# Unified understanding of the electron-phonon coupling strength for nanocarbon allotropes

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The electron-phonon coupling constant ( $\lambda$ ) of carbon nanotubes ( $\lambda = 0.006$ ) is much smaller than that of alkali-metal-doped C<sub>60</sub> crystals ( $\lambda = 0.6$ ). This difference may be due to the shape of the  $\pi$ -electron conjugated system: the former has a flat, whereas the latter has a sphere. In order to confirm the shape effects in  $\lambda$ , we have examined the magnitude of  $\lambda$  for a one-dimensional (1D) uneven peanut-shaped C<sub>60</sub> polymer that has an intermediate shape of the  $\pi$ -electron conjugated system between a carbon nanotube and a C<sub>60</sub> system, using femtosecond (fs) time-resolved pump-probe spectroscopy, because it can be expected to have an intermediate value of  $\lambda$  between them. Theoretical analysis of fs-transient refractivity obtained experimentally found the magnitude of  $\lambda$  of the 1D C<sub>60</sub> polymer film to be 0.02 as our expectation. This indicates that the shape of the  $\pi$ -electron conjugated system affects the magnitude of  $\lambda$  for nanocarbon allotropes significantly.

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## I. INTRODUCTION

Electron-phonon (e-ph) interaction is one of the most key ingredients for understanding the electronic properties of nanocarbon allotropes, because it is related to many physical phenomena such as superconductivity [1,2], charge-density-wave (CDW) [3], electron transport [4], and optical responses [5,6]. The e-ph coupling constant  $\lambda$  [7] has been determined for those systems so far [8–13]: For example,  $\lambda \simeq 0.6$  for A<sub>3</sub>C<sub>60</sub> [9] and  $\lambda \simeq 0.1$  for A<sub>1</sub>C<sub>60</sub> [10] with A = K, Rb, Cs, while  $\lambda \simeq 0.006$  for single-walled carbon nanotubes (CNTs) [12] and  $\lambda \simeq 5.4 \times 10^{-4}$  for multiwalled CNTs [13].

What causes the large difference in the magnitude of  $\lambda$ between CNTs and  $C_{60}$  systems? When both systems are compared to each other, the former CNTs have a flat-shaped  $\pi$ -electron conjugated system along the electron-conductive direction, whereas the latter C<sub>60</sub> have a spherically shaped  $\pi$ -electron one. This suggests that the shape of  $\pi$ -electron conjugation influences the magnitude of  $\lambda$  for nanocarbon allotropes. Given that C<sub>60</sub> molecules were coalesced to form a structure like CNTs via the generalized Stone-Wales (GSW) rearrangement [14], we could examine the shape effects on the magnitude of  $\lambda$  and thus confirm the above speculation. Although it was reported that C<sub>60</sub> molecules react with each other to form a polymer inside a CNT [15], it is reasonable to consider that the value of  $\lambda$  in the C<sub>60</sub> polymer thus formed is significantly modified by the exterior tubes: The shape effects on  $\lambda$  are not directly observed in the systems. Fortunately, the recent discovery of a one-dimensional (1D) uneven peanut-shaped  $C_{60}$  polymer enables us to confirm the speculation described above.

The 1D  $C_{60}$  polymer has been synthesized from electronbeam (EB) irradiation of pristine  $C_{60}$  films via the GSW transformations between adjacent  $C_{60}$  molecules, as shown in Fig. 1 [16–19], thus it is expected that the shape effects on  $\lambda$  can be observed in such systems because of no exterior tube. The 1D  $C_{60}$  polymer has a cross-linked structure roughly close to that of the P08 C<sub>120</sub> isomer (obtained via GSW) [17]. In other words, the coalescence among  $nC_{60}$  molecules gives rise to form a 1D  $(C_{60})_n$  polymer (*n* is a positive integer). In case of n = 2, there has been reported to be 24 C<sub>120</sub> structural isomers obtained via GSW, and one of which is equivalent to the shortest capped (5,5)CNT [20]. Because the 1D peanut-shaped  $C_{60}$  polymer has an intermediate shape of a  $\pi$ -electron conjugated system between (5,5)CNT and pristine solid  $C_{60}$ , it can be expected that the 1D polymer has an intermediate value between 0.006 and 0.6. In addition, the 1D  $C_{60}$  polymer exhibits fascinating phenomena such as geometrical curvature effects on Tomonaga-Luttinger liquid states [21], CDW transition at 50 K [22,23], and anomaly in the resistivity at temperatures lower than 50 K [24]. In particular, the latter two properties are also related to the e-ph interaction and thus inspire us to examine the magnitude of  $\lambda$  for the 1D C<sub>60</sub> polymers.

In the present work we report on the  $\lambda$  value of the 1D uneven peanut-shaped C<sub>60</sub> polymer by examining the temperature dependence of the relaxation time of photoexcited electrons in the polymer, and confirm that the shape of  $\pi$ -electron conjugation affects the magnitude of  $\lambda$  for nanocarbon allotropes.

In Sec. II we describe briefly experimental details. In Sec. III we show measured transient reflectivity changes and confirm that these results are consistent with the previously measured one [22]. From these data we estimate the e-ph coupling constant using a theoretical model. In Sec. IV we confirm that the estimated length of the 1D  $C_{60}$  polymer from the magnitude of the e-ph coupling is quite similar to the previous data obtained by the resistivity measurement [24]. We also compare the magnitude of the e-ph coupling of the 1D  $C_{60}$  polymer with that of other nanocarbon allotropes. In Sec. V we conclude our paper.

## **II. EXPERIMENTS**

Metallic 1D uneven peanut-shaped  $C_{60}$  polymers were formed from 3-kV electron-beam irradiation of a 500-nm-thick pristine  $C_{60}$  film deposited on a cesium iodide (CsI) substrate

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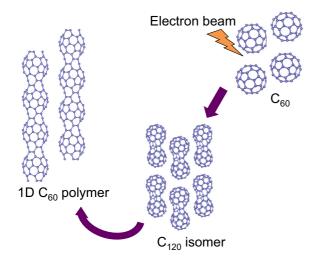


FIG. 1. (Color online) Schematic illustration showing the synthesis of the 1D  $C_{60}$  polymer. When the electron beam is irradiated to  $C_{60}$ , two  $C_{60}$  are coalesced with each other via the GSW transformation and transformed into  $C_{120}$  isomers, followed by the synthesis of the 1D  $C_{60}$  polymer.

(15 mm in diameter, 2 mm thick) for approximately 200 h in an ultrahigh vacuum (UHV) chamber (a base pressure  $<10^{-6}$  Pa) equipped with a high-resolution (0.25 cm<sup>-1</sup>) Fourier-transform infrared (FT-IR) spectroscopy (Mattson Research Series). The formation of the 1D  $C_{60}$  polymers was confirmed by FT-IR spectra measured in situ [25]. A slight blueshift of the optical phonon frequency was observed due to the bundling of the polymers [26]. In addition, the 1D structured  $C_{60}$  polymer has been confirmed using electron diffraction [27]. Furthermore, we have measured the ultraviolet-visible absorption spectra of pristine C<sub>60</sub> and 1D  $C_{60}\ polymers$  films. The pristine  $C_{60}\ shows$  intense peaks at 350 nm (3.54 eV) and 450 nm (2.75 eV), corresponding to the HOMO-1 and HOMO bands, respectively, whereas the 1D  $C_{60}$  polymers film shows that the absorption bands are broadened and continuum, where the absorption edge spread over to a wavelength longer than 1100 nm (less than 1.14 eV) [29]. Similar band structures have been observed in the photoemission spectra: Some intensive narrow bands become broadened after the EB irradiation, and the valence electronic states spread toward the Fermi level, clearly indicating the absence of the energy gap [30]. Thereafter, the 1D polymer film on CsI was taken out of the UHV chamber, and mounted on a copper holder within a liquid helium flow cryostat. We used a femtosecond (fs) UV/NIR pump-probe spectroscopy to measure the relaxation time of photoexcited electrons. The light sources were generated from a mode-locked cavity-dumped Ti:sapphire laser with a repetition rate of 270 kHz, and a pulse duration of 120 fs. The UV (3.0 eV) pump pulses with a fluence of 50 mJ/cm<sup>2</sup> were produced by the second harmonic generation (SHG) of the laser. The nonequilibrium electron transients induced by the pump were evaluated by the reflectivity changes of the NIR (1.5 eV) probe pulses whose fluence was an order of magnitude weaker than that of the pump. The excited carriers created by the pump pulse are thermalized within a

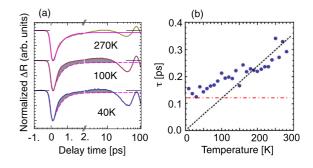


FIG. 2. (Color online) (a) The normalized reflectivity  $\Delta R$  as a function of delay time *t* of 1D C<sub>60</sub> polymers at *T* = 40, 100, and 270 K. The shaded areas (gray) indicate the differences from the fitting obtained at *T* = 270 K. The oscillations at *t* > 10 ps are due to the acoustic pulses. (b) *T* dependence of  $\tau$ . The dashed line is the fits to the data using Eq. (2). The dashed-dotted line (red) is the pulse width of 120 fs.

few hundred fs through the electron-electron scatterings and obey the Fermi-Dirac distribution function with an electron temperature higher than the lattice temperature [31]. These excess carriers above the Fermi level would be re-excited by absorbing the probe pulses, which leads to the transient changes in the reflectivity. Although the 1D C<sub>60</sub> polymer was polycrystalline on the CsI substrate (crystalline size: 8–18 nm from XRD), the transient responses were not affected by grain boundaries but reflected the carrier relaxation dynamics peculiar to the quasi-1D structures [22].

### **III. RESULTS AND ANALYSES**

## A. Carrier relaxation dynamics

Figure 2(a) shows the typical transient reflectivity changes  $(\Delta R)$  of the 1D C<sub>60</sub> polymer film obtained at T = 40, 100,and 270 K. The outline of the data is similar to those measured previously [22]. The sudden decrease in  $\Delta R$  at  $t \approx 0$  fs is caused by the electrons excitation by the pump pulse. The subsequent decay reflects their relaxation via e-ph scatterings, which will be characterized by its decay time below. There is also a slow decay component, where the oscillations of  $\Delta R$  becomes dominant at delay time t longer than 10 ps. Such oscillations have often been observed in the pumpprobe reflection measurements of the thin films [32] and are attributed to a result of the moving acoustic pulses generated by the instantaneous relaxation of electrons. Hereafter we neglect the oscillation component for evaluating the  $\lambda$  value directly from the nonequilibrium electrons relaxation. Based on the two kinds of decay components, fast and slow components, we extracted the associated relaxation time by fitting the measured reflectance with the following equation:

$$\Delta R \propto A_{\text{fast}} \exp\left(-\frac{t}{\tau}\right) + C_{\text{slow}}.$$
 (1)

Here  $\tau$  denotes the relaxation time corresponding to the fast component, whereas  $A_{\text{fast}}$  and  $C_{\text{slow}}$  are the amplitude of the fast and slow component, respectively. We assume the relaxation time of the slow component to be infinite because of its long decay time on the order of nanoseconds.

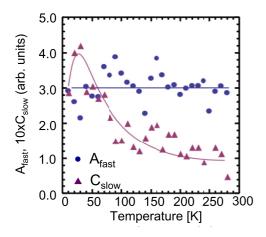


FIG. 3. (Color online) The amplitudes of both  $A_{\text{fast}}$  (solid circle) and  $C_{\text{slow}}$  (solid triangle) defined in Eq. (1) as a function of temperature. The solid curves are guides to the eyes.

To enhance the difference in  $\Delta R$  among the temperatures [see Fig. 2(a)], we plot a curve (solid line at T = 270 K and dashed lines at T = 40 and 100 K) obtained by a fit to the data at T = 270 K using Eq. (1): the differences are indicated by the shaded areas. As T decreases, the decay time of  $\tau$ became faster. To analyze the relaxation dynamics in more detail, we examined T dependence of  $\tau$ , as shown in Fig. 2(b). At temperatures above 100 K,  $\tau$  exhibited a quasilinear T dependence, which implies that the relaxation dynamics is governed by the e-ph interaction. According to the equation derived by Kabanov and Alexandrov [33], the  $\tau$  caused by the e-ph interaction in a metal can be expressed by the following relationship:

$$\tau = \frac{2\pi k_B T}{3\hbar\lambda\langle\omega^2\rangle}.$$
(2)

Here  $k_B$  is the Boltzmann constant,  $\omega$  is the phonon frequency, and  $\hbar$  is the reduced Planck's constant. In addition,  $\lambda \langle \omega^n \rangle = 2 \int_0^\infty \alpha^2 F(\omega') (\omega')^{n-1} d\omega'$ , where  $\alpha^2 F(\omega)$  is the Eliashberg function. Since the relaxation time at T > 100 K is larger than the pulse width  $\sim 120$  fs, it is reasonable to ignore the effect of the pulse width on the relaxation time. Using Eq. (2), we obtained  $\lambda \langle (\hbar \omega)^2 \rangle = 100 \text{ (meV)}^2$ . The magnitude of  $\lambda$ is discussed later. We next examine the slow component to complete the temperature dependent analysis.

Figure 3 shows the T dependence of both  $A_{\text{fast}}$  (solid circle) and  $C_{\text{slow}}$  (solid triangle) components. As T decreases from room temperature,  $C_{\rm slow}$  increases gradually until ~80 K and drastically increases at temperatures below 80 K, while  $A_{\text{fast}}$  fluctuates and seems to be independent of T. The T dependence of the slow component suggests another relaxation process that dominates the carrier relaxation at temperatures below 80 K. We have previously reported that a phonon bottleneck mechanism considering the CDW gap formation dominates the carrier relaxation of the 1D C<sub>60</sub> polymer film at temperatures below 50 K [22,23]. In the experiments, a similar temperature dependence of the amplitudes has been observed: The amplitude of the slow component increases as Tdecreases, while that of the fast one is independent of T [22]. It is reasonable to consider that the T dependence of  $C_{\text{slow}}$ obtained in the present study was due to the same mechanism.

## B. Estimation of $\lambda$

To evaluate the magnitude of  $\lambda$  for the 1D C<sub>60</sub> polymer, we assume that the Eliashberg function has the phonon density-of-states (DOS) shape, as assumed in Ref. [34]. Prior to estimating the upper limit of  $\lambda$ , we consider the following shape of DOS:

$$F(\omega) = A \left(\frac{\omega}{\omega_D}\right)^2 \theta(\omega_D - \omega) + B\delta(\omega - \omega_E), \quad (3)$$

where  $\omega_D$  is the cutoff frequency of acoustic phonon modes, and  $\omega_E$  is the typical optical phonon frequency.  $\theta(\omega)$  is the Heaviside step function. Normalization condition  $\int_0^\infty F(\omega)d\omega = 1$  gives  $A = 3(1-B)/\omega_D$ .  $\lambda$  is a decreasing function of B in the present model. To estimate the upper limit of  $\lambda$ , we first consider the case of B = 0, i.e., only the interaction between electrons and acoustic phonons contributes to the magnitude of  $\lambda$ . Then we obtained  $\lambda \langle (\hbar \omega)^2 \rangle / \lambda = (\hbar \omega_D)^2 / 2 \simeq$  $1.0 \times 10^3 \,(\text{meV})^2$  when we set  $\hbar \omega_D = 45 \,\text{meV} \,[23,35]$ . Thus we evaluated the magnitude of  $\lambda$  to be 0.1, which is comparable to that for  $AC_{60}$ . We next consider how the presence of the optical phonons lowers the magnitude of  $\lambda$ . We previously reported that the optical phonon mode with energy  $\hbar\omega_E =$  $1340 \text{ cm}^{-1} = 166 \text{ meV}$  has a large e-ph coupling [36]. When we set B = 0.5 and 1, we obtain  $\lambda \langle (\hbar \omega)^2 \rangle / \lambda = 5.5 \times 10^3$  and  $5.5 \times 10^4$  (meV)<sup>2</sup>, respectively. Then these values provide  $\lambda =$ 0.02 and 0.002, respectively. Here the latter is the lower limit of  $\lambda$ . Since both the acoustic and optical phonons contribute to the e-ph interaction, it is reasonable to determine that the value of  $\lambda$  is 0.02 [37], which supports our prediction that the 1D  $C_{60}$  polymer has an intermediate value of between AC<sub>60</sub> crystals and CNTs on the basis of the shape dependence of  $\lambda$ .

### **IV. DISCUSSION**

The transient reflectivity in the present sample can be nicely fitted by a single-exponential decay, which means that excited electrons decay by emitting their energy into a single thermal bath. In the estimation of the e-ph coupling, we assumed that the oscillation period for the Debye frequency and the optical phonon frequency is ~0.1 and ~0.03 ps, respectively. These periods are clearly shorter than the observed relaxation time shown in Fig. 2(b). Therefore, it is reasonable to consider a phonon bath as the corresponding thermal bath. In fact, the relaxation time is linearly proportional to *T*, which is consistent with the relaxation dominated by the e-ph interaction discussed in Ref. [33]. Furthermore, the interaction between polymers is very weak, which has been evidenced by IR spectroscopy [26]. This fact eliminates the effect of the electronic transfer between polymers on the carrier relaxation.

We can validate our estimation of  $\lambda$  quantitatively via the determination of the length of the 1D C<sub>60</sub> polymer, using the theory of Devos and Lannoo [38]. They examined the e-ph couplings for molecular crystals systematically, using density-functional theory within the local-density approximation and reported that  $\lambda$  can be factorized into  $\lambda = N(E_F)V$ , where  $N(E_F)$  is the electron DOS at the Fermi level  $(E_F)$  and V is the intramolecular e-ph coupling. They showed that the value of V is inversely proportional to the number of  $\pi$  electrons  $(N_{\pi})$  of atoms involved in the  $\pi$  states of the molecule, and derived the relationship  $V = 1800/N_{\pi}$  meV. For example,

TABLE I. Electron-phonon coupling constant  $\lambda$  of nanocarbon allotropes. The kind of chemical bonding between C<sub>60</sub> is also shown.

C <sub>60</sub> compound	λ	Bonding (C <sub>60</sub> -C <sub>60</sub> )
$\overline{A_3C_{60} (A = K, Rb)}$	0.6 (0.5)	van der Waals
$Na_4C_{60}$	0.3	Single C-C $\sigma$ bond
$AC_{60} (A = K, Rb)$	$\sim 0.1$	2+2 cycloadditional bond
1D C <sub>60</sub> polymer	0.02	Coalesced GSW bond
Single-walled CNT	0.006	_
Multiwalled CNT	$5.4 \times 10^{-4}$	_

V = 300 meV for C<sub>6</sub>H<sub>6</sub> and V = 50 meV for C<sub>60</sub>, though a slight deviation of the relation in the  $C_{60}$  may be attributed to a curvature effect due to its spherical shape. Since the interaction between adjacent 1D  $C_{60}$  polymers are governed by the van der Waals interaction, the 1D  $C_{60}$  polymers can be regarded as a molecular crystal. It is reasonable to set  $N(E_F) \simeq 10/\text{eV/spin}$ , because  $N(E_F)$  of the 1D C<sub>60</sub> polymer is slightly smaller than that of A<sub>3</sub>C<sub>60</sub> [39]. Then we evaluate  $V = \lambda/N(E_F)$  to be 2 meV, which gives a value of  $N_{\pi} = 900$ . Since there are  $60\pi$  electrons per one C<sub>60</sub>, the value of  $N_{\pi} = 900$  suggests that the 1D C<sub>60</sub> polymer may consist of 15 C<sub>60</sub> molecules at short, on which a conducting electron can move without any scattering. In fact, the length of tube consisting of 15  $C_{60}$ molecules is in coincidence with the mean hopping distance of 8–13 nm obtained for the 1D  $C_{60}$  polymer film [24]. This strongly supports our estimation of the  $\lambda$  value. From the relation  $\lambda = 1800 N(E_F)/N_{\pi}$ , the longer polymer has the smaller e-ph coupling, which would be verified by using other samples prepared on such as mica substrates instead of CsI substrates [28].

The concept of chemical bonding may provide us a unified understanding of the shape effects on the magnitude of  $\lambda$ for nanocarbons. Table I summarizes the value of  $\lambda$  for various nanocarbon allotropes, along with the classification of chemical bonding between adjacent C<sub>60</sub> molecules. Since the strength of the chemical bonding can be ordered as follows: van der Waals < single C-C  $\sigma$  bond < 2 + 2 cycloadditional bond (two C-C  $\sigma$  bonds) < coalesced GSW bond (not only more than two C-C  $\sigma$  bonds but also C-C  $\pi$  bonds) [see also Figs. 4(a)-4(d)], it can be said that a stronger chemical bonding gives rise to a larger  $N_{\pi}$  in association with an enhancement in the electron hopping probability between adjacent C<sub>60</sub> molecules. This results in a small value of V and  $\lambda$ , as described above for the 1D C<sub>60</sub> polymers. The single-walled (or multiwalled) CNT is one extreme case [Fig. 4(e)].

Although we did not consider any lattice model for the 1D  $C_{60}$  polymer in the present study, we have already employed several kinds of models for the 1D  $C_{60}$  polymers, using the GSW transformation [40]. It would be theoretically interesting to investigate the e-ph interaction matrix elements for the excited electrons in those models. If the matrix elements involving the optical and/or acoustic phonons are strong, the nature of the interaction may be attributed to polar and/or

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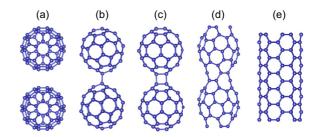


FIG. 4. (Color online) Schematic illustration of nanocarbon allotropes: two C<sub>60</sub> between which are connected via (a) van der Waals interaction, (b) single C-C  $\sigma$ -bond, (c) 2 + 2 cycloadditional bond, (d) 1D C<sub>60</sub> polymer having coalesced GSW bond, and (e) singlewalled CNT.

deformation coupling, respectively. Unfortunately, calculations both of the phonon spectra and of the e-ph coupling constant for all the models require a large amount of computational time, owing to a large number of carbon atoms in a unit cell. This is in progress.

Alternatively, electron transport measurements would be useful to investigate the nature of the e-ph interaction in low-dimensional systems. In fact, the e-ph interaction in the metallic CNTs has been studied using an atomic force microscope as an electrical probe [4]. A similar technique would enable us to clarify the nature of the e-ph interaction in the 1D C<sub>60</sub> polymers, while we postulate that the nature of the interaction has an intermediate character between solid C<sub>60</sub> and single-walled CNTs.

### V. SUMMARY

In summary, we have examined carrier relaxation dynamics of the 1D C<sub>60</sub> polymers, using femtosecond time-resolved pump-probe spectroscopy, and obtained the e-ph coupling constant ( $\lambda$ ) to be 0.02, which supports our speculation that the 1D peanut-shaped C<sub>60</sub> polymer has an intermediate value of between A<sub>x</sub>C<sub>60</sub> and CNTs on the basis of the shape of a  $\pi$ -electron conjugated system. This implies that the magnitude of an e-ph coupling constant for nanocarbon allotropes is significantly affected by the shape of a  $\pi$ -electron conjugated system. We believe that this finding becomes useful for a unified understanding of electron-phonon interaction for nanocarbon allotropes.

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