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Antiferroelectricity in thin-film ZrO2 from first principles

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Density-functional calculations are performed to investigate the experimentally reported field-induced phase transition in thin-film ZrO_2 [J. Müller *et al.*, [Nano Lett.](http://dx.doi.org/10.1021/nl302049k) [12](http://dx.doi.org/10.1021/nl302049k), [4318](http://dx.doi.org/10.1021/nl302049k) [\(2012\)](http://dx.doi.org/10.1021/nl302049k)]. We find a small energy difference of ∼1 meV*/*f.u. between the nonpolar tetragonal and polar orthorhombic structures, characteristic of antiferroelectricity. The requisite first-order transition between the two phases, which atypically for antiferroelectrics have a group-subgroup relation, results from coupling to other zone-boundary modes, as we show with a Landau-Devonshire model. Tetragonal $ZrO₂$ is thus established as a lead-free antiferroelectric with excellent dielectric properties and compatibility with silicon. In addition, we demonstrate that a ferroelectric phase of ZrO₂ can be stabilized through epitaxial strain, and suggest an alternative stabilization mechanism through continuous substitution of Zr by Hf.

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Zirconia $(ZrO₂)$ is a high-k dielectric [\[1\]](#page-3-0), chemically and structurally similar to $HfO₂$, and likewise is a candidate for dynamic random access memory (DRAM) applications [\[2,3\]](#page-3-0) and complementary metal-oxide-semiconductor (CMOS) devices [\[4,5\]](#page-3-0). Bulk $ZrO₂$ has a high-symmetry cubic $(Fm3m)$ structure [Fig. [1\(a\)\]](#page-1-0) above 2400 K, and a tetragonal $(P4₂/nmc)$ structure [Fig. [1\(b\)\]](#page-1-0) between 2400 and 1200 K [\[6\]](#page-3-0). The tetragonal structure is related to the cubic structure by freezing in an unstable X_2^- mode [\[7\]](#page-3-0) and is nonpolar. Below 1200 K, $ZrO₂$ is monoclinic ($P₁/c$) [Fig. [1\(c\)\]](#page-1-0). The first-order transition from the tetragonal phase to the monoclinic phase changes the coordination number of Zr from 8 to 7 and increases the volume by ∼5%.

In light of the extensive research which has been conducted over the past 50 years on this relatively simple dielectric, the recent report of antiferroelectriclike double-hysteresis loops in thin film $ZrO₂$ [\[8\]](#page-3-0) at first seems rather surprising. In thin film $ZrO₂$, the tetragonal-monoclinic transition temperature is suppressed and the structure is tetragonal at room temperature $[9-11]$; in contrast, thin film $HfO₂$ is monoclinic at room temperature and exhibits simple dielectric behavior. The field-induced polar phase in $ZrO₂$, which appears above a critical field on the order of 2 MV*/*cm, is isostructural with the ferroelectric phases that have been observed in thin films of HfO2 doped with Al [\[12\]](#page-3-0), Y [\[13\]](#page-3-0), Gd [\[14\]](#page-3-0), Si [\[15,16\]](#page-4-0), and Sr [\[17\]](#page-4-0), as well as in $(Hf_{1/2}Zr_{1/2})O_2$ thin films [\[18,19\]](#page-4-0). The structure of the polar phase is orthorhombic $(Pca2_1)$ [\[20\]](#page-4-0) and corresponds to a distortion of the high-symmetry cubic structure, as depicted in Fig. [1.](#page-1-0)

Antiferroelectrics have recently been the subject of increasing interest [\[21\]](#page-4-0). The characteristic electric-field-induced transition from a nonpolar to a strongly polar phase is the source of functional properties and promising technological applications. Nonlinear strain and dielectric responses due to the phase switching are useful for transducers and electro-optic applications [\[22,23\]](#page-4-0). The shape of the doublehysteresis loop suggests applications in high-energy storage capacitors [\[24,25\]](#page-4-0). In addition, an electrocaloric effect can be observed in systems with a large entropy change between the two phases [\[26\]](#page-4-0). While most attention has focused on Pb $ZrO₃$ and related perovskites $[27]$, a recent theoretical materials design search [\[28\]](#page-4-0) suggested that there are many more antiferroelectric compounds to be discovered.

In this Rapid Communication, we use first-principles calculations to provide clear evidence that the tetragonal phase of $ZrO₂$ is an antiferroelectric material, and that the behavior observed in thin films is intrinsic. We find a remarkably small energy difference of ∼1 meV per formula unit (meV*/*f.u.) between the nonpolar tetragonal and polar orthorhombic structures, which is a key characteristic of antiferroelectricity [\[29–32\]](#page-4-0). We show with a polynomial expansion of the energy that the requisite first-order transition between nonpolar and polar phases, which in the present case, atypically for antiferroelectrics, have a group-subgroup relation, results from coupling to other zone-boundary modes. This mechanism for antiferroelectricity provides a basis for broader searches for antiferroelectric materials with optimal functional properties. In addition, we demonstrate that the polar phase of $ZrO₂$ can be stabilized through epitaxial strain, and we discuss the possibility of ferroelectricity in the solid solution $(Zr_{1-x}Hf_x)O_2$.

Density-functional theory (DFT) calculations are performed using version 7.4.1 of ABINIT [\[33\]](#page-4-0) and the localdensity approximation (LDA). We use a plane-wave energy cutoff of 680 eV, and a $4 \times 4 \times 4$ Monkhorst-Pack sampling of the Brillouin zone [\[34\]](#page-4-0) for all structural optimizations. Polarization was calculated on a $10 \times 10 \times 10$ grid using the modern theory of polarization [\[35\]](#page-4-0) as implemented in ABINIT. We used norm-conserving pseudopotentials from the Bennett-Rappe library [\[36\]](#page-4-0) with reference configurations, Zr $((Kr)4d^{0}5s^{0})$, Hf $((Xe)4f^{14}5d^{0.5}6s^{0})$, and O $((He)2s^{2}2p^{4})$, generated by the OPIUM code [\[37\]](#page-4-0).

Table [I](#page-1-0) reports relaxed bulk energies and lattice constants for the experimentally observed bulk phases of $ZrO₂$. Our results are in good agreement with previous first-principles results [\[38–41\]](#page-4-0). The monoclinic phase is the lowest-energy structure, consistent with experiments. Experimental lattice constants [\[20,42–44\]](#page-4-0) are underestimated by 1%, typical for LDA. The spontaneous polarization of the orthorhombic structure was found to be $P_r \sim 58 \mu C/cm^2$.

We begin by discussing the energy-lowering distortions of the cubic phase of ZrO_2 . The zone-boundary X_2^- mode

FIG. 1. (Color online) Experimentally reported phases of $ZrO₂$: (a) Cubic $(Fm\overline{3}m)$, (b) tetragonal $(P4_2/nmc)$, (c) monoclinic $(P2₁/c)$, and (d) orthorhombic $(Pca2₁)$.

corresponds to an antipolar displacement of oxygen atoms in the \hat{x} direction. It is the single instability exhibited in the first-principles phonon dispersion of the high-symmetry cubic structure [\[45\]](#page-4-0) and, as mentioned above, leads to the tetragonal $P4_2/nmc$ phase with amplitude 0.319 Å and fourfold axis along \hat{x} . We find that the tetragonal structure is a local minimum, with no unstable modes. Therefore, even though the symmetry analysis shows that the transition from the *z*ˆ-polarized tetragonal structure to the orthorhombic *Pca*21 phase is allowed to be second order, the field-induced transition is expected to be first order.

The distortion relating the tetragonal $P4_2/nmc$ structure to the polar orthorhombic *Pca*21 phase can be decomposed into symmetry-adapted modes of the high-symmetry cubic

TABLE I. Energy ΔE (meV/f.u.), volume expansion $\Delta V/V$ (%), and lattice parameters (Latt.) in \AA (β : monoclinic angle) of the experimental phases of $ZrO₂$.

Phase	ΔE				$\Delta V/V$ Latt. This work Prev. work ^a	Expt.
Fm3m	0	0	a ₀	5.03	5.04	5.03
$P4_2/nmc -47.8$		2.0	$\mathfrak a$	5.04	5.03	5.09
			$\mathcal{C}_{0}^{(n)}$	5.12	5.10	5.19
Pca2 ₁	-48.2	3.6	a	5.22	5.26	5.26
			b	5.02	5.07	5.07
			\mathcal{C}	5.04	5.08	5.08
$P2_1/c$	-82.0	7.2	$\mathfrak a$	5.09	5.11	5.14
			h	5.20	5.17	5.20
			\mathcal{C}	5.24	5.27	5.31
			β	99.39	99.21	99.17

a References [\[40,41\]](#page-4-0).

 b References [\[20,42–44\]](#page-4-0).

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TABLE II. Wave vector (*q*) in reciprocal lattice units $(2\pi/a_0)$ and eigenvector of the structural modes contained in the $Pca2₁$ structure, specified by the independent nonzero displacements. With Zr at the origin, O_1 at $a_0/4(111)$, and O_2 at $a_0/4(111)$. Q_{th} and Q_{expt} are the computed and experimental mode amplitudes, defined as described in the text.

 $Fm\overline{3}m$ structure. Mode amplitudes are computed with respect to eigenvectors normalized to 1 \AA . Table II shows excellent agreement between first-principles (Q_{th}) and experimental (Q_{expt}) values of the mode amplitudes. There are six nonzero modes in addition to the X_2^- mode: $\Gamma_4^-, X_{5,x}^+, X_{5,y}^+, X_{5,z}^+, X_5^$ and X_3^- [\[46\]](#page-4-0). The zone-center Γ_4^- mode corresponds to a polar displacement of the zirconium atoms relative to the oxygen atoms in the \hat{z} direction. The three X_5^+ modes involve antipolar displacements of planes of oxygen atoms. The amplitudes of the X_5^- and X_3^- modes are relatively small.

We explore the field-induced tetragonal-to-orthorhombic transition by performing relaxations of the structures obtained by linear interpolation of the atomic positions, holding fixed the value of the polar mode amplitude $Q_{\Gamma_4^-}$. Figure 2 shows the energy of the polar structure as a function of $Q_{\Gamma_4^-}$. For small values of $Q_{\Gamma_4^-}$, the energy of the system increases, as expected for freezing in a stable mode. The induced polar structure has an orthorhombic space group *Aba*2, different from that of the polar orthorhombic *Pca*21 phase, and relaxes back to the tetragonal phase if the constraint is removed. At $Q_{\Gamma_4} \sim 0.134$ Å, we find an energy cusp with magnitude $\delta \vec{E} \sim 35$ meV/f.u., which separates the *Aba*2 structure with $Q_{X_{5,y}^+} \neq 0$ and $Q_{X_{5,x,z}^+} = 0$, from the $Pca2_1$ phase with $Q_{X^+_{5,x,y,z}} \neq 0$ [\[47\]](#page-4-0). Above this threshold value, the system volume expands (∼1.5%), and the structure relaxes to the

FIG. 2. (Color online) First-principles calculations (squares) and Landau-Devonshire model (solid line) of the energy profile between tetragonal and orthorhombic phases.

FIG. 3. (Color online) First-principles calculations (circles and squares) and model fitting (solid and dashed lines). (a) Energy of the polar *Aba2* structure as a function of $Q_{\Gamma_4^-}$ with $Q_{X_{5,y}^+} = 0$. (b) $Q_{\Gamma_4^-}$ as a function of $Q_{X_{5,x,y,z}^+} = Q_{X_5^+}$ (circles) and $Q_{X_{5,y}^+} = Q_{X_5^+}$ (squares). (c) Energy of the polar $Pca2_1$ structure as a function of $Q_{X_{5,x,y,z}^+} = Q_{X_5^+}$ (circles) and $Q_{X_{5,y}^+} = Q_{X_5^+}$ (squares).

polar orthorhombic *Pca*21 phase if the constraint is removed. Assuming that the critical field E_C required to overcome the energy barrier is given by $E_C \sim \delta E/V_C P_C$ (with $P_C =$ 36 μ C/cm² and $V_C = 32.9 \text{ Å}^3/\text{f.u.}$, the first-principles values of polarization and volume at the barrier), we estimate the critical field as $E_C \sim 4.7$ MV/cm.

In order to characterize this unusual energy surface, we describe the energetics of a single orthorhombic *Pca*21 variant based on an expansion around the minimum-energy tetragonal P_2/nmc structure using modes of the high-symmetry $Fm3m$ structure as a basis set:

$$
E_T(Q_{\Gamma_4^-}, Q_{X_{5,a,b,c}^+}) = E_T^0 + a_1 Q_{\Gamma_4^-}^2 + a_2 Q_{\Gamma_4^-} Q_{X_{5,y}^+} + S(Q_{X_{5,x}^+}, Q_{X_{5,z}^+}), \qquad (1)
$$

where *S* is a function to be determined below. The invariance of the term $a_2Q_{\Gamma_4^-}Q_{X_{5,y}^+}$ originates in the fact that the expansion is around a structure of lower symmetry than the structure used to construct and label the modes (for details, see the Supplemental Material [\[48\]](#page-4-0)).

The values of the coefficients are obtained by fitting the model to first-principles calculations. By freezing in $Q_{\Gamma_4^-}$ in the tetragonal phase, and relaxing the structure with $Q_{X^+_{5,y}}$ constrained to zero, we extract the values of E_T^0 and a_1 [Fig. $3(a)$]. For the two sets of structures obtained by freezing in $Q_{X_{5,y}^+}$, and separately $Q_{X_{5,x}^+} = Q_{X_{5,y}^+} = Q_{X_{5,z}^+}$, and relaxing, we obtain the energies and the relaxed values of $Q_{\Gamma_4^-}$. By fitting the relaxed values of $Q_{\Gamma_4^-}$ to the expression $Q_{\Gamma_4^-}$ $-(a_2/2a_1)Q_{X_{5,y}^+}$ obtained for both structures by minimizing Eq. (1) with respect to $Q_{\Gamma_4^-}$, we extract the value of a_2 [Fig. $3(b)$]. Finally, we use the computed energies of these two sets of structures to obtain *S*(0, *Q*, 0) = $a_3 Q^2$ and *S*(*Q*, *Q*, *Q*) as a spline interpolation [Fig. $3(c)$].

Using the model we can reproduce the energy cusp of Fig. [2](#page-1-0) by plotting

$$
\min_{\mathcal{Q}_{X^{\pm}_{5,x,y,z}}} E_T\big(\mathcal{Q}_{\Gamma_4^-},\mathcal{Q}_{X^{\pm}_{5,x,y,z}}\big)
$$

as a function of Q_{Γ_4} . As shown in Fig. [2,](#page-1-0) the model is able to capture the essential elements of the energy landscape: the position and height of the energy barrier, the two local minima and the energy difference between them. The disagreement between first principles and the model can be attributed to the truncation of the expansion in Eq. (1) . The cusp is caused by the multiple local minima displayed by the energy as a function of the X_5^+ mode, with a small change in the value of $Q_{\Gamma_4^-}$ causing a switch between two local minima (Fig. 4).

We investigate the effect of epitaxial strain on the relative stability of the tetragonal, orthorhombic, and monoclinic phases. Epitaxial strain is simulated through "strained-bulk" calculations [\[49,50\]](#page-4-0). We denote the epitaxially (*e*) strained $Pca2₁$ structure as $ePca2₁$, and indicate the matching plane by a prefix labeling the normal to the plane (for example, normal vector t_a yields a - $ePca2_1$).

Figure [5](#page-3-0) shows the calculated epitaxial strain diagram for $ZrO₂$. As expected from the dissimilar lattice constants, the equilibrium energy strains (σ) for the three phases and various matching planes are quite different ($\sigma_c^M = 2.3\%$ and $\sigma_a^M = 3.8\%$ for monoclinic, $\sigma_a^O = 0.0\%$ and $\sigma_b^O = 1.9\%$ for orthorhombic, and $\sigma_c^T = 0.2\%$ for tetragonal phases, respectively). The computed values of σ are in excellent agreement with the estimates obtained by comparing the relevant relaxed lattice vectors with the reference lattice constant $a_0 = 5.03$ Å, as discussed in Ref. [\[31\]](#page-4-0).

For large values of tensile strain (*>*1%), the monoclinic phases are highly favorable in energy. As the strain is decreased, the relative stability of the monoclinic phase is reduced and the tetragonal and orthorhombic phases become

FIG. 4. (Color online) Energy (E_T) as a function of $Q_{X_{5,y}^+} =$ $Q_{X_5^+}$ (solid line) and $Q_{X_{5,x,y,z}^+} = Q_{X_5^+}$ (dashed line) at fixed values of the mode amplitude $Q_{\Gamma_4^-}$. The red arrows show the global energy minimum for each value of $Q_{\Gamma_4^-}$.

FIG. 5. (Color online) Epitaxial strain diagram for $ZrO₂$. FE and PE refer to ferroelectric and paraelectric ground state.

favorable. In the strain range from -1.0% to $+1.5\%$, the tetragonal and orthorhombic phases are very close in energy. In this regime, if the appearance of the lower-energy monoclinic phase is suppressed by surface effects $[9-11]$ or other means [\[51,52\]](#page-4-0), the system is expected to be antiferroelectric. For large values of compressive strain (*<*−1%), the ferroelectric structure a - e P *ca* $2₁$ is favored. While within the accuracy of our calculations it is not possible precisely to predict the critical strains that will be observed in experiments, we expect stabilization of ferroelectricity at accessible values of compressive strain.

Finally, as an illuminating comparison with $ZrO₂$, we consider the stability of the corresponding tetragonal, orthorhombic, and monoclinic structures in $HfO₂$. The calculated lattice constants for the structures of $HfO₂$ are shown in Table III. The smaller ion size of Hf⁺⁴ compared to Zr^{+4} explains the reduced lattice constants of $HfO₂$, in good agreement with previous calculations [\[53\]](#page-4-0). The computed spontaneous polarization of 60 μ C/cm² is consistent with previous results [\[54\]](#page-4-0). The main difference between $ZrO₂$ and $HfO₂$ is the higher relative energy of the tetragonal structure in $HfO₂$, so that the polar orthorhombic structure is favored over the nonpolar tetragonal phase by ∼23 meV*/*f.u. This suggests that isovalent substitution of Zr by Hf will favor the polar orthorhombic phase in thin films of $(Zr_{1-x}Hf_x)O_2$, consistent with Refs. [8[,18,19\]](#page-4-0).

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TABLE III. Energy ΔE (meV/f.u.), lattice constants (\AA), monoclinic angle (β), and volume expansion $\Delta V/V$ (%) for the ZrO₂-type phases of HfO₂.

Phase	ΛE	a		c		$\Delta V/V$
Cubic		4.89				θ
Tetragonal	-22.5	4.90		4.95		1.3
Orthorhombic	-45.1	5.07	4.88	4.89		3.4
Monoclinic	-93.2	4.95	5.06	5.08	99.53	7.0

Estimation of the minimum energy strains for the relevant structures of HfO₂ gives similar values to the case of $ZrO₂$ $(\sigma_c^M = 2.4\% \text{ and } \sigma_a^M = 3.7\% \text{ for monoclinic, } \sigma_a^O = -0.1\%$ and $\sigma_b^0 = -1.0\%$ for orthorhombic, and $\sigma_c^T = 0.2\%$ for tetragonal structures, respectively). Therefore, the epitaxial strain diagram should be similar to that of $ZrO₂$ (Fig. 5) but with the epitaxial strain curve of the tetragonal structure shifted higher by ∼23 meV*/*f.u. Based on this, we speculate that ferroelectricity would be observed in $HfO₂$ over a wide range of epitaxial strain if the monoclinic ground state were suppressed.

In summary, the experimentally observed field-induced ferroelectric transition corresponds to an intrinsic behavior of thin-film $ZrO₂$. The tetragonal-to-orthorhombic antiferroelectric transition is explained as a field-induced first-order transition where the polar mode is stabilized from the cubic phase by a coupling with two zone-boundary modes. Our results suggest that a ferroelectric phase could be favored over the antiferroelectric phase by appropriate epitaxial strain or isovalent substitution of Zr for Hf. The absence of toxic elements and the compatibility of $ZrO₂$ with silicon make these results especially relevant for technological applications.

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