and then charge vs temperature was recorded during warming (1.5 K/min). The same procedure was followed to measure P under magnetic field; however, a magnetic field (50 kOe) was applied at the lowest temperature and kept on during measurements.

III. RESULTS AND DISCUSSION

A. Room temperature synchrotron based x-ray diffraction

The RT laboratory based XRD and neutron powder diffraction (NPD) of the prepared samples were reported earlier [38]; which confirm the phase purity. It is, however, rather difficult to comment conclusively on the exact crystal symmetry from those results. Indeed, the diffraction pattern is equally fitted using the cubic (Ia-3) or orthorhombic (Pcab) space groups (not shown here). This ambiguity is due to very small distortion in the crystal structure at RT. Hence, we performed SXRD measurements. The RT Rietveld refined SXRD pattern of α -Mn₂O₃ using the *Pcab* space group is shown in Fig. 1(a). The refined reliability parameters are given inside Fig. 1(a). The refined lattice parameters a = 9.4016(5), b = 9.4052(4),and c = 9.3985(2) Å are consistent with earlier reports [41]. We also fitted the data using cubic symmetry. For the sake of clarity, Bragg reflections [vertical lines in Fig. 1(a)] for cubic as well as orthorhombic symmetries are denoted with vertical pink and green color lines, respectively. From Fig. 1(a), it is clear that there are many additional peaks of small intensity that are not allowed in the cubic space group. For example, at $2\theta \sim 24.30^{\circ}$ [see inset of Fig. 1(a)], there is a weak peak



FIG. 1. (a) Rietveld refinement of room temperature (RT) SXRD data. The green and pink vertical lines represent Bragg's reflections for orthorhombic and cubic symmetry, respectively. Inset shows the zoomed view of SXRD to illustrate the orthorhombic peak. (b) Schematic of crystal structure at RT.

denoted as (034), and $h^2 + k^2 + l^2 = 25$, which is allowed in the *Pcab* space group but is forbidden in *Ia*-3 by the extinction rules. The presence of 0kl(l = 2n) planes is supporting the orthorhombic space group [41]. We also observed peaks at d = 3.134 and 1.167 Å which are consistent with Geller *et al.* [28]. We were not able to resolve the splitting of the $h^2 + k^2 + l^2 = 64$ {008} and 66 {811} peaks at RT, but nevertheless a peak asymmetry can be observed even at RT. We can thus conclude without ambiguity on the orthorhombic symmetry of Mn₂O₃ at RT. The schematic view of the Mn₂O₃ crystal structure is pictured in Fig. 1(b). It consists of 35 Mn atoms per unit cell in an octahedral environment. All the octahedra are distorted, even at RT. The nearest-neighbor octahedra are bond sharing while each oxygen atom is surrounded by four Mn atoms.



FIG. 2. (a) Magnetic susceptibility (dc) vs temperature behavior at 500 Oe in different protocols; zero field cooled warming (zfc), field cooled cooling (fcc), and field cooled warming (fcw). The inset shows fcc and fcw curves near the 25 K transition. (b) First derivative of fcw magnetization at 500 Oe and M(H) at 2 K.

B. Magnetic susceptibility

The temperature-dependent dc magnetization results are presented in Fig. 2. Two clear magnetic transitions can be seen in Fig. 2(a); one at 80 K and another at 25 K, in agreement with earlier reports [30]. The later transition is merely observed in samples prepared by solid-state reaction [30]. The fact that we observe a clear transition in polycrystalline samples corroborates the good quality of our samples. As mentioned earlier, the 25 K transition was suggested to be of first-order nature [30] but no clear evidence was given. To further explore the nature of this transition, we performed zero field cooled (zfc) warming, field cooled cooling (fcc), and field cooled warming (fcw) magnetic measurements at 500 Oe. A clear hysteretic behavior is observed in fcc and fcw curves between 20 and 30 K [see inset of Fig. 2(a)]. This observation clearly establishes the first-order nature of the 25 K transition. The inverse magnetic susceptibility follows a Curie-Weiss law between 200 and 300 K (not shown here). The calculated effective paramagnetic moment (μ_{eff}) per Mn³⁺ ion is found to be $4.87 \,\mu_{\rm B}$, which is approximately equal to the theoretical spin-only value of $4.90 \,\mu_{\rm B}$ for a high-spin Mn³⁺ ion and is consistent with earlier work [32].

The first derivative of the fcw magnetic susceptibility is shown in Fig. 2(b). It evidences a sharp transition around both 80 and 25 K. Moreover, there is a broad feature around 50 K, and a tail-like feature (shaded area) above the sharp peak at 80 K that ends around 110 K. This tail should be put into perspective with the remanence of the (100) magnetic Bragg peak up to over 100 K, seen in neutron scattering by Regulski et al. [36]. These effects can be interpreted as the sign of a magnetic order parameter onset around 110 K. Whether this onset is associated with a true long-range order or a short-range one is an open discussion. Indeed, on one hand, heat capacity measurements do not exhibit any sign of a phase transition in the 105-115 K temperature range, while clear anomalies can be seen at 80 K [42]; on the other hand Regulski et al. [36] argue against a short-range order as the width of the (100) Bragg peak is the same below and above the 80 K magnetic transition. The isothermal magnetization M(H) at 2 K is shown in the inset of Fig. 2(b), which perfectly illustrates the linear behavior expected from an antiferromagnet.

C. Dielectric permittivity

The α -Mn₂O₃ is a semiconducting material with high resistivity even at RT, which is the prerequisite for dielectric studies. The temperature and frequency dependent complex dielectric behavior is measured from 5 to 300 K. The temperature-dependent real part of the dielectric permittivity at 100 kHz during warming (1 K/min) is presented in Fig. 3(a). A clear transition is observed at 80 K which coincides with the AFM transition and suggests a correlation between magnetic and dielectric behavior in this material. This transition is frequency independent and is shown in the Supplemental Material [43]. The change in magnitude of the dielectric permittivity is very small below 120 K. The $tan\delta$ value is also very small below 120 K ($<10^{-3}$ at 100 kHz; see the Supplemental Material [43]) which infers the insulating behavior of the studied sample. The sharp change in the dielectric permittivity at 80 K is similar as observed in other well-known ME materials, e.g., YMnO₃, LuMnO₃, MnO, and BaMnF₄, etc. [44–46]. Such a decrease in the dielectric constant at a magnetic transition is sometimes (partly) attributed to the geometrical frustration, as in some RMnO₃ [44]. No visible change in the dielectric permittivity was, however, detected around 25 K.

To see the effect of a magnetic field on the transition temperature, we measured the dielectric behavior under different magnetic fields (0–80 kOe). The data at some selected fields are presented in the inset of Fig. 3(a). As one can see, there is no shift in the transition temperature (80 K) in the dielectric permittivity, nor any field-induced transition at 25 K. This invariance of T_N suggests the robustness of the AFM ordering at least up to 80 kOe. The derivative of the real part of the dielectric permittivity with respect to temperature, measured at zero magnetic field and 100 kHz, is presented in Fig. 3(b). As expected, it exhibits a clear peak at 80 K, whereas a feeble



FIG. 3. (a) Temperature variation of real part of dielectric permittivity at 100 kHz at selected magnetic fields (0 and 5 kOe) during warming (1 K/min). Inset shows dielectric permittivity near T_N at selected fields. (b) Derivative of dielectric permittivity measured without magnetic field at 100 kHz; inset shows magnetodielectric (MD) at 10 K.

change in slope can be noticed around 25–30 K. Importantly, Fig. 3(b) also evidences a cusplike feature (shaded area) above the 80 K transition that lasts up to 110 K. The isothermal magnetodielectric (MD) behavior at 10 K is shown in the inset of Fig. 3(b) where MD is defined as $(\varepsilon'_H - \varepsilon'_{H=0})/\varepsilon'_{H=0}$. The magnitude of MD is rather small but in agreement with other ME materials [47,48]. The comparison of the M(H) and MD results confirms the existence of a ME coupling in this material.

D. Electric polarization

To explore the ferroelectric behavior of this material, temperature-dependent remanent electric *P* is measured as described in the Experiment section. Figure 4 illustrates the variation of *P* with temperature. From this figure, it is clear that *P* decreases sharply above 80 K and becomes *T* independent only around 110 K. The *P* reversal is observed by reversing the polarity of the poling electric field, thus confirming the ferroelectric behavior of α -Mn₂O₃. The remanent *P* is ~40 μ C/m² at 10 K, comparable in amplitude with other multiferroics including Cr₂O₃ [8,9]. A similar experiment is performed in the presence of a 50 kOe magnetic



FIG. 4. Temperature-dependent remanent electric polarization (P) with positive and negative poling electric fields. The blue curve represents P at 50 kOe magnetic field.

field and a clear effect of the field is observed below AFM ordering, confirming the ME coupling in this material. The pyroelectric coefficient (dP/dT) vs temperature is presented in the Supplemental Material [43]. Additionally, a convex downward feature near 15 K and peaklike behavior at 24 K are observed, consistent with magnetization results [Fig. 2(b)]. Finally, there is an additional feature around 50 K; i.e., *P* does not vary smoothly with increasing temperature but has a concave upwards slope around 50 K. This change occurs at the same temperature where the crossing in intensity between the (110) and the (111) magnetic peaks is observed in neutron scattering [35]. Similarly, it is associated with the anomalies seen in the 1% Fe doped Mossbauer results [30] as well as our magnetization data.

E. Temperature-dependent structural evolution

In order to understand the structural correlations between the magnetic and dielectric properties, we performed T-dependent SXRD down to 6 K. Our analysis reveals that the crystal structure remains orthorhombic down to 6 K. However, the distortion increases with decreasing temperature, and the splitting of some peaks can clearly be seen at higher 2θ angles. The RT SXRD pattern is very well refined within the Pcab space group. This group belongs to the mmm point group (centrosymmetric) and does not allow electric polarization. Our experimental results, however, demonstrate the emergence of an electric polarization below 110 K, thus suggesting a lowersymmetry, noncentrosymmetric space group at least below 110 K. At this point one should remember two things: first that the polarization measurements using the pyroelectric method see only changes in the polarization and not absolute values; second, that the crystal structure of the RMn₂O₅ compounds, a benchmark example of ME materials, was believed to be centrosymmetric for a long time [10], but was recently found noncentrosymmetric even at RT. Such a discovery required a revision of the origin of the polarization in the magnetically ordered state as a spin-enhanced process rather than a spininduced one [48]. We thus performed a symmetry analysis and



FIG. 5. (a–d) Rietveld refinement of SXRD patterns at some selected temperatures. (e) Normalized intensity of selected 2θ range to illustrate the splitting of {800} and {811} peaks at selected temperatures.



FIG. 6. Temperature-dependent (a) lattice parameters and (b) lattice distortion (b/c) (left y axis) and unit cell volume (right y axis). Evolution of bond lengths as a function of temperature: (c) $Mn_{4(1)}-O_{5(2)}$ (d) $Mn_{3(1)}-O_{5(2)}$, (e) $Mn_{3(1)}-O_{4(1)}$; inset in (c) shows a schematic representation of oxygen $O_{5(2)}$ with four Mn ions at 65 K.

looked for *Pcab* subgroups allowing an electric polarization. The *Pca2*₁ group is the only polar maximal subgroup of *Pcab* losing only one generator (index 2). We thus analyzed the temperature-dependent SXRD patterns using both space groups–*Pcab* as well as *Pca2*₁.

In $Pca2_1$ the 8b and 8c Wyckoff positions split into 4a + 4a; that is, the Mn₃, Mn₄, Mn₅, and all oxygens split into double setting. The calculated diffraction in $Pca2_1$ allows additional Bragg reflection (130) around $2\theta = 15.45^{\circ}$ but we did not observe any peak intensity (whatever the temperature) at this 2θ value within our experimental resolution. The refined unit cell parameters remained the same in both space groups.

The Rietveld refined SXRD patterns using $Pca2_1$ at some selected temperatures are shown in Figs. 5(a)-5(d). They reveal good agreement between calculated and observed patterns. To show the distinction/similarity in our refined pattern using both space groups, we presented the refinement at 80 K using $Pca2_1$ and Pcab in Figs. 5(b) and 5(c). We have not observed any significant difference in the refinements and unit cell parameters. The main intensity peak does not undergo any splitting down to lowest temperature; in fact, the maximum changes are observed at the higher 2θ range. Hence, for visual inspection, the diffraction pattern is normalized with the main peak intensity. The normalized intensity of the selected 2θ range, mainly related to {800} and {811} family, is presented in Fig. 5(e) at selected temperatures. In addition to the $h^2 + k^2 + l^2 = 64$ and 66 there are peaks with $h^2 + k^2 + l^2 = 65$. It is worth mentioning here that the x and y scales for all patterns in Fig. 5(e) are the same. Figure 5(e) distinctly demonstrates the change/distortion in lattice with decreasing temperature. For instance, at 300 K, there is no clear splitting in the above-mentioned peaks but there are some asymmetric features, whereas splitting enhances progressively with decreasing temperature. The vertical red arrow in the 240 K data is shown as a guide to the eyes to demonstrate that this height increases with decreasing temperature. One can clearly notice the change in the peak intensities across magnetic transitions; e.g., in the 2θ range $39.60^{\circ}-40.10^{\circ}$, there were only two visible peaks up to 140 K, whereas one additional peak starts appearing in the middle of these two peaks below 110 K (see peaks for 100-60 K). With further decrease in temperature (50-6 K), the intensity of the (800) peak becomes less visible because the relative intensity of these peaks increases; see especially at 20 K. These results infer the change in lattice distortion across the magnetic transitions and speaks in favor of a magnetoelastic coupling.

The evolution of the unit cell parameters with temperature is presented in Figs. 6(a) and 6(b). Near RT the three lattice parameters have similar values, as expected when one is

TABLE I. Bond lengths and bond angles around oxygen (O_5) at selective temperatures across magnetic transition.

Bonds/Angles	105 K	65 K	6 K
Bond lengths (Å)			
$Mn_2-O_{5(1)}$	2.0348 (6)	1.9911 (7)	1.9644(4)
Mn ₂ -O ₅₍₂₎	1.9993(7)	1.9956(6)	2.0389(6)
$Mn_{3(1)}-O_{5(2)}$	1.8686(1)	1.8545(4)	1.8147(5)
$Mn_{3(2)}-O_{5(1)}$	1.9166(7)	1.9272(2)	1.9303(1)
$Mn_{4(1)}-O_{5(2)}$	1.9741(1)	2.0218(7)	2.0522(5)
$Mn_{4(2)}-O_{5(1)}$	1.9507(2)	1.9757(1)	1.9949(6)
$Mn_{5(1)}-O_{5(2)}$	2.2309(4)	2.1990(3)	2.1805(2)
$Mn_{5(2)}-O_{5(1)}$	2.1835(4)	2.2067(1)	2.23708
Bond angles (deg)			
$Mn_{5(1)}-O_{5(2)}-Mn_{4(1)}$	95.2275	94.7489(2)	93.8494
$Mn_{5(1)}-O_{5(2)}-Mn_{3(1)}$	115.6564	116.5622	119.2851
$Mn_{3(1)}-O_{5(2)}-Mn_{2(1)}$	136.0936	136.0947	135.812
Mn ₂₍₁₎ -O ₅₍₂₎ -Mn ₄₍₁₎	101.9773	101.1528	98.3620

close to a cubic to orthorhombic phase transition ($T_c = 308$ K [30,31]). However, they differ significantly with decreasing temperature [see Fig. 6(a)]. The distortion parameter (b/c) increases with decreasing temperature [Fig. 6(b)] with a maximum plateau between 80 and 110 K, then it starts to decrease, consistent with earlier reports [37]. A small but clear change in all the lattice parameters as well as in the unit cell volume is observed at the 80 K and 25 K magnetic transitions.

The temperature evolution of selected Mn bond lengths obtained from refined patterns is presented in Figs. 6(c)-6(e). The temperature evolution is different for the different Mn-O bonds. For example, the $Mn_{4(1)}$ -O₅₍₂₎ bond length decreases with increasing temperature [Fig. 6(c)] whereas the $Mn_{3(1)}-O_{5(2)}$ bond length increases with increasing temperature [Fig. 6(d)]. A clear change in these bond lengths can be seen around 80 K with a tail-like feature up to 110 K. The bond lengths of $Mn_{3(1)}$ - $O_{4(1)}$ are displayed in Fig. 6(e). These bond lengths show very clear anomalies at \sim 30 and \sim 80 K, that coincide with the magnetic transitions. In addition, these bond lengths show a broad peaklike feature between 80 and 110 K. A schematic view of the four Mn atoms surrounding the $O_{5(2)}$ is shown in the inset of Fig. 6(c). The details of some bond lengths and bond angles at selected temperatures are given in Table I.

Our SXRD results confirm that α -Mn₂O₃ crystallize in an orthorhombic symmetry at RT; however, these data do not present any direct experimental evidence favoring the polar $Pca2_1$ space group over the nonpolar Pcab one, while our polarization results clearly imply a symmetry lowering from Pcab to a polar subgroup. Moreover, the refined structures in the two space groups are associated with different temperature evolution of the Mn-O bond lengths. Indeed, while the $Pca2_1$ -refined structure exhibit clear signs of the three anomalies (the two magnetic transitions at 25 K and 80 K and the tail like trend at polar/magnetoelectric transition at 110 K), these signatures are not as obvious when the structure is refined in the Pcab space group; see Fig. S4 in the Supplemental Material [43]. These observations suggest that $Pca2_1$ could be the correct space group.

The dielectric/polarization and magnetic measurements clearly reveal the magnetoelectric nature of α -Mn₂O₃ through various signs of coupling between the magnetic and polar orders. The electric polarization emerges around 110 K that is at higher temperature than the main transition observed in dielectric and magnetization results (~80 K). However, a clear tail like feature is observed in the derivative of magnetic susceptibility at the onset of the polar order, coherent with the earlier report on the presence of magnetic peak (100) up to \sim 110 K. Despite our efforts the exact nature of this transition remains unclear. Indeed, Regulski et al. [36] pointed out that the width of the (100) magnetic peak below and above 80 K is temperature independent and hence cannot be related to short-range magnetic ordering. They suggested the existence of a structural distortion at ~110 K as seen in our SXRD results and consistent with our polarization results. In our hypothesis of $Pca2_1$ space group, the observed electric polarization up to 110 K could originate from spin components due to magnetoelectric coupling like in $Dy_2Mn_2O_5$ [49]. The one to one correspondence of the features in the magnetic, dielectric, electric polarization and structural properties as well as the sensitivity of the polarization to an applied magnetic field demonstrate the coupling between these different order parameters in α -Mn₂O₃.

IV. CONCLUSION

To summarize, we performed a comprehensive study on α -Mn₂O₃ to explore its multiferroicity and magnetoelastic coupling by investigating its structural, magnetic, dielectric and polarization behavior. The structural analysis confirmed the orthorhombic structure at room temperature that remains invariant down to 6 K. However, the lattice distortion (b/c) increases with decreasing temperature and shows a maximum around 100 K. Clear changes in lattice parameters observed across the magnetic transitions support the importance of the magnetoelastic coupling. Associated with the magnetic transition, a clear anomaly is observed in the dielectric permittivity. Similarly, the electric polarization can be tuned by a magnetic field, showing the existence of a ME coupling in this material. The emergence of an electrical polarization suggest that the structural space group could be $Pca2_1$, a polar subgroup of *Pcab*. Our detailed studies thus unambiguously established multiferroicity and magnetoelastic coupling in a unique binary perovskite material α -Mn₂O₃ above liquid nitrogen temperature and suggest the search for such coupling in related materials.

Note added. During the review process of our manuscript, we became aware of a report on the multiferroicity in ξ -Mn₂O₃ which is stabilized at high-pressure and high-temperature conditions [see Cong *et al.* [50].

ACKNOWLEDGMENTS

The authors would also like to thank Prof. E. V. Sampathkumaran for the encouragement and Dr. Vengadesh for his help in performing initial dielectric and magnetic measurements. The authors are also thankful to Dr. Archana Sagdeo and Mr. M. N. Singh for their help during low-temperature SXRD measurements.

- [1] M. Fiebig, J. Phys. D: Appl. Phys. **38**, R123 (2005).
- [2] S.-W. Cheong and M. Mostovoy, Nat. Mater. 6, 13 (2007).
- [3] W. Eerenstein, N. D. Mathur, and J. F. Scott, Nature (London, UK) 442, 759 (2006).
- [4] D. Khomskii, Physics 2, 20 (2009), and references therein.
- [5] I. E. Dzyaloshinskii, Zh. Eksp. Teor. Fiz. 37, 881 (1959) [Sov. Phys. JETP 10, 628(1960)].
- [6] D. N. Astrov, Zh. Eksp. Teor. Fiz. 38, 984 (1960) [Sov. Phys. JETP 11, 708 (1960)].
- [7] G. T. Rado and V. J. Folen, Phys. Rev. Lett. 7, 310 (1961).
- [8] A. Iyama and T. Kimura, Phys. Rev. B 87, 180408(R) (2013).
- [9] T. Kimura, T. Goto, H. Shintani, T. Arima, and Y. Tokura, Nature (London, UK) 426, 55 (2003).
- [10] N. Hur, S. Park, P. A. Sharma, J. S. Ahn, S. Guha, and S.-W. Cheong, Nature (London, UK) 429, 392 (2004).
- [11] G. Lawes, A. B. Harris, T. Kimura, N. Rogado, R. J. Cava, A. Aharony, O. Entin-Wohlman, T. Yildrim, M. Kenzelmann, C. Broholm, and A. P. Ramirez, Phys. Rev. Lett. 95, 087205 (2005).
- [12] K. Taniguchi, N. Abe, T. Takenobu, Y. Iwasa, and T. Arima, Phys. Rev. Lett. 97, 097203 (2006).
- [13] Q. Zhang, K. Singh, C. Simon, L. D. Tung, G. Balakrishnan, and V. Hardy, Phys. Rev. B 90, 024418 (2014).
- [14] Y. Yamasaki, S. Miyasaka, Y. Kaneko, J. P. He, T. Arima, and Y. Tokura, Phys. Rev. Lett. 96, 207204 (2006).
- [15] K. Singh, V. Caignaert, L. C. Chapon, V. Pralong, B. Raveau, and A. Maignan, Phys. Rev. B 86, 024410 (2012).
- [16] K. Singh, C. Simon, E. Cannuccia, M. B. Lepetit, B. Corraze, E. Janod, and L. Cario, Phys. Rev. Lett. 113, 137602 (2014).
- [17] K. Singh, T. Basu, S. Chowki, N. Mahapotra, K. K. Iyer, P. L. Paulose, and E. V. Sampathkumaran, Phys. Rev. B 88, 094438 (2013).
- [18] S. Ghara, E. Suard, F. Fauth, T. T. Tran, P. S. Halasyamani, A. Iyo, J. Rodríguez-Carvajal, and A. Sundaresan, Phys. Rev. B 95, 224416 (2017).
- [19] Y. S. Oh, S. Artyukhin, J. J. Yang, V. Zapf, J. W. Kim, D. Vanderbilt, and S.-W. Cheong, Nat. Commun. 5, 3201 (2014).
- [20] M. Fiebig, T. Lottermoser, D. Meier, and M. Trassin, Nat. Rev. Mater. 1, 16046 (2016), and references therein.
- [21] K. Ramesh, L. Chen, F. Chen, Y. Liu, Z. Wang, and Y. F. Han, Catal. Today 131, 477 (2007).
- [22] D. Saurel, C. Simon, A. Pautrat, C. Martin, C. Dewhurst, and A. Brület, Phys. Rev. B 82, 054427 (2010).
- [23] T. Goto, T. Kimura, G. Lawes, A. P. Ramirez, and Y. Tokura, Phys. Rev. Lett. 92, 257201 (2004).
- [24] G. H. Lee, H. S. Huh, J. W. Jeong, J. B. Cho, H. K. Seung, and C. R. Hyeong, J. Am. Chem. Soc. 124, 12094 (2002).
- [25] Y. Deng, Z. Li, Z. Shi, H. Xu, F. Peng, and G. Chen, RSC Adv. 2, 4645 (2012).
- [26] T. Yamashita and A. Vannice, J. Catal. 163, 158 (1996).
- [27] S. V. Ovsyannikov, A. M. Abakumov, A. A. Tsirlin, W. Schnelle, R. Egoavil, J. Verbeeck, G. T. Van, K. V. Glazyrin,

M. Hanfland, and L. Dubrovinsky, Angew. Chem., Int. Ed. **52**, 1494 (2013).

- [28] S. Geller, J. A. Cape, R. W. Grant, and G. P. Espinosa, Phys. Lett. A 24, 369 (1967).
- [29] L. Pauling and M. D. Shappell, Z. Kristallogr. 75, 128 (1930).
- [30] R. W. Grant, S. Geller, J. A. Cape, and G. P. Espinosa, Phys. Rev. 175, 686 (1968).
- [31] S. Geller and G. P. Espinosa, Phys. Rev. B 1, 3763 (1970).
- [32] R. G. Meisenheimer and D. L. Cook, J. Chem. Phys. **30**, 605 (1959).
- [33] S. Geller, R. W. Grant, J. A. Cape, and G. P. Espinosa, J. Appl. Phys. 38, 1457 (1967).
- [34] J. Cable, M. Wilkinson, E. Woolan, and W. Koehler, Oak Ridge National Laboratory Physics Division Semiannual Progress Report No. ORNL-2302, 1957, p. 43.
- [35] R. R. Chevalier, G. Roult, and E. F. Bertaut, Solid State Commun. 5, 7 (1967).
- [36] M. Regulski, R. Przenioslo, I. Sosnowska, D. Hohlwein, and R. Schneider, J. Alloys Compd. 362, 236 (2004).
- [37] E. Cockayne, I. Levin, H. Wu, and A. Llobet, Phys. Rev. B. 87, 184413 (2013).
- [38] M. Chandra, S. Yadav, S. Rayaprol, and K. Singh, in *Proceedings of DAE Solid State Physics Symposium*, 2017, Mumbai, AIP Conf. Proc. No. 1942 (AIP, New York, 2018), p. 110023.
- [39] A. K. Sinha, A. Sagdeo, P. Gupta, A. Upadhyay, A. Kumar, M. N. Singh, R. K. Gupta, S. R. Kane, A. Verma, and S. K. Deb, J. Phys: Conf. Ser. 425, 072017 (2013).
- [40] A. P. Hammersley, S. O. Svensson, and A. Thompson, Nucl. Instrum. Methods Phys. Res. A 346, 312 (1994).
- [41] S. Geller, Acta Crystallogr. B 27, 821 (1971).
- [42] R. A. Robie and B. S. Hamingway, J. Chem. Thermodyn. 17, 165 (1985).
- [43] See Supplemental Material at http://link.aps.org/supplemental/ 10.1103/PhysRevB.98.104427 for temperature-dependent dielectric constant and tan δ at different frequencies; pyroelectric coefficient (dP/dT) vs temperature, and temperature-dependent bond lengths using the *Pcab* space group.
- [44] T. Katsufuji, S. Mori, M. Masaki, Y. Moritomo, N. Yamamoto, and H. Takagi, Phys. Rev. B 64, 104419 (2001).
- [45] M. S. Seehra and R. E. Helmick, Phys. Rev. B 24, 5098 (1981).
- [46] G. A. Samara and P. M. Richards, Phys. Rev. B 14, 5073 (1976).
- [47] K. Singh, M.-B. Lepetit, Ch. Simon, N. Bellido, S. Pailhes, J. Varignon, and A. De Muer, J. Phys.: Condens. Matter 25, 416002 (2013).
- [48] T. D. Sparks, M. C. Kemei, P. T. Barton, R. Seshadri, E-D. Mun, and V. S. Zapf, Phys. Rev. B. 89, 024405 (2014); S. Sharma, T. Basu, A. Shahee, K. Singh, N. P. Lalla, and E. V. Sampathkumaran, *ibid.* 90, 144426 (2014).
- [49] V. Balédent, S. Chattopadhyay, P. Fertey, M. B. Lepetit, M. Greenblatt, B. Wanklyn, F. O. Saouma, J. I. Jang, and P. Foury-Leylekian, Phys. Rev. Lett. **114**, 117601 (2015).
- [50] J. Cong, K. Zhai, Y. Chai, D. Shang, D. D. Khalyavin, R. D. Johnson, D. P. Kozlenko, S. E. Kichanov, A. M. Abakumov, A. A. Tsirlin *et al.*, Nat. Commun. 9, 2996 (2018).