

Probing the $J_{\text{eff}} = 0$ ground state and the Van Vleck paramagnetism of the Ir^{5+} ions in layered $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$

S. Agrestini,^{1,2} C.-Y. Kuo,^{1,*} K. Chen,^{3,†} Y. Utsumi,^{1,‡} D. Mikhailova,^{1,4} A. Rogalev,⁵ F. Wilhelm,⁵ T. Förster,⁶ A. Matsumoto,⁷ T. Takayama,^{7,8} H. Takagi,^{7,8,9} M. W. Haverkort,^{1,10} Z. Hu,¹ and L. H. Tjeng¹

¹Max Planck Institute for Chemical Physics of Solids, Nöthnitzerstr. 40, 01187 Dresden, Germany

²ALBA Synchrotron Light Source, E-08290 Cerdanyola del Vallès, Barcelona, Spain

³Institute of Physics II, University of Cologne, Zùlpicher Str. 77, 50937 Cologne, Germany

⁴Leibniz Institute for Solid State and Materials Research (IFW) Dresden e.V., Helmholtzstr. 20, D-01069 Dresden, Germany

⁵ESRF-The European Synchrotron, 71 Avenue des Martyrs, 38000 Grenoble, France

⁶Hochfeld-Magnetlabor Dresden (HLD-EMFL), Helmholtz-Zentrum Dresden-Rossendorf, 01314 Dresden, Germany

⁷Department of Physics and Department of Advanced Materials, University of Tokyo, 7-3-1 Hongo, Tokyo 113-0033, Japan

⁸Max Planck Institute for Solid State Research, Heisenbergstr. 1, 70569 Stuttgart, Germany

⁹Institute for Functional Matter and Quantum Technologies, University of Stuttgart, Pfaffenwaldring 57, 70569 Stuttgart, Germany

¹⁰Institute for theoretical physics, Heidelberg University, Philosophenweg 19, 69120 Heidelberg, Germany



(Received 22 February 2018; revised manuscript received 9 May 2018; published 29 June 2018)

We report a combined experimental and theoretical x-ray magnetic circular dichroism (XMCD) spectroscopy study at the $\text{Ir-}L_{2,3}$ edges on the Ir^{5+} ions of the layered hybrid solid state oxide $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$ with the K_2NiF_4 structure. From theoretical simulation of the experimental $\text{Ir-}L_{2,3}$ XMCD spectrum, we found a deviation from a pure $J_{\text{eff}} = 0$ ground state with an anisotropic orbital-to-spin moment ratio ($L_x/2S_x = 0.43$ and $L_z/2S_z = 0.78$). This deviation is mainly due to multiplet interactions being not small compared to the cubic crystal field and due to the presence of a large tetragonal crystal field associated with the crystal structure. Nevertheless, our calculations show that the energy gap between the singlet ground state and the triplet excited state is still large and that the magnetic properties of the Ir^{5+} ions can be well described in terms of singlet Van Vleck paramagnetism.

DOI: [10.1103/PhysRevB.97.214436](https://doi.org/10.1103/PhysRevB.97.214436)

I. INTRODUCTION

The class of iridium based oxides has attracted tremendous attention in recent years. The presence of strong spin-orbit coupling (SOC) in the $5d$ shell and associated entanglement of the spin and orbital degrees of freedom may lead to unexpected exotic electronic states. In the present work, we focus on the layered $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$, a material which we were able to synthesize as a single phase and stoichiometric without oxygen deficiency. The two parent compounds Sr_2IrO_4 and Sr_2CoO_4 have very different physical properties, despite having the same crystal structure. Sr_2IrO_4 is a canted antiferromagnet with $T_N = 240$ K, where the strong SOC leads to an insulating state [1]. The pseudospin $J_{\text{eff}} = 1/2$ state has been proposed as the ground state of the Ir^{4+} ions in Sr_2IrO_4 [1], with the magnetic interactions described by a Heisenberg model [2] akin to the spin-1/2 Hamiltonian used to represent the magnetic dynamics in La_2CuO_4 . Hence electron doped Sr_2IrO_4 has raised a huge interest as a possible analog of the hole doped cuprate high-temperature superconductors [3]. Sr_2CoO_4 , instead, is a metallic ferromagnet ($T_c = 250$ K) with the Co^{4+} ions in the

$S = 1$ spin state [4]. Due to the very different electronic and magnetic properties of the two end compounds, Sr_2IrO_4 and Sr_2CoO_4 , the solid state solution $\text{Sr}_2\text{Co}_x\text{Ir}_{1-x}\text{O}_4$ is expected to exhibit interesting physics.

In an early study of the $\text{Sr}_2\text{Co}_x\text{Ir}_{1-x}\text{O}_4$ system, where the authors were able to replace only 30% of Ir ions by Co ions, the observed increase of the effective magnetic moment was interpreted in terms of $\text{Ir}^{4+}/\text{Co}^{4+}$ valence states with Co ions in the intermediate spin state similar to Sr_2CoO_4 [5]. However, a theoretical work [6] proposed that the introduced Co ions in the Sr_2IrO_4 matrix would prefer to have a lower valence state ($3+$) than the $4+$ high valence and hence induce a charge state change of the Ir ions. Indeed, following our recent successful synthesis of $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$, we were able to find experimental evidence in support of the $\text{Ir}^{5+}/\text{Co}^{3+}$ scenario [7,8]. Such a scenario is very interesting, because the magnetic ground state of Ir^{5+} has been recently the subject of debate.

In the limit of strong SOC and large on-site Coulomb energy U , ions with $d^4-t_g^4$ configuration are expected to be in the $J_{\text{eff}} = 0$ ground state. However, the large SOC and U limit of d^4 ions has been questioned by theoretical studies, which proposed that strong intersite hopping may lead to superexchange interactions large enough to cause a magnetic condensation of Van Vleck excitons and the onset of novel magnetic states [9–11]. Later works suggested that the interatomic exchange in Ir^{5+} double perovskites might be too weak to overcome the singlet-triplet gap [12,13]. From the experimental point of view, a recent work reported an antiferromagnetic long-range

*Present address: National Synchrotron Radiation Research Center, 101 Hsin-Ann Road, Hsinchu 30076, Taiwan.

†Present address: Synchrotron SOLEIL, L'Orme des Merisiers, Saint-Aubin, 91192 Gif-sur-Yvette, France.

‡Present address: Institute of Physics, Bijenička 46, 10000 Zagreb, Croatia.

order in the double perovskite Sr_2YIrO_6 [14]. To explain the observed magnetic order Gao *et al.* argued that the noncubic crystal field would cause a suppression of the excitation gap and, as a result, the breakdown of the $J_{\text{eff}} = 0$ state [14]. Subsequent band structure calculations [12,15] suggested the noncubic crystal field to be weak and the origin of the magnetism reported in the double perovskite Sr_2YIrO_6 and Ba_2YIrO_6 is still debated [15–19].

There is a clear need of a clarification of whether a large noncubic crystal field could provide a route for the realization of the excitonic condensation in Ir^{5+} . The case of $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$ is important because its layered structure produces a large tetragonal crystal field acting on the Ir^{5+} ion. Hence this compound allows us to test the robustness of the $J_{\text{eff}} = 0$ state of Ir^{5+} with respect to a large noncubic crystal field. Here we used XMCD spectroscopy, a powerful element-specific technique that allows one to study the magnetic ground state of different ions independently. The experimental data are combined with full atomic multiplet calculations which allowed us to probe the effect the stability of the $J_{\text{eff}} = 0$ state with respect to different perturbations, such as a noncubic crystal field, covalency, and intra-atomic multiplet interactions.

II. EXPERIMENT

Synthesis of the layered $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$ was carried out from stoichiometric powder mixtures of homemade Co_3O_4 with IrO_2 (Umicore) and SrCO_3 (Alfa Aesar, 99.99%) at 1200°C in air for 80 h. Co_3O_4 was obtained by thermal decomposition of $\text{Co}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$ at 700°C in an oxygen flow. In order to obtain fully oxidized $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$ samples for spectroscopic studies, postannealing in steel autoclaves at 400°C and 5000 bar O_2 pressure was performed for five days. The phase analysis and the determination of the unit cell parameters were performed using x-ray powder diffraction [7]. Transition metal cations Co and Ir are in edge-sharing oxygen octahedra that are elongated along the c axis. Single crystals of Sr_2IrO_4 were grown by the flux method.

The XMCD spectra at the $\text{Ir-}L_{2,3}$ edges were measured at the beamline ID12 [20] of ESRF in Grenoble (France) with a degree of circular polarization of about 97%. Spectra were recorded using the bulk sensitive total fluorescence yield detection mode. The XMCD signal was measured in a magnetic field of 17 T with the sample kept at a temperature of 2 K. The $\text{Ir-}L_{2,3}$ x-ray absorption spectra for right and left circularly polarized light were corrected for self-absorption effects. The $\text{Ir } L_3/L_2$ edge-jump intensity ratio $I(L_3)/I(L_2)$ was then normalized to 2.22 [21]. This takes into account the difference in the radial matrix elements of the $2p_{1/2}$ -to- $5d(L_2)$ and $2p_{3/2}$ -to- $5d(L_3)$ transitions. Magnetization measurements in pulsed fields up to 58 T were made using a pair of compensated pickup coils. The pulsed field data was then scaled with low field data obtained by a SQUID MPMS magnetometer.

III. RESULTS

As a first step in our investigation of $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$, one needs to make sure that the Ir magnetism probed by the XMCD technique originates from the Ir^{5+} ions and not from Ir^{4+}

ions' impurities. For this purpose we have performed XMCD measurements also on pure Sr_2IrO_4 , as reference for Ir^{4+} ions sitting on the same local environment as in the investigated compound.

In Fig. 1 we report the results of the $\text{Ir-}L_{2,3}$ x-ray absorption spectroscopy experiments carried out on a polycrystalline pellet of $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$ and on a single crystal of Sr_2IrO_4 at $T = 2$ K and magnetic field $B_{\text{app}} = 17$ T. The x-ray absorption spectra were taken using circular polarized light with the photon helicity being either parallel or antiparallel with respect to the applied magnetic field. The difference spectrum, called XMCD, and the sum spectrum, called XAS, of $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4(\text{Sr}_2\text{IrO}_4)$ are shown in Fig. 1 as blue and red (magenta and green) curves, respectively. The XAS of $\text{Sr}_2\text{ScIrO}_6$ (orange) is also reported in Fig. 1, as an Ir^{5+} reference [22]. In the case of Sr_2IrO_4 the spectra were collected in both normal and grazing incidence with the magnetic field forming an angle of 90° ($B \perp ab$) and 20° ($B \parallel ab$), respectively, with the ab plane. The XMCD of Sr_2IrO_4 measured for $B \perp ab$ is very tiny, 20 times smaller in size than the XMCD measured for $B \parallel ab$, in agreement with the large magnetic anisotropy revealed by magnetization measurements [23]. In Fig. 1 we show only the Sr_2IrO_4 data collected in grazing incidence. The XMCD spectrum of Sr_2IrO_4 is in very good agreement with the data published in literature [24].

It is important to notice that at both the $\text{Ir-}L_3$ and the $\text{Ir } L_2$ edges the XAS of $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$ lies about 1.3 eV higher in energy than the XAS of Sr_2IrO_4 , as illustrated by the vertical dotted lines in Fig. 1. The energy position of the $\text{Ir-}L_3$ XAS of $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$ is the same as that of the Ir^{5+} reference $\text{Sr}_2\text{ScIrO}_6$. It is well known that x-ray absorption spectra at the transition-metal $L_{2,3}$ edges are highly sensitive to the valence state. An increase of the valence of the metal ion by one results in a shift of the $L_{2,3}$ XAS spectra to higher energies by 1 eV or more, as shown by XAS studies on many oxides [25–28], including iridium oxides [22,29,30]. This shift is due to a final state effect in the x-ray absorption process. The energy difference between a $5d^n$ ($5d^5$ for Ir^{4+}) and a $5d^{n-1}$ ($5d^4$ for Ir^{5+}) configuration is approximately $\Delta E = E(2p^6 5d^{n-1} \rightarrow 2p^5 5d^n) - E(2p^6 5d^n \rightarrow 2p^5 5d^{n+1}) = U_{pd} - U_{dd} \sim 1-2$ eV, where U_{dd} is the Coulomb repulsion energy between two $5d$ electrons and U_{pd} the one between a $5d$ electron and the $2p$ core hole. The energy position of the XAS, hence, confirms that the 50% replacement of the iridium ions with cobalt ions in the Sr_2IrO_4 matrix induces an increase of the valence of the remaining iridium ions from $4+$ to $5+$.

The XMCD signal of $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$ at the $\text{Ir-}L_3$ exhibits a similar line shape as that of Sr_2IrO_4 . One might then wonder whether the XMCD signal of $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$ is due to Ir^{4+} impurities considering that the Ir^{5+} ions should be Van Vleck ions in the strong spin-orbit coupling limit. However, by looking carefully at the XMCD spectra one can notice that the energy position is different: the $\text{Ir-}L_3$ XMCD peak of $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$ occurs at about 1.1 eV higher energies than that of Sr_2IrO_4 . Furthermore, the XMCD spectra of the two samples at the $\text{Ir-}L_2$ edge have completely different line shapes: $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$ shows a double peak feature with positive intensity, while Sr_2IrO_4 exhibits only one peak with positive intensity on the high energy side of the $\text{Ir-}L_2$, and a slight negative intensity

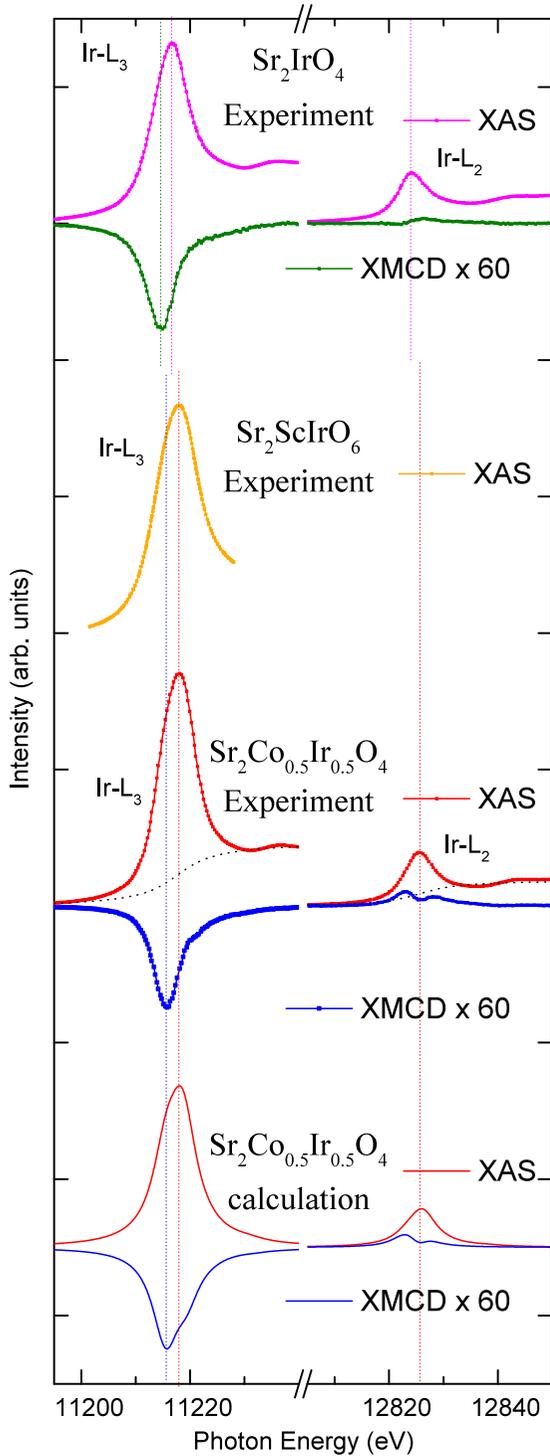


FIG. 1. Experimental Ir- $L_{2,3}$ XAS and XMCD spectra of $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$, red and blue circles, respectively, and of Sr_2IrO_4 , magenta and green circles, respectively, and XAS of $\text{Sr}_2\text{ScIrO}_6$ (orange). The spectra were measured at $T = 2$ K and $B_{\text{app}} = 17$ T. The vertical dotted lines illustrate the energy shift between the spectra of the Ir^{4+} and Ir^{5+} samples. The black dotted curves represent the edge jumps. Bottom: calculated Ir- $L_{2,3}$ XAS (red line) and XMCD (blue line) of $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$.

on the low energy side. These differences in energy position at the Ir- L_3 edge and in spectral line shape at the Ir- L_2 edge

demonstrate that the XMCD signal of $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$ cannot be due to the presence of Ir^{4+} impurities, but is related to the field induced magnetism of the Ir^{5+} ions. We also would like to note that our $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$ XMCD spectrum has different details in the line shape in comparison to the ones reported for $\text{Sr}_2\text{Fe}_{0.5}\text{Ir}_{0.5}\text{O}_4$ and $\text{Sr}_2\text{In}_{0.5}\text{Ir}_{0.5}\text{O}_4$ [30].

The large difference in intensity of the dichroic signal between the Ir L_3 and L_2 edges shown in Fig. 1 indicates clearly that the Ir ions have a significant unquenched orbital moment [31]. In order to extract directly from the spectrum the ratio of orbital and spin moments we have used the sum rules for XMCD developed by Thole *et al.* [31] and Carra *et al.* [32]. The sum rules can be summarized as

$$\frac{L_z}{2S_z + 7T_z} = \frac{2}{3} \frac{\int_{L_{2,3}} (\sigma^+ - \sigma^-) dE}{\int_{L_3} (\sigma^+ - \sigma^-) dE - 2 \int_{L_2} (\sigma^+ - \sigma^-) dE}, \quad (1)$$

where S_z and L_z are the spin and orbital contributions to the local magnetic moment, respectively, and T_z is the intra-atomic magnetic dipole moment. Advantage of this sum rule is that it does not require a saturation of the magnetic moment and can hence provide important information. Applying the sum rules to the Ir- $L_{2,3}$ XMCD spectrum of $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$ gives a ratio $L_z/(2S_z + 7T_z) = 0.45(1)$ for Ir^{5+} . This value is close to the ratio $L_z/2S_z = 0.5$ predicted for a $J_{\text{eff}} = 0$ system, if one neglects T_z . However, neglecting T_z in iridates can be very misleading. In fact, taking in account that T_z increases going from $3d$ to $4d$ and, further, $5d$ transition metals and that $S = 1$ for LS Ir^{5+} ions, then $7T_z$ in $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$ might be actually comparable to $2S_z$.

In order to circumvent this uncertainty problem related to T_z , we have performed configuration-interaction cluster calculations using the Quany Package [33–35]. The desired information can then be directly extracted from these calculations once the calculations can successfully produce an accurate simulation of the experimental XAS and XMCD spectral line shapes. The method uses an IrO6 cluster, which includes explicitly the full atomic multiplet interaction, the hybridization of Ir with the ligands, the crystal field acting on the Ir ion, and the noncubic crystal field acting on the ligands. The hybridization strengths and the crystal field acting on the oxygen ligands were extracted *ab initio* by DFT calculations carried out using the full-potential local-orbital code FPLO [36]. The noncubic crystal field acting on the Ir ion was fine-tuned to best fit the experimental XAS and XMCD spectra. The parameters used in the calculations are listed in Ref. [37]. Since we are dealing with a polycrystalline sample, we simulated the experimental data by summing two calculated spectra: one for light with the Poynting vector in the xy plane and one with the Poynting vector along the z axis, with a weighting ratio 2:1. As explained in a more detailed way later, an exchange field of 16 meV parallel to the magnetic field was introduced in the Hamiltonian in order to reproduce the size of the experimental XMCD signal.

The calculated Ir- $L_{2,3}$ XAS and XMCD spectra are plotted in Fig. 1 as solid red and blue curves, respectively. One can clearly see that the line shapes of the measured Ir- $L_{2,3}$ spectra are very well reproduced by our simulations. The nice agreement between theory and experimental data is also quantitative:

the calculated isotropic [38] ratio $L_z/(2S_z + 7T_z) = 0.46$ is essentially identical to the value of 0.45 extracted from the application of the sum rules to our experimental spectra. From the separate application of the spin and orbital sum rules to our Ir $L_{2,3}$ XMCD data we obtained $2S_z + 7T_z = 0.1\mu_B$ and $L_z = 0.047\mu_B$. These values are in excellent agreement with the calculated isotropic values ($2S_z + 7T_z = 0.09\mu_B$ and $L_z = 0.044\mu_B$) given by our simulations.

The best fit to our experimental spectra is obtained for the t_{2g} orbitals split by an effective tetragonal crystal field of $\Delta_{t_{2g}}^{\text{eff}} = -325$ meV, where this effective crystal field includes the effect of the hybridization with the oxygen ions. The negative sign indicates that the d_{xy} orbital is lower in energy than the d_{xz} and d_{yz} ones. A similar negative $\Delta_{t_{2g}}^{\text{eff}}$ was observed also in the case of Sr_2IrO_4 [39,40]. The value of $\Delta_{t_{2g}}^{\text{eff}}$ is more than 10 times larger than the trigonal t_{2g} splitting (10–20 meV) estimated for the double perovskite Sr_2YIrO_6 [15]. Such a large splitting is of the same order of the SOC (~ 0.4 eV) and should have consequences for the magnetic properties. Indeed, our calculations reveal that there is a strong anisotropic effect: the ratio between the orbital and spin moments is $L_x/2S_x = 0.35$ and $L_z/2S_z = 0.49$ for the magnetic field applied in the xy plane and along the z axis, respectively. Furthermore, the calculated intra-atomic magnetic dipole moment was found to be large along the z direction, i.e., $T_z/S_z = -0.35$, and nearly negligible in the xy plane, i.e., $T_x/S_x = -0.03$. These are the values obtained in the presence of the 16 meV exchange field. Switching off this exchange field, we obtain $L_x/2S_x = 0.43$ and $L_z/2S_z = 0.78$, the values more relevant for low field experiments.

Furthermore, also the parameters used in the calculations to obtain a good fit to the experimental spectra reveal that the $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$ system is strongly covalent and thus far from ionic. Consistent with the high-valence state of the Ir ion, the charge transfer energy is negative, i.e., $\Delta_{\text{CT}} \sim -1.5$ eV. The consequence is that only 7.1% of the ground state of the Ir^{5+} ion has in fact the $5d^4$ character, while the configurations $5d^5\bar{L}$ and $5d^6\bar{L}^2$, where \bar{L} denotes a ligand hole, are dominant. See Table I. On average, the number of electrons in the d bands is $n_e = 5.73$, i.e., almost two electrons are transferred from the oxygens to the Ir ions.

The question now arises to what extent the $J_{\text{eff}} = 0$ state is an accurate description of the ground state of the Ir^{5+} ion in $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$ in view of the presence of the large tetragonal field splitting, strong covalency, and participation of the e_g orbitals in addition to the t_{2g} . To this end it is

TABLE I. Weight of various configurations for the Ir^{5+} ground state.

$5d^4$	$5d^5\bar{L}$	$5d^6\bar{L}^2$	$5d^7\bar{L}^3$
7.0%	30.8%	43.3%	18.8%

instructive to calculate with full atomic multiplet theory the relevant expectation values of the Ir ion quantum numbers for several scenarios as listed in Table II (no exchange field). Starting with an ionic $\text{Ir}^{5+} 5d^5$ ion in octahedral symmetry (O_h) with a large value for the octahedral crystal field splitting of $10Dq = 10$ eV, we find that J_{eff} is 0.07. Here we defined $\mathbf{J}_{\text{eff}} = \mathbf{L}_{\text{eff}} + \mathbf{S}$, where the \mathbf{L}_{eff} operator is obtained by rotating the orbital basis of the \mathbf{L} operator to the cubic harmonics. The rotation matrix was modified to only keep the t_{2g} subset of the d eigenorbitals. After projecting out the e_g orbitals, the angular momentum operator is rotated back to the spherical harmonics. As the covalence mixes the d and ligand orbitals, in order to find good quantum numbers we calculated the expectation value of the $\mathbf{J}_{\text{eff}}^2$ operator acting on both the Ir- d and ligand- d shell, i.e., acting on the total IrO_6 cluster. While the ideal $J_{\text{eff}} = 0$ is the value one obtains when only the t_{2g} orbitals span the Hilbert space, i.e., when the e_g orbitals are completely projected out by making $10Dq$ infinitely large, the $J_{\text{eff}} = 0.07$ value for $10Dq = 10$ eV indicates that this is already close to the ideal situation. We have also calculated the $L_x/2S_x$ (and $L_z/2S_z$) ratio and found a value of 0.49, which is very close to the expected 0.50 number for the pure $J_{\text{eff}} = 0$ state. The magnetic susceptibility is calculated at 1.0×10^{-3} emu/mole/Oe.

Next we lower the octahedral crystal field splitting to the value we find in $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$, namely $10Dq = 3$ eV. See Table II. We find $J_{\text{eff}} = 0.69$, which indicates that we are far away from the ideal $J_{\text{eff}} = 0$ state. Still being in the ionic model, i.e., the hybridization with the oxygen ligands have not been included, this finding indicates that the e_g orbitals contribute significantly to the ground state of the Ir^{5+} ion. This mixing in of the e_g orbitals does not take place on the one-electron level since e_g and t_{2g} belong to different irreducible representations in O_h , but it does take place on the multielectron level via the full atomic multiplet interactions. These multiplet interactions, characterized by the Slater F^2 and F^4 integrals, are indeed not at all small compared to the e_g - t_{2g} crystal field splitting ($10Dq$) and their effect cannot be ignored. The $L_x/2S_x$ (and $L_z/2S_z$) ratio reduces

TABLE II. Calculated J_{eff} , orbital-to-spin ratio, and magnetic susceptibility for different scenarios.

Local symm.	$10Dq^{\text{eff}}$ (eV)	Cov.	J_{eff}	$L_x/2S_x$ $B \parallel x$	$L_z/2S_z$ $B \parallel z$	T_x/S_x $B \parallel x$	T_z/S_z $B \parallel z$	χ_x (emu/mole/Oe)	χ_z (emu/mole/Oe)
O_h	10.0	Ionic	0.07	0.49	0.49	-0.12	-0.12	1.0×10^{-3}	1.0×10^{-3}
O_h	3.0	Ionic	0.69	0.38	0.38	-0.30	-0.30	6.8×10^{-4}	6.8×10^{-4}
D_{4h}	10.0	Ionic	0.34	0.40	0.59	-0.04	-0.34	1.3×10^{-3}	4.9×10^{-4}
D_{4h}	3.0	Ionic	0.77	0.28	0.63	-0.20	-0.50	8.2×10^{-4}	4.3×10^{-4}
O_h	10.0	Covalent	0.06	0.68	0.68	-0.07	-0.07	3.5×10^{-4}	3.5×10^{-4}
O_h	3.0	Covalent	0.29	0.54	0.54	-0.12	-0.12	8.1×10^{-4}	8.1×10^{-4}
D_{4h}	10.0	Covalent	0.77	0.31	0.92	0.02	-0.25	9.2×10^{-4}	1.7×10^{-4}
D_{4h}	3.0	Covalent	0.53	0.43	0.78	-0.03	-0.35	1.1×10^{-3}	3.9×10^{-4}

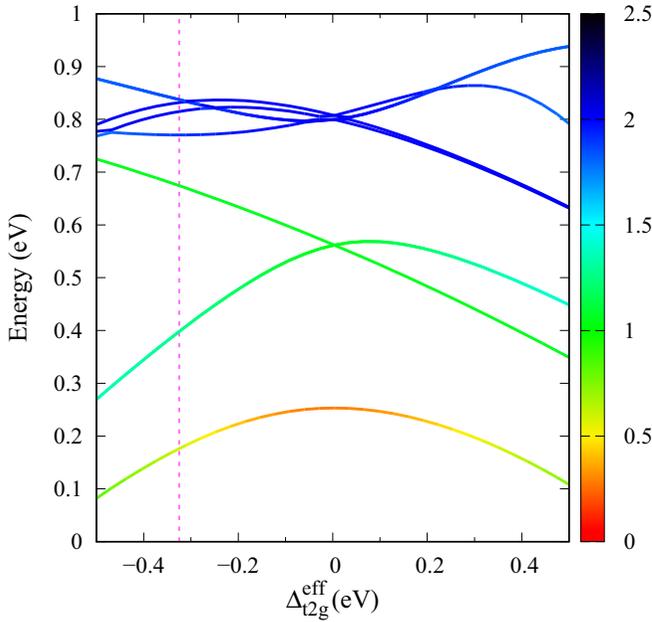


FIG. 2. Energy level diagram of the Ir^{5+} (d^4) ion as a function of the effective tetragonal crystal field in a D_{4h} local symmetry and $10Dq^{\text{eff}} = 3.0$ eV. The vertical magenta line indicates the $\Delta_{t2g}^{\text{eff}}$ of $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$ as obtained by the simulation of the XAS and XMCD spectra. The color lines represent the evolution of the expectation value of J_{eff} versus $\Delta_{t2g}^{\text{eff}}$. The value of J_{eff} ranges from 0 (red) to 2.5 (black) as indicated by the palette on the right.

to 0.38, and also the magnetic susceptibility decreases to 6.8×10^{-4} emu/mole/Oe, i.e., numbers that deviate strongly from those of the pure $J_{\text{eff}} = 0$ state.

The influence of the noncubic crystal field on the $J_{\text{eff}} = 0$ state is also listed in Table II. The calculations were performed in D_{4h} symmetry. Already in the ionic and large $10Dq$ limit we can observe that the tetragonal crystal field as it is present in $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$ causes $J_{\text{eff}} = 0.34$ together with a strong anisotropy in the magnetic properties: $L_x/2S_x = 0.40$ vs $L_z/2S_z = 0.59$ and $\chi_x = 1.3 \times 10^{-3}$ vs $\chi_z = 4.9 \times 10^{-4}$ emu/mole/Oe. Still in the ionic limit but now reducing to the realistic $10Dq = 3$ eV value, the tetragonal crystal field brings the system truly far away from the $J_{\text{eff}} = 0$ situation: $J_{\text{eff}} = 0.77$ together also with strong anisotropy, i.e., $L_x/2S_x = 0.28$ vs $L_z/2S_z = 0.63$ and $\chi_x = 8.2 \times 10^{-4}$ vs $\chi_z = 4.3 \times 10^{-4}$ emu/mole/Oe.

The effect of covalency is also systematically investigated in Table II. Keeping the same effective octahedral and tetragonal crystal field splittings as in the ionic calculations, we can observe that the hybridization of the Ir $5d$ orbitals with the O $2p$ ligands has a strong effect on the values for all relevant quantum numbers of the Ir ion: J_{eff} , $L_x/2S_x$, $L_z/2S_z$, χ_x , and χ_z all deviate appreciably from the ionic case. It is difficult to find a trend here and the only message we can learn from Table II is that one has to calculate it explicitly for each case of interest.

Given the fact that the $J_{\text{eff}} = 0$ state is no longer valid in the presence of finite octahedral crystal field splitting, strong tetragonal crystal field interaction, as well as covalency, we now study to what extent the Ir^{5+} ion can still be described

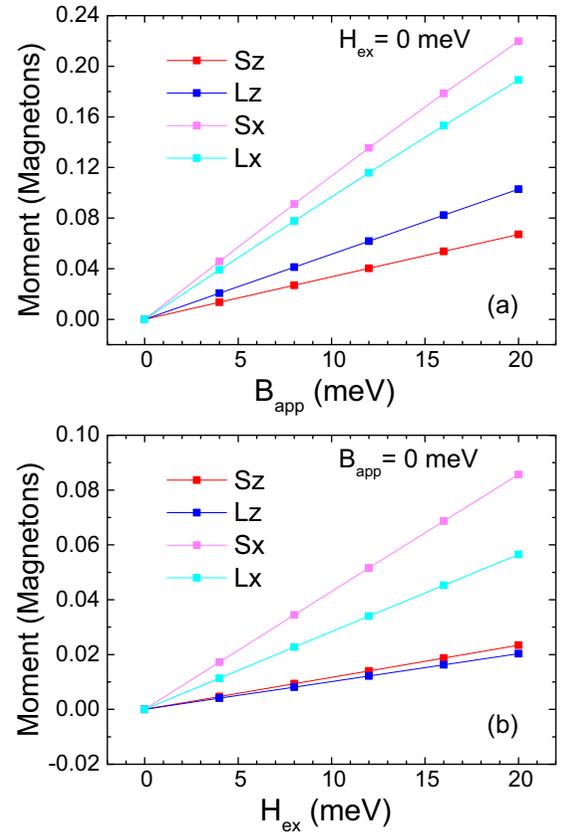


FIG. 3. Calculated spin and orbital moments as a function of the applied field (a) or exchange field (b).

as a singlet Van Vleck paramagnetic ion. In Fig. 2 we show the energy level diagram of the Ir^{5+} ion in the IrO_6 cluster as a function of the effective tetragonal crystal field using otherwise the parameters we found for $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$. The tetragonal distortion causes a splitting of the triplet $J_{\text{eff}} = 1$ excited state. However, the singlet ground state is still well below the first excited state $J_{\text{eff}} = 1$ even for very large values of $\Delta_{t2g}^{\text{eff}}$. So for $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$ where we found that $\Delta_{t2g}^{\text{eff}}$ is of the same order as the spin-orbit coupling, the excitation gap (235 meV) between the singlet and the triplet states is only slightly reduced from the value expected in a cubic symmetry ($3/4 \text{ SOC} = 300$ meV). All magnetic moments are thus of the Van Vleck type and induced by magnetic fields and exchange interactions. The colors of the curves in Fig. 2 indicate the value of J_{eff} as described in the above sections.

In our model we did not consider the effect of the hopping with the nearest neighbor metal ions, which could make the $J_{\text{eff}} = 1$ and the $J_{\text{eff}} = 2$ excited states dispersive in momentum and, hence, effectively reduce the singlet-triplet gap for certain momenta. However, the fact we could nicely simulate the XAS and the XMCD spectral shape indicates that these excited states remained fully gapped and not directly populated. It should be mentioned that a recent resonant inelastic x-ray scattering experiment determined the dispersion of the triplet and quintet states in the double perovskites $(\text{Ba}, \text{Sr})_2\text{YIrO}_6$ to be less than 50 meV [19].

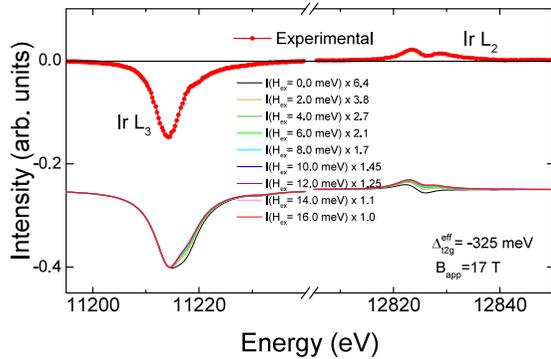


FIG. 4. Ir- $L_{2,3}$ XMCD simulations (solid lines) calculated using different values of H_{ex} , together with the experimental XMCD spectrum of $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$ (red circles). For a better comparison of the line shapes, the simulated spectra are normalized to the XMCD peak height, with the normalization factor also indicated in the legend.

Next we calculate the magnetic properties of the Ir^{5+} ion versus magnetic field. As shown in Fig. 3, the calculated spin and orbital moments (and the XMCD signal) are zero for both the applied magnetic field and the exchange field equal to zero, and increase linearly as a function of the applied field or exchange field. This is consistent for a Van Vleck system. We have also calculated the susceptibility as a function of temperature. In the calculations we considered the thermal population of the Ir^{5+} energy levels using a Boltzmann distribution. The susceptibility was found to be temperature independent with a value of $\chi = 8.3 \times 10^{-4}$ emu/mole/Oe, which is the isotropic average of the values listed in Table II. This calculated value is in nice agreement with the value of the Van Vleck susceptibility measured by SQUID in Ba_2YIrO_6 ($\chi_{VV} = 7.51 \times 10^{-4}$ emu/mol/Oe) [16], Sr_2YIrO_6 ($\chi_{VV} = 6.6 \times 10^{-4}$ emu/mol/Oe) [17], and NaIrO_3 ($\chi_{VV} = 19 \times 10^{-4}$ emu/mol/Oe) [41]. Hence from the above results we conclude that, despite the $J_{\text{eff}} = 0$ state being quite perturbed, the Ir^{5+} ions in $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$ are still describable as Van Vleck ions.

Finally, we would like to discuss the exchange field in our simulations. We introduced in the Hamiltonian an exchange field parallel to the magnetic field, since the calculated XMCD signal in an applied field of 17 T is six times smaller than the measured one. An exchange field of about 16 meV is needed in order to reproduce the size of the experimental XMCD spectrum. This is illustrated in Fig. 4, where we show the XMCD spectra calculated for 17 T magnetic field plus the presence of varying strengths of the exchange field. It is also important to note that the introduction of a strong exchange field is necessary to obtain a good fit to the experimental XMCD spectrum: for zero exchange field the line shape of the XMCD spectrum cannot be properly simulated, both at the L_3 and L_2 edges, no matter the value of the tetragonal distortion.

To unveil the origin of the 16 meV exchange field in our 17 T XMCD experiment, we measured the magnetization and magnetic susceptibility of our $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$ sample using pulsed magnetic fields up to 58 T at 2 K. The results are

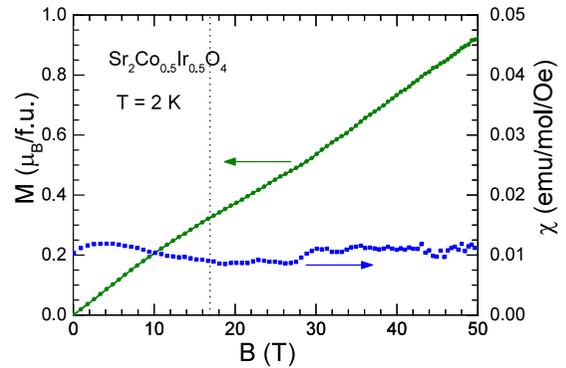


FIG. 5. Magnetization of $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$ as a function of magnetic field, together with the derived magnetic susceptibility.

displayed in Fig. 5. We can see a practically regular linear increase of the magnetization with field, yielding about $0.3\mu_B$ per formula unit (f.u.) at 17 T. The magnetic susceptibility varies around 0.01 emu/mole/Oe over the entire magnetic field range. This value is an order of magnitude larger than the magnetic susceptibility of the Ir^{5+} ion, which was calculated to be 8.3×10^{-4} emu/mole/Oe on the basis of our XMCD analysis, with similar values from the Ba_2YIrO_6 [16], Sr_2YIrO_6 [17], and NaIrO_3 [41] compounds. This in turn implies that the large magnetic susceptibility of the $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$ compound is caused by mostly the high-spin Co^{3+} ions, and that the $0.3\mu_B$ per f.u. magnetization at 17 T is associated with the canting of these antiferromagnetically ordered Co ions.

With 0.5 Co per f.u. we thus find that the 17 T magnetic field induces a canted magnetic moment of about $0.6\mu_B$ per Co. The presence of 16 meV exchange field at 17 T felt by the Ir ions can then be attributed to the presence of canted moments at the Co sites. With each Ir ion coordinated by four nearest-neighbor Co ions, we then have an exchange field of 4 meV per Co neighbor having $0.6\mu_B$ moment, i.e., about 7 meV per neighbor $\times \mu_B$. This seems to be not so unreasonable when considering, for example, the case of NiO, where an exchange field is found of 19 meV per Ni neighbor with $2\mu_B$ [42,43], i.e., about 9.5 meV per neighbor $\times \mu_B$. The considerations we just have made should of course be made self-consistently. With the 16 meV exchange field, we calculate that the Ir ions acquire about $0.16\mu_B$ magnetic moment, i.e., $0.08\mu_B$ per f.u. This leaves $0.3\mu_B - 0.08\mu_B = 0.22\mu_B$ moment for the Co ions per f.u., or $0.44\mu_B$ per Co ion, i.e., about 9 meV per neighbor $\times \mu_B$. The exchange field strength per neighbor per μ_B is then quite similar to the NiO case. However, the agreement is likely to be fortuitous considering the fact that we have not evaluated the energetics of the virtual excitations involved in these superexchange type of interactions. Nevertheless, the numbers are not unreasonable and may serve as a first order estimate.

IV. CONCLUSIONS

We have investigated the local magnetism of the Ir^{5+} ions in the layered hybrid solid state oxide $\text{Sr}_2\text{Co}_{0.5}\text{Ir}_{0.5}\text{O}_4$ by employing a combined experimental and theoretical x-ray magnetic circular dichroism (XMCD) spectroscopy study at the Ir- $L_{2,3}$ edges. From simulations of the experimental

XMCD spectrum we found that the orbital-to-spin moment ratio is significantly reduced compared to the value expected for a pure $J_{\text{eff}} = 0$ ground state. We show that the combination of atomic multiplet interactions, large tetragonal crystal field, and high covalency brings the system away from the ideal $J_{\text{eff}} = 0$ scenario. Nevertheless, our calculations show also that the excitation gap between the singlet ground state and the triplet excited state is still very large and that the Ir^{5+} ions exhibit magnetic properties, as a function of both temperature and applied field, which are typical for Van Vleck ions.

ACKNOWLEDGMENTS

We gratefully acknowledge the ESRF staff for providing beamtime. The research in Dresden was partially supported by the Deutsche Forschungsgemeinschaft (DFG) through SFB 1143. K.C. benefited from support of the DFG via Project SE 1441/1-2. We acknowledge the support of the Hochfeld-Magnetlabor Dresden (HLD) at the Helmholtz-Zentrum Dresden-Rossendorf (HZDR), member of the European Magnetic Field Laboratory (EMFL).

- [1] B. J. Kim, H. Jin, S. J. Moon, J.-Y. Kim, B.-G. Park, C. S. Leem, J. Yu, T. W. Noh, C. Kim, S.-J. Oh *et al.*, *Phys. Rev. Lett.* **101**, 076402 (2008).
- [2] G. Jackeli and G. Khaliullin, *Phys. Rev. Lett.* **102**, 017205 (2009).
- [3] F. Wang and T. Senthil, *Phys. Rev. Lett.* **106**, 136402 (2011).
- [4] J. Matsuno, Y. Okimoto, Z. Fang, X. Z. Yu, Y. Matsui, N. Nagaosa, M. Kawasaki, and Y. Tokura, *Phys. Rev. Lett.* **93**, 167202 (2004).
- [5] A. J. Gatimu, R. Berthelot, S. Muir, A. W. Sleight, and M. Subramanian, *J. Solid State Chem.* **190**, 257 (2012).
- [6] X. Ou and H. Wu, *Phys. Rev. B* **89**, 035138 (2014).
- [7] D. Mikhailova, Z. Hu, C.-Y. Kuo, S. Oswald, K. M. Mogare, S. Agrestini, J.-F. Lee, C.-W. Pao, S.-A. Chen, J.-M. Lee *et al.*, *Eur. J. Inorg. Chem.* **2017**, 587 (2017).
- [8] S. Agrestini, C.-Y. Kuo, D. Mikhailova, K. Chen, P. Ohresser, T. W. Pi, H. Guo, A. C. Komarek, A. Tanaka, Z. Hu *et al.*, *Phys. Rev. B* **95**, 245131 (2017).
- [9] G. Khaliullin, *Phys. Rev. Lett.* **111**, 197201 (2013).
- [10] O. N. Meetei, W. S. Cole, M. Randeria, and N. Trivedi, *Phys. Rev. B* **91**, 054412 (2015).
- [11] J. Chaloupka and G. Khaliullin, *Phys. Rev. Lett.* **116**, 017203 (2016).
- [12] K. Pajskr, P. Novák, V. Pokorný, J. Kolorenč, R. Arita, and J. Kuneš, *Phys. Rev. B* **93**, 035129 (2016).
- [13] C. Svoboda, M. Randeria, and N. Trivedi, *Phys. Rev. B* **95**, 014409 (2017).
- [14] G. Cao, T. F. Qi, L. Li, J. Terzic, S. J. Yuan, L. E. DeLong, G. Murthy, and R. K. Kaul, *Phys. Rev. Lett.* **112**, 056402 (2014).
- [15] S. Bhowal, S. Baidya, I. Dasgupta, and T. Saha-Dasgupta, *Phys. Rev. B* **92**, 121113 (2015).
- [16] T. Dey, A. Maljuk, D. V. Efremov, O. Kataeva, S. Gass, C. G. F. Blum, F. Steckel, D. Gruner, T. Ritschel, A. U. B. Wolter *et al.*, *Phys. Rev. B* **93**, 014434 (2016).
- [17] L. T. Corredor, G. Aslan-Cansever, M. Sturza, K. Manna, A. Maljuk, S. Gass, T. Dey, A. U. B. Wolter, O. Kataeva, A. Zimmermann *et al.*, *Phys. Rev. B* **95**, 064418 (2017).
- [18] J. Terzic, H. Zheng, F. Ye, H. D. Zhao, P. Schlottmann, L. E. De Long, S. J. Yuan, and G. Cao, *Phys. Rev. B* **96**, 064436 (2017).
- [19] M. Kusch, V. M. Katukuri, N. A. Bogdanov, B. Büchner, T. Dey, D. V. Efremov, J. E. Hamann-Borrero, B. H. Kim, M. Krisch, A. Maljuk *et al.*, *Phys. Rev. B* **97**, 064421 (2018).
- [20] A. Rogalev and F. Wilhelm, *Phys. Met. Metall.* **116**, 1285 (2015).
- [21] B. Henke, E. Gullikson, and J. Davis, *At. Data Nucl. Data Tables* **54**, 181 (1993).
- [22] K. Baroudi, C. Yim, H. Wu, Q. Huang, J. H. Roudebush, E. Vavilova, H.-J. Grafe, V. Kataev, B. Buechner, H. Ji *et al.*, *J. Solid State Chem.* **210**, 195 (2014).
- [23] T. Takayama, A. Matsumoto, G. Jackeli, and H. Takagi, *Phys. Rev. B* **94**, 224420 (2016).
- [24] D. Haskel, G. Fabbris, M. Zhernenkov, P. P. Kong, C. Q. Jin, G. Cao, and M. van Veenendaal, *Phys. Rev. Lett.* **109**, 027204 (2012).
- [25] C. T. Chen and F. Sette, *Phys. Scr.* **1990**, 119 (1990).
- [26] C. Mitra, Z. Hu, P. Raychaudhuri, S. Wirth, S. I. Csiszar, H. H. Hsieh, H.-J. Lin, C. T. Chen, and L. H. Tjeng, *Phys. Rev. B* **67**, 092404 (2003).
- [27] T. Burnus, Z. Hu, M. W. Haverkort, J. C. Cezar, D. Flahaut, V. Hardy, A. Maignan, N. B. Brookes, A. Tanaka, H. H. Hsieh *et al.*, *Phys. Rev. B* **74**, 245111 (2006).
- [28] T. Burnus, Z. Hu, H. H. Hsieh, V. L. J. Joly, P. A. Joy, M. W. Haverkort, H. Wu, A. Tanaka, H.-J. Lin, C. T. Chen *et al.*, *Phys. Rev. B* **77**, 125124 (2008).
- [29] J.-H. Choy, D.-K. Kim, S.-H. Hwang, G. Demazeau, and D.-Y. Jung, *J. Am. Chem. Soc.* **117**, 8557 (1995).
- [30] M. A. Laguna-Marco, P. Kayser, J. A. Alonso, M. J. Martínez-Lope, M. van Veenendaal, Y. Choi, and D. Haskel, *Phys. Rev. B* **91**, 214433 (2015).
- [31] B. T. Thole, P. Carra, F. Sette, and G. van der Laan, *Phys. Rev. Lett.* **68**, 1943 (1992).
- [32] P. Carra, B. T. Thole, M. Altarelli, and X. Wang, *Phys. Rev. Lett.* **70**, 694 (1993).
- [33] M. W. Haverkort, M. Zwierzycki, and O. K. Andersen, *Phys. Rev. B* **85**, 165113 (2012).
- [34] Y. Lu, M. Höppner, O. Gunnarsson, and M. W. Haverkort, *Phys. Rev. B* **90**, 085102 (2014).
- [35] M. W. Haverkort, G. Sangiovanni, P. Hansmann, A. Toschi, Y. Lu, and S. Macke, *Europhys. Lett.* **108**, 57004 (2014).
- [36] K. Koepnik and H. Eschrig, *Phys. Rev. B* **59**, 1743 (1999).
- [37] IrO₆ cluster parameters [eV]: $U_{dd} = 1.0$, $U_{pd} = 2.0$, ionic crystal field $10D_q = 1.6$, $\Delta t_{2g} = -0.27$, $\Delta e_g = -0.35$, charge transfer energy $\Delta_{CT} = -1.5$, hybridization $V(b_{1g}) = 5.30$, $V(a_{1g}) = 4.93$, $V(b_{2g}) = 2.70$, $V(e_g) = 2.64$, ligand-crystal field $10Dq^{\text{lig}} = 0.43$, $\Delta_{eg}^{\text{lig}} = -1.10$, $\Delta_{t2g}^{\text{lig}} = -0.54$, spin-orbit coupling $\zeta_{5d} = 0.4$, exchange field $H_{ex} = 0.016$ and magnetic field 17 T. The Slater integrals were reduced to 70% of Hartree-Fock values. The effective splittings of the $5d$ orbitals were

- estimated to be $10Dq^{\text{eff}} = 3.0$, $\Delta t_{2g}^{\text{eff}} = -0.325$, and $\Delta e_g^{\text{eff}} = -0.265$.
- [38] Taking into account that the sample is polycrystalline, the isotropic value of the $L_z/(2S_z + 7T_z)$ ratio was calculated as $(L_z + 2L_x)/[2(S_z + 2S_x) + 7(T_z + 2T_x)]$.
- [39] S. Agrestini, C.-Y. Kuo, M. Moretti Sala, Z. Hu, D. Kasinathan, K.-T. Ko, P. Glatzel, M. Rossi, J.-D. Cafun, K. O. Kvashnina *et al.*, *Phys. Rev. B* **95**, 205123 (2017).
- [40] N. A. Bogdanov, V. M. Katukuri, J. Romhányi, V. Yushankhai, V. Kataev, B. Büchner, J. van den Brink, and L. Hozoi, *Nat. Commun.* **6**, 7306 (2015).
- [41] M. Bremholm, S. Dutton, P. Stephens, and R. Cava, *J. Solid State Chem.* **184**, 601 (2011).
- [42] R. E. Dietz, G. I. Parisot, and A. E. Meixner, *Phys. Rev. B* **4**, 2302 (1971).
- [43] M. T. Hutchings and E. J. Samuelsen, *Phys. Rev. B* **6**, 3447 (1972).