

Ultrathin magnetite in Fe₃O₄/MgO superlattices: Investigating the enhanced thin film magnetic moment

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The electrical, crystallographic, and magnetic properties of ultrathin magnetite (Fe₃O₄) have been studied in detail, by employing superlattice structures of Fe₃O₄/MgFe₂O₄ and Fe₃O₄/MgO on a variety of substrates. By careful analysis of their properties, the influence of substrate stoichiometry, Fe₃O₄ thin film thickness, antiphase boundaries on the magnetic properties can be separated. In particular, the controversial enhanced magnetic moment in ultrathin films (<5 nm) was confirmed to be related to the substrate stoichiometry, specifically the migration of oxygen vacancies into the Fe₃O₄ thin films. The multilayer concept can be employed with many other such systems and offers methods of tuning the properties of thin magnetic oxides.

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I. INTRODUCTION

Transition metal oxides are interesting materials to study the complex interaction between electron spin, charge, and orbital order. These interactions are at the heart of novel multifunctional oxides used in spin valves and other devices reliant on electron spin interaction [1]. One such oxide is magnetite (Fe₃O₄), a ferromagnet at room temperature which undergoes a metal insulator transition (Verwey transition) around 120 K [2,3]. In recent years Fe₃O₄ attracted renewed attention not just from a fundamental point of view, as the mechanism behind the Verwey transition is better understood [3–9], but also for the observed changes in the intrinsic physical properties of ultrathin films once the thickness is reduced to well below 10–15 nm [10–16]. For spintronic applications, particularly if the oxide is used as a spin filter, the oxide thickness will be only a few nm [17]. Hence any alterations of the magnetic structure in such ultrathin films is of major interest. Most noteworthy, for Fe₃O₄ an enhanced magnetic moment has been frequently observed in layers thinner than 5 nm [11]. There have been many attempts to explain the enhancement, ranging from interface magnetic moments [10,11,18] to measurement artifacts caused by a residual substrate magnetization [19]. A combined experimental and theoretical study of oxidized thin metallic iron films grown on GaAs suggested that oxygen vacancies play a major role [16].

We have developed an experimental approach to investigate the origin of the enhanced moment in ultrathin Fe₃O₄ grown on oxide substrates by employing a set of stacked multilayer samples as illustrated in Fig. 1. Our primary aim was to improve the signal to noise (SNR) in magnetic measurements for ultrathin films, by increasing the number of layers. This concept has been used previously to enhance the SNR of magnetic measurements and to exclude the influence of interface effects on the superparamagnetic behavior of ultrathin

Fe₃O₄ films by the use of 2 nm MgO spacers [20]. We have adapted the method to investigate the size effects by systematic variation of magnetic properties with individual Fe₃O₄ film thickness, as well as altering the material of the spacer layers. The size confinement was achieved employing not only MgO but also MgFe₂O₄ as spacer layers, and the validity of this approach was confirmed by x-ray reflection (XRR), Raman spectroscopy, and transmission electron micrographs (TEM) confirming the sample structure.

Using multilayers it is possible to independently alter certain sample properties such as the density of antiphase boundaries by replacing the MgO spacers with MgFe₂O₄. As growth conditions and individual substrate quality can also influence the properties of Fe₃O₄, thicker reference layers (B samples) were grown on the same substrate by partially shadowing the samples during the growth of the spacer layers. Our approach allowed for a direct comparison of samples having the same total Fe₃O₄ thickness (d_{Total}), with one part (A) consisting of three, separated layers with an individual Fe₃O₄ layer thickness ($d_{\text{Fe}_3\text{O}_4} = d_{\text{Total}}/3$) and uninterrupted part (B) with $d_{\text{Fe}_3\text{O}_4} = d_{\text{Total}}$ (see Fig. 1).

We will demonstrate that the substrate stoichiometry, in particular the density of oxygen vacancies, is the main cause of the frequently observed enhanced magnetic moment in ultrathin Fe₃O₄ grown on oxides. We will also demonstrate that by employing the correct substrate treatment, with optimized growth conditions to avoid over or under oxidization, combined with *in situ* capping to prevent surface oxidization during exposure [21,22], layers as thin as 5 nm still show Verwey transitions. The previously reported [23] disappearance of the transition below 20 nm is hence not an inherent property of Fe₃O₄ thin films but rather due to changes in the sample stoichiometry induced by surface oxidization, interface reduction, or nonideal growth conditions.

II. EXPERIMENTAL DETAILS

Ultrathin epitaxial Fe₃O₄ films with and without spacer layers of MgFe₂O₄ were grown using an oxygen-plasma assisted MBE system (DCA MBE M600, Finland) with a

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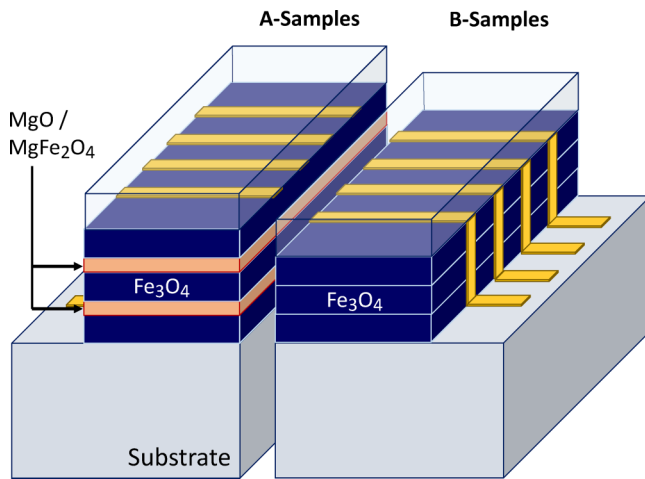


FIG. 1. Schematic representation of the sample sets under investigation. Several multilayers were grown on MgO(001) and SrTiO₃(001) substrates. The thickness of the individual Fe₃O₄ layers was varied from 2–6 nm, and the spacer layers were ≈ 2 nm thick MgO or MgFe₂O₄. Half of each sample (part B) was shadowed during the growth of the spacer layers, leading to continuous Fe₃O₄ films of 3–18 nm, grown simultaneously with the multilayer samples. All samples have been capped by MgO to prevent surface oxidation and cut *ex situ* to independently measure parts A and B. Gold contacts were deposited prior to capping in order to facilitate electrical characterization.

base pressure of 2×10^{-10} Torr. The films were grown on (100) oriented MgO (1 × 1 cm) single-crystalline substrates. The substrates were cleaned by sonication in acetone and isopropanol and then annealed *in situ* at 600 °C for thirty minutes in ultrahigh vacuum (UHV) followed by a further two hours in 1.3×10^{-5} Torr oxygen partial pressure. To grow the magnetite thin film layers, an Fe flux was produced by e-beam evaporating from metallic iron (99.99%), which was directed onto the substrate with a very low growth rate (0.08 Å/s), while the films were simultaneously oxidized by molecular oxygen in the chamber. The substrate temperature was kept at 250 °C. In contrast to most previous reports on plasma assisted MBE grown Fe₃O₄ the oxygen reactive atmosphere was solely molecular oxygen. The crystalline quality of thicker layers was assessed by XRD, and XRR thickness measurements were used to calibrate growth rates prior growth of the multilayer structures.

Subsequent to each Fe₃O₄ layer, MgFe₂O₄ or MgO layers were grown on top with nominal 2 nm thickness. In order to grow the MgFe₂O₄ interlayer, co-deposition was utilized. Material was e-beam evaporated from pure metallic Fe and ceramic MgO in separate sources onto the substrates. Corresponding samples without the interlayers, but with the same total amount of Fe₃O₄, were also prepared simultaneously by partially shadowing the samples for comparison (B samples). Gold contacts were patterned *in situ* using a shadow mask. Finally, an MgO capping layer of approximately 20 nm was grown at a rate of 0.1 Å/s on top of the gold contacts to avoid unwanted oxidation (see Fig. 1). Post growth samples have been cut by a diamond saw to separate the A and B part for further measurements.

To directly image the sample structures, TEM measurements of selected sample cross sections have been performed. The samples were first coated with 5 nm of gold. They were then loaded into a Carl Zeiss Auriga dual beam FIB/SEM equipped with a micromanipulator needle and gas injection system. A 100 nm platinum strap was deposited on the surface by e-beam followed by a further 1 μ m with I-beam in order to protect the region during milling. A cross section of the surface was extracted and transferred to a TEM grid. The ion beam was then used to thin the lamella to approximately 60 nm. An FEI Titan operating at 300 kV was used to image the sample in both bright field TEM and HAADF-STEM modes.

The overall sample thickness of capped single layers as well as all multilayers were also analyzed by x-ray reflection using a Bruker D8 Advance with monochromized Cu-K α x rays. These measurements were used to independently measure total film thickness, as well as confirm the sharpness of the superlattice structures themselves. Details of the XRR fits and a table of all sample geometries analyzed are found in the Supplemental Material [24].

Raman measurements were performed with a Renishaw inVia Raman microscope. Polarization dependent measurements were performed in backscattering geometry with $z(x_{100}, x_{100})\bar{z}$ and $z(x_{100}, y_{010})\bar{z}$ configuration. To improve the elastically scattered background as well as any contributions from the E_g modes of the substrates, the difference between those measurements has been analyzed. This way only contributions with A_{1g} symmetry are considered. Both a 488 nm Ar laser and a 532 nm solid state laser have an incident power of 10 mW and integration time of 10 min for samples on MgO substrates and 40 min for samples on SrTiO₃. The latter measurements required a numerical subtraction of the intense substrate signal prior to further analysis. The detailed procedure is outlined in the Supplemental Material [24]. The Raman line shape was analyzed by least square fitting of a Voigt profile. The total full width at half maximum and peak position were analyzed.

The resistivity of each film was measured as a function of temperature (300 K to 75 K) in a linear four point probe geometry using the gold contact patches deposited *in situ* prior to MgO capping. The magnetic hysteresis loops of the films were measured using a Quantum Design physical properties measurements system (PPMS). The PPMS is equipped with a vibrating sample magnetometer (VSM) with a sensitivity of 5×10^{-7} emu which was used to measure the saturation magnetization.

III. RESULTS

A. Sample structure

Selected multilayers were analyzed by TEM to confirm the nominal sample geometry and assess potential problems of, e.g., Mg interdiffusion. Figure 2 shows TEM images of two multilayer samples with three Fe₃O₄ layers (nominal thickness 4 nm), spaced by MgO and MgFe₂O₄ (nominal thickness 2 nm), respectively. In the sample with MgO spacers, the individual layers can be clearly distinguished by the difference in the lattice structure of MgO and Fe₃O₄. For the sample with MgFe₂O₄ spacers, only indirect confirmation on the layering can be seen, as the bright field image is dominated by the

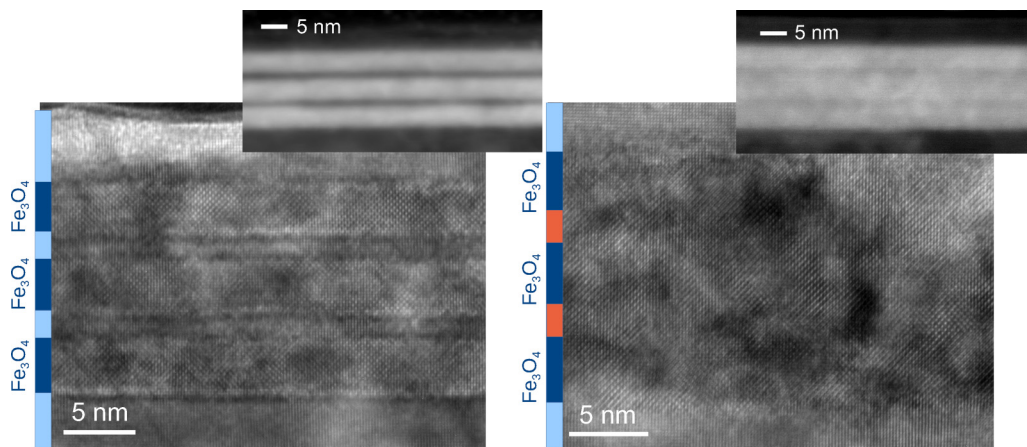


FIG. 2. TEM micrographs of multilayered samples. The left panel shows a sample with 4 nm Fe₃O₄ layers using 2 nm MgO spacers; the right panel shows a sample with 1.5 nm MgFe₂O₄ spacers. The insets are taken in HAADF mode where the Z contrast between the Fe₃O₄ and MgO is apparent. The Fe₃O₄/MgFe₂O₄ sample shows less contrast as there is still significant amounts of iron in the spacer. Likewise the different crystal structure of the MgO spacer is clearly seen, while the similarity between the Fe₃O₄ and MgFe₂O₄ crystal structures makes it difficult to observe the layering by TEM.

positions of the oxygen atoms. As the latter do not differ in position or density between MgFe₂O₄ and Fe₃O₄, there is no contrast between spacer and Fe₃O₄ layers.

In high angle annular dark field (HAADF) scanning TEM (STEM) images (Fig. 2 insets), which is more sensitive to Z contrast, the spacer layers can be observed as weak darker bands, due to the reduced Fe content of the MgFe₂O₄ spacer. In addition the total sample thickness between the MgO substrate and MgO capping layer is consistent with the nominal sample geometry as indicated schematically in Fig. 2.

Due to the identical symmetry of the MgFe₂O₄ and Fe₃O₄ oxygen sublattice we could not confirm sharp interfaces in multilayers with MgFe₂O₄ spacers. To confirm the presence of defined MgFe₂O₄ layers, as well as for independent thickness confirmation all samples have been analyzed by XRR, and measured data have been fitted to extract total and individual layer thickness. Figure 3 shows the result for the sample analyzed by TEM for the multilayered part and the single layer reference part A. The data show a clear difference between the single and multilayer part and the data are consistent with the fits assuming a perfect periodic multilayer with sharp interfaces. Thickness values extracted from the XRR measurements have been used to calculate the Fe₃O₄ resistivity and magnetization subsequently. A full description of the fitting procedure and complete list of samples and their nominal and measured geometry is found in the Supplemental Material [24].

Typically the measured thickness was within 10% of the nominal thickness. Deviations between nominal and XRR-fitted thickness for parts A and B have also been used to estimate the magnitude of the thickness error, dominating the error of the magnetization measurements for ultrathin films. For samples where TEM images were available, no deviation from the expected thickness and the one seen in TEM were found.

For ultrathin films XRD measurements are not suitable to analyze the crystalline quality. Alternatively Raman spectroscopy was used here to assess the Fe₃O₄ crystalline quality

and via analysis of size effects in the Raman shift also to indirectly confirm sharp interfaces in samples with MgFe₂O₄ interlayers.

Of particular focus was the A_{1g} mode of Fe₃O₄ around 668 cm⁻¹. The mode position, width, and shape is known to be sensitive to the crystalline quality [25,26], strain [27,28], number of APBs [29,30], as well as the oxidization state [26,31]. In addition, for ultrathin layers we expect a size dependence similar to what is seen for Fe₃O₄ nanoparticles [32]. Figure 4(a) shows the set of spectra from 4 and 12 nm single layers and a layered sample of 3 × 4 nm Fe₃O₄ with 1.5 nm

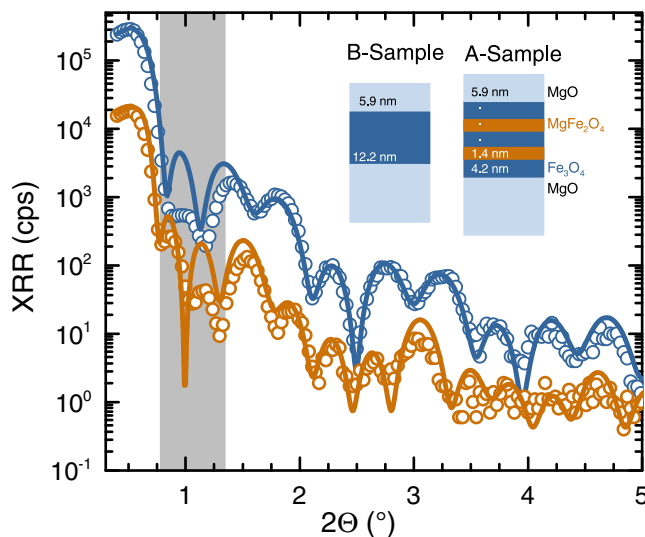


FIG. 3. The raw data of the XRR measurements and the least square fits for a sample with and without MgFe₂O₄ spacers. The XRR measurements differ due to the different total film thickness and difference in internal structure. The data for the multilayered sample have been divided by 10 for better visualization. The area marked in gray is dominated by instrumental, systematic errors from a nonideal primary x-ray beam and residual Au contacts.

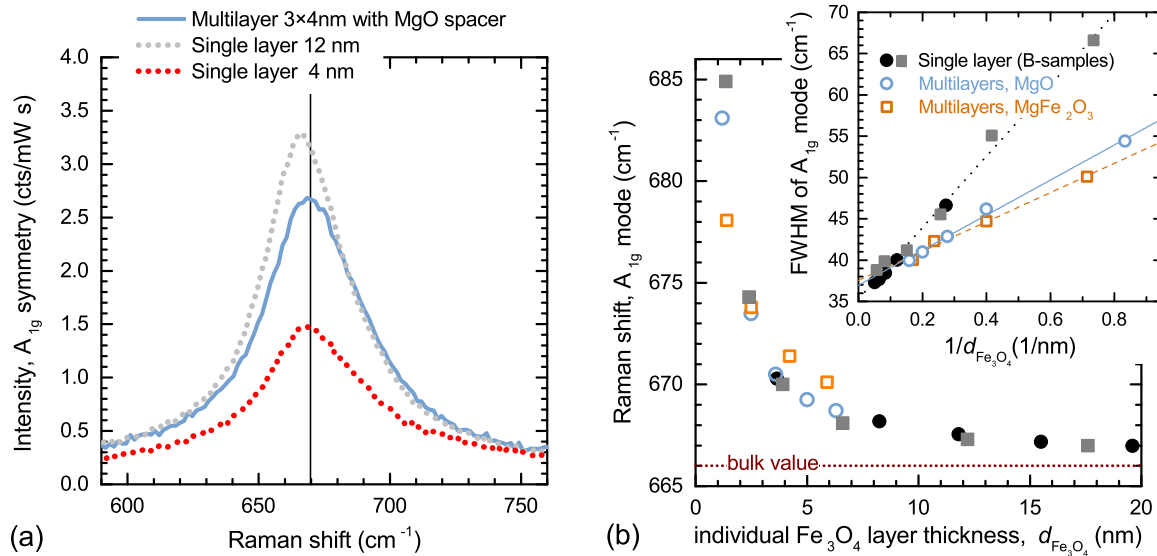


FIG. 4. The Raman spectra (a) reveal distinct size related shifts in the Fe_3O_4 A_{1g} mode. The vertical line indicates the center of the peak in the multilayer sample which is aligned with that of the single 4 nm layer sample. The continuous 12 nm thick B sample shows a Raman shift closer to the bulk value. (b) shows the fitted peak position of all samples as a function of the individual Fe_3O_4 layer thickness. The inset shows the width of the mode versus reciprocal individual layer thickness. The peak width of an oxygen breathing mode should strongly depend on the oxygen stoichiometry linking the distinctly different behavior for single and multilayered samples (closed and open symbols, respectively) to an altered stoichiometry. APBs have also been linked to an increased FWHM, and their reduction can explain the difference between the samples with MgFe_2O_4 and MgO spacers (open squares and circles, respectively).

MgFe_2O_4 spacer layers. The measured Raman intensity scales with the scattering volume, as the penetration depth of the Raman laser is well above the actual film thickness. The peak position for the single 4 nm layer is shifted to higher wave numbers compared to the 12 nm thick sample. The peak position for the multilayer sample is found to be identical to that of the 4 nm sample, confirming that the Fe_3O_4 layers are well separated individual layers. Figure 4(b) shows the peak position for the complete sample set as a function of individual layer thickness ($d_{\text{Fe}_3\text{O}_4}$), illustrating that the latter determines the A_{1g} peak position in our high quality Fe_3O_4 samples. Strain effects on the energetic position of the Raman mode can be excluded as all films are equally strained, fully confined to the substrate lattice. Equally no asymmetry, which would indicate over-oxidation and consequent $\gamma - \text{Fe}_2\text{O}_3$ formation [21], is seen in the Raman modes of the capped samples.

While the peak position is mostly affected by the individual layer thickness, independent of number of layers or type of spacer layer, the peak width is found to be significantly different between single and multilayer samples. The A_{1g} mode was found to be more symmetric and sharper for either thicker layers, or samples employing the spacer layers, in particular if MgFe_2O_4 spacer layers are used. The Raman data hence not only confirm the expected layering in the samples but also highlight that by employing a multilayer with similar individual Fe_3O_4 thickness, the signal to noise is significantly improved in the measurement when compared to a single layer due to the increased volume of the sample. They also confirm that by replacing the MgO spacers with MgFe_2O_4 spacers, the film quality further improves while maintaining the integrity of the layer confinement, thus such samples can be used to investigate the enhancement of the magnetic moment as we can now separate size effects from other changes such as the

number of antiphase boundaries (APBs), as well as improving the SNR by tripling the amount of material in the measurement.

We have to stress that all thin films, multi- or single layers will be epitaxially fully strained with an in-plane lattice constant of $2 \times a_{\text{MgO}}$. The lattice mismatch between MgFe_2O_4 and Fe_3O_4 or MgO is even lower than that of Fe_3O_4 and MgO . Hence even films up to 100 nm will be fully strained [33]. Hence the A_{1g} peak position itself is initially offset from bulk values due to the strain within the film. For $d_{\text{Fe}_3\text{O}_4} < 5$ nm an additional blueshift is observed. The origin of the latter cannot be attributed to a single cause, as multiple effects can result in such a shift. One possibility would be Mg diffusion. The corresponding A_{1g} mode of MgFe_2O_4 is found at higher frequencies (707 cm^{-1}) [34]. To explain the magnitude of the shift by formation of $\text{Mg}_x\text{Fe}_{1-x}\text{O}_4$, we would need Mg concentrations x of 0.08 to 0.19 for the 6 nm and 3 nm samples, respectively, assuming a linear dependence on mode position and x . The obtained values for x are inconsistent with the resistance and magnetoresistance data from the same films, as a significant reduction in conductivity and increase in magnetoresistance is expected for Mg doped Fe_3O_4 [35]. In addition the position is found to equally shift to higher wave numbers for multilayers with MgO and MgFe_2O_4 spacers and for single layers on SrTiO_3 and GaAs [27]. In the latter two cases there is no source of Mg. It is hence more likely that the observed peak shift is directly related to the size reduction. For Fe_3O_4 hollow spheres, a blueshift in the A_{1g} mode has been observed with decreasing diameter and wall thickness [32]. A direct comparison with smaller Fe_3O_4 nanoparticles is difficult due to the large differences in the stoichiometry between chemically synthesized nanoparticles and high quality epitaxial thin films. The effect of dimensional confinement on Raman spectra is well studied for silicon, where in porous silicon the size

can be independently controlled by the crystallinity, but also for many oxide nanoparticles [36]. In the case of Fe₃O₄ in particular, no systematic study of the Raman size effect has been performed yet. While first order Raman spectroscopy typically only probes the Brillouin zone center (Γ point), the size reduction in the probed sample can lead to a non-negligible momentum transfer of the incident photon, resulting in an effective averaging of phonon modes with nonzero momentum in the Raman measurement [36]. Depending on the phonon dispersion of the mode under investigation, this can either lead to a blueshift (minimum at the Γ point) or redshift (maximum at the Γ point). The phonon dispersion of Fe₃O₄ has been previously investigated, but particular focus was on the low energy phonon modes, believed to be of importance for the Verwey transition [9,37,38]. Calculated phonon dispersions including the higher energy modes show both modes with maxima and minima in the energy region of interest [39]. Unfortunately they have not been assigned to individual modes at the zone center to verify our explanation. In case the observed shift with decreasing film thickness is caused by such a Raman size effect, rather than changes in the stoichiometry or crystalline order of the thin film an increase in line width is also expected, consistent with our measurements. The inset in Fig. 4 includes linear fits of the observed line width with reciprocal layer thickness. For all cases the expected linear behavior is observed. The Raman size effect is significantly less pronounced than for nanoparticles, as our films are only confined in one dimension. Compared to, e.g., CeO₂ nanoparticles [36] where the corresponding slope of the 466 cm⁻¹ mode was found to be 125 cm⁻¹nm, we observe 43, 21, 18 cm⁻¹nm for Fe₃O₄ single layers, multilayer with MgO and MgFe₂O₄ spacer, respectively. We have to highlight already a stark difference in the slope between multilayer and single layer samples, which will be discussed at a later stage.

Even with the evidence for the Raman size effect, other possibilities need to be addressed: The charge and crystallographic reordering at the Verwey transition is known to cause a blueshift of the A_{1g} mode itself [40]. It is possible that the Fe₃O₄ thin film does not undergo the Verwey transition but remains in the low conducting charge ordered phase. However, the Raman line width is dramatically increased in our case and the Verwey transition temperature is typically reduced, the thinner the film. Finally the shifts are most prominent for the thinnest films. It is therefore also possible that we observe evidence of quantum confinement affecting the electronic properties, which due to strong electron-phonon coupling in Fe₃O₄ alters the vibrational states.

In summary, we have to stress that whatever the cause of the changed energetic position of the A_{1g} mode, it is maintained in the multilayered samples and can be used to confirm that the size confinement in the multilayers is fundamentally the same as in single layers. This is of particular use in the samples with MgFe₂O₄ spacers, where TEM images only show poor contrast between the two materials. The Raman analysis provides evidence in that case that there is little intermixing and the Fe₃O₄ layers are well separated from the MgFe₂O₄ layers. Secondly the Raman line width for samples as thin as 6 nm in the case of multilayers and 12 nm in the case of single layers is already as low as typically found in bulk films

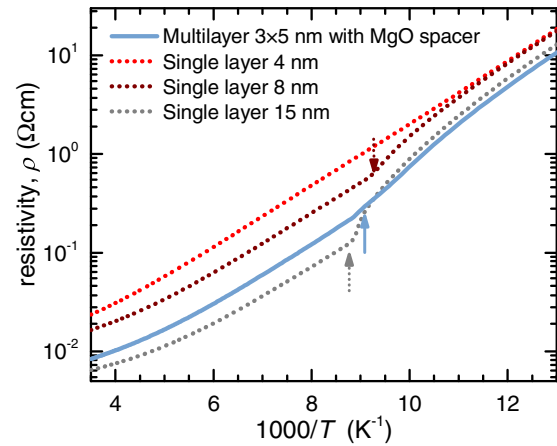


FIG. 5. Verwey transition in thin films: Resistivity of B samples with thickness of 15, 8, and 4 nm (dotted lines). For the 15 nm and the 8 nm samples, the Verwey transition is clearly observed as indicated by the arrows. In comparison an A sample (3×5 nm Fe₃O₄ with 2 nm MgO, solid line) with 5 nm individual Fe₃O₄ thickness is also shown. The improved layer quality in this case already leads to a lower resistivity than seen in a 8 nm single layer, with the Verwey transition still occurring. However there is a size induced reduction in the resistivity and shift in the Verwey transition compared to the corresponding B sample ($d_{\text{Fe}_3\text{O}_4} = 15$ nm). All films have been grown on MgO(001).

(35–40 cm⁻¹). This is direct evidence of a very high crystalline quality in all our Fe₃O₄ films.

B. Resistance and magnetoresistance

Electrical measurements (Fig. 5) on the B samples show very good conductivity (6 mΩcm at 300 K) and most importantly a sharp Verwey transition at 113.2 ± 0.5 K for 15-nm-thick layers. Even in 8-nm-thick samples a Verwey transition at 107.7 ± 0.5 K can be observed, confirming the very high quality of the Fe₃O₄ ultrathin layers, once they are protected from further *ex situ* oxidation in air. In A samples with MgFe₂O₄ and MgO spacer layers the resistivity increases if compared to their corresponding continuous B samples. However if one compares the resistivity of multilayered samples with continuous layers of similar individual layer thickness, the resistivity in the multilayered samples is significantly reduced. One such example is shown in Fig. 5. Remarkably, even in samples with 5 nm individual layer thickness the Verwey transition can be observed. These results significantly lower the previously reported Fe₃O₄ minimum thickness at which the transition is observed and give a first indication that the multilayered ultrathin films have a better crystalline quality than single layers of the same thickness.

In epitaxial thin films the presence of antiphase boundaries (APBs) is known to affect the resistivity of the films but more importantly their magnetoresistance [15,41–43]. The formation of the APBs is of particular importance in samples grown on MgO(001), where single height steps can increase their number significantly [42,44–46]. To investigate how the formation of APBs is altered in our multilayers we performed magnetoresistance measurements exemplary shown in Fig. 6. The sample with MgFe₂O₄ spacer layers shows reduced

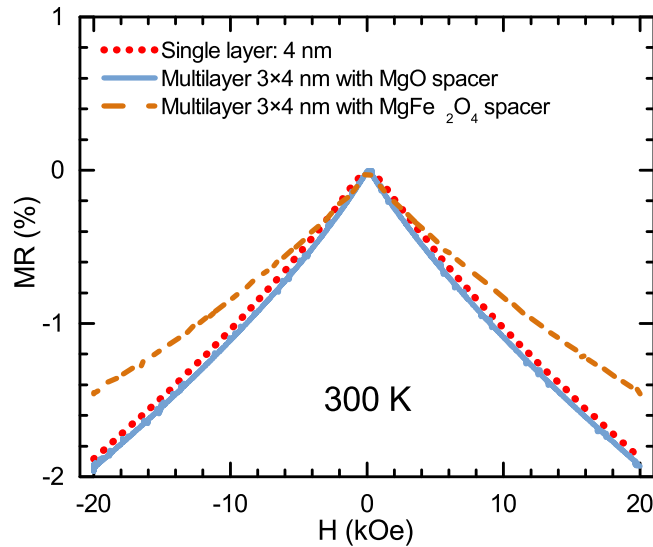


FIG. 6. Magnetoresistance at 300 K of a set of samples with 4-nm-thick Fe_3O_4 layers, either as single layer (red, dotted line), as three layers spaced with 2 nm MgO (blue, solid line) or with 1.5 nm MgFe_2O_4 (orange, dashed line). All films have been grown on $\text{MgO}(001)$.

MR compared to either one with MgO spacers or a single 4-nm-thick layer. It was previously confirmed that as the number of APBs decreases, the thicker the Fe_3O_4 layer [15,41]. By introducing a spacer layer of similar lattice geometry to Fe_3O_4 itself, e.g., spinel MgFe_2O_4 or MgAl_2O_4 , the number of APBs in the second and third Fe_3O_4 layer are effectively reduced. The multilayer concept therefore allows the decoupling of size effects from the density of APBs for samples grown on the same substrate. Indeed the measured MR in the $\text{Fe}_3\text{O}_4/\text{MgFe}_2\text{O}_4$ multilayer on an MgO substrate is already lower than that reported for 10 nm Fe_3O_4 on MgAl_2O_4 symmetry matched substrates, where APBs are only formed due to misfit dislocations due to the higher lattice mismatch [47–49]. The MR data already suggests that in the case of MgO spacer layers, the natural roughness of the MgO spacer prevents the reduction in APBs with thickness and might even lead to an increase in number. While our measurements cannot provide absolute values for the APB densities we can estimate the relative change in densities. Assuming the MR is solely caused by APBs and proportional to the APB density, the 3×4 nm sample using 1.5 nm MgFe_2O_4 spacers already has a 30% lower APB density compared to the corresponding single layer of sample with MgO spacers. As only the second and third Fe_3O_4 layer will be actually different from a single layer, this implies the APB density can be reduced by the same amount in a single Fe_3O_4 layer if a 5 nm MgFe_2O_4 buffer layer would be used.

C. Magnetic moment

After confirming that the multilayer structure preserves the size confinement in Fe_3O_4 layers, and that by employing MgFe_2O_4 spacer layers the density of APBs can be reduced compared to single layers of similar individual Fe_3O_4 layer thickness, we are now able to investigate if the Fe_3O_4 thin films

do show an enhanced magnetic moment. Due to the increased total volume of thin film by employing multilayers, systematic errors in thin film magnetic measurements (due to, e.g., small contributions from para- or ferromagnetic impurities in the substrate) are significantly reduced. Equally the re-optimized growth conditions produced high quality thin films, where the crystalline structure is not only stoichiometric, confirmed by the Raman line width close to those of bulk samples, but also the Fe_3O_4 characteristic Verwey transition is observed even for ~ 5 nm thin films. Figure 7 shows the results magnetic moment measurements of the multilayered and single layer Fe_3O_4 films. While in the single layers we observe the increased magnetic moment for the thinnest sample, the finding is not confirmed in the multilayers which have better SNR. On the contrary, the magnetic moment is reduced for both samples with a high density of APBs (MgO spacer) and those with better magnetic domain ordering (MgFe_2O_4 spacer).

IV. DISCUSSION

Our results so far confirm that the increased magnetic moment does not originate from an inherent size effect as it is not observed for the multilayer samples. Due to the very high crystalline quality of the samples, as confirmed by the presence of the Verwey transition and the closeness of the A_{1g} Raman mode peak width to bulk values, we can also exclude that structural defects affected the measurements. This also excludes the possibility that interface states have the effect of contributing to the enhanced moment, as the number of interfaces in our multilayers is increased. Any interface effect would consequently also scale in proportion to the number of layers.

This leaves only the direct influence of the substrates on the measurements. It has been previously suggested that Fe or other magnetic impurities affect the results [19]. Despite our best efforts we could not detect any magnetic signal from plain substrates at room temperature, where the increased moment was also seen in the single layer samples. We therefore suggest another mechanism, supported by our results. One of our key findings is that the single Fe_3O_4 films show an increased line width in the A_{1g} Raman mode compared to its multilayered counterparts as evident by a much steeper slope in the FWHM vs inverse film thickness plots (see Fig. 4). Both size effects and the number of APBs can affect the Raman line width, but both the size confinement and number of APBs are expected to be similar for the single Fe_3O_4 on MgO layer and the triple layer with MgO spacers. The contributions of APBs to the FWHM is much smaller as evident by the small difference in the FWHM between MgO and MgFe_2O_4 spacers, even though their APB density significantly differs as seen in MR data (see Fig. 6). The significantly broader mode of the single layer, and larger slope in the width vs reciprocal layer thickness analysis, must hence originate from increased stoichiometric disorder. The A_{1g} mode around 670 cm^{-1} only involves movement of oxygen atoms [37,39,50], suggesting that a disordered oxygen sublattice is affecting it the most.

Over-oxidation of Fe_3O_4 , e.g., at the surface is known to lead to the formation of $\gamma - \text{Fe}_2\text{O}_3$ [21,22,31]. As maghemite has a smaller magnetic moment than Fe_3O_4 an oxygen surplus would lead to a reduction in the magnetic moment.

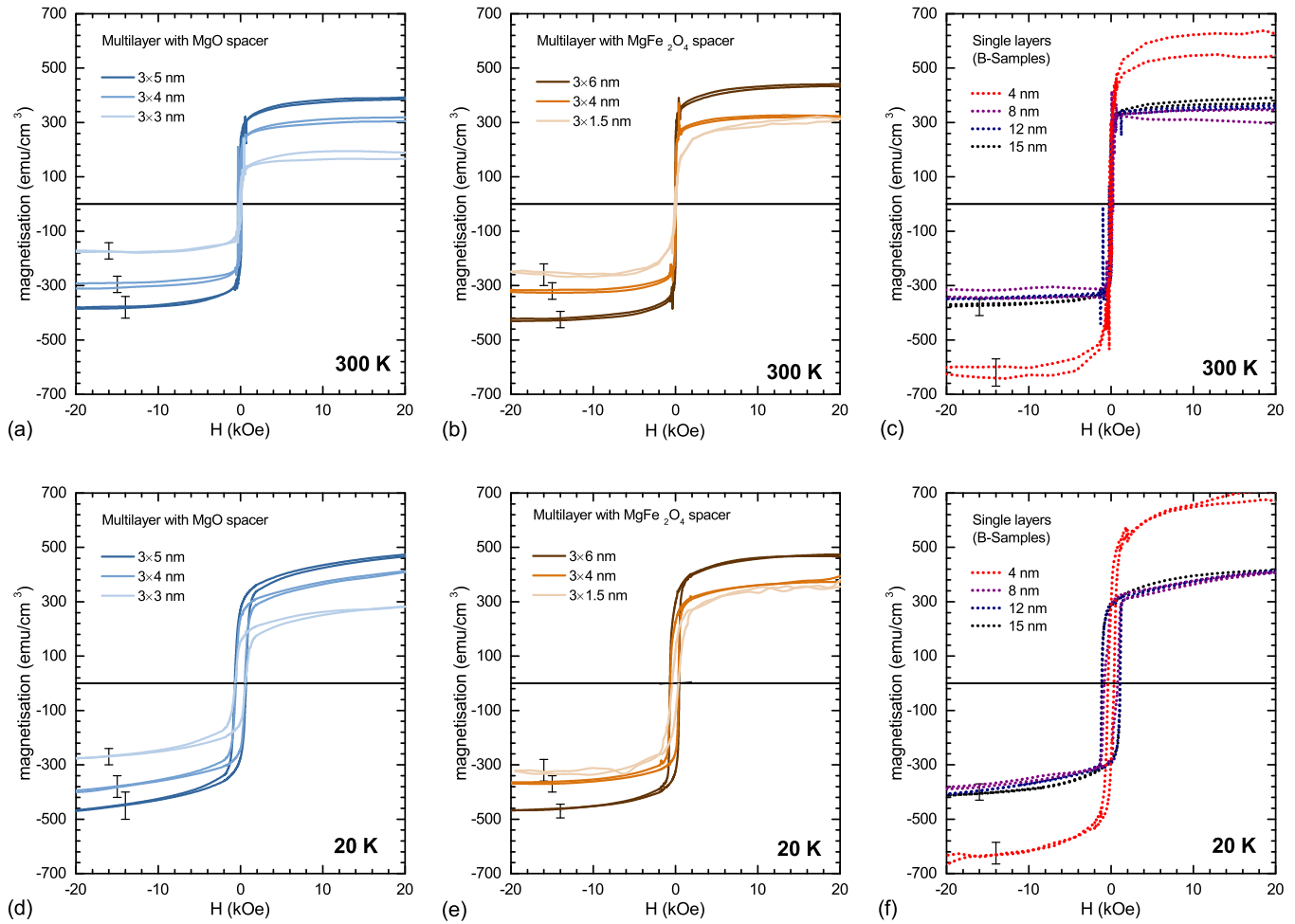


FIG. 7. Measurements of the magnetic moments of multi- and single layered samples at 300 K (top row) and 20 K (bottom row). An increased magnetic moment was only observed for single layered films with $d < 5$ nm.

Conversely an under-oxidization could lead to an increase in the magnetic moment and could be the root cause of the observations. The most likely point defect to lead to an effective under-oxidization are oxygen vacancies. During the formation of the Fe_3O_4 layers, oxygen can migrate into the substrate, effectively reducing the first layers of the Fe_3O_4 films—providing an under stoichiometric substrate surface. In our multilayers this mechanism will only affect the first layer but not the subsequent layers separated by stoichiometric spacer layers of MgO or MgFe_2O_4 . This mechanism can also explain why the effect is not seen universally as substrate quality and preparation conditions between groups varies. In particular the substrate cleaning steps are typically performed by *in situ* vacuum annealing, which is known to be capable to increase the density of oxygen vacancies.

In many transition metal oxides the creation of oxygen vacancies upon vacuum annealing is well studied [51]. However, few studies explicitly discuss this for MgO , where the effect is more localized to the MgO surface layers [52,53]. We therefore prepared a set of test Fe_3O_4 layers on $\text{SrTiO}_3(001)$ substrates, so that the number of bulk oxygen vacancies could be more readily controlled [54,55]. Figure 8 compares the magnetic moment of a 6-nm-thin single Fe_3O_4 layer on two different SrTiO_3 substrates: one was vacuum annealed

at 600°C and the other at 800°C . While the one on the low temperature annealed SrTiO_3 substrates shows a similar magnetic moment to the films grown on MgO shown above, the sample grown on the high temperature annealed SrTiO_3 substrate has a significantly enhanced moment. The high temperature annealing of the SrTiO_3 substrates creates a significant quantity of bulk oxygen vacancies [56], clearly verified by discoloration of the substrate (see Supplemental Material S5). The presence of these vacancies, in particular at the surface region, leads to the already suspected reduction of the Fe_3O_4 thin film, thus increasing the magnetic moment. The annealing temperature of 600°C and 800°C was carefully chosen, in order to avoid changes in surface morphology and step density seen for higher temperature annealing of STO [57,58], while at the same time altering the concentration of oxygen vacancies.

A line shape analysis of the A_{1g} mode for these films is more complicated on STO grown films due to a strong substrate background signal. The latter was numerically removed as detailed in the Supplemental Material [24]. Figure 8 shows the mode and a least squares fit for each sample. While the position of the mode is quite similar ($670.1 \pm 0.5 \text{ cm}^{-1}$ and $668.4 \pm 0.7 \text{ cm}^{-1}$) the mode width significantly differs between the sample grown on the 600°C or 800°C annealed

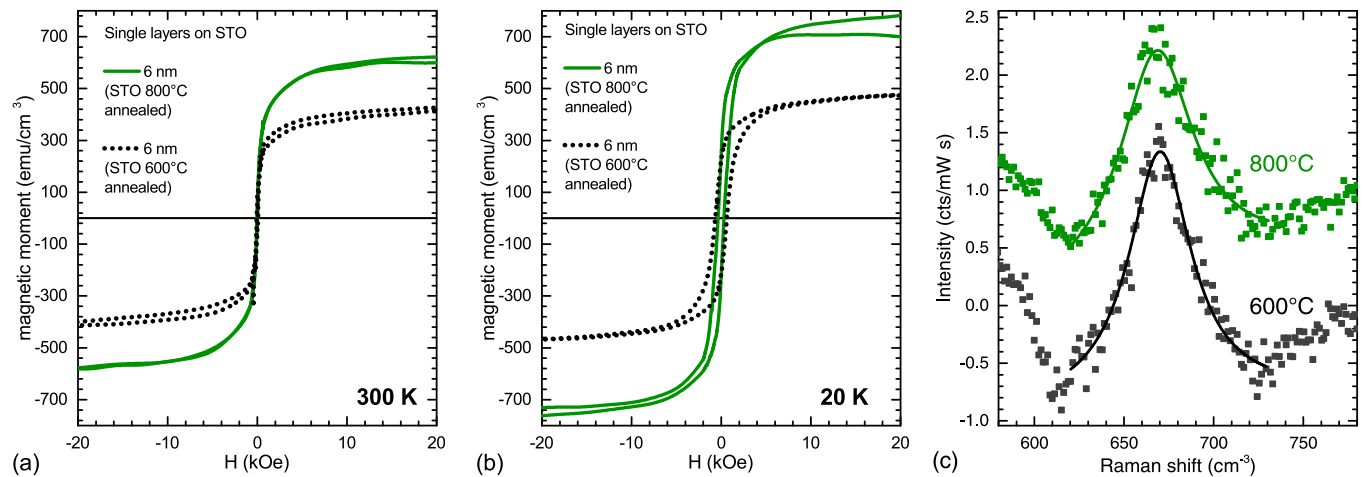


FIG. 8. Magnetic moment for 6-nm-thick single layer Fe₃O₄ samples on SrTiO₃ measured at 300 K (a) and 20 K (b). The measurements for films grown on two different SrTiO₃ substrates are shown, where one was vacuum annealed at 600 °C and one at 800 °C. (c) shows the A_{1g} Raman mode of the magnetite thin films for the same two samples. The one grown on the substrate annealed at 800 °C shows a broader A_{1g} mode associated with a more off-stoichiometric Fe₃O₄ lattice.

substrate ($42.3 \pm 3 \text{ cm}^{-1}$ and $50 \pm 4 \text{ cm}^{-1}$, respectively). This is fully consistent with our model of an altered Fe/O stoichiometry in films directly grown on the substrate with more oxygen vacancies. The A_{1g} mode is very sensitive to the oxygen stoichiometry and disorder, as the probed vibration is a breathing mode of the oxygen octahedra [37]. Over oxidization in the form of $\gamma\text{-Fe}_2\text{O}_3$ formation could lead equally to a broadening of the mode but is accompanied with a blueshift to higher wave numbers and stark asymmetry in the peak [21]. In contrast we observe a small redshift to lower wave number in addition to the significant broadening, further suggesting a reduction of the film. Interestingly on similarly strained 5 nm Fe₃O₄ thin film with a low number of APBs grown on GaN, no enhanced magnetic moment has been observed [59]. This illustrates that in nonoxide substrates, where by definition oxygen vacancies cannot migrate from the substrate, reduction of the Fe₃O₄ thin film can also be prevented.

Our findings suggest, that for ultrathin Fe₃O₄ films the substrate stoichiometry can thus substantially alter the magnetic properties of the films as oxygen can migrate into the substrate bulk, leaving a Fe₃O₄ film with an increased number of oxygen vacancies. It is evident that the enhanced moment is only seen in films well below 8 nm and only in single layers. While the moment for films in this study is lower than in a previous study on MgO(100) [11], the magnitude of the increased moment of a single layer of 4 nm compared to a 8 nm film is confirmed. The wide spread of data points in the inset of Fig. 9 illustrates that the moment of thin film Fe₃O₄ does depend on many parameters such as crystalline quality and APB density. Our data clearly indicates that the frequently observed increased moment is not inherent to a size effect in high quality Fe₃O₄ but rather linked to an altered oxygen stoichiometry induced by the substrate. The enhancement is therefore not seen for films grown in multilayers, where the substrate is decoupled from the thin films by the stoichiometric spacer layers. It is equally absent if the substrate itself does not possess a large number of oxygen vacancies, such as seen in films grown on Al₂O₃ [14].

From a phenomenological point of view the specific magnetic moment of several bulk magnetic iron oxides and iron can

be compared (see Fig. 9), showing that as a general trend the moment increases with increased iron content. However the magnetization of off-stoichiometric films will highly depend on the exact nature of the oxygen vacancy or possible defect complexes. One of the few theoretical studies on the impact of oxygen vacancies in Fe₃O₄ found no significant change in the overall magnetization, however only isolated single oxygen vacancies (V_O) have been considered [61]. A more recent combined experimental and theoretical study on Fe₃O₄ grown on GaAs(100) concluded that oxygen vacancies do affect the ferrimagnetic coupling between iron on tetrahedral and octahedral sites, explaining the net increase in the magnetic

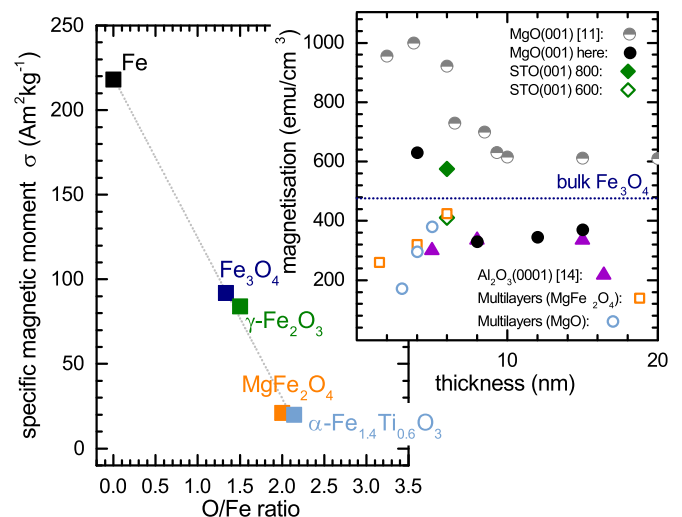


FIG. 9. Specific magnetic moment of bulk materials as a function of oxygen/iron ratio for selected iron oxides in a spinel or closely related lattice. Values for iron are included for comparison. The graph illustrates a general trend in bulk materials of an increased moment in lattices with reduced oxygen content (data taken from J. M. D. Coey [60]). The inset summarizes the dependency of the magnetic moment of Fe₃O₄ thin films investigated here and compared to previous literature data for the growth on MgO and Al₂O₃ surfaces.

moment [16]. Likewise in thicker Fe₃O₄ films grown by PLD in various oxygen partial pressure conditions, the films grown at the lower p_{O_2} have shown higher magnetization. That study was performed on Si(001), where strain affects the magnetic properties and the phase purity of the films was much worse than the MBE grown films investigated here [62]. In related oxides, namely spinel ferrites such as NiFe₂O₄ grown on STO, oxygen vacancies have been previously discussed as a root cause for the enhanced magnetic moment, but the effects of the cation disorder of the two inequivalent cations Ni and Fe were concluded to be the more dominant contribution [63]. In binary Fe₃O₄ the latter would be equivalent to an increase in Fe⁺³-A, Fe⁺³-B, and Fe⁺²-B site disorder. Site disorder is more likely in a defective lattice with a high number of oxygen vacancies. Alternatively, there could be a phase separation into stoichiometric Fe₃O₄ in the film and metallic iron or FeO at the substrate/film interface. The possibility of an interface FeO layer has been previously discussed in films grown on MgO(001) [64]. However, a paramagnetic FeO layer would not result in an enhanced magnetic moment in such a thin film. We therefore suggest that the oxygen poor interface layer is better described as a Fe₃O₄ layer with a high number of oxygen vacancies, rather than a formation of the paramagnetic wüstite phase. This is consistent with the data of Bertram *et al.* [64] as their assessment is based on XRR and XPS data, which are sensitive to the density of the interface layer and charge states of Fe but not necessarily to the crystal structure.

V. CONCLUSION

By analyzing the electric, magnetic, and crystallographic properties of ultrathin Fe₃O₄ films grown on MgO(001), in the form of Fe₃O₄/MgO and Fe₃O₄/MgFe₂O₄ superlattices, we were able to demonstrate that the frequently observed increased magnetic moment is not an inherent property of Fe₃O₄ but related to changes in the Fe₃O₄ stoichiometry on oxide substrates with a high number of oxygen vacancies. Thus we can confirm that the oxygen vacancy mechanism suggested by Huang *et al.* [16] is also responsible for the increase in ultrathin Fe₃O₄ grown on oxide substrates. However the root cause for the oxygen vacancy formation in Fe₃O₄ is a high density of oxygen vacancies in the surface region of the substrate, rather than deposition conditions. Optimizing the latter for thick films does not lead to high quality thin films due to these substrate interactions. Indeed our Fe₃O₄ ultrathin films grown at a reduced growth rate from metallic Fe, and

using only molecular oxygen as the oxidizing agent, are of high crystalline quality, which was indicated by the presence of the Verwey transition even in films as thin as 6 nm and by the sharp A_{1g} Raman modes.

In superlattice structures the magnetic moment is reduced in thin films below 6 nm thickness, while in single layers grown directly on the substrate surface the magnetic moment can be increased. The link between the presence of oxygen vacancies in the bulk or at the surface of the substrate and an enhanced magnetic moment in thin films was demonstrated for 5 nm films grown on SrTiO₃ substrates, where the density of oxygen vacancies was increased by vacuum annealing of the substrate prior growth.

Secondly we were able to show that the use of MgFe₂O₄ spacer layers reduced the number of APBs, leading to a weaker magnetoresistance and sharper A_{1g} Raman modes. This suggests that using MgFe₂O₄ as a buffer layer between MgO and Fe₃O₄ can be employed to significantly improve the properties of even single layer Fe₃O₄, as it will not only reduce the APBs but can also limit the diffusion of oxygen vacancies by increasing the separation of the Fe₃O₄ films from potentially oxygen poor substrates. Other oxide buffer layers [65] or metallic layers have already been effective in improving thin film Fe₃O₄, but the similar lattice structure of MgFe₂O₄ and Fe₃O₄ should be better suited to reduce the number of APBs and strain at the same time as acting as an oxygen diffusion barrier.

While we only investigated Fe₃O₄ films, the proposed mechanism of magnetic moment alterations in ultrathin films by oxygen vacancy diffusion is likely to be found in other magnetic oxides. For example similarly enhanced moments have been found for spinel NiFe₂O₄ and NiCo₂O₄ [63,66,67]. Recent studies have shown that the suggested mechanism of a change in cation distribution in such thin film ternary oxides does not consistently explain the increased moment [66]. We therefore suggest that similar to the presented findings for Fe₃O₄ the *substrate stoichiometry* could play an important role in these related spinels.

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