Role of local structural distortion in driving ferroelectricity in GdCrO₃

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Temperature-dependent synchrotron x-ray diffraction and extended x-ray absorption fine structure (EXAFS) studies were performed to understand the role of structural characteristics in driving the magnetoelectric multiferroic properties of $GdCrO_3$. The results suggest that the distortion in the structure appears to be associated with the off-center displacement of Gd atoms together with octahedral rotations via displacement of the oxygen ions in $GdCrO_3$. Further, a comparative EXAFS study of $GdCrO_3$ with a similar system, $YCrO_3$, suggests that the oxygen environment of Gd in $GdCrO_3$ is different from that of Y in $YCrO_3$, which results in an orthorhombic $Pna2_1$ structure in $GdCrO_3$ in contrast to the monoclinic $P2_1$ structure in $YCrO_3$.

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I. INTRODUCTION

Magnetoelectric multiferroics have drawn great interest in recent years due to their multifunctionality for a wide variety of potential device applications in modern technologies [1–4]. The family of rare-earth chromites (RCrO₃) has been recognized as promising systems for multiferroicity at reasonably high temperatures [5–8]. But the conflicting observations of the ferroelectric behavior at relatively high temperature and the average centrosymmetric lattice (Pbnm) and magnetic structure (G-type) in these systems remained a puzzling issue in this series of compounds [5,9]. GdCrO₃, a member of the RCrO₃ family, shows magnetic and ferroelectric transitions simultaneously at around 169 K (T_M) [5] in contrast to YCrO₃ having a ferroelectric transition at 473 K and a magnetic transition at 140 K [10]. In addition, the strength of the electric polarization in GdCrO₃ is one order of magnitude less than that of YCrO₃ [5,10]. Our recent report on temperature-dependent x-ray diffraction (XRD) studies along with first-principles density-functional-theory calculations showed that GdCrO₃ possesses noncentrosymmetric orthorhombic $Pna2_1$ structure [11], in contrast to the monoclinic $P2_1$ (noncentrosymmetric) structure in YCrO₃ as reported earlier [10,12], which leads to a weaker polar property in GdCrO₃ compared to YCrO₃. Phonon instability studies in the high-symmetry cubic perovskite structure showed that there is a strong similarity between GdCrO₃ and YCrO₃ as the polar distortion in both the systems is associated with the R-ion displacement [10,11,13]. Furthermore, as the ferroelectric instability is very weak in these systems, the local noncentrosymmetry has been suggested to play a crucial role in driving ferroelectricity [10,12]. Thus, it is essential to study the short-range structural order in these systems to understand the origin of their ferroelectric properties. Since extended x-ray absorption fine structure (EXAFS) is a powerful tool for local structure investigations, we performed detailed EXAFS studies in conjunction with temperature-dependent XRD to understand the structure of $GdCrO_3$. These studies show that a strong correlation exists between the presence of local distortion and its implication in the ferroelectric ordering and the global structure of the system. Further, we also discuss a comparative EXAFS study of $GdCrO_3$ with the similar chromite system, $YCrO_3$.

II. EXPERIMENTAL DETAILS

Polycrystalline samples of GdCrO₃ (YCrO₃) were prepared by the solid-state synthesis technique using stoichiometric proportions of Gd_2O_3 (Y_2O_3) and Cr_2O_3 and details are described elsewhere [14,15]. Phase purity of the samples was confirmed by powder XRD measurements carried out in a D8 advanced diffractometer equipped with Cu K_{α} radiation. Temperaturedependent XRD measurements were carried out at the XRD1 beamline at the Elettra synchrotron radiation facility using photons with a wavelength of 0.85507 Å. Rietveld refinements of the diffraction patterns were performed using the FULLPROF package. Temperature-dependent EXAFS measurements were carried out at P-65 beamline at PETRA-III synchrotron source, DESY, Hamburg, Germany. Both incident (I_0) and transmitted (I_t) photon intensities were recorded simultaneously using ionization chambers filled with appropriate gases at Gd L_3 (7243 eV) and Cr K (5989 eV) edges in GdCrO₃ and Y K (17038 eV) and Cr K edges in YCrO₃. The raw data collected were background subtracted and normalized to extract EXAFS signals through a series of steps using the ATHENA software [16]. Thereafter, the fitting of the EXAFS spectrum with a specific model obtained from basic crystallographic information was carried out using the ARTEMIS software [16]. The software computes the theoretical spectrum from the given model using the ATOMS and FEFF6 programs [16–18].

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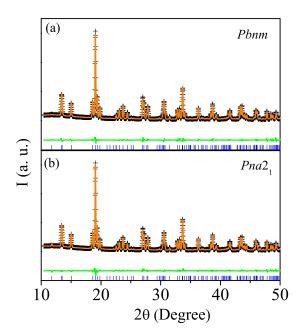


FIG. 1. Room-temperature x-ray powder diffraction patterns (symbols) of $GdCrO_3$ with corresponding refinement patterns (solid curve) using the (a) Pbnm and (b) $Pna2_1$ space groups.

III. RESULTS AND DISCUSSION

Rietveld refinement of the XRD patterns of GdCrO₃ were performed using both centrosymmetric Pbnm and noncentrosymmetric Pna2₁ space groups throughout the entire (studied) temperature range (100-300 K). It was found that both the space groups fit with better goodness-of-fit parameters in the entire range. The XRD pattern acquired at 300 K along with corresponding Rietveld refinement data using Pbnm and $Pna2_1$ space groups are depicted in Figs. 1(a) and 1(b), respectively. The reliability parameters obtained for the Pbnm space group are $R_w \sim 0.084$, $R_{\rm expt} \sim 0.041$, and $\chi^2 \sim 4.16$ and the corresponding parameters for the $Pna2_1$ structure are $R_w \sim 0.083$, $R_{\rm expt} \sim 0.041$, and $\chi^2 \sim 4.37$. The similarity in reliability parameters is possibly due to the small structural changes between the two space groups as $Pna2_1$ is a subgroup of Pbnm. To verify the structure, we used the program PLATON [19,20] and found that the symmetry of the structure is better described in $Pna2_1$ throughout the entire temperature range. The noncentrosymmetric Pna2₁ structure in GdCrO₃ was also supported by density-functional theory calculations, favoring the polar nature of the system as reported earlier [11].

In addition to x-ray diffraction studies, temperature-dependent EXAFS measurements were also performed to extract information about the local structure around selected atoms. The EXAFS technique is useful to provide valuable information about the structural peculiarities and allows one to verify different structural models. Figures 2(a) and 2(b) show k^3 -weighted EXAFS data in k space at Cr K and Gd L_3 edges, respectively, acquired at room temperature for GdCrO₃, along with corresponding fittings superimposed on it. The magnitudes of Fourier transforms of the corresponding spectra along with fittings are shown in Figs. 2(c) and 2(d), respectively. Various contributions in different regions of the spectra are marked in the figures. The scattering contributions

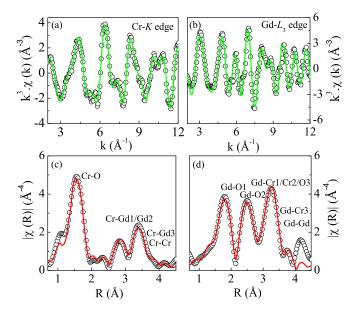


FIG. 2. k^3 -weighted EXAFS data in k space (open circles) at (a) the Cr K edge and (b) the Gd L_3 edge acquired at room temperature for GdCrO₃, along with corresponding fittings (solid lines). The magnitudes of Fourier transforms spectra (open circles) of Cr K edge and Gd L_3 edge along with the corresponding fitting (solid lines) are shown in (c) and (d) respectively. Various contributions in different regions are marked in the figures.

for atomic shells were derived considering both Pbnm and Pna2₁ crystal structures and in both cases the EXAFS data fit well throughout the entire (measured) temperature range having an R factor of ~ 0.01 for the Cr K edge and ~ 0.009 for the Gd L_3 edges. This corroborates that distortions in the structure are very small, in agreement with the XRD results as discussed above. For the Cr K edge, fits were confined to a krange of $3 < k < 12 \text{ Å}^{-1}$ and an R range of 1.2 < R < 4 Å. In this region Cr K EXAFS originates from scattering of photoelectrons from the nearest-neighbor octahedral oxygens of Cr, three Gd subshells with two, four, and two coordination numbers, respectively, and a Cr-Cr coordination shell with four atoms and some multiple scattering contributions. The fitting of the Gd L_3 edge was done in the k range of 3 < $k < 12.5 \text{ Å}^{-1}$ and an R range of 1.1 < R < 4 Å, to model Gd-O and Gd-Cr distributions. The Gd-O distribution is more complex, consisting of three subshells with four, two, and six oxygens, respectively. The Gd-Cr contribution is also split into three subshells with two, four, and two Cr atoms. Additionally, a single Gd-Gd shell is considered with coordination number four. During the fitting procedure, the coordination number was kept fixed, while bond length and mean-square relative displacement (MSRD) $[\sigma^2 = \langle (r - \langle r \rangle)^2 \rangle]$ were used as free parameters. It is seen that MSRD for short (O1) and long (O3) Gd-O bonds are highly correlated; therefore, there is a single σ^2 for them and a separate one for intermediate oxygens (O2). Similarly, two σ^2 were chosen for Gd-Cr bonds: one for short and long bonds in the bc plane and another for intermediate bonds. σ^2 are the most sensitive to modes contributing to radial motions, basically depending only on the local vibrational structure [17].

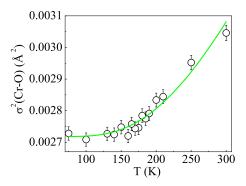


FIG. 3. Temperature dependence of mean-square relative displacement (MSRD) (σ^2) of Cr-O bond correlation of GdCrO₃. The solid line represents the fitting using the Einstein model [Eq. (1)].

The temperature dependencies of σ^2 for all the shells were fitted with the Einstein model, which considers the bond vibrations as harmonic oscillations with a single effective frequency proportional to the Einstein temperature, θ_E , as given by the relation [17,21]

$$\sigma^{2}(T) = \sigma_{0}^{2} + \left(\frac{\hbar^{2}}{2\mu k_{B}\theta_{E}}\right) \coth\left(\frac{\theta_{E}}{2T}\right),\tag{1}$$

where σ_0^2 is the static contribution, T is in kelvins, and μ is the reduced mass of the bond pair. The Einstein temperature θ_E is a measure of the stiffness of the bonds. Fitting of the temperature dependence of σ^2 using the Einstein model for Cr-O bond correlation is shown in Fig. 3. The best fitting yields $\sigma_0^2 = 0.0012(4)$ and $\theta_E = 818(11)$ K. The large value

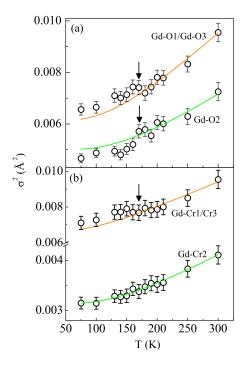


FIG. 4. (a) Temperature dependence of MSRD (σ^2) of (a) Gd-O bond correlations and (b) Gd-Cr bond correlations in GdCrO₃. The solid lines represent the fitting using the Einstein model [Eq. (1)]. The arrows indicate the magnetic/ferroelectric transition temperature.

of θ_E indicates the rigidness of the CrO₆ octahedra [22]. Furthermore, no anomalous change in the σ^2 is observed below 169 K, where both magnetic and ferroelectric ordering are observed.

The fitting of the temperature dependence of σ^2 for Gd-O subshells using Eq. (1) is shown in Fig. 4(a). Fitting yielded $\sigma_0^2 = 0.0015(4)$ and $\theta_E = 391(15)$ K for Gd-O1/O3 bond correlations and $\sigma_0^2 = 0.0011(4)$ and $\theta_E = 458(26)$ K for Gd-O2 bond correlation. The relatively low values of θ_E for Gd-O subshells compared to that for the Cr-O shell suggest that the Gd-O bonds are weaker than Cr-O bonds; i.e., GdO₁₂ polyhedra are not as rigid as CrO₆ octahedra. Furthermore, there is clear deviation of σ_0^2 from the expected behavior in the region near the magnetic/ferroelectric transition temperature, suggesting the presence of structural anomalies around the transition. The temperature-dependent variation of σ^2 for Gd-Cr bond correlations along with fitting using Eq. (1) are shown in Fig. 4(b), giving $\sigma_0^2=0.0044(1)$ and $\theta_E=293(7)$ K for Gd-Cr₁/Cr₃ bonds and $\sigma_0^2=0.0015(3)$ and $\theta_E=428(6)$ K for the Gd-Cr₂ bond. The σ^2 for Gd-Cr₁/Cr₃ bonds which are in the bc plane show an anomaly around the transition, possibly due to the Gd displacements caused by the magnetostriction effect associated with the Gd³⁺-Cr³⁺ interaction [11,23], whereas the Gd-Cr₂ bond shows no anomalous behavior.

For a further understanding of the structural properties, we have extracted various bond lengths from EXAFS fitting and Rietveld refinement of the XRD pattern, and the temperature dependence of average Cr-O bond lengths is depicted in Fig. 5(a). It is observed that the average bond lengths obtained from both EXAFS and XRD analysis are very close and show no anomalies around the transition temperature. Figures 5(b)-5(d) represent the individual bond lengths (Cr-O₁/O₂/O₃) extracted from the XRD analysis, in which O1 occupies the apex and O_2/O_3 occupy the base of the Cr O_6 octahedra. The bond length of Cr-O₁ contracts with decreasing temperature, while in the ab plane the Cr-O₂ bond length decreases and the Cr-O₃ bond length increases with decreasing temperature. The Cr-O₃ bond shows anomalous behavior around the transition, indicating the magnetostriction effect, consistent with the Raman spectroscopy study as reported earlier [11]. The temperature-dependent variation of the bond angles with temperature is shown in Figs. 5(e)-5(g). With decreasing temperature the axial angle (Cr-O₁-Cr) increases and one of the equatorial angles (Cr-O₂-Cr) decreases, whereas the other equatorial angle (Cr-O₃-Cr) increases. The bond angles also exhibit anomalous behavior near the transition temperature, suggesting that there exist distortions in the octahedra probably associated with the off-center displacements of oxygens via octahedral rotations [6,7]. It is known that the off-center displacement of oxygens, generally in the ab plane, is the most important factor for the Dzyaloshinskii-Moriya interaction in the system [24,25].

The temperature-dependent variation of Gd-O and Gd-Cr bond lengths obtained from EXAFS and XRD analyses are shown in Figs. 6(a) and 6(b), respectively. Except for Gd-O₃, all the Gd-O and Gd-Cr bonds obtained from EXAFS analysis match reasonably well with that extracted from the Rietveld refinement of XRD. The Gd-O3 bond obtained from EXAFS fitting is shorter by \sim 0.07 Å than that determined from XRD. Discrepancies between EXAFS and diffraction results may

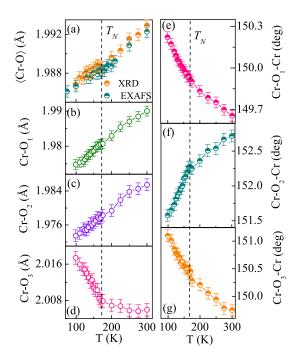


FIG. 5. (a) Temperature dependence of average Cr-O atomic bond length in $GdCrO_3$ obtained from EXAFS and XRD analyses. Temperature dependence of bond lengths $(Cr-O_1/O_2/O_3)$ and bond angles $(Cr-O_1/O_2/O_3-Cr)$, obtained from Rietveld refinement of XRD are shown in (b)–(g), respectively. The vertical dashed lines correspond to the magnetic/ferroelectric transition temperature.

arise from either physical or fictitious effects. The systematic errors in the EXAFS data analysis may originate from the correlation between distances and energy scale parameters. On the other hand, discrepancies can also be possible due to the differences between local and long-range structures as observed in various systems like $La_{1-x}Sr_xMnO_3$ [26],

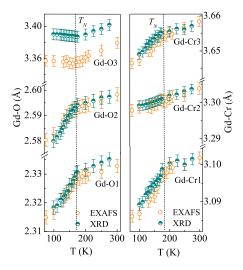


FIG. 6. Temperature dependence of (a) various Gd-O bond lengths and (b) Gd-Cr bond lengths obtained from EXAFS and XRD analyses. The vertical dashed lines correspond to the magnetic/ferroelectric transition temperature.

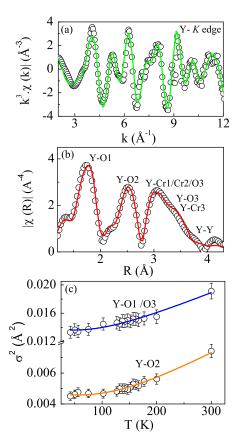


FIG. 7. (a) k^3 -weighted EXAFS data in k space and (b) magnitude of Fourier transform of the data at Y K edge acquired at room temperature for YCrO₃, along with corresponding fittings (solid lines). Various contributions in different regions are marked in the figure. (c) Temperature dependence of MSRD (σ^2) of Y-O bond correlations. The solid lines represent the fitting using the Einstein model [Eq. (1)].

 $La_{1-x}Ca_xMnO_3$ [27], $La/PrCoO_3$ [28], and $Na_{0.5}Bi_{0.5}TiO_3$ [29]. In the present system, the nearly equal lengths of Gd-O3 and Gd-Cr bonds may lead to the suppression of the Gd-O3 contribution. However, the contraction is observed only for one distance (Gd-O3 bond), while all the other bond lengths show a general agreement between XRD and EXAFS results. Further, though there is a difference in the magnitude of the Gd-O3 bond length obtained from XRD and EXAFS analyses, their temperature evolutions are similar. This points towards the true shorter distance of the Gd-O3 bond length, which may be possibly due to the deviations of local structure from the average structure. The change in local structure is mainly associated with the Gd-O coordination sphere, whereas the Cr-O coordination remains less affected, suggesting that the structural distortion is dominated by Gd displacements. Further, all the Gd-O bonds and Gd-Cr bonds except Gd-Cr2 show slope changes around the transition temperature. This anomalous behavior can be understood through magnetostriction effect, which plays a role in ferroelectric distortion [11,23].

For comparison the EXAFS measurements were carried out on a similar chromite system, YCrO₃. Figures 7(a) and 7(b) depict k^3 -weighted EXAFS data in k space and magnitude of Fourier transform of the data, respectively, acquired at the Y K

edge at room temperature for YCrO3, along with corresponding fitting superimposed on these. The scattering contributions for atomic shells were derived and fitted considering the Pbnm crystal structure and R-factor was found to be \sim 0.01 for the entire (measured) temperature range. Fits were confined to the k range of $3 < k < 12.5 \text{ Å}^{-1}$ and an R range of $1.15 < R < 12.5 \text{ Å}^{-1}$ 4 Å. Here the EXAFS was fitted with three nearest-neighbor O shells of four coordinations each, three Y-Cr shells with two, four, and two atoms, respectively, and a single Y shell consisting of four neighbors. A total of five σ^2 parameters were chosen in a similar way to that used for fitting Gd EXAFS in GdCrO₃. The fitting of the temperature dependence of σ^2 for Y-O subshells using Eq. (1) is shown in Fig. 7(c). From the fitting, it is found that $\sigma_0^2 = 0.0083(2)$ and $\theta_E = 341(7)$ K for Y-O1/O3 bond distributions and $\sigma_0^2 = 0.0015(8)$ and $\theta_E = 432(5)$ K for the Y-O2 bond distribution. The relatively low values of θ_E for Y-O subshells indicate YO₁₂ polyhedra are also not rigid enough, like GdO₁₂ in GdCrO₃. These results suggest a close analogy between GdCrO₃ and YCrO₃. However, the Gd-O environment (three subshells with four, two, and six oxygens, respectively) in GdCrO₃ is different from the Y-O environment (three subshells with four oxygens each) in YCrO₃, which leads to an orthorhombic Pna2₁ structure in GdCrO₃ in contrast to the monoclinic P2₁ structure in YCrO₃ [10,11]. The good fitting of EXAFS data using the *Pbnm* structure is possibly due to the tiny distortions in these systems, giving rise to very weak polarizations (GdCrO₃, \sim 0.7 $\mu \text{C/cm}^2$; YCrO₃, ~3 $\mu \text{C/cm}^2$) [5,10].

The distortion in the structure is associated with the offcentering displacement of Gd atoms together with octahedral rotations via movements of the oxygen ions, which lift certain symmetries of the centrosymmetric Pbnm structure [6,7] and possibly stabilize a local noncentrosymmetric Pna2₁ structure. Figure 8 illustrates a schematic representation of the displacement of oxygens around the CrO6 octahedron (octahedral rotation), as indicated by the arrows, and dashed circles represent the possible displacement of Gd atoms. The distortion is dominated mainly by Gd displacements, indicating that Gd-O bond polarization plays a major role in driving ferroelectricity. Moreover, it is found that GdCrO₃ prefers $Pna2_1$ symmetry as discussed earlier, suggesting that some sort of long-range positional disorder is always present in the system [11]. The magnetic coupling leads to extra distortion in the system via the magnetostriction effect and

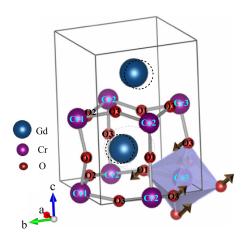


FIG. 8. Visualization of displacement of oxygens around the CrO₆ octahedron (octahedral rotation), as indicated by the arrows and dashed circles, representing possible displacement of Gd atoms in GdCrO₃.

plays a complementary role in the enhancement of ferroelectric polarization [8,11,23].

IV. CONCLUSION

In conclusion, through detailed structural investigations using temperature-dependent XRD and EXAFS studies, it is found that CrO₆ octahedra in GdCrO₃ are rather rigid and the Gd-O bonds are weaker than the Cr-O bonds. Octahedral rotations along with Gd displacements lead to noncentrosymmetric $Pna2_1$ structure in GdCrO₃. The distortion is dominated by Gd displacements, indicating that Gd-O bond polarization plays a major role in driving ferroelectricity in this system.

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