Strain-induced electronic phase transition and strong enhancement of thermopower of TiS₂

Atanu Samanta, Tribhuwan Pandey, and Abhishek K. Singh

Materials Research Center, Indian Institute of Science, Bangalore 560012, India (Received 19 February 2014; revised manuscript received 6 June 2014; published 3 November 2014)

Using first-principles density functional theory calculations, we show a semimetal to semiconducting electronic phase transition for bulk TiS_2 by applying uniform biaxial tensile strain. This electronic phase transition is triggered by charge transfer from Ti to S, which eventually reduces the overlap between Ti-(d) and S-(p) orbitals. The electronic transport calculations show a large anisotropy in electrical conductivity and thermopower, which is due to the difference in the effective masses along the in-plane and out-of-plane directions. Strain-induced opening of band gap together with changes in dispersion of bands lead to threefold enhancement in thermopower for both p- and n-type TiS_2 . We further demonstrate that the uniform tensile strain, which enhances the thermoelectric performance, can be achieved by doping TiS_2 with larger iso-electronic elements such as Zr or Hf at Ti sites.

DOI: 10.1103/PhysRevB.90.174301

PACS number(s): 72.20.Pa, 71.15.Mb

I. INTORDUCTION

Transition-metal dichalcogenides (TMDs) are layered materials, which are well known for their unique electronic and optical properties [1,2]. While within a layer, the metals and chalcogens form strong ionic-covalent bonds, these layers are held together by weak van der Waals (vdW) interactions. Depending upon the composition, TMDs offer a wide range of functional materials such as metals [3], semimetals [4], semiconductors [5], insulators [6], and superconductors [7]. Among the TMDs, TiS₂ has been the focus of extensive research due to its potential applications as cathode materials for lithium-ion batteries [8–11]. Previous experimental studies [12] have reported that nearly stoichiometric TiS_2 shows a large power factor value of $37.1 \mu W/K^2$ cm at room temperature that is comparable with the best thermoelectric material Bi_2Te_3 [13]. The large power factor originates from a sharp increase in the density of states just above the Fermi energy as well as the intervalley scattering of charge carriers [12,14,15]. However, the semimetallic nature of TiS₂ gives rise to bipolar effects which are not desirable for thermoelectric applications [16].

The electronic structure of TiS₂ is very unique and even a slight change can significantly influence thermoelectric properties. It has been shown that 0.04% Mg doping at the Ti site in TiS₂ causes a 1.6 times increase in thermopower at 300 K [17]. Recently, it has been discovered that the electronic structures of semiconducting TMDs $[MX_2(M =$ Mo, W; X = S, Se, Te)] are very sensitive to the applied pressure/strain, which causes an electronic phase transition from semiconductor to metal [18–21]. In addition few-layer and monolayer TMDs show a wide range of tunability in electronic and magnetic properties by application of strain [22-26]. Contrary to that, TiS₂ remains semimetallic up to a compressive hydrostatic pressure of 20 GPa [27]. So far there have been no studies on the effect of uniform tensile strain on the electronic properties of TiS_2 . Here, we carry out a study of the electronic structure of TiS_2 as a function of applied uniform biaxial tensile strain (BTS). The material undergoes electronic phase transition from semimetal to a small band gap (<0.15 eV) semiconductor. Most interestingly, the semiconducting strained TiS₂ exhibits nearly a fourfold enhancement in thermopower compared to the unstrained phases. We also explore the possibility of generating such a strain by doping with larger atoms at Ti sites. Iso-electronic Hf and Zr turn out to be the best dopants to generate the 2% tensile strain producing the same enhanced thermoelectric properties as obtained by straining TiS₂.

II. COMPUTATIONAL DETAILS

Structural optimization, total energy, and electronic structure calculations were performed using first-principles density functional theory (DFT). The ionic cores are described by the all-electron projector augmented wave potentials [28,29] and the Perdew-Burke-Ernzerhof [30] generalized gradient approximation (GGA) to the electronic exchange and correlation as implemented in the Vienna ab initio simulation package (VASP) [28,29,31]. We have incorporated the vdW interaction using Grimme's DFT-D2 method [28,29,31,32] (where D2 refers to the second generation of this method) for all the calculations. In this approach, the dispersion energy is described by an atom-pairwise interaction potential $E_{\text{disp}} = -\frac{1}{2} \sum_{i \neq j} f_{\text{damp}}(R_{i,j}) \frac{C_6^{i,j}}{R_{i,j}^6}$, where $R_{i,j}$ and $C_6^{i,j}$ are the distance and isotropic dipole-dipole dispersion coefficient between a pair of atoms, respectively [32]. The accuracy of the results crucially depends on the approximations of the dispersion coefficients and on the cutoff of damping function(f_{damp}) at short distances [32-35]. The parameters used in the DFT-D2 method are taken from the empirical force field of Grimme [32]. The lattice parameters obtained using this approach are in very good agreement with experiments for various materials [35,36]. However, this approach overestimates the binding/cohesive energy for most of the cases including TMDs [19,35,36].

The unit cell was optimized using the conjugate gradient scheme until the forces on every atom were $\leq 10^{-3}$ eV/Å. For transport calculations, Boltzmann transport theory [37,38] was used, which enables calculation of the temperature and doping-level-dependent thermopower and other transport parameters from the electronic structure. All the transport properties were calculated within the constant scattering time approximation (CSTA) [37,38]. The CSTA is based on the assumption that the scattering time does not vary strongly with energy. It also does not consider any assumptions on temperature and doping

level dependence of the scattering time. Within the CSTA, the energy dependence of transport function is described through both the density of states and carrier velocity. In this theory, the motion of an electron is treated semiclassically, and its group velocity in a specific band is given by

$$\nu_{\alpha}(i,\mathbf{k}) = \frac{1}{\hbar} \frac{\partial \epsilon(i,\mathbf{k})}{\partial \mathbf{k}_{\alpha}},\tag{1}$$

where $\epsilon(i, \mathbf{k})$ and \mathbf{k}_{α} are the *i*th energy band and α th component of wave vector \mathbf{k} , respectively. From group velocity $\nu_{\alpha}(i, \mathbf{k})$ the thermopower and electrical conductivity can be obtained as

$$S_{\alpha\beta}(T,\mu) = \frac{1}{eT} \frac{\int \nu_{\alpha}(i,\mathbf{k})\nu_{\beta}(i,\mathbf{k})(\epsilon-\mu) \left[-\frac{\partial f_{\mu}(T,\epsilon)}{\partial \epsilon}\right] d\epsilon}{\int \nu_{\alpha}(i,\mathbf{k})\nu_{\beta}(i,\mathbf{k}) \left[-\frac{\partial f_{\mu}(T,\epsilon)}{\partial \epsilon}\right] d\epsilon},$$
(2)

$$\frac{\sigma_{\alpha\beta}(T,\mu)}{\tau(i,\mathbf{k})} = \frac{1}{V} \int e^2 \nu_{\alpha}(i,\mathbf{k}) \nu_{\beta}(i,\mathbf{k}) \left[-\frac{\partial f_{\mu}(T,\epsilon)}{\partial \epsilon} \right] d\epsilon, \quad (3)$$

where e, T, V, τ , and f_{μ} are the electronic charge, temperature, volume of the unit cell, relaxation time, and Fermi-Dirac distribution function, respectively. The Brillouin zone was sampled by a well-converged $35 \times 35 \times 21$ Monkhorst-Pack k mesh [39]. Subsequently, the group velocities were obtained by Fourier interpolation [40,41] of band energies on a converged denser **k** grid. These values are used in Eqs. (2) and (3), to calculate the transport properties as implemented in BOLTZ-TRAP code [42]. This approach has been shown to provide a good estimate of thermopower as a function of temperature and carrier concentration in a variety of thermoelectric materials without any adjustable parameters [43–49].

III. RESULTS AND DISCUSSION

TiS₂ crystallizes in a hexagonal structure with the space group $P\bar{3}m1$ [50,51]. The unit cell and corresponding Brillouin zone along with high symmetry points are shown in Figs. 1(a) and 1(b), respectively. The optimized lattice parameters a =b = 3.4 Å, c = 5.77 Å, and z = 0.246 are in good agreement with previously reported values [50–53]. Calculated electronic band structure and corresponding density of states (DOS) of bulk TiS_2 are shown in Fig. 1(c). The valence band maxima (VBM) and conduction band minima (CBM) at Γ point originates from the S(p) and Ti(d) orbitals, respectively. The overlap of these bands renders TiS₂ semimetallic. The extent of this indirect band overlapping is of 0.2 eV as shown in Fig. 1(c). The valence bands Γ^{-3}/Γ^{+3} and Γ^{-2} mainly originate from S- $(p_x)/S-(p_y)$ and S- (p_z) orbitals [54], respectively. Among them, the first two bands are almost dispersionless along Γ to A in comparison to the highly dispersive third band. This anomaly in dispersion comes from the interlayer interaction of $S_{-}(p_z)$ orbitals, which makes the third valence band more dispersive. The conduction bands at L (L⁺¹) and Γ points originate from Ti-(d_z) and Ti-($d_{x^2-y^2}/d_{xy}$) [54–56] orbitals, respectively. The conduction band along Γ to A is less dispersive due to weaker interaction between two different layers. On the other hand, due to stronger in-plane interaction of Ti- d_z orbitals, the band at the L point is highly dispersive. Therefore, the effective mass along Γ to A is



FIG. 1. (Color online) (a) Side view of the bulk unit cell of TiS_2 and (b) the corresponding Brillouin zone along with high symmetry points. Band structure, total, and partial density of states per formula unit (f.u.) at (c) 0%, (d) 2%, (e) 5%, and (f) 8% strain values. Fermi energy is referenced to 0 eV.

very high compared to the L symmetric points. This large difference in the effective mass makes the transport properties very anisotropic.

It has been shown that strain can affect the electronic structure significantly by modifying the extent of the hybridization [19]. As shown previously [27], the compressive hydrostatic pressure does not affect the electronic phase of TiS₂. Therefore, we studied the effect of 2%–8% uniform BTS on the electronic structure of TiS₂. The uniform BTS is applied along the *x* and *y* axis. Subsequently, each structure is relaxed at the fixed cell volume. The uniform BTS was calculated as $\epsilon = (a-a_0)/a_0$, where a_0 and *a* are the equilibrium and instantaneous lattice parameters, respectively. The corresponding band structure and DOS are shown in Figs. 1(d) and 1(f). The change in intralayer Ti-S and interlayer S-S distance with strain, is shown in Fig. 2(a). The intralayer bonding between S-(*p*)/Ti-(*d*)



FIG. 2. (Color online) (a) Change in bond length as a function of applied strain. (b) Change in band gap as a function of applied strain along different high symmetry directions. (c) and (d) Bader charge analysis as a function of strain.

orbitals is much stronger than the interaction between layers. Therefore, the S-S distance between two layers increases faster compared to the intralayer Ti-S distance with increasing strain, as shown in Fig. 2(a).

Next, we analyze the electronic band structures under the applied uniform BTS, which causes the electronic phase transition in this material. All the band gaps were calculated with respect to the VBM at different strain values, irrespective of the symmetry of the VBM. The VBM remains at Γ point for all the strain values. While the band gaps along Γ -A and $\Gamma - \Gamma$ decrease, the gap along Γ -L increases linearly with increasing strain as shown in Fig. 2(b). At strain level lower than 7%, TiS₂ has an indirect gap between the Γ and the L points. When the strain exceeds the 7%, the band gap shifts to Γ -A. With the increasing strain, the conduction band minima (CBM) at L becomes flatter due to decrease in the overlap of the S(p) and Ti(d) orbitals, caused by an increase in Ti-S bond lengths. Therefore, the bands around the Fermi energy in the conduction band become very heavy. The DOS at the Fermi level in the conduction band also become sharp, as shown in Figs. 1(c)-1(f). It is known that this type of 2D nature of DOS enhances the thermoelectric performance, significantly [47,57]. The third valance band along K-Γ-A moves up faster than the first and second valence bands, with increasing strain. This anomalous behavior is caused by the faster increase in the S-S interlayer distance than the S-Ti intralayer distance. Eventually, above 8% of applied strain the material becomes a small indirect-to-direct gap semiconductor. In order to gain further insight on the semimetal-semiconductor transition, we compute the Bader [58,59] charge with FFT grid $400 \times 400 \times 240$ and energy cutoff 1000 eV including all-electron charge (core + valence), which are plotted in Figs. 2(c) and 2(d). The Bader charge analysis shows that with the applied strain charge transfer occurs between the intralayer Ti to S atom, thereby increasing



FIG. 3. (Color online) (a) and (b) Conductivity anisotropy $(\sigma_{ii}/\sigma_{av})$ as a function of carrier concentration for *p*-(solid lines) and *n*-type (dashed lines) TiS₂ at 0 and 5% strain, respectively. (c) and (d) Anisotropy in thermopower as a function of carrier concentration for *p*-(solid lines) and *n*-type (dashed lines) TiS₂ at 0 and 5% strain. (e) Dependence of average thermopower on the carrier concentration under different strain. (f) Power factor divided by relaxation time (in units of W/m K² s) as a function of carrier concentration. All these calculations are carried out at 300 K.

the ionic bond character between Ti and S, which leads to the band gap opening.

It has been proposed that the semimetal to semiconductor transition can enhance thermoelectric performance [57,60,61]. Therefore, we study the effect of applied uniform BTS on the transport properties of TiS2 within the constant scattering time approximation (CSTA) [37,38]. The ratio of the component of the conductivity tensor σ_{ii} (where i = x, y, and z) to the average conductivity σ_{av} (($\sigma_{xx} + \sigma_{vv} + \sigma_{zz}$)/3), which is independent of the relaxation time is calculated and shown in Fig. 3(a). The conductivity is the highest along the in-plane direction (x, y) for unstrained TiS₂ at 300 K over all the carrier concentration range. Compared to in-plane directions, both conduction and valance bands along the out-of-plane (z) direction (Γ -A in reciprocal space) have larger effective mass for electron and hole, leading to remarkable anisotropy in conductivity. While the hole conductivity increases significantly, with the applied uniform BTS, the electron conductivity remains nearly the same as shown in Fig. 3(b). Interestingly, upon application of uniform BTS for p-type TiS₂, the high conductivity direction changes from in-plane to out of plane in the carrier concentration range of $(5 \times 10^{19} - 10^{21} \text{ cm}^{-3})$. At a very high carrier concentration (excess 5×10^{22} cm⁻³) in-plane conduction starts to dominate again. On the other hand for n-type TiS₂, the conduction direction remains in-plane over all the range of carrier concentration. However, the difference between in-plane and out-of-plane conductivity gets reduced compared with the unstrained case. A similar behavior of conductivity has also been observed in other layered compounds such as unstrained CrSi₂ and FeSi₂ [44,48].

The reason for this dependence of anisotropy in conductivity on the applied uniform BTS can be found in the band structure as shown in Figs. 1(c) and 1(e). At 0% strain in Fig. 1(c), the valence band is flatter along Γ -A, leading to lower hole mobility. On the other hand the conduction band has two pockets at M and L symmetry points, leading to higher mobility for electrons, which is consistent with the conductivity behavior as shown in Fig. 3(a). With the increasing strain the third valence band, which is highly dispersive moves up and finally crosses the flatter band at around 5% strain. This lowers the effective mass of the holes dramatically, leading to a drastic increase in out-of-plane conductivity, which starts to dominate over the in-plane conductivity as shown in Fig. 3(b). On the other hand, with the applied uniform BTS the dispersion of the conduction band reduces, causing a decrease in the anisotropy in conductivity, as shown in Fig. 3(b). However, for a complete understanding of electrical conductivity, one needs to estimate the effect of different scattering mechanisms on relaxation time (τ) . The carrier transport in a system is mainly hampered by scattering of charge carriers due to phonons (acoustic, optical, and polar optical) and point defects [62-64]. For nonparabolic bands the average band mass (density-of-states effective mass for each pocket) is defined as $m_b = (m_L m_T^2)^{1/3}$ where m_L and m_T are longitudinal and transverse effective masses, respectively [62,64]. According to the deformation potential theory of Bardeen [65] the relaxation time $\tau \propto m_b^{-3/2}$ [38,64–66]. The band mass of the charge carriers increases with increasing strain, which leads to the decrease in the relaxation time as well as conductivity. At low temperatures, charge carriers are scattered mostly by the point defects [62,63], which can further reduce the relaxation time. All the scattering mechanisms described above will affect the transport properties. Furthermore, for chalogenides at high temperatures, the scattering is mainly dominated by phonons [62,64]. Since our calculation does not involve explicitly the effect of these scattering mechanisms, the results present the upper limit of the transport properties. However, it is shown experimentally that irrespective of all these scattering effects, TiS₂ still shows good electrical conductivity [12,67]. Thereby, CSTA can be a good approximation for TiS₂.

In order to estimate the effect of uniform BTS on thermoelectric performance of TiS_2 , we have calculated the thermopower of the unstrained and strained TiS_2 . The direction-dependent thermopower at 300 K as a function of carrier concentration for 0 and 5% strain is shown in Figs. 3(c)and 3(d), respectively. Like conductivity, the thermopower also exhibits large anisotropy at low concentration for both *p*- and *n*-type carriers. However, at high carrier concentration the thermopower shows reduced anisotropy. This anisotropy comes from the nonparabolic nature of bands which gets modified by applying strain. The density of states at 5% strain becomes sharp near the Fermi energy at the conduction band as shown in the Fig. 1(e), which results in large thermopower for a given carrier concentration. Furthermore, the band structure of TiS₂ is composed of heavy and light mass bands near the Fermi energy that also contributes to high thermopower values. A similar phenomenon has also been observed for many other good thermoelectric materials [44,45,49].

At low carrier concentration and 0% strain, the thermopower attains high values along out of plane for both *p*-type and *n*-type cases. The thermopower increases with the applied strain attaining a maxima at 5% strain. With the strain TiS₂ shows large thermopower for both *p*-type and *n*-type doping along the out of plane and in-plane direction, respectively. This improved performance comes from opening of the band gap and changes in the band dispersion with the applied strain.

Next, we study the effect of applied uniform BTS on the average of all the components of thermopower, as it enables the comparison of our results with the polycrystalline samples. The average thermopower is defined as 1/3 of trace of the thermopower tensor ($S_{av} = (S_{xx} + S_{yy} + S_{zz})/3$). The average thermopower as a function of carrier concentration shows a peak, which shifts with the applied strain as shown in Fig. 3(e). This peak arises due to bipolar effect. For a good thermoelectric performance a material should exhibit unipolar behavior over a wide range of carrier concentration [16]. As shown in Fig. 3(e), the calculated averaged thermopower for p and n type at 300 K, shows unipolar behavior for a large range of carrier concentration compared to the unstrained case. For *p*-type TiS₂ at 5% strain highest thermopower can be achieved at moderate carrier concentrations, typically $1-5 \times 10^{20}$ cm⁻³. We report for n-type TiS₂ efficient thermoelectric performance can be achieved in the carrier concentration range of $1-7 \times 10^{20}$ cm⁻³ depending upon temperature. The above predicted doping level is equivalent to approximately 0.006-0.03 holes and 0.006–0.04 electrons per unit cell, respectively. Previous experimental study of Sams et al. [68] has demonstrated that TiS₂ can be heavily doped substitutionally at both Ti and S sites. Based on this experimental observation it can be assumed that the doping level proposed here should be readily achievable.

As discussed above due to the presence of heavy band and multicarrier transport, n type leads to higher thermopower than p type even at higher carrier concentrations. With the applied strain, the thermopower increases nearly four times in comparison to the unstrained case, for both types of the carriers. However, strain not only affects the thermopower but also the electrical conductivity (σ). In order to quantify the effect of strain on thermoelectric performance (ZT = $S^2\sigma/\kappa_{\text{total}}$, where κ_{total} is the total thermal conductivity) it is important to calculate power factor $(S^2\sigma)$. The power factor is defined as the ability of a material to produce useful electrical power at a given temperature difference. The large power factor is indicative of better thermoelectric performance. As a first approximation for power factor $(S^2\sigma)$, here the quantity $S^2\sigma/\tau$ is calculated at 300 K for *n*-type TiS_2 as shown in Fig. 3(f). This quantity also increases drastically (three times) in comparison to the unstrained case. The figure of merit ZT closely follows the overall change in the thermoelectric performance, hence can also be enhanced by three times assuming no drastic changes in thermal conductivity and relaxation time.

It is well known that LDA/GGA tends to underestimate the band gap [19,69]. In order to test the effect of band gap underestimation on thermoelectric performance, we applied the scissors operator and adjusted the band gap to 0.30 eV for 5% strained TiS₂. The scissors operator was applied by rigidly shifting the conduction bands up and the valence bands down by 0.075 eV around the midgap energy. After the rigid



FIG. 4. (Color online) (a) The average thermopower as a function of the carrier concentration for p- (solid lines) and n-type (dashed lines) at 300 K. (b) Total density of state of Hf (x = 0.25) and Zr (x = 0.25) doped TiS₂, respectively. For comparison purposes the DOS of bulk TiS₂ is also shown in the same figure.

shift of the bands we performed the transport calculations. The maximum value of $S^2\sigma/\tau$ is 8×10^{11} W/m K² s, which is nearly twice than the value at the band gap 0.15 eV. Therefore, the GGA results serve as a lower limit for the thermoelectric performance.

Having shown the advantages of applied strain on the electronic and thermoelectric properties of TiS₂, next we look at the feasibility of generating such kinds of strain in experiments. In order to achieve uniform BTS intrinsically in bulk TiS₂, the material can be grown epitaxially on a substrate with a larger lattice parameter. However, the presence of the interface would dominate the electronic properties, masking the intrinsic changes in properties by BTS. Another more easier route could be doping with the iso-electronic atoms of larger size, which could induce the required BTS. We have studied doping of TiS₂ using a $2 \times 2 \times 2$ supercell with the iso-electronic Hf and Zr on the Ti site. The volume relaxation of the doped structures was carried out by including the vdW interaction between the layers. The fully relaxed structure expands due to the larger size of the Zr and Hf. This eventually generates in-plane tensile strain of the order of 2% and 1.98% corresponding to a doping of 0.25 Zr and Hf atom per formula unit, respectively. The out-of-plane strain is nearly zero for both the dopants. The calculated thermopower of iso-electronic doped systems matches very well with that of 2% tensile strained bulk TiS_2 for both *p*- and *n*-type doping, as shown in Fig. 4(a). The DOS [Fig. 4(b)] does not show any significant changes far away from Fermi energy with iso-electronic doping. However, the reduction in DOS around the Fermi energy is similar to the one caused by 2% tensile strain. For TiS_2 , better thermoelectric performance can be achieved at optimized strains (2%-5%) as well as optimized carrier concentrations (approximately, 0.006-0.03 holes and 0.006–0.04 electrons per unit cell). As shown experimentally [68], TiS₂ can be heavily doped with V, Mo, Ta, and Ce at the Ti site. Since the ionic radii of these dopants are larger than the Ti-ionic radius, these dopants can generate lattice strain and charge carriers in the system. For practical realization of strain values proposed here, one can perform co-doping by isovalent and p- or n-type dopants, having larger ionic radius than Ti.

The lattice part of thermal conductivity κ_l is an another key parameter, which influences the thermoelectric performance



FIG. 5. Phonon dispersion (a) bulk TiS_2 and (b) $Ti_{0.94}Zr_{0.06}S_2$.

of a material. Zr and Hf turn out to be the best iso-electronic dopant, which enhance the thermoelectric performance of TiS₂. In order to get a qualitative estimate of thermal conductivity of Zr doped TiS₂, we have calculated phonon dispersion of undoped TiS₂ and doped Ti_{1-x}Zr_xS₂ (x = 0.06and 0.17), which are shown in Figs. 5(a) and 5(b). The density functional perturbation theory (DFPT) in VASP is used to calculate force constants for phonon calculation. Then, the phonon dispersions are calculated using the Parlinski-Li-Kawazoe method as implemented in the PHONOPY package [70]. The phonon group velocity is calculated from the relation $v_g = \frac{\partial \omega}{\partial a}$ as shown in Table I, where q is the wave vector. According to the single mode relaxation time approximation of the Boltzmann equation, the total thermal conductivity (κ_i) contributed by each phonon mode can be represented as $\kappa_i(q) = C_i(q)v_i^2(q)\tau_i(q)$, where C_i , v_i , and τ_i are the specific heat, the group velocity, and the phonon relaxation time of the i^{th} mode, respectively. κ_l is directly proportional to the square of phonon group velocity, implying that the decrease in phonon group velocity will reduce the lattice thermal conductivity. As presented in Table I, with the increasing Zr concentration the phonon group velocity significantly decreases along in-plane and out-of-plane direction, which will lower the thermal conductivity. This reduction in phonon group velocity is due to the mass difference between Zr and Ti atoms. Furthermore, the large differences in atomic mass and size can increase the phonon scattering [71,72], which will lead to further reduction of thermal conductivity.

TABLE I. Effect of doping on transverse v_{Trans} , longitudinal v_{Long} , and average sound velocities. Sound speeds in m/s are extracted from phonon dispersion data. v_{Trans} and v_{Long} denote speed corresponding to transverse and longitudinal modes, respectively.

System	Direction	Sound velocity (m/s)		
		v_{TA1}	v_{TA2}	v_{LA}
TiS ₂	Г-М	2510.54	3693.01	5332.23
	Г-А	2000.64	2000.64	3735.94
Average		2255.59	2846.82	4534.08
$Ti_{0.94}Zr_{0.06}S_2$	Г-М	2247.00	2856.00	3587.00
	Г-А	2070.33	2070.33	3542.00
Average		2158.66	2463.16	3564.50
$Ti_{0.83}Zr_{0.17}S_2$	Г-М	1964.02	1643.40	1893.21
	Г-А	1925.02	2058.09	3362.30
Average		1944.52	1850.74	2627.75



FIG. 6. (Color online) Carrier-concentration-dependent electronic part of the thermal conductivity scaled with respect to relaxation time, respectively.

Because of the semimetallic nature of TiS₂ the electronic thermal conductivity (κ_e) could hamper the thermoelectric performance. However, under strain TiS₂ exhibits an electronic phase transition from semimetal to semiconductor which can lower the κ_e . In Fig. 6 we present the κ_e scaled with respect to τ for bulk, 2% strain, and Zr doped TiS₂. As can be seen at lower carrier concentration we observe a significant reduction in κ_e for Zr doped TiS₂ and 2% strained TiS₂ in comparison to bulk TiS₂. On the other hand, at higher carrier concentration the thermal conductivity remains more or less the same for all the cases. Based on this analysis we can safely conclude that the total thermal conductivity ($\kappa_e + \kappa_l$)

- Q. H. Wang, K. Kalantar-Zadeh, A. Kis, J. N. Coleman, and M. S. Strano, Nat. Nanotechnol. 7, 699 (2012).
- [2] J. Wilson and A. Yoffe, Adv. Phys. 18, 193 (1969).
- [3] A. Ayari, E. Cobas, O. Ogundadegbe, and M. S. Fuhrer, J. Appl. Phys. 101, 014507 (2007).
- [4] A. H. Reshak and S. Auluck, Phys. Rev. B 68, 245113 (2003).
- [5] K. K. Kam and B. A. Parkinson, J. Phys. Chem. 86, 463 (1982).
- [6] B. Sipos, A. F. Kusmartseva, A. Akrap, H. Berger, L. Forró, and E. Tutiš, Nat. Mater 7, 960 (2008).
- [7] E. Morosan, H. W. Zandbergen, B. S. Dennis, J. W. G. Bos, Y. Onose, T. Klimczuk, A. P. Ramirez, N. P. Ong, and R. J. Cava, Nat. Phys. 2, 544 (2006).
- [8] R. Friend and A. Yoffe, Adv. Phys. 36, 1 (1987).
- [9] M. Whittingham, Prog. Solid State Chem. 12, 41 (1978).
- [10] Z. Mao and R. White, J. Power Sources 43, 181 (1993).
- [11] A. H. Reshak, I. V. Kityk, and S. Auluck, J. Chem. Phys. 129, 074706 (2008).
- [12] H. Imai, Y. Shimakawa, and Y. Kubo, Phys. Rev. B 64, 241104 (2001).
- [13] S. Scherrer and H. Scherrer, Bismuth Telluride, Antimony Telluride, and Their Solid Solutions in CRC Handbook of Thermoelectrics, 1st ed. (CRC Press, London, 1995).
- [14] P. C. Klipstein, A. G. Bagnall, W. Y. Liang, E. A. Marseglia, and R. H. Friend, J. Phys. C: Solid State Phys. 14, 4067 (1981).
- [15] M. Koyano, H. Negishi, Y. Ueda, M. Sasaki, and M. Inoue, Phys. Status Solidi (b) 138, 357 (1986).

will decrease under doping. Thus, the observed enhancement in thermoelectric performance as shown in this work can be achieved experimentally by doping/co-doping of TiS_2 .

IV. CONCLUSIONS

In summary, we demonstrate in TiS_2 , the electronic phase transition from semimetallic to semiconducting by application of uniform biaxial tensile strain. The strain modifies the dispersion of bands, which improves the thermoelectric performance of the material, significantly. Our results show a large-strain-dependent anisotropy in electrical conductivity and thermopower at lower carrier concentration for both *p*and *n*-type carriers. The thermopower results suggest that 5% strained TiS₂ exhibits three times enhancement in thermoelectric properties. We further demonstrate the possibility of generating such strain by doping of TiS₂ with larger size atoms such as Zr and Hf at Ti sites. Moreover, experimental realization of strain, induced by doping, can serve as a powerful tool to enhance the electronic and thermoelectric properties of this promising material.

ACKNOWLEDGMENTS

This work was supported by the ADA under NPMASS and DST nanomission. The authors thank the Supercomputer Education and Research Centre and Materials Research Centre, IISc, for providing the required computational facilities for the above work.

- [16] C. Wood, Rep. Prog. Phys. 51, 459 (1988).
- [17] X. Y. Qin, J. Zhang, D. Li, H. Z. Dong, and L. Wang, J. Appl. Phys. **102**, 073703 (2007).
- [18] M. Dave, R. Vaidya, S. Patel, and A. Jani, Bull. Mater. Sci. 27, 213 (2004).
- [19] S. Bhattacharyya and A. K. Singh, Phys. Rev. B 86, 075454 (2012).
- [20] E. Scalise, M. Houssa, G. Pourtois, V. Afanasev, and A. Stesmans, Nano Res. 5, 43 (2012).
- [21] W. S. Yun, S. W. Han, S. C. Hong, I. G. Kim, and J. D. Lee, Phys. Rev. B 85, 033305 (2012).
- [22] Y. Zhou, Z. Wang, P. Yang, X. Zu, L. Yang, X. Sun, and F. Gao, ACS Nano 6, 9727 (2012).
- [23] Y. Ma, Y. Dai, M. Guo, C. Niu, Y. Zhu, and B. Huang, ACS Nano 6, 1695 (2012).
- [24] P. Johari and V. B. Shenoy, ACS Nano 6, 5449 (2012).
- [25] T. Li, Phys. Rev. B 85, 235407 (2012).
- [26] H. Shi, H. Pan, Y.-W. Zhang, and B. I. Yakobson, Phys. Rev. B 88, 205305 (2013).
- [27] B. Liu, J. Yang, Y. Han, T. Hu, W. Ren, C. Liu, Y. Ma, and C. Gao, J. Appl. Phys. **109**, 053717 (2011).
- [28] P. E. Blöchl, Phys. Rev. B 50, 17953 (1994).
- [29] G. Kresse and D. Joubert, Phys. Rev. B 59, 1758 (1999).
- [30] J. P. Perdew, K. Burke, and M. Ernzerhof, Phys. Rev. Lett. 78, 1396 (1997).
- [31] G. Kresse and J. Furthmller, Comp. Mater. Sci. 6, 15 (1996).

STRAIN-INDUCED ELECTRONIC PHASE TRANSITION ...

- [32] S. Grimme, J. Comput. Chem 27, 1787 (2006).
- [33] A. Tkatchenko and M. Scheffler, Phys. Rev. Lett. **102**, 073005 (2009).
- [34] J. Klimeš and A. Michaelides, J. Chem. Phys. 137, 120901 (2012).
- [35] T. Bučko, S. Lebègue, J. Hafner, and J. G. Ángyán, Phys. Rev. B 87, 064110 (2013).
- [36] J. He, K. Hummer, and C. Franchini, Phys. Rev. B 89, 075409 (2014).
- [37] N. W. Ashcroft and D. N. Mermin, *Solid State Physics*, 1st ed. (Thomson Learning, Toronto, 1976).
- [38] J. M. Ziman, *Principles of the Theory of Solids* (Cambridge University Press, Cambridge, 1964).
- [39] H. J. Monkhorst and J. D. Pack, Phys. Rev. B 13, 5188 (1976).
- [40] Y. Pei, X. Shi, A. LaLonde, H. Wang, L. Chen, and G. J. Snyder, Nature (London) 473, 66 (2011).
- [41] L. L. Boyer, Phys. Rev. B 19, 2824 (1979).
- [42] G. K. Madsen and D. J. Singh, Comput. Phys. Commun. 175, 67 (2006).
- [43] L. Jodin, J. Tobola, P. Pecheur, H. Scherrer, and S. Kaprzyk, Phys. Rev. B 70, 184207 (2004).
- [44] D. Parker and D. J. Singh, New J. Phys. 14, 033045 (2012).
- [45] D. J. Singh, Sci. Adv. Mater. 3, 561 (2011).
- [46] M.-S. Lee, F. P. Poudeu, and S. D. Mahanti, Phys. Rev. B 83, 085204 (2011).
- [47] D. Parker, X. Chen, and D. J. Singh, Phys. Rev. Lett. 110, 146601 (2013).
- [48] T. Pandey, D. J. Singh, D. Parker, and A. K. Singh, J. Appl. Phys. 114, 153704 (2013).
- [49] D. Parker and D. J. Singh, Phys. Rev. B 85, 125209 (2012).
- [50] R. Chianelli, J. Scanlon, and A. Thompson, Mater. Res. Bull. 10, 1379 (1975).
- [51] C. Riekel, J. Solid State Chem. 17, 389 (1976).
- [52] Y. Arnaud and M. Chevreton, J. Solid State Chem. 39, 230 (1981).
- [53] R. B. Murray and A. D. Yoffe, J. Phys. C: Solid State Phys. 5, 3038 (1972).

- [54] G. A. Benesh, A. M. Woolley, and C. Umrigar, J. Phys. C: Solid State Phys. 18, 1595 (1985).
- [55] F. R. Shepherd and P. M. Williams, J. Phys. C: Solid State Phys. 7, 4427 (1974).
- [56] L. F. Mattheiss, Phys. Rev. B 8, 3719 (1973).
- [57] L. D. Hicks and M. S. Dresselhaus, Phys. Rev. B 47, 12727 (1993).
- [58] G. Henkelman, A. Arnaldsson, and H. Jónsson, Comput. Mater. Sci. 36, 354 (2006).
- [59] W. Tang, E. Sanville, and G. Henkelman, J. Phys.: Condens. Matter 21, 084204 (2009).
- [60] M. Dresselhaus, G. Chen, M. Tang, R. Yang, H. Lee, D. Wang, Z. Ren, J.-P. Fleurial, and P. Gogna, Adv. Mater. 19, 1043 (2007).
- [61] J. Heremans, C. M. Thrush, Y.-M. Lin, S. Cronin, Z. Zhang, M. S. Dresselhaus, and J. F. Mansfield, Phys. Rev. B 61, 2921 (2000).
- [62] Y. I. Ravich, B. A. Efimova, and I. A. Smirno, *Semiconducting Lead Chalcogenides* (Plenum, New York, 1970).
- [63] Y. I. Ravich, B. A. Efimova, and V. I. Tamarchenko, Phys. Status Solidi (b) 43, 11 (1971).
- [64] S. Ahmad and S. D. Mahanti, Phys. Rev. B 81, 165203 (2010).
- [65] J. Bardeen and W. Shockley, Phys. Rev. 80, 72 (1950).
- [66] C. Herring and E. Vogt, Phys. Rev. 101, 944 (1956).
- [67] C. A. Kukkonen, W. J. Kaiser, E. M. Logothetis, B. J. Blumenstock, P. A. Schroeder, S. P. Faile, R. Colella, and J. Gambold, Phys. Rev. B 24, 1691 (1981).
- [68] W. Sams, N. Lowhorn, T. Tritt, E. Abbott, and J. Kolis, in *Ther-moelectrics*, 2005. ICT 2005. 24th International Conference on Thermoelectrics (IEEE, Piscataway, 2005) pp. 99–101.
- [69] L. J. Sham and M. Schlüter, Phys. Rev. Lett. 51, 1888 (1983).
- [70] A. Togo, F. Oba, and I. Tanaka, Phys. Rev. B 78, 134106 (2008).
- [71] T. Shiga, T. Hori, and J. Shiomi, Jpn. J. Appl. Phys. 53, 021802 (2014).
- [72] X. Yan, W. Liu, H. Wang, S. Chen, J. Shiomi, K. Esfarjani, H. Wang, D. Wang, G. Chen, and Z. Ren, Energy Environ. Sci. 5, 7543 (2012).